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## **Electronic Supplementary Information**

## Enhanced Photoelectrocatalytic performance of α-MnO<sub>2</sub> by Sb

## and N charge compensation

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Fig. S1 The SEM images of 2% (Sb,N)-MnO<sub>2</sub>(a), 4% (Sb,N)-MnO<sub>2</sub>(b) and 8% (Sb,N)-MnO<sub>2</sub>(c)

nanorods



Fig. S2 The enlarged XRD patterns of pure MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> with different doping ratios nanorods

	MnO <sub>2</sub>	2%	4%	6%	8%
	2	(Sb,N)-MnO <sub>2</sub>	(Sb,N)-MnO <sub>2</sub>	(Sb,N)-MnO <sub>2</sub>	(Sb,N)-MnO <sub>2</sub>
Group space	I4/m	I4/m	I4/m	I4/m	I4/m
Lattice Parameter a/Å	9.786	9.779	9.746	9.728	9.694
2 heta	37.666	37.582	37.547	37.502	37.456
FWHM	0.267	0.310	0.347	0.353	0.533
Crystal size/nm	33	28	25	24	16
BET surface area/m <sup>2</sup> ·g <sup>-1</sup>	37.098	49.286	67.609	103.642	58.984
Pore Volume /cm <sup>3</sup> ·g <sup>-1</sup>	0.114	0.161	0.2603	0.452	0.189

Table S1 Group space, lattice parameter, crystal size, BET surface area and pore volume of pureMnO2 and (Sb,N)-MnO2 with different doping ratios nanorods



Fig. S3 XPS survey spectra of 6% (Sb,N)-MnO<sub>2</sub> nanorods



Fig. S4 LSV curves of pure MnO<sub>2</sub>, Sb-MnO<sub>2</sub>, N-MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> electrodes



Fig. S5 CV curves of pure MnO<sub>2</sub>, Sb-MnO<sub>2</sub>, N-MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> electrodes



Fig. S6 EIS curves of pure MnO<sub>2</sub>, Sb-MnO<sub>2</sub>, N-MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> electrodes



Fig. S7 Transient photocurrent response of pure MnO<sub>2</sub>, Sb-MnO<sub>2</sub>, N-MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> electrodes



**Fig. S8** Relationship curves of photocatalytic time and the MB degradation efficiency of pure MnO<sub>2</sub>, Sb-MnO<sub>2</sub>, N-MnO<sub>2</sub> and (Sb,N)-MnO<sub>2</sub> nanorods



Fig. S9 UV-Vis absorbance spectra (a)and plots of  $(ahv)^{1/2}$  vs photon energy (b) of different ratios of pure MnO<sub>2</sub> and different ratios of (Sb,N)-MnO<sub>2</sub>