

# Interface Engineering Triggered by Carbon Nanotubes-Supported Multiple Sulfides for Boosting Oxygen Evolution

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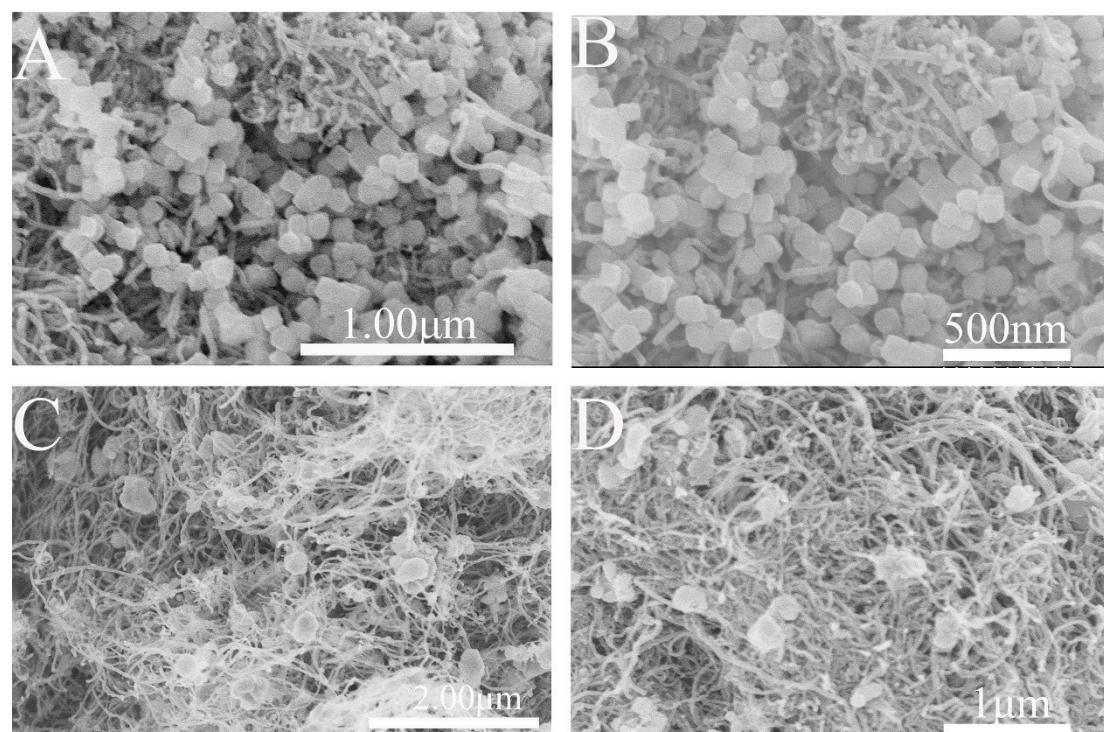
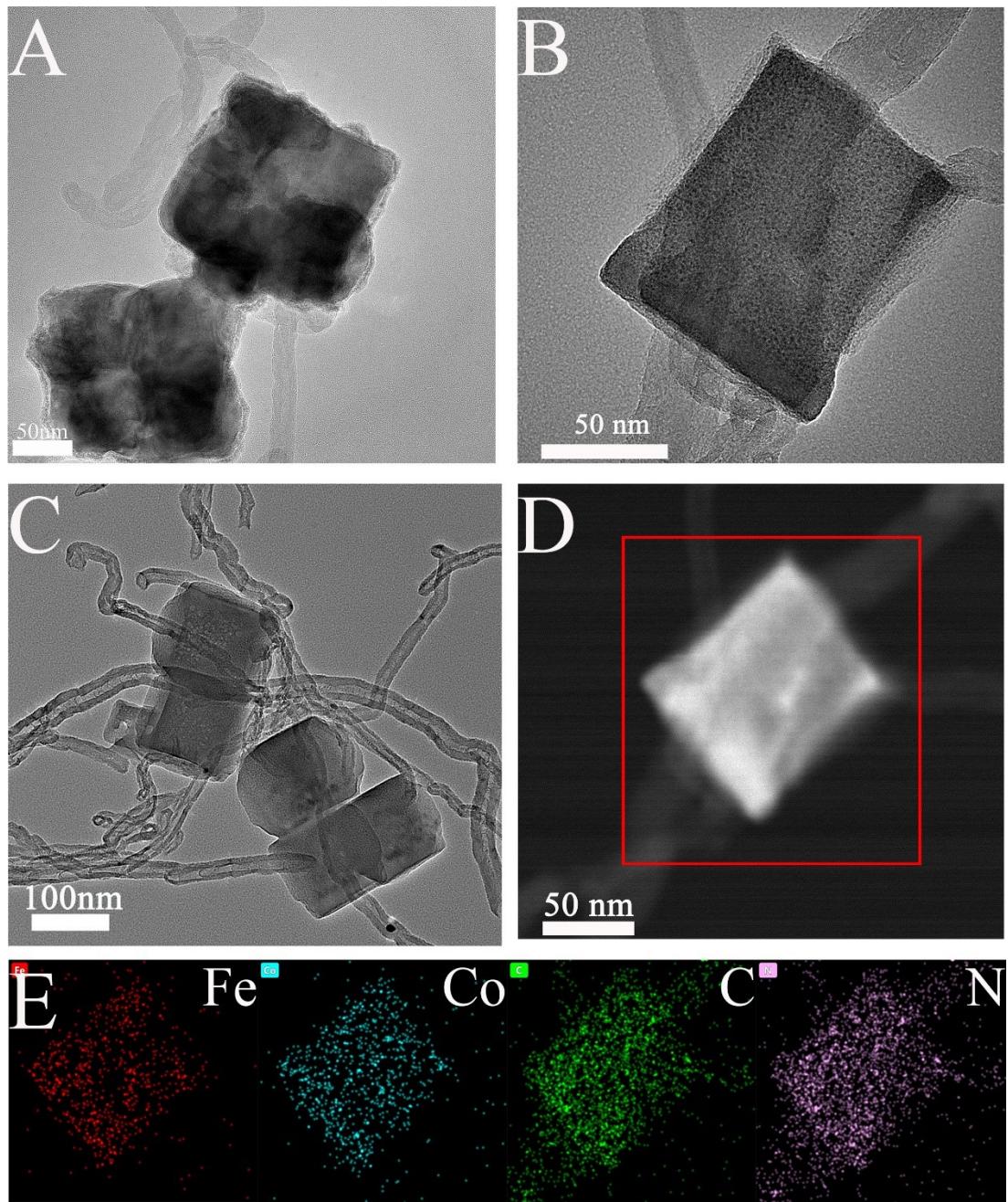
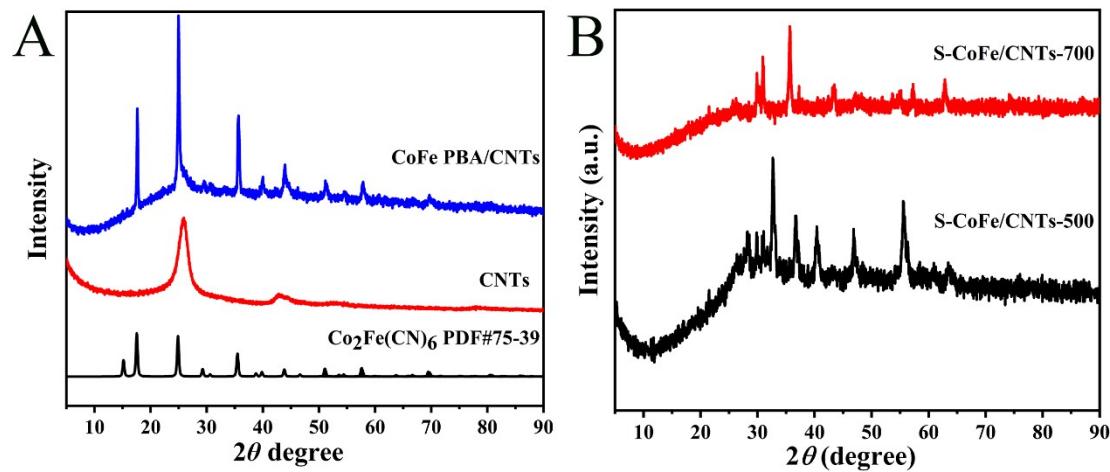


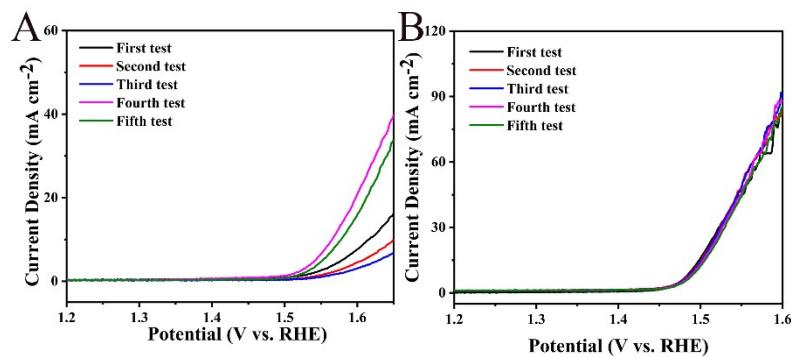
Figure S1 SEM image of CoFe PBA/CNTs (A, B) and S-CoFe/CNTs (C, D)



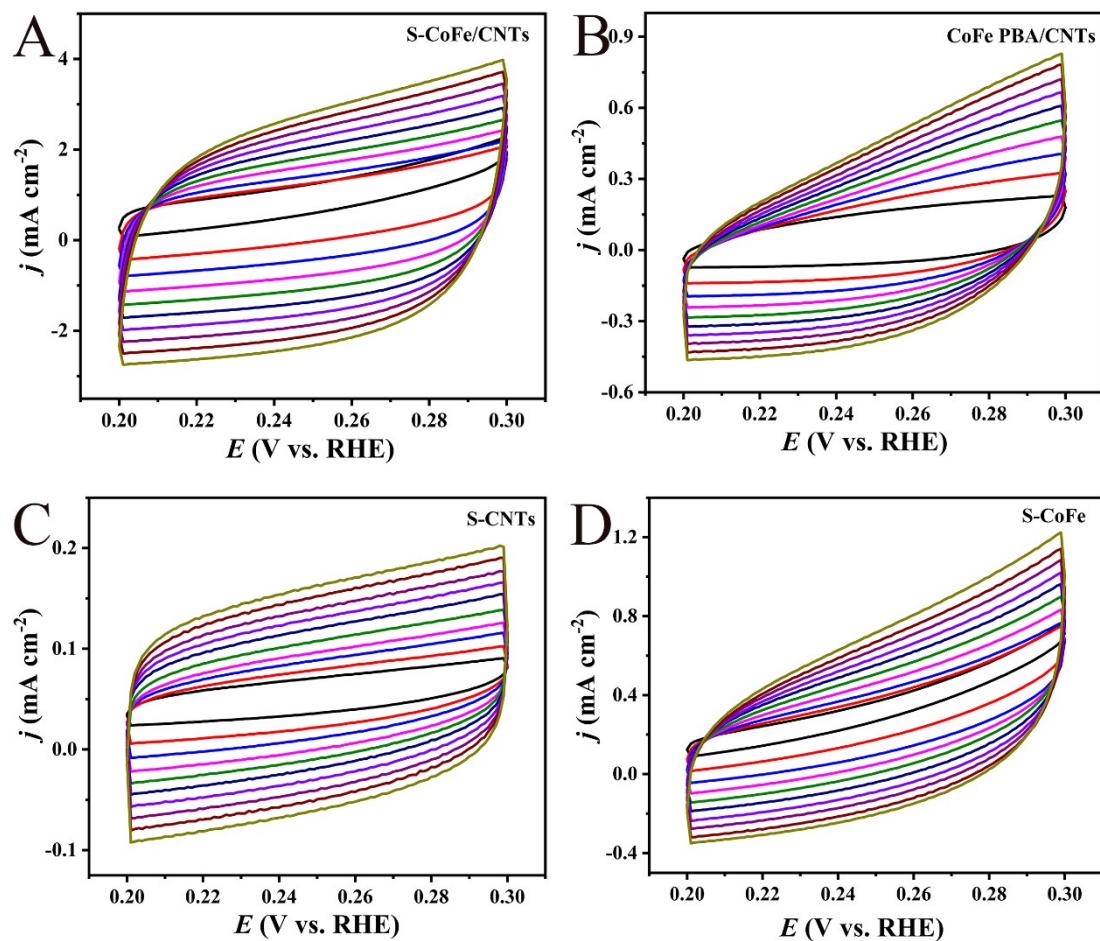
**Figure S2** TEM images (A) of multi-individual S-CoFe/CNTs , (B) one individual CoFe PBA/CNTs , (C) multi-individual CoFe PBA/CNTs (D) Dark field TEM of CoFe PBA/CNTs, (E) EDS elemental mapping of the CoFe PBA/CNTs refers to the signals of Co, Fe, C and N respectively as marked in (C).



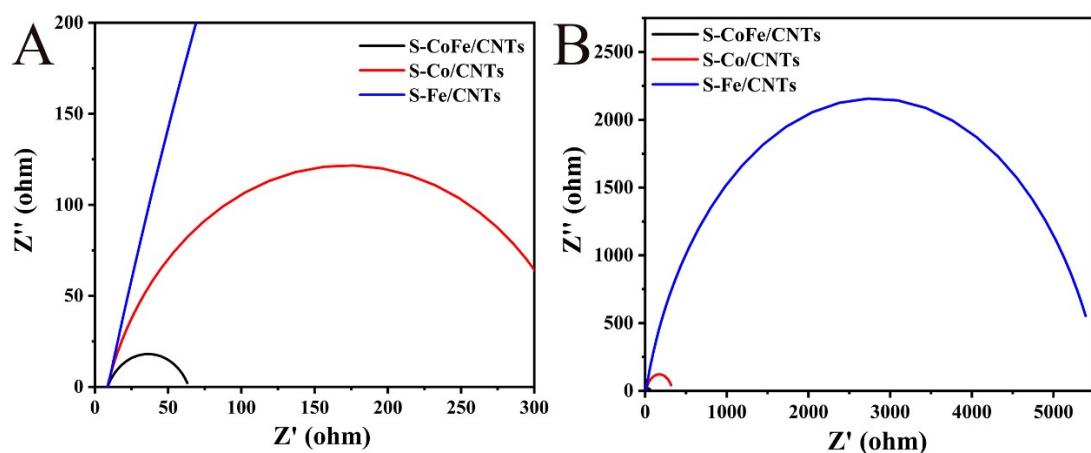
**Figure S3** (A) XRD patterns of CoFe PBA/CNTs samples; (B) XRD diffractograms of S-CoFe/CNTs-700 and S-CoFe/CNTs-500



**Figure S4** (A) The polarization curve of S-CoFe /CNTs prepared at 400°C for five tests ; (B) The polarization curve of S-CoFe /CNTs prepared at 500°C for five tests



**Figure S5** CV curves of the (A) S-CoFe/CNTs, (B) CoFe PBA/CNTs, (C) S-CNTs and (D) S-CoFe at different scan rates from 10 to 100mV s<sup>-1</sup>.



**Figure S6** (A) local enlarged images of image (B); (B) Nyquist diagrams of EIS for S-CoFe/CNTs, S-Co/CNTs and S-Fe/CNTs;

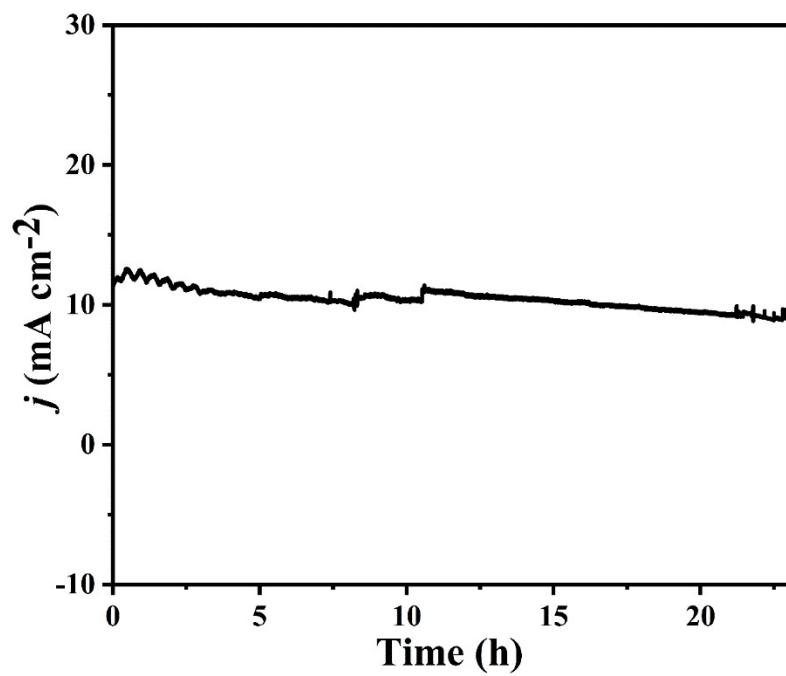


Figure S7 Amperometric i-t curve of S-CoFe/CNTs

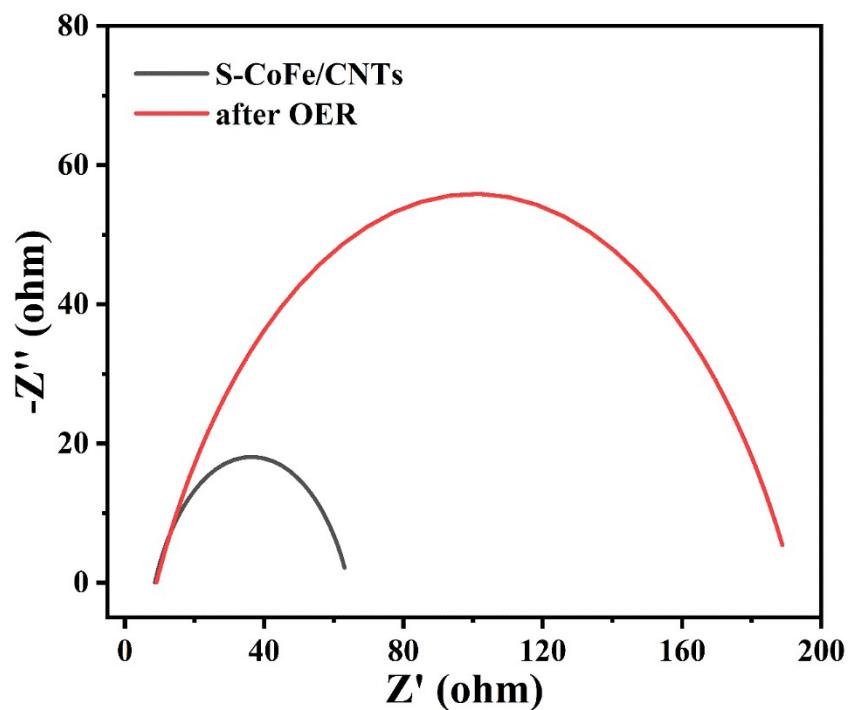
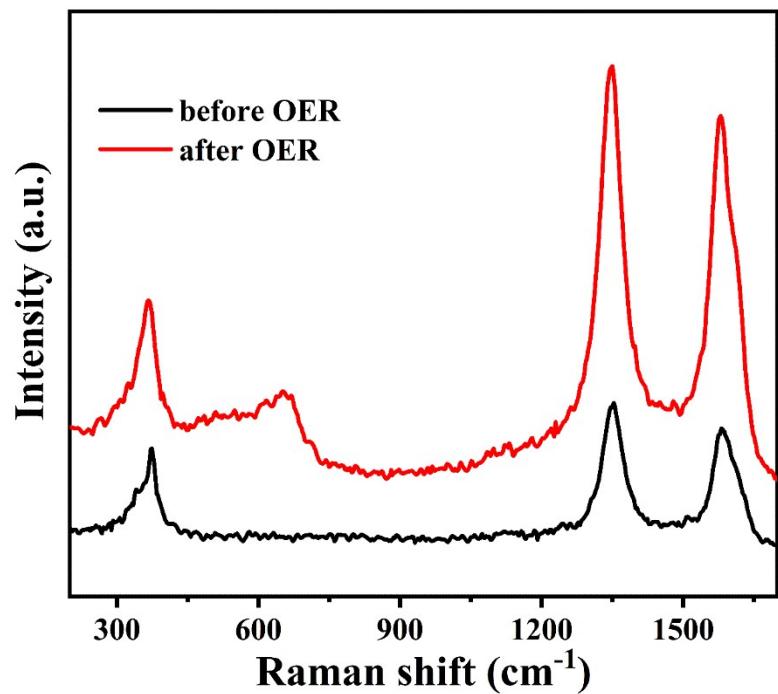
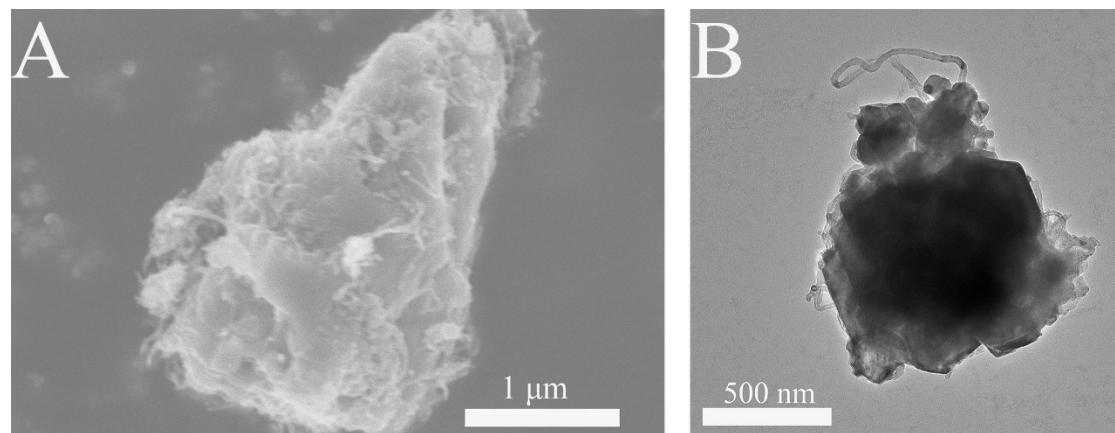


Figure S8 Nyquist diagrams of EIS for S-CoFe/CNTs before and after the OER test



**Figure S9** Raman spectra of S-CoFe/CNTs before after OER test



**Figure S10** (A) SEM image and (B) TEM image of S-CoFe/CNTs after OER test

**Table S1** The comparison of other OER catalysts derived from Prussian blue analogue in alkaline medium.

Materials	Overpotential (mV) @10mA cm <sup>-2</sup>	Reference
S-CoFe/CNTs	253	This work
CoFe@NC-NCNT-H	380	1
CoSe <sub>2</sub> NBs	335	2
PBA@POM	440	3
NiCoO <sub>x</sub> -400	280	4
Ni <sub>x</sub> Co <sub>3-x</sub> O <sub>4</sub> /NF	287	5
Ni <sub>2</sub> Fe-O	370	6
CoFe PBA/CoS <sub>2</sub> -12	301	7
FeCoP-400	261	8
Fe <sub>0.5</sub> Co <sub>0.5</sub> MoO <sub>4</sub> -xSx	263	9
CoP-NPC	308	10
PB@Met-700	330	11

**Table S2** The comparison of OER performance of other TMSs catalysts in alkaline medium.

Materials	Overpotential (mV) @10 mA cm <sup>-2</sup>	Reference
S-CoFe/CNTs	253	This work
Cu <sub>9</sub> S <sub>5</sub> /NF	298	12
Co <sub>0.25</sub> Fe <sub>0.75</sub> S <sub>2</sub>	270	13
Ni-Fe-OH/Ni <sub>3</sub> S <sub>2</sub> /NF	268	14
CNTs/N-Cu <sub>2</sub> S-	280	15
5cyc		
Ni <sub>3</sub> S <sub>4</sub> /N,P-HPC	370	16
CuS <sub>2</sub> nanorods	270	17
CoS <sub>2</sub> @MXene	270	18
Fe/C-doped-MoS <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub> -450	270	19

**Table S3** The Integral table of Co and Fe peak area in XPS data of catalyst before and after OER test

	Before OER test		After OER test	
	2p1/2	2p3/2	2p1/2	2p3/2
Co <sup>2+</sup> /Co <sup>3+</sup>	4.172	4.176	0.475	0.477
Fe <sup>2+</sup> /Fe <sup>3+</sup>	2.705	2.701	0.538	0.531

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