

## Supporting Information

### Porphyrin-Based Metallacage for Enhanced Photodynamic Therapy

Jingjing Jiao<sup>a</sup>, Jing He<sup>a</sup>, Mengmeng Li<sup>a</sup>, Jingxia Yang<sup>a</sup>, Hong Yang<sup>a</sup>, Xiaoqing Wang<sup>b,\*</sup>, Shiping Yang<sup>a,\*</sup>

<sup>a</sup>The Key Laboratory of Resource Chemistry of Ministry of Education, Shanghai Frontiers Science Center of Biomimetic Catalysis, Shanghai Normal University, Shanghai 200234, China;

<sup>b</sup>Department of Chemistry, College of Science, North University of China, Taiyuan 030051, China.

\*Corresponding authors. E-mail address: [xqwang@nuc.edu.cn](mailto:xqwang@nuc.edu.cn), [shipingy@shnu.edu.cn](mailto:shipingy@shnu.edu.cn)

### Contents

<b>1. Materials and reagents</b> .....	2
<b>2. Characterization</b> .....	2
<b>3. Method</b> .....	2
3.1 Synthesis of TAPP, Zn-TAPP and PM.....	2
3.2 Synthesis of PM@mPEG.....	2
3.3 Therapeutic effect of PDT in vivo .....	2
<b>4. Figures and Table</b> .....	3
Fig. S1 Composite route diagram of PM .....	3
Fig. S2 <sup>1</sup> H NMR of TAPP, Zn-TAPP and PM .....	3-4
Fig. S3 <sup>13</sup> C NMR of TAPP and Zn-TAPP .....	5
Fig. S4 FT-ICR MS of PM .....	6
Fig. S5 Fluorescence emission spectra of Zn-TAPP and PM .....	7
Fig. S6 Stability of PM@HAase-mPEG in Saline, FBS and RPMI-1640.....	8
Fig. S7 UV-vis spectra of PM@mPEG .....	9
Fig. S8 SEM and DLS of PM@mPEG .....	9
Fig. S9 Gelatin test .....	9
Fig. S10 The release of HAase from PM@HAase-mPEG by BCA protein assay .....	10
Fig. S11 Cellular uptake of materials by ICP at different time .....	10
Fig. S12 Dark toxicity of PM@mPEG .....	10
Table. S1 PDT efficiency of TAPP and porphyrin-based MOFs .....	11
Fig. S13 Quantitative analysis of AM/PI staining .....	11
Fig. S14 Hemolysis test .....	11
Fig. S15 Photos of the mice after various treatment .....	12
Fig. S16 Photos of the tumors after various treatment .....	12
Fig. S17 Changes in weight of 4T1 tumor-bearing mice .....	13
Fig. S18 H&E staining of mice organs .....	13

## 1. Materials and Reagents

Solvents and raw materials are available on the market without further processing and purification. Nitrobenzaldehyde was purchased from TCL. Acetic anhydride, pyridine and hydrated zinc acetate were purchased from Greagent. Stannous chloride, 2-pyridinecarboxaldehyde and trifluoromethanesulfonate were purchased from Adamas. Trypsin-EDTA, PBS, DMEM high glucose, fetal bovine serum (FBS) and RPMI-1640 medium were purchased from Adamas. 4T1 and RAW 264.7 cells were obtained from ATCC. Calcein AM, propidium iodide (PI) were purchased from Shanghai Yeasen BioTechnologies co.,Ltd. 2,7-dichlorodihydrofluorescein diacetate (DCFH-DA) was purchased from APExBIO.

## 2. Characterization

<sup>1</sup>H NMR spectra were probed on 400 MHz Bruker spectrometer using deuterated DMSO or DMF as the solvent. The fourier transform ion cyclotron resonance mass spectrometry (FT-ICR MS) was measured by Solarix 7.0T. Scanning electron microscope (SEM) images were collected on a Zeiss EVO MA 25/LS 25. Zeta potential and dynamic light scattering (DLS) measurements were measured using ZEN3690, Malvern Instrument Ltd. UV-vis measurements were obtained on a DU730. Inductively coupled plasma (ICP, Vista MPX ICP) was used for quantitative determination of zinc. Fluorescent measurements were obtained using Shimadzu F-4500 spectrofluorophotometer. Confocal laser scanning microscopy (CLSM) images were acquired by a SP5 II Leica confocal microscope. Hematoxylin and eosin (H&E) staining images were obtained with an ECLIPSE Ci-L microscope. TUNEL, CD31, HIF-1a staining images were obtained with a panoramic MIDI digital slice scanner.

## 3. Methods

### 3.1 Synthesis of TAPP, Zn-TAPP and PM

Synthesis of 5,10,15,20-tetraphenyl (4-aminophenyl) porphyrin (TAPP).

4-nitrobenzaldehyde (3.6 g, 23.8 mmol) and 4 mL acetic anhydride were dissolved in 100 mL propionic acid. The mixture was heated to reflux, and 1.6 mL pyrrole was slowly dripped into the solution, then the solution kept reflux for 1 h. After the mixture cooled to room temperature, the filter residue was washed with methanol and vacuum dried to get black powder. The powder was dissolved in 100 mL pyridine and the solution was refluxed for 1.5 h. The product was filtered and vacuum dried, purple powder tetra-nitrophenyl porphyrin (TNPP) was obtained. At last, TNPP was dissolved in hot HCl solution and reduced by SnCl<sub>2</sub>·2H<sub>2</sub>O. After cooling, the precipitation was neutralized with NaOH affording to the final product as purple powder (yield:69.9%). <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>, 400 MHz): 8.87 (8H, s, H4), 7.83 (8H, d, H3), 7.00 (8H, d, H2), 5.55(8H, s, H1), -2.75(2H, s, H5). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 101 MHz):148.65, 135.61, 131.03, 128.86, 120.72, 112.67. Synthesis of Zinc (II) 5,10,15,20-tetrakis(4-aminophenyl)-21H, 23H-porphine (Zn-TAPP).

TAPP (100 mg, 148 μmol) was dissolved in mixed solvent DMF/CHCl<sub>3</sub> (15 mL/45 mL) and methanol solution of hydrated zinc acetate (130 mg, 593 μmol) was added. The mixture was refluxed for 24 h, and cooled to room temperature, washed with water and pumped to obtain 77 mg green solid product (yield: 70.2%). <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>, 400 MHz): 8.82 (8H, s, H1), 7.78 (8H, d, H2), 6.94 (8H, d, H3), 5.44(8H, s, H4). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 101 MHz):149.75, 148.00, 135.25, 131.29, 130.45, 120.88, 112.32.

Synthesis of porphyrin-based metallacage (PM).

A mixture of Zn-TAPP (30 mg, 0.04 mmol), 2-pyridinaldehyde (16.5 μL, 0.16 mmol), zinc trifluoromethanesulfonic acid (20 mg, 0.56 mmol) and 5 mL DMF was heated at 70 °C for 12 h. PM was obtained by addition of ethyl ether into the reaction mixture as a purple solid 27 mg, about 43% yield. <sup>1</sup>H-NMR (DMF-*d*<sub>7</sub>, 400 MHz): 10.14 (24H, s, H6), 9.04 (24H, H13), 8.97 (24H, H1), 8.78 (24H, d, H13\*), 8.37 (24H, t, H2), 8.31 (24H, H9), 8.20 (24H, H3), 8.00 (24H, d, H4), 7.85 (24H, d, H9\*), 7.21 (24H, d, H8), 5.69 (24H, br, H8\*).

### 3.2 Synthesis of PM@mPEG.

The nanoparticles of PM@mPEG were prepared through matrix-encapsulation method. Briefly, 400 μL of DMF solution containing PM (4 mg) was injected into 4 mL deionized water containing DSPE-mPEG2000 (24 mg). Then the mixture was stirred at 600 rpm for 15 min. After dialysis for 48 h, an aqueous solution of PM@mPEG nanoparticles was obtained.

### 3.3 Therapeutic effect of PDT *in vivo*.

A 4T1 tumor bearing mouse model was established by subcutaneously inoculating 100 μL 4T1 cells with a density of 3×10<sup>7</sup> cells/mL into the right hindquarters of the mice. After the tumor grew to the size of mung bean, the mice were randomly divided into six groups (n = 3 for each group): (1) PBS; (2) L; (3) PM@mPEG; (4) PM@HAase-mPEG; (5) PM@mPEG + L; (6) PM@HAase-mPEG + L. The nanoparticles were injected into the mice at a dose of 20 mg/kg PM by caudal vein every four days and irradiated with a 660 nm laser for 10 min after 12 h injection each time. Mice weight and tumor volume were monitored daily. After 14 days, the mice were necked and sacrificed, and the tumors, heart, liver, spleen, lungs and kidneys were collected. Tumors from each group were weighed and photographed to show the inhibitory effect of each treatment on the tumors. Tumors and important organs sliced were collected for H&E, TUNEL, CD31 and HIF-1a staining, respectively.

#### 4. Figures and Table

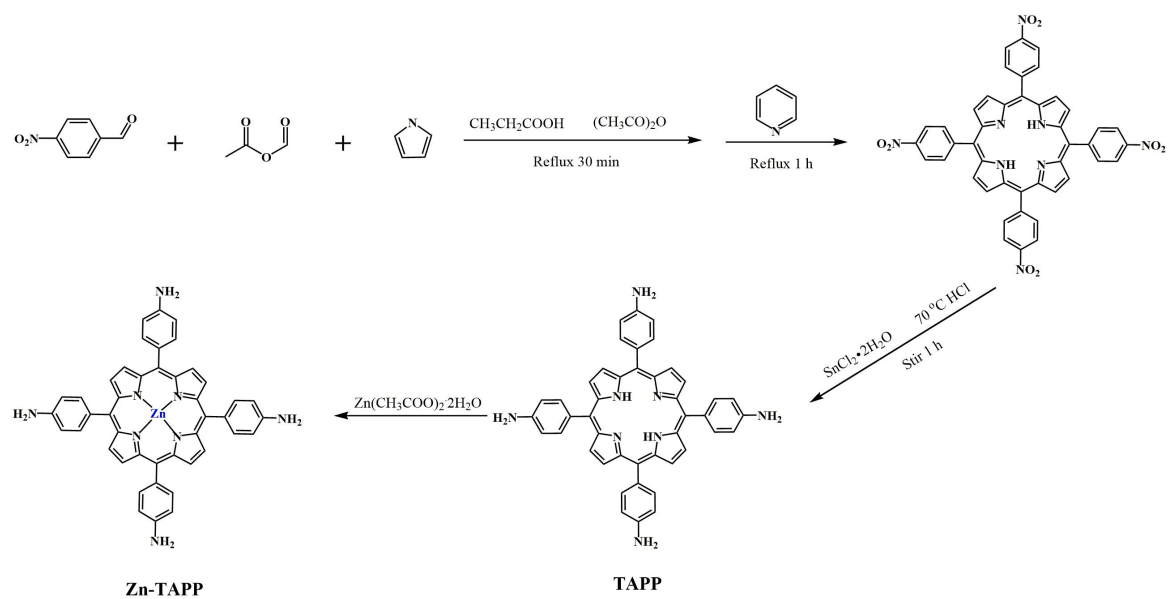
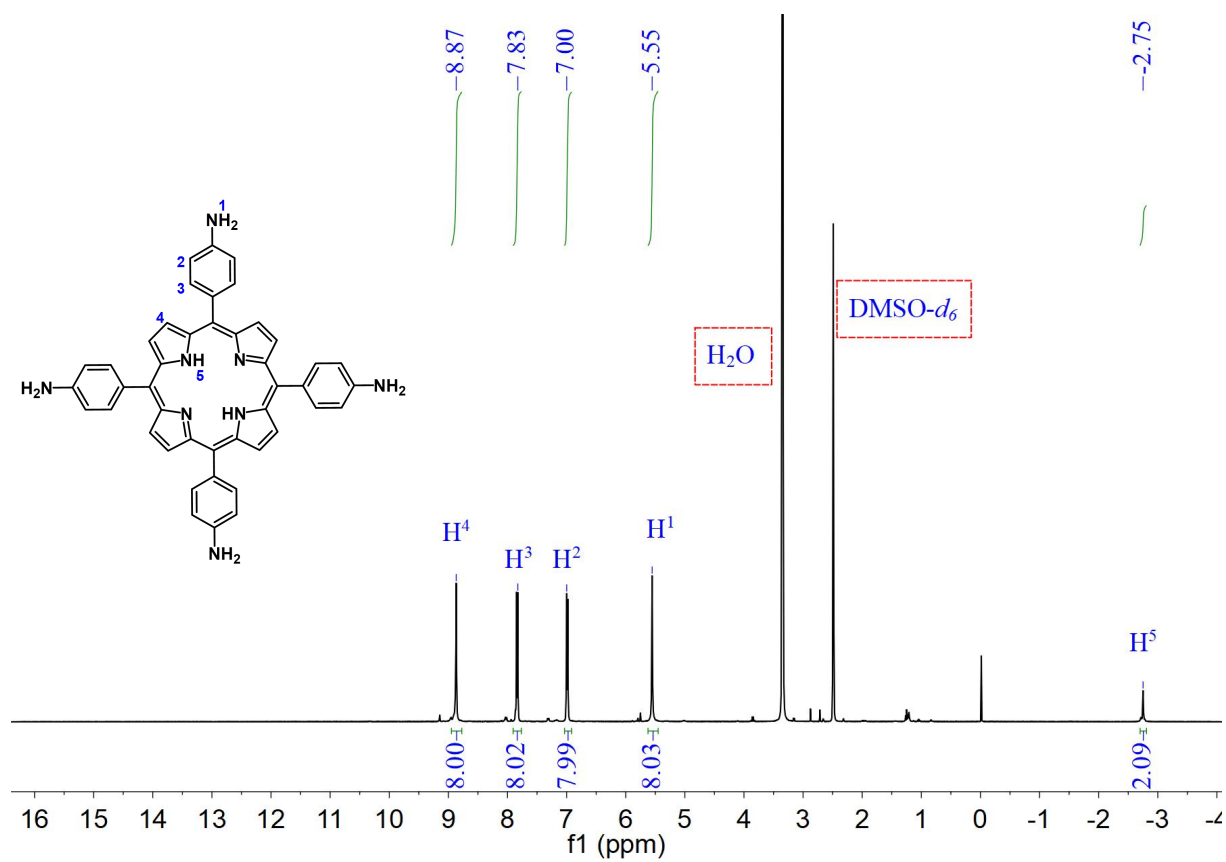


Fig. S1 Synthesis route of ligand.



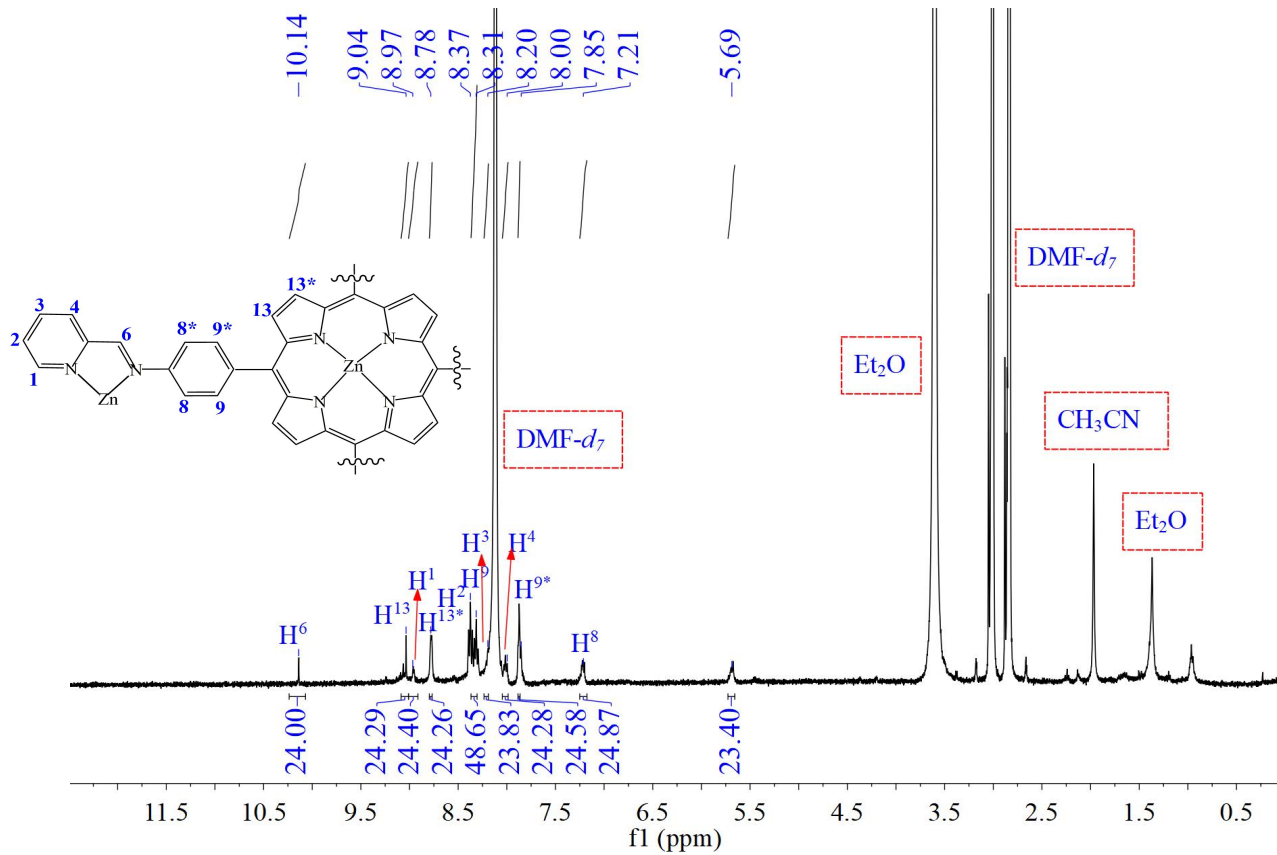
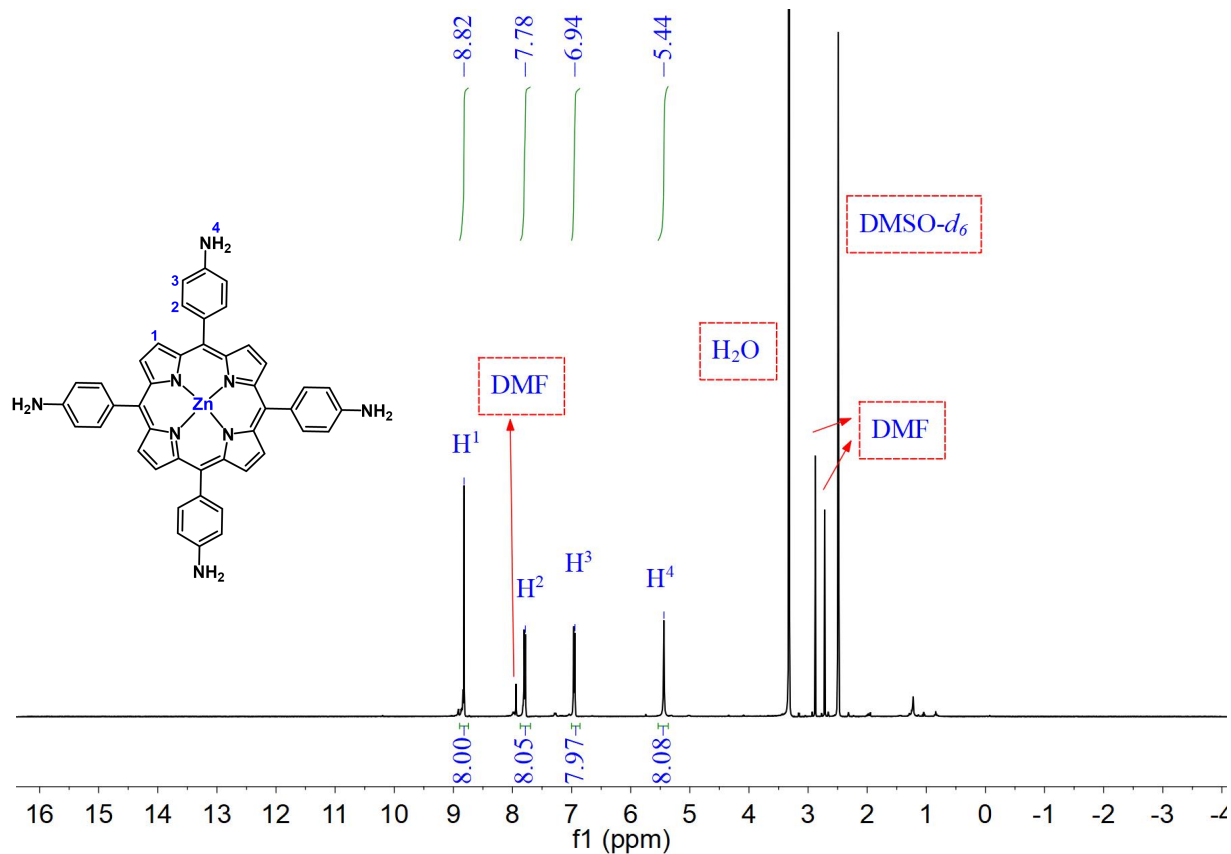


Fig. S2 <sup>1</sup>H NMR spectra of TAPP, Zn-TAPP and PM.

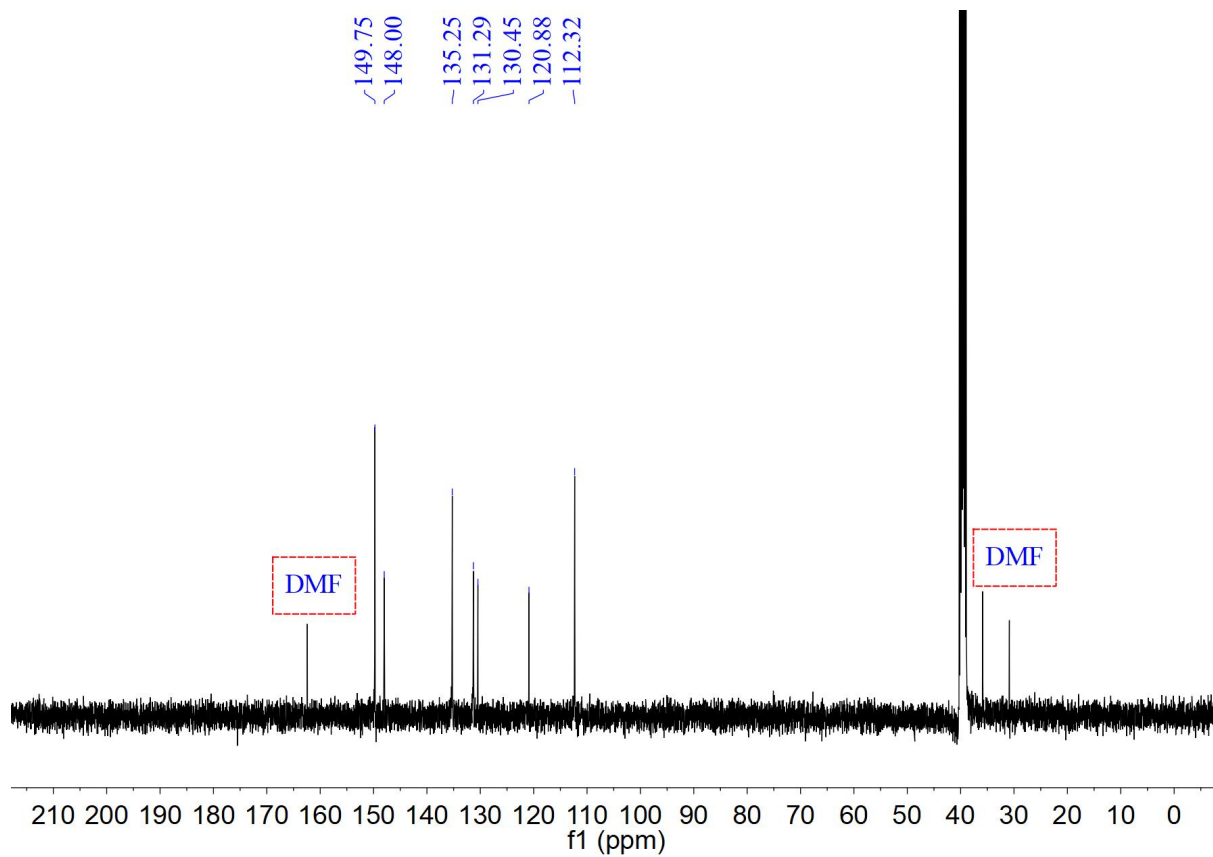
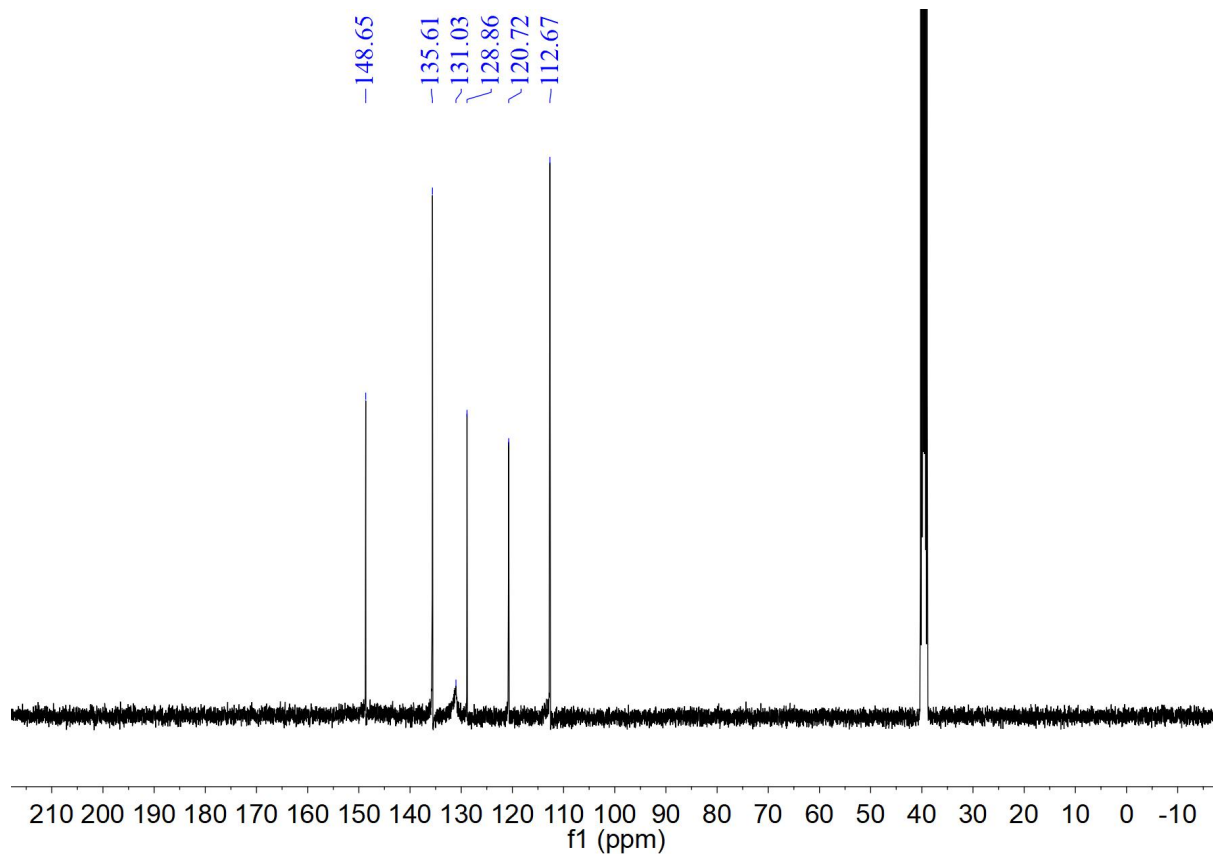
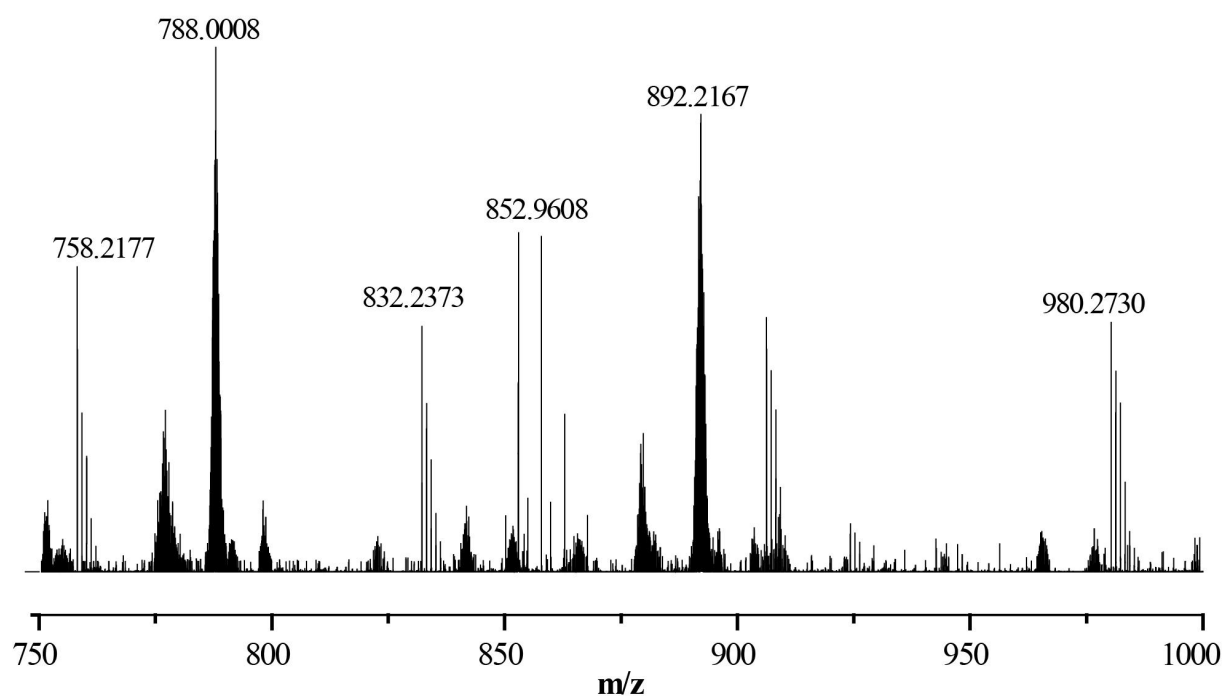
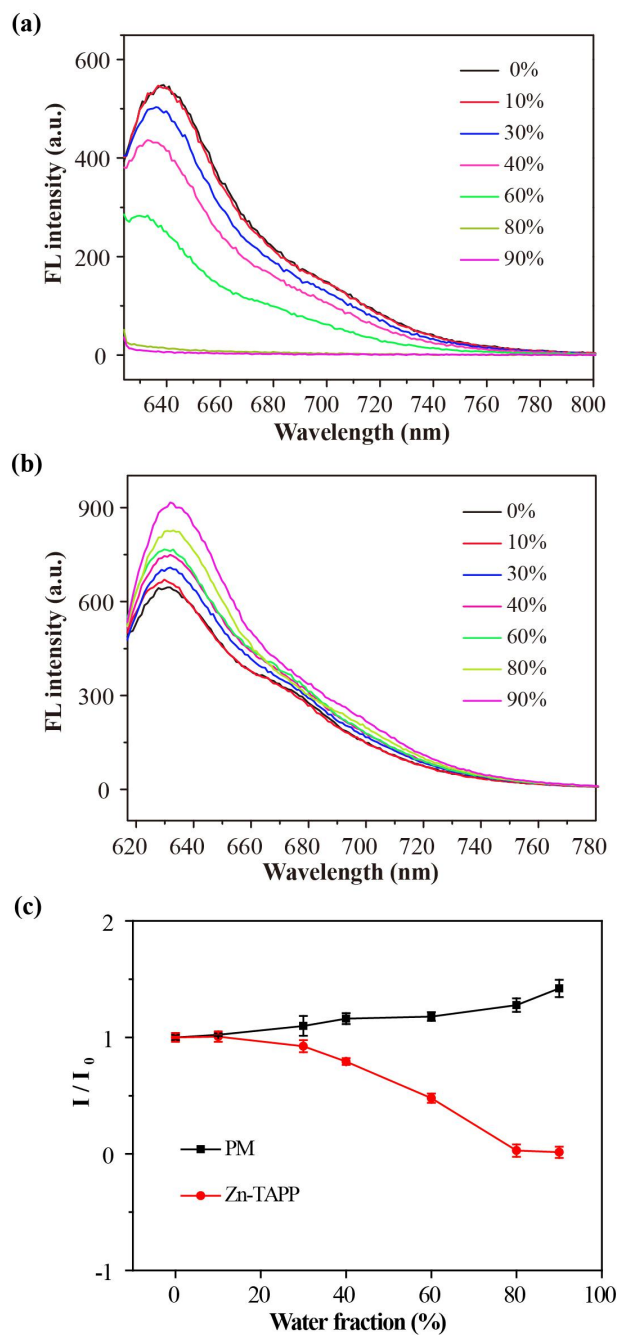


Fig. S3  $^{13}\text{C}$  NMR spectra of TAPP and Zn-TAPP in  $\text{DMSO-}d_6$ .

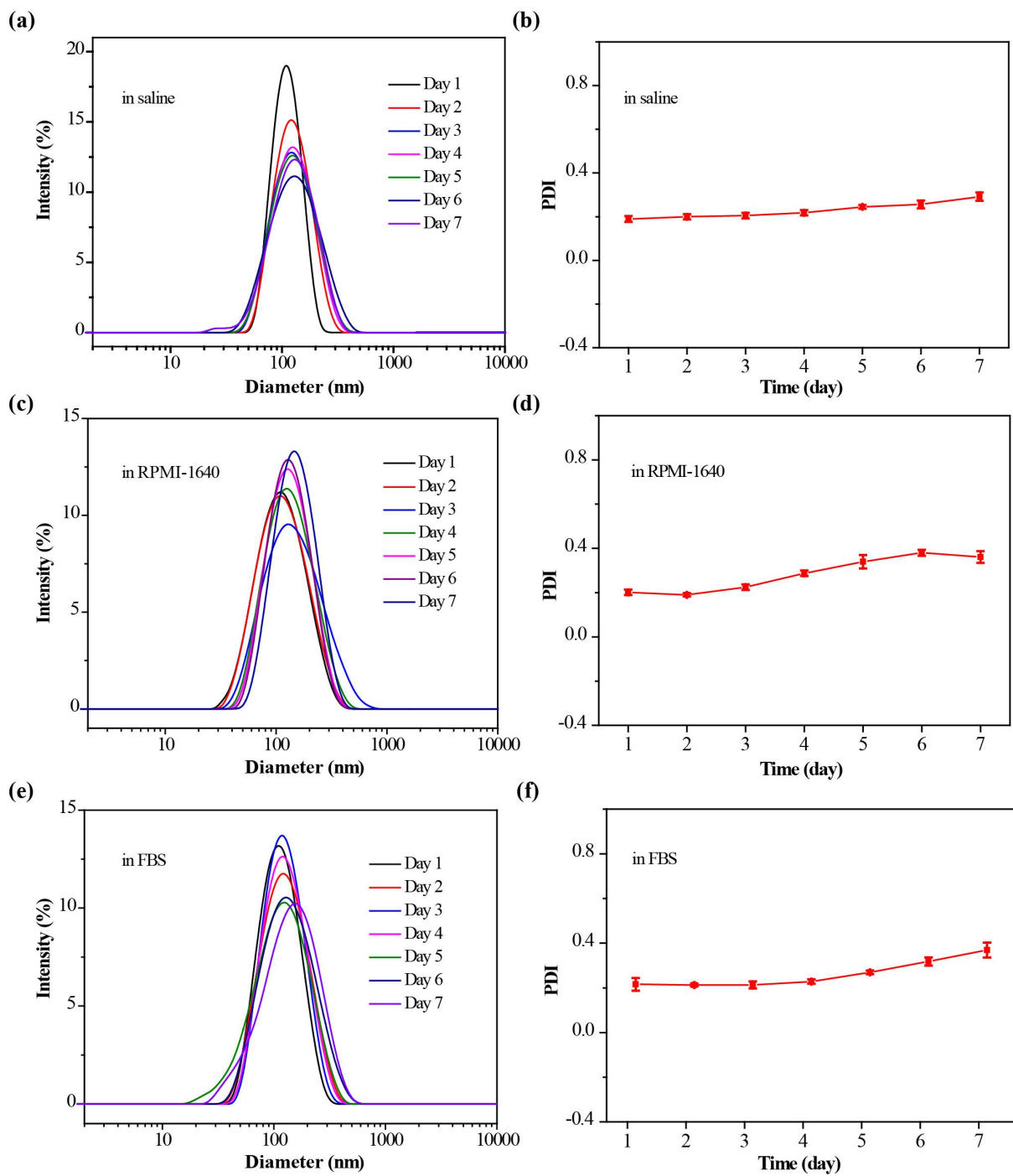


m/z	fragment
758.22	$[\text{PM}(\text{OTf})_5 \cdot 7\text{DMF}]^{11+}$
788.00	$[\text{PM}(\text{OTf})_4 \cdot 24\text{DMF} \cdot \text{H}_2\text{O}]^{12+}$
832.24	$[\text{PM}(\text{OTf})_6 \cdot 4\text{DMF} \cdot 2\text{H}_2\text{O}]^{10+}$
852.96	$[\text{PM}(\text{OTf})_6 \cdot 3\text{DMF} \cdot 21\text{H}_2\text{O}]^{10+}$
892.22	$[\text{PM}(\text{OTf})_6 \cdot 12\text{DMF} \cdot 3\text{H}_2\text{O}]^{10+}$
980.27	$[\text{PM}(\text{OTf})_7 \cdot \text{DMF}]^{9+}$

Fig. S4 Mass spectrum of PM.



**Fig. S5** The fluorescence spectra of Zn-TAPP (a) and PM (b) at the same concentration of Zn-TAPP (84  $\mu$ M) in DMF/H<sub>2</sub>O mixtures with water fractions range from 0% to 90%. (c) The fluorescence intensity ratios of Zn-TAPP and PM in DMF/H<sub>2</sub>O mixtures.  $I_0$  and  $I$  are intensities in pure DMF and DMF/H<sub>2</sub>O mixtures, respectively. Data are presented as means  $\pm$  S.D (n=3).



**Fig. S6** PM@HAase-mPEG particle size distribution and PDI changes in (a, b) saline, (c, d) RPMI-1640 and (e, f) FBS within 7 days. Data are presented as means  $\pm$  S.D (n=3).



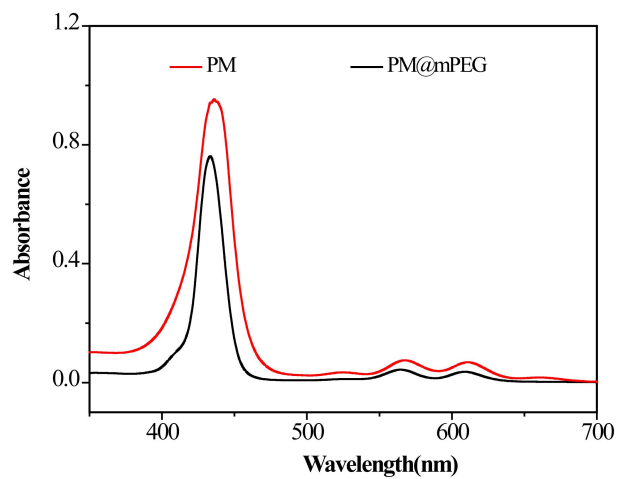


Fig. S7 UV-vis spectra of PM and PM@mPEG.

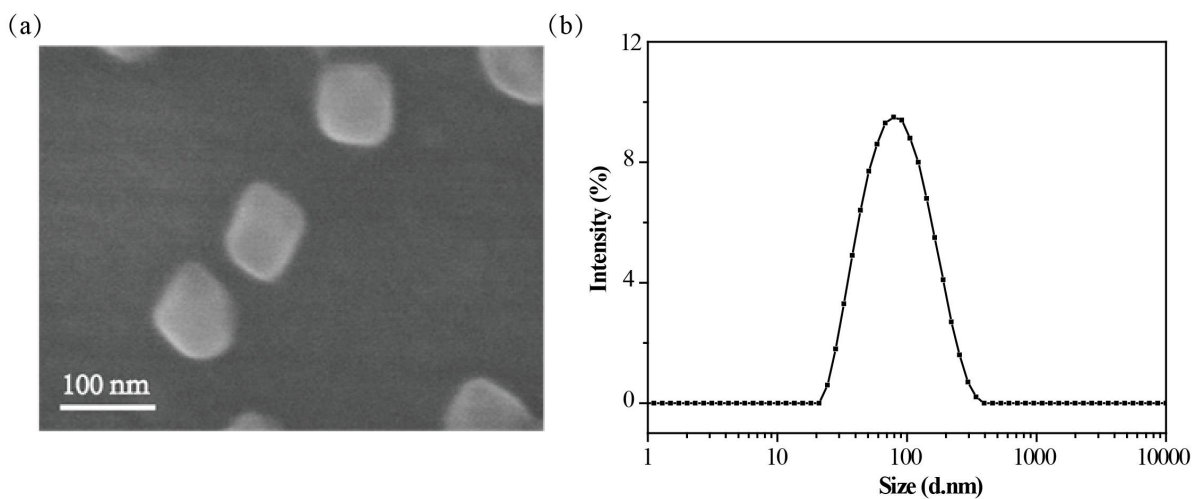


Fig. S8 (a) SEM image and (b) DLS distribution of PM@mPEG.

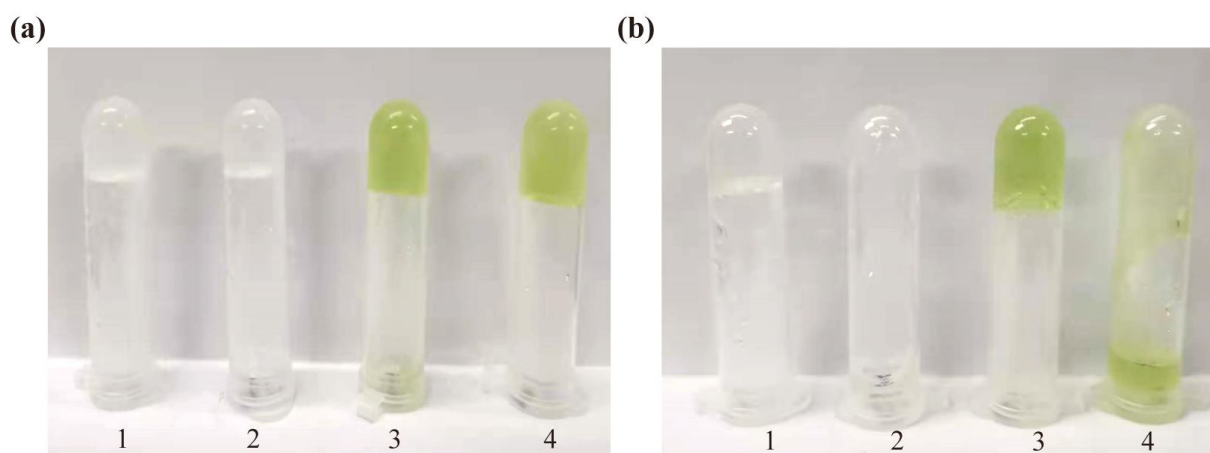
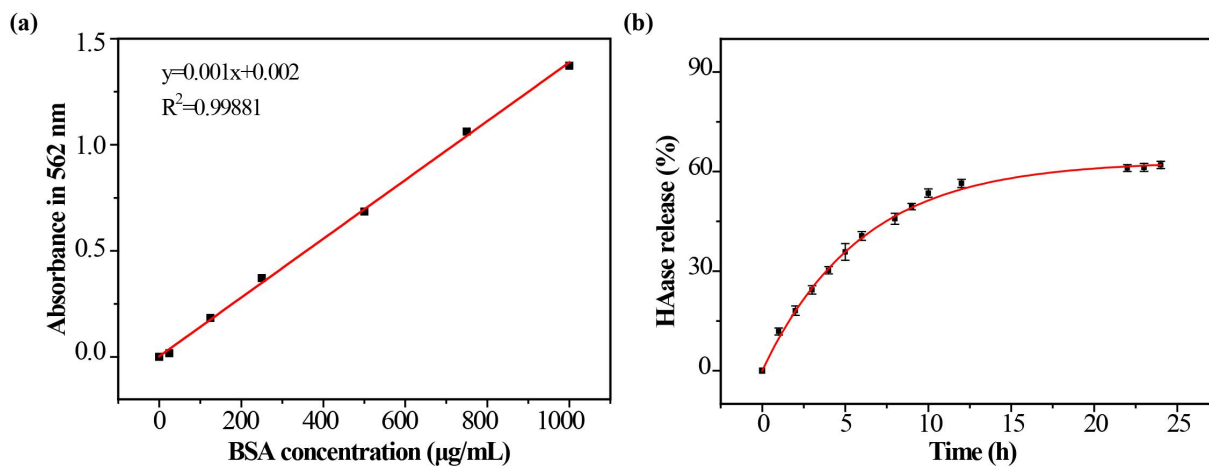
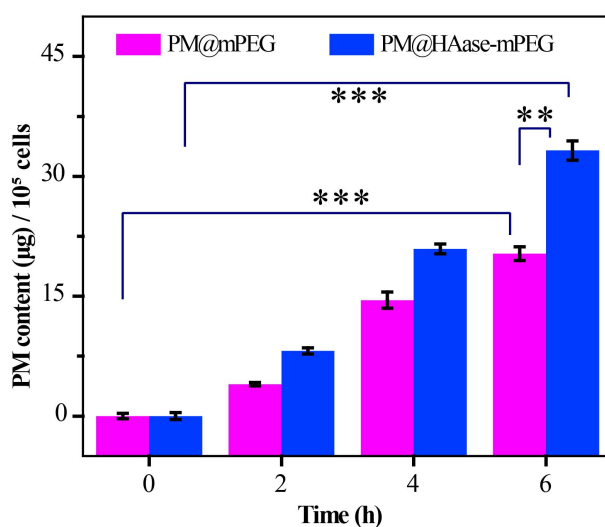


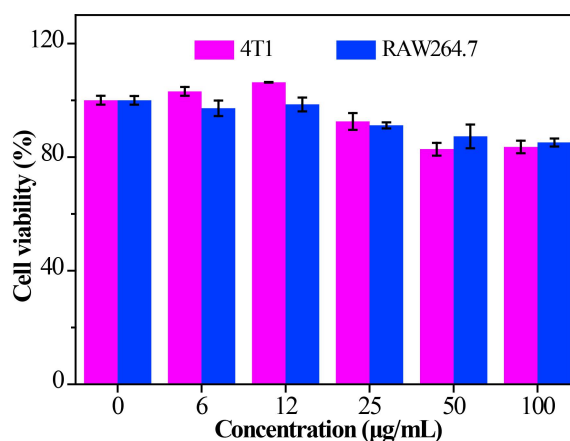
Fig. S9 Gelatin test experimental images (a) at the beginning of addition of materials and (b) 4 h later, respectively. (1) gelatin + HA; (2) gelatin + HA + HAase; (3) gelatin + HA + PM@mPEG; (4) gelatin + HA + PM@HAase-mPEG.



**Fig. S10** The release of HAase from PM@HAase-mPEG by BCA protein assay (a) standard curve and (b) HAase release profile at 37 °C for different time. Data are presented as means  $\pm$  S.D (n=3).



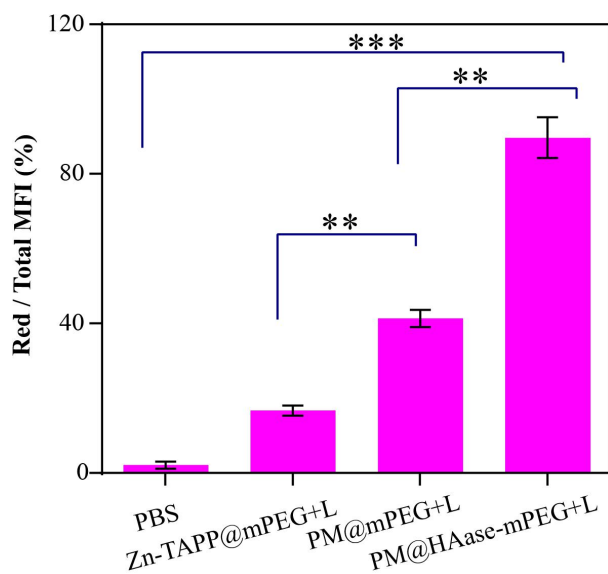
**Fig. S11** Cellular uptake of PM@mPEG and PM@HAase-mPEG after 2, 4 or 6 hours of incubation. The concentrations were determined by measuring Zn content inside cells by ICP. Data are presented as means  $\pm$  SD (n = 3). \*\*p < 0.01, \*\*\*p < 0.001.



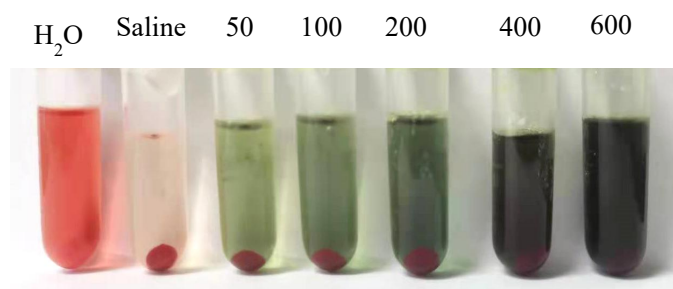
**Fig. S12** Cell viabilities of 4T1 and RAW 264.7 cells incubated with different concentration of PM@mPEG in dark for 24 h. Data are presented as means  $\pm$  SD (n=3).

	$C_p$	PDT efficiency
PM@PEG	42 $\mu$ M	55%
PM@PEG	84 $\mu$ M	67%
Zn-TAPP	84 $\mu$ M	31%
PCN-224	68 $\mu$ M	22%
Hf-TCPP NMOF	51 $\mu$ M	50%
NPMOF	75 $\mu$ M	25%

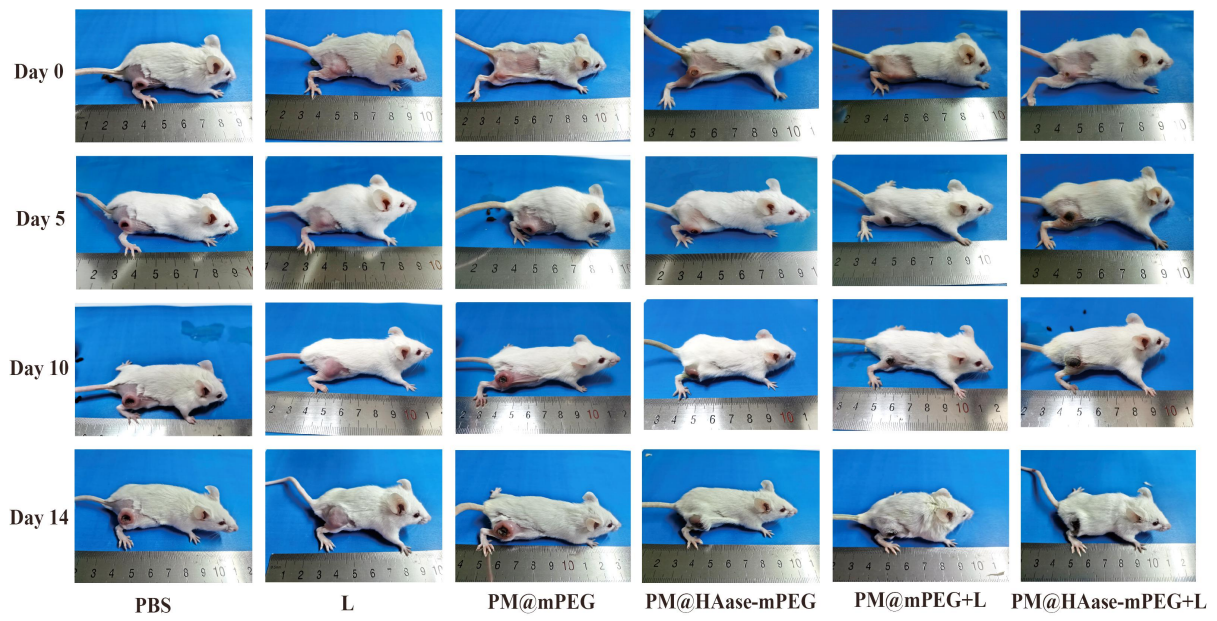
**Table S1** PDT efficiencies for TAPP ligand and MOFs reported in the literature.  $C_p$  represents the porphyrin concentration.



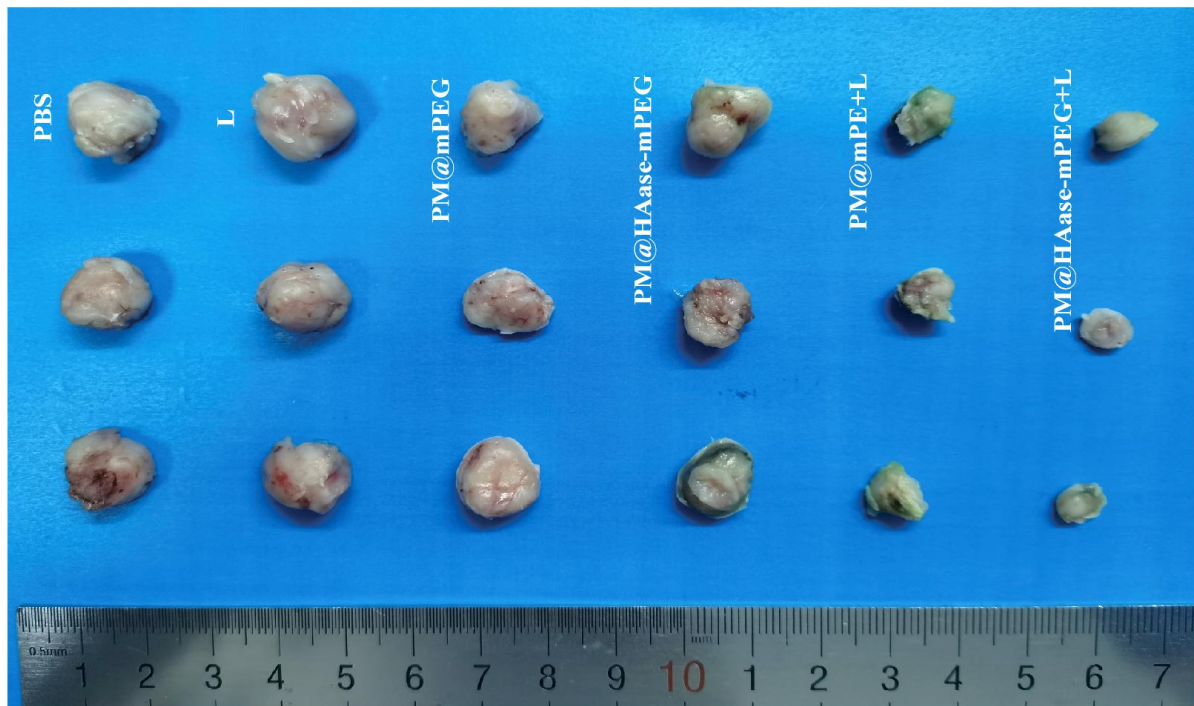
**Fig. S13** Quantitative analysis of red fluorescence intensity ratio in calcein-AM /PI staining of each group. Data are presented as means  $\pm$  SD (n = 3). \*\*\*p < 0.001.



**Fig. S14** Red blood cells were incubated with various concentrations of PM@HAase-mPEG for 2 h. PBS and pure water were employed as the negative and positive control, respectively.



**Fig. S15** Photographs of tumor-bearing mice on days 0, 5, 10 and 14 days of during follow-up treatment.



**Fig. S16** Photographs of the tumors after treatment for 14 days.

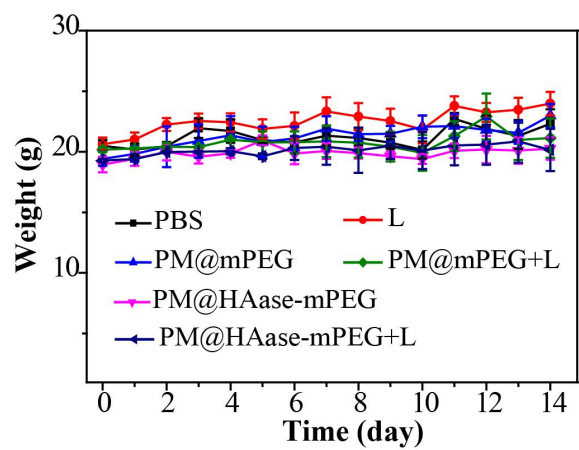


Fig. S17 Body weight of 4T1 tumor-bearing mice during various treatment for 14 days. Data are presented as means  $\pm$  S.D (n=3).

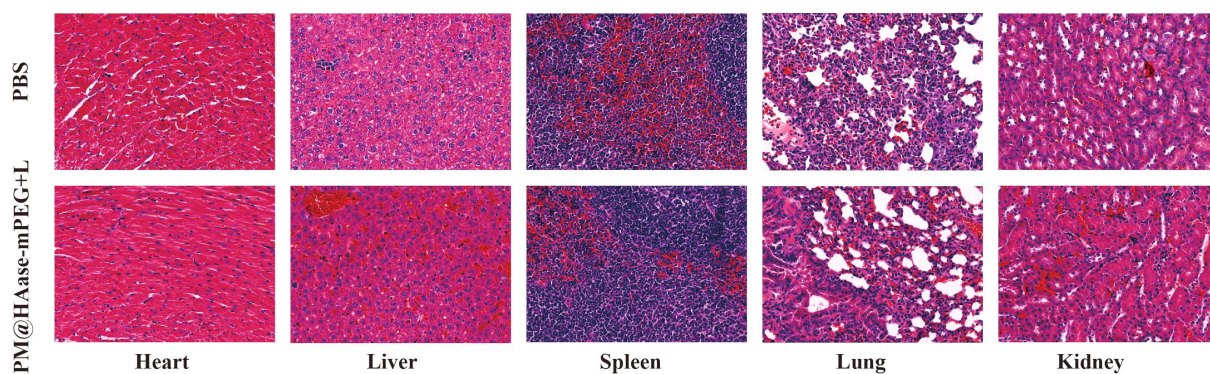


Fig. S18 H&E staining of mice heart, liver, spleen, lung and kidney after treatment by PM@HAase-mPEG with irradiation.