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Supporting Information

Transition metal phosphide of nickel and cobalt modified $Zn_{0.5}Cd_{0.5}S$ for efficient photocatalytic hydrogen evolution with visible light irradiation

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Photocatalyst	Electron donor	amount of catalyst (mg)	Light Power	H ₂ evolution (μmol·h ⁻¹)	Ref.
ZnCdS/NiCoP	Na ₂ S/Na ₂ SO ₃	10 mg	5 W (LED)	12.80	This work
NP-CN	CH ₃ OH	50 mg	$\lambda > 420 \text{ nm} (Xe)$	75.00	[1]
g-C ₃ N ₄ -Co ₂ P	TEOA	50 mg	$\lambda > 420 \text{ nm} (\text{Xe})$	28.00	[2]
$Co_2P/ZnIn_2S_4$	Na ₂ S/Na ₂ SO ₃	5 mg	$\lambda > 420 \text{ nm} (\text{Xe})$	39.65	[3]
Co ₂ P/S-CN	TEOA	2 mg		0.40	
Fe ₂ P/S-CN	TEOA	2 mg	$\lambda > 420 \text{ nm} (Xe)$	0.32	[4]
Ni ₂ P/S-CN	TEOA	2 mg		0.42	
NixCo1-xP/rGO/CN	Poly Lactic	10 mg	$\lambda > 420 \text{ nm} (Xe)$	5.80	[5]
MOF-Fe-Ni-P	TEOA	5 mg	$\lambda > 420 \text{ nm} (Xe)$	27.10	[6]
CoP/g-C ₃ N ₄	Methanol	50 mg	UV - Vis (Xe)	52.00	[7]
CoP/BP	Na ₂ S/Na ₂ SO ₃	10 mg	$\lambda > 420 \text{ nm} (Xe)$	2.00	[8]
NiP/Cd _{0.5} Zn _{0.5} S	Na ₂ S/Na ₂ SO ₃	50 mg	$\lambda > 420 \text{ nm} (Xe)$	65.00	[9]
NiO/Ni ₂ P/CN	TEOA	5 mg	$\lambda > 420 \text{ nm} (Xe)$	0.50	[10]

 Table S1. This experiment is compared with the hydrogen production of the previously reported Metal phosphating compounds.

2.



Fig. S1. Hydrogen production rates with different photocatalyst.

1.

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