## Supplementary Information

## One-pot preparation of multicomponent photocatalyst with

## $(\mathrm{Zn}, \mathrm{Co}, \mathrm{Ni})(\mathrm{O}, \mathrm{S}) / \mathrm{Ga}_{2} \mathrm{O}_{3}$ nanocomposites to significantly enhance hydrogen production

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Fig. S1 Reusability performance of ZNC-10G nanocomposite for five-cycle photocatalytic HER

Table S1 EDS elemental analysis of ZNC-10G nanocomposite

| Elements | Atomic <br> percentages |
| :---: | ---: |
| Zn | $39.97 \%$ |
| Ni | $0.86 \%$ |
| Co | $1.01 \%$ |
| Ga | $2.82 \%$ |
| O | $28.73 \%$ |
| S | $26.61 \%$ |

The energy of single photon (Ep) with wavelength of $\lambda$ is calculated using the equation below:
$E p=h c / \lambda$
where h is Planck's constant ( $6.6 \times 10^{-34} \mathrm{~J} . \mathrm{s}$ ), c is the speed of light ( $3 \times 10^{8} \mathrm{~m} / \mathrm{s}$ ), and $\lambda$ is the wavelength of the incident light.

The total energy of incident light $(\mathrm{Et})$ is calculated using the equation below:
$E t=P A t$
where $P$ is the power density of the incident light $\left(\mathrm{W} / \mathrm{m}^{2}\right)$, A is the irradiation area $\left(\mathrm{m}^{2}\right)$, and t is the duration of the incident light irradiation (s).

The number of incident photons $(\mathrm{N})$ is determined by the following equation:
$N=E t / E p=P A t \lambda / h c$
The apparent quantum yield (AQY) is calculated by using the following equation:

$$
\begin{aligned}
A Q Y= & \frac{2 x \text { number of envolved hydrogen molecules }}{n u m b e r ~ o f ~ i n c i d e n t ~ p h o t o n s ~} \times 100 \% \\
& =\frac{2 n \text { Na hc }}{P A t \lambda} \times 100 \%
\end{aligned}
$$

where n is the amount of hydrogen evolved (mol), NA is Avogadro's constant ( $6.02 \times 1023 / \mathrm{mol}$ ).
Based on the formula, the calculated AQY was 10.4 \%.

