

Supplementary Information

The Pt/g-C₃N₄-CNS catalyst via in-situ synthesis process with excellent performance for methanol electrocatalytic oxidation reaction

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Preparation of carbon nanosheets:

The anhydrous sodium carbonate powder is used as the hard template, and the carbon nanosheets are prepared by the chemical vapor deposition method. First, the Na_2CO_3 powder is put into a ceramic boat, and the boat is placed in a tube furnace. The Ar was then passed into the tube furnace. The heating furnace is heated to $700\text{ }^\circ\text{C}$ at a heating rate of $10\text{ }^\circ\text{C}/\text{min}$. Hydrogen (H_2) and acetylene (C_2H_2) with a flow ratio of 5:1 are introduced into the furnace to deposit carbon nanosheets. The deposition reaction was maintained for 40 minutes, and then the furnace was naturally cooled to room temperature in an Ar- H_2 environment. The deposited product was washed with deionized water. Finally, powdered carbon nanosheets were obtained by drying the product at $80\text{ }^\circ\text{C}$ for 24 hours.

Activity calculation process

Specific activity (mA/cm^2) values were calculated from the effective area of the electrode (cm^2) and the measured current I (mA) :

$$\text{Specific activity} = \text{current density} = I/V$$

Mass activity ($\text{mA}/\text{mg}_{\text{Pt}}$) values were calculated from the electrocatalyst loading m (mg) and the measured current density j (mA/cm^2) :

$$\text{Mass activity} = j/m$$

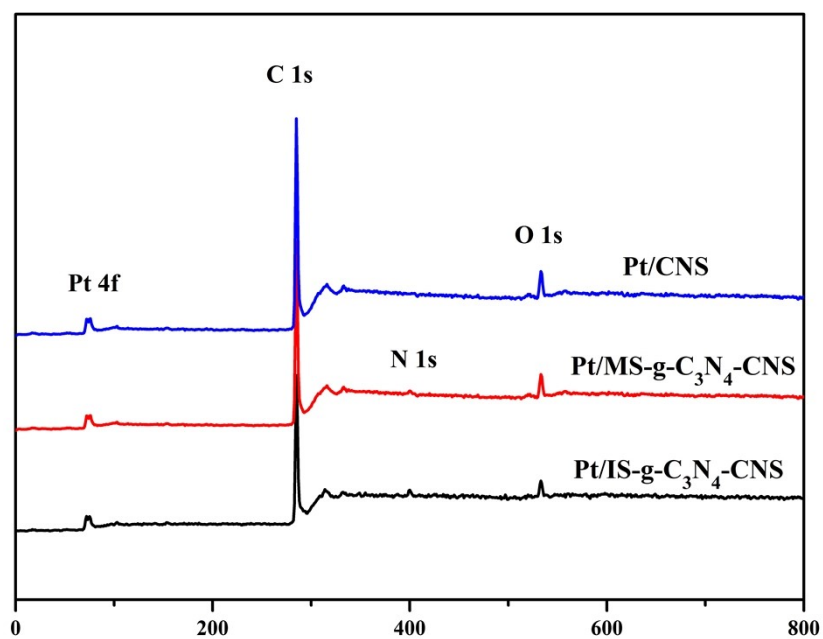


Fig. S1 XPS survey spectra of Pt/IS-g-C₃N₄-CNS, Pt/MS-g-C₃N₄-CNS and Pt/CNS.

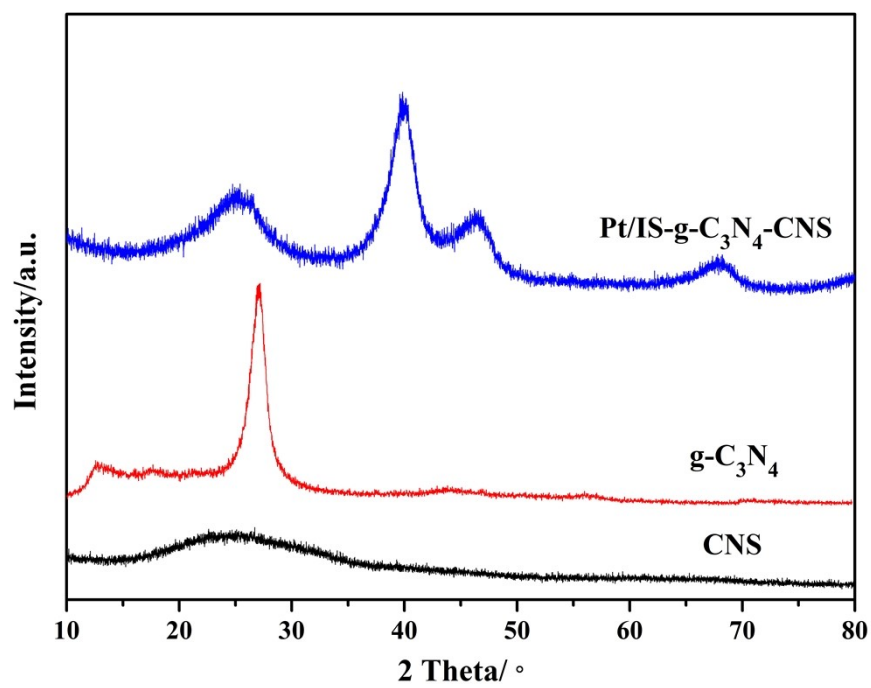


Fig.S2 The XRD patterns of Pt/IS-g-C₃N₄-CNS, g-C₃N₄ and CNS.

Table S1 XPS elemental analysis of Pt/g-C₃N₄-CNS samples.

Samples	C at.%	N at.%	O at.%	Pt at.%
Pt/IS-g-C ₃ N ₄ -CNS	90.09	3.57	5.31	1.02
Pt/MS-g-C ₃ N ₄ -CNS	90.36	2.89	5.62	1.13
Pt/CNS	93.18	--	5.58	1.24