

## Electronic Supplementary Material (ESI) for New Journal of Chemistry.

### Supporting Information

## Microporous Polycarbazole Frameworks with Large Conjugated $\pi$ System for Cyclohexane Separation from Cyclohexane-Containing Mixtures

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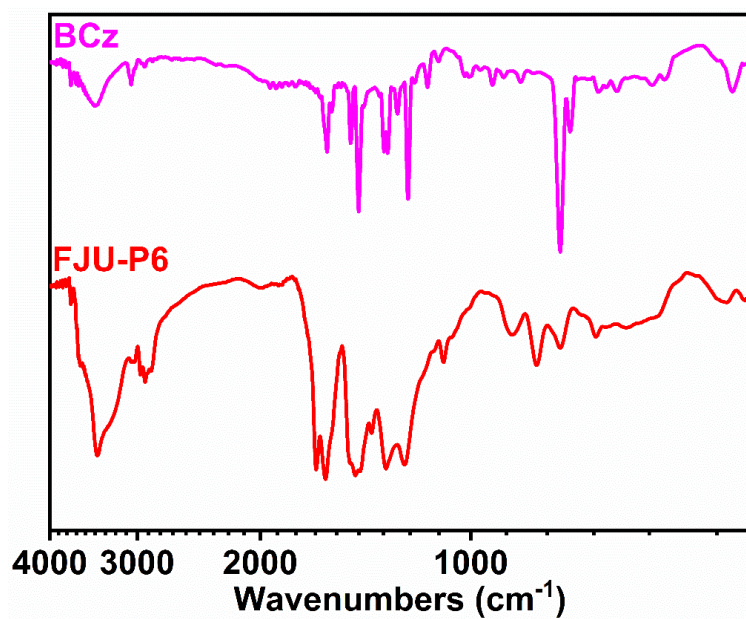


Figure S1. FT-IR spectra of BCz and FJU-P6.

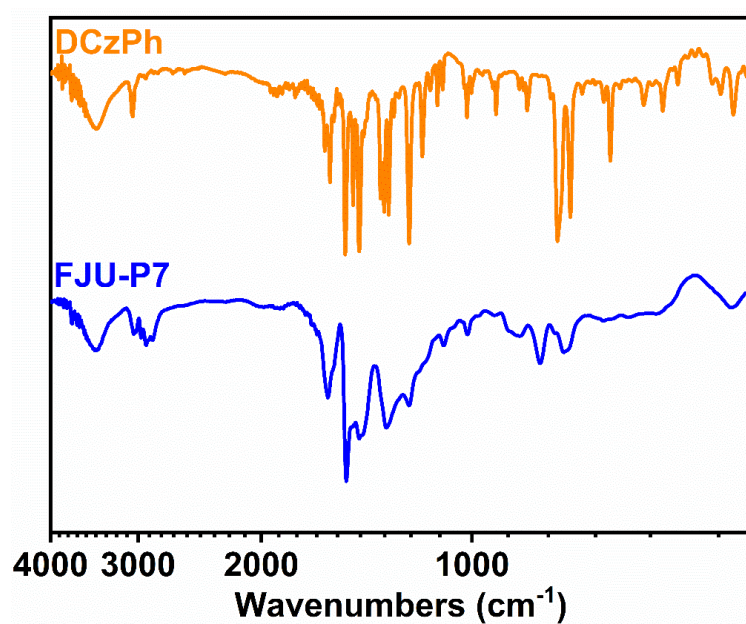


Figure S2. FT-IR spectra of DCzPh and FJU-P7.

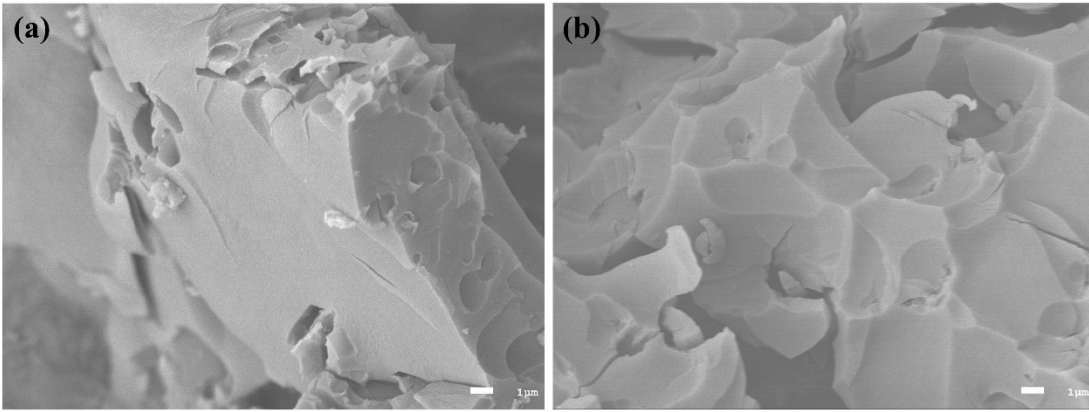


Figure S3. SEM pictures of (a) FJU-P6 and (b) FJU-P7.

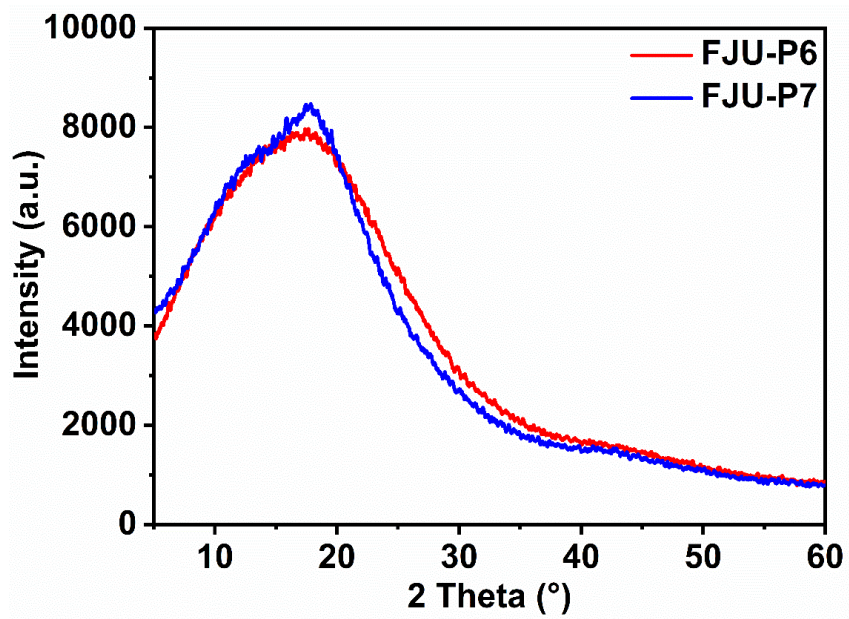


Figure S4. PXRD patterns of FJU-P6 and FJU-P7.

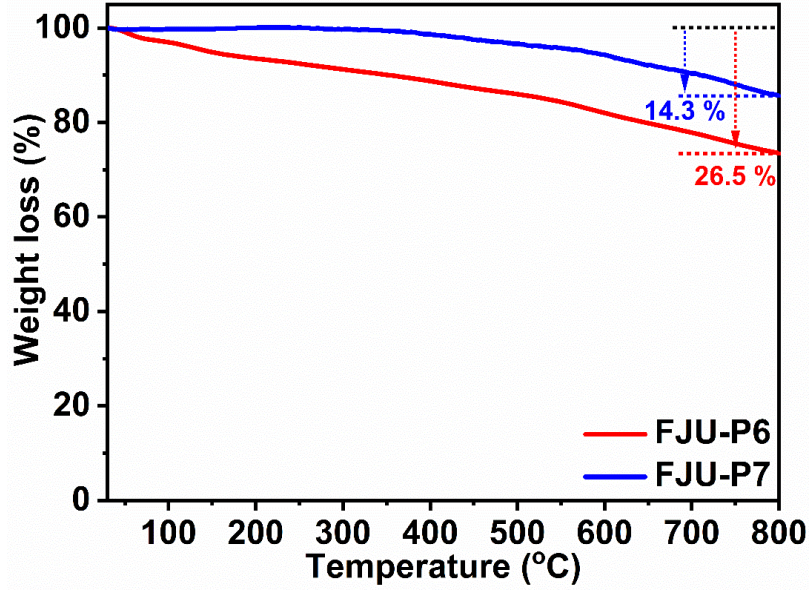


Figure S5. TGA curves of FJU-P6 and FJU-P7.

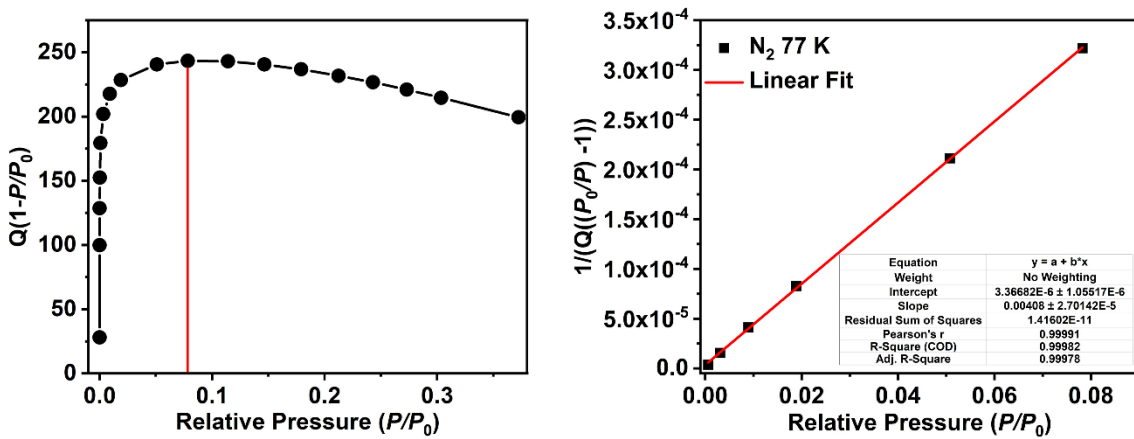


Figure S6. BET plots for FJU-P6. Points are selected based on the first consistency criterion.

$$S_{BET} = 1 / (3.36682 \times 10^{-6} + 0.00408) / 22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 1066 \text{ m}^2 \text{ g}^{-1}$$

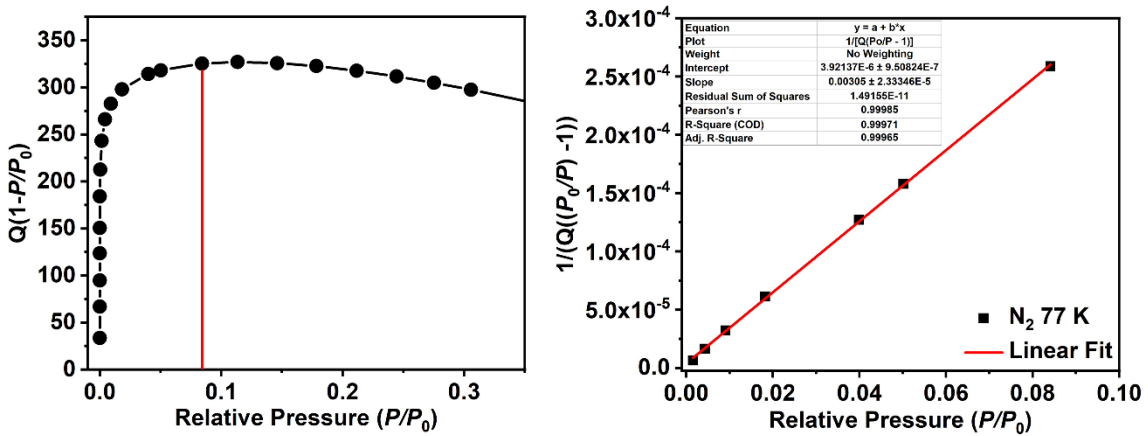


Figure S7. BET plots for FJU-P7. Points are selected based on the first consistency criterion.

$$S_{BET} = 1 / (3.92137 \times 10^{-6} + 0.00305) / 22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 1425 \text{ m}^2 \text{ g}^{-1}$$

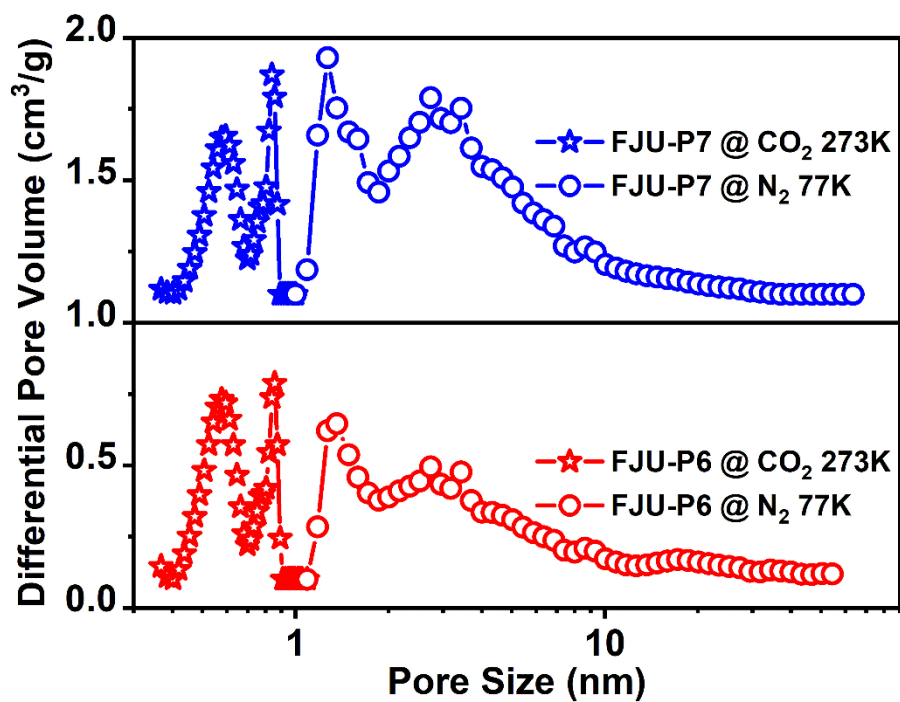


Figure S8. The pore-size distribution (PSD) of FJU-P6 and FJU-P7 from the 273 K CO<sub>2</sub> and 77 K N<sub>2</sub> isotherms based on the no-local density functional theory (NLDFT) model.

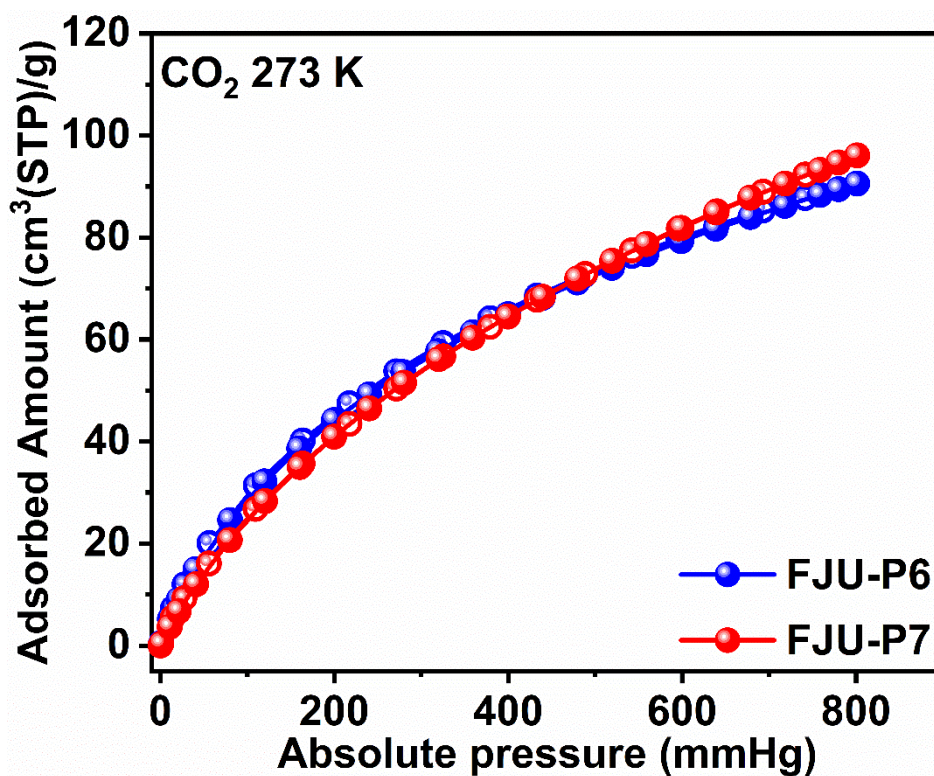
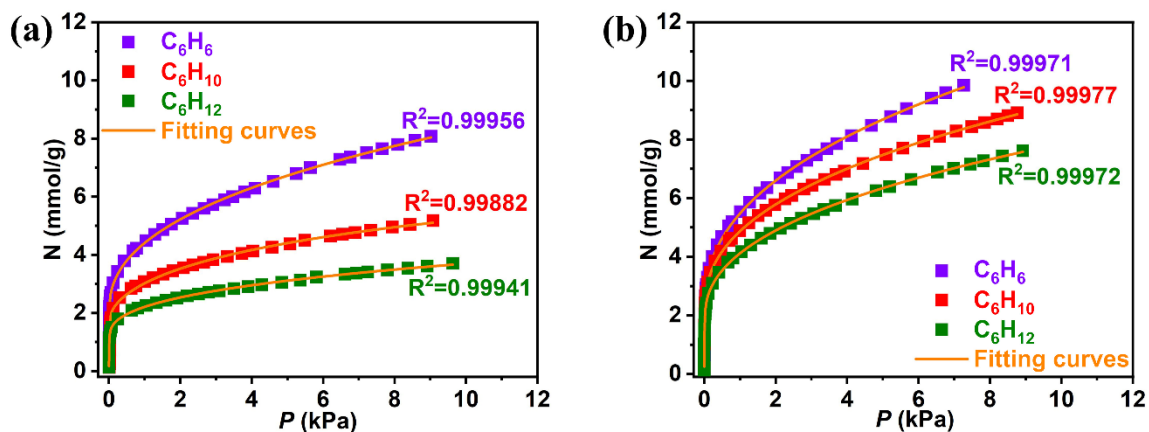
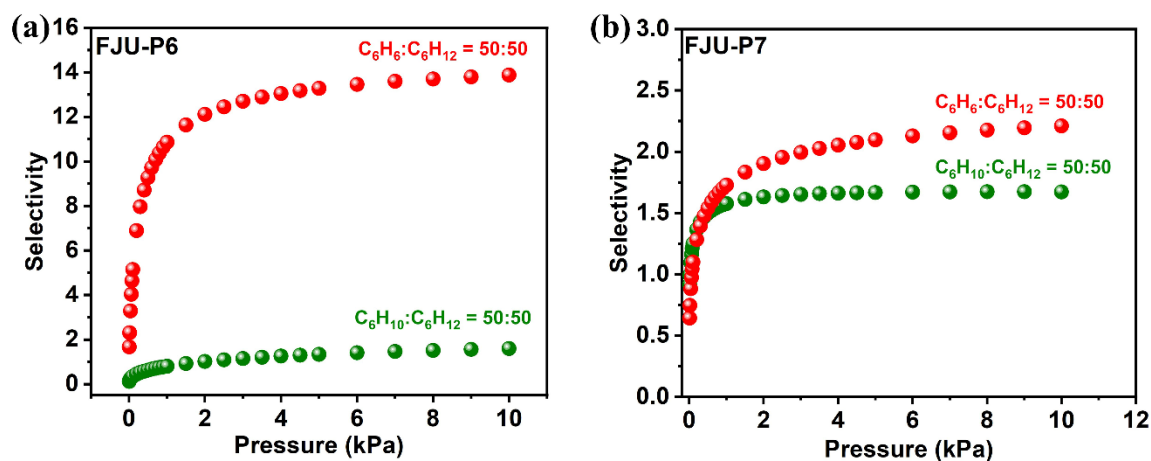


Figure S9. CO<sub>2</sub> adsorption isotherms of FJU-P6 (red) and FJU-P7 (blue) at 273 K.





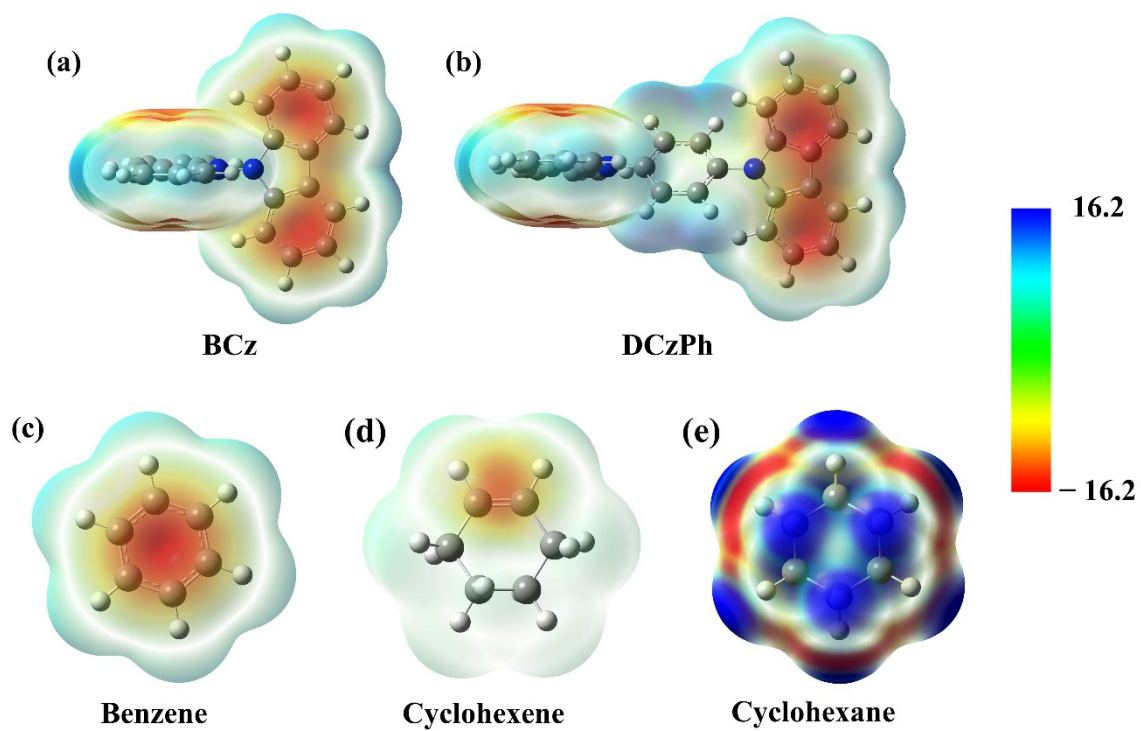
**Figure S10.** DSLF fitting of the  $C_6H_6$  (violet),  $C_6H_{10}$  (red) and  $C_6H_{12}$  (olive) sorption data at 294 K on FJU-P6 (a) and FJU-P7 (b).



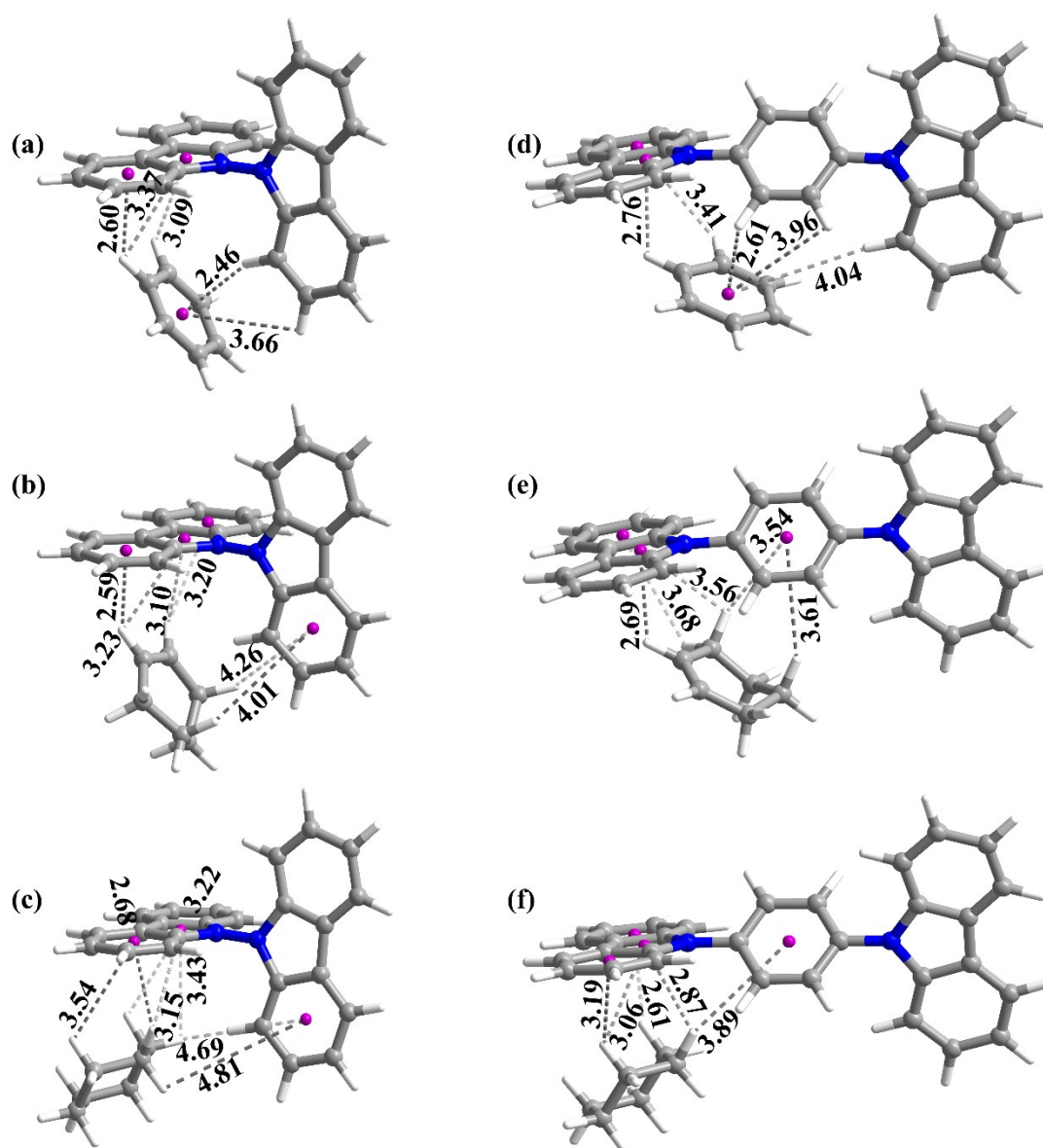
**Figure S11.** IAST selectivities for equivalent  $C_6H_6/C_6H_{12}$  (red) and  $C_6H_{10}/C_6H_{12}$  (olive) of FJU-P6 (a) and FJU-P7 (b) at 294 K.

**Table S1.** Equation parameters for the DSLF isotherm model.

Polymers	Adsorbates	$N_1^{max}$ (mmol g <sup>-1</sup> )	$b_1$ (kPa <sup>-1</sup> )	$1/n_1$	$N_2^{max}$ (mmol g <sup>-1</sup> )	$b_2$ (kPa <sup>-1</sup> )	$1/n_2$
FJU-P6	benzene	2.02758	9122.53765	1.71824	212.92323	0.01131	0.42889
	cyclohexene	1.68155	3.47291E8	5.64168	10.64258	0.14395	0.54079
	cyclohexane	1.12413	946.0946	1.24002	144.52421	0.00758	0.37874
FJU-P7	benzene	2.24083	10744.94006	2.04497	168.13274	0.01962	0.44012
	cyclohexene	2.17533	2181.68026	1.56111	166.3786	0.01962	0.4296
	cyclohexane	1.83063	1685.06854	1.41181	219.64403	0.01053	0.42814



**Figure S12.** Electrostatic potential maps around of (a) **BCz**, (b) **DCzPh**, (c) benzene, (d) cyclohexene and (e) cyclohexane. The gradation on the scale bar is in kcal mol<sup>-1</sup>.



**Figure S13.** C-H $\cdots$  $\pi$  interaction geometries for **BCz** with (a) benzene, (b) cyclohexene, (c) cyclohexane, and **DCzPh** with (d) benzene, (e) cyclohexene, (f) cyclohexane. Dot lines indicate C-H $\cdots$  $\pi$  interactions, the unit of distance is Å. (C gray, N blue, H white, dummy atom: center of the arene ring, violet).