

**Supporting Information**

**Iron Sulphide Rice Grain Nanostructures as Potential Electrocatalysts for  
Improved Oxygen Evolution Reaction**

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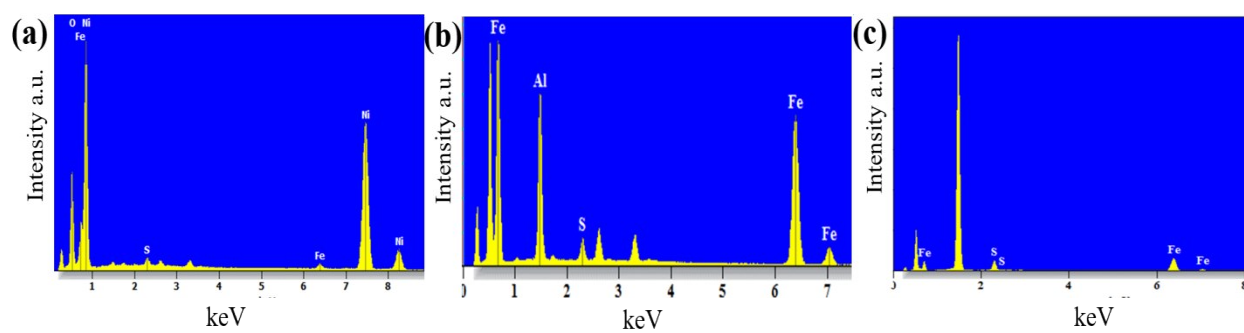
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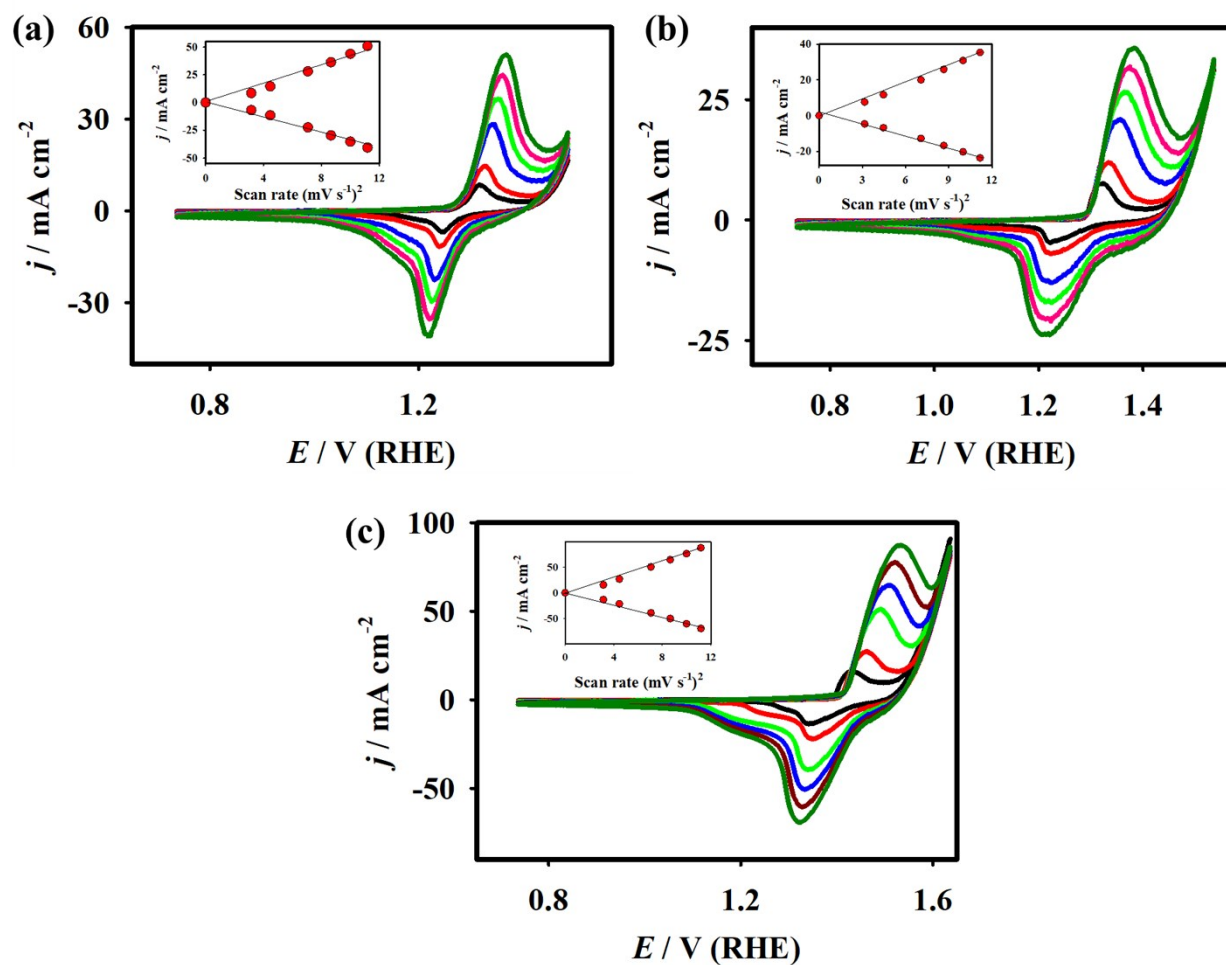
**Table S1.** Comparison of the as-developed FeS nanostructured electrodes in this study and other electrodes reported in the literature.

Materials	Overpotential (V) at 10 mA cm <sup>-2</sup>	Electrolyte	Tafel (mV dec <sup>-1</sup> )	Mass activity (A g <sup>-1</sup> )	Ref
CaFe <sub>0.5</sub> Co <sub>0.5</sub> O <sub>3</sub>	0.26	0.1 M KOH	59.0	0.55	1
Fe <sub>2</sub> /Co <sub>1</sub> -GNCL	0.35	1.0 M KOH	70.0	-	2
Ni-O-G	0.22	1.0 M KOH	42.0	0.46	3
Ni-Fe LDH	0.25 (20 mA cm <sup>-2</sup> )	1.0 M KOH	71.0	-	4
SiW <sub>9</sub> Co <sub>3</sub> [h]@ZIF-67	0.42	1.0 M KOH	93.9	-	5
Co <sub>2</sub> Mo <sub>3</sub> O <sub>8</sub> @NC-800	0.42	1.0 M KOH	87.5	4.1	6
Fe-Co-O	0.26	1.0 M KOH	53.0	-	7
Ag@Co(OH) <sub>x</sub>	0.25	1.0 M KOH	76.0	-	8
HXP@NC800	0.31	1.0 M KOH	48.0	-	9
FeS-NFS NF	0.22	1.0 M KOH	64.3	1.6	This study
FeS-NPS NF	0.30	1.0 M KOH	72.5	3.6	This study
FeS-RGS NF	0.20	1.0 M KOH	54.2	5.4	This study

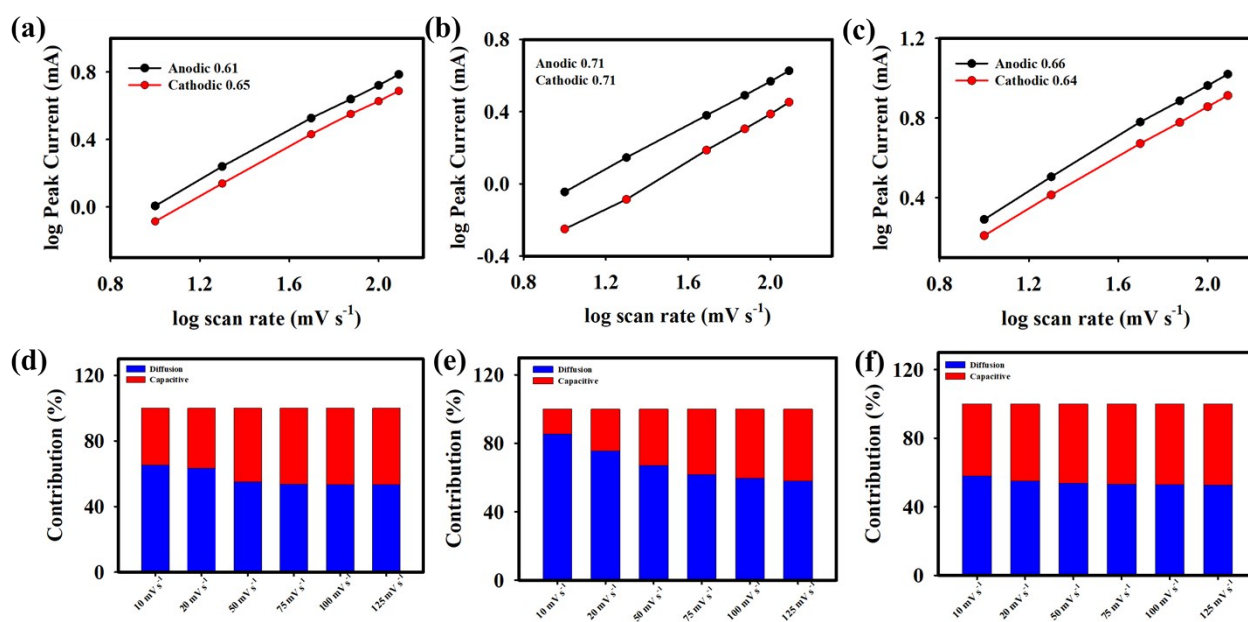
GNCL: graphitized nitrogen-doped carbon layer; Ni-O-G: individual nickel (Ni) bonded to oxygen sites on graphene-like carbon; LDH: layered-double-hydroxide; ZIF: zeolitic imidazole frameworks; NC: nitrogen-rich carbon; HXP: 2D nickel-based MOF hexagonal nanoplate; NC: nanocrystal; NFS: nanoflowers; NF: nickel foam; NPS: nanoparticles; RGS: rice grain structures;



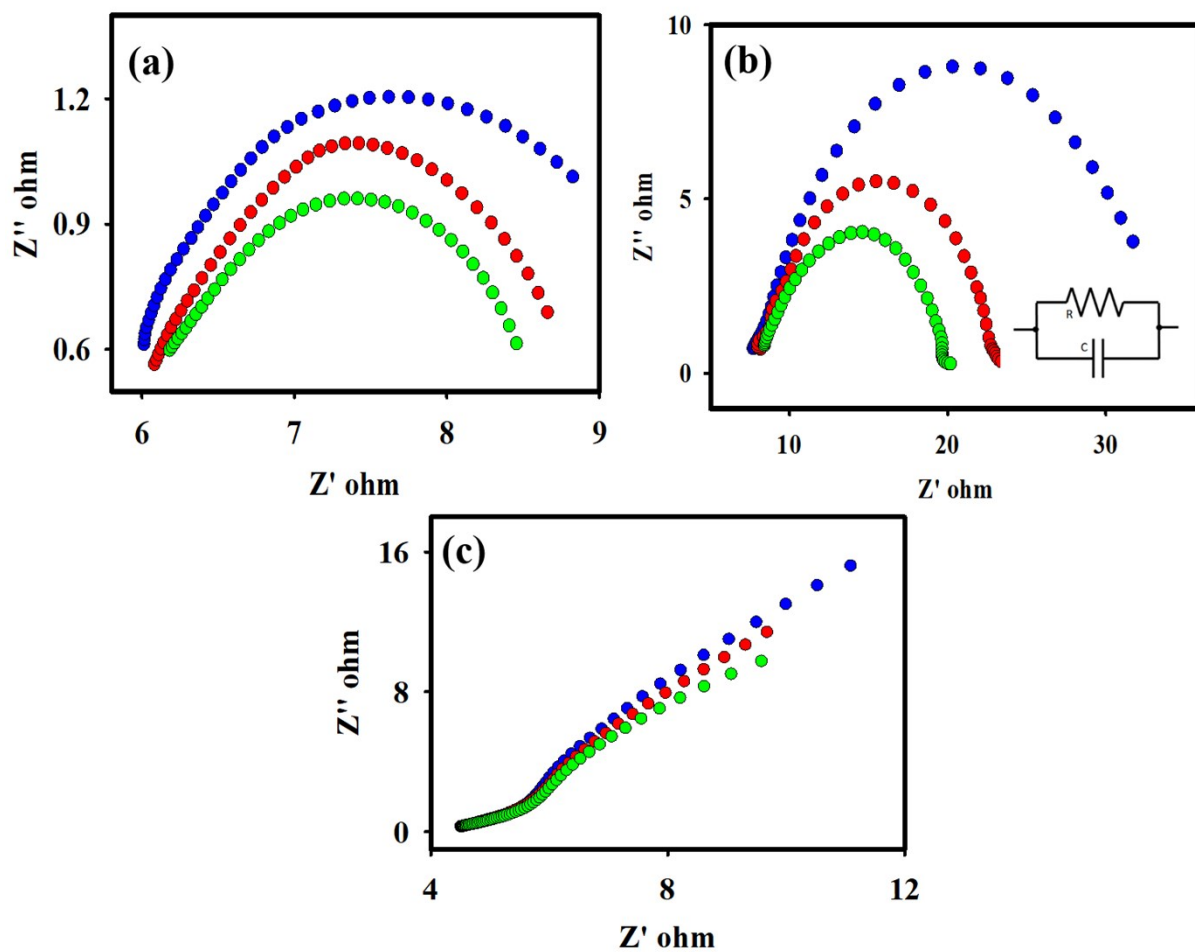
**Fig. S1.** EDX spectra of FeS-RGS (a), FeS-NFS (b) and FeS-NPS (c) nanomaterials.



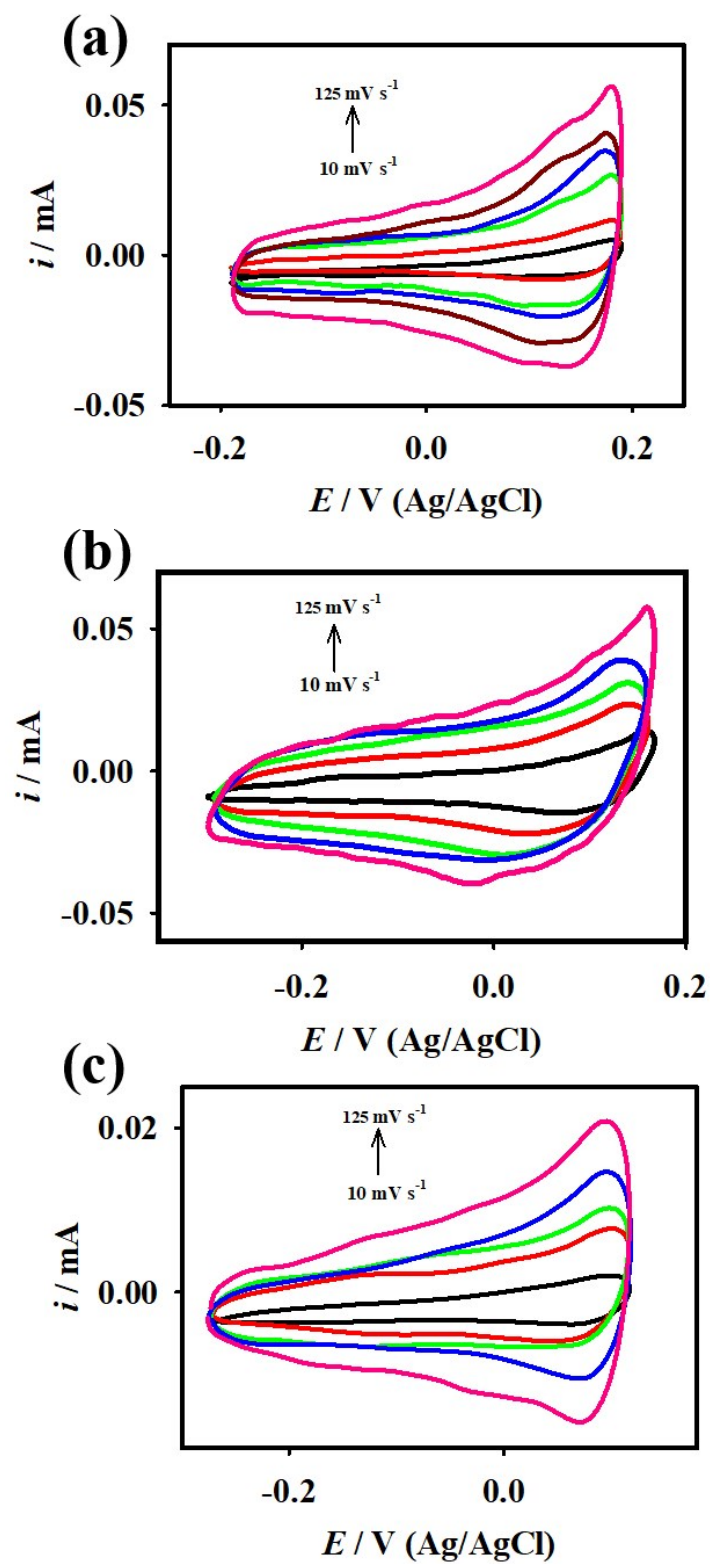
**Fig. S2.** CV curves of the FeS-RGS|NF (a), FeS-NFS|NF (b) and FeS-NPS|NF (c) electrodes recorded at different scan rates, starting from 10  $\text{mV s}^{-1}$  to 125  $\text{mV s}^{-1}$ . Inset showing the plot of peak currents vs square root of the scan rates.



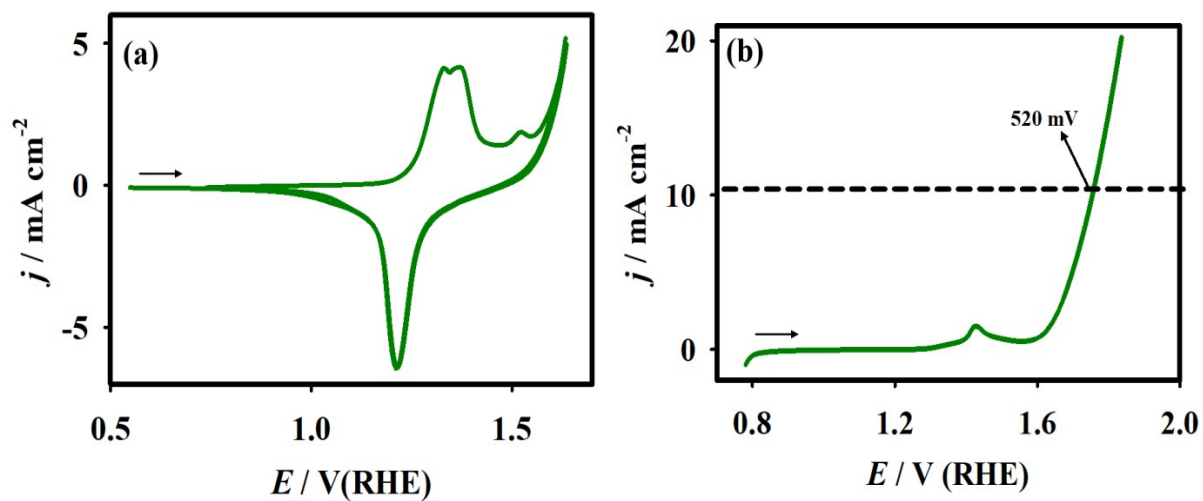
**Fig. S3.** Plot of logarithm peak current against logarithm scan rate measured for FeS-RGS|NF (a), FeS-NFS|NF (b) and FeS-NPS|NF (c) electrodes, derived from Fig. S2. Contribution ratio between diffusion controlled and capacitance at the FeS-RGS|NF (d), FeS-NFS|NF (e) and FeS-NPS|NF (f) electrodes.



**Fig. S4.** Nyquist plot of the FeS-RGS|NF (a), FeS-NFS|NF (b) and FeS-NPS|NF (c) electrodes at the applied potential of 1.49 V (blue), 1.51 V (red) and 1.53 V (green) in 1.0 M KOH.

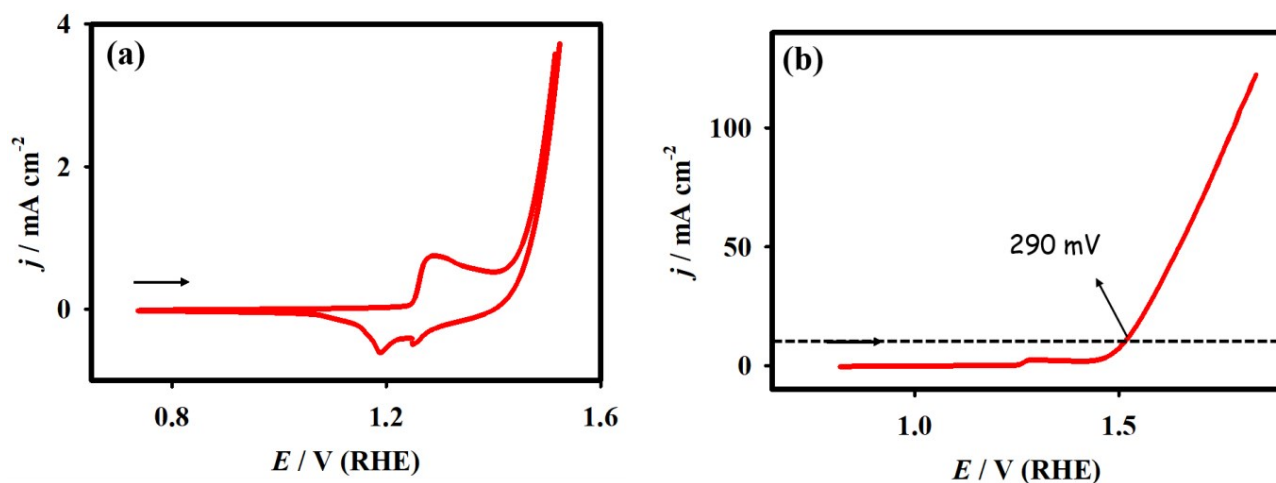


**Fig. S5.** CV curves of the FeS-RGS|NF (a), FeS-NFS|NF (b) and FeS-NPS|NF (c) electrodes recorded for the electrochemical double layer capacitance measurements at various scan rates.

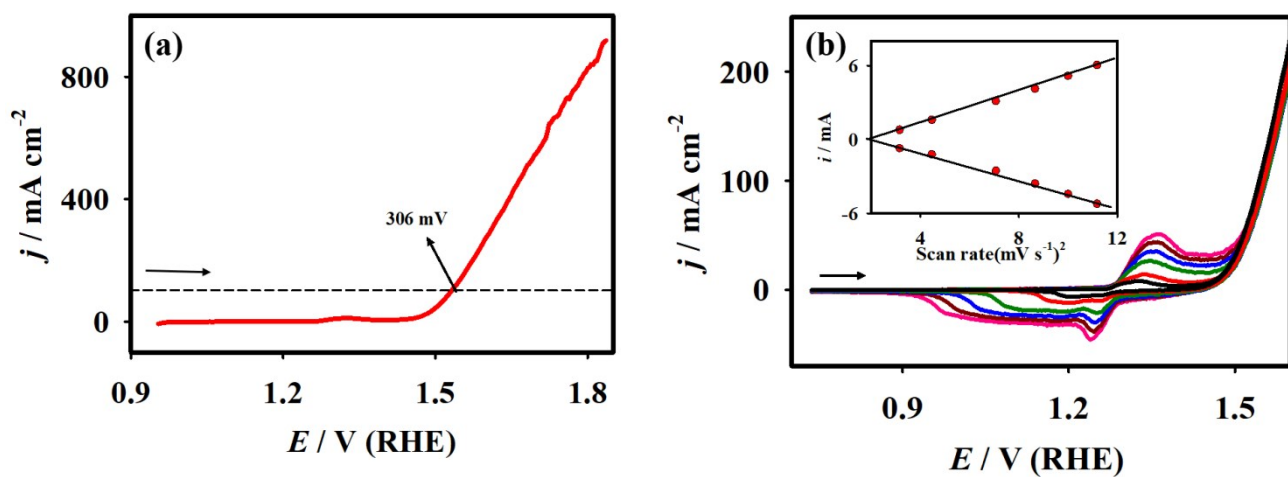


**Fig. S6.** CV curve (a) and LSV curve (b) of the CoS|NF electrodes recorded in 1.0 M KOH at a scan rate of 20 mV s<sup>-1</sup>.

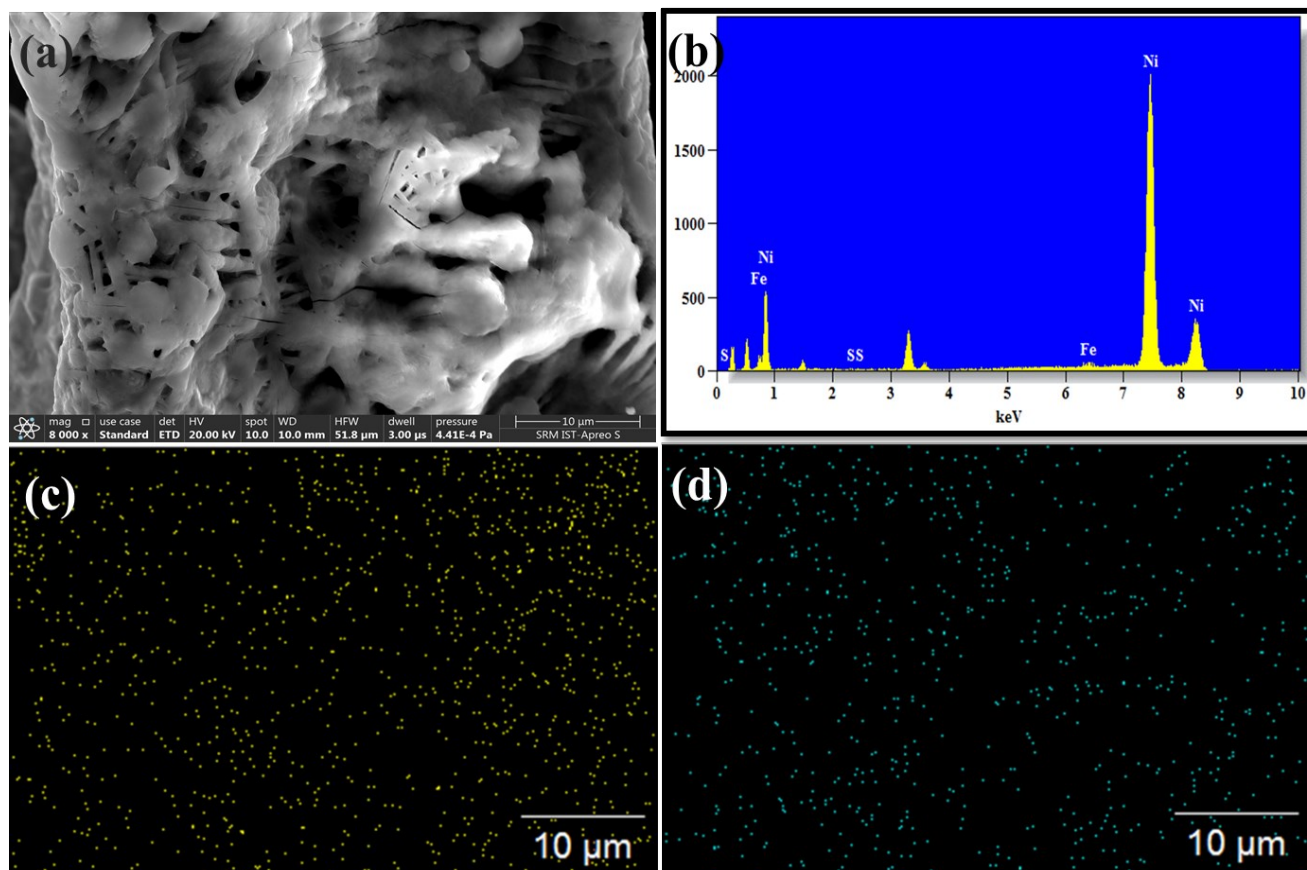




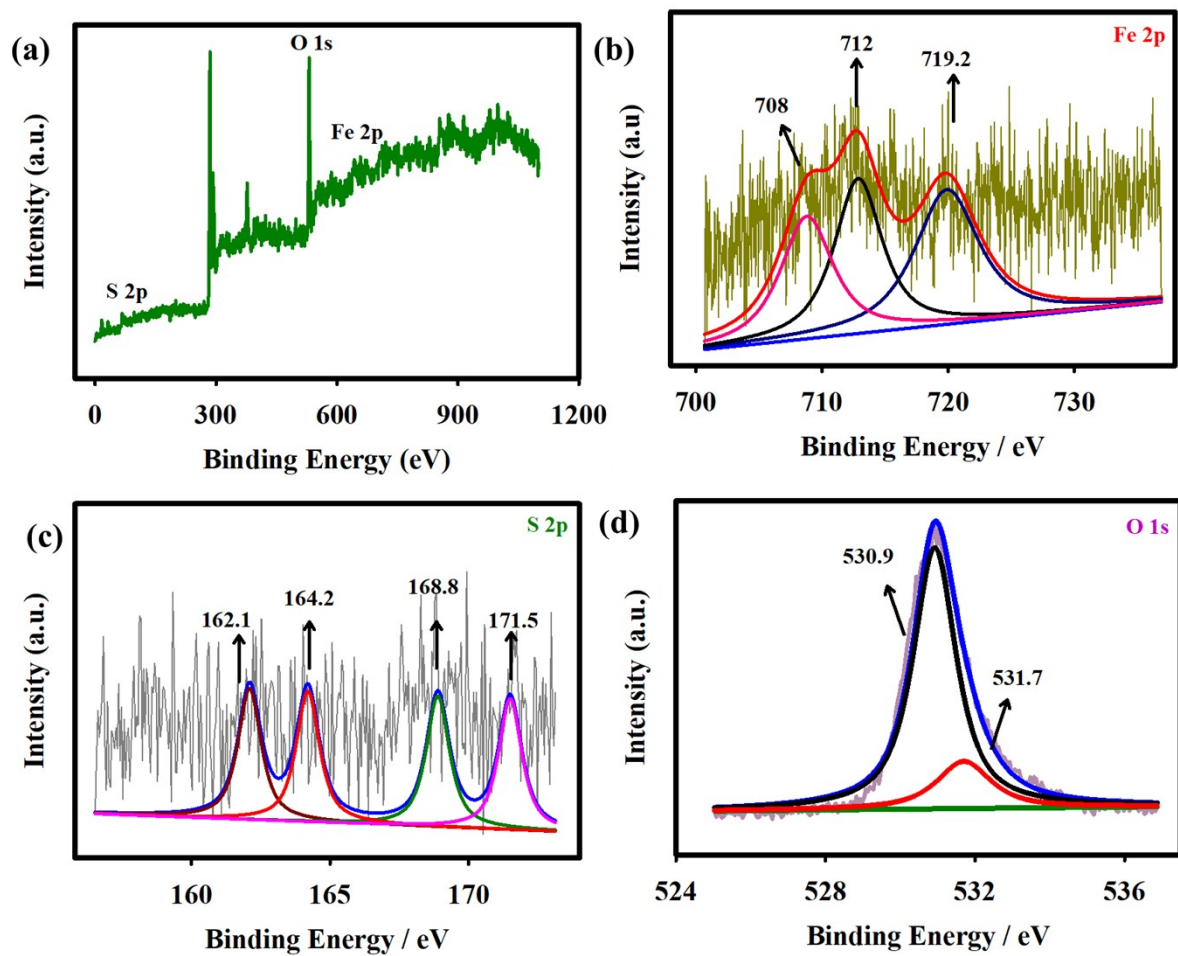
**Fig. S7.** CV curve (a) and LSV curve (b) of the FeCoS/NF electrodes recorded in 1.0 M KOH at a scan rate of 20 mV s<sup>-1</sup>.



**Fig. S8.** LSV curve (a) and CV curves (b) of the FeS-RGS|NF electrode recorded in 30% KOH. The inset shows the plot of peak current vs square root of scan rates.



**Fig. S9.** SEM image (a), EDX spectra (b) and elemental mapping of Fe (c) and S (d) for the FeS-RGS|NF electrode measured after had a long-term stability test.



**Fig. S10.** Full scan XPS analysis of the FeS-RGS|NF electrode after had a long-term durability test (a) and its high resolution XPS spectra in the regions of Fe 2p (b), S 2p (c), and O 1s (d).

## References

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