

Supporting Information

for

Oxidation and Degradation of WS₂ Monolayers Grown by NaCl-Assisted Chemical Vapor Deposition: Mechanism and Prevention

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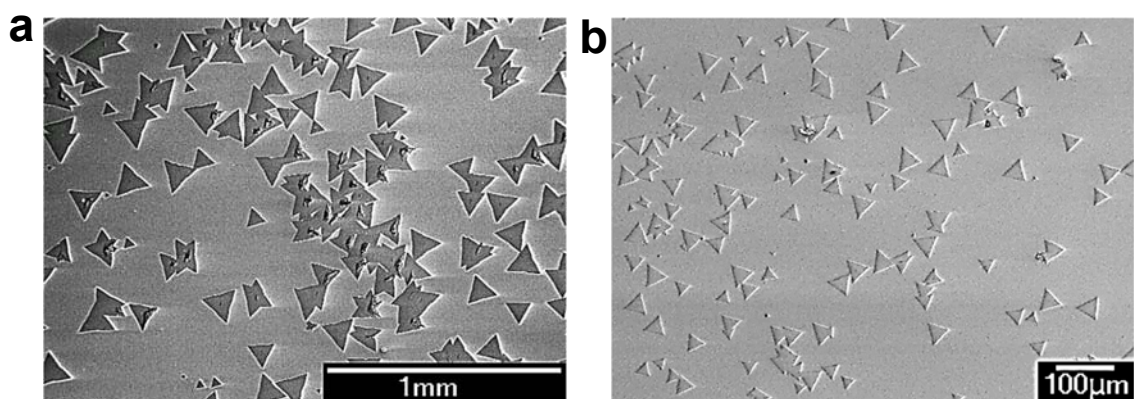


Figure S1. SEM images of WS₂ samples made (a) with NaCl assistant and (b) without NaCl assistant.

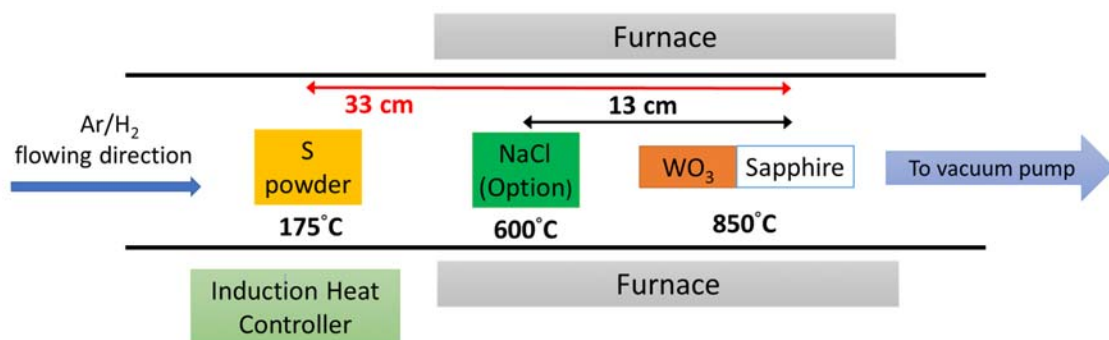


Figure S2. Schematics of the homemade chemical vapour deposition in this work. NaCl is an optional ingredient in this study (please find the details of the experiment).

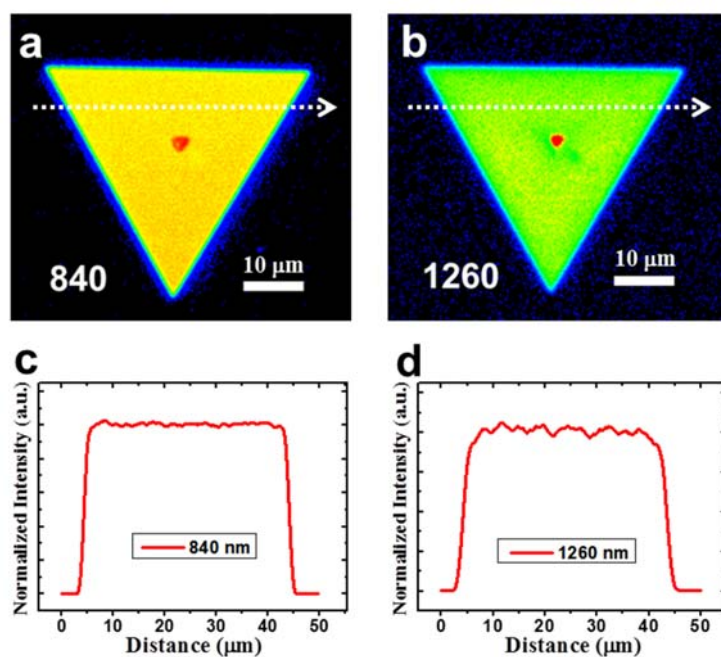


Figure S3. Intensity profile analysis of Figure 1d,f. Both intensity profiles are flat, indicating that there was no photooxidation caused by the SHG laser scanning. In (d), the slight fluctuation is due to the 1260 nm wavelength excitation, and the signal-to-noise ratio is worse than that of 840 nm.

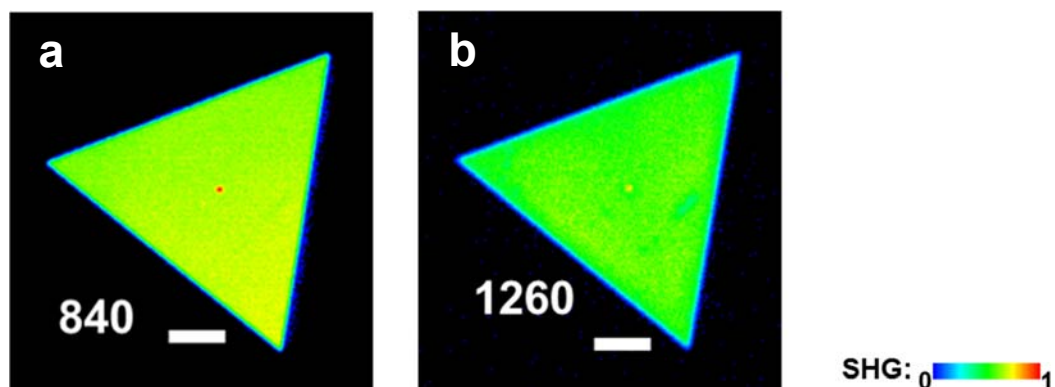


Figure S4. SHG images of a fresh NaCl-assisted WS₂ monolayer before the PL mapping excited at (a) 840 and (a) 1260 nm with fluence of $\sim 3.1 \times 10^6$ and 5.1×10^7 Jm⁻², respectively. They did not show any dark spots, indicating that there was no photooxidation caused by the SHG laser scanning. SHG intensity is normalized and shown with a rainbow color scale. Scale bar, 10 μ m.

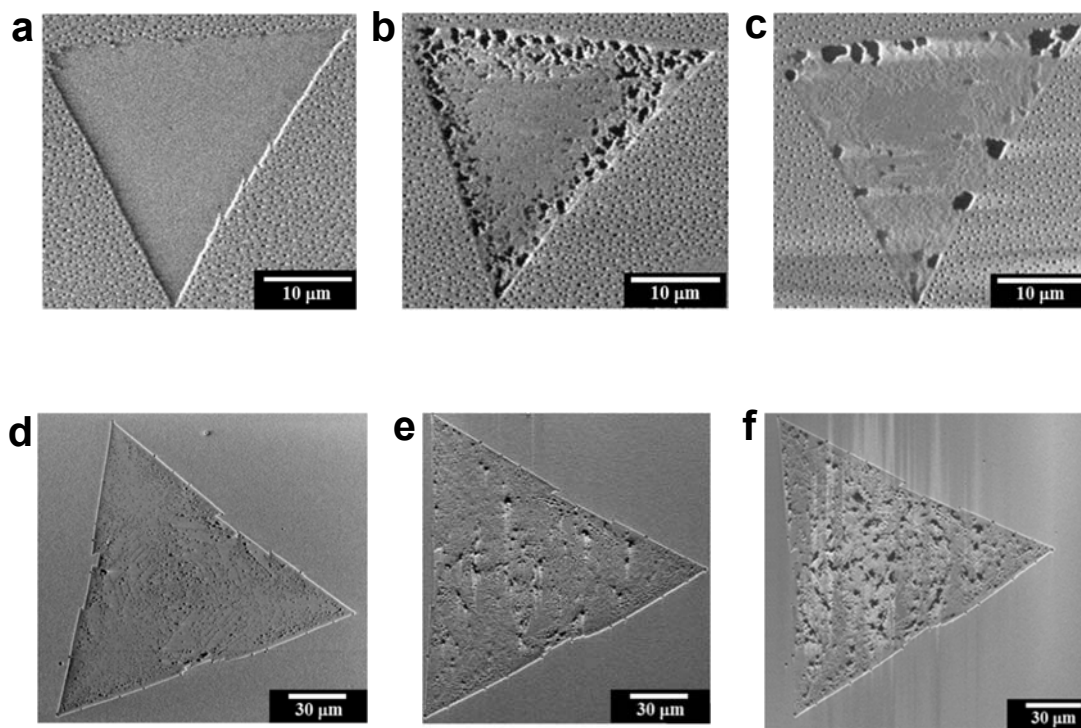


Figure S5. SEM images of a pure and a NaCl-assisted WS₂ monolayer stored under fluorescent light and 55% RH for (a,d) 1 day, (b,e) 3 days, and (c,f) 7 days, respectively.

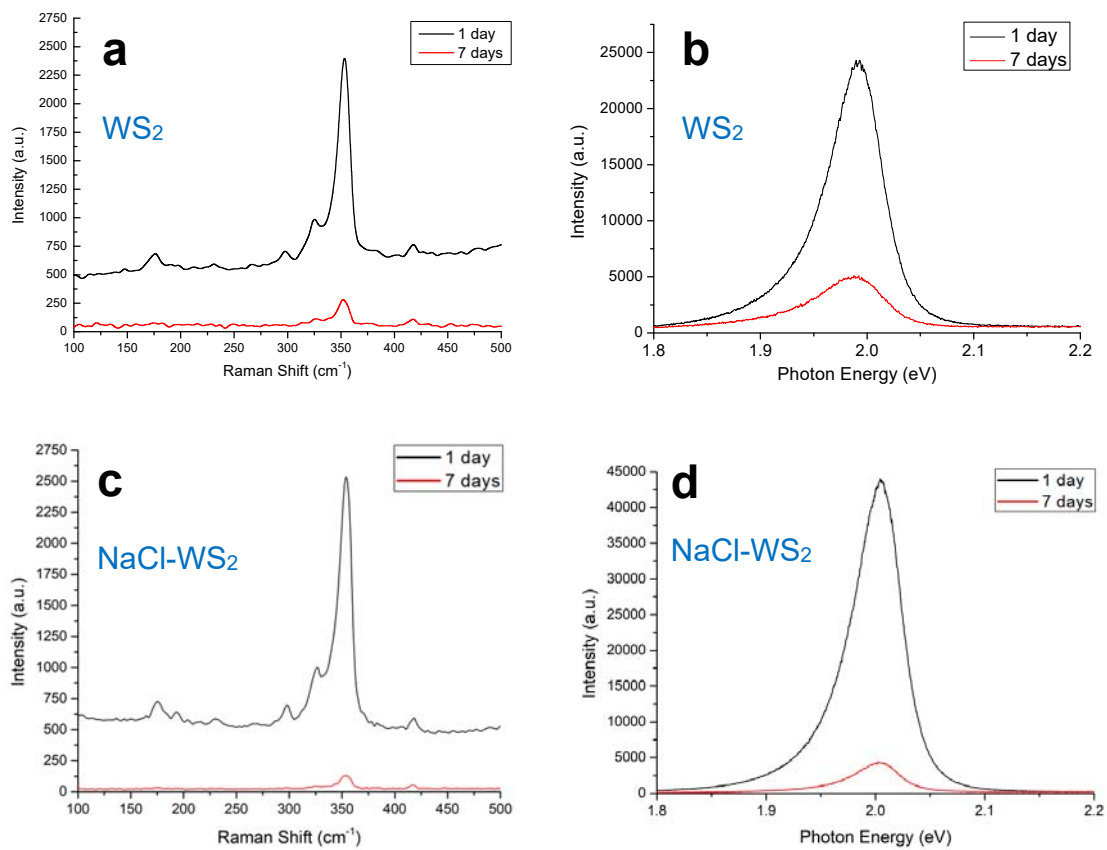


Figure S6. (a,c) Raman and (b,d) PL spectra of the WS₂ samples without and with NaCl assistant, respectively, stored under fluorescent light and 55% RH for 1 day and 7 days. The measurement point covers the black area in the SEM image (i.e., the transparent area in the optical microscopy (OM) images).

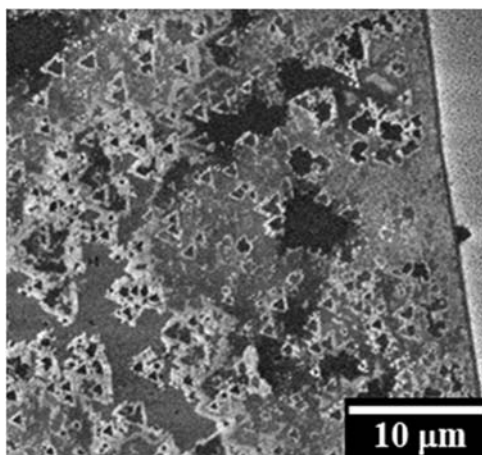


Figure S7. Magnified SEM image of the selected area of WS₂ sample with NaCl assistant stored under fluorescent light and 55% RH for 7 days. There are small triangles, which is similar to pure WS₂ caused by surface defects.^{1,2,3}

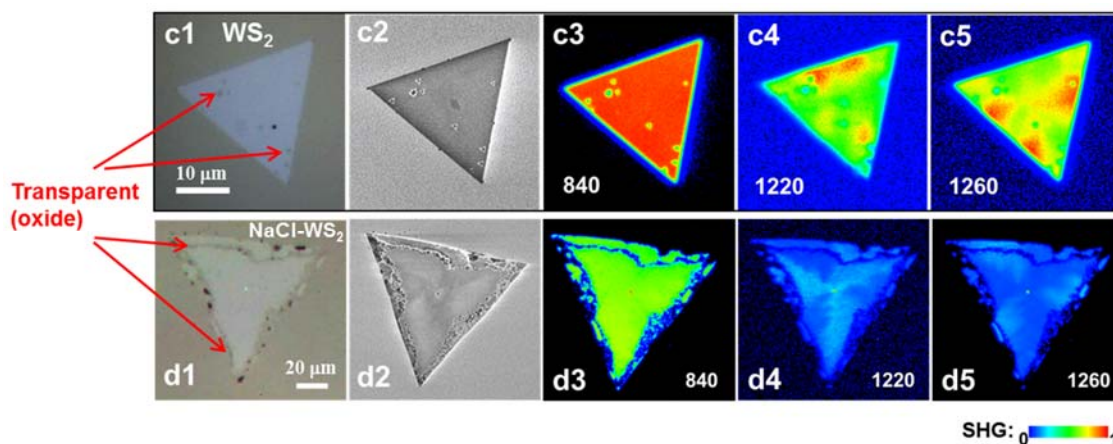


Figure S8. WS₂ and NaCl-assisted WS₂ monolayers stored under fluorescent light and 55% RH about 1 day were characterized by optical microscopy (c1,d1), SEM (c2,d2), and SHG (c3-c5,d3-d5), respectively. SHG images were excited at 840, 1220, and 1260 nm with fluence of $\sim 3.1 \times 10^6$, 4.6×10^7 , and 5.1×10^7 Jm⁻², respectively. SHG intensity is normalized and shown with a rainbow color scale.

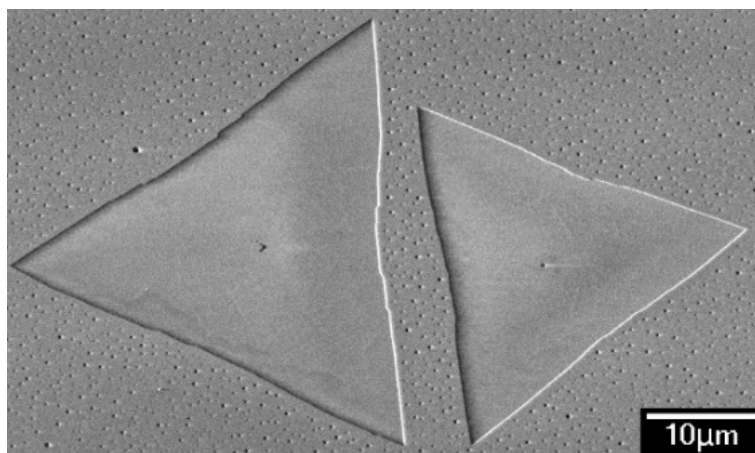


Figure S9. SEM image of the NaCl-assisted WS₂ flakes stored in a Schlenk tube pumped to vacuum (ca. 1.7×10^{-3} Torr) and exposed to fluorescent light for 14 days. There is no visible change in the NaCl-assisted WS₂ monolayers.

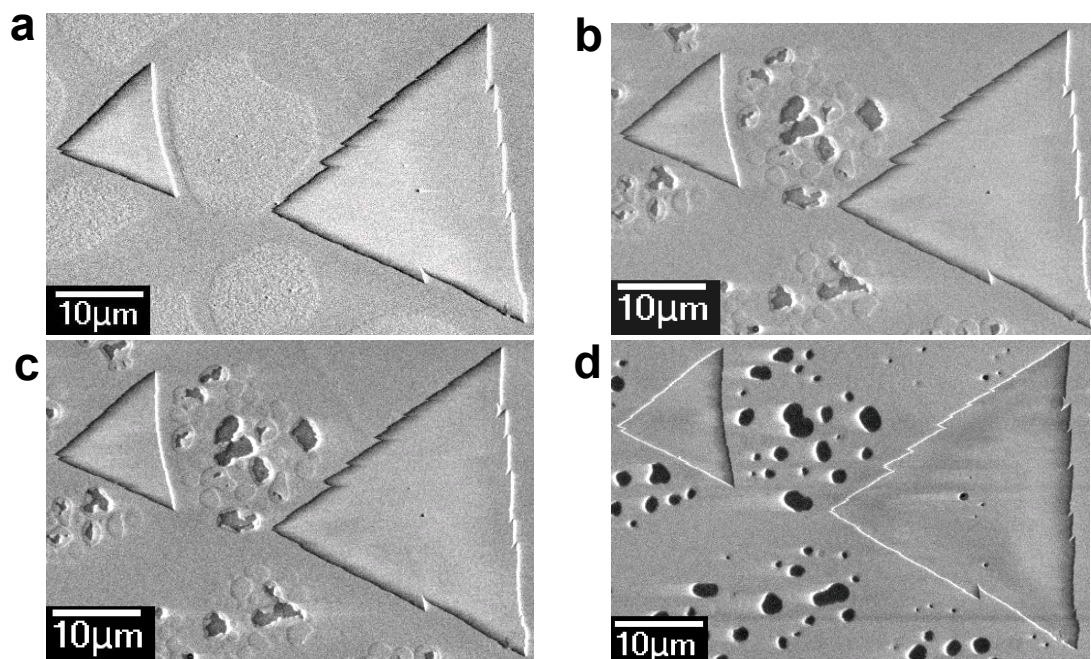


Figure S10. SEM images of pure WS₂ sample stored at dark and 99% RH for (a) 1 day, (b) 3 days, (c) 7 days, and (d) 14 days.

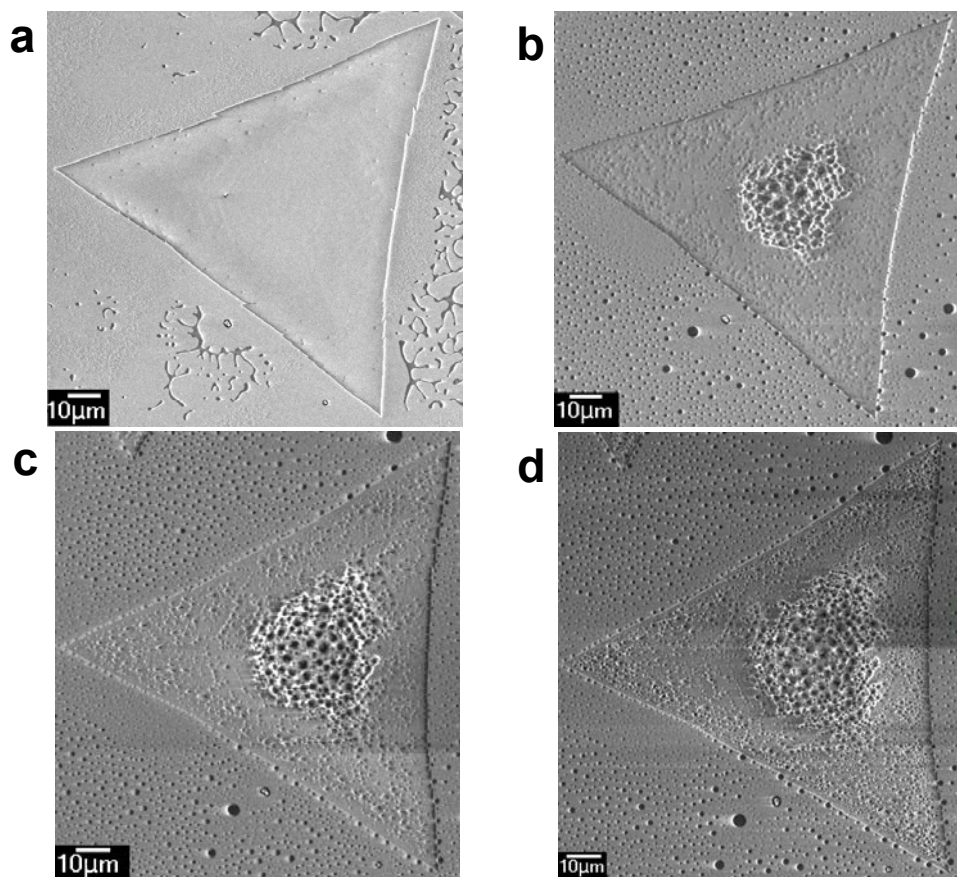


Figure S11. SEM images of WS₂ sample with NaCl assistant stored at dark and 99% RH for (a) 1 day, (b) 3 days, (c) 7 days, and (d) 14 days.

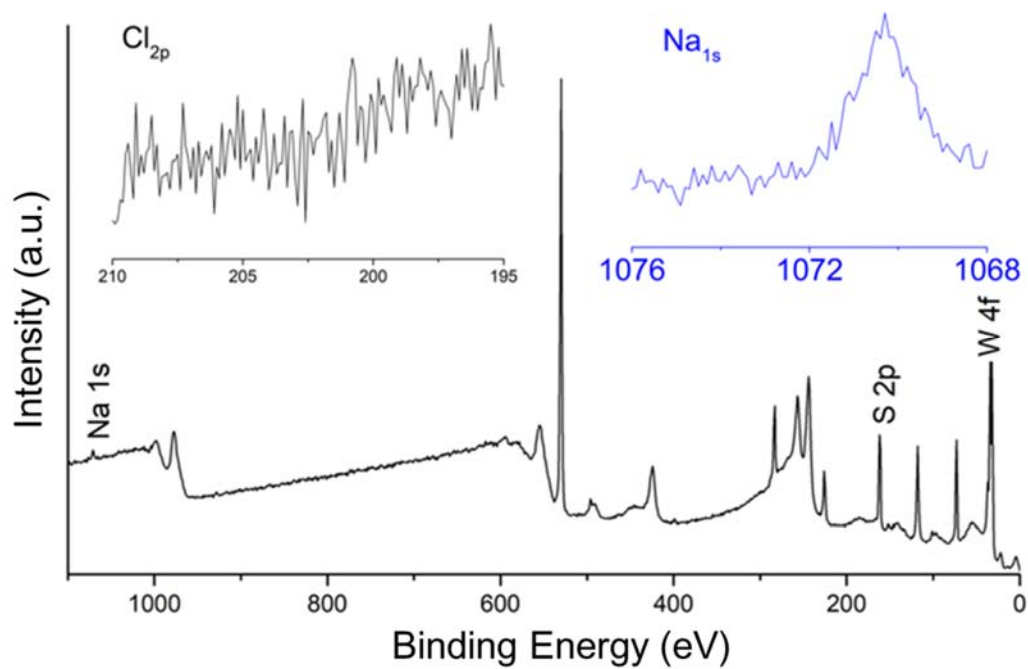


Figure S12. XPS survey spectra of the WS₂ samples grown by NaCl-assisted CVD collected after the degradation process. The insets show the high resolution XPS spectra of the energy ranges of Na 1s and Cl 2p. No Cl 2p signal is observed.⁴

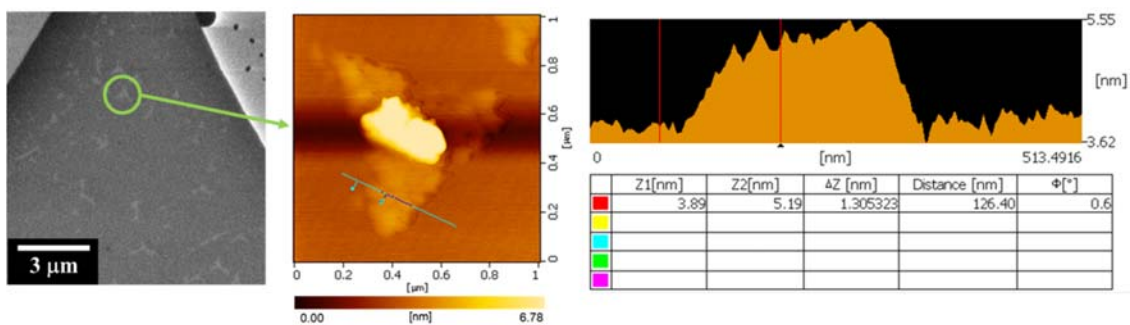


Figure S13. SEM image and enlarged AFM image of a NaCl-assisted WS₂ monolayer stored under fluorescent light and 55% RH for 1 day. There was a small raised area in the middle of the triangular oxide island. The height of this island is ~1.3 nm.

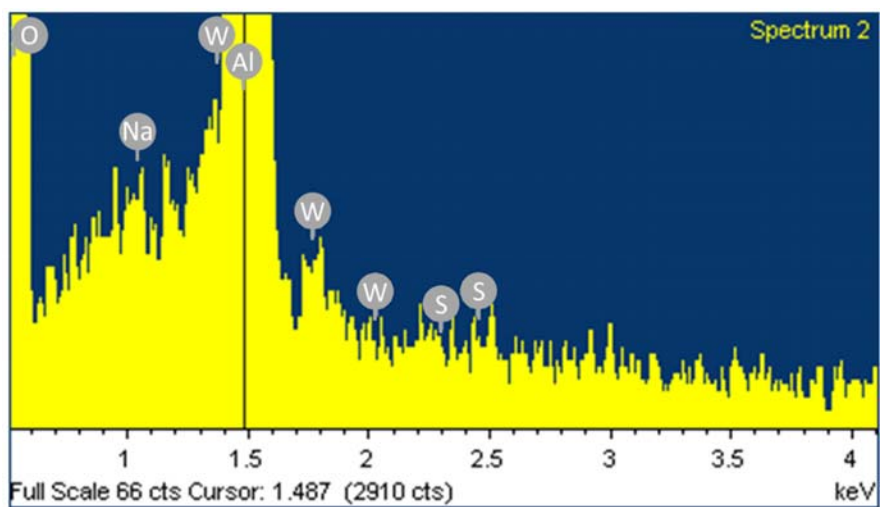


Figure S14. Energy-dispersive X-ray spectroscopy (EDS) analysis on the big dot in Figure 4g and found the presence of O, Na, W, Al, and S elements.^{5,6}

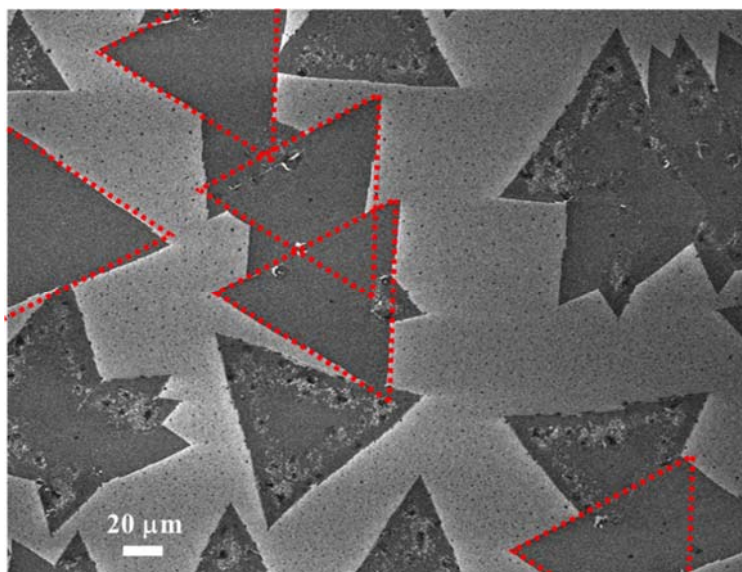


Figure S15. SEM image of the NaCl-assisted WS₂ monolayers in a specific orientation marked by red dashed triangles, which have anti-photooxidation properties.

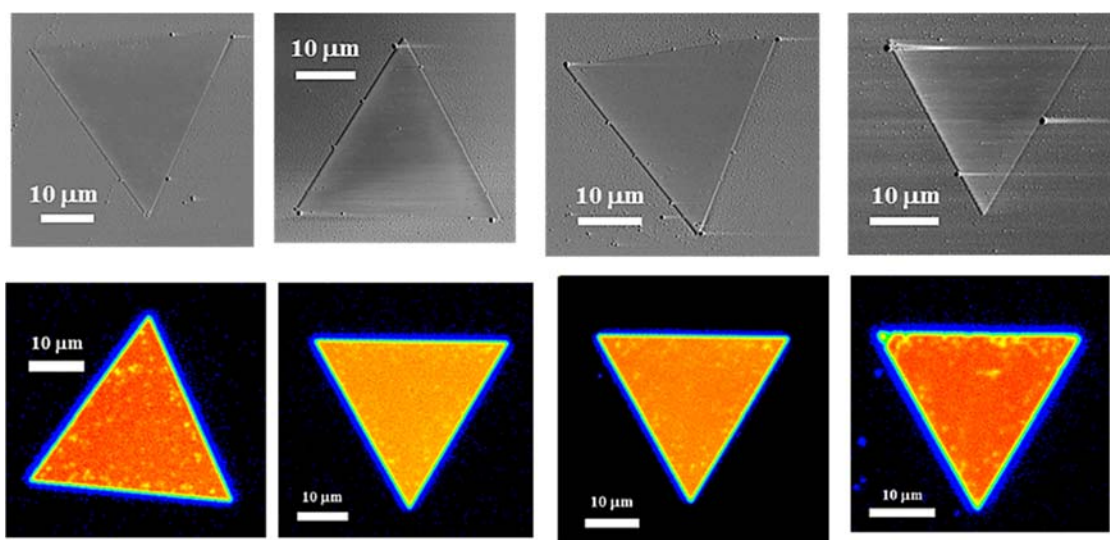


Figure S16. SEM and SHG images of Type I NaCl-assisted WS₂ flakes after 7.5 hours of fluorescent light exposure with fluence of $\sim 4.5 \times 10^4 \text{ Jm}^{-2}$ at 55% RH. These are eight different flakes. These images show the photooxidation effect of WS₂ grown with 0.5 mg NaCl assistant.

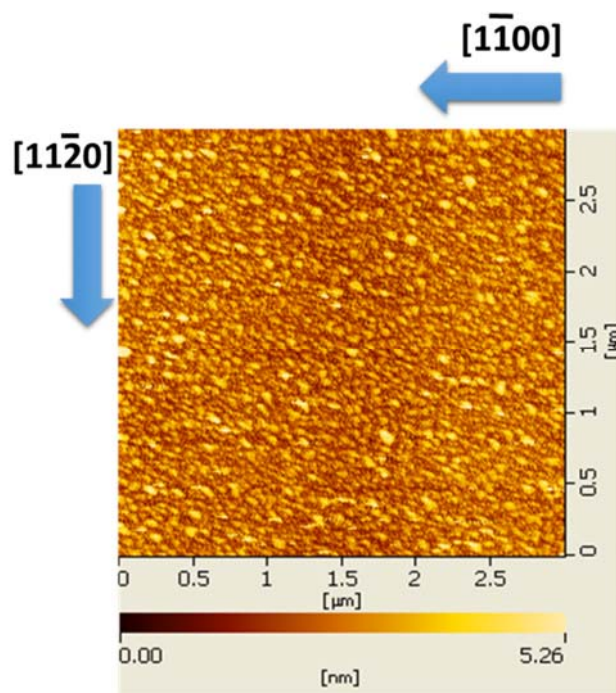


Figure S17. AFM image of the sapphire substrate, which went through a complete CVD growth process, except that WO_3 and sulfur powder precursors were not added. The growth temperature of our system is 850 °C, which is lower than the 950 °C indicated in the literature.^{7,8} This makes the sapphire substrate not produce obvious steps.

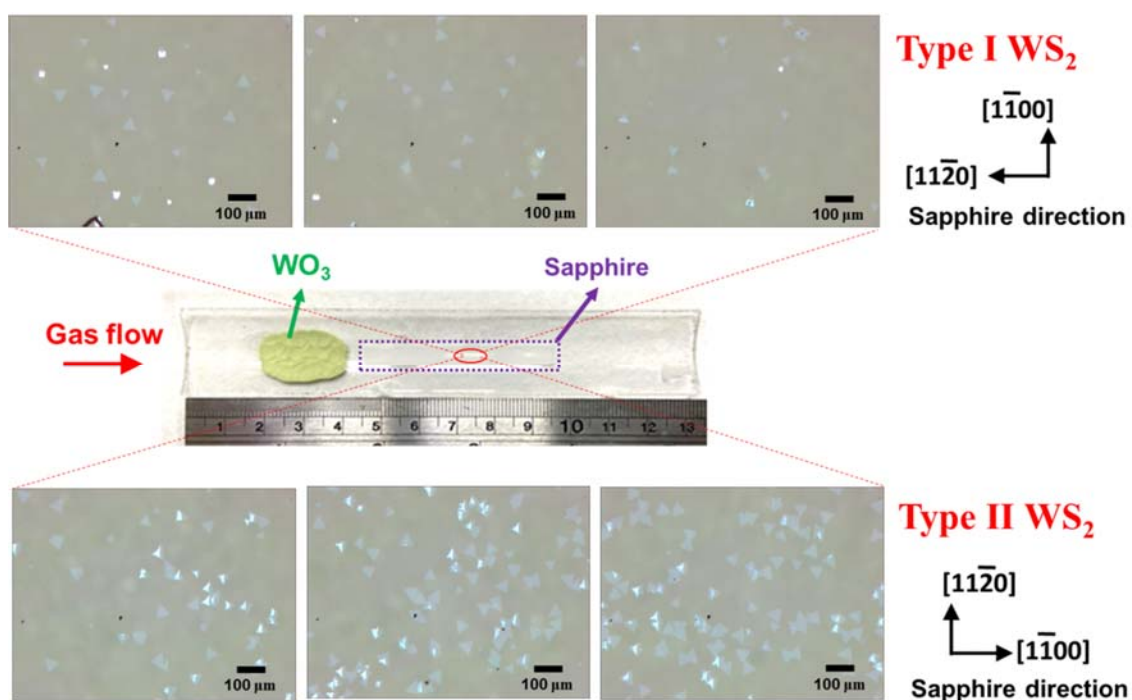


Figure S18. The OM images show the growth density of Type I and Type II WS₂. The Type II WS₂ has a higher growth density.

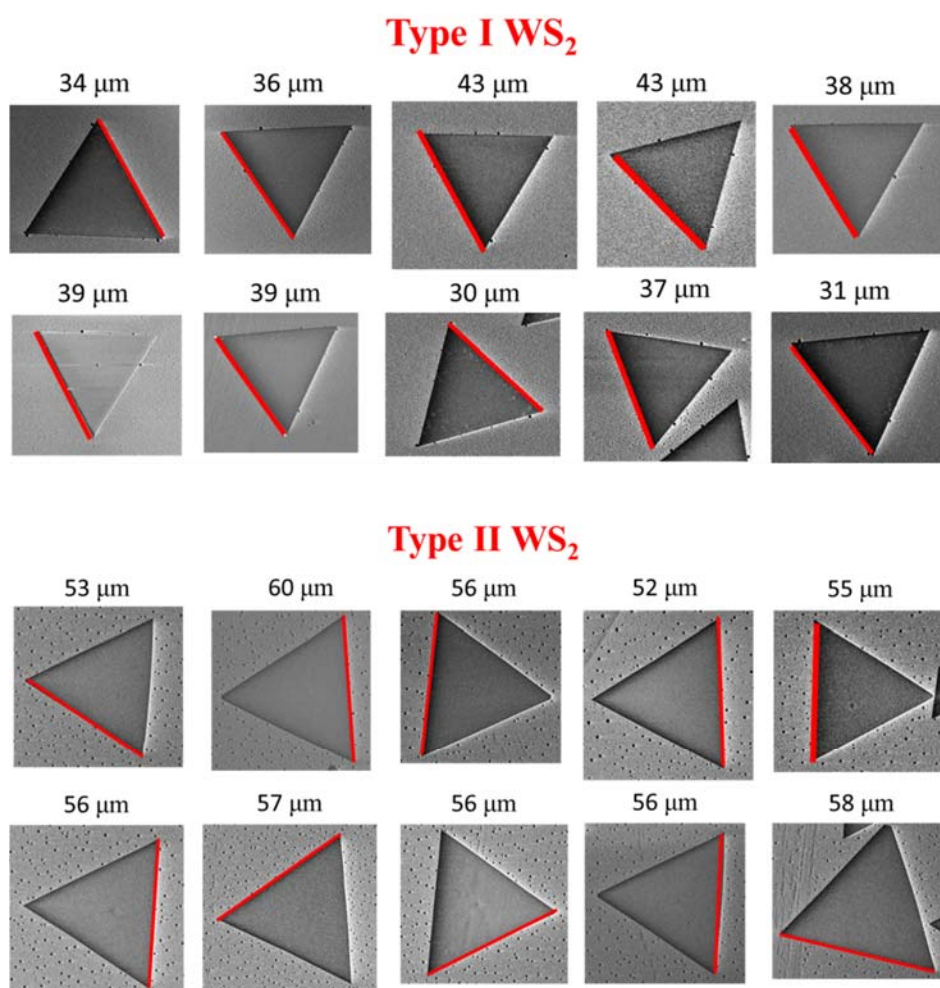


Figure S19. The sizes of Type I and Type II WS₂ are counted from the SEM images. The average size of Type I and Type II WS₂ is 37 and 56 μm, respectively. The red lines represent the side length.

References

1. Lin, Y.-C.; Li, S.; Komsa, H.-P.; Chang, L.-J.; Krasheninnikov, A. V.; Eda, G.; Suenaga, K. Revealing the Atomic Defects of WS₂ Governing Its Distinct Optical Emissions. *Adv. Funct. Mater.* **2018**, *28*, 1704210.
2. Lin, Y. C.; Bjorkman, T.; Komsa, H. P.; Teng, P. Y.; Yeh, C. H.; Huang, F. S.; Lin, K. H.; Jadczyk, J.; Huang, Y. S.; Chiu, P. W., et al. Three-fold rotational defects in two-dimensional transition metal dichalcogenides. *Nat. Commun.* **2015**, *6*, 6736.
3. Jimmy C. Kotsakidis, Quianhui Zhang, Amadeo L. Vazquez de Parga, Marc Currie, Kristian Helmerson, D. Kurt Gaskill, Michael S. Fuhrer, Oxidation of Monolayer WS₂ in Ambient Is a Photoinduced Process, *Nano Lett.* **2019**, *19*, 5205–5215.
4. Cai, Z.; Shen, T.; Zhu, Q.; Feng, S.; Yu, Q.; Liu, J.; Tang, L.; Zhao, Y.; Wang, J.; Liu, B.; Cheng, H.-M. Dual-Additive Assisted Chemical Vapor Deposition for the Growth of Mn-Doped 2D MoS₂ with Tunable Electronic Properties. *Small* **2020**, *16*, 1903181.
5. Lim, Y. V.; Wang, Y.; Kong, D.; Guo, L.; Wong, J. I.; Ang, L. K.; Yang, H. Y. Cubic-shaped WS₂ nanopetals on a Prussian blue derived nitrogen-doped carbon nanoporous framework for high performance sodium-ion batteries. *J. Mater. Chem. A* **2017**, *5*, 10406–10415.
6. Rostami, S.; Azizi, S. N.; Ghasemi, S. Preparation of an efficient electrocatalyst for oxalic acid oxidation based on Ag-doped ZSM-5 nanozeolites synthesized from bagasse. *J. Electroanal. Chem.* **2017**, *788*, 235–245.
7. Liang Chen, Bilu Liu, Mingyuan Ge, Yuqiang Ma, Ahmad N. Abbas, Chongwu Zhou, Step-Edge-Guided Nucleation and Growth of Aligned WSe₂ on Sapphire via a Layer-over-Layer Growth Mode. *ACS Nano* **2015**, *9*, 8368–8375.
8. Stefano Curiotto, Dominique Chatain, Surface morphology and composition of c-, a- and m-sapphire surfaces in O₂ and H₂ environments. *Surface Science* **2009**, *603*, 2688–2697.