Electronic Supplementary Information

Synthesis of Pd Nanorod Arrays on Au Nanoframes for Excellent Ethanol Electrooxidation

Qinru Yun,^a Juan Xu,^a Tingcha Wei,^{a,b} Qifeng Ruan,^c Xingzhong Zhu*^{a,b} and Caixia Kan*^{a,b}

 ^a College of Science, Nanjing University of Aeronautics and Astronautics, Nanjing 211106, China. E-mail: cxkan@nuaa.edu.cn; xzzhu@nuaa.edu.cn
^b Key Laboratory of Aerospace Information Materials and Physics (NUAA), MIIT, Nanjing 211106, China
^c Engineering Product Development, Singapore University of Technology and Design,

Singapore 487372, Singapore

Supplementary figures



Fig. S1 (a) TEM image of Au NBPs with average waist width of 26 ± 2 nm and length of 83 ± 3 nm. (b) TEM image of the Au NBP@Ag nanorods with average diameter of 27 ± 2 nm and length of 121 ± 4 nm. (c) Extinction spectra of Au NBPs and Au NBP@Ag nanorods.



Fig. S2 The HAADF-STEM image of a single Au NF@Pd array.



Fig. S3 HRTEM image of a single Au NBP-embedded Au NF.



Fig. S4 XRD patterns of the Au NFs and Au NF@Pd arrays, respectively.



Fig. S5 FDTD simulations with Pd arrays only deposited on the outside Au NFs. The longitudinal dipolar plasmon wavelength of the starting Au NBP@Au nanoframe is 745 nm.(a) Schematic models utilized in the simulations. (b) Simulated extinction spectra for the nanostructure.



Fig. S6 (a) Extinction spectra of Au NF@Pd nanostructures synthesized with different surfactants. (b–d) TEM images of Au NF@Pd nanostructures synthesized with CTAB, CTAC and CPC, respectively.



Fig. S7 (a) Extinction spectra of Au NF@Pd arrays synthesized at different temperatures. (b–d) TEM images of Au NF@Pd arrays synthesized at 30 °C, 65 °C and 90 °C, respectively.



Fig. S8 (a) Extinction spectra of Au NF@Pd arrays with different reaction times. (b–d) TEM images of Au NF@Pd arrays synthesized at 0.5 h, 3 h and 17 h, respectively.



Fig. S9 CV curves of commercial Pd/C, Au NBP@Pd array, Au NF@Pd and Au NF@Pd array in a N₂-saturated (a) aqueous KOH and (b,c) mixture of ethanol and KOH, respectively. The currents shown in (a,c) and (b) are normalized by the Pd mass loaded and the ECSA values, respectively.



Fig. S10 (a) Au 4f and (b) Pd 3d XPS spectra for the Au NF@Pd arrays after electrocatalysis.



Fig. S11 TEM image of the Au NF@Pd arrays after CA measurement. The sample was collected by ultrasonicated the working electrode after CA test.

Supplementary tables

Table S1 The contents of Au, Pd, and Ag elements in different Au-Pd catalysts before andafter the electrocatalysis.

	Au (µg)		Pd (µg)		Ag (µg)	
	before	after	before	after	before	after
Au NBP@Pd array	1.018	0.87	0.278	0.227	0.0496	0.0433
Au NF@Pd	1.063	0.914	0.305	0.232	0.1634	0.1355
Au NF@Pd array	1.232	0.974	0.3	0.262	0.1637	0.1357