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## Supplementary information

## Investigation of enhanced electro-catalytic HER/OER performances of copper tungsten oxide@reduced graphene oxide nanocomposites in alkaline and acidic media

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## Supplementary data:

Figure S-1: XRD analysis of pure (a) CuWO<sub>4</sub> nanoparticles and (b) rGO sheets.

Figure S-2. EDAX studies equipped with TEM (a) without sample and (b) with sample.

Figure S-3. CV curves of the CuWO<sub>4</sub>@rGO nanocomposites in 0.5M KOH for HER.

Figure S-4. CV curves of the CuWO<sub>4</sub>@rGO nanocomposites in 0.5M H<sub>2</sub>SO<sub>4</sub> for HER.

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Figure S-7. Cyclic stability of the CuWO<sub>4</sub>@rGO electrodes in 0.5M KOH at 50 mV/s.

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Table S-1: Experimental details and results of the synthesized CuWO<sub>4</sub>@rGO nanocomposites for HER

Table S-2: Experimental details and results of the synthesized CuWO<sub>4</sub>@rGO nanocomposites for OER



Figure S-1: XRD analysis of pure (a) CuWO<sub>4</sub> nanoparticles and (b) rGO sheets.



Figure S-2. EDAX studies equipped with TEM (a) without sample and (b) with sample.



Figure S-3. CV curves of the CuWO<sub>4</sub>@rGO nanocomposites in 0.5M KOH for HER.



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Figure S-5. CV sweeps of the CuWO<sub>4</sub>@rGO nanocomposites in 0.5M KOH for OER.



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Figure S-7. Cyclic stability of the CuWO<sub>4</sub>@rGO electrodes in 0.5M KOH at 50 mV/s.



Figure S-8. Plot for energy loss during CA measurements at fixed anodic potentials with time.



Figure S-9. (a) FTIR, (b) SEM and (c) TEM analysis of  $CuWO_4@rGO$  nanocomposites after electro-catalysis for OER in alkaline electrolyte.

Table S-1: Comparison of present experimental details and results of the synthesized CuWO<sub>4</sub>@rGO nanocomposites for HER against previous reports.

| Electrolyte                             | Working Electrode   | Reference<br>Electrode | Tafel               | References  |
|---|---|------------------------|---------------------|---|
| 0.5М<br>КОН                             | Glassy Carbon with<br>CuWO <sub>4</sub> @rGO<br>nanocomposites      | Ag/AgCl                | ~212 mV/dec         | Present work  |
| 0.5M<br>H <sub>2</sub> SO <sub>4</sub>  | Glassy Carbon with<br>CuWO <sub>4</sub> @rGO<br>nanocomposites      | Ag/AgCl                | ~192 mV/dec         | Present work  |
| 0.5M<br>H <sub>2</sub> SO <sub>4</sub>  | Glassy Carbon with<br>ZnMoO <sub>4</sub>                            | Ag/AgCl                | ~230 mV/dec         | Materials Letters, 2021, 284, 128996  |
| 0.1 M<br>KOH                            | Glassy Carbon with<br>NiWO <sub>4</sub> @rGO                        | Ag/AgCl                | ~250 mV/dec         | Materials Letters, 2019, 240, 51-54   |
| 0.1 M<br>KOH                            | Glassy Carbon with<br>NiWO <sub>4</sub>                             | Ag/AgCl                | ~475 mV/dec         | Materials Letters, 2019, 240, 51-54   |
| 1 М КОН                                 | Screen-printed<br>carbon electrodes<br>with SrWO <sub>4</sub>       | Hg/HgO                 | 138 – 167<br>mV/dec | Journal of the Taiwan<br>Institute of Chemical<br>Engineers, 2021, 126, 145-<br>153 |
| 0.5 M<br>H <sub>2</sub> SO <sub>4</sub> | Carbon paper with<br>commercial CoWO4<br>and CoWO <sub>4-x</sub> @C | Hg/HgO                 | 225 – 254<br>mV/dec | AppliedCatalysisB:Environmental,2019,259,118090                                     |

Table S-2: Comparison of present experimental details and results of the synthesized  $CuWO_4@rGO$  nanocomposites for OER against previous reports.

| Electrolyte                            | Working Electrode  | Reference<br>Electrode | Tafel       | References  |
|--|--|------------------------|-------------|---|
| 0.5M<br>KOH                            | Glassy Carbon with<br>CuWO <sub>4</sub> @rGO<br>nanocomposites                           | Ag/AgCl                | ~110 mV/dec | Present work  |
| 0.5M<br>H <sub>2</sub> SO <sub>4</sub> | Glassy Carbon with<br>CuWO <sub>4</sub> @rGO<br>nanocomposites                           | Ag/AgCl                | ~315 mV/dec | Present work  |
| 1.0 M<br>KOH                           | Glassy Carbon with CuWO <sub>4</sub> nanoparticles                                       | Ag/AgCl                | ~190 mV/dec | ChemElectroChem, 2018, 5, 3938-3945   |
| 0.5 M<br>KOH                           | Glassy Carbon with<br>Cu-Ni@rGO<br>nanoparticles   | Ag/AgCl                | ~190 mV/dec | Materials Letters, 2020, 260, 126969  |
| 0.5 M<br>KOH                           | Glassy Carbon with<br>Cu-Ni nanoparticles  | Ag/AgCl                | ~378 mV/dec | Materials Letters, 2020, 260, 126969  |
| 1 М КОН                                | Screen-printed<br>carbon electrodes<br>with SrWO <sub>4</sub>                            | Hg/HgO                 | ~218 mV/dec | Journal of the Taiwan<br>Institute of Chemical<br>Engineers, 2021, 126, 145-<br>153 |
| 1 М КОН                                | Ni Foam with<br>NiCo <sub>2</sub> O <sub>4</sub> @NiWO <sub>4</sub>                      | Ag/AgCl                | ~217 mV/dec | International Journal of<br>Hydrogen Energy, 2019, 44,<br>2883-2888                 |
| 1 М КОН                                | Glassy carbon<br>electrode with<br>Bi <sub>2</sub> WO <sub>6</sub> /TiO <sub>2</sub> -Ag | Ag/AgCl                | ~231 mV/dec | Applied Surface Science, 2021, 569, 150918  |