

## **Ni(OH)<sub>2</sub> Coated CoMn-Layered Double Hydroxide Nanowires as Efficient Water Oxidation Electrocatalysts**

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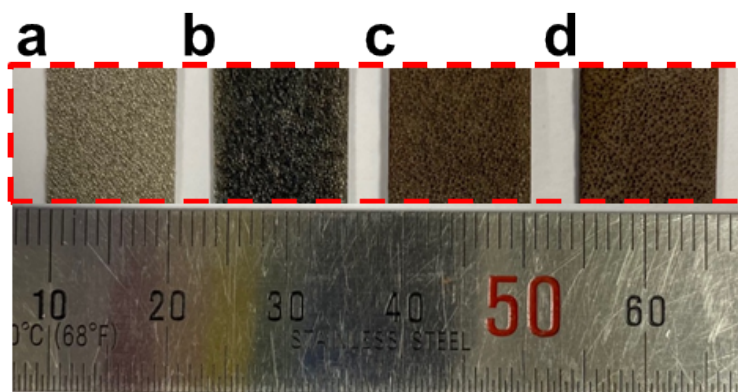
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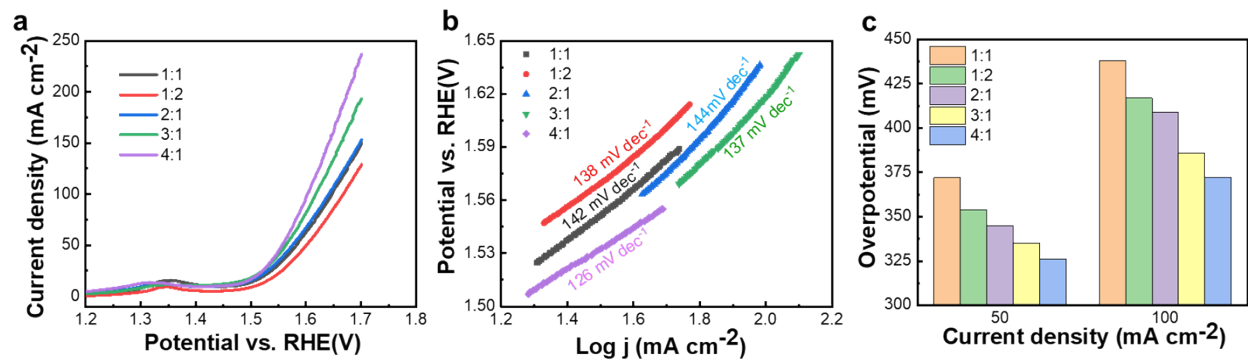
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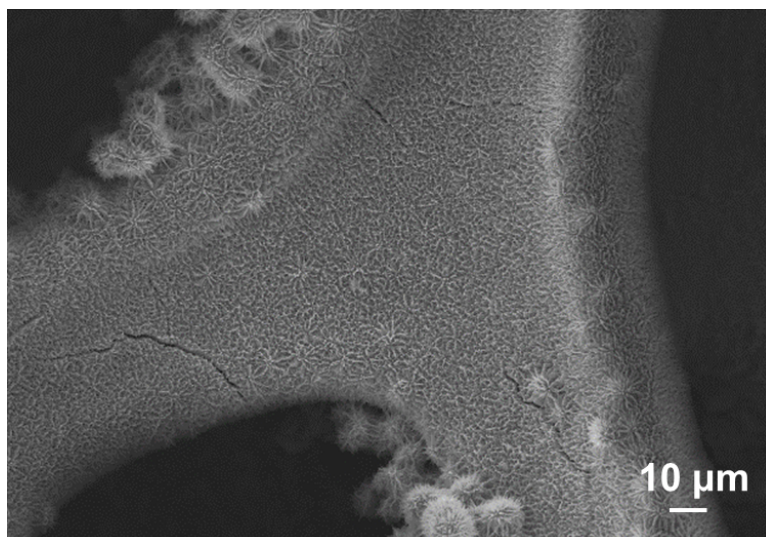
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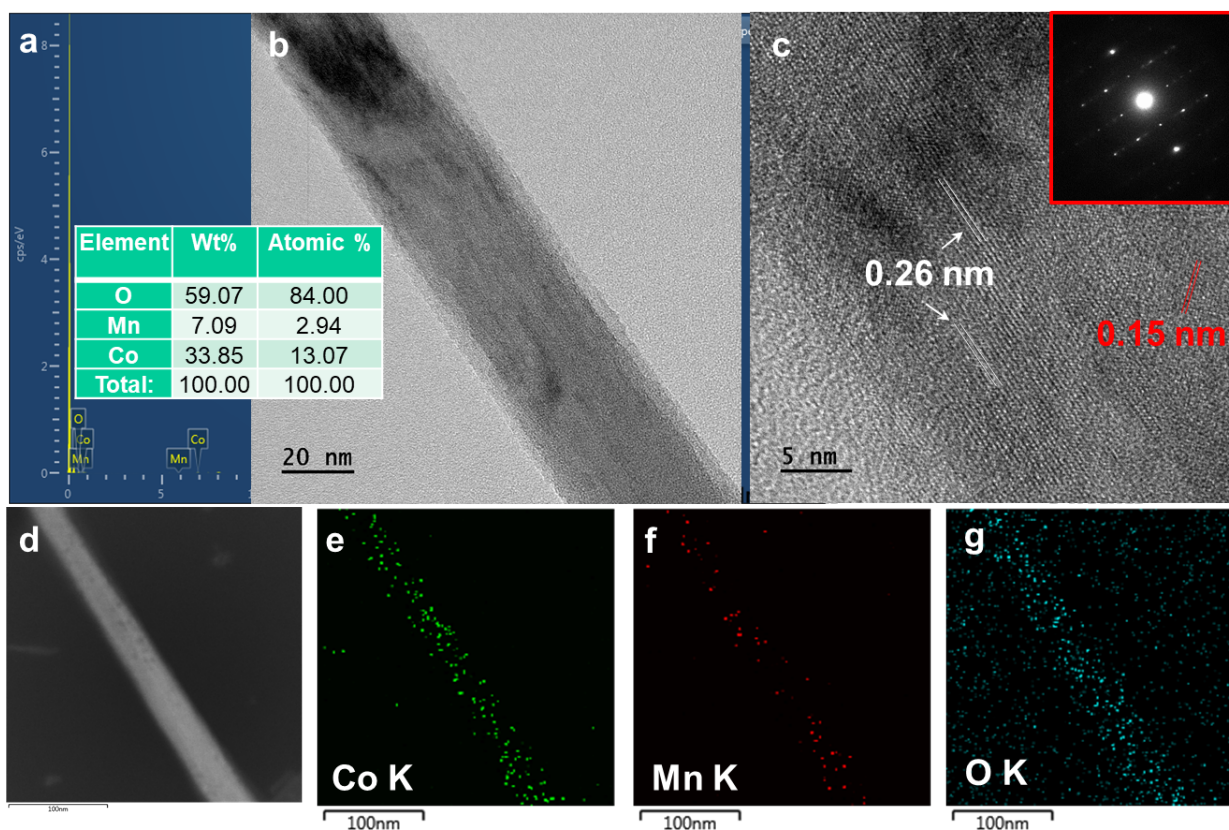
**Fig.S1** Optical photographs of catalysts (a) bare NF, (b) Ni(OH)<sub>2</sub>/NF, (c) CoMn-LDH/NF, and (d) Ni(OH)<sub>2</sub>@CoMn-LDH/NF.



**Fig. S2** (a) OER polarization curves at a scan rate of 1 mV s<sup>-1</sup>, (b) Corresponding Tafel plots, (c) Overpotential values at the current density of 50 and 100 mA cm<sup>-2</sup> with various molar ratios of Co: Mn of CoMn-LDH in 1 M KOH.



**Fig. S3.** FE-SEM image of Ni(OH)<sub>2</sub>@CoMn-LDH nanowires at low magnification.



**Fig. S4** (a) EDS spectrum (b) TEM image (c) HRTEM image and inset is SAED pattern (d-g) Elemental mapping images for CoMn-LDH nanowires.

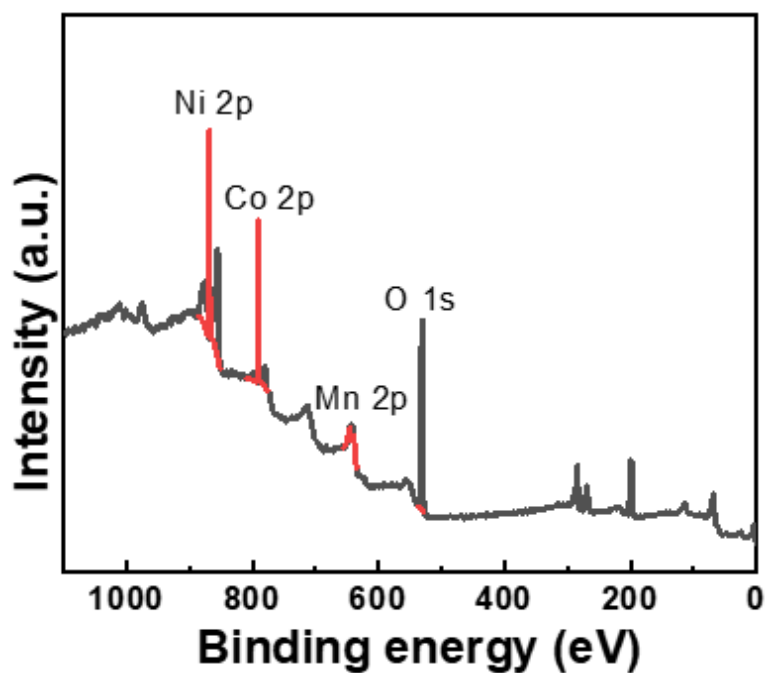


Fig. S5 XPS survey spectrum of Ni(OH)<sub>2</sub>@CoMn-LDH/NF.

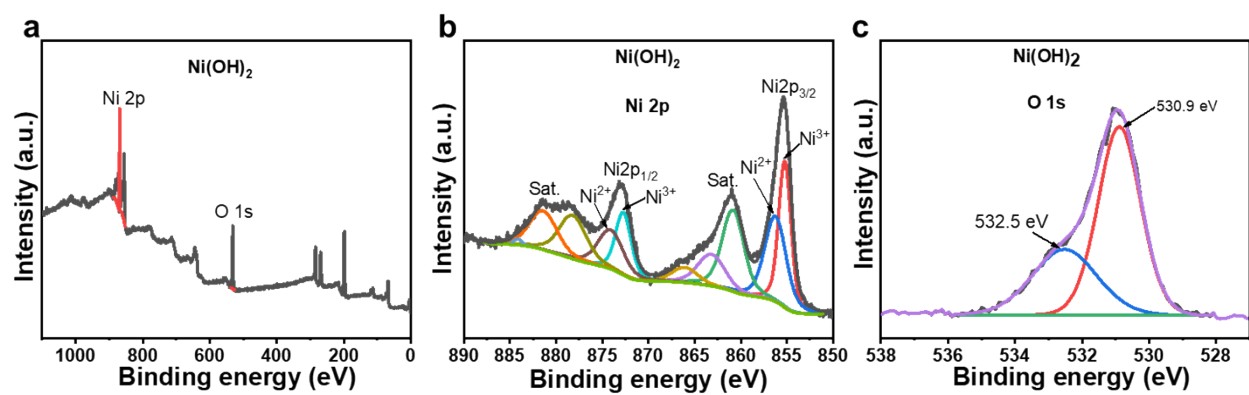
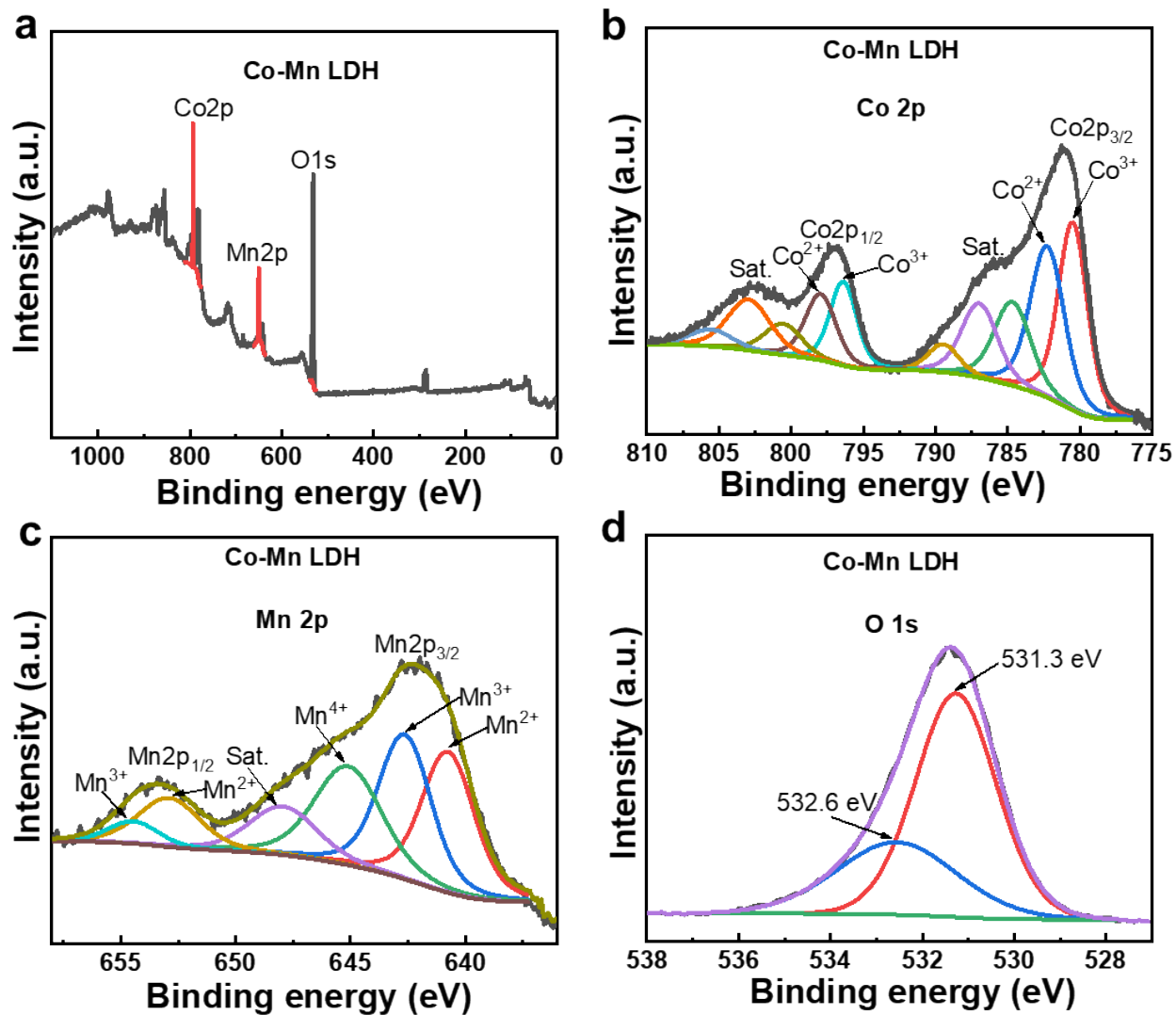
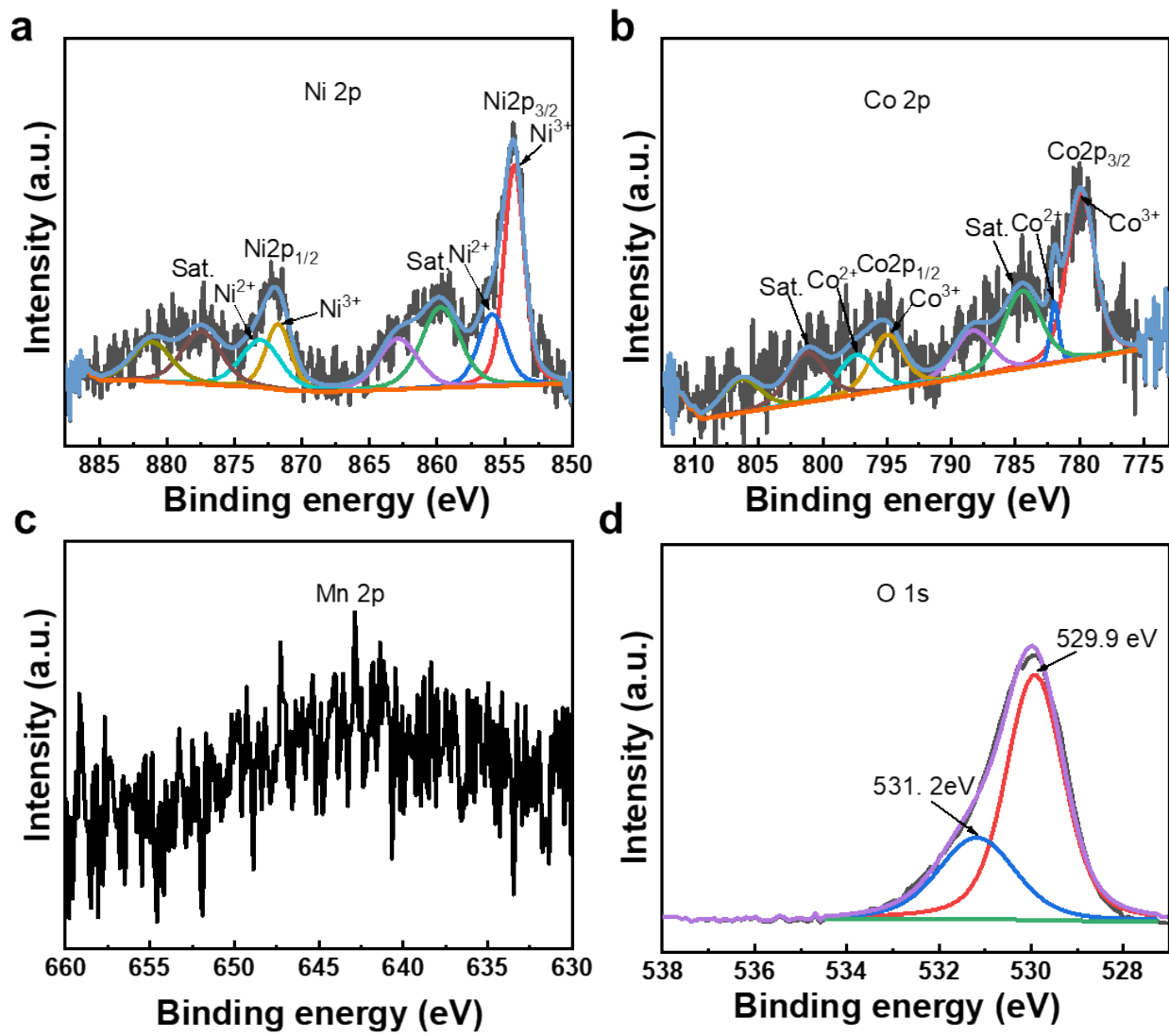


Fig. S6 XPS spectra of Ni(OH)<sub>2</sub>/NF. (a) XPS survey spectrum and high-resolution XPS spectra of (b) Ni 2p, and (c) O 1s.



**Fig. S7** XPS spectra of CoMn-LDH/NF. (a) XPS survey spectrum and high-resolution XPS spectra of (b) Co 2p, (c) Mn 2p, and (d) O 1s.



**Fig. S8** High-resolution XPS spectra of Ni(OH)<sub>2</sub>@CoMn-LDH/NF after 25 h stability test. (a) Ni 2p, (b) Co 2p, (c) Mn 2p, and (d) O 1s.

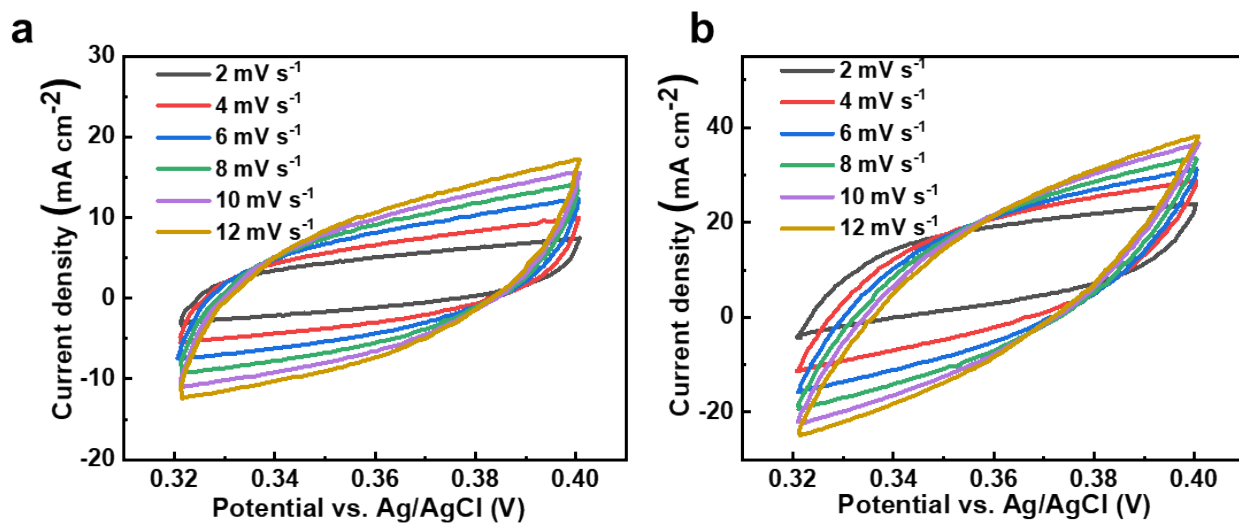


Fig. S9 CV curves at different scan rates of (a) CoMn-LDH, and (b)  $\text{Ni(OH)}_2$ .

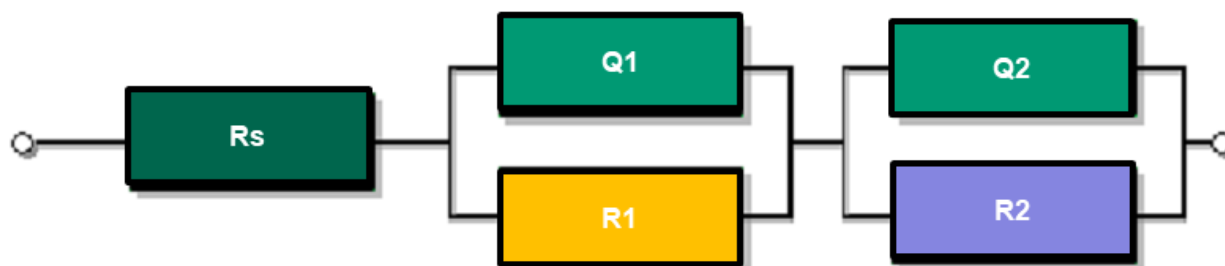


Fig. S10 The equivalent circuit for the simulation of EIS

**Table S1** Comparison of OER performances of various LDH-based catalysts in alkaline solution previously reported OER electrocatalysts

Electrocatalysts	Substrate	Electrolyte	Overpotential (mV) <sup>a</sup>	Ref.
CoMn-LDH	NF <sup>b</sup>	1 M KOH	326 <sup>c</sup>	This work
Ni(OH) <sub>2</sub> @CoMn-LDH	NF	1 M KOH	250 <sup>c</sup>	This work
CoMn LDH	GC <sup>d</sup>	1 M KOH	324	1
CoMn LDH	-	1 M KOH	325	2
CoMn LDH/g-C <sub>3</sub> N <sub>4</sub>	-	1 M KOH	350 <sup>e</sup>	2
CoMn-LDH/MWCNT <sup>d</sup>	-	1 M KOH	300	3
CoMn-LDH	-	0.1 M KOH	386	4
CoMn LDH/CNT	-	0.1 M KOH	355	4
CoFe LDH	NF	1 M KOH	300	5
NiFe LDH	GC	1M KOH	375	6
NiFe LDH/CNT	GC	1 M KOH	320	7
NiMn LDH/NiCo <sub>2</sub> O <sub>4</sub>	NF	1 M KOH	310	8
Ni <sub>3</sub> Mn <sub>1</sub> LDH	RDE <sup>f</sup>	1 M KOH	350	9
Ni <sub>5</sub> Mn LDH/MWCNT	RDE	1 M KOH	350	9
CoNi LDH	CP <sup>g</sup>	1 M KOH	367	10
CoNi LDH	ITO <sup>h</sup>	1M KOH	333	11

Overpotential (mV)<sup>a</sup>: at 10 mA cm<sup>-2</sup>

NF<sup>b</sup>: Ni Foam

326<sup>c</sup>: at 30 mA cm<sup>-2</sup>

GC<sup>d</sup>: Glass Carbon



350 e<sup>-</sup>: at 39.23 mA cm<sup>-2</sup>

RDE <sup>f</sup>: Rotating Disk Electrode

CP <sup>g</sup>: Carbon Paper

ITO <sup>h</sup>: Indium Tin Oxide

## References

1. Jagadale, A. D. *et al.* Ultrathin nanoflakes of cobalt-manganese layered double hydroxide with high reversibility for asymmetric supercapacitor. *J. Power Sources* **306**, 526–534 (2016).
2. Arif, M. *et al.* Coupling of Bifunctional CoMn-Layered Double Hydroxide@Graphitic C3N4 Nanohybrids towards Efficient Photoelectrochemical Overall Water Splitting. *Chem. - An Asian J.* **13**, 1045–1052 (2018).
3. Jia, G. *et al.* Formation of Hierarchical Structure Composed of (Co/Ni)Mn-LDH Nanosheets on MWCNT Backbones for Efficient Electrocatalytic Water Oxidation. *ACS Appl. Mater. Interfaces* **8**, 14527–14534 (2016).
4. Liu, Z. *et al.* CoMn Layered Double Hydroxides/Carbon Nanotubes Architectures as High-Performance Electrocatalysts for the Oxygen Evolution Reaction. *ChemElectroChem* **3**, 906–912 (2016).
5. Feng, L. *et al.* A Highly Active CoFe Layered Double Hydroxide for Water Splitting. *Chempluschem* **82**, 483–488 (2017).
6. Gao, X., Long, X., Yu, H., Pan, X. & Yi, Z. Ni Nanoparticles Decorated NiFe Layered Double Hydroxide as Bifunctional Electrochemical Catalyst. *J. Electrochem. Soc.* **164**, H307–H310 (2017).

7. Gong, M. *et al.* An advanced Ni-Fe layered double hydroxide electrocatalyst for water oxidation. *J. Am. Chem. Soc.* **135**, 8452–8455 (2013).
8. Yang, L. *et al.* NiMn layered double hydroxide nanosheets/NiCo<sub>2</sub>O<sub>4</sub> nanowires with surface rich high valence state metal oxide as an efficient electrocatalyst for oxygen evolution reaction. *J. Power Sources* **392**, 23–32 (2018).
9. Sumboja, A., Chen, J., Zong, Y., Lee, P. S. & Liu, Z. NiMn layered double hydroxides as efficient electrocatalysts for the oxygen evolution reaction and their application in rechargeable Zn-air batteries. *Nanoscale* **9**, 774–780 (2017).
10. Liang, H. *et al.* Hydrothermal continuous flow synthesis and exfoliation of NiCo layered double hydroxide nanosheets for enhanced oxygen evolution catalysis. *Nano Lett.* **15**, 1421–1427 (2015).
11. Zhang, C. *et al.* Layer-by-layer assembly of exfoliated layered double hydroxide nanosheets for enhanced electrochemical oxidation of water. *J. Mater. Chem. A* **4**, 11516–11523 (2016).