

## **Suppressing vanadium dissolution of V<sub>2</sub>O<sub>5</sub> via in situ polyethylene glycol intercalation towards ultralong lifetime room/low-temperature zinc-ion batteries**

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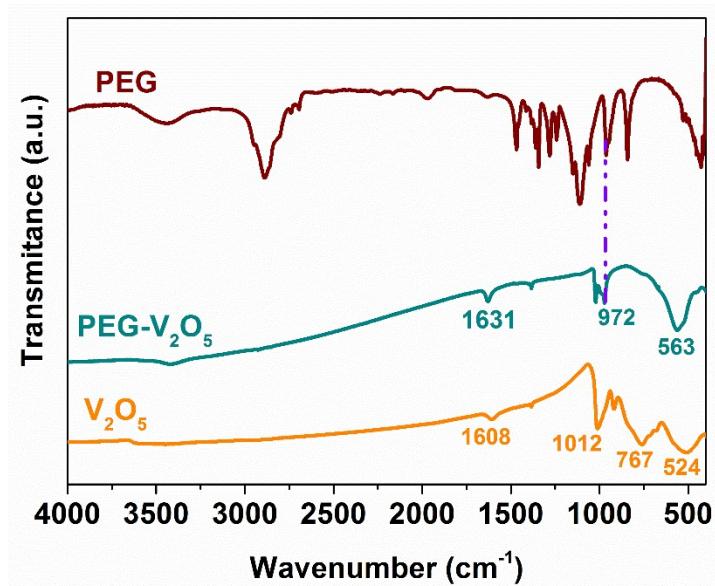
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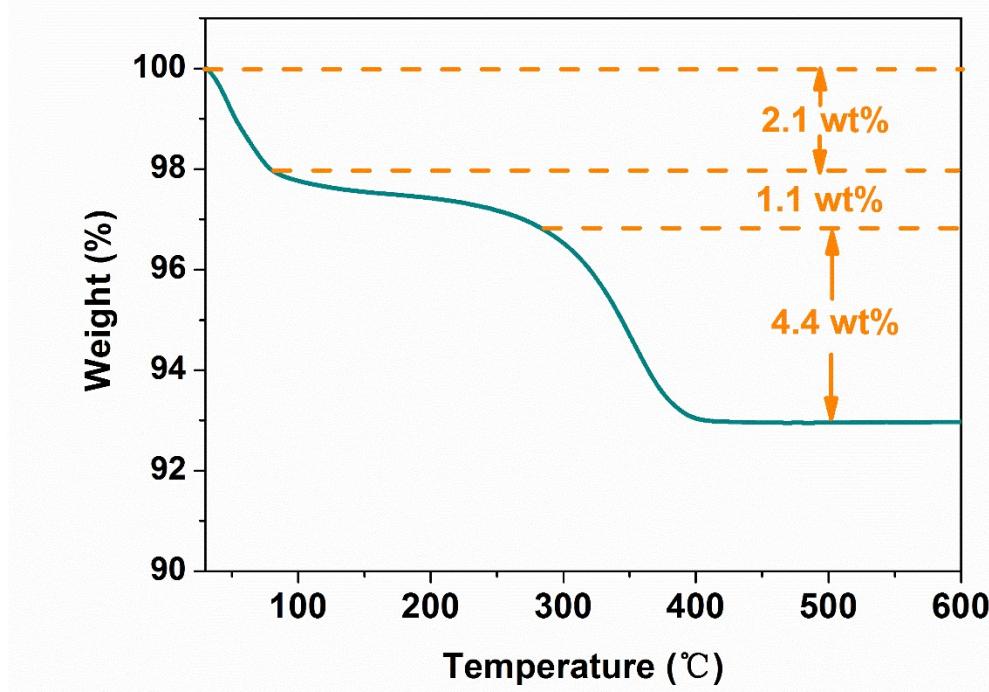
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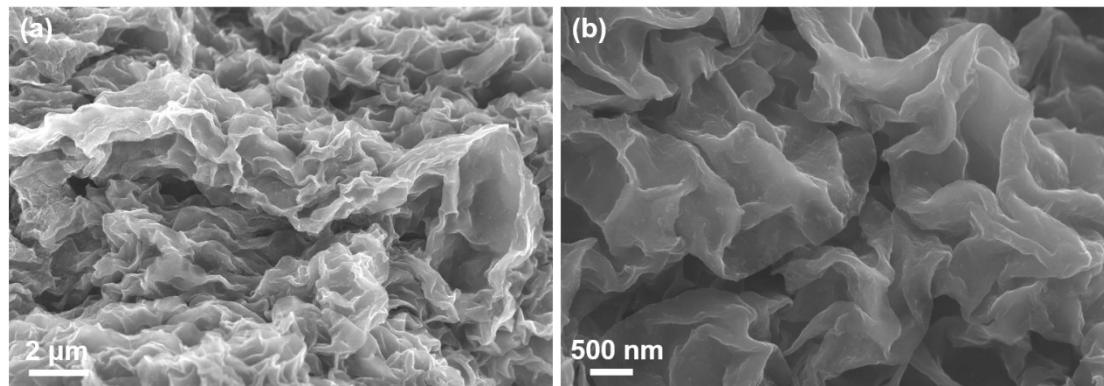
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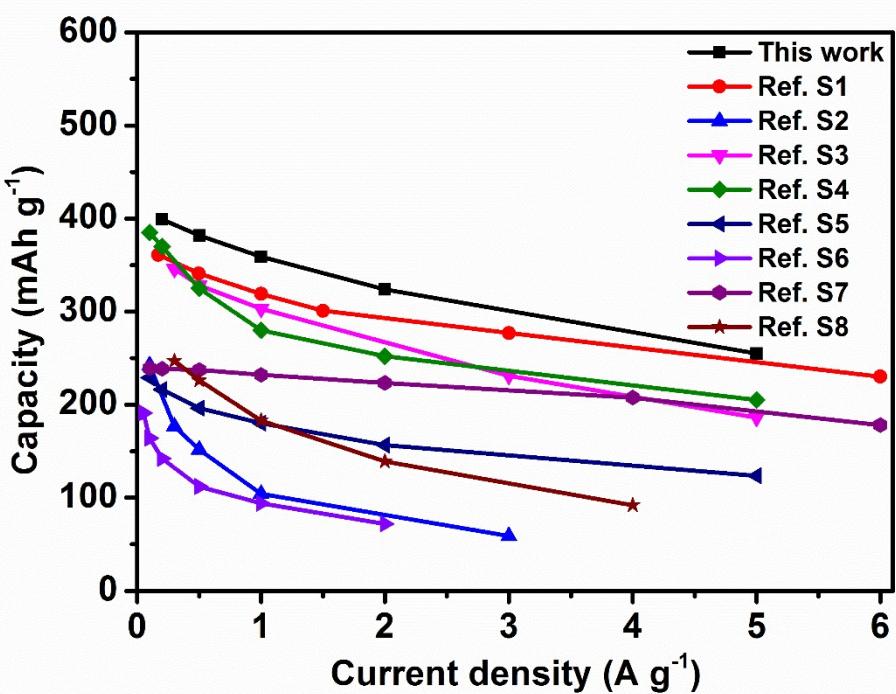
**Figure S1.** FTIR spectra of PEG, PEG- $\text{V}_2\text{O}_5$  and raw  $\text{V}_2\text{O}_5$ .



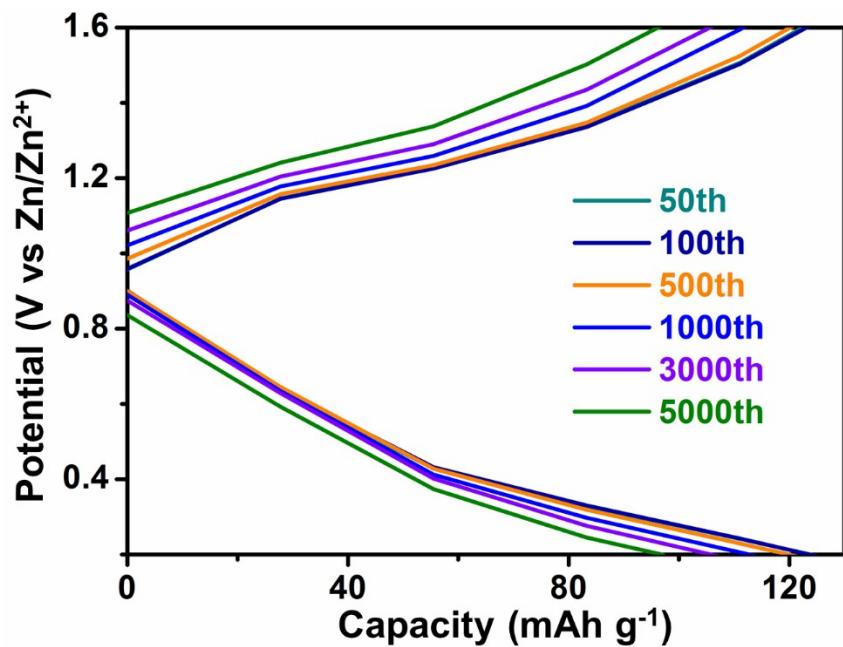
**Figure S2.** The TG curves of PEG- $\text{V}_2\text{O}_5$ .



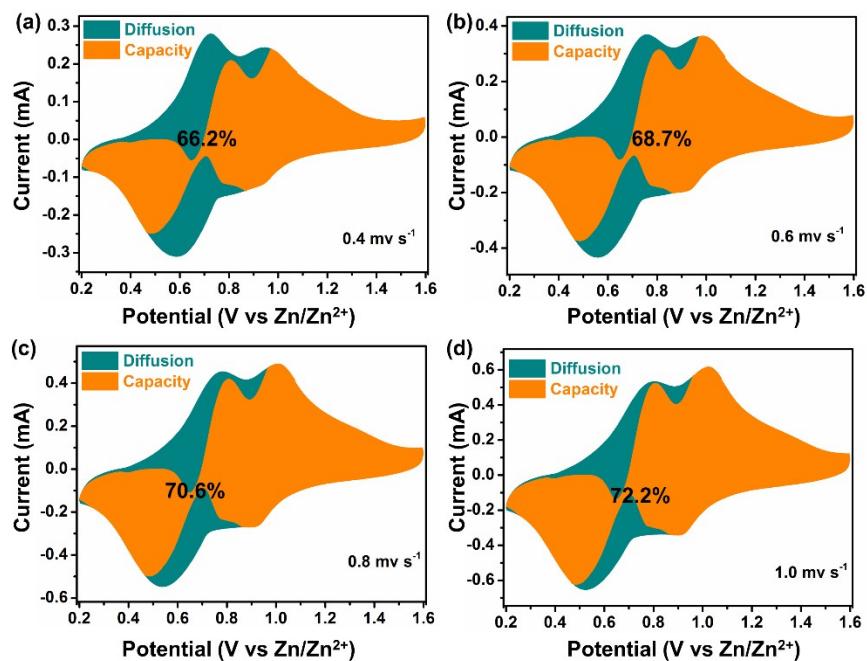
**Figure S3.** (a, b) SEM images of HVO.



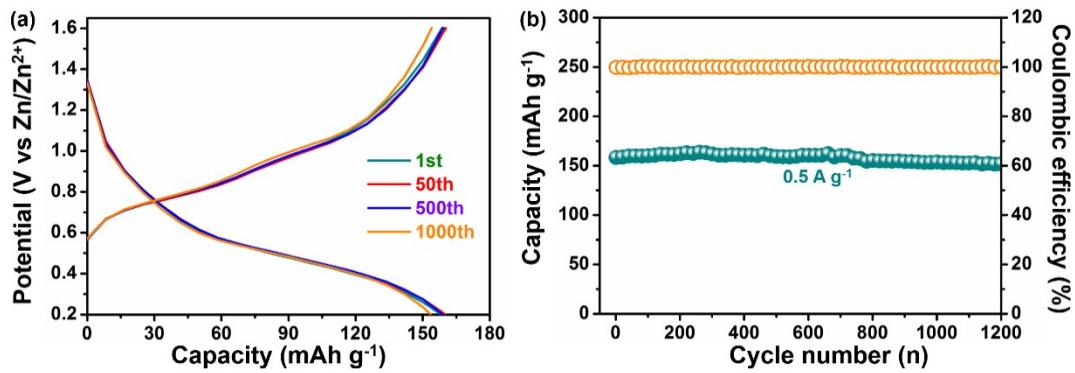
**Figure S4.** Comparison of the rate capability of the PEG-V<sub>2</sub>O<sub>5</sub> with other previously reported ZIBs cathodes.



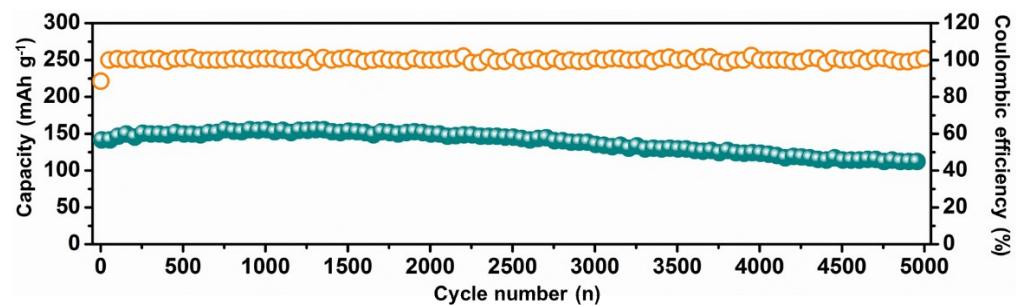
**Figure S5.** Galvanostatic charge-discharge profiles during different cycling numbers at  $10.0 \text{ A g}^{-1}$ .



**Figure S6.** Separation of the capacitive and diffusion-controlled current contribution at different scan rates.



**Figure S7.** Zinc-ion storage performance of PEG-V<sub>2</sub>O<sub>5</sub> electrode at a low temperature of -10 °C: (a) The charge-discharge curves and (b) cycling performance at 0.5 A·g<sup>-1</sup>.



**Figure S8.** Long cycling performance of PEG-V<sub>2</sub>O<sub>5</sub> electrode at a low temperature of -10 °C at 2.0 A g<sup>-1</sup>.

**Table S1.** Comparison of the diffusion coefficient ( $D_{Zn^{2+}}$ ) with other reported cathode materials.

Cathode Materials	$D_{Zn^{2+}}$ (cm <sup>2</sup> s <sup>-1</sup> )	Ref.
PEG-V <sub>2</sub> O <sub>5</sub>	$10^{-9} \sim 10^{-11}$	This work
Mn <sub>0.15</sub> V <sub>2</sub> O <sub>5</sub> •nH <sub>2</sub> O	$10^{-10} \sim 10^{-12}$	[9]
$\alpha$ -MnO <sub>2</sub>	$10^{-12} \sim 10^{-17}$	[2]
LiFePO <sub>4</sub>	$10^{-15} \sim 10^{-19}$	[10]
MnO <sub>2</sub> nanospheres	$10^{-12} \sim 10^{-15}$	[11]
$\delta$ -MnO <sub>2</sub>	$10^{-12} \sim 10^{-15}$	[12]
PANI-VOH	$10^{-13} \sim 10^{-16}$	[3]
Al <sub>0.2</sub> V <sub>2</sub> O <sub>5</sub>	$10^{-12} \sim 10^{-13}$	[13]

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