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Electronic Supplementary Information

Biocompatible Hole Scavenger–Assisted Graphene Oxide

Dots for Photodynamic Cancer Therapy

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1. Topographic analysis of NGODs



Fig. S1 (a) AFM image of the NGODs distributed on a mica substrate. (b) Height profile along the line in panel

a

2. Full-range XPS spectrum of NGODs



Fig. S2 Full-range XPS spectrum of NGODs.

3. Chemical composition of NGODs

Table S1 Atomic ratios of (N 1s)/(C 1s) and (O 1s)/(C 1s) determined from the full-range XPS spectra (Fig. S2), nitrogen functionality composition determined from the N 1s XPS spectrum (Fig. 1c), and carbon bonding composition determined from the C 1s XPS spectrum (Fig. 1d) for NGODs.

Atomic ratio (%)	Nitrogen functionality composition (% of C1s)				
N 1s / C 1s	Pyridine	Amino	Pyrrolic	Quaternary	Amide
16	0.47	1.9	4.4	4.7	4.4
Atomic ratio (%)	Carbon bonding composition (%)				
O 1s / C 1s	C–C	C–N	С–О	C=O	О-С=О
58	56	13	5.5	21	5.5



Fig. S3 FTIR absorption spectrum of NGODs.



Fig. S4 UPS spectrum of NGODs. The electron onset binding energy values, E_{B1} and E_{B2} , were determined from the intercepts of the extrapolated tangent lines on the abscissa at low and high binding energies, respectively. The UPS width constituted the difference between E_{B1} and E_{B2} . The spectrum was obtained under He I light irradiation at 21.2 eV.

6. Spectrum of PDT light source



Fig. S5 Spectrum of tungsten–halogen incandescent light, obtained from the Carl Zeiss Microscopy Online C ampus website (http://zeiss-campus.magnet.fsu.edu/articles/lightsources/tungstenhalogen.html).

7. H₂O₂ generation from GO and NGOD suspensions



Fig. S6. H_2O_2 generation from the GO and NGOD suspensions containing 0.2 mM AA and 10 µg mL⁻¹ of the GO or NGODs. The irradiation was conducted using the tungsten–halogen incandescent light at 40 mW cm⁻² for 10 min. The plot presents the mean value of the data, whereas the error bars on each plot represent the standard error of the mean. The P values are indicated as *, **, and *** for P < 0.05, P < 0.01 and P < 0.001, respectively.

8. Viability of the cancer and normal cells under H₂O₂ treatment



Fig. S7 Viabilities of the cancer (PC-9 and OECM-1) and normal (L929 and hNOK) cells after treatments with H_2O_2 of various dosages. Viability was analyzed using the MTT assay 72 h after the treatments. The plot presents the mean value of the data, whereas the error bars on each plot represent the standard error of the mean. The P values are indicated as *, **, and *** for P < 0.05, P < 0.01 and P < 0.001, respectively.



Fig. S8 Population variation of the viable cells (encompassed by quadrant [-,-]) with time according to the data of Fig. 4b, in which AnnV/PI assays were used for analysis. The plot presents the mean value of the data, whereas the error bars on each plot represent the standard error of the mean.



Fig. S9 EPR spectra of TEMPL solutions for ${}^{1}O_{2}$ generation analysis. (a) Nonirradiated and irradiated TEMPL solution. (b) Nonirradiated and irradiated NGOD–TEMPL solution. (c) Nonirradiated and irradiated NGOD–TEMPL–AA solution. The inset of panel (b) compares the spectra of the irradiated NGOD, RB, and MB solutions containing TEMPL. The concentrations of NGODs, RB, and MB were 10 μ g mL⁻¹, that of AA was 0.1 mM, and that of TEMPL was 85 mM.

12. EPR for O₂^{-•}, HO₂•, and OH• generation analysis



Fig. S10 EPR spectra of DMPO solutions for $O_2^{-\bullet}$, HO_2^{\bullet} , and OH^{\bullet} generation analysis. (a) Nonirradiated and irradiated NGOD–DMPO–AA solution. (b) Nonirradiated and irradiated NGOD–DMPO–AA solution. The concentrations of NGODs were 10 µg mL⁻¹, that of AA was 0.1 mM, and that of DMPO was 100 mM.

12. Amplex[®] Red method for H₂O₂ quantification



Fig. S11 Schematic of reaction process using Amplex[®] Red method to quantify the amount of produced H₂O₂.