

Supplementary Information

Air-stable, efficient electron doping of monolayer MoS₂ by salt–crown ether treatment

Hiroto Ogura,^a Masahiko Kaneda,^a Yusuke Nakanishi,^a Yoshiyuki Nonoguchi,^b
Jiang Pu,^c Mari Ohfuchi,^d Toshifumi Irisawa,^e Hong En Lim,^a Takahiko Endo,^a
Kazuhiro Yanagi,^a Taishi Takenobu^c and Yasumitsu Miyata^{*a}

^a *Department of Physics, Tokyo Metropolitan University, Hachioji, 192-0397, Japan*

^b *Faculty of Materials Science and Engineering, Kyoto Institute of Technology, Kyoto
606-8585, Japan*

^c *Department of Applied Physics, Nagoya University, Nagoya, 464-8603, Japan*

^d *Fujitsu Laboratories Ltd., Atsugi, 243-0197, Japan*

^e *Nanoelectronics Research Institute, AIST, Tsukuba, 305-8562, Japan*

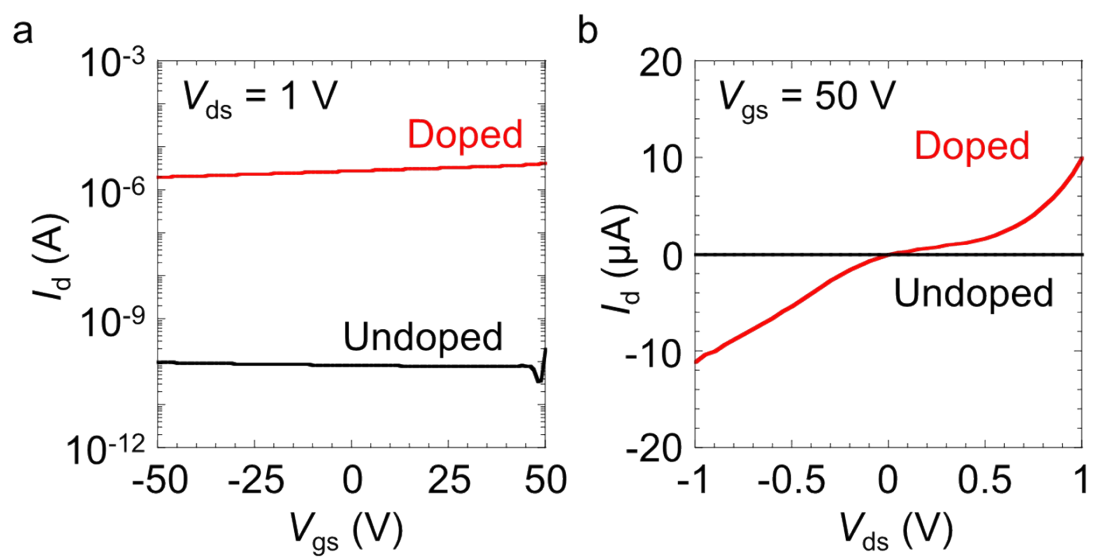


Figure S1. (a, b) The transfer (a) and output (b) curves for undoped and doped monolayer WS₂ using 100 mM dopant solution.

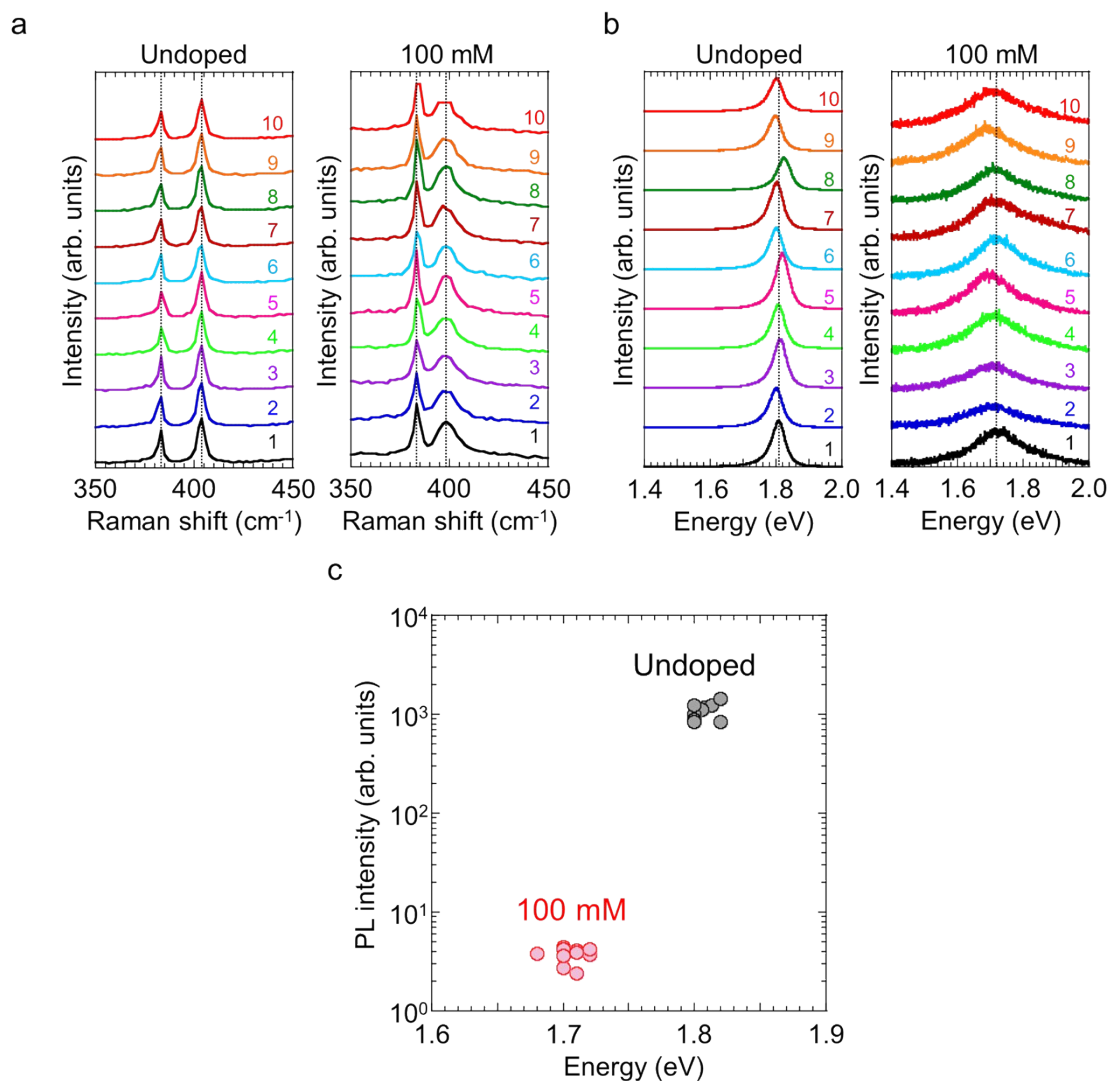


Figure S2. (a, b) Raman (a) and PL (b) spectra of undoped and doped MoS₂ crystals taken at different 10 points. (c) Distributions of PL peak energy and intensity of undoped and 100 mM doped MoS₂ taken at different 10 points.

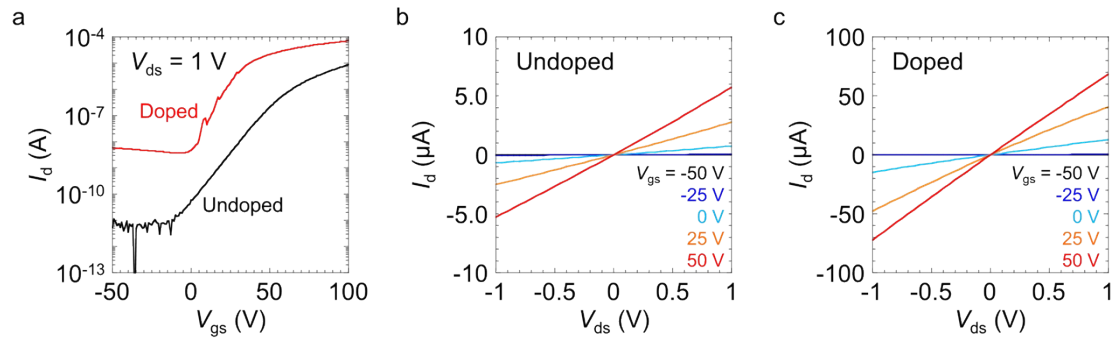


Figure S3. (a) Transfer curves of the device with photoresist coating only in the middle of the channel before and after the doping using 100 mM dopant solution drawn on a logarithmic scale. (b, c) Output curves of the device with a photoresist coating only in the middle of the channel: (b) before and (c) after the doping process using 100 mM dopant solution.

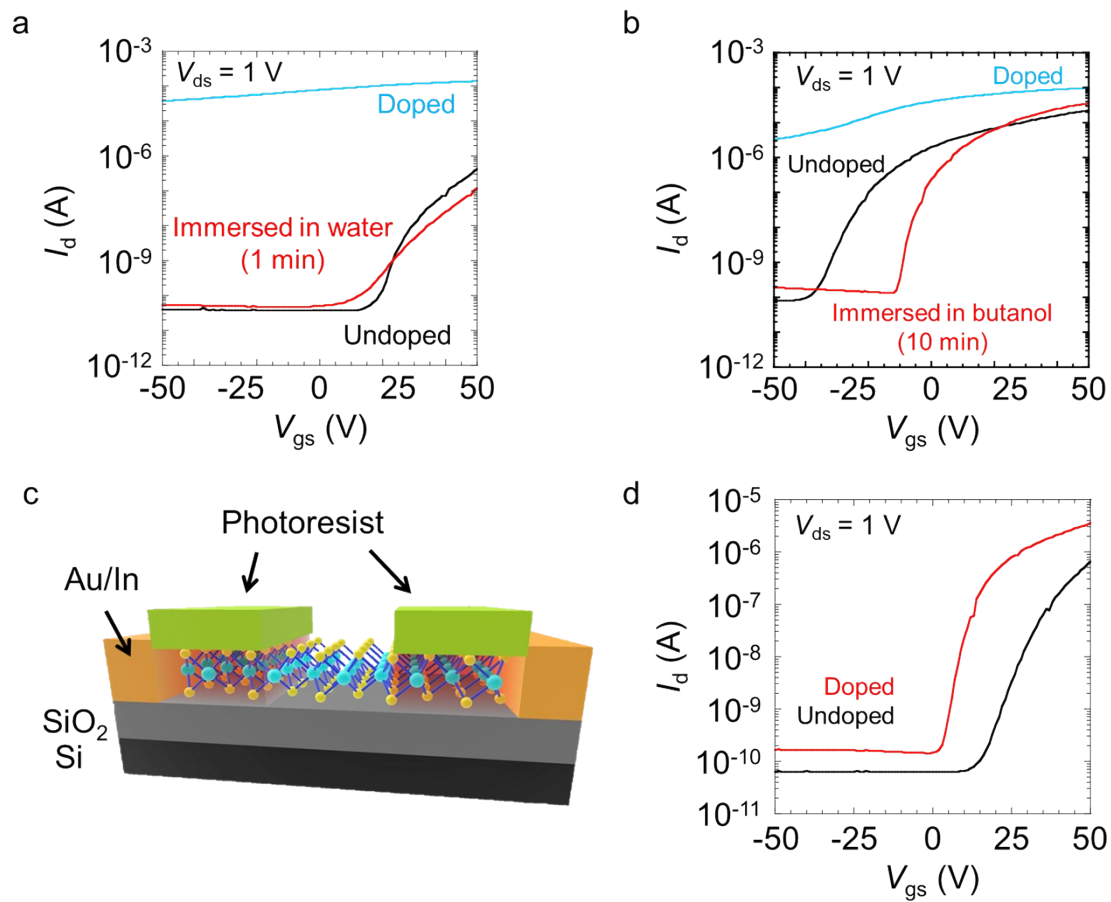


Figure S4. (a, b) Reversible transfer characteristics of a KOH/benzo-18-crown-6-doped device using 100 mM (a) or 10mM (b) dopant solution. The transfer curves returned to those of undoped sample after immersion in water (a) or butanol (b). (c, d) Schematic (c) and transfer characteristics (d) of an n⁺/i/n⁺ patterned device using 100 mM dopant solution with photoresist coating only close to the source and drain electrodes.

Table S1. Comparison of the carrier density and air stability of chemically doped monolayer (ML) and few-layer (FL) MoS₂ using different n-type dopants. The carrier densities were estimated by Hall effect for this work, by the electric-field effect with the parallel plate model for Ref. 1, 2, 5, and the shift of threshold voltage for Ref. 3, 4, 6.

	Dopants	Materials	Carrier density (cm ⁻²)	Air stability
This work	KOH/benzo-18-crown-6	ML MoS ₂	$\sim 3.4 \times 10^{13}$	~ 24 days
Fang <i>et al.</i> ¹	Potassium	FL MoS ₂	$\sim 1.0 \times 10^{13}$	
Kiriya <i>et al.</i> ²	Benzyl viologen	FL MoS ₂	$\sim 1.2 \times 10^{13}$	~ 9 days
Andleeb <i>et al.</i> ³	<i>p</i> -Toulene sufonic acid	FL MoS ₂	$\sim 2.5 \times 10^{12}$	~ 10 days
Rai <i>et al.</i> ⁴	Amorphous titanium suboxide	ML MoS ₂	$\sim 7.4 \times 10^{12}$	~ 30 days
Rosa <i>et al.</i> ⁵	Poly(vinyl-alcohol)	FL MoS ₂	$\sim 4.0 \times 10^{12}$ ($\sim 8.0 \times 10^{12}$ after annealing)	~ 16 hours (> 30 days with Al ₂ O ₃ encapsulation)
Zhang <i>et al.</i> ⁶	Pentamethylrhodocene dimer	ML MoS ₂	$\sim 2.4 \times 10^{12}$	

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