Dual-Functional Porous MOFs with Hierarchical Guest Encapsulation for Room-Temperature Phosphorescence and White-Light-Emitting

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Figure S1. PXRD of PCN-921:as-synthesized (black line) and stimiulate (red line). Some diffraction peaks of the simulated curve are absent in the as-synthesized curve due to the crystal orientation. This phenomenon is consistent with the previous literature.^[1]



Figure S2. IR of H₈ETTB (red line) and PCN-921 (black line).



Figure S3. PXRD of PCN-921 MOF in different solvents.



Figure S4. TGA of PCN-921 MOF before (black line) and after (blue lline) acetonitrile-exchanged.



Figure S5. Fluorescence lifetime 3.78 ns of PCN-921 at Ex: 300 nm.



Figure S6. Quantum yield 93.6% of PCN-921 MOF.



Figure S7. PXRD of coronene.



Figure S8. Phosphorescent lifetime 3.72 ns of PCN-921 MOF.



Figure S9. The instantaneous phosphor of PCN-921. There is almost no long afterglow.



Figure S10. (a) fluorescence of package x mg-coronene@PCN-921 with different contents. (b) CIE: 5mg coronene@PCN-921 (0.30,0.39) (Brick red), 10mg coronene@PCN-921 (0.28,0.38) (Black), 15mg coronene@PCN-921 (0.26,0.36) (Origin), 20mg coronene@PCN-921 (0.25,0.29) (Red).



Figure S11. Phosphorescence of different amounts of coronene were capsulated.



Figure S12. PXRD of PCN-921 (black line), 5mg coronene@PCN-921 (red line) and 0.06wt% RhB@coronene@PCN-921 (blue line).



Figure S13. ¹H NMR spectra of coronene, RhB, PCN-921, and RhB@coronene@PCN-921 contained DMSO-d6.



Figure S14. IR of coronene, RhB, PCN-921, and RhB@coronene@PCN-921.



Figure S15. SEM-EDS mapping of RhB@coronene@PCN-921.



Figure S16. TGA of RhB, 5mg coronene@PCN-921 and RhB@coronene@PCN-921.



Figure S17. UV-Vis leaching test of RhB@coronene@PCN-921.

Param	Valuen s	Std. Dev.ns	Param	Value	Std. Dev.	Rel.%
τ_1	2.4750	0.06896	B1	1744.662	90.1142	41.20
τ_2	4.6931	0.08466	B2	1312.974	94.8295	58.80
			А	0.182		
			χ^2	1.124		

Table S1. Fluorescence lifetime 3.78 ns of PCN-921 MOF.

Fitting Range Low 198

Fitting Range High 1640

Table S2. The encapsulated	guest molecules in RTP and	WLED MOFs materials.
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Name	Ligand	Coordinated	Guests	Phospho	LED(Q	Ref.
		metal ions		r-	Y)	
				escence		
				lifetime		
				(ms)		

Zn(II) based		Zn ²⁺	-	2000(10	-	2
coordination				K)		
polymer (CP)						
Eu-Cd-CP		Zn^{2+}	Eu ³⁺	454(RT)	-	3
	ноос	Cd^{2+}			_	3
{(H ₂ Bpy)[Cd ₃	ноос соон	Cd^{2+}	H ₂ Bpy ²⁺ (H	Two	-	4
$(BTC)_2] \cdot 2H_2O$			₂ Bpy=	months		
} _n	НООС СООН		deprotonat	(RT)		
			ed 4,4'-			
			bipyridin)			
ZIF-8 and		Zn^{2+}	N M	0.0184	-	5
MOF-5			Pt-N	(77K)		
	ноос- Соон					
Cu ₄ I ₄ and	N N	Cu^+	Cu ₄ I ₄	0.0152	-	6
$[Cu_3Pz_3]_2$				(50K)		
(Pz=pyrazolat	N=N					
e)						
[Ag ₃ (dmtrz) ₂ (H ₃ C CH ₃	Ag^+	-	25.6	-	7
CN)] _n	N//N			(RT)		
Zn(II)-based	Q N	Zn^{2+}	0	0.15	-	8
MOFs	H		NO	(RT)		
[Cd ₂ (ptz)(squa	N	Cd^{2+}	O I	0.13	-	9
rate)(OH)(H ₂			00	(RT)		
O) ₂] _n			U O			
ZIF-8		Zn^{2+}	-	0.11	White	10
	NN			(RT)	(4.73%)	
$[(Ag_4I_4)(bix)]_n$		Ag⁺	-	1.26	-	11
(b1x=1,4-	7			(10K)		
bis(imidazole-						
1-						
ylmethyl)benz						
ene)						
[AgL] _n ∙nH2O	<u> </u>	Ag^+	-	2.60	White	12
(L=4-	Ğz			(RT)	(10.86%	
cyanobenzoat	<u> </u>)	

e)	N.					
$[Ag(tz)]_{\infty}(tz=$	$\langle \rangle$	Ag^+	-	4.59	-	13
triazole)	N-N			(RT)		
[CdLi(IPA) ₂](CdLi	-	32	-	14
Me_2NH_2)				(RT)		
(IPA =	·		Mn	1.6-10.5	-	15
isophthalic				(RT)		
acid, Me ₂ NH ₂						
=						
dimethylamin						
e)						
[Cd(µ-	0,0	Cd^{2+}	-	250 and	-	16
mimc) ₂ (H ₂ O)]	H ₃ C-N			430		
n	∖′			(10K)		
${[Cd_3(\mu_5-$	0,0	Cd^{2+}		170 and	-	17
btc) ₂ (μ -	0			_ 760		
pbptz)]·2DMF	0 0			- (10K)		
} _n						
[Zn(u-6ani)],	0	Zn ²⁺		430-	-	2
	O			1110(10		2
	₩N			K)		
	-	Cd ²⁺		340-830	-	17
				(10K)		17
				290(RT)		17
				2,0(111)		17
		Pb ²⁺	-	4.17	-	18
[Pb ₂ (EBTC)((10K)		
DMSO) ₃]						
	0 0					
	EBTC					
Cd(II)-based	0,00	Zn^{2+}	-	202(RT)	-	19
		Cd^{2+}	-	75(RT)		19
	-					
[Zn(TPA)(D	нооссоон	Zn^{2+}	pyridine	472(RT)	-	20
MF)]						
	TPA					

Zn(II)/Cd(II)-		Zn^{2+}	-	1796	-	19
based	0 0 0			(77K)		
	Ť			1321		19
				(293K)		
				94		19
				(413K)		
		Cd^{2+}	-	324	-	19
				(293K)		
Zn/Cd-		Zn^{2+}	-	472(293	-	20
terephthalate	$\rightarrow \rightarrow \rightarrow$			K)		
(TPA)	ő 🛁 ö			106	-	20
				(373K)		
				30		20
				(293K)		
				475	-	19
				(293K)		
		Cd ²⁺	-	158(293	-	19
				K)		
Cd-Eu\Tb\Gd-	O O	Cd^{2+}	-	489	-	21
CPs	0			(77K)		
			En Cd. O-	427	-	21
			Eu _x Cu _{I-x} Oγ	(293K)		
				10.54	-	21
				(413K)		
			$\mathrm{Tb}_x Cd_{1\text{-}x} O_7$	312	-	21
				(293K)		
				57.66		21
				(293K)		
Ln-CPs	HOOC	Cd^{2+}	Eu ³⁺	10.54	-	21
				(RT)		
			Tb ³⁺	57.66	-	21
				(RT)		
zinc iso-		Zn^{2+}	rhodamine B	926.56	-	22
phthalic acid	ноос Соон			(RT)		
(IPA)				97.55		22
based MOF				(RT)		
(denoted as						

NH+/Na+/K+-	ноос-Соон	$\mathrm{NH_4^+}$	-	586(RT)	-	23
TPA		Na ⁺		504(RT)		23
	_	K+-		585(RT)	-	23
[Pb ₂ (EBTC)(DMSO) ₃]	ноос соон соон	Pb ²⁺	-	4.17 (RT)	-	18
Cd-TCPA	соон Соон Ноос	Cd ²⁺	[(CH ₃) ₂ NH 2] ⁺ cations, which are generated from in situ decomposit ion of	472 (RT)	White	24
{[Cd2(tipa)2C]		Zn ²⁺	DIVIF -	-(77K)		25
$\frac{1.6 \text{ DMF}}{1.6 \text{ DMF}}$		Lii		(7714)		23
4] 0 2111) [Cd^{2+}	_			25
C@Zn(ZIF-8)	N~N	Zn^+		7400	-	26
	NN			(RT)		
				22400		26
				(RT)		
C@Zn(ZIF-8)	N	Zn ²⁺	Gd[(Pyr)4c yclen] (Pyr pyrenol)	(77K)	-	26
Cd(m- BDC)(BIM)	0	Cd^{2+}		755 (293K)	-	27
				554		27
Cd(m-BDC)		Cd ²⁺	H ₂ O/benzi	698(RT)	_	27
(H ₂ O)	ноос соон		midazole(404(RT)		27
	BDC		BIM)			



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