

## A selective fluorescent probe based on citrate doped polypyrrole for dual determination of $\text{VO}^{2+}/\text{Fe}^{3+}$ in biological samples

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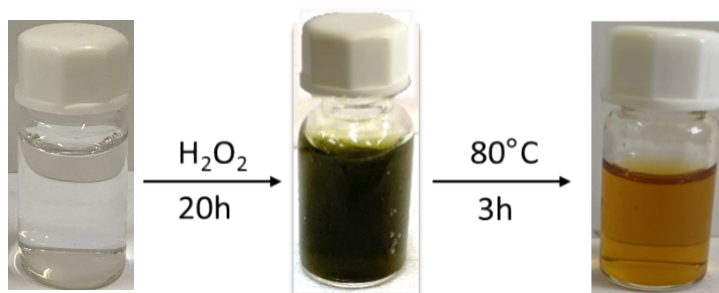


Fig. S1. Synthesis procedure of water-soluble PPy-Cit conjugated polymer

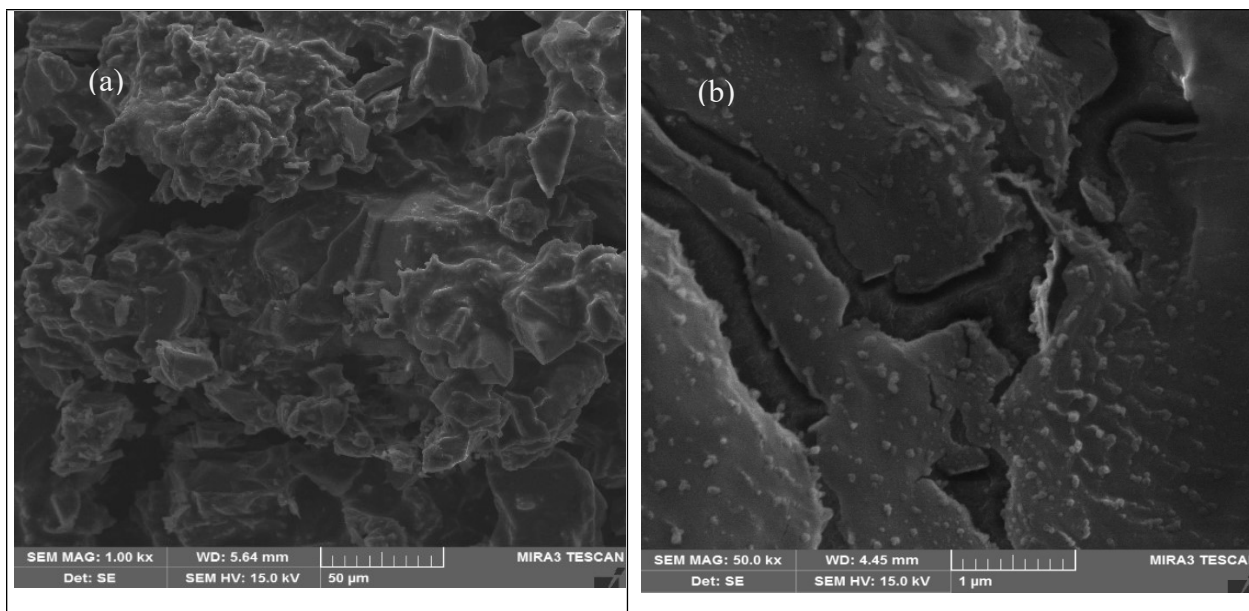


Fig. S2. SEM image of PPy-Cit polymer in 1000 (a) and 50000 (b) magnifications.

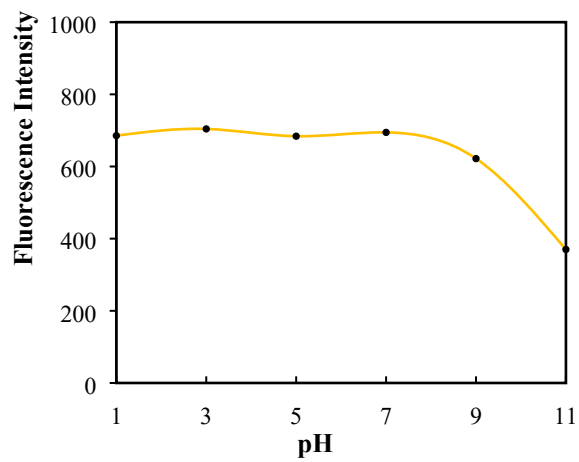


Fig. S3. Fluorescence intensity of 250 times diluted PPy-Cit solution in different pH, excitation wavelength = 395 nm and emission wavelength = 460 nm.

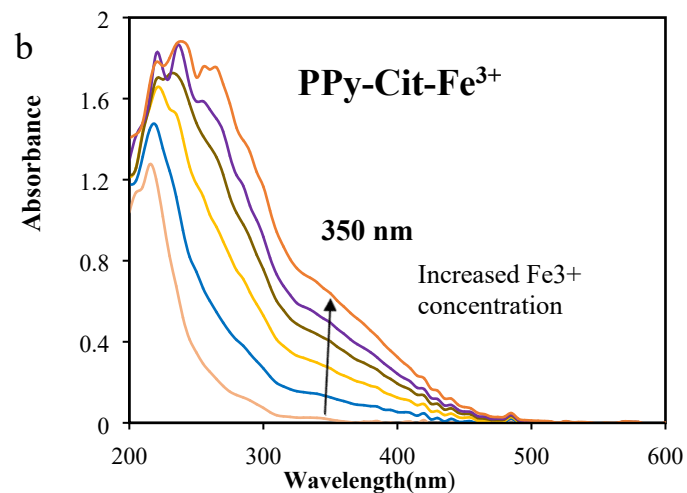
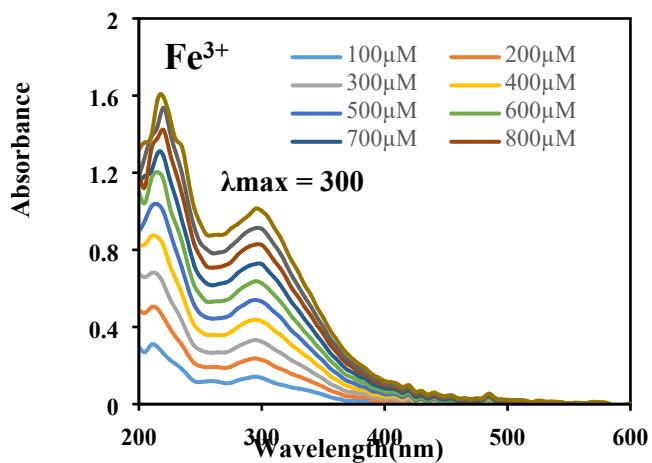
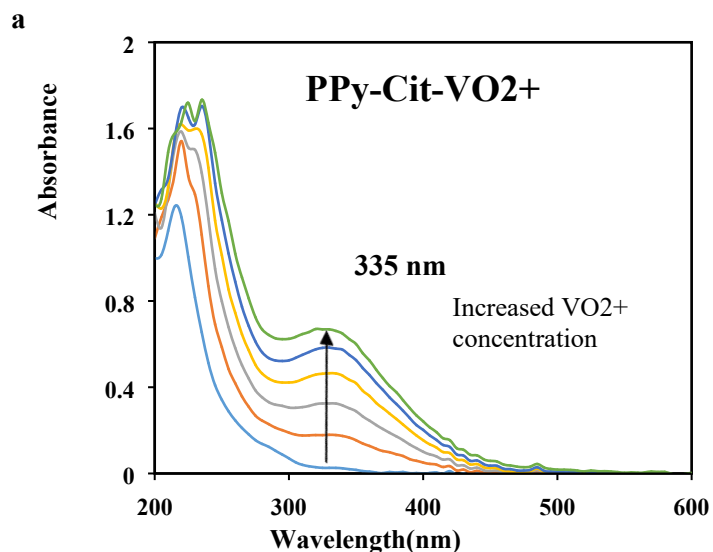
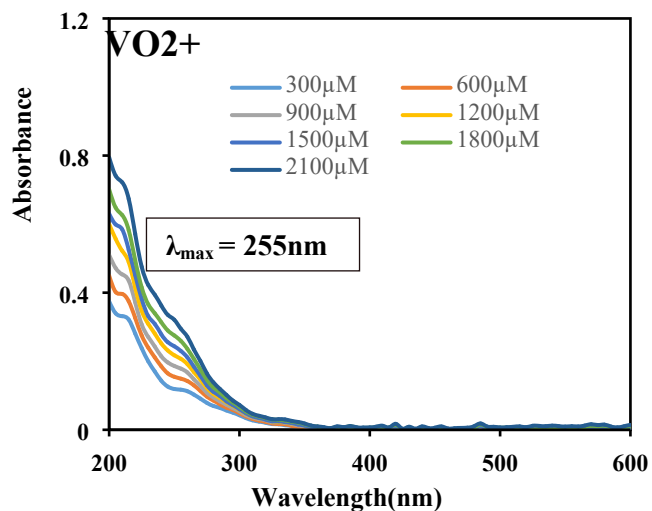


Fig. S4. UV-vis spectrum of the added ions at different concentrations and PPy-Cit in presence of added ions, a:  $\text{VO}_2^+$  (0, 250, 500, 750, 1000, 1250  $\mu\text{M}$ ); b:  $\text{Fe}^{3+}$  (0, 150, 250, 350, 450 and 550  $\mu\text{M}$ ).

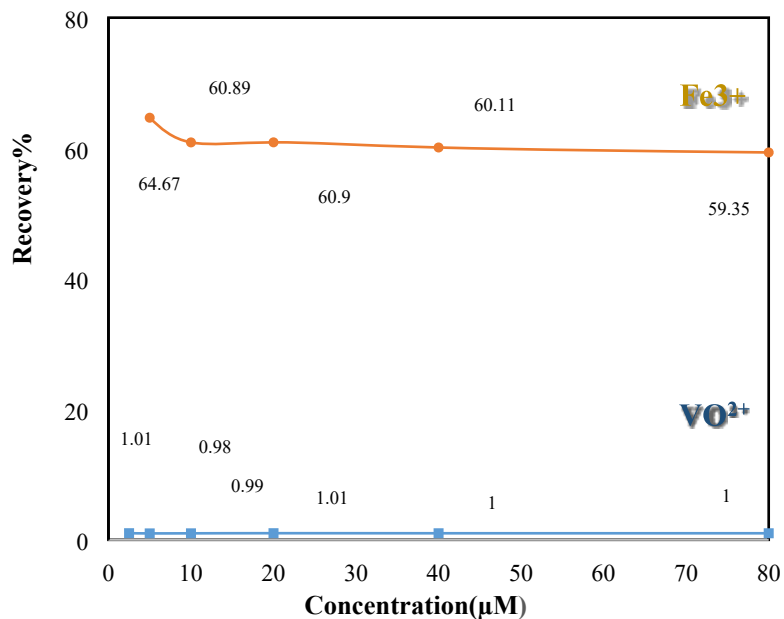


Fig. S5. Fluorescence intensity recovery percent, after adding EDTA (100 μM) to PPy-Cit-VO<sup>2+</sup> (lower curve) and PPy-Cit-Fe<sup>3+</sup> (upper curve) system.

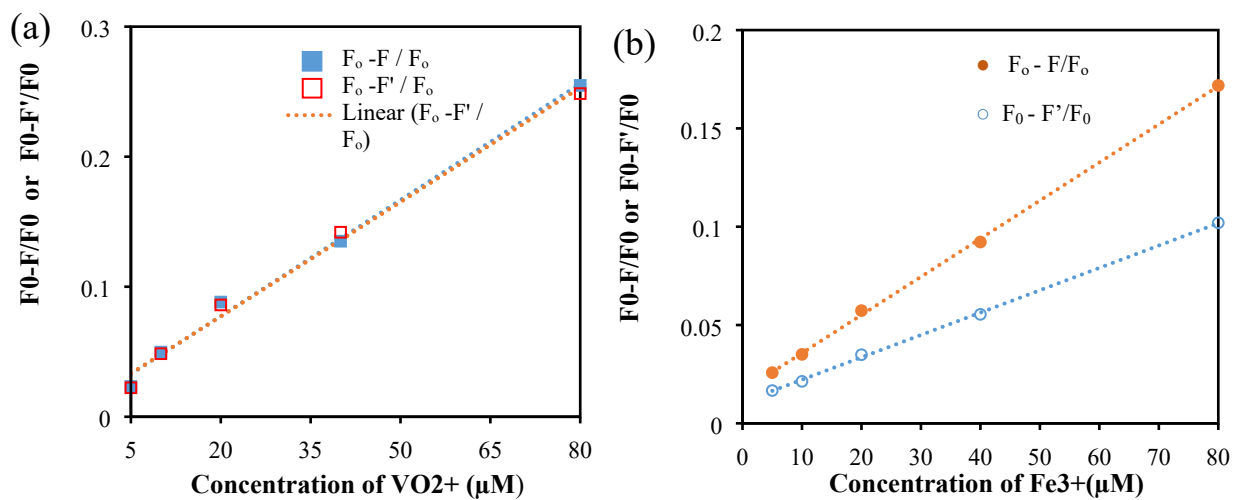


Fig. S6. Calibration curve before (solid symbol) and after (hollow symbol) adding EDTA (equal concentration of EDTA to VO<sup>2+</sup> and Fe<sup>3+</sup> concentration) to PPy-Cit-VO<sup>2+</sup> system (a), and PPy-Cit-Fe<sup>3+</sup> system (b); where F and F' are fluorescence intensity before and after adding EDTA to PPy-Cit-VO<sup>2+</sup> and PPy-Cit-Fe<sup>3+</sup>, respectively.

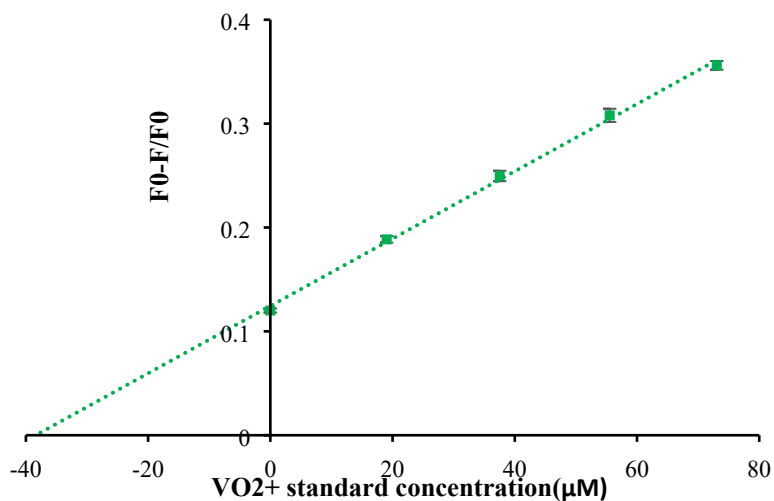


Fig. S7. Standard addition method for analysis of VO<sup>2+</sup> in vanadyl sulfate tablet (addition of 0, 25.6, 50.9, 75.9 and 100.5 μM). The error bars represent the triplicate measurements.

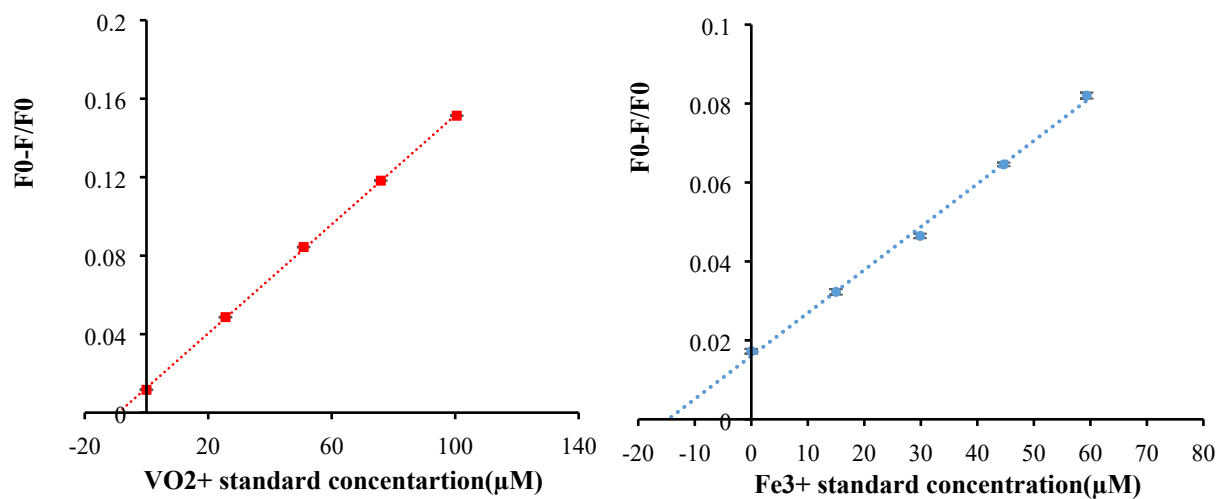


Fig. S8. Standard addition method for analysis of VO<sup>2+</sup> in serum 2 (addition of 0, 25.6, 50.9, 75.9 and 100.5 μM) (A) and Fe<sup>3+</sup> in 30μM spiked serum 6 (B) (addition of 0, 15, 22.7, 44.7 and 59.3μM)

**Table S1.** Relative response ( $F_0-F/F_0$ ) of PPy-Cit sensor to different concentrations of  $VO^{2+}$  in the presence of  $Fe^{3+}$  ions at various delay times.

Concentration ( $\mu M$ )		% Quenching after		Measurement ( $\mu M$ )		Recovery %	
$VO^{2+}$	$Fe^{3+}$	0.5 min	20 min	$VO^{2+}$ after 0.5 min	$Fe^{3+}$ after 20 min	$VO^{2+}$	$Fe^{3+}$
5	5	3.1	5.6	5.23 ( $\pm .11$ ) <sup>a</sup>	5.65 ( $\pm .01$ )	104.6	113
10	20	4.6	12.6	10.18 ( $\pm .09$ )	19.06 ( $\pm .05$ )	101.8	95.5
20	40	7.7	24	20.57( $\pm .04$ )	39.98( $\pm .1$ )	102.9	100.0

<sup>a</sup> Standard deviation for three replicate