Supporting Information

Hollow microspherical Bi₂MoO₆/Zn-Ti layered double hydroxide heterojunction for efficient visible-light photocatalytic degradation of organic contaminants

Xiu-Jie Yang,* Lunan Wu, Ruona Hu, Junqi Xing, Guohao Zhou, Shufen Lu, Jun Wu, Peng Li, Dong Liu*

State Key Laboratory of Heavy Oil Processing, School of Chemical Engineering, China University of Petroleum (East China), Qingdao, Shandong 266580, China



Figure S1 Standard curve of RhB solution.



Figure S2 XRD pattern of h-Bi₂MoO₆/(Zn/Ti)LDH-2%.



Figure S3 N₂ adsorption-desorption isotherm of (Zn/Ti)LDH.

Table	S1.	The	surface	area,	pore	volum	e and	pore	diameter	of	sam	oles.
1	~				p • • •			p • • •		~ -	r	

Sample	Surface Area (m ² ·g ⁻¹)	Pore Volume (cm ³ ·g ⁻¹)	Pore Diameter (nm)	
h-Bi ₂ MoO ₆	16.0	0.142	30.5	
h-Bi ₂ MoO ₆ /(Zn/Ti)LDH-1.5%	17.7	0.106	19.3	
h-Bi2MoO6/(Zn/Ti)LDH-2%	20.3	0.172	17.0	
h-Bi2MoO6/(Zn/Ti)LDH-3%	22.6	0.095	15.1	
h-Bi2MoO6/(Zn/Ti)LDH-4%	27.1	0.129	16.5	
(Zn/Ti)LDH	134.3	0.310	7.2	



Figure S4 Adsorption performance of the as-prepared samples.

Table S2. Photodegradation performance of the as-prepared samples compared with other related photocatalysts.

		~ 1	D 11			
Sample	Light source	Catalyst concentration (g·L ⁻¹)	Pollutant concentration (mg·L ⁻¹)	Degradation time(min)	Degradation rate(%)	Ref.
Bi ₂ MoO ₆ /Ag ₂ O	300W Xe (λ > 420 nm)	0.6	10	60	95	[1]
Ag/Bi ₂ MoO ₆	300W Xe (λ > 420 nm)	1	5	210	97	[2]
PtCo-Bi ₂ MoO ₆	300W Xe (λ > 420 nm)	0.5	10	30	95	[3]
Ag/Ag ₃ PO ₄ /Bi ₂ Mo O ₆	300W Xe (λ > 420 nm)	0.2	5	80	100	[4]
ZnFe ₂ O ₄ /Bi ₂ MoO ₆	150W Xe (λ > 420 nm)	1	10	240	99.9	[5]
BiFeO ₃ /Bi ₂ MoO ₆	150W Xe (λ > 420 nm)	1	10	300	93.84	[6]

$\delta\text{-}Bi_2O_3/Bi_2MoO_6$	500WXe	0.6	5	180	100	[7]
Fe ₃ O ₄ /SiO ₂ /Bi ₂ Mo O ₆	$300W \text{ Xe} (\lambda > 420 \text{ nm})$	1	10	120	100	[8]
SiO ₂ /Bi ₂ MoO ₆	500W Xe (λ > 420 nm)	0.2	5	60	100	[9]
Cl-doped Bi ₂ MoO ₆	300W Xe (λ > 420 nm)	1	5	160	99.5	[10]
Bi ₂ MoO ₆ /(Zn/Ti)L DH	500W Xe (λ > 400 nm)	1	10	60	100	This work



Figure S5 Photocatalytic performance of $h-Bi_2MoO_6/(Zn/Ti)LDH-2\%$ and the mixture.

References

[1] J. Zhang, H. Liu, Z. Ma, J. Mol. Catal. A-Chem., 2016, 424, 37.

[2] P. Suebsom, A. Phuruangrat, S. Suwanboon, S. Thongtem, T. Thongtem, Inorg. Chem. Commun., 2020, 119, 108120.

- [3] Y. Lu, X. Li, C. Han, L. Ge, S. Fang, P Qiu, RSC Adv., 2016, 6, 84485.
- [4] X. Lin, J. Hou, S. Jiang, Z. Liu, M. Wang, G. Che, RSC Adv., 2015, 5, 104815.

[5] C. Zhao, C. Shao, X. Li, X. Li, R. Tao, X. Zhou, Y. Liu, J. Alloys Compd., 2018, 747, 916.

[6] R. Tao, C. Shao, X. Li, X. Li, S. Liu, S. Yang, C. Zhao, Y. Liu, J. Colloid Interface Sci., 2018, 529, 404.

- [7] Y. Yin, F. Li, Q. Zhan, D. Jiang, R. Chen, Mater. Res. Bull., 2018, 103, 47.
- [8] X. Hou, Y. Tian, X. Zhang, S. Dou, L. Pan, W. Wang, Y. Li, J. Zhao, J. Alloys Compd., 2015, 638, 214.
- [9] X. Lin, X. Guo, D. Liu, Q. Wang, H. Zhai, L. Chang, Mater. Res. Bull., 2015, 63, 72.
- [10] A. Phuruangrat, P. Dumrongrojthanath, B. Kuntalue, S. Thongtem, T. Thongtem, Mater. Lett.,2017, 196, 256.