

## Two new photochromic supramolecular compositions based on viologen: photocontrolled fluorescence, aniline detection and inkless erasable printing performance

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Synthesis of 3-BCBPY·2Cl and 4-BCBPY·2Cl.

3-BCBPY·2Cl: Dissolve 4,4'-bipyridine (10 mmol, 1.56 g) and 3-cyanobenzyl chloride (20 mmol, 3.92 g) in 25mL of N, N'-dimethylformamide (DMF), and stir at 110°C for 6h. Filtration yielded a light yellow product. The obtained product was washed 3 times with hot DMF and recrystallized with methanol to finally obtain a pale yellow solid. The yield was 90%.

4-BCBPY·2Cl: The synthesis process is similar to 3-BCBPY·2Cl except that 4-cyanobenzyl chloride (20 mmol, 3.92 g) is used instead of 4-cyanobenzyl chloride (20 mmol, 3.92 g). A yellow solid is obtained. The yield was 90%.

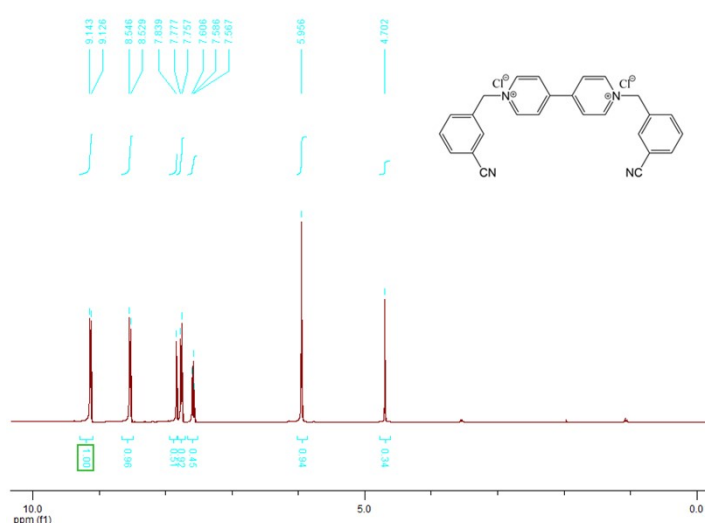


Figure S1. The <sup>1</sup>H NMR spectrum of 3-BCBPY·2Cl in D<sub>2</sub>O (600 MHz).

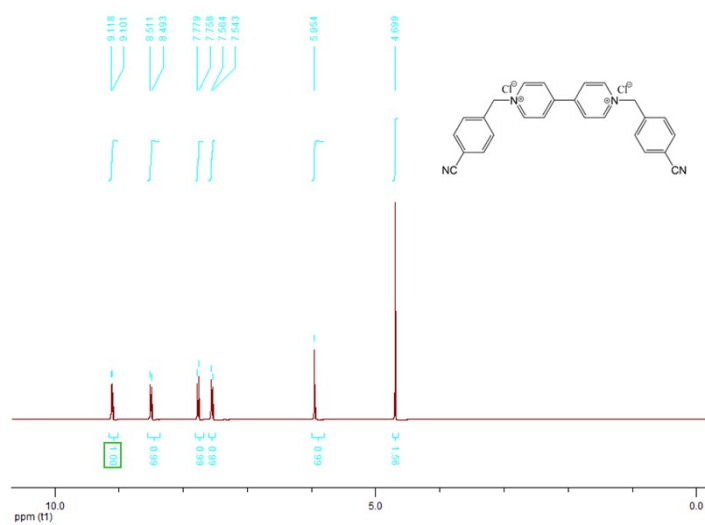


Figure S2. The  $^1\text{H}$  NMR spectrum of 4-BCBPY·2Cl in  $\text{D}_2\text{O}$  (600 MHz).

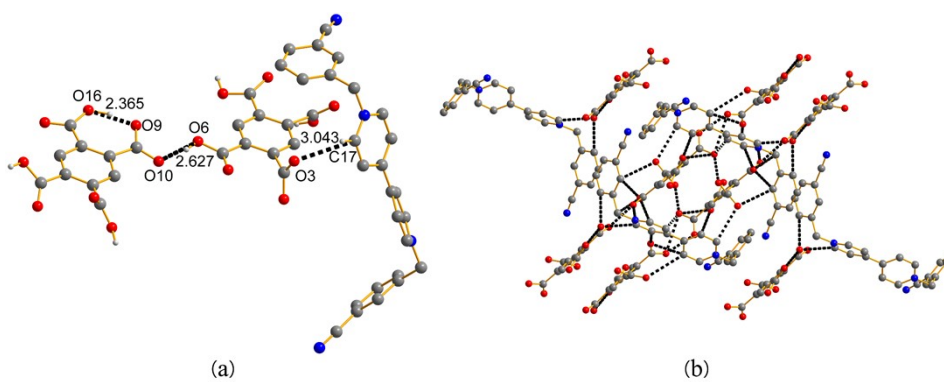


Figure S3. (a) Schematic diagram of partial hydrogen bond of Compound **1**; (b) The hydrogen bond network of Compound **1**.

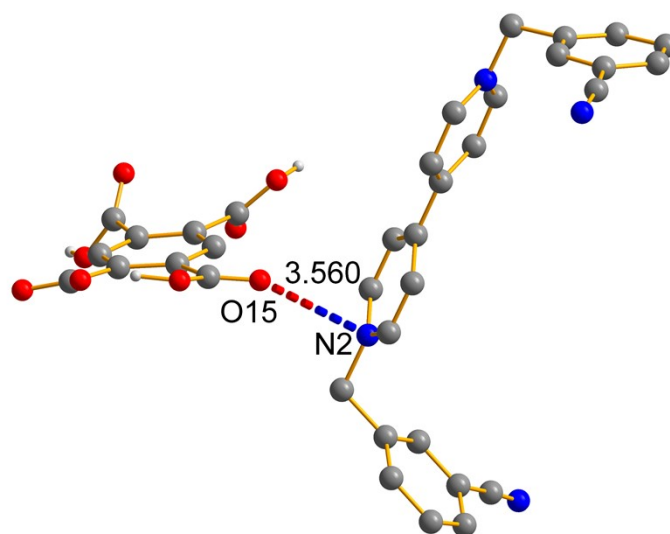


Figure S4. ET distances for Compound **1**.

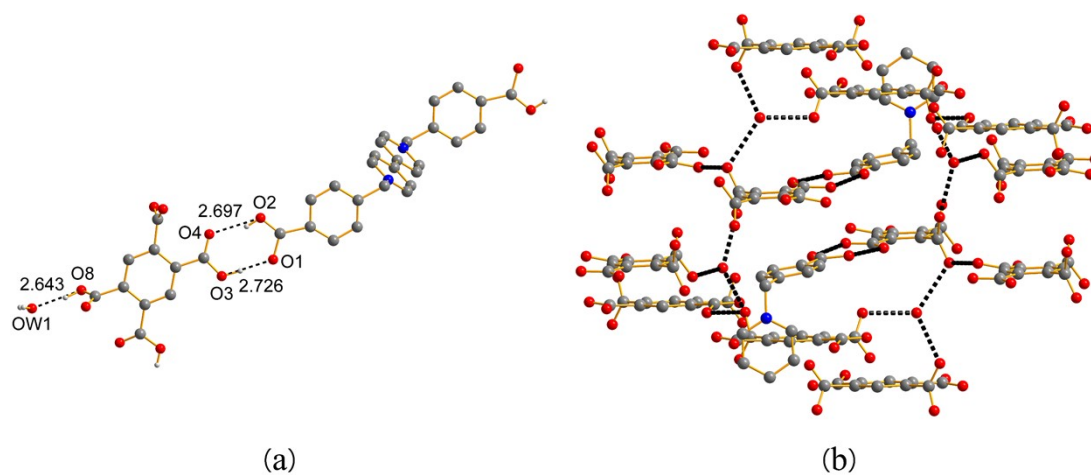


Figure S5. (a) Schematic diagram of partial hydrogen bond of Compound 2; (b) The hydrogen bond network of Compound 2.

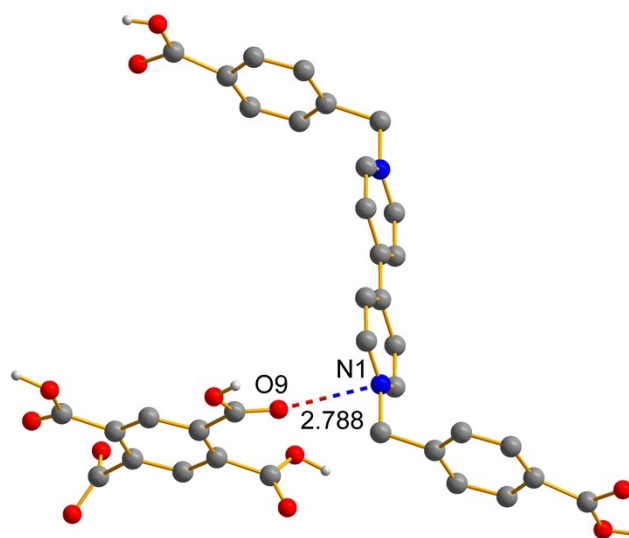


Figure S6. ET distances for Compound 2.

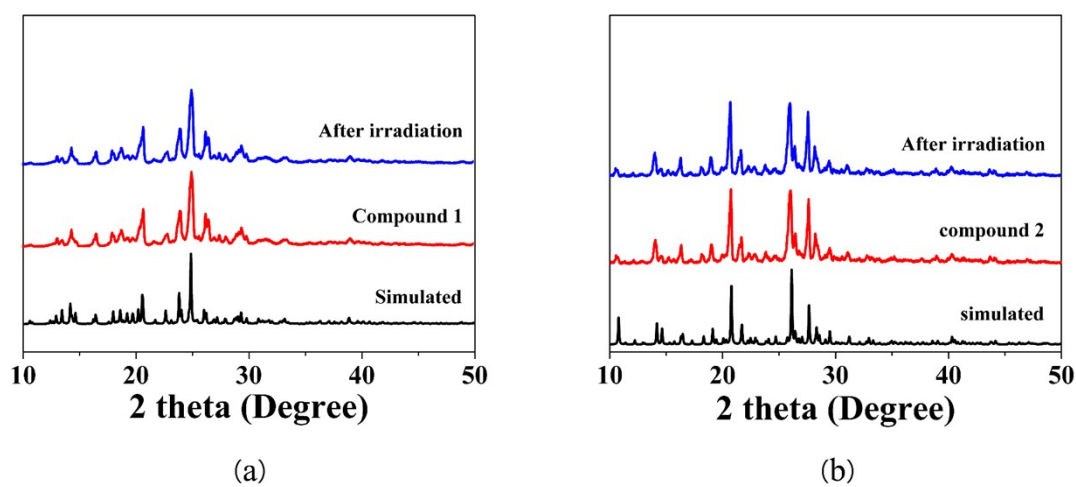


Figure S7. The PXRD spectra before and after irradiation for 1(a) and 2(b).

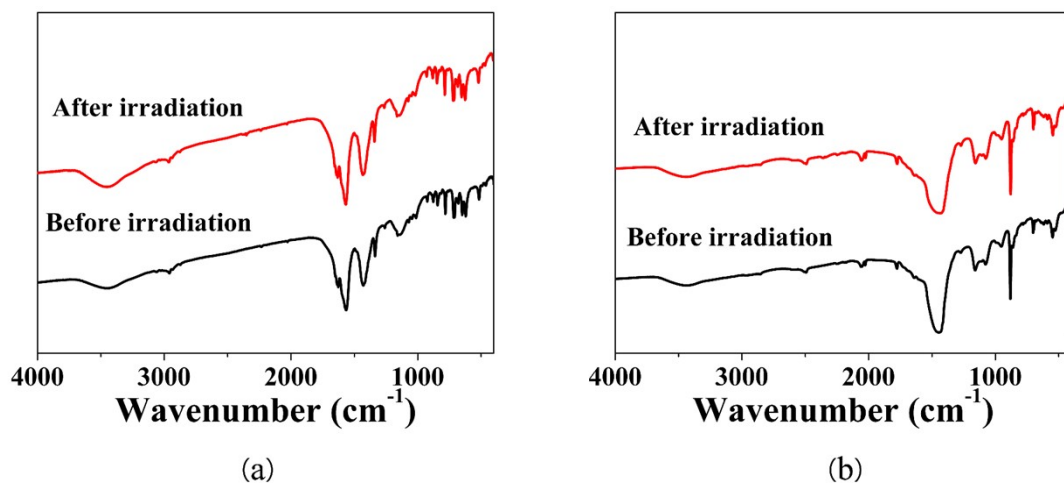


Figure S8. FT-IR spectra of Compounds 1(a) and 2(b).

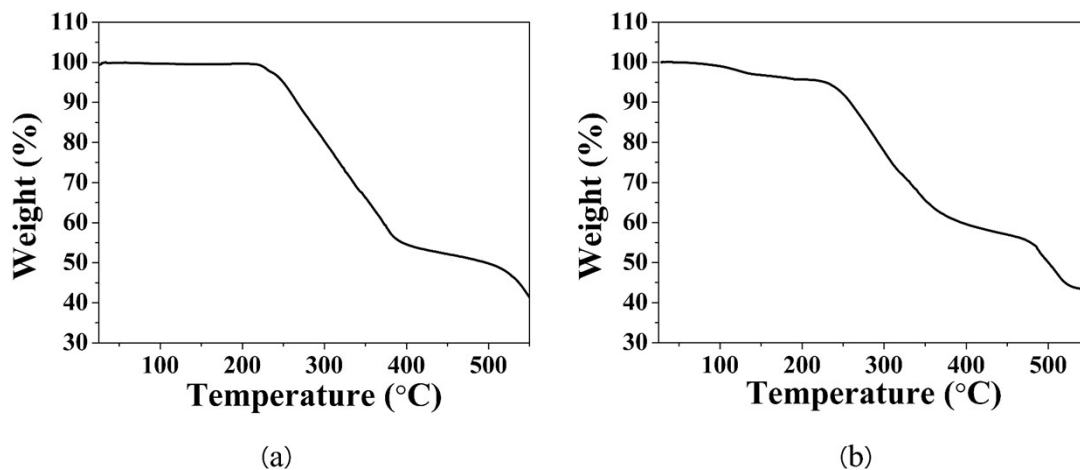


Figure S9. Thermogravimetric curves of Compounds 1(a) and 2(b).

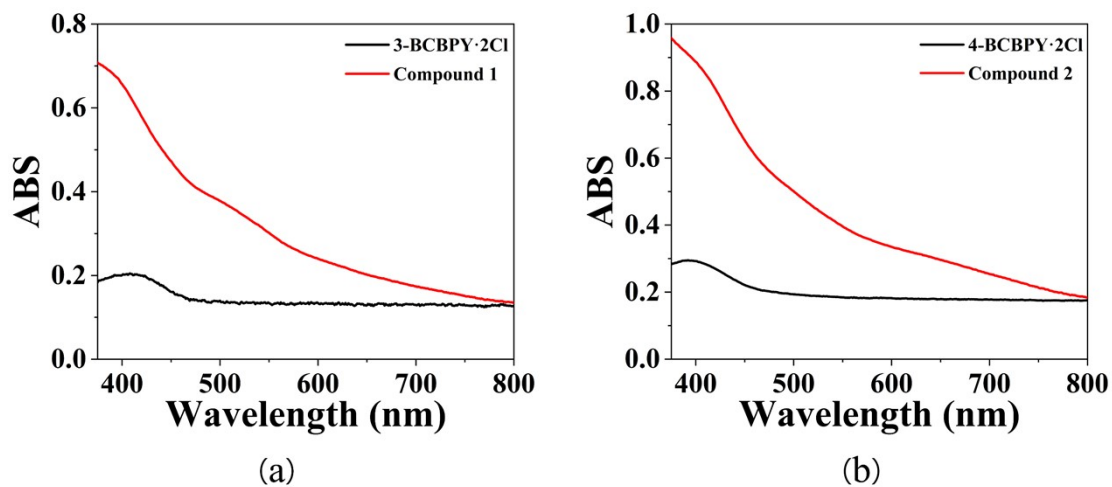
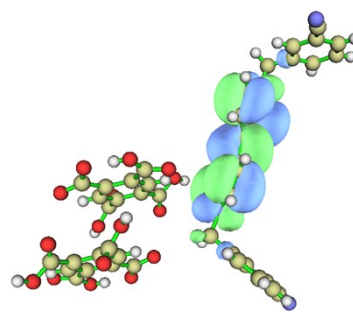


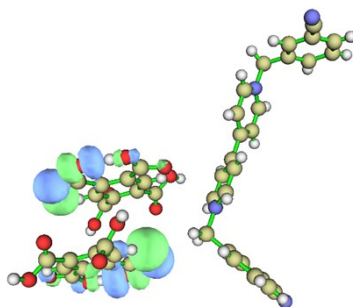
Figure S10. (a) UV-Vis spectra of 3-BCBPY.2Cl and compound 1; (b) UV-Vis spectra of 4-

BCBPY.2Cl and compound 2.

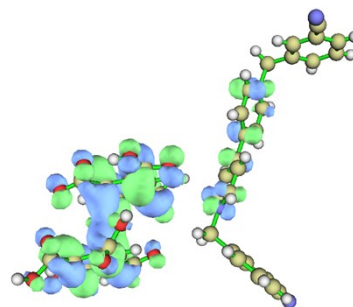
Compound 1



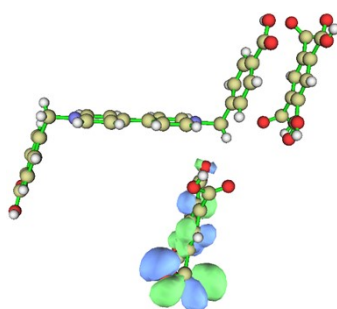
LUMO



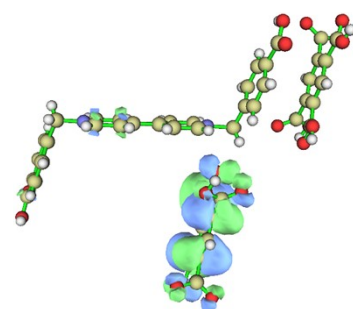
HOMO-1



LUMO+1

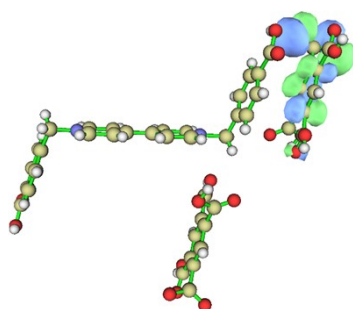


HOMO

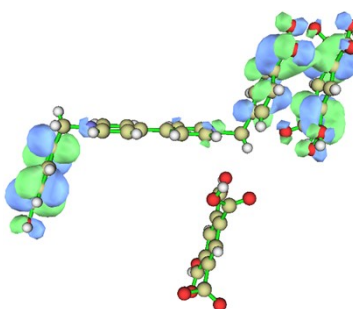


LUMO+4

Compound 2



HOMO-1



LUMO+3

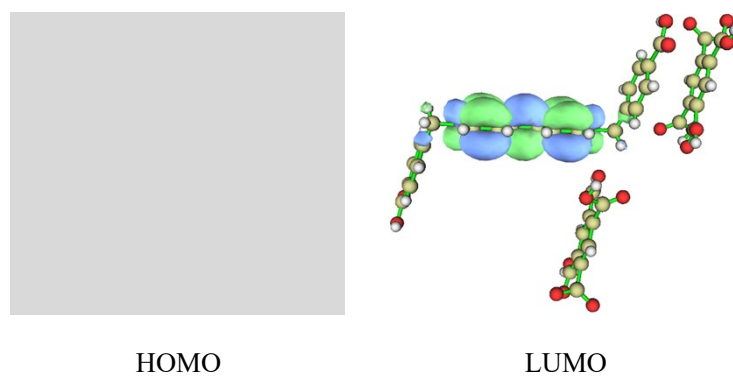


Figure S11. Calculated HOMO and LUMO orbitals of compounds **1** and **2**.

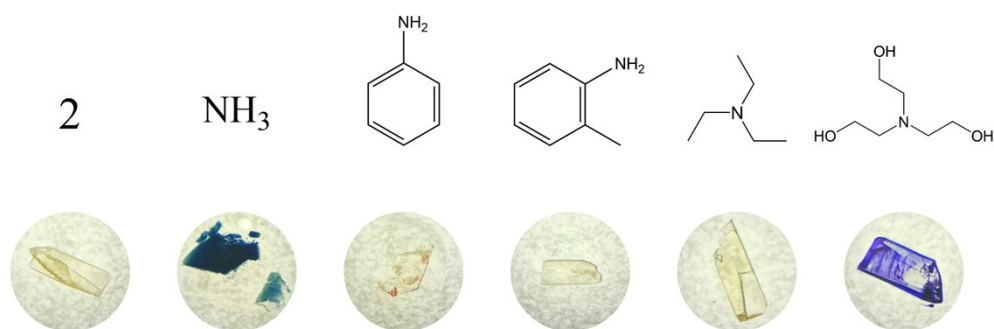


Figure S12. Photographs of Compound **2** before and after exposure to different amines.

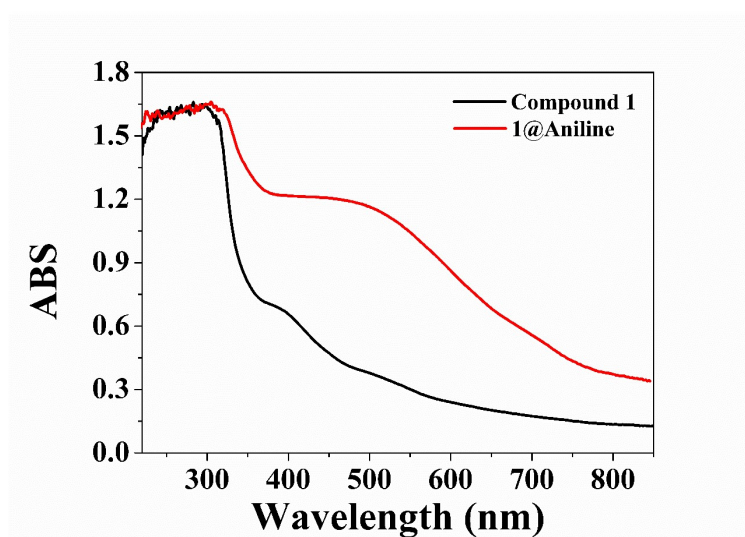


Figure S13. The UV-Vis spectra of Compounds **1** and **1@Aniline**.

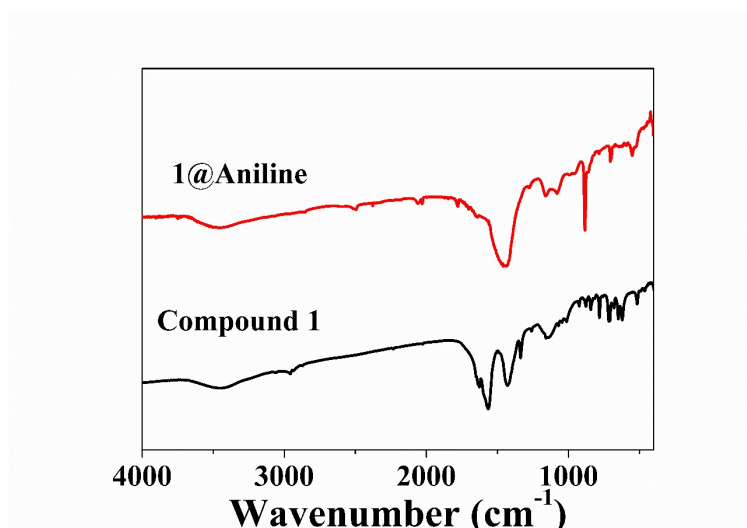


Figure S14. FT-IR spectra of Compound 1 and 1@Aniline.

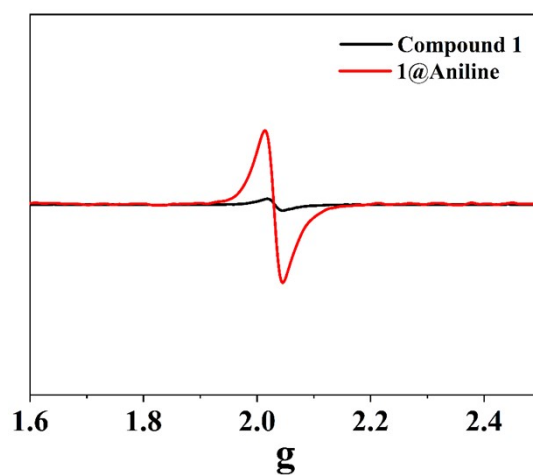


Figure S15. ESR spectra of Compounds 1 and 1@Aniline.

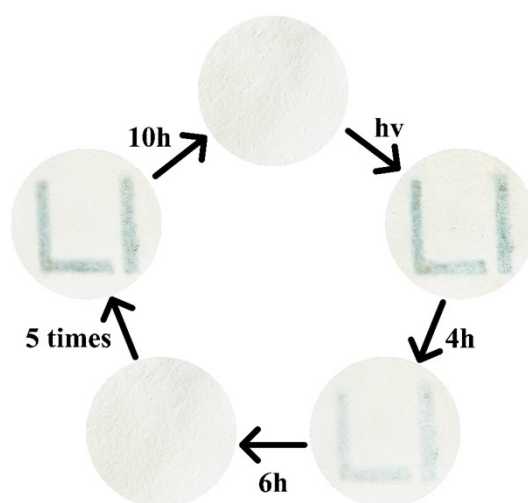


Figure S16. Photograph of 1@ethanol after repeated printing.

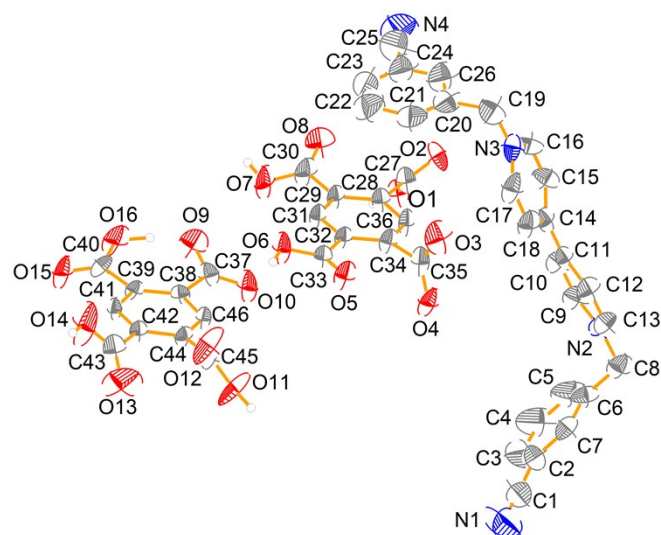


Figure S17. The asymmetric units of the structures (thermal ellipsoids) for Compound **1**.

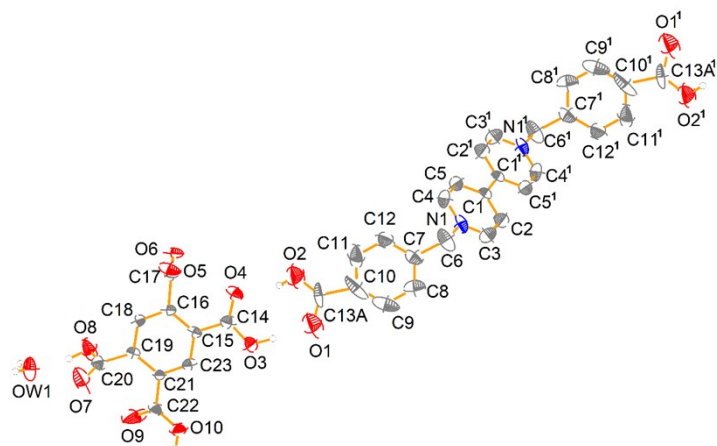


Figure S18. The asymmetric units of the structures (thermal ellipsoids) for Compound **2**

(Symmetry code:  $1/2-x, -y, -z$ ).



Table S1. Crystal data and structure refinement for **1** and **2**.

Compound	<b>1</b>	<b>2</b>
Empirical formula	C <sub>46</sub> H <sub>29.58</sub> N <sub>4</sub> O <sub>16</sub>	C <sub>46</sub> H <sub>35</sub> N <sub>2</sub> O <sub>21</sub>
Formula weight	894.32	951.76
Temperature/K	296.15	296.15
Crystal system	triclinic	monoclinic
Space group	<i>P</i> -1	<i>P</i> 2 <sub>1</sub> / <i>c</i>
<i>a</i> /Å	9.7493(16)	9.6809(14)
<i>b</i> /Å	14.816(2)	15.526(2)
<i>c</i> /Å	14.827(2)	13.6809(17)
$\alpha$ /°	81.262(3)	90
$\beta$ /°	81.825(3)	94.407(3)
$\gamma$ /°	72.098(3)	90
Volume/Å <sup>3</sup>	2003.9(5)	2050.2(5)
<i>Z</i>	2	2
$\rho_{\text{calc}}/\text{cm}^3$	1.482	1.542
$\mu/\text{mm}^{-1}$	0.114	0.124
$2\theta$ range for data collection/°	2.908 to 54.868	3.974 to 55.382
Goodness-of-fit on <i>F</i> <sup>2</sup>	1.036	1.062
Final R indexes [ <i>I</i> ≥ 2σ ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0910, <i>wR</i> <sub>2</sub> = 0.2500	<i>R</i> <sub>1</sub> = 0.0896, <i>wR</i> <sub>2</sub> = 0.1823
Final R indexes [all data]	<i>R</i> <sub>1</sub> = 0.1507, <i>wR</i> <sub>2</sub> = 0.2980	<i>R</i> <sub>1</sub> = 0.2066, <i>wR</i> <sub>2</sub> = 0.2192

Table S2. O/C–H···O interactions Geometry (Å, °) for **1** and **2**.

	D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)	
<b>Compound 1</b>	C13-H13...O5 <sup>1</sup>	0.93	2.34	3.175(5)	148.6	
	C13-H13...O1 <sup>2</sup>	0.93	2.63	3.201(6)	120.1	
	C12-H12...O1 <sup>2</sup>	0.93	2.55	3.161(6)	123.5	
	C12-H12...O1A <sup>2</sup>	0.93	2.24	3.143(9)	163.3	
	C10-H10...O13 <sup>3</sup>	0.93	2.16	3.088(6)	173.1	
	C18-H18...O8 <sup>2</sup>	0.93	2.23	3.120(5)	159.6	
	C7-H7...O5 <sup>1</sup>	0.93	2.57	3.402(6)	149.3	
	C7-H7...O3 <sup>1</sup>	0.93	2.58	3.313(6)	135.9	
	C9-H9...O9 <sup>4</sup>	0.93	2.31	3.220(6)	165.2	
	C15-H15...O12 <sup>3</sup>	0.93	2.42	3.332(6)	165.5	
	C15-H15...O13 <sup>3</sup>	0.93	2.75	3.380(7)	126.0	
	C8-H8A...O5 <sup>1</sup>	0.97	2.59	3.389(5)	139.3	
	C17-H17...O3	0.93	2.42	3.043(7)	124.1	
	O1-H1...O4 <sup>4</sup>	0.82	1.62	2.373(6)	151.7	
	C16-H16...N4 <sup>5</sup>	0.93	2.35	3.219(10)	156.0	
	C5-H5...O9 <sup>4</sup>	0.93	2.59	3.351(6)	139.6	
	O16-H16A...O9	1.04(5)	1.34(5)	2.364(5)	168(4)	
	O6-H6...O10	0.88(6)	1.76(6)	2.627(4)	170(6)	
	O11-H11...O15 <sup>2</sup>	1.00(7)	1.72(7)	2.704(4)	168(6)	
	O11-H11...O16 <sup>2</sup>	1.00(7)	2.52(7)	3.152(5)	121(5)	
	O14-H14...O2 <sup>6</sup>	0.86(2)	1.79(2)	2.649(6)	177(7)	
	O14-H14...O1 <sup>6</sup>	0.86(2)	2.54(6)	3.093(7)	123(5)	
	O14-H14...O2A <sup>6</sup>	0.86(2)	2.03(4)	2.797(10)	148(6)	
	O14-H14...O1A <sup>6</sup>	0.86(2)	2.36(5)	3.009(9)	133(6)	
	O7-H7A...O4 <sup>7</sup>	0.84(2)	1.91(5)	2.678(4)	152(9)	
	<b>Compound 2</b>	OW1-HW1A...O5 <sup>1</sup>	0.85	1.86	2.685(5)	162.3
		OW1-HW1B...O6 <sup>2</sup>	0.85	1.99	2.798(4)	158.3
O2-H2A...O4		0.82	1.91	2.697(8)	159.6	
O8-H8A...OW1		0.92(5)	1.77(6)	2.644(5)	158(5)	
O10-H10...O6 <sup>3</sup>		0.84(6)	1.74(6)	2.572(4)	171(6)	
O3-H3A...O1		0.98(7)	1.75(7)	2.727(5)	176(6)	

Symmetry codes of 1: <sup>1</sup>2-x, 1-y, 1-z; <sup>2</sup>1+x, +y, +z; <sup>3</sup>+x, -1+y, 1+z; <sup>4</sup>1-x, 1-y, 1-z; <sup>5</sup>1-x, -y, 1-z; <sup>6</sup>+x, 1+y, -1+z; <sup>7</sup>-1+x, +y, +z.

Symmetry codes of 2: <sup>1</sup>1-x, 2-y, 1-z; <sup>2</sup>1-x, 1/2+y, 1/2-z; <sup>3</sup>-1+x, +y, +z.

Table S3. Atomic Occupancy for **1** and **2**.

	<b>Atom</b>	<b>Occupancy</b>
<b>Compound 1</b>	O1	0.584(8)
	O1A	0.416(8)
	O2	0.584(8)
	O2A	0.416(8)
	H1	0.584(8)
<b>Compound 2</b>	C13A	0.497(11)
	C13B	0.503(11)
	O2	0.5
	H2A	0.5

Table S4. Results of the TDDFT/B3LYP-lanl2dz calculations related to **1** and **2**.

	Excited states	Orbital description	$\lambda_{\text{max}} / \text{nm}$		f
			Exp.	Calc.	
Compound <b>1</b>	8	HOMO <sub>-1</sub> →LUMO (51.20%)	404	364	0.0034
	2	HOMO <sub>-1</sub> →LUMO <sub>+1</sub> (54.24%)	618	514	0.0001
Compound <b>2</b>	8	HOMO→LUMO <sub>+4</sub> (64.78%)	406	377	0.0737
	7	HOMO <sub>-1</sub> →LUMO <sub>+3</sub> (46.67%)		379	0.0651
	1	HOMO→LUMO (70.67%)	624	622	0.0001