

## Supporting Information

### Preparation and Characterization of $\text{Li}_{1.12}\text{K}_{0.05}\text{Mn}_{0.57}\text{Ni}_{0.24}\text{Nb}_{0.02}\text{O}_2$ Cathode Material with Highly Improved Rate Cyclic Performance for Lithium Ion Batteries

Cong Liu,<sup>1</sup> Shuang Zhang,<sup>1</sup> Yuanyuan Feng,<sup>1</sup> Xiaowei Miao,<sup>1</sup> Gang Yang<sup>\*,1</sup> Jie Li<sup>\*,2</sup>

<sup>1</sup>. Suzhou Key Laboratory of Functional Ceramic Materials, Changshu Institute of Technology, Changshu, 215500, P.R. China

<sup>2</sup>. Department of Energy, Politecnico di Milano, Campus Bovisa - Via Lambruschini 4a, Milan, 20156, Italy

\* Corresponding authors

E-mail: [gyang@cslg.edu.cn](mailto:gyang@cslg.edu.cn)

[jie1.li@polimi.it](mailto:jie1.li@polimi.it)

**Table 1** Refined lattice parameters of LMN and LMN-K/Nb.

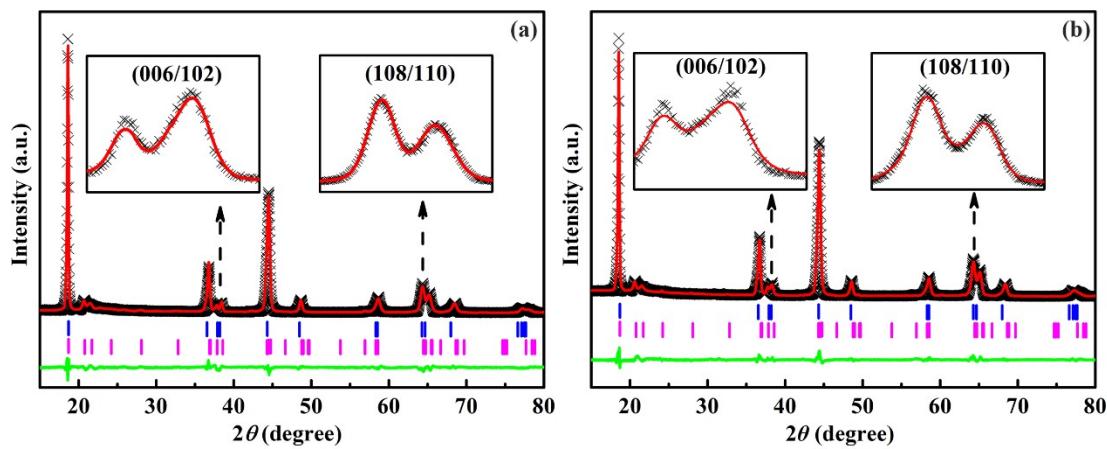
sample	LMN		LMN-K/Nb	
	LMO	LMNO	LMO	LMNO
Lattice parameters	<i>a</i>	4.9641	2.8606	4.9558
(Å)	<i>b</i>	8.5631		8.5827
	<i>c</i>	5.0241	14.2608	5.0337
				14.2754
$I_{003}/I_{104}$		2.283		1.656
Reliability and weighted factors	$R_{wp}$ (%)	9.6		5.4
	$R_p$ (%)	6.8		5.2
	$\chi^2$	2.1		1.8

**Table S2** The fitting parameters for the Nyquist plots.

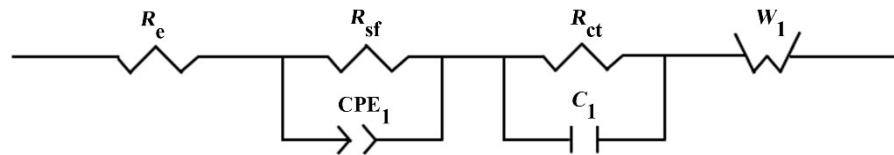
	3 <sup>rd</sup> cycle discharge to 3.5 V			50 <sup>th</sup> cycle discharge to 3.5 V		
	$R_e/\Omega$	$R_{sf}/\Omega$	$R_{ct}/\Omega$	$R_e/\Omega$	$R_{sf}/\Omega$	$R_{ct}/\Omega$
LMN	1.83	45.52	505.71	2.18	180.53	1325.32
LMN-K/Nb	1.47	13.58	90.63	2.02	50.36	266.16

**Table S3** Comparison of electrochemical performance of Li-rich oxides.

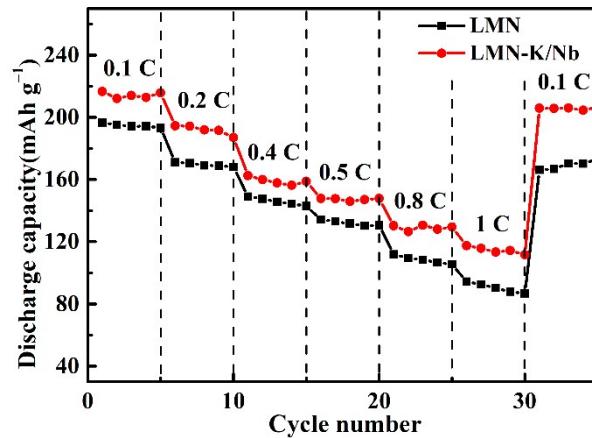
Materials	Initial coulomb efficiency	Capacity-current density	Cycling performance (cycles)	Ref.
$\text{Li}_{1.12}\text{K}_{0.05}\text{Mn}_{0.57}\text{Ni}_{0.24}\text{Nb}_{0.02}\text{O}_2$	77.2%	244 mAh g <sup>-1</sup> at 0.1 C 161 mAh g <sup>-1</sup> at 2 C 145 mAh g <sup>-1</sup> at 5 C	96% at 0.1 C (100) 88% at 2 C (200)	This work
$\text{Li}_{1.117}\text{K}_{0.05}\text{Mn}_{0.583}\text{Ni}_{0.25}\text{O}_2$	71.7%	229 mAh g <sup>-1</sup> at 0.1 C 145 mAh g <sup>-1</sup> at 2 C 100 mAh g <sup>-1</sup> at 5 C	98% at 0.1 C (100) 95% at 2 C (200)	[1]
$\text{Li}_{1.2}\text{Ni}_{0.133}\text{Co}_{0.133}\text{Mn}_{0.534}\text{O}_{1.8}\text{F}_{0.2}$	74%	212 mAh g <sup>-1</sup> at 0.1 C 150 mAh g <sup>-1</sup> at 0.2 C 50 mAh g <sup>-1</sup> at 1 C	80% at 0.1 C (100)	[2]
$\text{Li}_{1.17}\text{Ni}_{0.25}\text{Mn}_{0.58}\text{Mg}_{0.02}\text{O}_2$	78.1%	228 mAh g <sup>-1</sup> at 0.2 C 137 mAh g <sup>-1</sup> at 2 C 104 mAh g <sup>-1</sup> at 5 C	95% at 0.2 C (100)	[3]
$\text{Li}_{1.2}\text{Ni}_{0.18}\text{Co}_{0.18}\text{Mn}_{0.18}\text{Ti}_{0.26}\text{O}_2$	60%	175 mAh g <sup>-1</sup> at 0.1 C 105 mAh g <sup>-1</sup> at 1 C 82 mAh g <sup>-1</sup> at 2 C	97% at 0.1 C (182)	[4]
Li <sub>3</sub> PO <sub>4</sub> -coated $\text{Li}_{1.2}\text{Mn}_{0.54}\text{Ni}_{0.13}\text{Co}_{0.13}\text{O}_2$	75.78%	226 mAh g <sup>-1</sup> at 0.1 C	78% at 1 C (100)	[5]
Li <sub>3</sub> PO <sub>4</sub> -coated $\text{Li}_{1.2}\text{Mn}_{0.6}\text{Ni}_{0.2}\text{O}_2$	77%	240 mAh g <sup>-1</sup> at 0.1 C 159 mAh g <sup>-1</sup> at 1 C	88% at 0.1 C (50) 84% at 1 C (50)	[6]
LiAlO <sub>2</sub> -coated $\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.56}\text{O}_2$	74.9%	237 mAh g <sup>-1</sup> at 0.06 C 213 mAh g <sup>-1</sup> at 0.2 C 112 mAh g <sup>-1</sup> at 5 C	91% at 0.2 C (100)	[7]
$\text{Li}_{1.2}\text{Mn}_{0.6}\text{Ni}_{0.2}\text{O}_2$ (Sol-freeze-drying method)	61%	242 mAh g <sup>-1</sup> at 0.2 C 148 mAh g <sup>-1</sup> at 2 C 84 mAh g <sup>-1</sup> at 5 C	96% at 0.2 C (100)	[8]
$\text{Li}_{1.2}\text{Ni}_{0.2}\text{Mn}_{0.6}\text{O}_2$ (Electrolyte additive)	70.2%	238 mAh g <sup>-1</sup> at 0.1 C 170.0 mAh g <sup>-1</sup> at 1 C 100 mAh g <sup>-1</sup> at 5C	80% at 2 C (200)	[9]
$\text{Li}_{1.2}\text{Mn}_{0.55}\text{Ni}_{0.15}\text{Co}_{0.1}\text{O}_2$ (Electrolyte additive)	70.03%	248mAh g <sup>-1</sup> at 0.1 C	61% at 0.1 C (200)	[10]



**Fig. S1** Rietveld refinement of the XRD patterns for LMN (a) and LMN-K/Nb (b).



**Fig. S2** The equivalent circuit model for Nyquist plots.



**Fig. S3** The rate performance of the LMN and LMN-K/Nb in full cell at 0.1 C, 0.2 C, 0.4 C, 0.5 C, 0.8 C and 1 C.

## References

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