

Supporting information

MoS₂ stacking matters: 3R polytype significantly outperforms 2H MoS₂ for hydrogen evolution reaction

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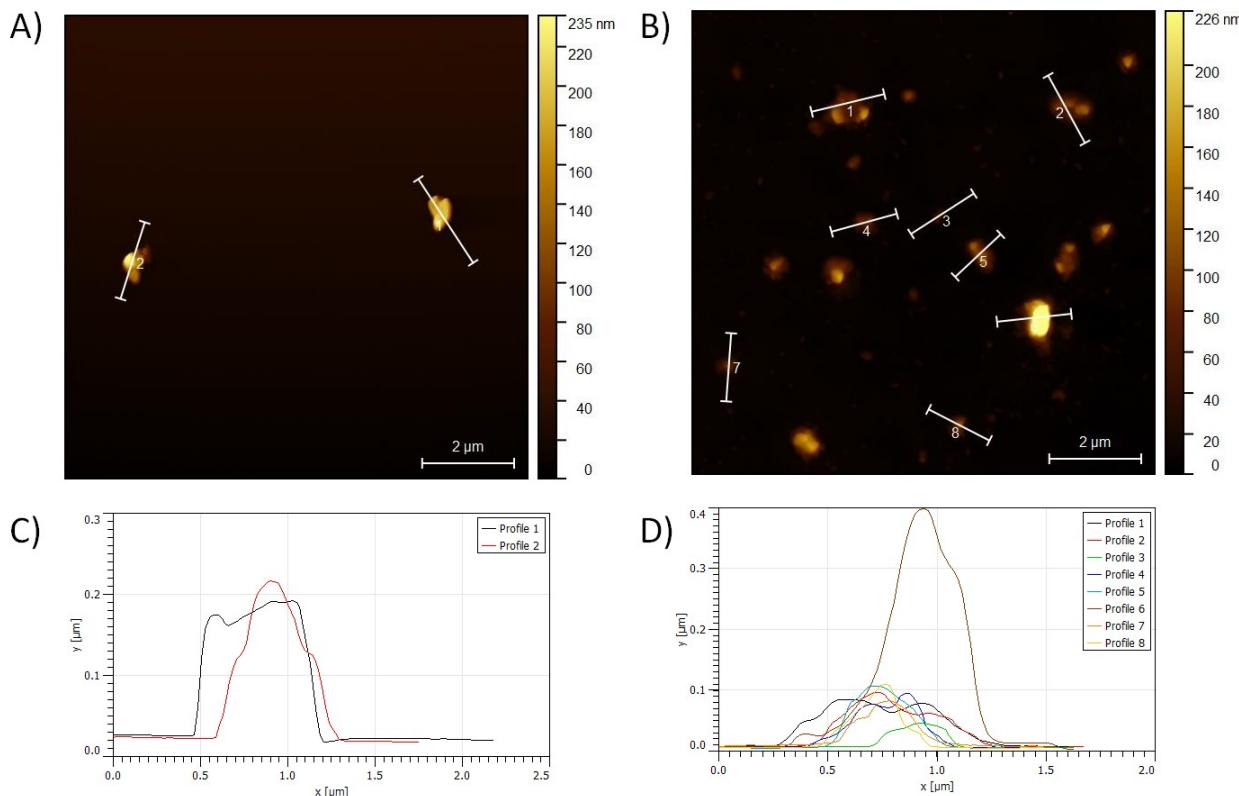


Figure S1. Atomic force microscopy image of A) MoS₂ 2H exf. together with C) extracted height profiles. Atomic force microscopy image of B MoS₂ 3R exf. together with D) extracted height profiles.

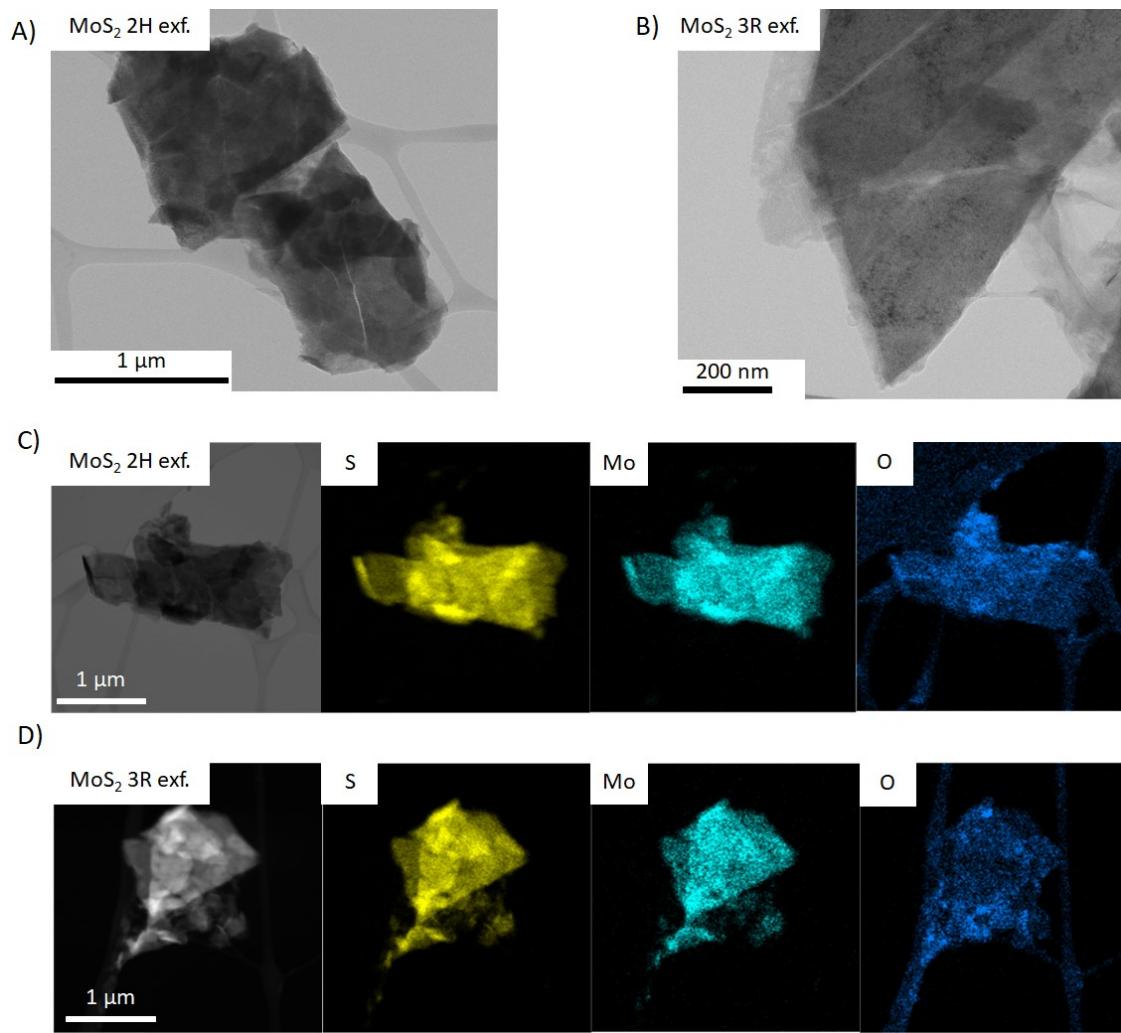


Figure S2. TEM images of A) MoS₂ 2H exf.; B) MoS₂ 3R exf.. Elemental composition maps for C) MoS₂ 2H exf.; D) MoS₂ 3R exf..

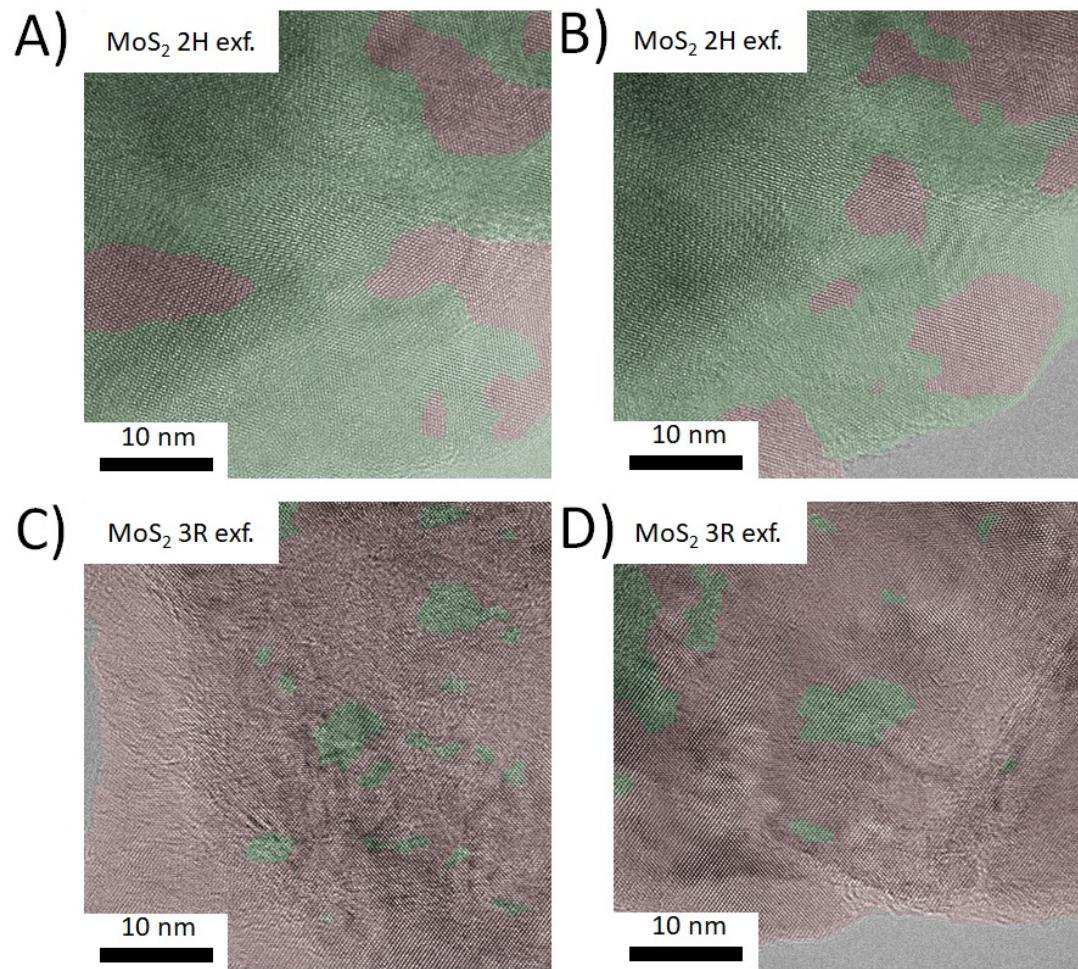


Figure S3. HR-TEM images showing the representation of trigonal prismatic (green) and octahedral (red) coordination of central atom in A) and B) MoS₂ 2H exf. And C) and D) MoS₂ 3R exf..

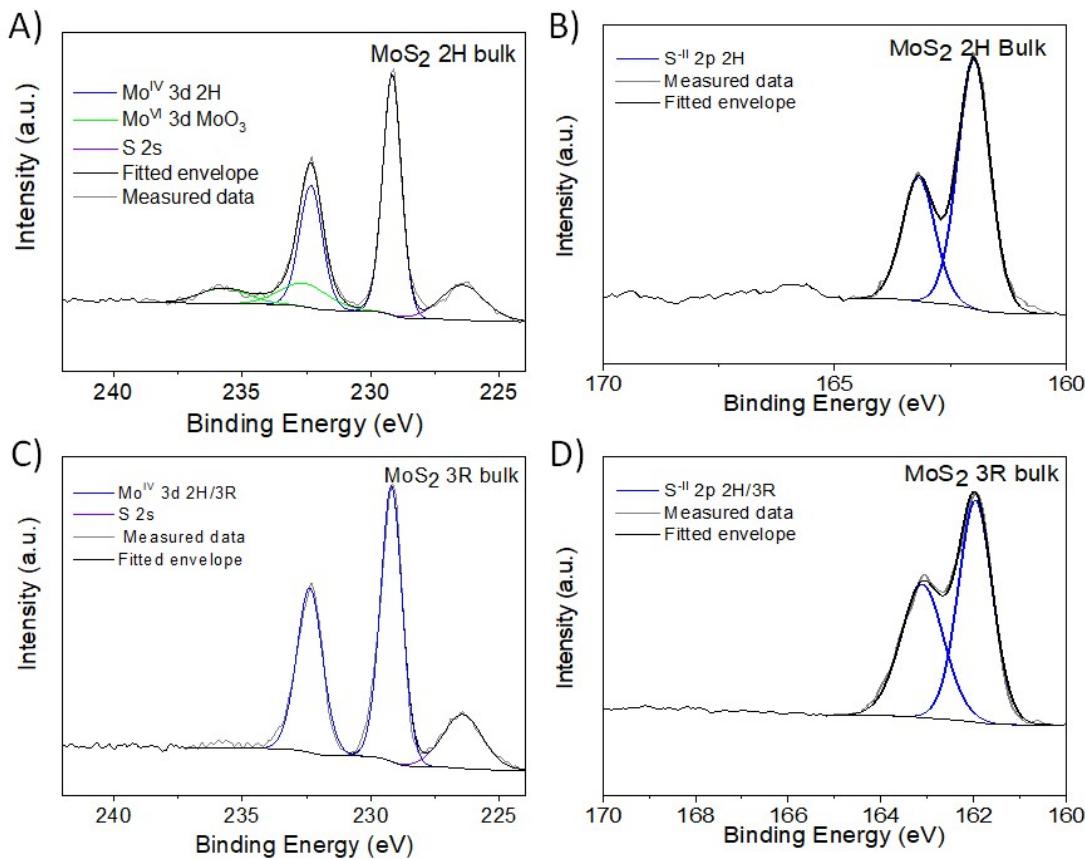


Figure S4. Core-level spectra of Mo 3d regions for A) MoS₂ 2H Bulk and C) MoS₂ 3R Bulk and core-level spectra of S 2p regions for B) MoS₂ 2H Bulk and D) MoS₂ 3R Bulk.

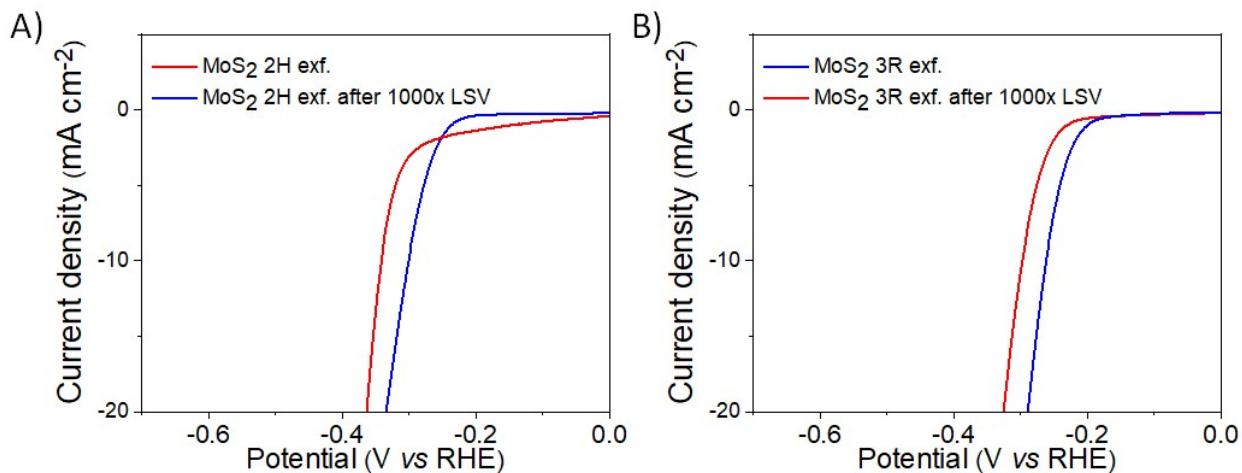


Figure S5. HER stability test after 1000 LSV cycles for A) MoS₂ 2H exf. and B) MoS₂ 3R exf.

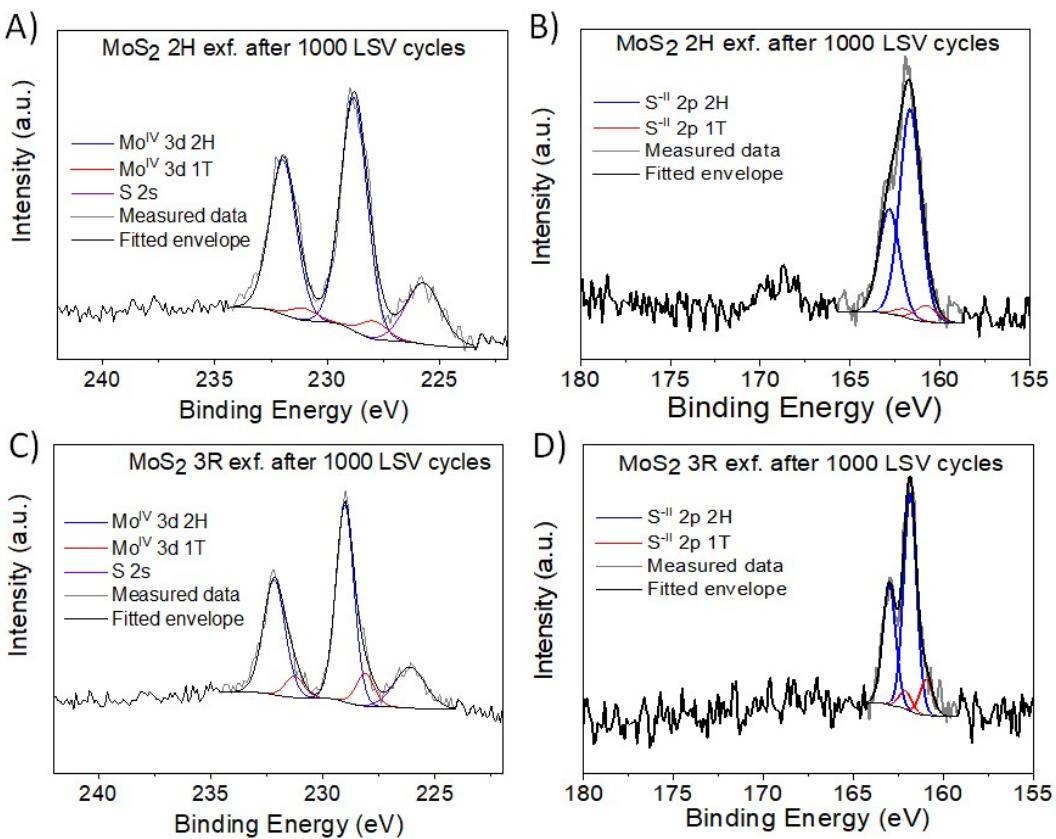


Figure S6. Core-level XP spectra after stability test of Mo 3d region for A) MoS₂ 2H exf. and C) MoS₂ 3R exf. and core-level XP spectra of S 2p region for B) MoS₂ 2H exf. and D) MoS₂ 3R exf.

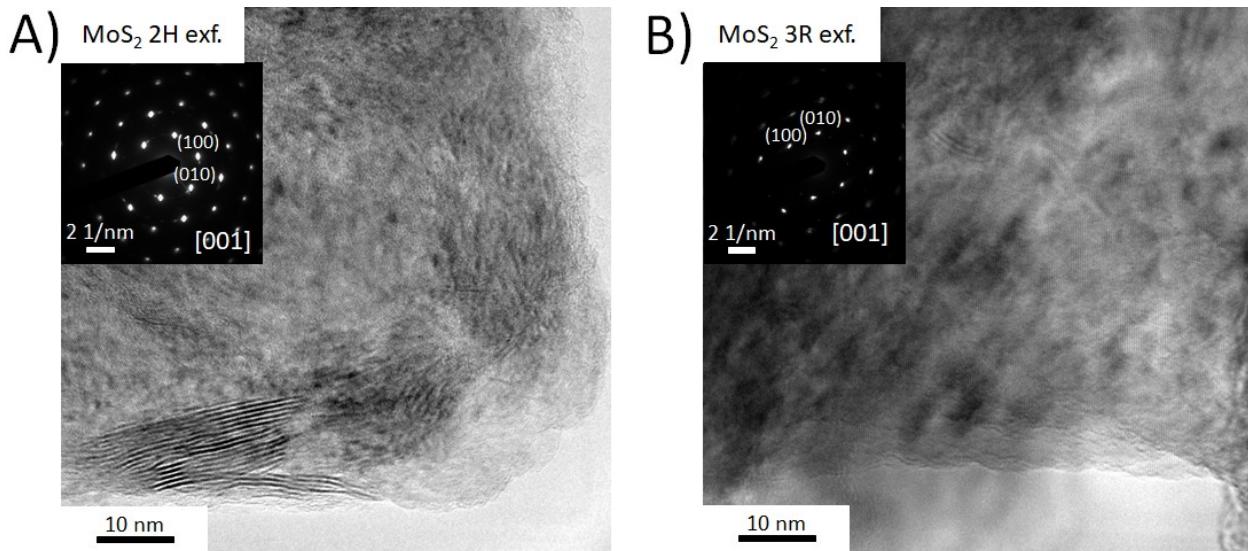


Figure S7. HR-TEM images after 1000 LSV cycles for A) MoS₂ 2H exf. and B) MoS₂ 3R exf.. The insets show selected area electron diffraction patterns.

Table S1. Concentration of individual chemical states in bulk and exfoliated MoS₂ materials obtained by XPS.

Sample	2H/3R (at. %)	1T (at. %)	MoO ₂ (at. %)	MoO ₃ (at. %)
MoS ₂ 2H bulk	85.3	-	-	14.7
MoS ₂ 3R bulk			-	-
MoS ₂ 2H exf.	64.1	20.8	-	15.1
MoS ₂ 3R exf.	27.0	55.9	17.1	-
MoS ₂ 2H exf. after 1000 LSV cycles	92.8	7.2	-	-
MoS ₂ 3R exf. after 1000 LSV cycles	86.2	13.8	-	-

Table S2. Comparison of various MoS₂ catalysts for HER based on the polytype.

Catalyst	Electrolyte	Overpotential (at - 10 mA cm ⁻²)	Tafel slope (mV dec ⁻¹)	Ref.
Exfoliated 3R MoS ₂	0.5M H ₂ SO ₄	250 mV	58	This paper
Strained 2H MoS ₂	0.5M H ₂ SO ₄	170 mV	60	¹
Basal plane activated MoS ₂	0.5M H ₂ SO ₄	300 mV	49	²
Mesoporous 2H MoS ₂	0.5M H ₂ SO ₄	218 mV	62	³
Mesoporous 1T MoS ₂	0.5M H ₂ SO ₄	153 mV	43	³
Modified 1T MoS ₂	0.5M H ₂ SO ₄	271 mV	61	⁴
Co-doped 1T MoS ₂	1M KOH	240 mV	68	⁵
Ni and O ₂ doped 1T MoS ₂	1M KOH	46 mV	52	⁶
Graphene/1T MoS ₂ composite	0.5M H ₂ SO ₄	103 mV	46	⁷
Sulfur intercalated 1T' MoS ₂	0.5M H ₂ SO ₄	205 mV	50	⁸
1T' monolayers	0.5M H ₂ SO ₄	191 mV	142	⁹

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