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Supporting Information

Effect of Preparation Conditions and Co-Pi Groups as Noble Metalfree Redox Mediator and Hole Extractor to Boost Photoelectrochemical Water Oxidation for 1D Nanorod α-Fe₂O₃

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Figure S1 SEM image of cross-section of Fe₂O₃ at 600°C calcination temperature photoanodes.



Figure S2 EDS scanning spectra of (a)α-Fe₂O₃ and (b)α-Fe₂O₃/Co-Pi photoanodes.

Table S1 Contents of corresponding elements in α -Fe₂O₃ and α -Fe₂O₃/Co-Pi photoanodes obtained from EDS.

Element	Wt%		At%	0
	a-Fe ₂ O ₃	α-Fe ₂ O ₃ /Co-Pi	a-Fe ₂ O ₃	α-Fe ₂ O ₃ /Co-Pi
ОК	6.21	5.97	21.29	20.71
FK	3.10	3.13	8.96	9.15
Si k	7.47	8.32	14.58	16.45
Sn K	54.03	57.17	24.95	26.73
Ca K	4.09	4.37	5.60	6.05
Fe K	25.08	21.04	24.62	20.91



Figure S3 TEM image with 50000x magnification of Fe_2O_3 (a) without or (b) with dashed frame. Inside the dashed frame was Fe_2O_3 with short nanorod morphology.



Figure S4 XRD for FTO substrate, Fe_2O_3 calcined at various temperatures of 500, 525, 550, 575, and 600°C, SnO_2 (black five-pointed star), α -Fe₂O₃(red snowflake).

Sample	Fe	22 O 3	Fe ₂ O ₃ /Co-Pi	
Name	Atomic %	PPAt. %	Atomic %	PP At. %
C 1s	36.67	41.8	35.76	44.28
Fe 2p	19.04	5.43	14.02	4.71
O 1s	44.28	52.77	43.46	48.54
Co 2p	0	0	4.74	1.09
Р 2р	0	0	2.02	1.38

Table S2 The content of each element in Fe_2O_3 and Fe_2O_3/Co -Pi photoanodes obtained from XPS.



Figure S5 UV-vis absorption spectra of Fe₂O₃ prepared with various hydrothermal time of 2, 4, 6, 8, and 10 h.



Figure S6 UV-vis absorption spectra of Fe₂O₃ prepared with various concentration of Fe(NO₃)₃·9H₂O precursor of 2, 4, 6, 8, 10, and 12 mM.



Figure S7 UV-vis absorption spectra of Fe₂O₃ calcined at various temperatures of 500, 525, 550, 575, and 600°C.



Figure S8 LSV curves of bare α-Fe₂O₃ prepared with various hydrothermal time of 2, 4, 6, 8, and 10 h. The dotted line represents the dark state current density.



Figure S9 LSV curves of bare α-Fe₂O₃ prepared with various concentration of Fe(NO₃)₃·9H₂O precursor of 2, 4, 6, 8, 10, and 12 mM. The dotted line represents the dark state current density.



Figure S10 PEC stability test of 5h of bare Fe₂O₃ and Fe₂O₃/Co-Pi.

Fig. S10 is the PEC stability test of 5h of bare Fe_2O_3 and Fe_2O_3/Co -Pi. It could be seen from the figure that after the PEC stability test for 5 hours at $1.23V_{RHE}$, the bare Fe_2O_3 and Fe_2O_3/Co -Pi photoanodes both maintained the photocurrent density of 80% of the initial value.



Figure S11 LSV curves of (a) bare Fe₂O₃ and (b) Fe₂O₃/CoPi without and with Na₂SO₃.



Figure S12 EIS Nyquist plots of Fe₂O₃ calcined at various temperatures of 500, 525, 550, 575, and 600°C at a bias of 1.23 V_{RHE} under simulated solar light illumination.



Figure S13 Digital photo of bare Fe₂O₃ calcined at 600°C.



Figure S14 EIS Nyquist plots for Fe_2O_3 and Fe_2O_3/Co -Pi photoanodes after immersion in phosphate for various time of 3, 6, 9, and 12 h at a bias of 1.23 V_{RHE} under simulated solar light illumination.