

Supplementary Information

Construction of three dimensional S, N co-doped ZIF-67 derivative assisted by PEDOT nanowires and its application in rechargeable Zn-air batteries

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Supplementary Figures

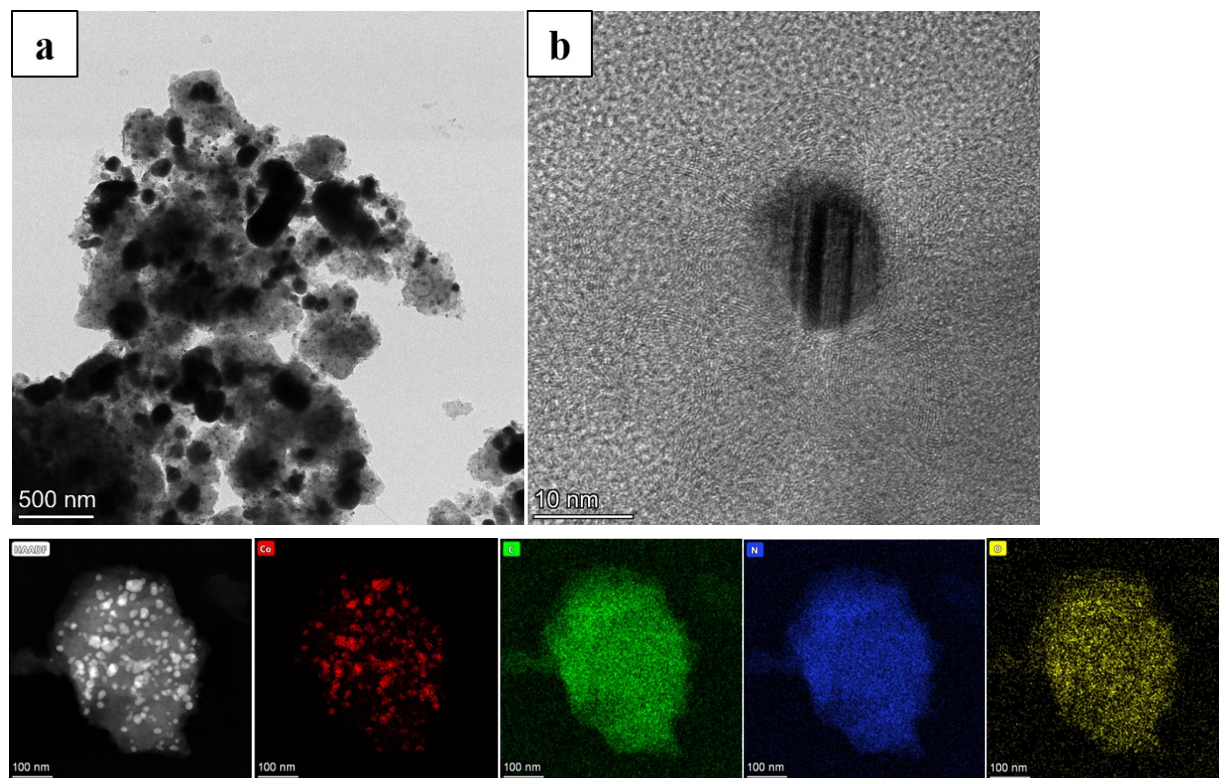


Fig. S1. TEM images (a, b) and elemental color mapping of the Co/C@N sample.

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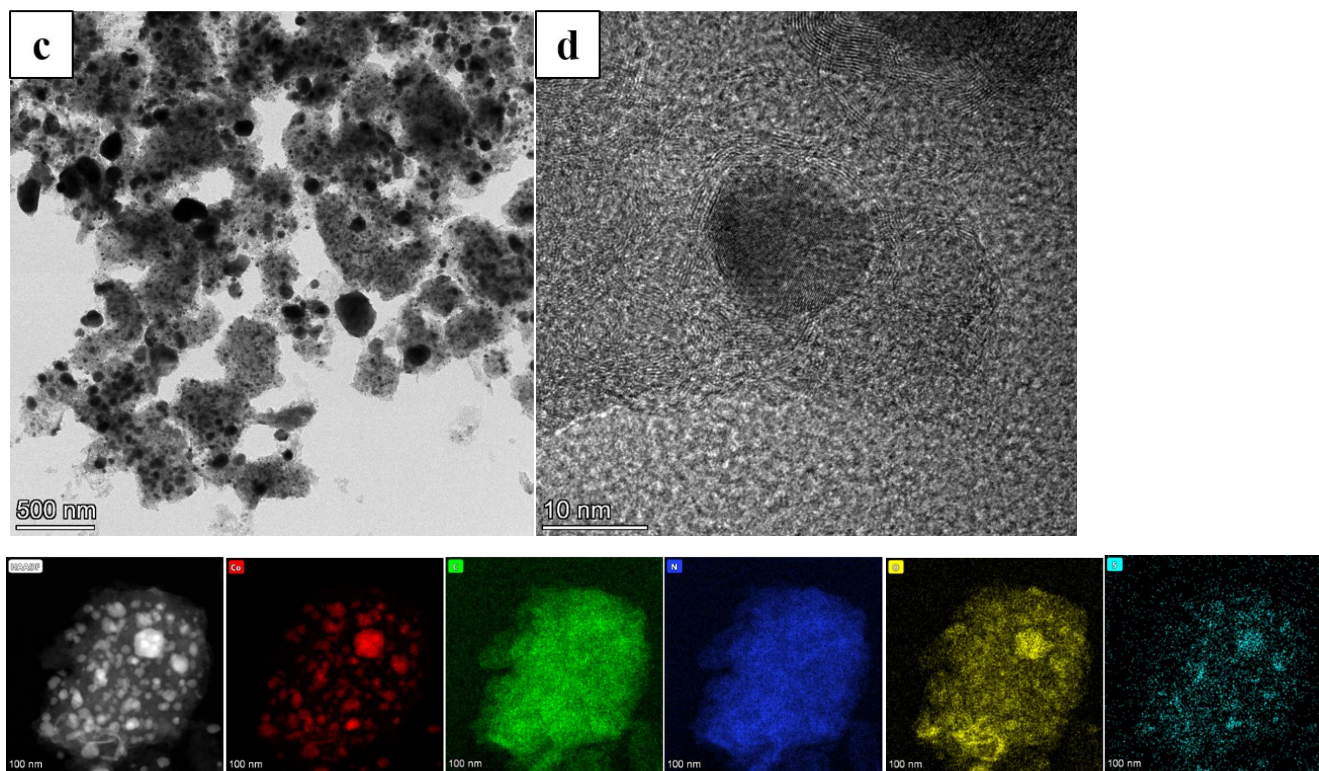


Fig. S2. TEM images (c, d) and elemental color mapping of the Co/C@NS sample.

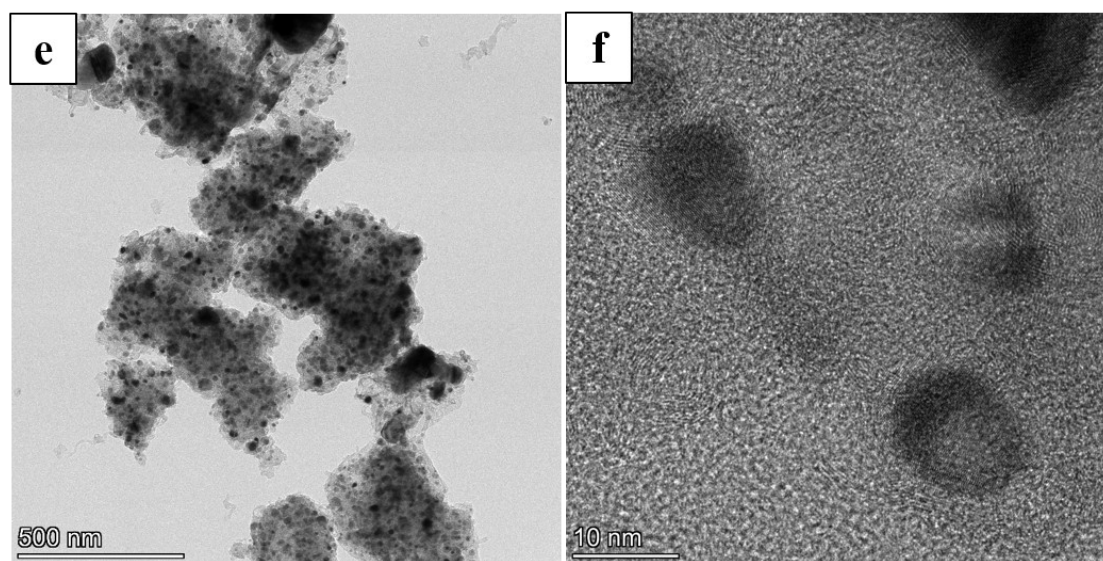


Fig. S3. TEM images for Co/C@NS NWs (e, f) catalysts.

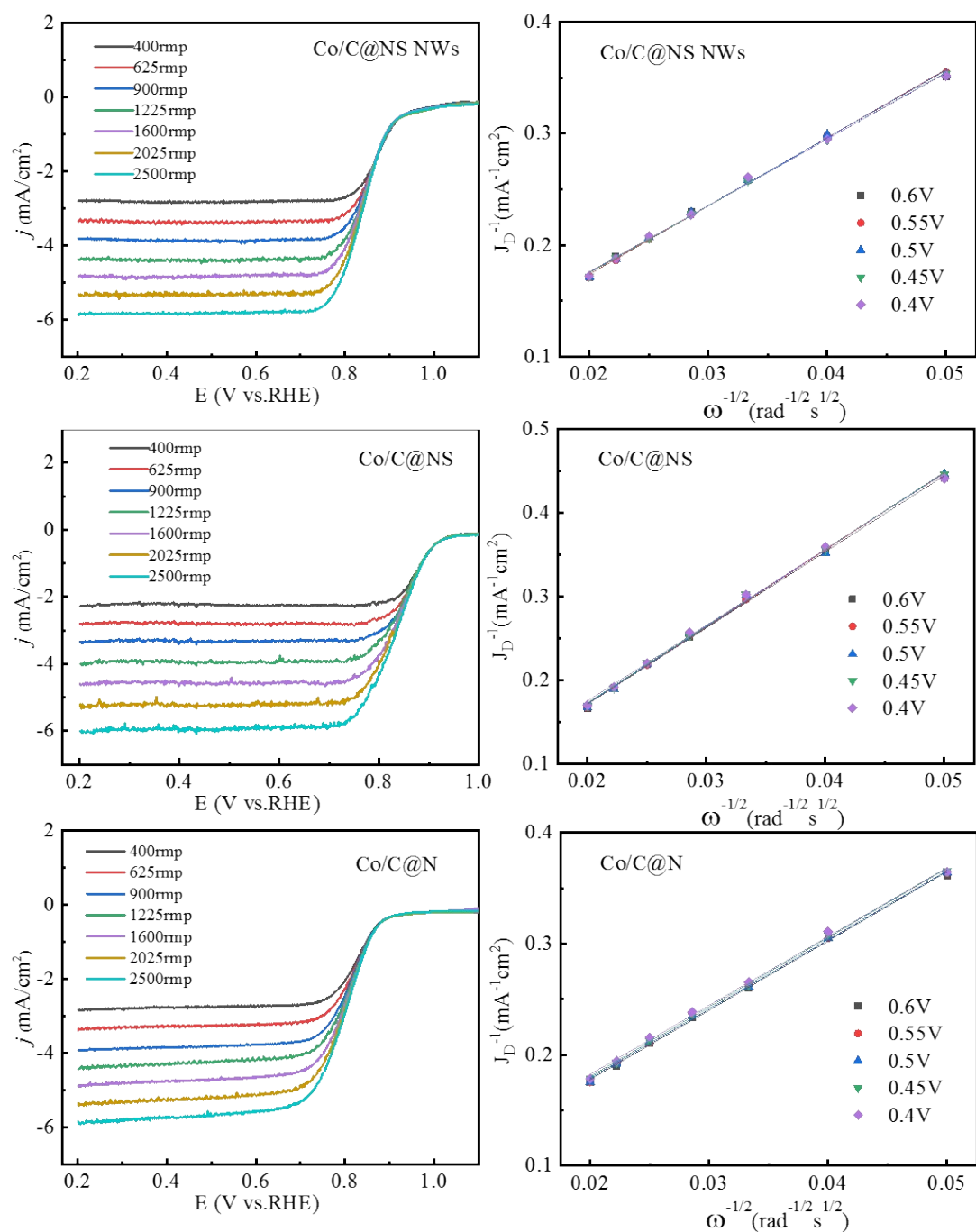


Fig. S4. RRDE curves of various catalysts at different rotation rates from 400 to 2500 rpm and the corresponding Koutecky-Levich (K-L) plots (j^{-1} vs. $\omega^{-1/2}$) of various catalysts between 0.40 V and 0.60 V.

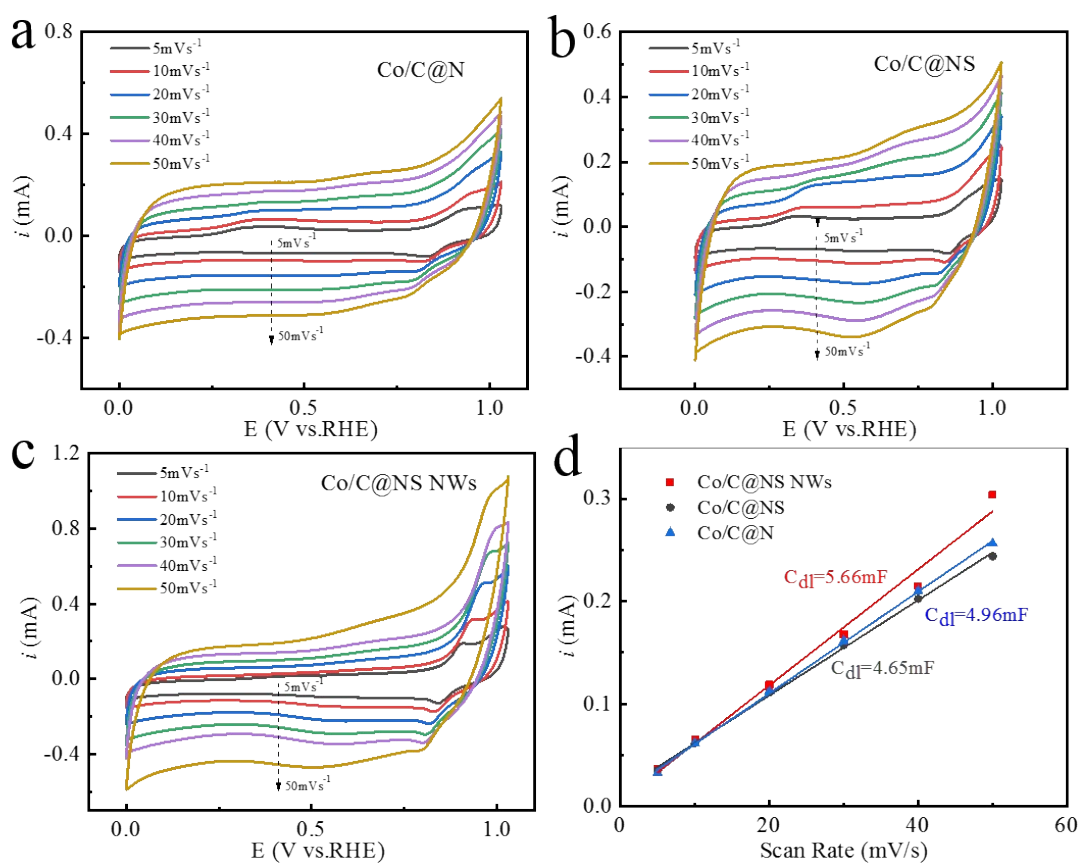


Fig. S5. (a - c) CV graphs of various catalysts measured at different scan rate from 5 to 50 mV s^{-1} . (d) Plots of the current versus the scan rate for Co/ C@N, Co/ C@NS and Co/ C@NS NWs.

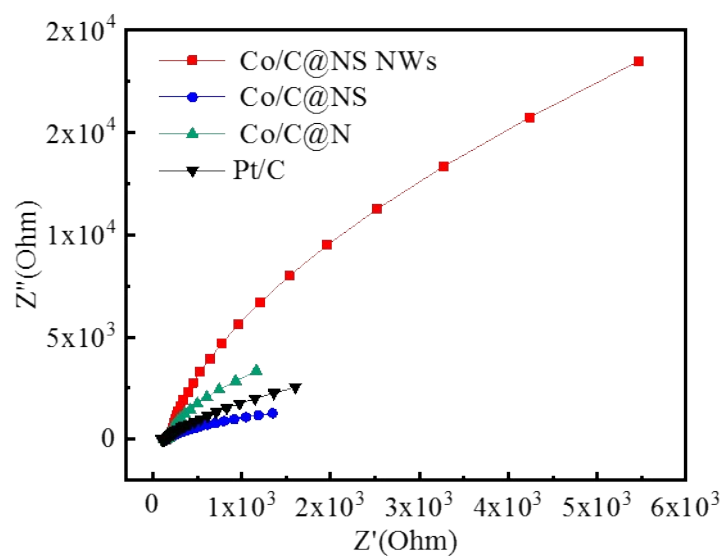


Fig. S6. EIS of various catalysts measured.

Supplementary table

Table S1. Comparison of the ORR/OER catalytic performances of Co/C@NS NWs with those of other reported works about ZIF-derived catalysts.

Sample	ORR			electron transfer number (n)	OER	Ref.
	E_{onset} vs. RHE (V)	$E_{1/2}$ vs. RHE (V)	J_L (mA cm ⁻²)		E (V) (J=10mA cm ⁻²)	
Co/ C@NS NWs	0.88	0.80	-4.8	3.86~3.94	1.4	this work
Co/ C@NS	0.87	0.81	-4.5	3.92-3.99	1.44	this work
Co/ C@N	0.86	0.79	-4.3	3.9-3.99	1.5	this work
Pt/C	0.92	0.81	-6	3.94-3.99	1.81	this work
Ag NWs-ZIF67	0.903	0.852	-	3.85	1.546	1
ZIF-67	0.75	0.66	4.32	3.87	1.68	2
Zn-Co-ZIF/GO-920	0.914	0.807	6.23	3.97-4.0	-	3
ZIF-67@NPC-2(2:1)	0.890	0.82	-4.94	4.03	1.64	2
Co₃N-PCL	0.863	0.822–0.846	4.64–5.22	3.97	-	4
P/Ni/Co/NC	0.872	0.775	5.02	-	-	5
Co-CNT	0.957	0.877	5.20	3.9	-	6
CS-HPCN1000-5	0.897	0.697	-3.75	3.9-4.0	-	7
CNT-900	0.934	0.819	4.98	3.99	-	8

Table S2. The performance of liquid rechargeable Zn-air batteries with various bifunctional electrocatalysts.

Sample	Open circuit potential (V)	Peak power density (mW.cm ⁻²)	Specific capacity at current density@J (mAh.g ⁻¹)	Electrolyte	Oxygen source	Ref.
Co/C@NS NWs	1.46	129.3	800@10	6 M KOH	air	This work
Co ₃ O ₄ -x/NG	1.49	166	700.6@10	6 M KOH	air	9
Co-NCNT/Ng-900	1.4	174.4	795.0@10	6 M KOH	air	9
CoDNG900	1.45	207.47	669 @10	6 M KOH	air	9
FeCo-N-C-700	1.39	150	518@10	6 M KOH	air	9
Co@hNCTs-800	1.45	149	746 @10	6 M KOH	air	9
NiCo _{2.148} O ₄ PNSs	1.46	83	700@10	6 M KOH	air	9
Co ₄ N@NC-2	1.48	74.3	769.4@5	6 M KOH	air	9
1-Pd/Co(OH) ₂	1.4	-	766@10	6 M KOH	air	9
CoNC (1:4)	1.33	126	852@10	6 M KOH	air	9
Co-MOF-800	1.38	144	671.6@10	6 M KOH	air	9
Co-MOF/LC-0.5 + 20Pt/C	1.44	126	-	6 M KOH+0.2M ZnCl ₂	air	10
MnO@Co-N/C	-	130.3	-	6M KOH+0.2 M Zn(Ac) ₂	air	11
SNC	1.24	94.8	-	Alkaline electrolyte	air	12

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