

Supplementary Information

Urchin-like NiCo₂O₄ hollow microspheres with oxygen vacancies synthesized by self-template for supercapacitors

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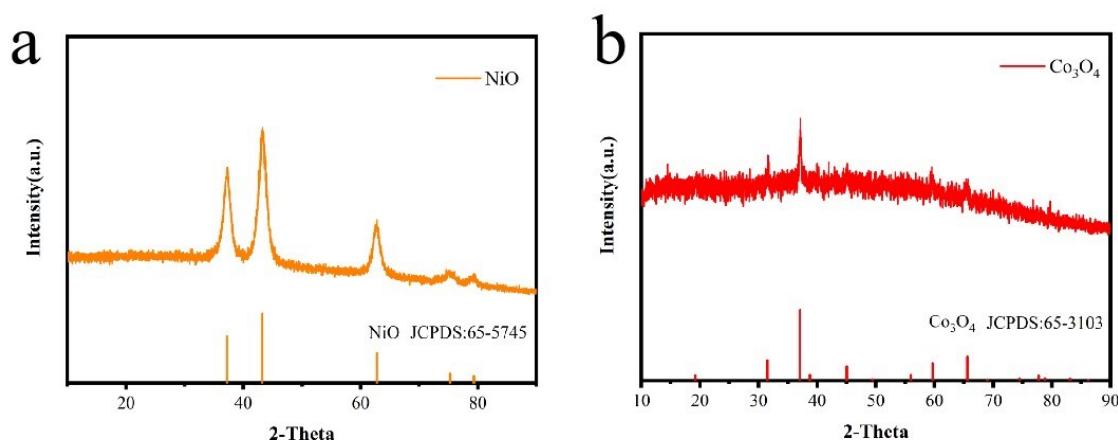


Figure S1. XRD curves of NiO (a) and Co₃O₄ (b).

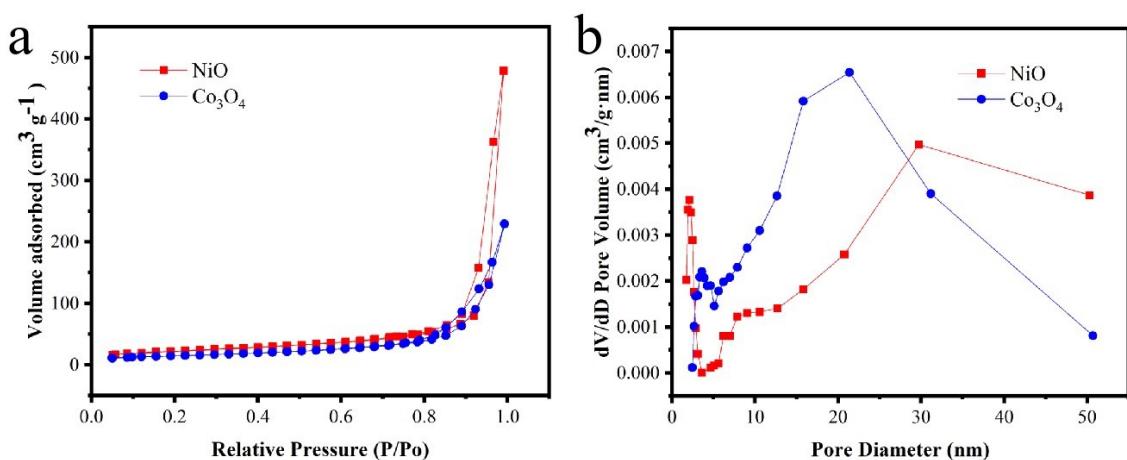


Figure S2. (a) N₂ adsorption /desorption curves and (b) pore size distribution curves

of NiO and Co₃O₄.

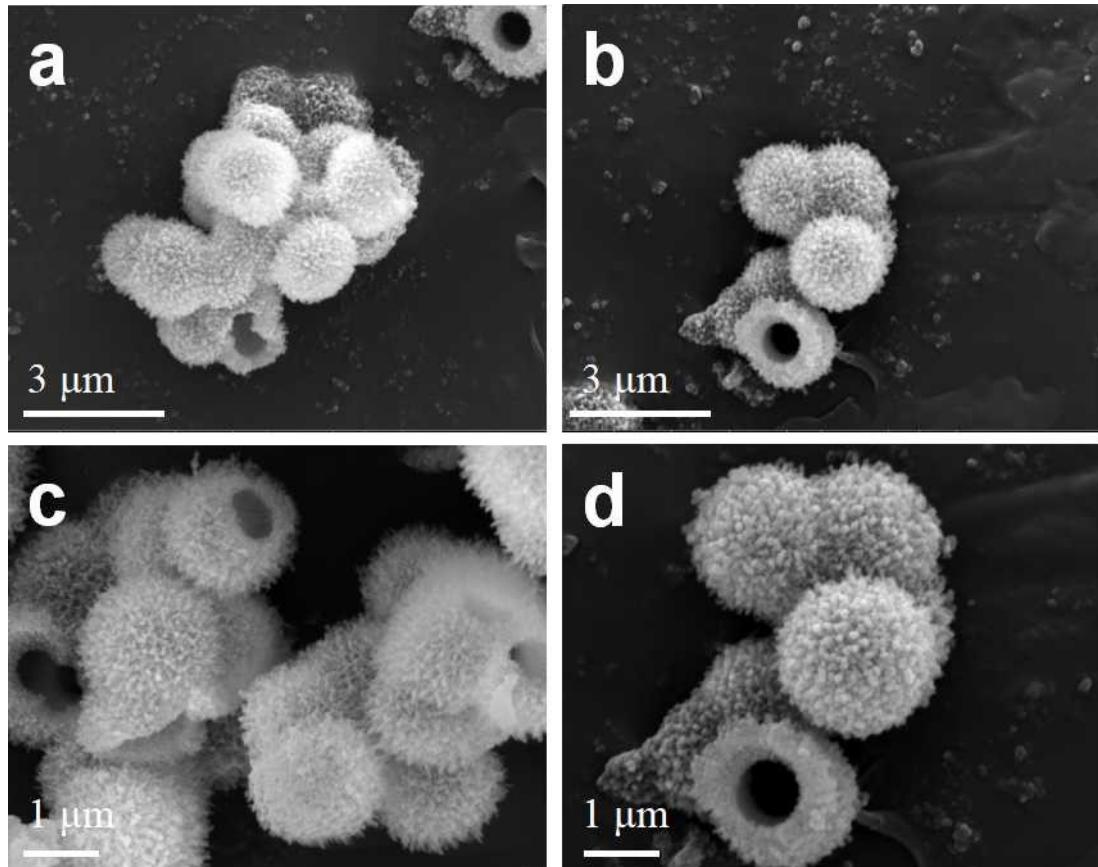


Figure S3. SEM image of NiO (a,c) and Co₃O₄ (b,d).

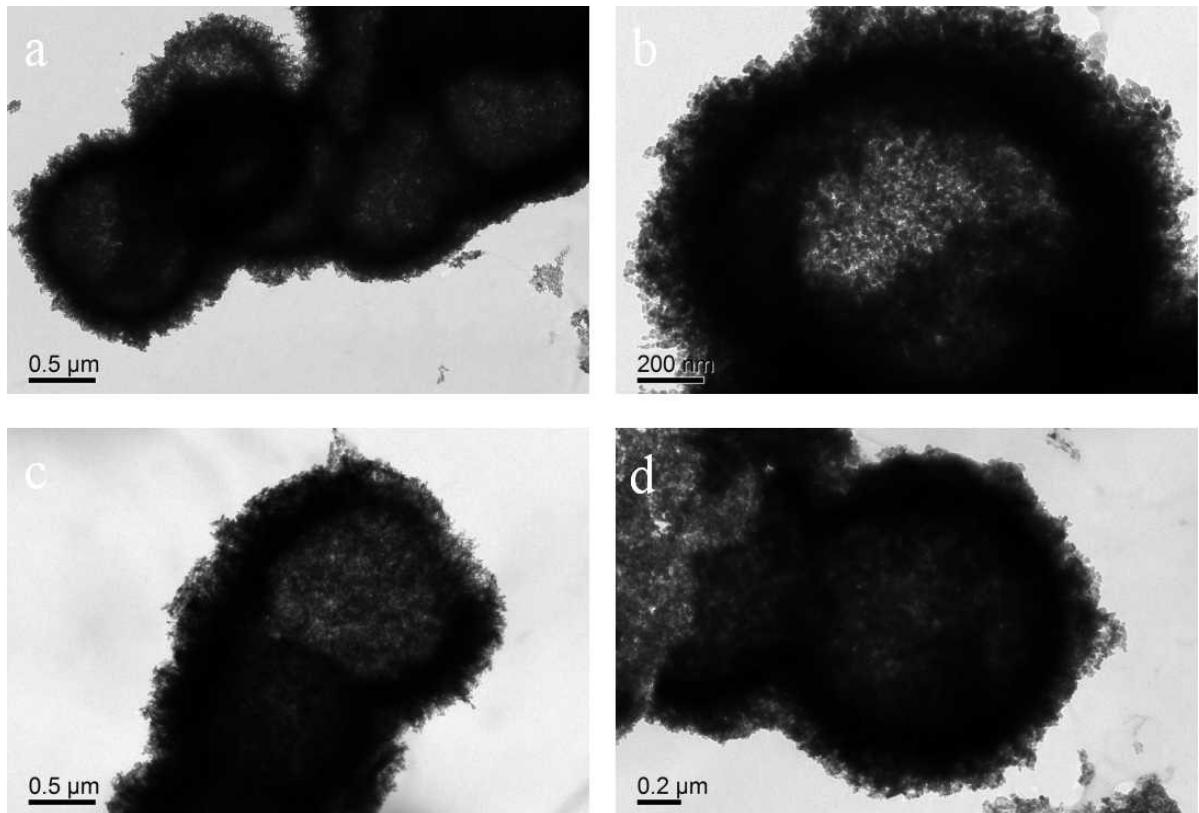


Figure S4. TEM image of NiO (a-b) and Co₃O₄ (c-d).

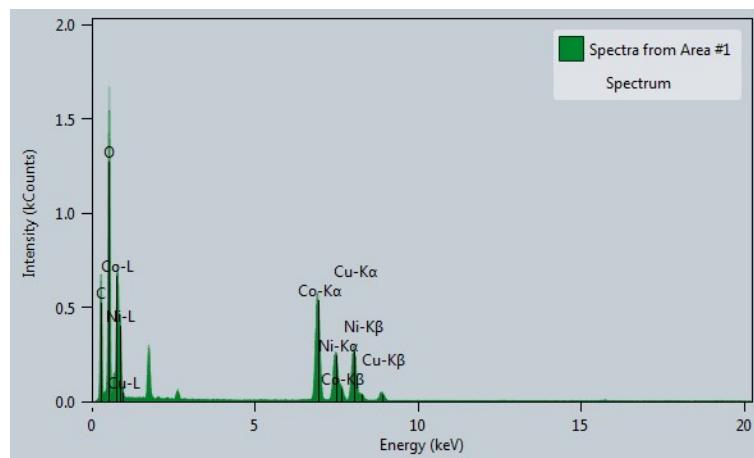


Figure S5. Energy dispersive X-Ray (EDX) spectrum of H-NiCo₂O₄.

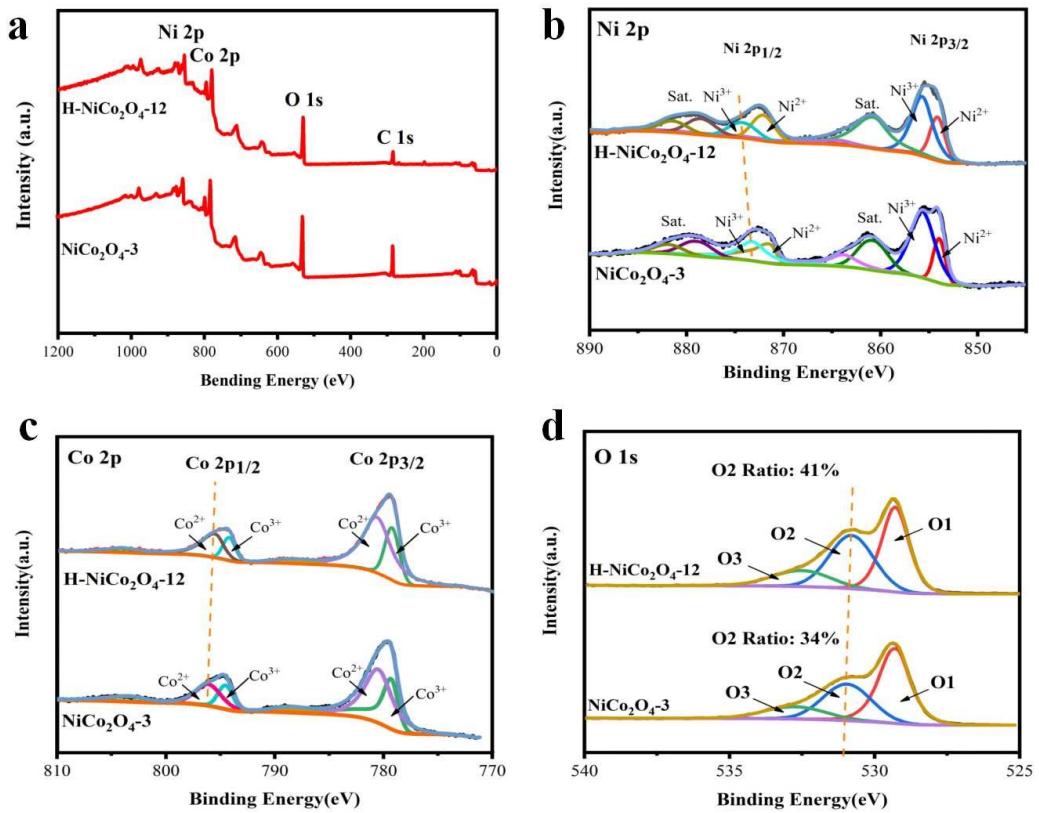


Figure S6. XPS full spectrum (a), high-resolution Ni 2p (b), Co 2p (c) and O 1s (d) of H-NiCo₂O₄-12 and NiCo₂O₄-3.

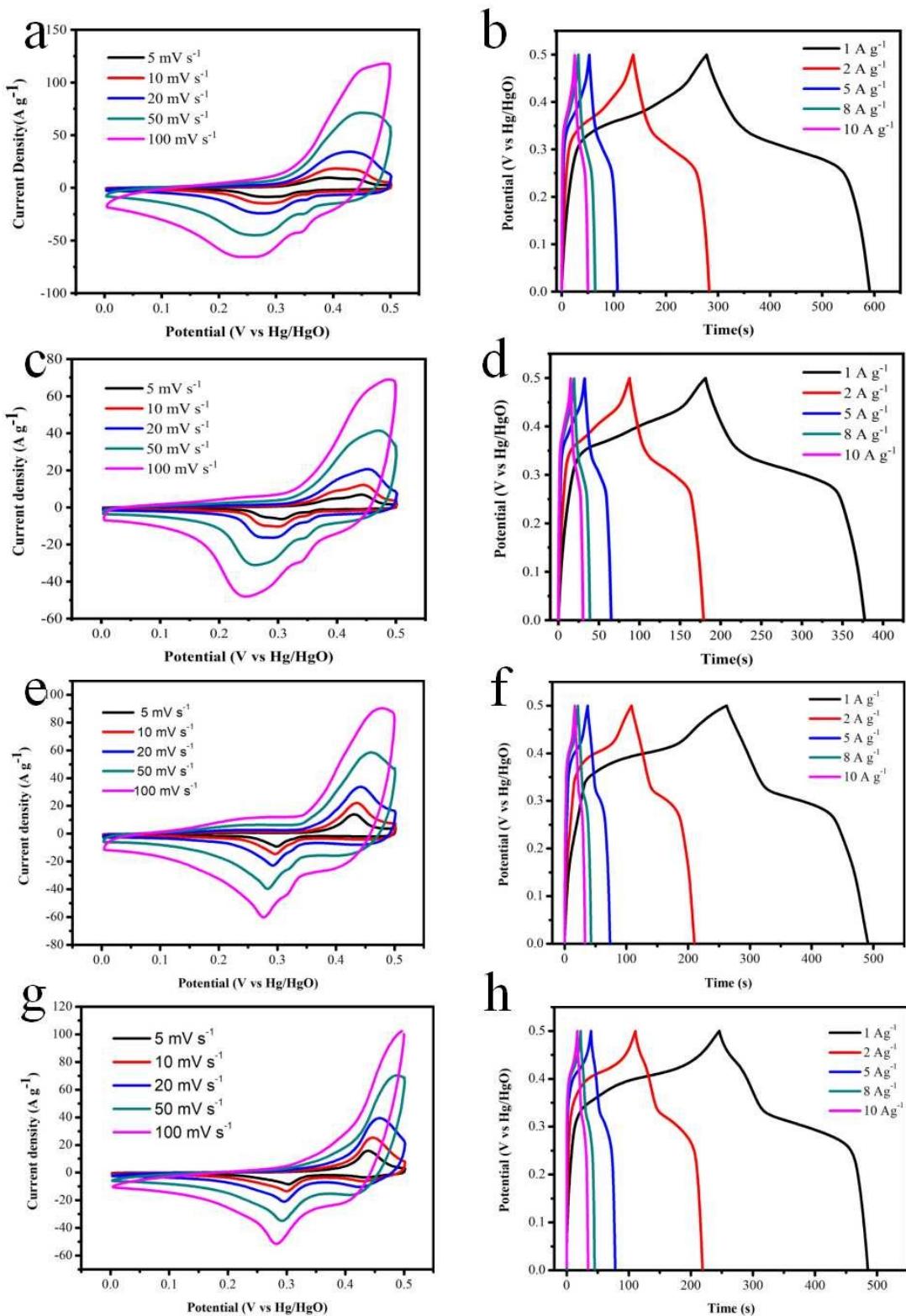


Figure S7. CV curves at various scan rates and GCD curves under different current densities of $\text{NiCo}_2\text{O}_4\text{-3}$ (a, b), $\text{NiCo}_2\text{O}_4\text{-1}$ (c, d), NiO (e, f) and Co_3O_4 (g, h), respectively.

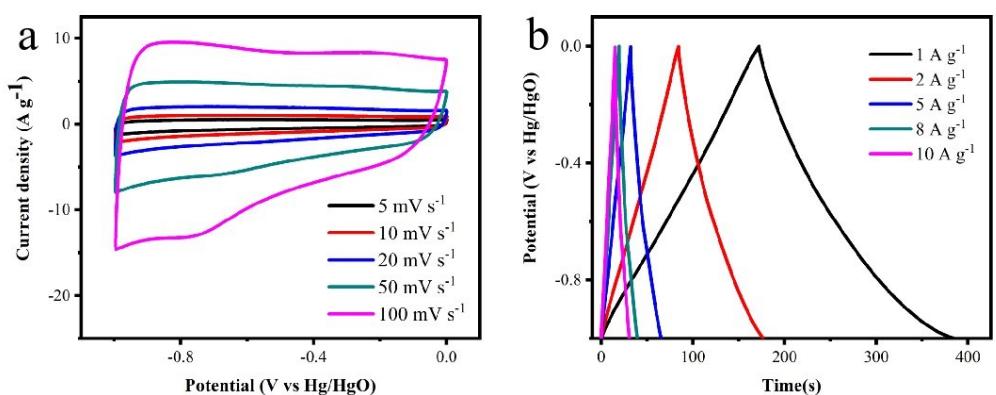


Figure S8. (a) CV curves at various scan rates and (b) GCD curves under different current densities of activated carbon.

Table S1 Comparison of specific capacitance between H-NiCo₂O₄ and reported NiCo₂O₄-based electrodes.

Sample	Specific capacitance	Rate		Cycle Stability	Ref.
		Capability			
NiCo ₂ O ₄ nanoflowers	702 F g ⁻¹ (5mV s ⁻¹)	-----		94.2% (5000 C)	¹
YS-NiCo ₂ O ₄	835.7 F g ⁻¹ (0.5 A g ⁻¹) 0 A g ⁻¹)	64 % (1 to 2	70.5 % (10000 C)		²
NiCo ₂ O ₄ @ CNFs	540 F g ⁻¹ (1A g ⁻¹) A g ⁻¹)	46 % (1 to 7	93.1 % (6000 C)		³
PNCO@FSSM	530 F g ⁻¹ (6 mA cm ⁻²)	-----	90.5 % (3000 C)		⁴
3D rGN/NiCo ₂ O ₄	708.36 F g ⁻¹ (1A g ⁻¹) A g ⁻¹)	82% (1 to 16	94.3 % (6000 C)		⁵
honeycomb-like NiCo ₂ O ₄ @NF	646.6 F g ⁻¹ (1A g ⁻¹) A g ⁻¹)	68% (1 to 9	96.5% (3000 C)		⁶
NiCo ₂ O ₄	725.7 F g ⁻¹ (1A g ⁻¹) 10 A g ⁻¹)	48.5% (1 to	70 % (5000 C)		⁷
H-NiCo ₂ O ₄	862 F g ⁻¹ (1A g ⁻¹) o 10 A g ⁻¹)	74.2% (1 t	80% (5000 C)	This work	

References

1. R. B. Waghmode, N. C. Maile, D. S. Lee and A. P. Torane, *Electrochim. Acta*, 2020, **350**, 136413.
2. L. Wang, X. Y. Jiao, P. Liu, Y. Ouyang, X. F. Xia, W. Lei and Q. L. Hao, *Appl. Surf. Sci.*, 2018, **427**, 174–181.
3. M. H. El-Shafei, A. G. El-Deen and A. Abd El-Moneim, *J Mater Sci: Mater Electron*, 2021, **32**, 15882–15897.

4. G.P. Kamble, A.S. Rasal, S.A. Mane, R.A. Chavan, J.-Y. Chang, Y.-C. Ling, S.S. Kolekar and A.V. Ghule, *RSC Adv.*, 2021, **11**, 3666-3672.
5. Y. Zhou, H. J. Liao, J. Li, H. X. Wang and Y. Wang, *Appl. Surf. Sci.*, 2021, **534**, 147598.
6. D. R. Kumar, K. R. Prakasha, A. S. Prakash and J. J. Shim, *J. Alloys Compd.*, 2020, **836**, 155370.
7. Q. Y. Wang, J. Y. Zhao, C. Zhang, X. Luo, J. Shao, M. Zhong, Z. H. Ye, P. D. Feng, X. L. Liu, K. Li and W. W. Zhao, *J. Alloys Compd.*, 2021, **852**, 156613.