

Supporting Information

Fabrication of Hollow ZrO_2 Nanoreactors Encapsulating Au- Fe_2O_3 Dumbbell Nanoparticles for CO Oxidation

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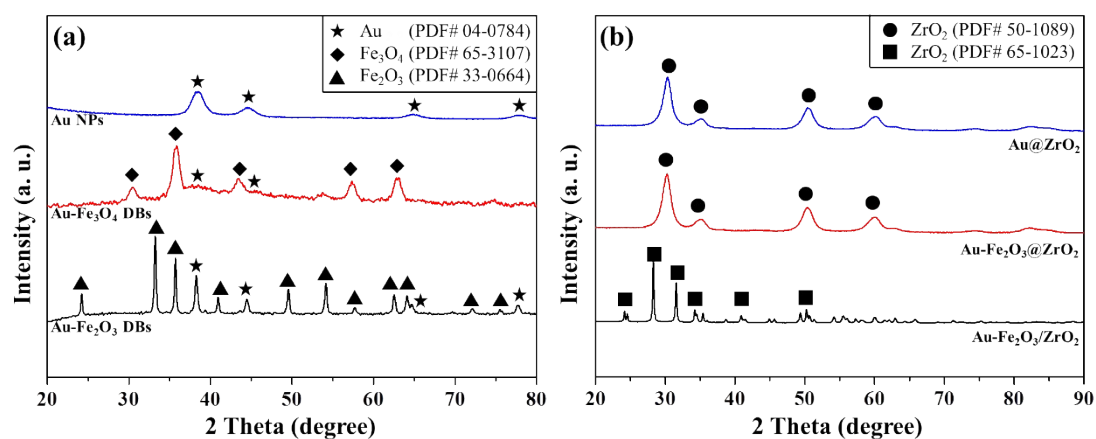


Fig. S1 XRD diffraction patterns of (a) Au NPs, Au- Fe_3O_4 DBs, and Au- Fe_2O_3 DBs obtained by calcination of Au- Fe_3O_4 at 700 °C; (b) Au@ ZrO_2 , Au- Fe_2O_3 @ ZrO_2 and Au- Fe_2O_3 / ZrO_2 catalysts.

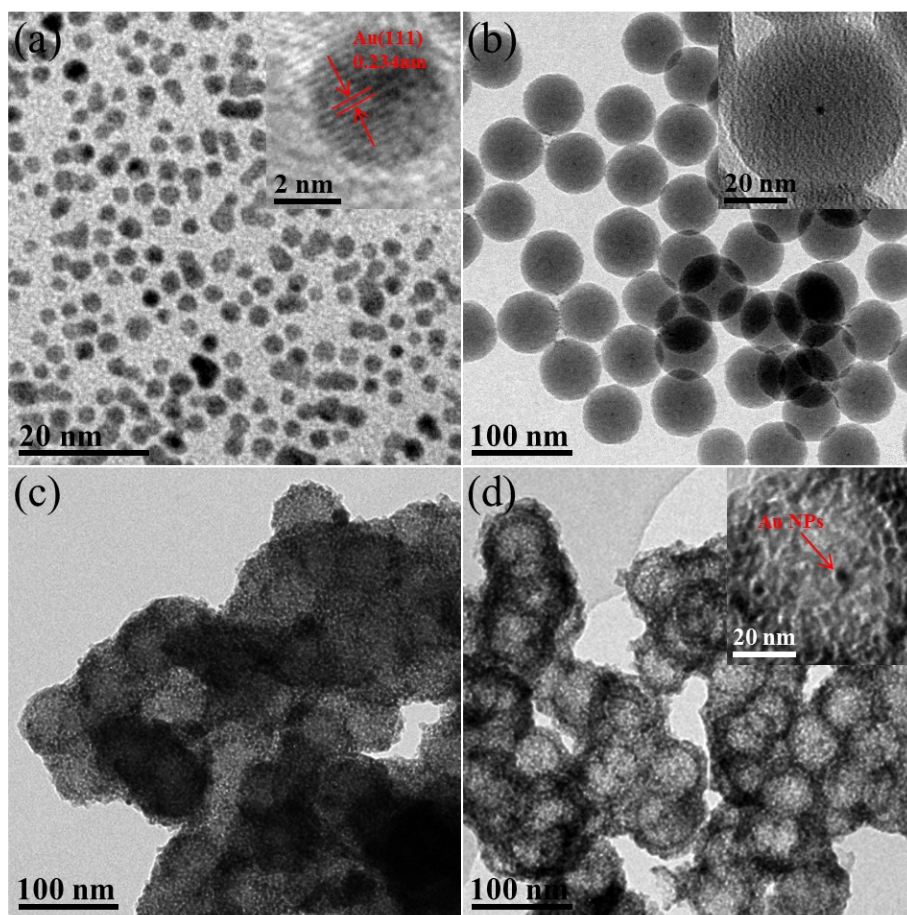


Fig. S2 TEM images show (a) Au NPs, (b) Au@SiO₂, (c) Au@SiO₂@ZrO₂, (d) Au@ZrO₂ yolk-shell nanoreactors. The insert of (a) show the HRTEM images of Au NPs with scale of 2 nm and TEM images of Au@SiO₂ and Au@ZrO₂ yolk-shell NPs with scales of 20 nm are shown in the insert of (b) and (d), respectively.

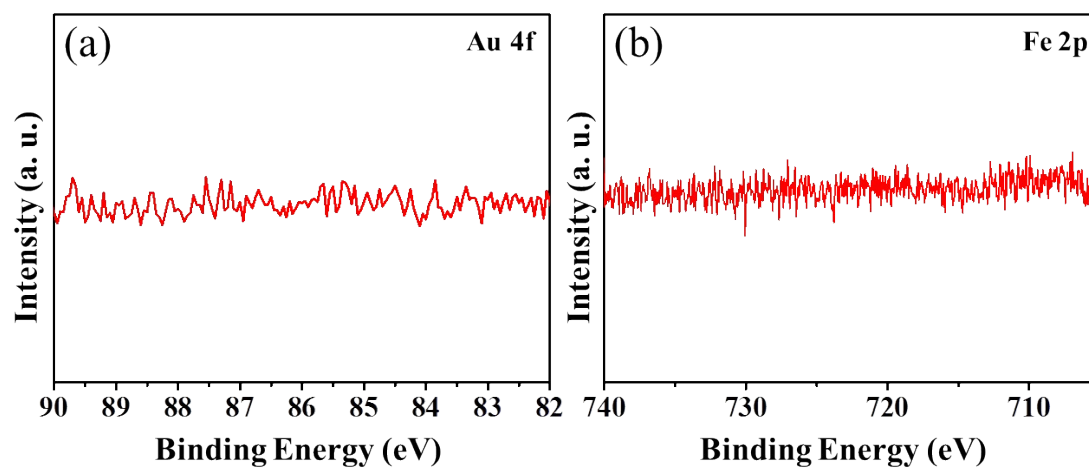


Fig. S3 (a) Au 4f spectra for Au-Fe₂O₃@ZrO₂ nanoreactors; (b) Fe 2p spectra for Au-Fe₂O₃@ZrO₂ nanoreactors.

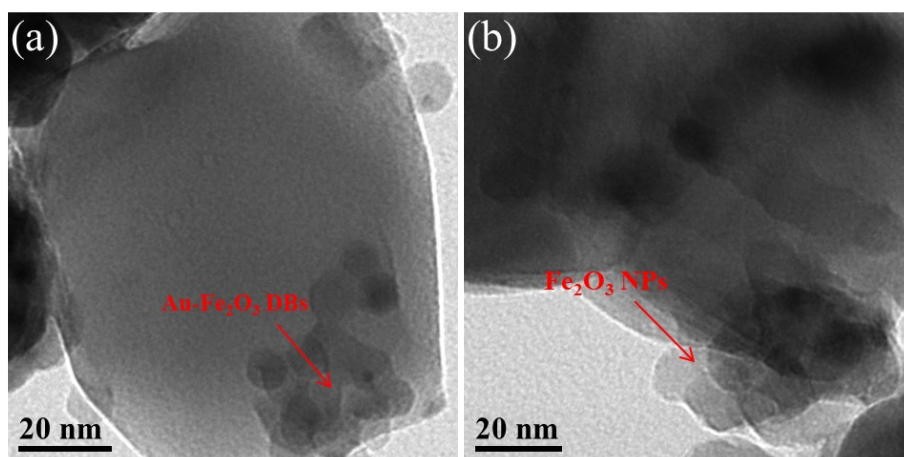


Fig. S4 TEM images show (a) Au-Fe₂O₃/ZrO₂ and (b) Fe₂O₃/ZrO₂ supported catalysts.

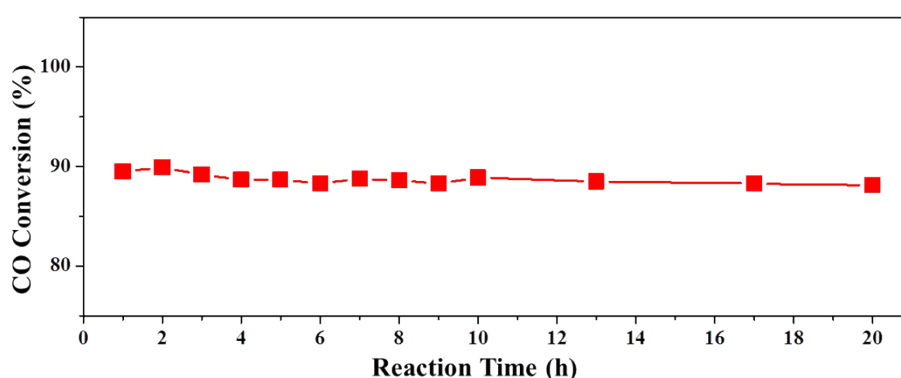


Fig. S5 Long-term stability of CO oxidation over Au-Fe₂O₃@ZrO₂ yolk-shell nanoreactors.

Table S1 Physical properties and catalytic performance of Au catalysts.

Cat.	Au loading/wt. %	Au d/nm	T ₅₀ /°C	TOF/h ⁻¹	Ref.
Au-Fe ₂ O ₃ @ZrO ₂	0.1	2.5	~ 85	1099.16	This work
Au/TiO ₂	1.0	3.8	-6	334.15	S1
Au/Fe ₂ O ₃	3.7	2.7	-16	64.17	S2
Au/Ni _{0.05} Al	3	2.4	13	281.39	S3
Au/TiO ₂ @ZrO ₂	7.5	8	60	375.18	S4
Au@CeO ₂	0.93	17	80	1205.5	S5
Au@ZrO ₂	6.8	14	150	724.15	S6
Au/ZnO	3.8	1.7	<-90	39.34	S7
Au/α-Fe ₂ O ₃	1.86	2.3	<20	543.67	S8
Au-CeO ₂ @ZrO ₂	0.55	15	200	19185.39	S9

TOF ≈ r (CO conversion) • d (Au diameter)^{S1}

r (CO conversion) defined as moles of CO converted per mol of total Au per hour at T₅₀.

References

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