

Supporting Information

Chiral diphenylacrylonitrile-perylene liquid crystal with circularly polarized luminescence in aggregated state

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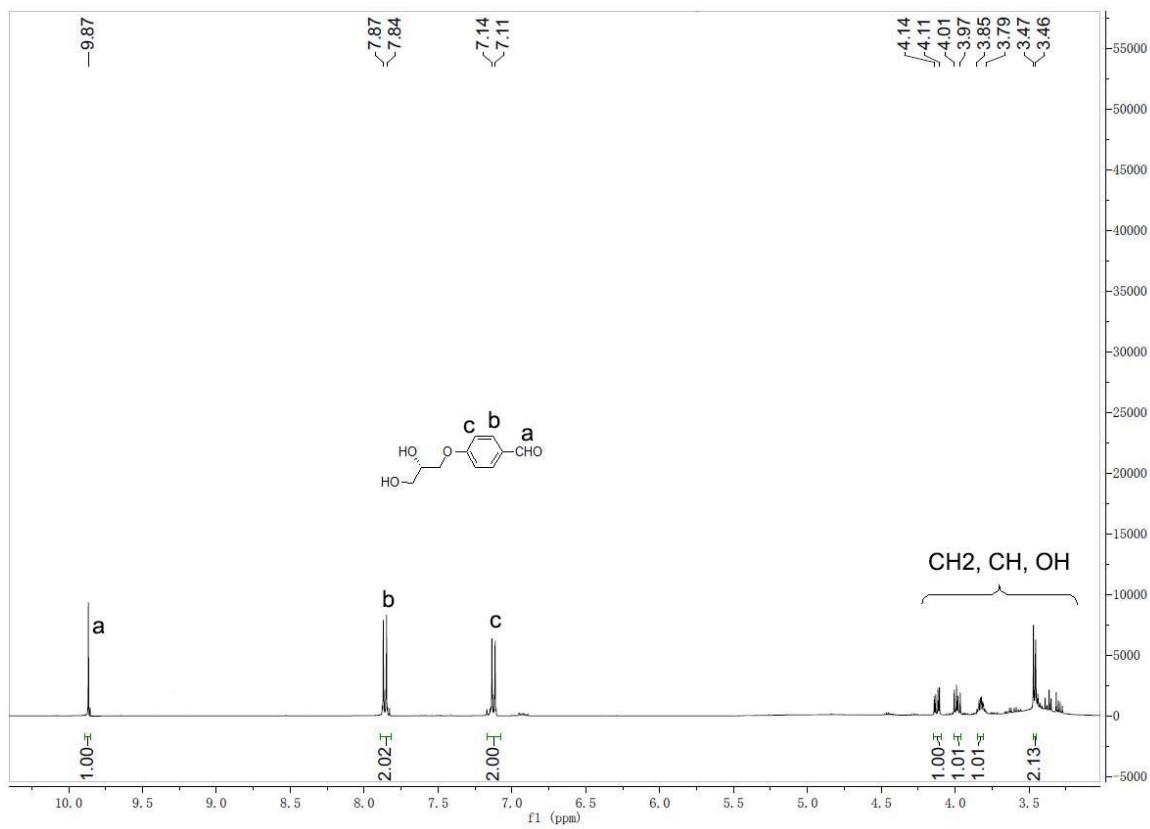


Figure S1. The ^1H NMR spectrum of compound 1

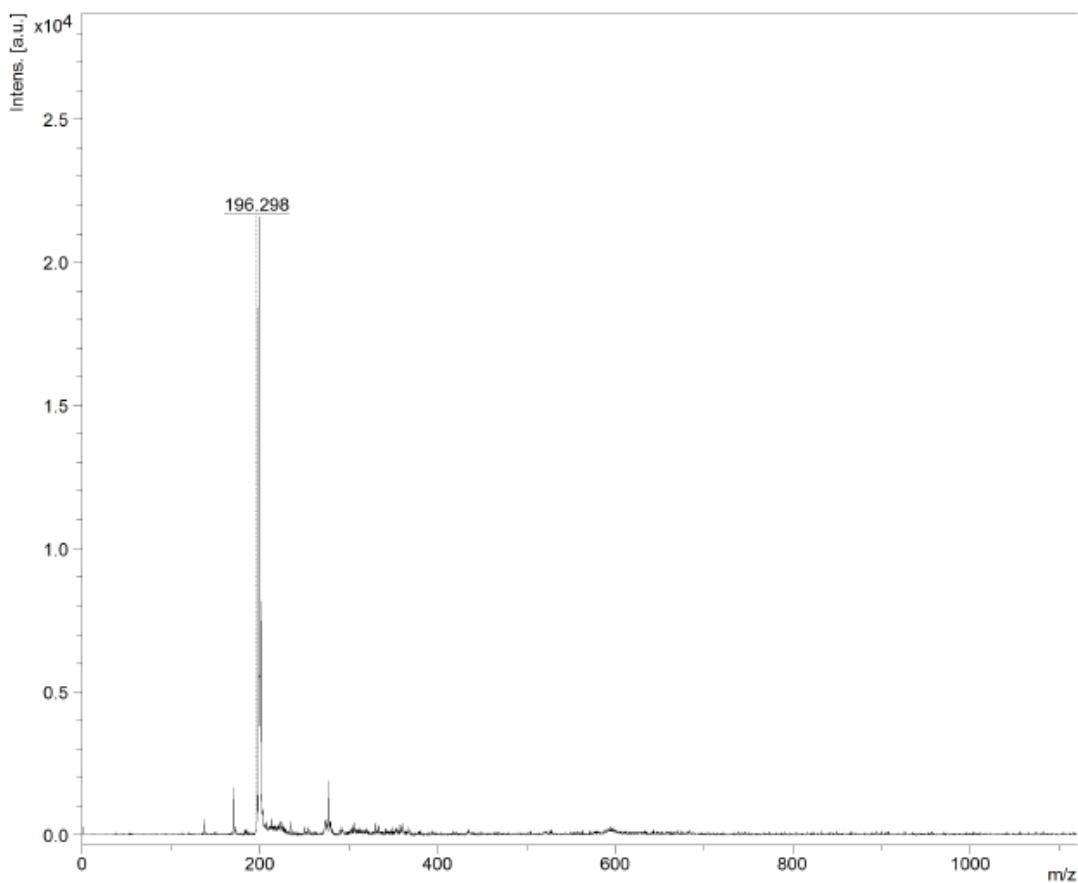


Figure S2. MALDI-TOF-MS spectrum of compound 1

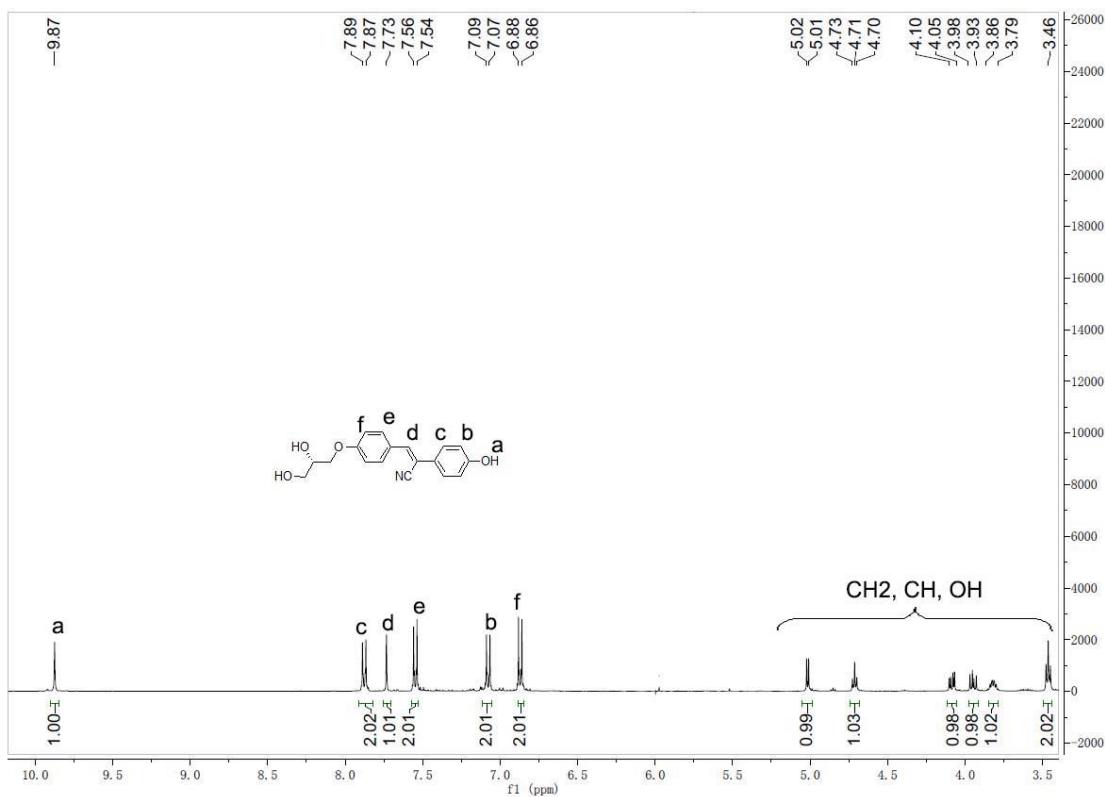


Figure S3. The ^1H NMR spectrum of compound 2

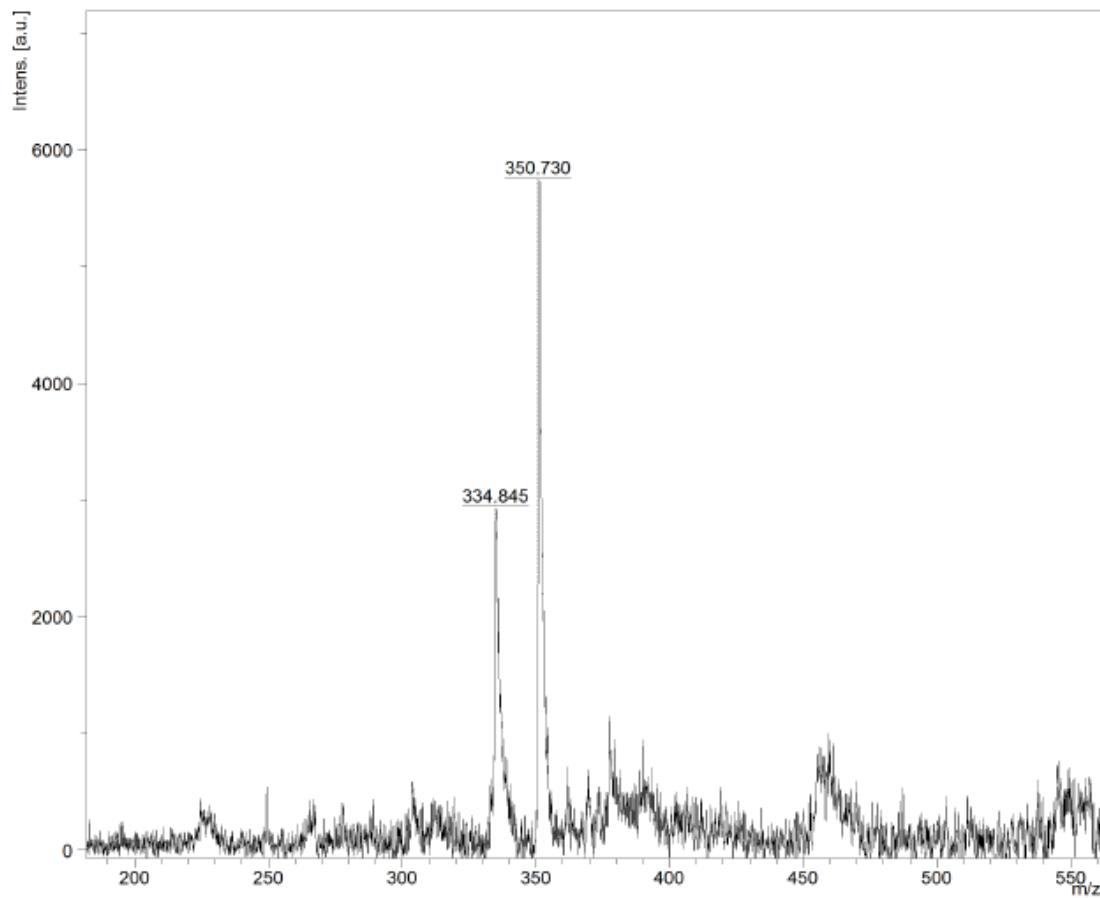


Figure S4. MALDI-TOF-MS spectrum of compound 2

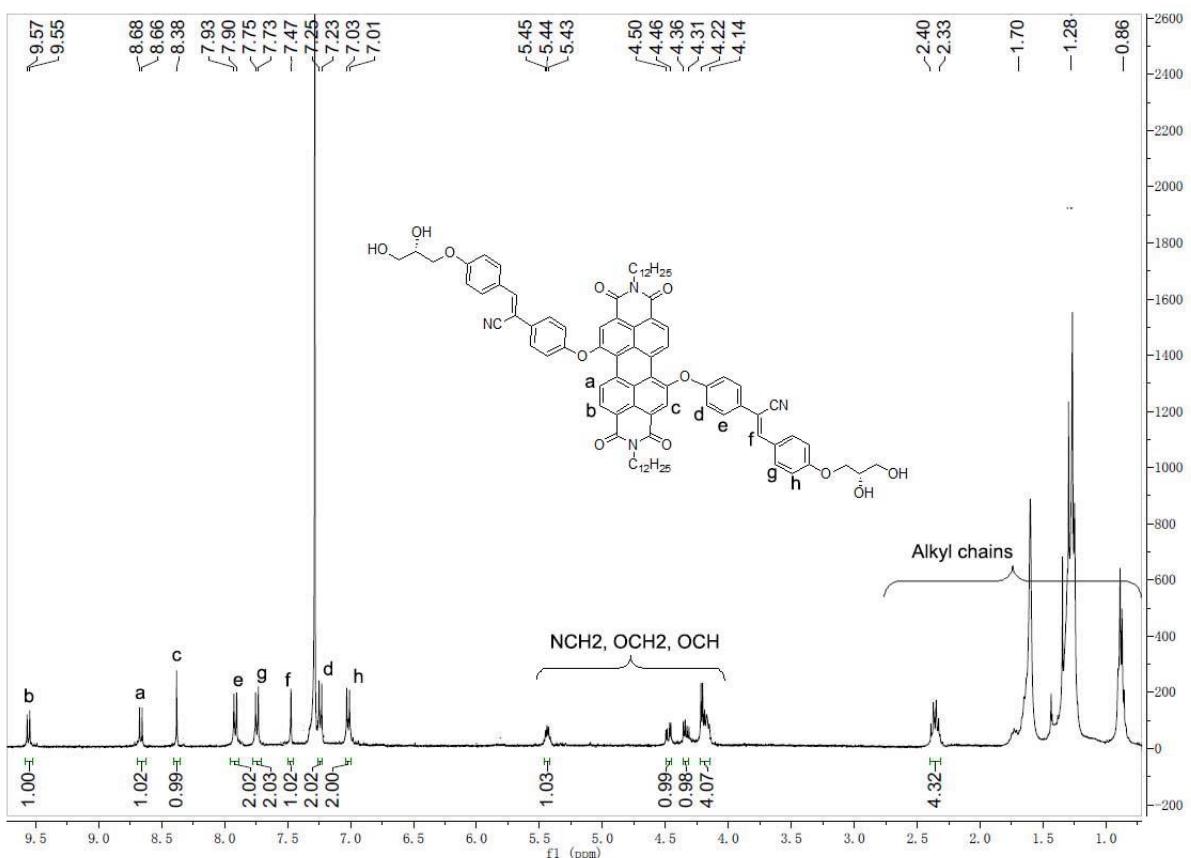


Figure S5. The ^1H NMR spectrum of compound 5

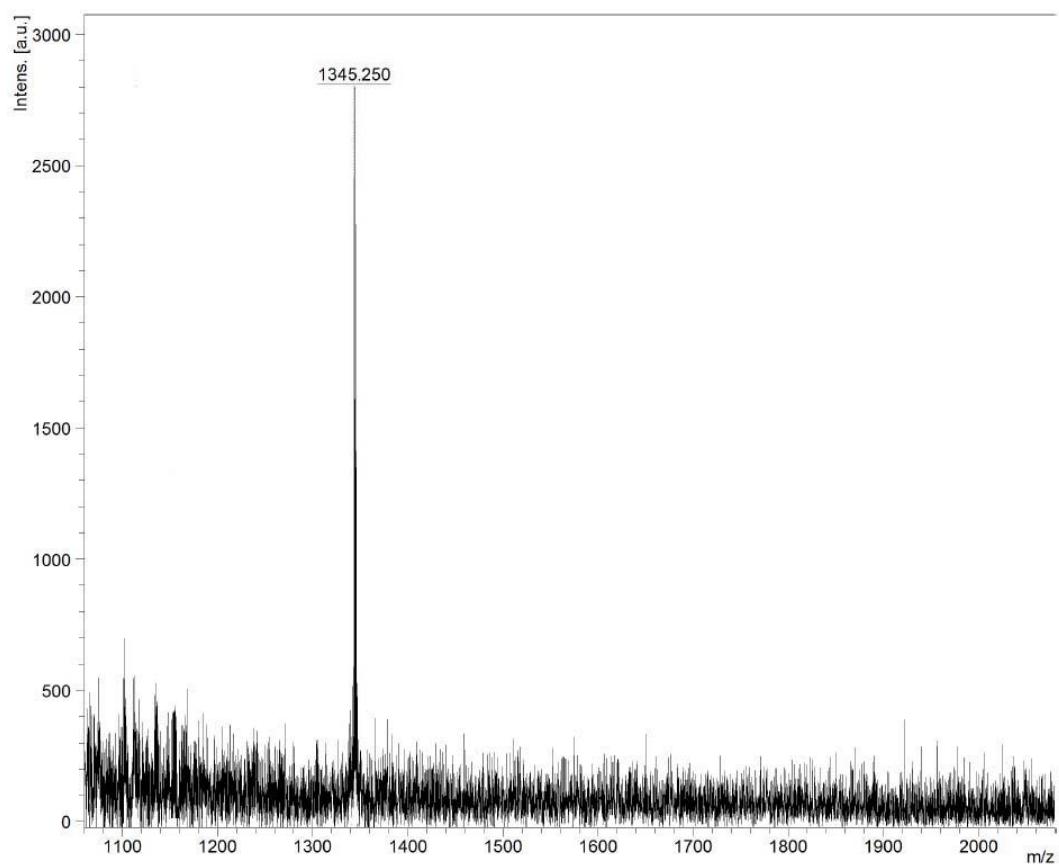


Figure S6. MALDI-TOF-MS spectrum of compound 5

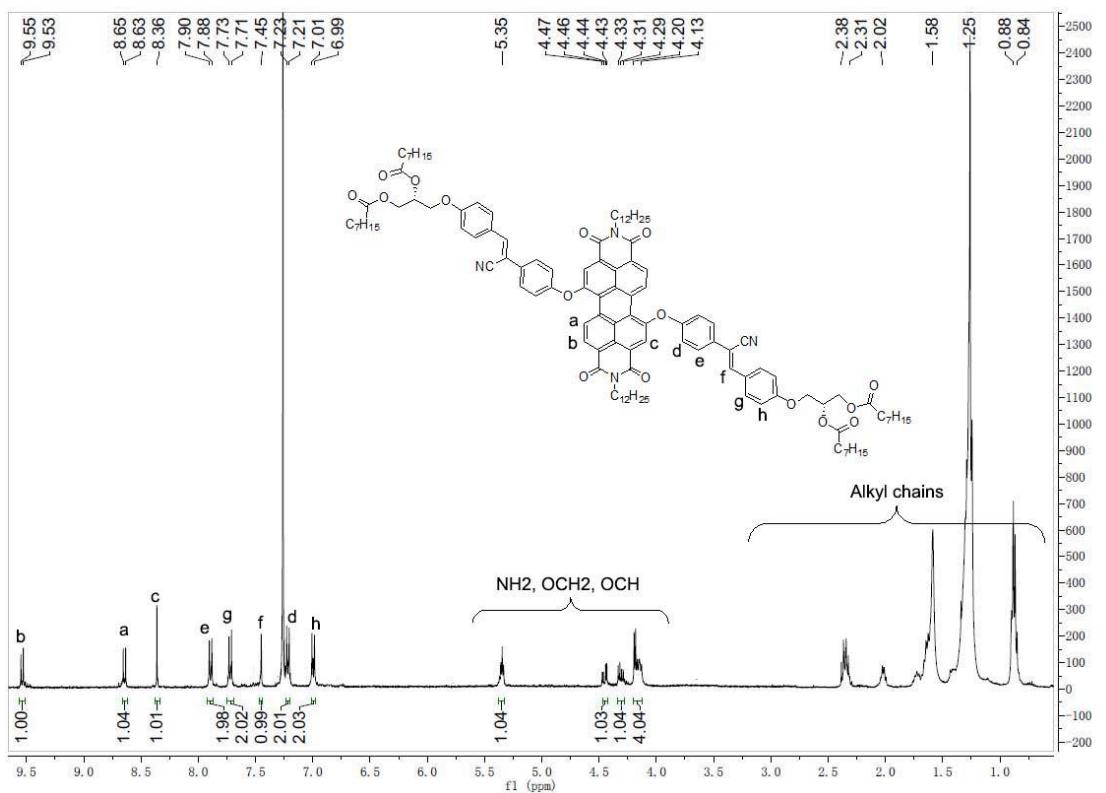


Figure S7. The ^1H NMR spectrum of CPL-P

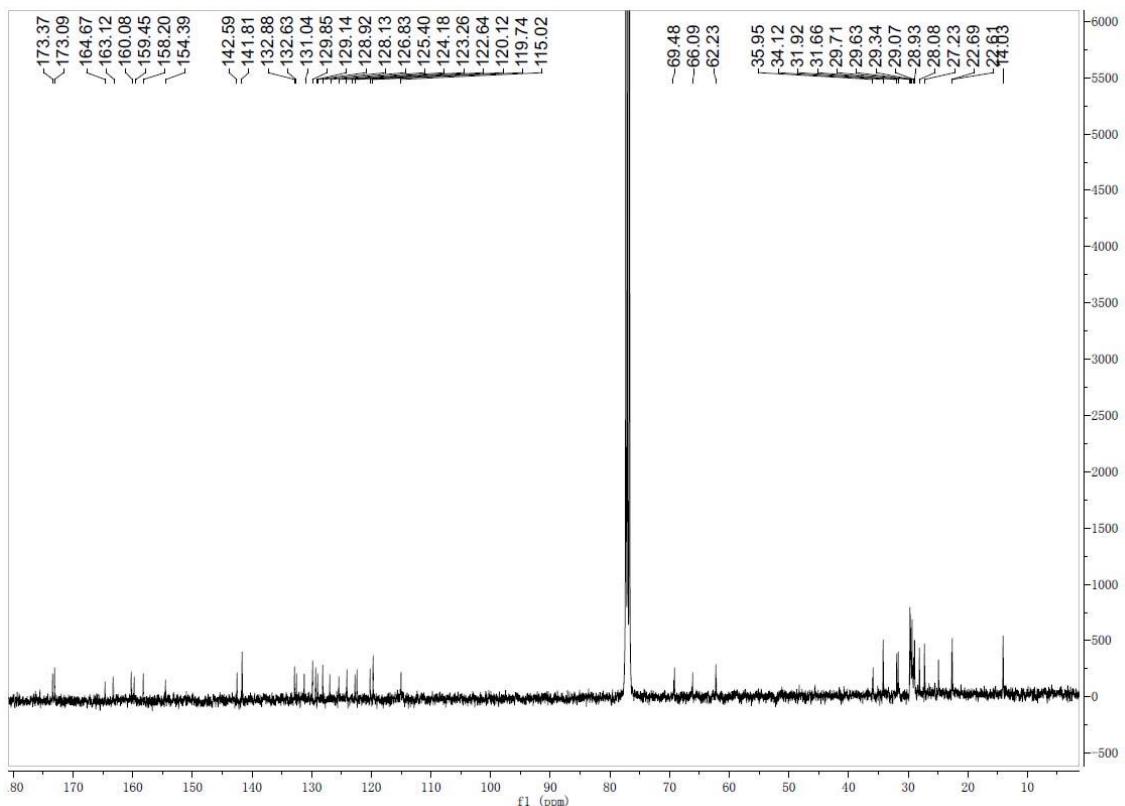


Figure S8. The ^{13}C NMR spectrum of CPL-P

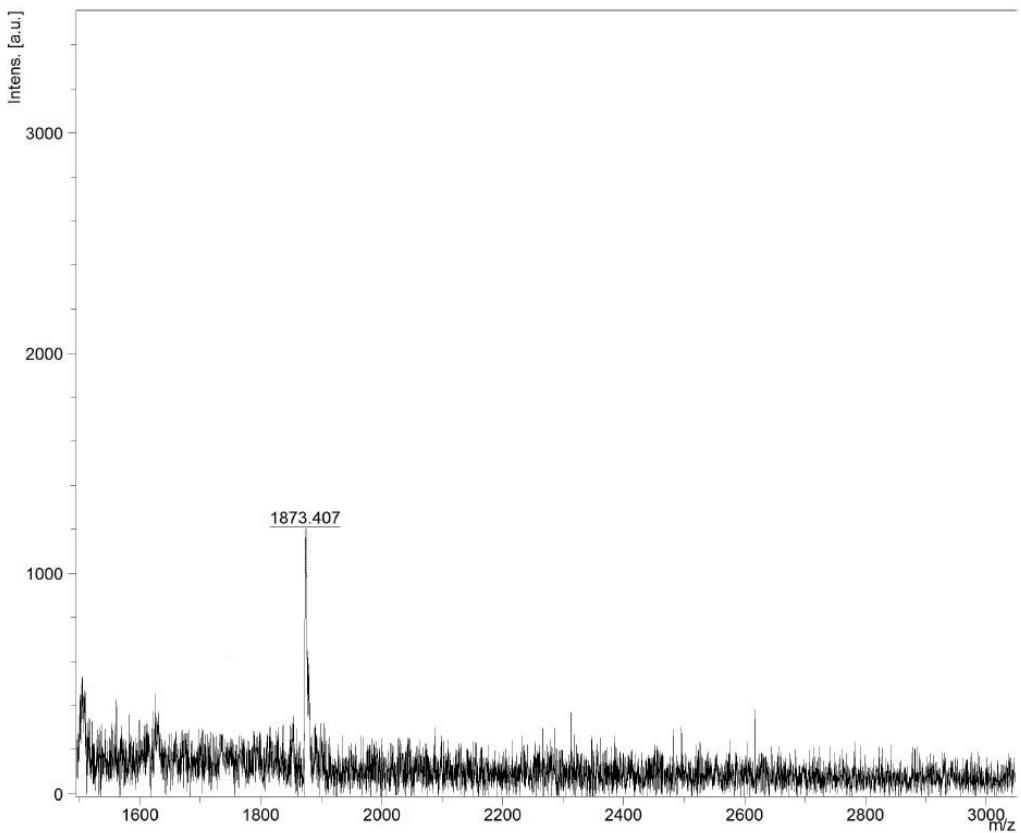


Figure S9. MALDI-TOF-MS spectrum of compound **CPL-P**

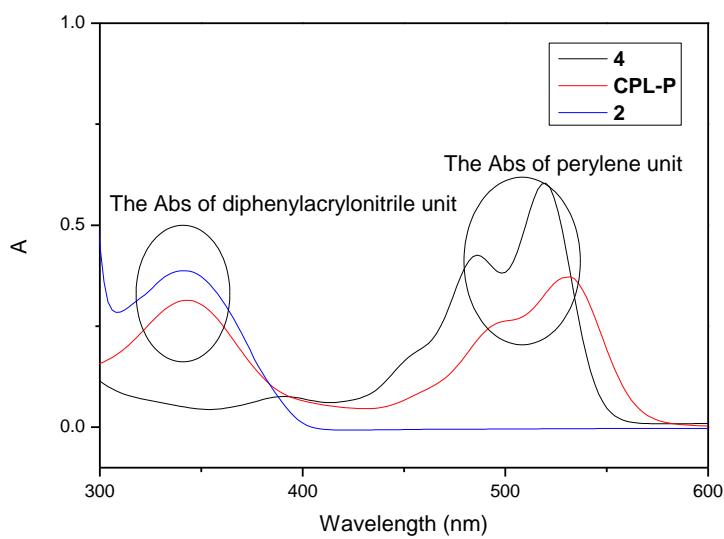


Figure S10 The absorption spectra of compounds **2**, **4** and **CPL-P** in THF solution (1×10^{-6} M)

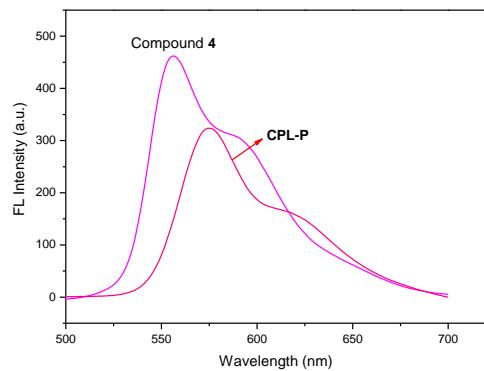


Figure S11 The emission spectra of compounds **5** and **CPL-P** in THF solution (2×10^{-6} M) with $\lambda_{\text{ex}} = 480$ nm.

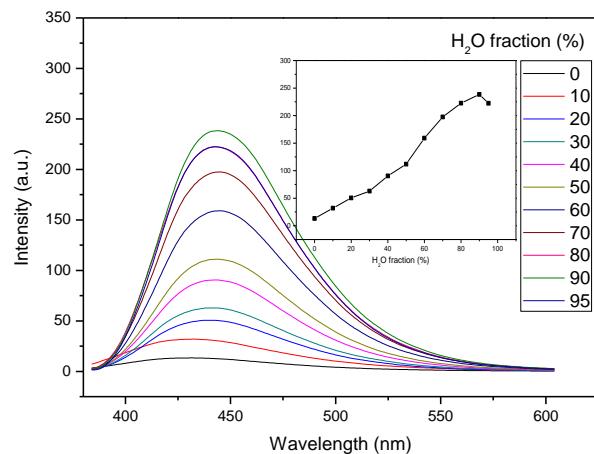


Figure S12 The emission spectra of precursor **2** with different fractions of H₂O in THF-H₂O system (5×10^{-6} M) with $\lambda_{\text{ex}} = 340$ nm. (inserted: Variation in intensity with fractions of H₂O in THF-H₂O system)

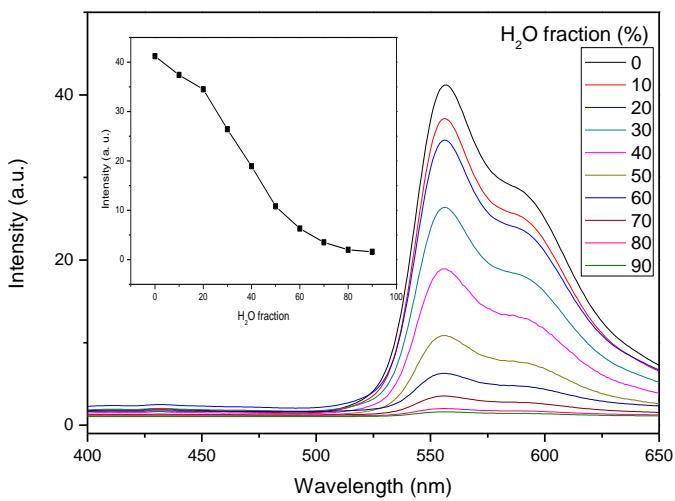


Figure S13 The emission spectra of sample **4** in THF/H₂O mixtures (2×10^{-6} M) with different H₂O fractions ($\lambda_{\text{ex}} = 340$ nm). (Inserted: Variation in intensity with H₂O fractions)

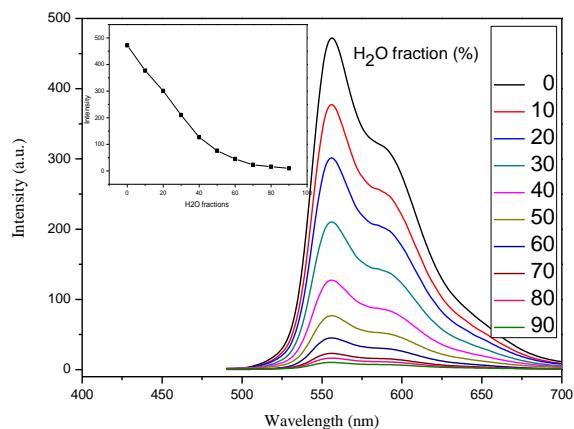


Figure S14 The emission spectra of sample **4** in THF/H₂O mixtures (2×10^{-6} M) with different H₂O fractions ($\lambda_{\text{ex}} = 480$ nm). (Inserted: Variation in intensity with H₂O fractions)

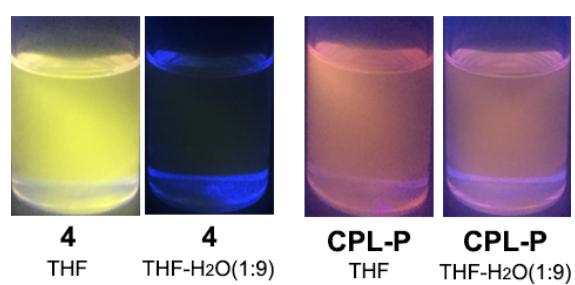


Figure 15 The fluorescence photographs of sample **4** and **CPL-P** under UV light ($\lambda_{\text{ex}} = 365 \text{ nm}$)

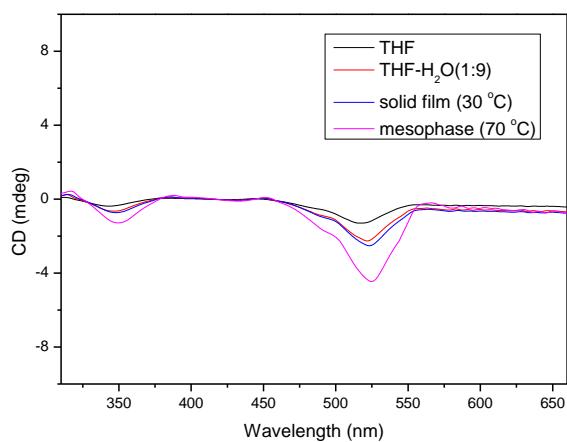
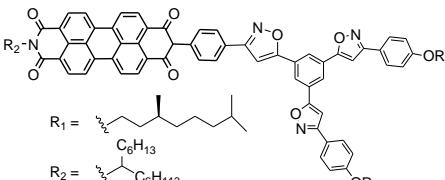
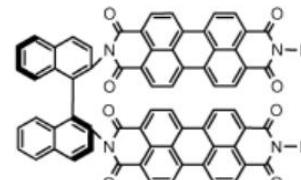
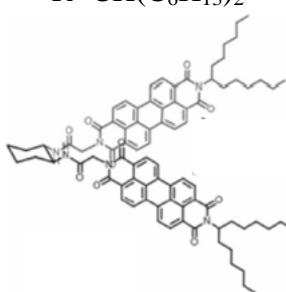
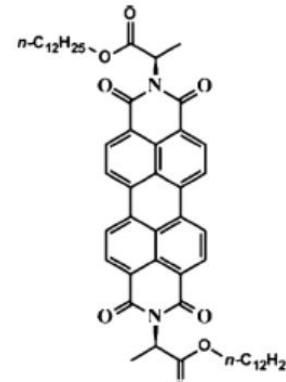
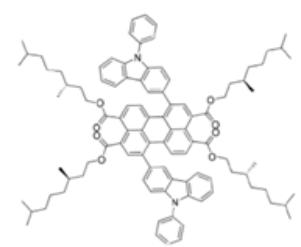
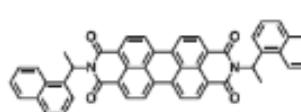
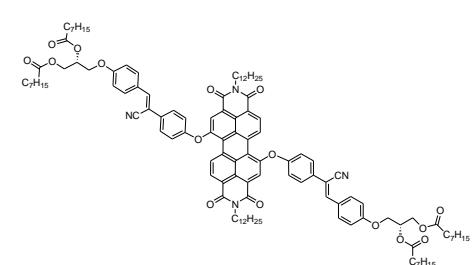
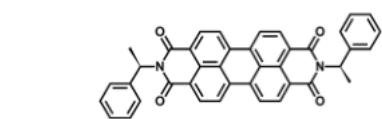


Figure S16. CD spectra of **CPL-P** in various phases

Table S1 Comparison of CPL properties for perylene derivatives

Molecular structure	CPL values in solution	CPL values in film	Reference
	Φ/g_{lum}	Φ/g_{lum}	
	0.07/0.007	not mentioned	T. Ikeda, T. Masuda, T. Hirao, J. Yuasa, H. Tsumatori, T. Kawai and T. Haino, <i>Chem. Commun.</i> , 2012, 48 , 6025-6027
	$0.88/3 \times 10^{-3}$	Very weak(no data mentioned)	H. Tsumatori, T. Nakashima and T. Kawai, <i>Org. Lett.</i> , 2010, 12 , 2362-2365
	$0.24/8 \times 10^{-3}$	0.035/not mentioned	J. Kumar, T. Nakashima, H. Tsumatori, M. Mori, M. Naito and T. Kawai, <i>Chem. Eur. J.</i> , 2013, 19 , 14090-14097
	0.67/0.02	not mentioned	F. Li, Y. Li, G. Wei, Y. Wang, S. Li and Y. Cheng, <i>Chem. Eur. J.</i> , 2016, 22 , 12910-12915
	$0.53/1.93 \times 10^{-4}$	$0.2/5.34 \times 10^{-4}$	J. Li, C. Yang, X. Peng, Q. Qi, Y. Li, W. Lai and W. Huang, <i>Org. Biomol. Chem.</i> , 2017, 15 , 8463-8471
	$0.06/3.9 \times 10^{-3}$	$0.04/1.7 \times 10^{-3}$	K. Watanab, A. Taniguchi, D. Kaji, N. Hara, T. Hosoya, A. Kanesaka, T. Harada, H. Nishikawa, Y. Imai, <i>Tetrahedron</i> , 2019, 75 ,

A. Taniguchi, D. Kaji, N.
 Hara, R. Murata, S.
 Akiyama, T. Harada, A.
 Sudo, H. Nishikawa, and Y.
 Imai, *RSC Adv.*, 2019, **9**,
 1976-1982.
 This work



0.76/not mentioned 0.09/2.0 $\times 10^{-3}$

0.30/5.6 $\times 10^{-4}$ 0.32/1.37 $\times 10^{-3}$