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## Scalable Manufacturing of Flexible and Highly Conductive

## Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>/PEDOT:PSS Thin Films for Electromagnetic Interference Shielding

Supplementary data



**Figure S1.** Sample mass spectrometry analysis of  $Ti_3C_2T_x$ 



**Figure S2.** Cross-section SEM images of (a) the pristine  $Ti_3C_2T_x$  and  $_{Ti3C2Tx}$ /PEDOT:PSS composite films containing (b) 10, (c) 25, (d) 35, and (e) 50 wt% of the polymer (scale is 2  $\mu$ m). The large vacancies shown in Figures S3a-b were observed everywhere in the dried film.



Figure S3. Reflectance and absorbance of pristine  $Ti_3C_2T_x$  and the composite films that contained various concentrations of polymer.



**Figure S4.** XRD patterns of (a) pure  $Ti_3C_2T_x$  and the composite films that contained (b) 25 and (c) 50 wt.% of PEDOT:PSS. (d) Compression between the 002 peaks of the pristine  $Ti_3C_2T_x$  and the composite films that contained various concentrations of polymer.



**Figure S5.** XPS results for the composite film containing 25 wt.% of PEDOT:PSS before and after post-treatment with sulfuric acid and methanol.

**Table S1.** Previous studies that used drop-casting or inkjet printing approach for PEDOT:PSS and its composites.

Material	Substrate	Method	Reference
Cu-Bi <sub>0.5</sub> Sb <sub>1.5</sub> Te <sub>3</sub> /PEDOT:PSS	Silicon dioxide (Activated by	Drop-casting	1
	plasma)		
PEDOT:PSS	Glass (Activated by UV/ozone)	Drop-casting	2
PEDOT:PSS/ethanol, and Graphene dots (GDs)/ PEDOT:PSS/ethanol	Fluorine-doped SnO <sub>2</sub>	Drop-casting	3
SnSe/PEDOT:PSS/ dimethyl sulfoxide	Glass	Drop-casting	4
(DMSO)	01		5
PEDOT:PSS coated Te nanorod/ PEDOT:PSS	Glass	Drop-casting	6
Bi <sub>2</sub> Te <sub>3</sub> nanosheet/PEDOT: PSS	Glass	Drop-casting	7
PEDOT: PSS	Silicon	Drop-casting	/ 
PEDOT: PSS	Glass water	Drop-casting	8
PEDOT: PSS	Silicon dioxide (Activated by plasma)	Drop-casting	9
PEDOT: PSS	Glass	Drop-casting	10
DMSO/PEDOT:PSS	Polydimethylsiloxane (PDMS)	Drop-casting	11
PEDOT:PSS)/waterborne polyurethane	Glass or elastomeric substrates	Drop-casting	12
(WPU)/ Carbon nanocoils (CNCs)	(Activated by plasma)	1 0	
PEDOT:PSS/ single-walled carbon nanotubes	Polyimide	Drop-casting	13
(SWCNTs)	2	1 0	
DMSO/PEDOT:PSS and ethylene glycol (EG)/	Polypropylene, glass, and	Drop-casting	14
PEDOT:PSS	poly(ethyleneterephthalate)	1 0	
	(PET)		
PEDOT:PSS/SWCNTs/ DMSO	Glass	Drop-casting	15
Bi <sub>2</sub> Te <sub>3</sub> nanosheet/PEDOT: PSS	Glass	Drop-casting	16
Cellulose/PEDOT:PSS/DMSO/glycerol	Glass	Drop-casting	17
PEDOT: PSS	Glass	Drop-casting	18
PEDOT: PSS/DMSO	Glass (Activated by aqua regia)	Drop-casting	19
Graphene oxide (GO)/PEDOT:PSS/DMSO	Glass (Activated with a mixture	Drop-casting	20
	of concentrated sulfuric acid and	1 0	
	hydrogen peroxide)		
PEDOT:PSS/ CNTs	Polyimide	Drop-casting	21
PEDOT: PSS/DMSO	Glass	Drop-casting	22
PEDOT: PSS	Glass	Drop-casting	23
Ag <sub>2</sub> Se NW/PEDOT:PSS	Glass	Drop-casting	24
PEDOT:PSS/ GDs	Fluorine-doped SnO <sub>2</sub>	Drop-casting	25
PEDOT-PSS/SWCNTs	PET	Inkjet	26
		printing	
PEDOT-PSS/vegetal glycerol	Polyimide	Inkjet	27
	-	printing	
PEDOT:PSS/DMSO	Pre-treated polyethylene	Inkjet	28
	naphthalate	printing	
PEDOT:PSS	Polyimide (Activated by plasma)	Inkjet	29
	- · · · · · · /	printing	
PEDOT:PSS	PET	Inkjet	30
		printing	
PEDOT-PSS/ ethanol/ diethylene glycol	Polyimide (Activated by plasma)	Inkjet	31
		printing	

	Average static contact	Average advancing	Average receding	
	angle	contact angle	contact angle	
Before treatment	81.6	89.2	69.6	
After treatment for 30 min	24.2	28.6	19.9	

**Table S2.** Average static, advancing, and receding contact angles of water on the polycarbonate sheet.

**Table S3.** Average static, advancing, and receding contact angles of the aqueous solution of  $Ti_3C_2T_x$  or  $Ti_3C_2T_x$ /PEDOT:PSS (25 wt. %) on activated polycarbonate sheet.

	Avaraga static	Average	Average
	contact angle	advancing contact angle	receding contact angle
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	17.3	21.1	14.2
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /PEDOT:PSS (25 wt.%)	15.9	19.3	0

**Table S4.** Previous studies of thickness, conductivity,  $SE_T$ , and specific EMI shielding effectiveness of various nanocomposites.

Material	Thickness	Conductivity (σ	SE <sub>T</sub> (dB)	SE <sub>T</sub> /Density/thick	Reference
	(mm)	cm <sup>-1</sup> )		ness (dB cm <sup>2</sup> g <sup>-1</sup> )	
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /PEDOT:PSS	0.011	340.5	42.1	19498	32
Ti <sub>2</sub> CT <sub>x</sub> /Polyvinyl alcohol	0.1	$2.0 \times 10^{-4}$	26	5136	33
(PVA)					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /Cellulose nanofiber	0.035	1.43	39.6	7029	34
(CNF)					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /Silver nanowires	0.0169	300	42.74	16724	35
(AgNW)/ Nanocellulose					
(NC)					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /Aramid nanofibers	0.012	-	34.71	21971	36
(ANF)					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /PVA	0.027	7.16	44.4	9343	37
Polystyrene/Graphene	2.5	1.25	29	322	38
Poly(3-hydrobutyrate-co-3-	0.018	-	45.9	19678	39
hydroxyvalerate) (PHBV)/					
AgNW					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /Polyacrylonitrile	0.045	92.68	32	4085	40
(PAN)/TiO <sub>2</sub>					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /rGO/Epoxy	2	6.95	56.4	9400	41
Graphene/PEDOT:PSS	0.8	6.84	70	841	42
V <sub>2</sub> O <sub>5</sub> /Polyaniline (PANI)	6	0.016	34.7	2770.3	43
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /AgNW/CNF	0.035	373.78	59.7	10647	44
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /Xanthan	0.00684	115.29	34.1	14,490	45
Poly vinylidene fluoride	0.9	0.074	32	259	46
(PVDF)/MWCNT					
Graphene/ PMMA	0.24	3.11	19	1042	47

Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /PANI	0.376	3.25	35.3	1700	48
Poly(oxymethylene) (POM)/	2	3.33	58.6	344.4	49
MWCNT					
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /CNF	0.047	7.39	24	2647	50
MWCNT/Polycarbonate	2.16	3	39	164	51
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /PEDOT:PSS	0.007	$2900 \pm 400$	55.42	38079	This work

**Table S5.** Compositions of the  $Ti_3C_2T_x$  MXene and  $Ti_3C_2T_x$  MXene/PEDOT:PSS solutions used for drop-casting. One mL of each of the solutions was used for covering 9 cm<sup>2</sup> of polycarbonate sheet.

Sample name	PEDOT:PSS content (mg)	Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> MXene content (mg)	Water content (mL)
Pure Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	-	15	1
10 wt.%	1.5	13.5	
25 wt.%	3.75	11.25	
35 wt.%	5.25	9.75	-
50 wt.%	7.5	7.5	

## **Equations**

1. The EMI reflection loss is calculated using:

$$SE_R = 168 + 10\log\left(\frac{\sigma}{\mu f}\right) = 10\log\left(\frac{1}{1-R}\right)$$
 (S1)

where  $\sigma$  is the conductivity,  $\mu$  is the magnetic permeability of the shield relative to air, and f is the frequency in Hz and R is the reflectance.

2. The EMI absorption loss:

$$SE_A = 8.68t(\sqrt{\frac{\sigma\omega\mu}{2}}) = 8.68\frac{t}{\delta} = 10\log(\frac{1-R}{T})$$
 (S2)

where t is the thickness of the shield,  $\omega$  is the angular frequency in rad/s,  $\delta$  is the skin depth of the shield, and T is the transmittance.

3. Reflectance (R):

$$R = S_{11}^{2}$$
 (S3)

where  $S_{11}$  is the reflected voltage magnitude divided by the incident voltage magnitude in port 1.

4. Transmittance (T):

$$T = S_{21}^{2}$$
 (S4)

where  $S_{21}$  is the transmitted voltage magnitude from port 1 to port 2 divided by the incident voltage magnitude in port 1.

5. Absorbance (A):

$$A = 1 - R - T \tag{S5}$$

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