

Electronic Supplementary Information

Value-added chemicals from biomass-derived furans: radical functionalizations of 5-chloromethylfurfural (CMF) by metal-free ATRA reactions

Rajesh Dasi,^a Daniel Schmidhuber,^b Lisa Marie Gronbach,^a
Julia Rehbein^{*b} and Malte Brasholz^{*a}

a) Institute of Chemistry – Organic Chemistry, University of Rostock, Albert-Einstein-Str. 3A,
18059 Rostock, Germany

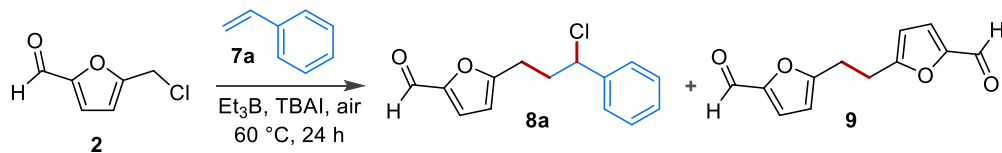
e-mail: malte.brasholz@uni-rostock.de

b) Institute of Organic Chemistry, University of Regensburg, Universitätsstr. 31,
93053 Regensburg, Germany

e-mail: julia.rehbein@chemie.uni-regensburg.de

1. Reaction optimization

Table ESI-1. Reaction optimization.



#	Et_3B (eq.)	TBAI (eq.)	7a (eq.)	solvent	conv. 2 [%] ^a	yield 8a [%] ^a	yield 9 [%] ^a
1	2.0	0	15	MeCN	29	0	0
2	2.0	1	15	MeCN	87	23	10
3 ^b	2.0	1	15	MeCN	0	0	0
4	2.0	0.5	8	MeCN	100	16	2
5	2.0	1	8	MeCN	100	30	2
6	2.0	2	8	MeCN	100	31	3
7	2.0	4	8	MeCN	100	40	3
8	2.0	6	8	MeCN	100	38	4
9	2.0	4	4	MeCN	95	30	2
10	2.0	4	10	MeCN	100	42	3
11	2.0	4	15	MeCN	82	42	6
12	2.0	4	20	MeCN	100	30	2
13	4.0	4	10	MeCN	100	33	5
14	1.0	4	10	MeCN	98	51	5
15	0.5	4	10	MeCN	96	74	3
16	0.25	4	10	MeCN	91	38	3
17	0.5	4	10	acetone	97	75 (78)^c	3
18	0.5	4	10	THF	74	50	2
19	0.5	4	10	DMF	63	28	2
20	0.5	4	10	DMSO	77	35	1

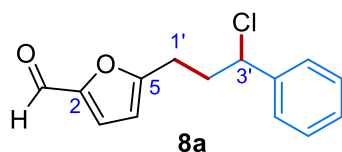
Reactions were performed on 0.30 mmol scale of **1**. [a] Determined by ^1H NMR spectroscopy against CH_2Br_2 as internal standard. [b] Reaction in the absence of O_2 . [c] Yield of isolated product after chromatography.

2. Synthetic experiments

General Experimental Procedure:

5-Chloromethylfurfural **2** (0.30 mmol, 1.0 equiv.) and tetrabutylammonium iodide (TBAI, 1.20 mmol, 4.0 equiv.) were combined in a 10 mL crimp cap vial and the vial was sealed. Dry acetone (2.00 mL) and styrene derivative **7** (or **10** or **12**, respectively; 3.00 mmol, 10 equiv., freshly distilled) were added and the mixture was heated to 60 °C. After 30 min, triethylborane (Et₃B, 1.0 M solution in hexanes, 0.15 mmol, 0.5 equiv.) was added. Then, the septum was pierced with a cannula to allow the capture of oxygen from the ambient air (the cannula remained in the septum membrane throughout the reaction). After 24 h of reaction time, the solvent was evaporated. The crude mixture was diluted with DCM and washed with water and brine (3 × 20 mL). The organic layer was collected and dried over anhydrous MgSO₄ and filtered, followed by concentration under reduced pressure. The crude mixture was subjected to column chromatography (silica gel, 230 – 400 mesh) to afford the desired compounds **8** (or **11** or **13**, respectively).

5-(3-Chloro-3-phenylpropyl)furan-2-carbaldehyde (**8a**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), styrene **7a** (345 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8a** (58 mg, 78%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.39$ (EtOAc/heptane 1:3)

¹H NMR (300 MHz, CDCl₃) $\delta = 2.37$ – 2.57 (m, 2 H, CH₂, C-1'), 2.79 – 3.00 (m, 2 H, CH₂, C-2'), 4.86 (dd, $J = 6.1, 8.3$ Hz, 1 H, C-3'), 6.27 (d, $J = 3.6$ Hz, 1 H, 4-H), 7.16 (d, $J = 3.5$ Hz, 1 H, 3-H), 7.27 – 7.40 (m, 5 H, Ar-H), 9.52 (s, 1 H, CHO).

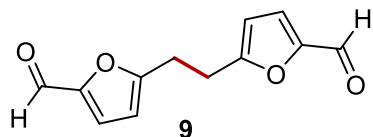
¹³C NMR (75 MHz, CDCl₃) $\delta = 26.2$ (t, C-1'), 37.7 (t, C-2'), 62.5 (d, C-3'), 109.5 (d, C-4), 123.3 (d, C-3), 127.0 (d, Ar), 128.7 (d, Ar), 128.9 (d, Ar), 140.9 (s, Ar), 152.2 (s, C-2), 161.7 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu} = 3030, 2920, 2815$ (C-H), 1670 (CO), $1515, 1395, 1195$ (C-Cl), $1020, 955, 800, 695$ cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₄H₁₃ClO₂⁺ [M]⁺: 248.0599, found: 248.0596.

From the reaction optimization experiments, a sample of furan dimer **9** could be obtained, in ~90% purity after chromatography.

5,5'-(ethane-1,2-diyl)bis(furan-2-carbaldehyde) (**9**)



Orange solid, **m.p.** 113 °C – 116 °C.

R_f = 0.20 (EtOAc /heptane 1:2)

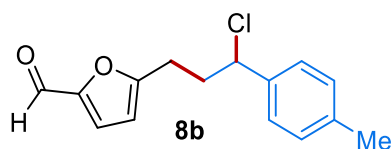
¹H NMR (500 MHz, CDCl₃) δ = 3.14 (s, 4 H, 2 × CH₂, C-1'), 6.24 (d, *J* = 3.5 Hz, 2 H, 4-H), 7.14 (d, *J* = 3.5 Hz, 2 H, 3-H), 9.52 (s, 2 H, CHO).

¹³C NMR (126 MHz, CDCl₃) δ = 26.6 (t, C-1'), 109.6 (d, C-4), 123.3 (d, C-3), 152.2 (s, C-2), 160.9 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 1676, 1521, 1222, 1024, 786, 775, 762, 757, 730, 667 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₂H₁₁O₄ [M+H]⁺: 219.0657, found: 219.0658.

5-(3-Chloro-3-(p-tolyl)propyl)furan-2-carbaldehyde



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-methylstyrene **7b** (400 μL, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μL, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8b** (52.7 mg, 67%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, **R_f** = 0.39 (EtOAc /heptane 1:3)

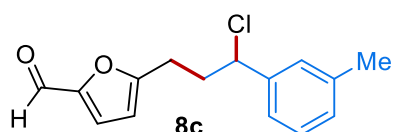
¹H NMR (300 MHz, CDCl₃) δ = 2.35 (s, 3 H, Ar-CH₃), 2.38–2.56 (m, 2 H, CH₂, C-1'), 2.79–2.99 (m, 2 H, CH₂, C-2'), 4.83 (dd, *J* = 6.1, 8.4 Hz, 1 H, CH, C-3'), 6.27 (d, *J* = 3.5 Hz, 1 H, 4-H), 7.14–7.16 (m, 1 H, 3-H), 7.18 (dt, *J* = 1.8 Hz, 0.6, 2 H, Ar-H), 7.24–7.29 (m, 2 H, Ar-H), 9.53 (s, 1 H, CHO).

¹³C NMR (126 MHz, CDCl₃) δ = 21.3 (q, Ar-CH₃), 26.3 (t, C-1'), 37.7 (t, C-2'), 62.6 (d, C-3'), 109.5 (d, C-4), 126.3 (d, C-3), 126.9 (d, Ar), 129.6 (d, Ar), 138.0 (s, Ar), 138.6 (s, Ar), 152.3 (s, C-2), 161.9 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3025, 2955, 2920 (C-H), 1715, 1675 (CO), 1515, 1395, 1205 (C-Cl), 1020, 955, 800, 755 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₅H₁₅ClO₂⁺ [M]⁺: 262.0755, found: 262.0762.

5-(3-Chloro-3-(*m*-tolyl)propyl)furan-2-carbaldehyde (**8c**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 3-methylstyrene **7c** (400 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8c** (66 mg, 84%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, *R_f* = 0.39 (EtOAc /heptane 1:3)

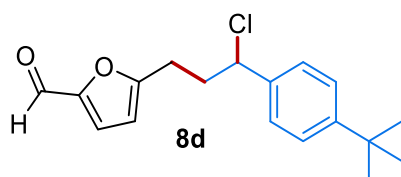
¹H NMR (500 MHz, CDCl₃) δ = 2.36 (s, 3 H, Ar-CH₃), 2.39–2.53 (m, 2 H, C-1'), 2.82–2.97 (m, 2 H, C-2'), 4.82 (dd, *J* = 5.8, 8.6 Hz, 1 H, CH, C-3'), 6.28 (d, *J* = 3.5 Hz, 1 H, 4-H), 7.13 (d, *J* = 7.5 Hz, 1 H, Ar-H), 7.16–7.21 (m, 3 H, 3-H, Ar-H), 7.22–7.29 (m, 1 H, Ar-H), 9.53 (s, 1 H, CHO).

¹³C NMR (126 MHz, CDCl₃) δ = 21.5(q, Ar-CH₃), 26.2 (t, C-1'), 37.7 (t, C-2'), 62.7 (d, C-3'), 109.5 (d, C-4), 123.4 (d, C-3), 124.1 (d, Ar), 127.7 (d, Ar), 128.8 (d, Ar), 129.5 (d, Ar), 138.7 (s, Ar), 140.9 (s, Ar), 152.2 (s, C-2), 161.8 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3115, 2930, 2830 (C-H), 1670 (CO), 1515, 1260, 1150 (C-Cl), 1020, 750, 690 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₅H₁₅ClO₂⁺ [M]⁺: 262.0755, found: 262.0762.

5-(3-(4-(*Tert*-butyl)phenyl)-3-chloropropyl)furan-2-carbaldehyde (**8d**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-tert-butyl styrene **7d** (550 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8d** (63 mg, 69%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.42$ (EtOAc/heptane 1:3)

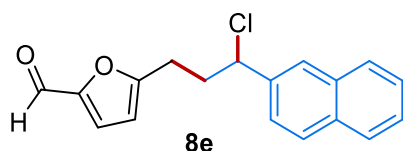
¹H NMR (300 MHz, CDCl₃) $\delta = 1.31$ (s, 9 H, CH₃), 2.37–2.57 (m, 2 H, CH₂, C-1'), 2.79–3.03 (m, 2 H, CH₂, C-2'), 4.86 (dd, $J = 6.1, 8.3$ Hz, 1 H, CH, C-3'), 6.27 (d, $J = 3.4$ Hz, 1 H, 4-H), 7.16 (d, $J = 3.5$ Hz, 1 H, 3-H), 7.28–7.40 (m, 4 H, Ar-H), 9.53 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) $\delta = 26.2$ (t, C-1'), 31.4 (q, 3 \times CH₃), 34.7 (s, *t*-Bu), 37.6 (t, C-2'), 62.5 (d, C-3'), 109.4 (d, C-4), 123.4 (d, C-3), 125.8 (d, Ar), 126.7 (d, Ar), 137.9 (s, Ar), 151.8 (s, Ar), 152.2 (s, C-2), 161.9 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu} = 2960, 2860, 2810$ (C-H), 1675 (CO), 1510, 1395, 1365, 1200 (C-Cl), 1020, 950, 800, 750 cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₈H₂₁ClO₂⁺ [M]⁺: 304. 1225, found: 304. 1225.

5-(3-Chloro-3-(naphthalen-2-yl)propyl)furan-2-carbaldehyde (**8e**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 2-Vinylnaphthalene **7e** (463 mg, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8e** (33.8 mg, 40%) after chromatography (silica, EtOAc/heptane 1:6).

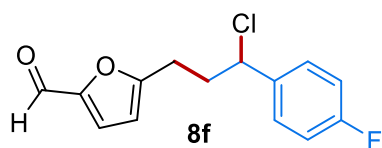
Pale yellow oil, $R_f = 0.37$ (EtOAc/heptane 1:3)

¹H NMR (300 MHz, CDCl₃) $\delta = 2.47$ –2.67 (m, 2 H, CH₂, C-1'), 2.83–3.03 (m, 2 H, CH₂, C-2'), 5.04 (dd, $J = 6.2, 8.3$ Hz, 1 H, CH, C-3'), 6.27 (d, $J = 3.5$ Hz, 1 H, 4-H), 7.16 (d, $J = 3.5$ Hz, 1 H, 3-H), 7.48–7.56 (m, 3 H, Ar-H), 7.78–7.90 (m, 4 H, Ar-H), 9.53 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) $\delta = 26.3$ (t, C-1'), 37.6 (t, C-2'), 62.8 (d, C-3'), 109.5 (d, C-4), 123.3 (d, C-3), 124.5 (d, Ar), 126.2 (d, Ar), 126.7 (d, Ar), 127.8 (d, Ar), 128.2 (d, Ar), 129.0 (d, Ar), 133.1 (s, Ar), 133.4 (s, Ar), 138.1 (s, Ar), 152.3 (s, C-2), 161.7 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu} = 3115, 3055, 2920, 2820$ (C-H), 1670 (CO), 1510, 1400, 1240 (C-Cl), 1020, 955, 750, 480 cm⁻¹.

5-(3-Chloro-3-(4-fluorophenyl)propyl)furan-2-carbaldehyde (**8f**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-fluorostyrene **7f** (330 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8f** (44 mg, 55%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.36$ (EtOAc/heptane 1:3)

¹H NMR (300 MHz, CDCl₃) $\delta = 2.35$ – 2.54 (m, 2 H, CH₂, C-1'), 2.80 – 2.99 (m, 2 H, CH₂, C-2'), 4.84 (dd, $J = 6.0, 8.4$ Hz, 1 H, CH, C-3'), 6.28 (d, $J = 3.5$ Hz, 1 H, 4-H), 6.99 – 7.08 (m, 2 H, Ar-H), 7.17 (d, $J = 3.6$, 1 H, 3-H), 7.30 – 7.40 (m, 2 H, Ar-H), 9.53 (s, 1 H, CHO).

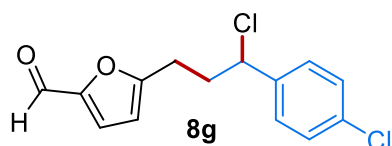
¹³C NMR (75 MHz, CDCl₃) $\delta = 26.2$ (t, C-1'), 37.8 (t, C-2'), 61.7 (d, C-3'), 109.5 (d, C-4), 115.7 (d, Ar), 115.9 (d, $J_{C,F} = 22.5$ Hz, Ar), 123.4 (d, C-3), 128.9 (d, $J_{C,F} = 7.5$ Hz, Ar), 136.9 (s, Ar), 152.3 (s, C-2), 161.5 (s, C-5), 162.4 (d, $J_{C,F} = 285.0$ Hz, Ar), 177.1 (d, CHO).

¹⁹F NMR (282 MHz, CDCl₃) $\delta = -112.99$ (s, 1 F, Ar-F).

IR: $\tilde{\nu} = 3115, 2915, 2820$ (C-H), 1670 (CO), $1600, 1395, 1220$ (C-Cl), $1160, 1020, 950, 760, 525$ cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₄H₁₂ClFO₂⁺ [M]⁺: 266.0504, found: 266.0503.

5-(3-Chloro-3-(4-chlorophenyl)propyl)furan-2-carbaldehyde (**8g**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-chlorostyrene **7g** (360 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8g** (55 mg, 65%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.36$ (EtOAc/heptane 1:3)

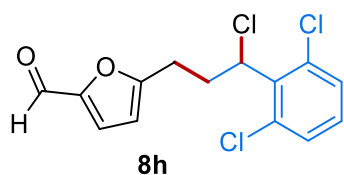
¹H NMR (300 MHz, CDCl₃) δ = 2.34–2.53 (m, 2, CH₂, C-1'), 2.79 – 3.01 (m, 2 H, CH₂, C-2'), 4.82 (dd, *J* = 6.1, 8.4 Hz, 1 H, CH, C-3'), 6.28 (d, *J* = 3.6 Hz, 1 H, 4-H), 7.17 (d, *J* = 3.6 Hz, 1 H, 3-H), 7.28–7.37 (m, 4 H, Ar-H), 9.53 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 26.1 (t, C-1'), 37.7 (t, C-2'), 61.6 (d, C-3'), 109.6 (d, C-4), 123.4 (d, C-3), 128.4 (d, Ar), 129.1 (d, Ar), 134.5 (s, Ar), 139.5 (s, Ar), 152.3 (s, C-2), 161.4 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3115, 2960, 2820 (C-H), 1670 (CO), 1515, 1395, 1190 (C-Cl), 1090, 1010, 950, 800, 620 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₄H₁₂Cl₂O₂⁺ [M]⁺: 282.0209, found: 282.0207.

5-(3-Chloro-3-(2,6-dichlorophenyl)propyl)furan-2-carbaldehyde (8h)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 2,6-dichlorostyrene **7h** (410 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8h** (33.2 mg, 35%) after chromatography (silica, EtOAc/heptane 1:4).

Pale yellow oil, *R_f* = 0.38 (EtOAc /heptane 1:3)

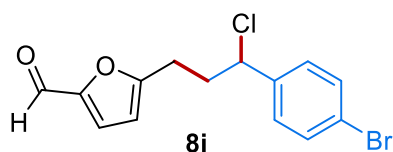
¹H NMR (300 MHz, CDCl₃) δ = 2.58–3.06 (m, 4 H, CH₂, C-1', C-2'), 5.71–5.80 (m, 1 H, CH, C-3'), 6.30 (d, *J* = 3.5 Hz, 1 H, 4-H), 7.12–7.21 (m, 2 H, 3-H, Ar-H), 7.25 – 7.36 (m, 2 H, Ar-H), 9.52 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 26.3 (t, C-1'), 33.2 (t, C-2'), 56.7 (d, C-3'), 109.6 (d, C-4), 123.2 (d, C-3), 128.7 (d, Ar), 130.0 (d, Ar), 131.0 (d, Ar), 134.6 (s, Ar), 135.0 (s, Ar), 136.0 (s, Ar), 152.3 (s, C-2), 161.3 (s, C-5), 177.2 (d, CHO).

IR: $\tilde{\nu}$ = 3115, 2925, 2815 (C-H), 1670 (CO), 1510, 1435, 1185 (C-Cl), 1020, 950, 760, 710 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₄H₁₁Cl₃O₂⁺ [M]⁺: 315.9819, found: 315.9815.

5-(3-(4-Bromophenyl)-3-chloropropyl)furan-2-carbaldehyde (**8i**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-bromostyrene **7i** (392 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8i** (69.4 mg, 71%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow colour oil, R_f = 0.35 (EtOAc /heptane 1:3)

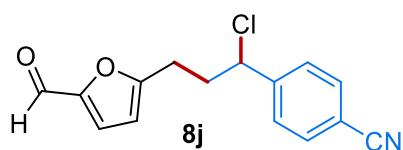
¹H NMR (300 MHz, CDCl₃) δ = 2.33–2.54 (m, 2 H, CH₂, C-1'), 2.81–3.00 (m, 2 H, CH₂, C-2'), 4.82 (dd, J = 6.1, 8.3 Hz, 1 H, CH, C-3'), 6.29 (d, J = 3.6 Hz, 1 H, 4-H), 7.18 (d, J = 3.6 Hz, 1 H, 3-H), 7.23–7.31 (m, 2 H, Ar-H), 7.47–7.54 (m, 2 H, Ar-H), 9.55 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 26.1 (t, C-1'), 37.7 (t, C-2'), 61.6 (d, C-3'), 109.6 (d, C-4), 122.6 (s, Ar), 123.4 (d, C-3), 128.7 (d, Ar), 132.1 (d, Ar), 140.0 (s, Ar), 152.3 (s, C-2), 161.4 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3120, 2925, 2815 (C-H), 1670 (CO), 1510, 1490, 1200 (C-Cl), 1070, 1010, 750, 620 cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₄H₁₂BrClO₂⁺ [M]⁺: 325.9704, found: 325.9701.

4-(1-Chloro-3-(5-formylfuran-2-yl)propyl)benzonitrile (**8j**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-cyanostyrene **7j** (360 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8j** (48.3 mg, 59%) after chromatography (silica, EtOAc/heptane 1:4).

Pale yellow oil, R_f = 0.30 (EtOAc /heptane 1:3)

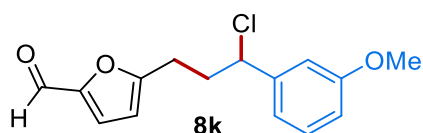
¹H NMR (500 MHz, CDCl₃) δ = 2.37–2.48 (m, 2 H, CH₂, C-1'), 2.84–2.98 (m, 2 H, CH₂, C-2'), 4.86 (dd, *J* = 5.9, 8.4 Hz, 1 H, CH, C-3'), 6.28 (d, *J* = 3.5 Hz, 1 H, 4-H), 7.16 (d, *J* = 3.5 Hz, 1 H, 3-H), 7.47–7.53 (m, 2 H, Ar-H), 7.62–7.68 (m, 2 H, Ar-H), 9.51 (s, 1 H, CHO).

¹³C NMR (126 MHz, CDCl₃) δ = 25.9 (t, C-1'), 37.5 (t, C-2'), 61.1 (d, C-3'), 109.7 (d, C-4), 112.5 (s, Ar), 118.4 (s, CN), 123.5 (d, C-3), 127.8 (d, Ar), 132.7 (d, Ar), 146.0 (s, Ar), 152.3 (s, C-2), 161.0 (s, C-5), 177.0 (d, CHO).

IR: $\tilde{\nu}$ = 3115, 2930, 2820 (C-H), 2230 (CN), 1670 (CO), 1515, 1190 (C-Cl), 1020, 950, 750, 560 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₅H₁₂ClNaNO₂⁺ [*M*+Na]⁺: 296.0449, found: 296.0449.

5-(3-Chloro-3-(3-methoxyphenyl)propyl)furan-2-carbaldehyde (**8k**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 3-methoxystyrene **7k** (416 μL, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μL, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8k** (60.8 mg, 73%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, *R_f* = 0.35 (EtOAc /heptane 1:3)

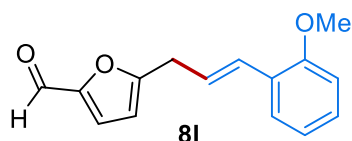
¹H NMR (300 MHz, CDCl₃) δ = 2.35–2.55 (m, 2 H, CH₂, C-1'), 2.78–2.99 (m, 2 H, CH₂, C-2'), 3.81 (s, 3 H, CH₃, OMe), 4.82 (dd, *J* = 6.2, 8.2 Hz, 1 H, CH, C-3'), 6.27 (d, *J* = 3.6 Hz, 1 H, 4-H), 6.82–6.87 (m, 1 H, Ar-H), 6.90–6.97 (m, 2 H, Ar-H), 7.16 (d, *J* = 3.5 Hz, 1 H, 3-H), 7.22–7.31 (m, 1 H, Ar-H), 9.52 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 26.2 (t, C-1'), 37.7 (t, C-2'), 55.4 (q, OMe), 62.4 (d, C-3'), 109.5 (d, C-4), 112.7 (d, Ar), 114.1 (d, Ar), 119.2 (d, Ar), 123.3 (d, C-3), 129.9 (d, Ar), 142.4 (s, Ar), 152.2 (s, C-2), 159.9 (s, C-5), 161.7 (s, Ar), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3115, 2940, 2835 (C-H), 1670 (CO), 1515, 1260, 1150 (C-Cl), 1020, 750, 690 cm⁻¹.

HRMS (ESI+): *m/z* calc.: C₁₅H₁₅ClO₃⁺ [*M*]⁺: 278.0704, found: 278.0703.

(E)-5-(3-(2-Methoxyphenyl)allyl)furan-2-carbaldehyde (**8l**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 2-methoxystyrene **7l** (416 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **8l** (30 mg, 41%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.36$ (EtOAc /heptane 1:3)

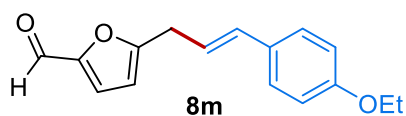
¹H NMR (300 MHz, CDCl₃) δ = 3.67 (d, $J = 7.1$ Hz, 2 H, CH₂, C-1'), 3.85 (s, 3 H, CH₃, OMe), 6.31 (td, $J = 7.1, 15.9$ Hz, 1 H, 2'-H), 6.32 (d, $J = 3.5$ Hz, 1 H, 4-H), 6.84–6.96 (m, 3 H, Ar-H, 3'-H), 7.17–7.24 (m, 2 H, 3-H, Ar-H), 7.42 (dd, $J = 1.7, 7.6$ Hz, 1 H, Ar-H), 9.55 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 32.7 (t- C-1'), 55.6 (q, CH₃, OCH₃), 109.4 (d, C-4), 111.0 (d, Ar), 120.8 (d, Ar), 123.4 (d, C-3), 123.9 (d, C-2'), 125.9 (s, Ar), 127.0 (d, Ar), 128.6 (d, C-3'), 128.9 (d, Ar), 152.2 (s, C-2), 156.7 (s, C-5), 162.0 (s, Ar), 177.3 (d, CHO).

IR: $\tilde{\nu} = 2935, 2835$ (C-H), 1670 (CO), 1510, 1490, 1395, 1240 (C-Cl), 965, 750 cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₅H₁₄NaO₃⁺ [M+Na]⁺: 265.0835, found: 265.0839.

(E)-5-(3-(4-ethoxyphenyl)allyl)furan-2-carbaldehyde (**8m**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-ethoxystyrene **7m** (450 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in Acetone (2.0 mL) gave, after 24 h reaction time, compound **8l** (23 mg, 30%) after chromatography (silica, EtOAc/heptane 1:6).

Pale yellow oil, $R_f = 0.36$ (EtOAc /heptane 1:3).

¹H NMR (300 MHz, CDCl₃) δ = 1.41 (t, $J = 7.0$ Hz, 3 H, CH₃), 3.62 (d, $J = 6.9$ Hz, 2 H, C-1'), 4.03 (q, $J = 7.0$ Hz, 2 H, OCH₂), 6.14 (td, $J = 6.9, 15.7$ Hz, 1 H, C-2'), 6.30 (d, $J = 3.6$ Hz, 1 H,

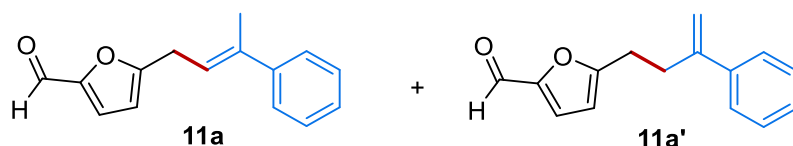
4-H), 6.49 (d, $J = 15.7$ Hz, 1 H, C-3'), 6.81–6.88 (m, 2 H, Ar-H), 7.19 (d, $J = 3.5$ Hz, 1 H, 3-H), 7.25–7.32 (m, 2 H, Ar-H), 9.55 (s, 1 H, CHO).

^{13}C NMR (75 MHz, CDCl_3) $\delta = 14.9$ (q, CH_3), 32.2 (t- C-1'), 63.6 (t- CH_2), 109.3 (d- C-4), 114.7 (d, Ar), 120.9 (d, C-2'), 123.4 (d, C-3), 127.6 (s, Ar), 129.5 (d, Ar), 133.1 (d, C-3'), 152.3 (s, C-2), 158.8 (s, C-5), 161.9 (s, Ar), 177.3 (d, CHO).

IR: $\tilde{\nu} = 3110, 2980, 2830$ (C-H), 1670 (CO), 1510, 1390, 1230 (C-Cl), 1175, 975, 755 cm^{-1} .

HRMS (ESI+): m/z calc.: $\text{C}_{16}\text{H}_{16}\text{NaO}_3^+$ $[\text{M}+\text{Na}]^+$: 279.0992, found: 279.0994.

(*E*)-5-(3-Phenylbut-2-en-1-yl)furan-2-carbaldehyde (**11a**) and 5-(3-Phenylbut-3-en-1-yl)furan-2-carbaldehyde (**11a'**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), α -methylstyrene **10a** (400 μL , 3.00 mmol, 10 equiv), Et_3B (1.0 M solution in hexanes, 150 μL , 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compounds **11a**, **11a'** (16 mg, 25%, inseparable mixture, ratio 1:1) after chromatography (silica, EtOAc/heptane 1:7).

Brown oil, $R_f = 0.40$ (EtOAc /heptane 1:3)

11a:

^1H NMR (300 MHz, CDCl_3) $\delta = 2.12$ (m, 3 H, CH_3 , 4'-H), 3.67 (d, $J = 7.4$ Hz, 2 H, CH_2 , 1'-H), 5.92 (qt, $J = 1.5, 7.3$ Hz, 1 H, 2'-H), 6.29 (d, $J = 3.5$ Hz, 1 H, 4-H), 7.18 (d, $J = 3.6$ Hz, 1 H, 3-H), 7.27–7.44 (m, 5 H, Ar-H), 9.55 (s, 1 H, CHO).

^{13}C NMR (126 MHz, CDCl_3) $\delta = 16.1$ (q, CH_3 , C-4'), 28.1 (t, C-1'), 108.9 (d, C-4), 119.8 (d, C-2'), 123.4 (d, C-3), 126.0 (d, Ar), 129.2 (d, Ar), 137.5 (s, C-3'), 138.4 (s, Ar), 152.1 (s, C-2), 163.0 (s, C-5), 177.0 (d, CHO).

11a':

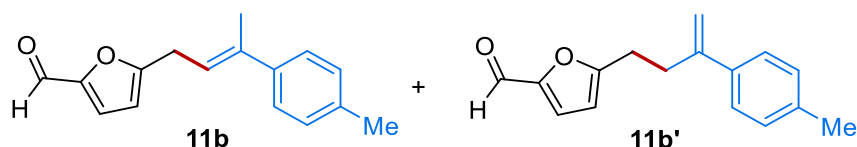
^1H NMR (300 MHz, CDCl_3) $\delta = 2.91$ (dt, $J = 2.3, 4.5$ Hz, 4 H, CH_2 , 1'-H, 2'-H), 5.08 (d, $J = 1.2$ Hz, 1 H, 4'-H^a), 5.30 (d, $J = 1.2$ Hz, 1 H, 4'-H^b), 6.21 (d, $J = 3.5$ Hz, 1 H, 4-H), 7.15 (d, $J = 3.5$ Hz, 1 H, 3-H), 7.27–7.44 (m, 5 H, Ar-H), 9.52 (s, 1 H, CHO).

^{13}C NMR (126 MHz, CDCl_3) $\delta = 27.5$ (t, C-1'), 33.3 (t, C-2'), 109.1 (d, C-4), 112.7 (t, C-4'), 123.4 (d, C-3), 125.7 (d, Ar), 129.0 (d, Ar), 140.1 (s, Ar), 146.5 (s, C-3'), 151.9 (s, C-2), 162.1 (s, C-5), 177.0 (d, CHO).

IR: $\tilde{\nu}$ = 3055, 2920, 2815 (C-H), 1670 (CO), 1510, 1395, 1190 (C-Cl), 1120, 965, 750, 695 cm^{-1} .

HRMS (ESI+): m/z calc.: $\text{C}_{15}\text{H}_{14}\text{NaO}_2^+$ $[\text{M}+\text{Na}]^+$: 249.0886, found: 249.0888.

(E)-5-(3-(p-tolyl)but-2-en-1-yl)furan-2-carbaldehyde (11b) and 5-(3-(p-tolyl)but-3-en-1-yl)furan-2-carbaldehyde (11b')



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4, α -dimethylstyrene **10b** (440 μL , 3.00 mmol, 10 equiv), Et_3B (1.0 M solution in hexanes, 150 μL , 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compounds **11b**, **11b'** (20.2 mg, 28%, inseparable mixture, ratio 40:60) after chromatography (silica, EtOAc/heptane 1:7).

Brown oil, R_f = 0.42 (EtOAc /heptane 1:3)

11b:

$^1\text{H NMR}$ (300 MHz, CDCl_3) δ = 2.09 (dt, J = 0.9, 1.5 Hz, 3 H, CH_3 , 4'-H), 2.35 (s, 3 H, Ar- CH_3), 3.65 (d, J = 7.4, 2 H, CH_2 , 1'-H), 5.89 (tt, J = 1.4, 7.3 Hz, 1 H, 2'-H), 6.28 (d, J = 3.5 Hz, 1 H, 4-H), 7.12–7.17 (m, 2 H, Ar-H), 7.18 (d, J = 3.5 Hz, 1 H, 3-H), 7.27–7.32 (m, 2 H, Ar-H), 9.54 (s, 1 H, CHO).

$^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ = 16.3 (q, CH_3 , C-4'), 21.2 (q, Ar- CH_3), 28.3 (t, C-1'), 109.0 (d, C-4), 119.9 (d, C-2'), 123.5 (d, C-3), 125.8 (d, Ar), 129.1 (d, Ar), 137.2 (s, C-3'), 137.6 (s, Ar), 138.5 (s, Ar), 152.0 (s, C-2), 163.1 (s, C-5), 177.2 (d, CHO).

11b':

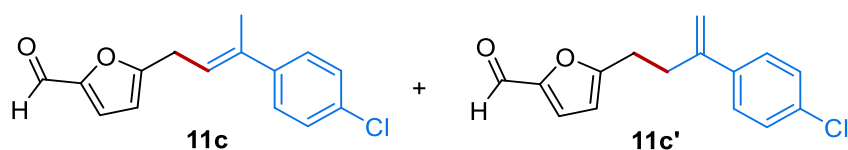
$^1\text{H NMR}$ (300 MHz, CDCl_3) δ = 2.35 (s, 3 H, Ar- CH_3), 2.89 (d, J = 0.7 Hz, 4 H, CH_2 , 1'-H, 2'-H), 5.03 (d, J = 1.2 Hz, 1 H, 4'-H^a), 5.28 (d, J = 1.2 Hz, 1 H, 4'-H^b), 6.21 (d, J = 3.5 Hz, 1 H, 4-H), 7.12–7.17 (m, 3 H, 3-H, Ar-H), 7.27–7.32 (m, 2 H, Ar-H), 9.52 (s, 1 H, CHO).

$^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ = 21.2 (q, CH_3 , Ar- CH_3), 27.6 (t, C-1'), 33.4 (t, C-2'), 109.2 (d, C-4), 112.8 (t, C-4'), 123.5 (d, C-3), 126.1 (d, Ar), 129.3 (d, Ar), 137.6 (s, Ar), 146.6 (s, C-3'), 152.0 (s, C-2), 163.1 (s, C-5), 177.1 (d, CHO).

IR: $\tilde{\nu}$ = 3020, 2915, 2815 (C-H), 1670 (CO), 1510, 1400, 1280, 1190 (C-Cl), 1015, 965, 795, 755 cm^{-1} .

HRMS (ESI+): m/z calc.: $\text{C}_{16}\text{H}_{16}\text{O}_2^+$ $[\text{M}]^+$: 240.1145, found: 240.1144.

(E)-5-(3-(4-Chlorophenyl)but-2-en-1-yl)furan-2-carbaldehyde (11c) and 5-(3-(4-Chlorophenyl)but-3-en-1-yl)furan-2-carbaldehyde (11c')



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-chloro- α -methylstyrene **10c** (430 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compounds **11c**, **11c'** (25.7 mg, 33%, inseparable mixture, ratio 60:40) after chromatography (silica, EtOAc/heptane 1:7).

Brown oil, $R_f = 0.40$ (EtOAc /heptane 1:3)

11c:

¹H NMR (300 MHz, CDCl₃) δ = 2.09 (s, 3 H, CH₃, 4'-H), 3.65 (d, J = 7.3 Hz, 2 H, CH₂, 1'-H), 5.91 (t, J = 7.3 Hz, 1 H, =CH, 2'-H), 6.28 (d, J = 3.5 Hz, 1 H, 4-H), 7.18 (d, J = 3.5 Hz, 1 H, 3-H), 7.27–7.36 (m, 4 H, Ar-H) 9.55 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 16.2 (q, CH₃, C-4'), 28.3 (t- C-1'), 109.1 (d, C-4), 121.4 (d, C-2'), 123.5 (d, C-3), 127.2 (d, Ar), 128.6 (d, Ar), 133.2 (s, Ar), 137.6 (s, Ar), 141.5 (s, C-3'), 152.3 (s, C-2), 161.7 (s, C-5), 177.2 (d, CHO).

11c':

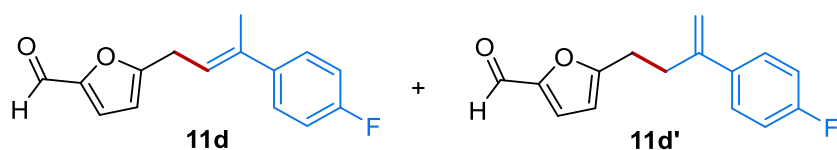
¹H NMR (300 MHz, CDCl₃) δ = 2.88 (s, 4 H, CH₂, 1'-H, 2'-H), 5.10 (d, J = 1.0 Hz, 1 H, 4'-H^a), 5.29 (d, J = 1.0 Hz, 1 H, 4'-H^b), 6.20 (d, J = 3.6 Hz, 1 H, 4-H), 7.14 (d, J = 3.5 Hz, 1 H, 3-H), 7.27–7.36 (m, 4 H Ar-H), 9.52 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 27.5 (t- C-1'), 33.3 (t- C-2'), 109.3 (d, C-4), 114.2 (t, C-4'), 123.5 (d, C-3), 127.6 (d, Ar), 128.8 (d, Ar), 133.7 (s, Ar), 139.0 (s, Ar), 145.8 (s, C-3'), 152.1 (s, C-2), 162.6 (s, C-5), 177.1 (d, C-6).

IR: $\tilde{\nu}$ = 3015, 2920, 2820 (C-H), 1670 (CO), 1510, 1400, 1190 (C-Cl), 1090, 1010, 795, 755, 545 cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₅H₁₃ClO₂⁺ [M]⁺: 260.0599, found: 260.0602.

(E)-5-(3-(4-fluorophenyl)but-2-en-1-yl)furan-2-carbaldehyde (11d) and 5-(3-(4-fluorophenyl)but-3-en-1-yl)furan-2-carbaldehyde (11d')



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 4-fluoro- α -methylstyrene **10d** (400 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compounds **11d**, **11d'** (21.2 mg, 29%, inseparable mixture, ratio 54:46) after chromatography (silica, EtOAc/heptane 1:7).

Brown colour oil, R_f = 0.40 (EtOAc /heptane 1:3)

11d:

¹H NMR (300 MHz, CDCl₃) δ = 2.09 (s, 3 H, CH₃, 4'-H), 3.65 (d, J = 7.3 Hz, 2 H, CH₂, 1'-H), 5.86 (qt, J = 1.4, 7.3 Hz, 1 H, 2'-H), 6.28 (d, J = 3.5 Hz, 1 H, 4-H), 7.01 (tdd, J = 2.1, 3.7, 8.7 Hz, 2 H, Ar-H), 7.18 (dd, J = 3.7 Hz, 1 H, 3-H), 7.32–7.40 (m, 2 H, Ar-H), 9.54 (s, 1 H, CHO).
¹³C NMR (75 MHz, CDCl₃) δ = 16.4 (q, CH₃, C-4'), 28.2 (t, C-1'), 109.3 (d, C-4), 115.1 (d, Ar), 115.3 (d, Ar), 120.8 (d, C-2'), 123.5 (d, C-3), 127.4 (d, Ar), 127.9 (d, Ar), 137.7 (s, C-3'), 139.2 (s, Ar), 152.3 (s, C-2), 162.7 (s, C-5), 164.2 (s, Ar), 177.1 (d, CHO).

¹⁹F NMR (282 MHz, CDCl₃) δ = -115.53, (s, 1 F, Ar-F)

11d':

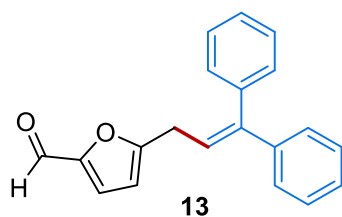
¹H NMR (300 MHz, CDCl₃) δ = 2.88 (s, 4 H, CH₂, 1'-H, 2'-H), 5.07 (d, J = 1.0 Hz, 1 H, 4'-H^a), 5.25 (d, J = 1.0 Hz, 1 H, 4'-H^b), 6.20 (d, J = 3.5 Hz, 1 H, 4-H), 7.01 (tdd, J = 2.2, 3.8, 8.7 Hz, 2 H, Ar-H), 7.14 (d, J = 3.5 Hz, 1 H, 3-H), 7.32–7.39 (m, 2 H, Ar-H), 9.52 (s, 1 H, CHO).
¹³C NMR (75 MHz, CDCl₃) δ = 27.5 (t, C-1'), 33.5 (t, C-2'), 109.1 (d, C-4), 113.6 (t, C-4'), 115.3 (d, Ar), 115.6 (d, Ar), 123.5 (d, C-3), 127.5 (d, Ar), 127.8 (d, Ar), 139.1 (s, Ar), 145.8 (s, C-3'), 152.1 (s, C-2), 161.9 (s, C-5), 163.9 (s, Ar), 177.2 (d, CHO).

¹⁹F NMR (282 MHz, CDCl₃) δ = -114.77, (s, 1 F, Ar-F)

IR: $\tilde{\nu}$ = 3020, 2925, 2815 (C-H), 1670 (CO), 1505, 1395, 1220 (C-Cl), 1160, 1015, 800, 760, 550 cm⁻¹.

HRMS (ESI+): m/z calc.: C₁₅H₁₃FO₂⁺ [M]⁺: 244.0894, found: 244.0892.

5-(3,3-Diphenylallyl)furan-2-carbaldehyde (**13**)



According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), 1,1-diphenylethylene **12** (529 μ L, 3.00 mmol, 10 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, compound **13** (25 mg, 29%) after chromatography (silica, EtOAc/heptane 1:7).

Pale yellow oil, $R_f = 0.49$ (EtOAc /heptane 1:3)

¹H NMR (500 MHz, CDCl₃) δ = 3.56 (d, $J = 7.5$ Hz, 2 H, CH₂, C-1'), 6.24 (t, $J = 7.6$ Hz, 1 H, =CH, C-2'), 6.29 (d, $J = 3.6$ Hz, 1 H 4-H), 7.16–7.23 (m, 3 H, Ph-H), 7.23–7.31 (m, 5 H, Ph-H), 7.32–7.42 (m, 3 H, 3-H, Ph-H), 9.54 (s, 1 H, CHO).

¹³C NMR (75 MHz, CDCl₃) δ = 29.5 (t- C-1'), 109.1 (d, C-4), 121.7 (d, C-2'), 123.4 (d, C-3), 127.5 (d, Ph), 127.7 (d, Ph), 128.4 (d, Ph), 128.6 (d, Ph), 129.8 (d, Ph), 139.2 (s, Ph), 141.9 (s, Ph), 145.2 (d, C-3'), 152.3 (s, C-2), 162.0 (s, C-5), 177.2 (d, CHO).

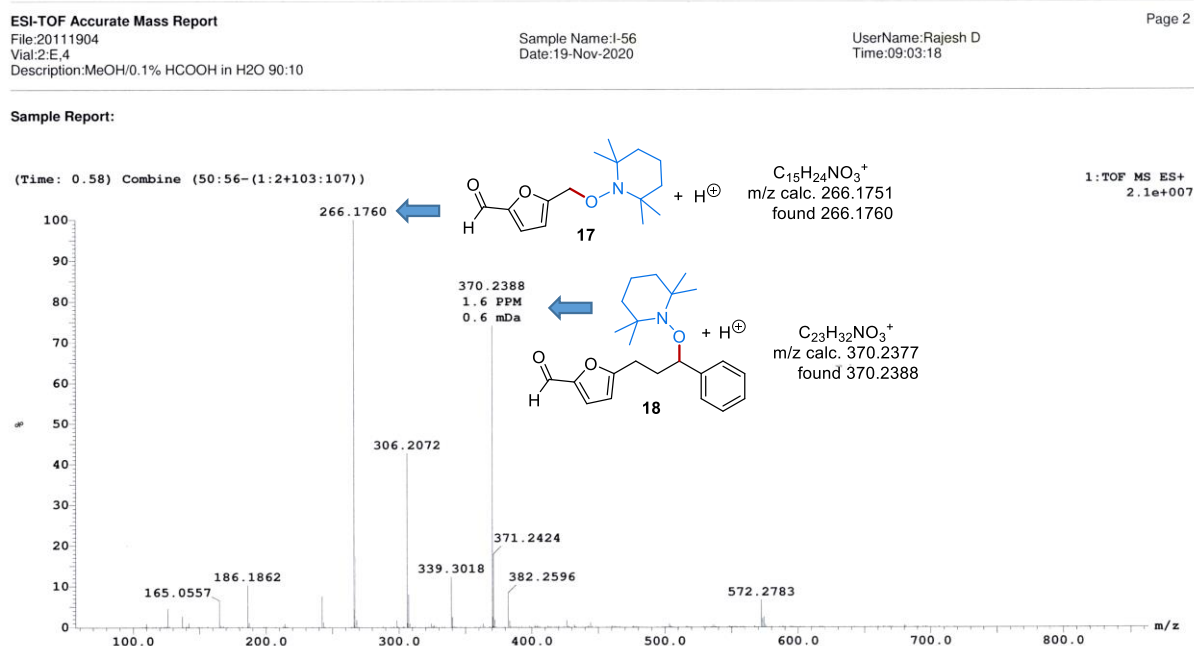
IR: $\tilde{\nu} = 3050, 3020, 2815$ (C-H), 1670 (CO), 1515, 1440, 1395, 1190 (C-Cl), 1020, 755, 695 cm^{-1} .

HRMS (ESI+): m/z calc.: C₂₀H₁₆O₂ [M]⁺: 288.1145, found: 288.1144.

3. TEMPO trapping experiments

a) Trapping experiment with TEMPO in the presence of styrene:

According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), styrene **7a** (345 μ L, 3.00 mmol, 10 equiv), TEMPO (46.9 mg, 0.30 mmol, 1 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, a crude product that was analyzed by ESI-TOF mass spectrometry. TEMPO adducts **17** and **18** were unambiguously confirmed.

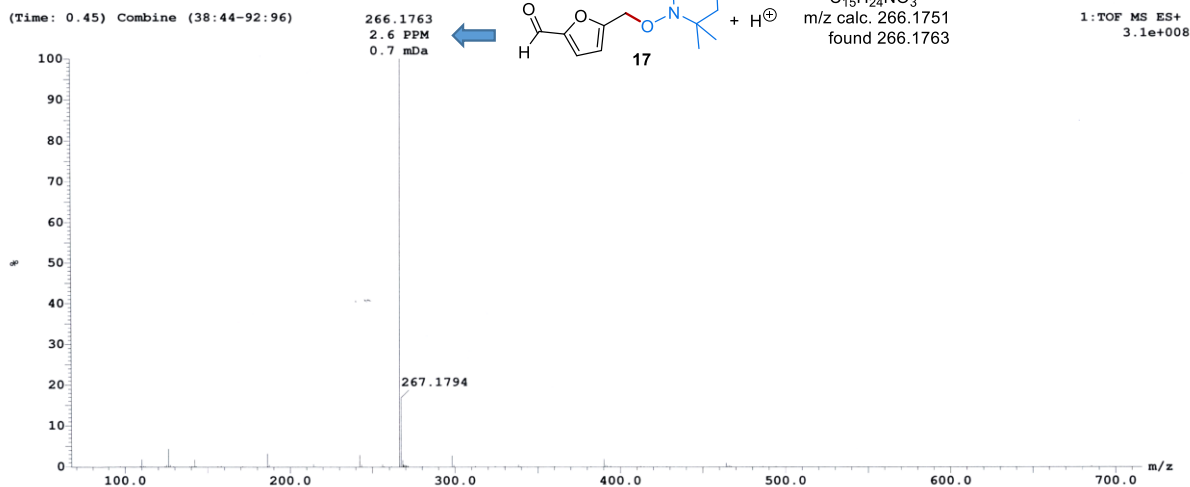


b) Trapping experiment with TEMPO in the absence of styrene:

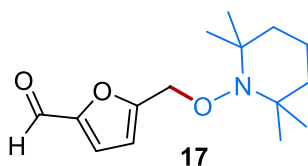
According to the General Procedure, **2** (43.4 mg, 0.30 mmol, 1.0 equiv), TBAI (443 mg, 1.20 mmol, 4.0 equiv), TEMPO (46.9 mg, 0.30 mmol, 1 equiv), Et₃B (1.0 M solution in hexanes, 150 μ L, 0.15 mmol, 0.5 equiv) in acetone (2.00 mL) gave, after 24 h reaction time, a crude product that was analyzed by ESI-TOF mass spectrometry. TEMPO adduct **17** was the only species present. Chromatography of the crude product (silica, EtOAc/heptane 1:6) gave pure compound **17** (35 mg, 44%).

Sample Report:

(Time: 0.45) Combine (38:44-92:96)



5-[[[(2,2,6,6-tetramethylpiperidin-1-yl)oxy]methyl]furan-2-carbaldehyde (17)

Colorless oil, $R_f = 0.42$ (EtOAc /heptane 1:3)

$^1\text{H NMR}$ (500 MHz, CDCl_3) $\delta = 1.11$ (s, 6 H, CH_3), 1.22 (s, 6 H, CH_3), 1.29 – 1.70 (m, 6 H, 3 \times CH_2), 4.84 (s, 2 H, CH_2 , C-1'), 6.51 (d, $J = 3.5$ Hz, 1 H, 4-H), 7.21 (d, $J = 3.5$ Hz, 1 H, 3-H), 9.61 (s, 1 H, CHO).

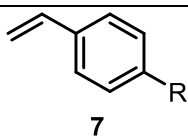
$^{13}\text{C NMR}$ (126 MHz, CDCl_3) $\delta = 17.1$ (t, CH_2), 20.2 (q, CH_3), 33.0 (q, CH_3), 39.8 (t, CH_2), 60.2 (s, $\text{C}(\text{CH}_3)_2$), 71.7 (t, C-1'), 110.9 (d, C-4), 122.2 (d, C-3), 152.4 (s, C-2), 158.5 (s, C-5), 177.8 (d, CHO).

HRMS (ESI+): m/z calc.: $C_{15}H_{24}NO_3^+$ $[M+H]^+$: 266.1751, found: 266.1763.

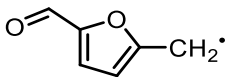
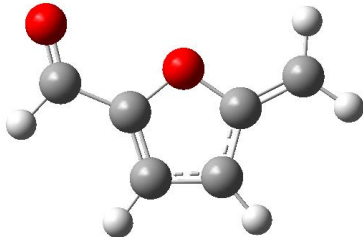
4. Computational studies

All calculations were performed with the Gaussian09 Revision E.01^[1] suite of programs. The calculations were carried out either on a uMP2/6-311+G**^{[2]-[17]} level of theory or a uM06-2X/6-311+G**^{[2]-[12],[18]} level of theory. Stationary points were confirmed as ground or transition states with the computation of the harmonic vibrational frequencies and evaluating the number of imaginary frequencies (0 for ground state, 1 for transition state). Further confirmation of the TS being associated to the reaction coordinate of interest was obtained by visualization of the imaginary frequency. Reported energies are unscaled Gibbs free energies that include a zero-point energy correction and are based on the frequency calculations within the harmonic oscillator approximation. To obtain information about spin densities and FMO energies population analyses were performed with the NBO6^[19] software.

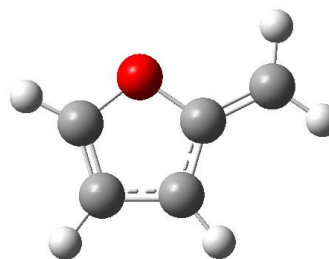
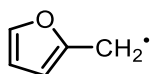
Table ESI-2. SOMO energies of radicals **3-6**, LUMO energies of 4-substituted styrenes **7** and energy gaps $\Delta\epsilon$ (in eV) calculated using NBOs at the uMP2/6-311+G** level of theory.

radicals	SOMO (eV)	3	4	5	6
	LUMO (eV)	$\Delta\epsilon$ (eV)	$\Delta\epsilon$ (eV)	$\Delta\epsilon$ (eV)	$\Delta\epsilon$ (eV)
R = H	1,762	-10,590	-9,929	-10,930	-10,533
R = Me	1,773	-10,601	-9,940	-10,941	-10,544
R = OMe	1,710	-10,538	-9,877	-10,878	-10,481
R = F	1,701	-10,529	-9,868	-10,869	-10,472
R = CN	1,192	-10,020	-9,359	-10,360	-9,963

Ground state optimizations

3			
			
Level of theory: uMP2/6-311+G**			
C	1.43203	-0.23181	-0.00036
C	1.38765	1.16412	-0.00044
C	0.02790	1.51339	0.00001
C	-0.66792	0.34829	0.00089
O	0.15794	-0.71657	0.00053
C	-2.13816	0.11712	0.00021
H	-2.69552	1.08331	0.00084
O	-2.67588	-0.93693	-0.00079
C	2.48883	-1.11985	0.00007
H	2.24724	1.81779	-0.00113
H	-0.40165	2.50568	0.00006
H	2.31083	-2.18626	0.00065
H	3.50070	-0.74011	-0.00065
Zero-point correction=		0.095818 (Hartree/Particle)	
Thermal correction to Energy=		0.102759	
Thermal correction to Enthalpy=		0.103703	
Thermal correction to Gibbs Free Energy=		0.063899	
Sum of electronic and zero-point Energies=		-380.950128	
Sum of electronic and thermal Energies=		-380.943187	
Sum of electronic and thermal Enthalpies=		-380.942243	
Sum of electronic and thermal Free Energies=		-380.982047	
	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	64.482	25.538	83.775

4



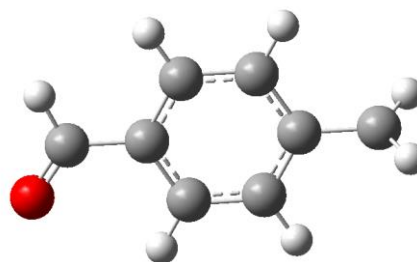
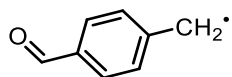
Level of theory: uMP2/6-311+G**

C	-1.46886	0.62471	0.00003
C	-1.32349	-0.71427	0.00002
O	-0.02473	-1.07518	0.00001
C	0.71710	0.08155	-0.00009
C	-0.15314	1.16180	-0.00007
C	2.09816	-0.01220	-0.00004
H	-2.40467	1.16326	0.00030
H	-2.03336	-1.52727	0.00006
H	0.14075	2.20074	-0.00028
H	2.69019	0.89180	0.00088
H	2.58631	-0.97663	-0.00008

Zero-point correction= 0.085339 (Hartree/Particle)
 Thermal correction to Energy= 0.090250
 Thermal correction to Enthalpy= 0.091194
 Thermal correction to Gibbs Free Energy= 0.056819
 Sum of electronic and zero-point Energies= -267.887282
 Sum of electronic and thermal Energies= -267.882371
 Sum of electronic and thermal Enthalpies= -267.881427
 Sum of electronic and thermal Free Energies= -267.915802

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	56.633	18.815	72.349

5



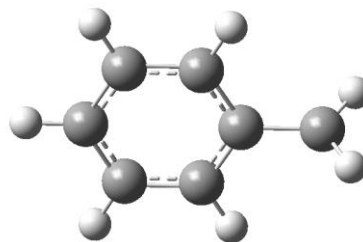
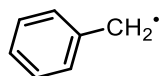
Level of theory: uMP2/6-311+G**

C	0.16485	1.34407	-0.01171
C	0.99118	0.20820	0.01554
C	0.41611	-1.07536	0.00901
C	-0.97498	-1.21155	0.01648
C	-1.81705	-0.07907	-0.00715
C	-1.22810	1.20016	-0.00627
C	2.46368	0.37960	-0.00500
O	3.26558	-0.53898	-0.00235
C	-3.29018	-0.23345	-0.00383
H	0.61216	2.33655	-0.01490
H	1.06710	-1.94539	0.02195
H	-1.42205	-2.20269	0.02288
H	-1.86648	2.07992	-0.02644
H	2.80913	1.43475	-0.02366
H	-3.81101	-0.57900	-0.88835
H	-3.86663	-0.00790	0.88491

Zero-point correction= 0.121599 (Hartree/Particle)
 Thermal correction to Energy= 0.130172
 Thermal correction to Enthalpy= 0.131116
 Thermal correction to Gibbs Free Energy= 0.088028
 Sum of electronic and zero-point Energies= -383.075480
 Sum of electronic and thermal Energies= -383.066906
 Sum of electronic and thermal Enthalpies= -383.065962
 Sum of electronic and thermal Free Energies= -383.109051

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	81.684	32.168	90.687

6



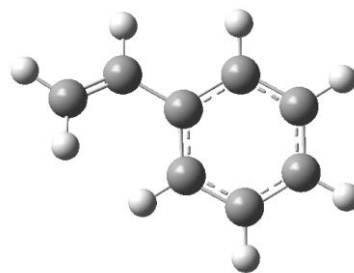
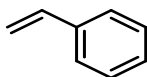
Level of theory: uMP2/6-311+G**

C	0.25383	-1.21103	-0.00868
C	-1.14890	-1.21201	-0.00911
C	-1.85399	-0.00000	-0.00002
C	-1.14891	1.21201	0.00914
C	0.25383	1.21103	0.00865
C	0.97311	-0.00000	0.00001
C	2.45496	0.00001	0.00000
H	0.79768	-2.15252	-0.01938
H	-1.68734	-2.15615	-0.01634
H	-2.94064	-0.00000	0.00000
H	-1.68734	2.15615	0.01633
H	0.79768	2.15252	0.01939
H	3.00817	0.22846	-0.90277
H	3.00819	-0.22848	0.90275

Zero-point correction= 0.111844 (Hartree/Particle)
 Thermal correction to Energy= 0.118610
 Thermal correction to Enthalpy= 0.119554
 Thermal correction to Gibbs Free Energy= 0.081236
 Sum of electronic and zero-point Energies= -270.003007
 Sum of electronic and thermal Energies= -269.996241
 Sum of electronic and thermal Enthalpies= -269.995297
 Sum of electronic and thermal Free Energies= -270.033615

	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	74.429	25.786	80.648

7a



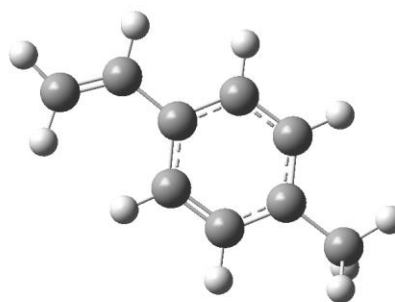
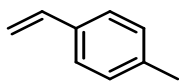
Level of theory: uMP2/6-311+G**

C	0.51829	-0.22196	-0.10104
C	-0.39180	-1.29025	-0.00008
C	-1.76581	-1.05422	0.10872
C	-2.25612	0.25685	0.08022
C	-1.36118	1.32806	-0.03871
C	0.01131	1.08991	-0.15430
C	1.96348	-0.50063	-0.18541
C	2.93281	0.32808	0.24558
H	-0.01276	-2.31005	0.02330
H	-2.45330	-1.89172	0.19417
H	-3.32377	0.44301	0.15961
H	-1.73696	2.34711	-0.08238
H	0.69574	1.92457	-0.28416
H	2.24346	-1.46951	-0.59925
H	3.97943	0.05716	0.15237
H	2.70232	1.28442	0.70644

Zero-point correction= 0.132065 (Hartree/Particle)
 Thermal correction to Energy= 0.139130
 Thermal correction to Enthalpy= 0.140074
 Thermal correction to Gibbs Free Energy= 0.100880
 Sum of electronic and zero-point Energies= -308.630904
 Sum of electronic and thermal Energies= -308.623839
 Sum of electronic and thermal Enthalpies= -308.622895
 Sum of electronic and thermal Free Energies= -308.662089

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	87.305	27.384	82.491

7b



Level of theory: uMP2/6-311+G**

C	1.03651	0.22751	-0.09632
C	0.19002	1.34769	-0.01307
C	-1.19783	1.19640	0.05706
C	-1.79027	-0.07501	0.00676
C	-0.94460	-1.19164	-0.09944
C	0.44270	-1.04572	-0.17530
C	2.49700	0.41448	-0.14871
C	3.40537	-0.48474	0.27464
C	-3.28616	-0.24146	0.11204
H	0.62748	2.34366	0.02472
H	-1.83077	2.07915	0.12749
H	-1.38176	-2.18640	-0.16633
H	1.06995	-1.92553	-0.29760
H	2.84530	1.37404	-0.53146
H	4.46826	-0.27763	0.20463
H	3.10807	-1.43690	0.70483
H	-3.80753	0.63298	-0.28793
H	-3.62253	-1.12325	-0.44106
H	-3.59295	-0.36519	1.15676

Zero-point correction= 0.159808 (Hartree/Particle)

Thermal correction to Energy= 0.168714

Thermal correction to Enthalpy= 0.169658

Thermal correction to Gibbs Free Energy= 0.125921

Sum of electronic and zero-point Energies= -347.803462

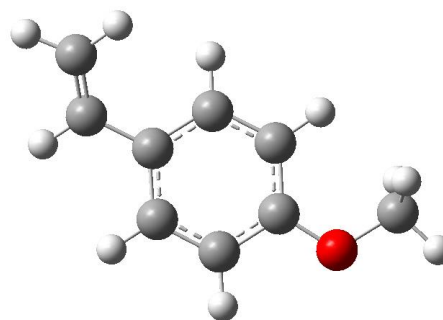
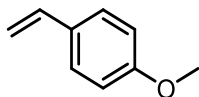
Sum of electronic and thermal Energies= -347.794556

Sum of electronic and thermal Enthalpies= -347.793611

Sum of electronic and thermal Free Energies= -347.837349

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	105.870	33.415	92.053

7n



Level of theory: uMP2/6-311+G**

C	-1.47152	0.20927	0.10068
C	-0.77294	1.42985	0.00449
C	0.61667	1.46307	-0.07053
C	1.35495	0.26980	-0.01365
C	0.68039	-0.95522	0.10123
C	-0.71787	-0.97153	0.18306
C	-2.94320	0.20548	0.15141
C	-3.72833	-0.80200	-0.27518
O	2.71098	0.40807	-0.08432
C	3.47489	-0.78804	-0.03979
H	-1.33206	2.36229	-0.04158
H	1.15590	2.40251	-0.15351
H	1.22262	-1.89138	0.17378
H	-1.22479	-1.92508	0.31024
H	-3.41298	1.10955	0.53966
H	-4.80886	-0.73572	-0.20082
H	-3.31056	-1.70452	-0.71235
H	4.51591	-0.47659	-0.11909
H	3.32331	-1.32098	0.90568
H	3.22543	-1.44871	-0.87778

Zero-point correction= 0.165092 (Hartree/Particle)

Thermal correction to Energy= 0.174794

Thermal correction to Enthalpy= 0.175738

Thermal correction to Gibbs Free Energy= 0.129979

Sum of electronic and zero-point Energies= -422.856986

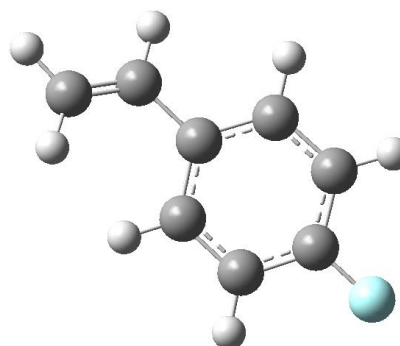
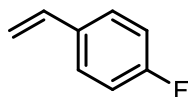
Sum of electronic and thermal Energies= -422.847284

Sum of electronic and thermal Enthalpies= -422.846340

Sum of electronic and thermal Free Energies= -422.892099

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	109.685	36.542	96.307

7f



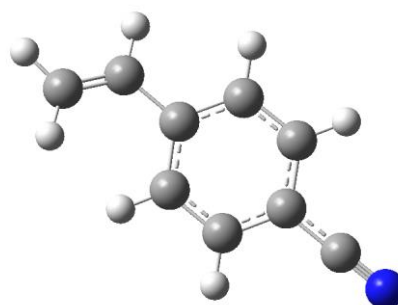
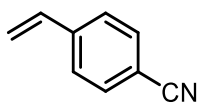
Level of theory: uMP2/6-311+G**

C	-0.98112	0.22912	-0.10640
C	-0.12969	1.34518	-0.01040
C	1.25759	1.19499	0.07235
C	1.78830	-0.09059	0.02938
C	0.98334	-1.22082	-0.08588
C	-0.40074	-1.05140	-0.17386
C	-2.44067	0.42593	-0.16562
C	-3.35402	-0.45930	0.27466
F	3.12829	-0.24791	0.09994
H	-0.56072	2.34322	0.02493
H	1.92171	2.04958	0.15522
H	1.44227	-2.20317	-0.13954
H	-1.03582	-1.92425	-0.29963
H	-2.78169	1.37872	-0.57045
H	-4.41557	-0.24862	0.19738
H	-3.06263	-1.40289	0.72727

Zero-point correction= 0.124005 (Hartree/Particle)
 Thermal correction to Energy= 0.131864
 Thermal correction to Enthalpy= 0.132808
 Thermal correction to Gibbs Free Energy= 0.091513
 Sum of electronic and zero-point Energies= -407.716615
 Sum of electronic and thermal Energies= -407.708756
 Sum of electronic and thermal Enthalpies= -407.707812
 Sum of electronic and thermal Free Energies= -407.749107

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	82.746	30.249	86.911

7j



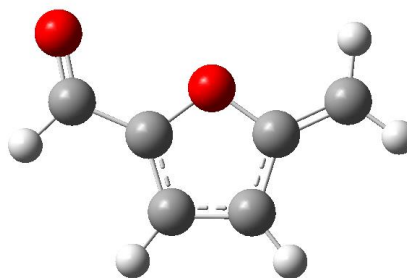
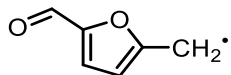
Level of theory: uMP2/6-311+G**

C	-1.31675	0.22203	-0.10033
C	-0.49775	1.36354	-0.01451
C	0.89148	1.25414	0.05200
C	1.48947	-0.01668	-0.01048
C	0.68782	-1.16802	-0.11383
C	-0.69899	-1.04040	-0.18546
C	-2.78131	0.37558	-0.14248
C	-3.66041	-0.54399	0.29790
C	2.91767	-0.14200	0.05954
N	4.08983	-0.24452	0.11708
H	-0.95972	2.34726	0.03265
H	1.51502	2.14021	0.12981
H	1.15749	-2.14552	-0.17786
H	-1.31038	-1.93108	-0.30395
H	-3.15521	1.32402	-0.52739
H	-4.72880	-0.36401	0.24000
H	-3.33455	-1.48451	0.73308

Zero-point correction= 0.130013 (Hartree/Particle)
 Thermal correction to Energy= 0.138982
 Thermal correction to Enthalpy= 0.139926
 Thermal correction to Gibbs Free Energy= 0.095906
 Sum of electronic and zero-point Energies= -400.671735
 Sum of electronic and thermal Energies= -400.662766
 Sum of electronic and thermal Enthalpies= -400.661822
 Sum of electronic and thermal Free Energies= -400.705842

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	87.212	33.634	92.648

3



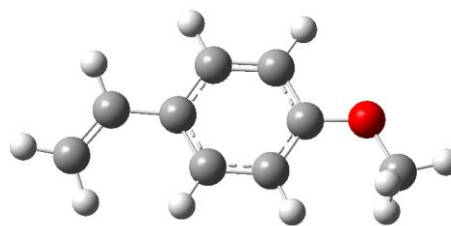
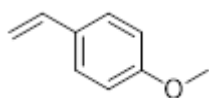
Level of theory: uM06-2X/6-311+G**

C	1.43187	-0.24576	-0.00021
C	1.38425	1.17083	0.00014
C	0.04194	1.52564	-0.00023
C	-0.68469	0.34186	0.00009
O	0.14619	-0.71967	0.00010
C	-2.12135	0.13617	0.00014
H	-2.69944	1.07897	0.00022
O	-2.66260	-0.94431	-0.00010
C	2.47752	-1.12315	-0.00002
H	2.24628	1.81823	0.00046
H	-0.38272	2.51771	-0.00026
H	2.30006	-2.18861	0.00007
H	3.48990	-0.74803	0.00009

Zero-point correction= 0.094526 (Hartree/Particle)
 Thermal correction to Energy= 0.101284
 Thermal correction to Enthalpy= 0.102228
 Thermal correction to Gibbs Free Energy= 0.062869
 Sum of electronic and zero-point Energies= -381.876224
 Sum of electronic and thermal Energies= -381.869466
 Sum of electronic and thermal Enthalpies= -381.868522
 Sum of electronic and thermal Free Energies= -381.907881

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	63.557	25.198	82.838

7n



Level of theory: uM06-2X/6-311+G**

C	-1.46177	0.20941	0.05720
C	-0.76122	1.42205	0.00146
C	0.62051	1.45857	-0.03569
C	1.35007	0.26851	-0.00873
C	0.67558	-0.95133	0.06227
C	-0.71471	-0.96742	0.09762
C	-2.93410	0.22447	0.07432
C	-3.74290	-0.81129	-0.14470
O	2.70092	0.39683	-0.04686
C	3.47547	-0.78342	-0.02469
H	-1.31756	2.35332	-0.02017
H	1.16210	2.39511	-0.08390
H	1.21619	-1.88730	0.10090
H	-1.21971	-1.92344	0.17322
H	-3.37967	1.19614	0.27452
H	-4.81834	-0.69349	-0.10564
H	-3.36732	-1.80119	-0.37853
H	4.51379	-0.46352	-0.07161
H	3.31016	-1.34789	0.89829
H	3.25128	-1.41962	-0.88657

Zero-point correction= 0.167054 (Hartree/Particle)

Thermal correction to Energy= 0.176315

Thermal correction to Enthalpy= 0.177259

Thermal correction to Gibbs Free Energy= 0.132197

Sum of electronic and zero-point Energies= -423.927653

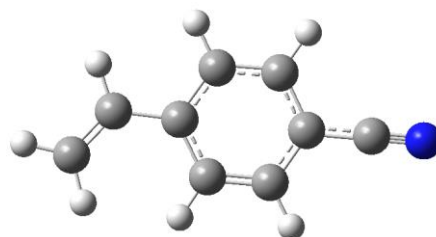
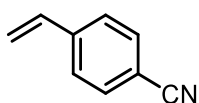
Sum of electronic and thermal Energies= -423.918392

Sum of electronic and thermal Enthalpies= -423.917448

Sum of electronic and thermal Free Energies= -423.962510

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	110.639	34.997	94.840

7j



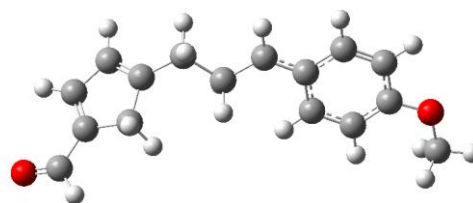
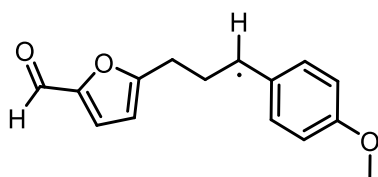
Level of theory: uM06-2X/6-311+G**

C	-1.30211	0.22364	0.00001
C	-0.48232	1.35702	0.00000
C	0.89985	1.24627	-0.00000
C	1.48782	-0.01868	-0.00001
C	0.68374	-1.16368	-0.00001
C	-0.69364	-1.03848	0.00000
C	-2.76431	0.40551	0.00001
C	-3.68376	-0.55762	0.00002
C	2.91816	-0.14673	-0.00001
N	4.06397	-0.25050	-0.00002
H	-0.93858	2.34060	0.00001
H	1.52605	2.12954	-0.00001
H	1.14835	-2.14176	-0.00001
H	-1.30103	-1.93482	0.00000
H	-3.09546	1.44055	0.00002
H	-4.73832	-0.31311	0.00002
H	-3.42943	-1.61111	0.00002

Zero-point correction= 0.132796 (Hartree/Particle)
 Thermal correction to Energy= 0.140470
 Thermal correction to Enthalpy= 0.141414
 Thermal correction to Gibbs Free Energy= 0.100256
 Sum of electronic and zero-point Energies= -401.688524
 Sum of electronic and thermal Energies= -401.680851
 Sum of electronic and thermal Enthalpies= -401.679906
 Sum of electronic and thermal Free Energies= -401.721064

	E (Thermal) KCal/Mol	CV Cal/Mol-Kelvin	S Cal/Mol-Kelvin
Total	88.146	29.944	86.623

14n



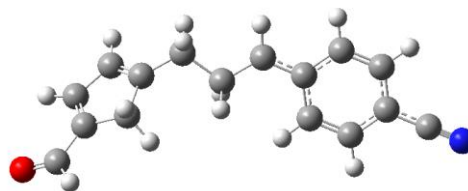
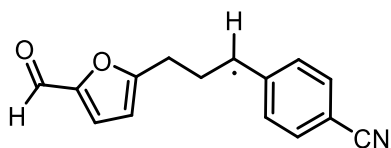
Level of theory: uM06-2X/6-311+G**

C	-4.76161	0.56479	-0.11771
C	-4.99689	-0.28257	0.90906
C	-3.88448	-1.21404	1.03244
C	-2.96061	-0.94483	0.08440
C	-3.43460	0.22413	-0.73591
C	-5.63956	1.65018	-0.55957
O	-6.70526	1.91716	-0.05661
C	-1.66866	-1.65604	-0.15871
C	-0.44398	-0.73836	-0.03977
C	0.83807	-1.47814	-0.24079
C	2.11510	-0.87781	-0.15531
C	3.29319	-1.64660	-0.36202
C	4.54356	-1.08495	-0.28415
C	4.68965	0.28167	0.00578
C	3.55485	1.06697	0.21460
C	2.29284	0.49345	0.13444
O	5.96472	0.73994	0.06084
C	6.16125	2.10780	0.35191
H	-5.87954	-0.25987	1.53521
H	-3.81523	-2.00340	1.76986
H	-3.52092	-0.04256	-1.79679
H	-2.73263	1.06573	-0.67957
H	-5.26440	2.24383	-1.41781
H	-1.57330	-2.48114	0.55259
H	-1.67603	-2.09813	-1.16242
H	-0.52568	0.07597	-0.77363
H	-0.45703	-0.25252	0.94598
H	0.78712	-2.53797	-0.46980
H	3.19367	-2.70327	-0.58622
H	5.43944	-1.67300	-0.44171
H	3.64148	2.12145	0.43995
H	1.42575	1.12181	0.30135
H	7.23719	2.26531	0.34946
H	5.69806	2.74355	-0.40899
H	5.75763	2.36340	1.33651

Zero-point correction= 0.289066 (Hartree/Particle)
 Thermal correction to Energy= 0.306414
 Thermal correction to Enthalpy= 0.307358
 Thermal correction to Gibbs Free Energy= 0.240443
 Sum of electronic and zero-point Energies= -769.878718
 Sum of electronic and thermal Energies= -769.861371
 Sum of electronic and thermal Enthalpies= -769.860426
 Sum of electronic and thermal Free Energies= -769.927341

	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	192.278	64.950	140.835

14j



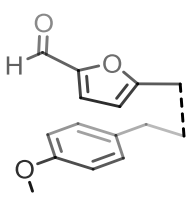
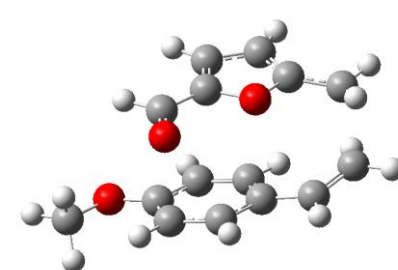
Level of theory: uM06-2X/6-311+G**

C	4.63929	0.50723	0.11109
C	4.83057	-0.35444	-0.91216
C	3.68274	-1.24577	-1.01410
C	2.78232	-0.93572	-0.05698
C	3.30967	0.21941	0.75040
C	5.56363	1.56324	0.53410
O	6.63023	1.78459	0.01300
C	1.46819	-1.59874	0.20555
C	0.27522	-0.64553	0.04820
C	-1.03180	-1.32999	0.27659
C	-2.28914	-0.70089	0.15570
C	-3.48201	-1.43305	0.39504
C	-4.71958	-0.84129	0.28411
C	-4.81840	0.51346	-0.07220
C	-3.65599	1.25971	-0.31421
C	-2.41640	0.66657	-0.20301
C	-6.10408	1.13257	-0.18819
N	-7.13805	1.63079	-0.28127
H	5.70515	-0.36881	-1.54960
H	3.57607	-2.03816	-1.74388
H	3.40429	-0.04649	1.81085
H	2.64079	1.08809	0.70249
H	5.22379	2.17655	1.39294
H	1.34882	-2.44333	-0.47866
H	1.45787	-2.00648	1.22346
H	0.37965	0.19424	0.74891
H	0.30086	-0.20067	-0.95566
H	-1.01665	-2.37858	0.55797
H	-3.40764	-2.47912	0.67010
H	-5.62232	-1.41005	0.46892
H	-3.74346	2.30367	-0.58878
H	-1.52812	1.25633	-0.39378

Zero-point correction= 0.254890 (Hartree/Particle)
 Thermal correction to Energy= 0.271485
 Thermal correction to Enthalpy= 0.272429
 Thermal correction to Gibbs Free Energy= 0.206957
 Sum of electronic and zero-point Energies= -747.640552
 Sum of electronic and thermal Energies= -747.623957
 Sum of electronic and thermal Enthalpies= -747.623013
 Sum of electronic and thermal Free Energies= -747.688485

	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	170.359	61.812	137.797

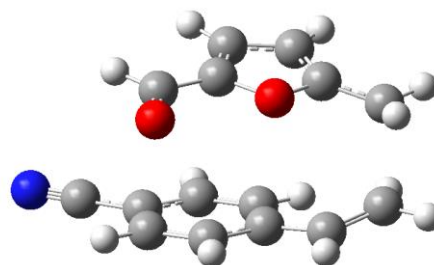
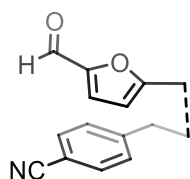
Transition state optimizations

TS 3 – 7n			
			
Level of theory: uM06-2X/6-311+G**			
C	-0.57331	1.75790	0.22973
C	-0.66656	1.55678	1.58771
C	-1.87545	0.87522	1.81073
C	-2.46224	0.69876	0.56137
O	-1.66572	1.24691	-0.39105
C	0.48244	2.37189	-0.54860
O	0.49401	2.49028	-1.75241
C	-3.59489	-0.01486	0.12602
C	-2.78206	-1.99944	-0.44785
C	-1.65836	-1.75367	-1.18920
C	-0.33423	-1.49915	-0.65221
C	0.02103	-1.77144	0.68408
C	1.26831	-1.44673	1.17353
C	2.21448	-0.83261	0.34346
C	1.89689	-0.57096	-0.98928
C	0.63639	-0.90565	-1.46789
H	1.32083	2.73395	0.07827
H	0.06920	1.85951	2.31683
H	-2.28699	0.54044	2.74925
H	-4.29810	-0.34969	0.87747
H	-3.98582	0.20953	-0.85667
H	-2.70714	-2.28459	0.59500
H	-3.69810	-2.28842	-0.94881
H	-1.76968	-1.60612	-2.25958
H	0.38487	-0.65890	-2.49364
H	-0.69245	-2.24519	1.34787
H	1.54299	-1.65191	2.20109
H	2.60003	-0.08646	-1.65296
O	3.40254	-0.52751	0.92241
C	4.38306	0.09684	0.11792
H	4.66938	-0.54308	-0.72204
H	4.02719	1.05919	-0.26320
H	5.24422	0.25692	0.76238
$\nu_{\text{imag}} = -545.37 \text{ cm}^{-1}$			

Zero-point correction= 0.263768 (Hartree/Particle)
 Thermal correction to Energy= 0.280102
 Thermal correction to Enthalpy= 0.281046
 Thermal correction to Gibbs Free Energy= 0.218548
 Sum of electronic and zero-point Energies= -805.800117
 Sum of electronic and thermal Energies= -805.783784
 Sum of electronic and thermal Enthalpies= -805.782839
 Sum of electronic and thermal Free Energies= -805.845337

	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	175.766	62.792	131.538

TS 3 – 7j



Level of theory: uM06-2X/6-311+G**

C	0.47722	1.76205	-0.22764
C	0.43007	1.55939	-1.58678
C	1.58523	0.82839	-1.92217
C	2.28256	0.63114	-0.73528
O	1.60077	1.20677	0.28735
C	-0.46724	2.41750	0.66160
O	-0.33625	2.50984	1.85913
C	3.43341	-0.10967	-0.40501
C	2.63506	-2.05582	0.33131
C	1.58942	-1.74673	1.15618
C	0.23252	-1.47497	0.71544
C	-0.22109	-1.78091	-0.58072
C	-1.49602	-1.43758	-0.98865
C	-2.35751	-0.77668	-0.10544
C	-1.93375	-0.48245	1.19471
C	-0.65720	-0.83315	1.59445
C	-3.66750	-0.38840	-0.54232
H	-1.35006	2.82885	0.13724
H	-0.35681	1.89680	-2.24381
H	1.89306	0.48214	-2.89573
H	4.04158	-0.48665	-1.21694
H	3.93143	0.12051	0.52656
H	2.46598	-2.38096	-0.68814
H	3.58769	-2.33907	0.76213
H	1.79343	-1.55829	2.20569
H	-0.31865	-0.56638	2.58911
H	0.43094	-2.29185	-1.27803
H	-1.83556	-1.67113	-1.99027
H	-2.59788	0.03827	1.87354
N	-4.71320	-0.06825	-0.90161

$\nu_{\text{imag}} = -557.87 \text{ cm}^{-1}$

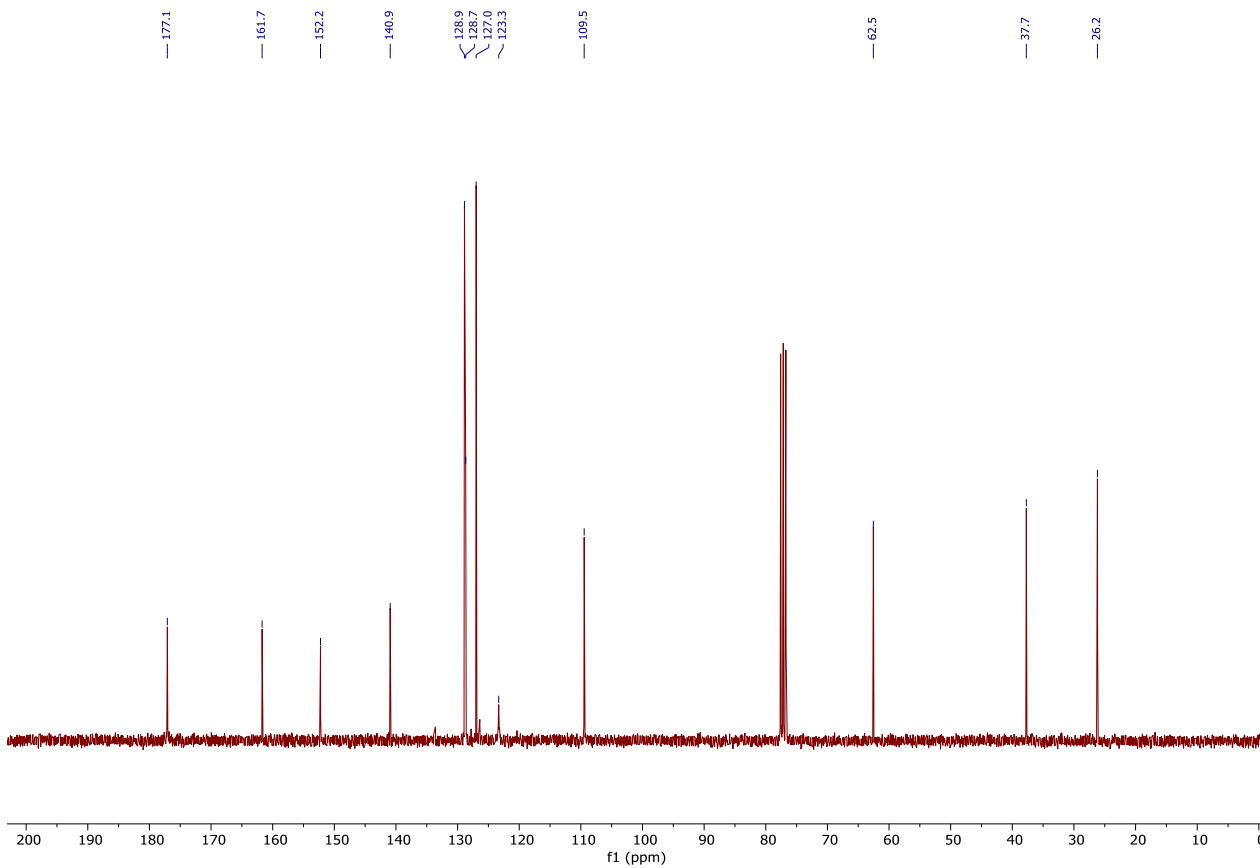
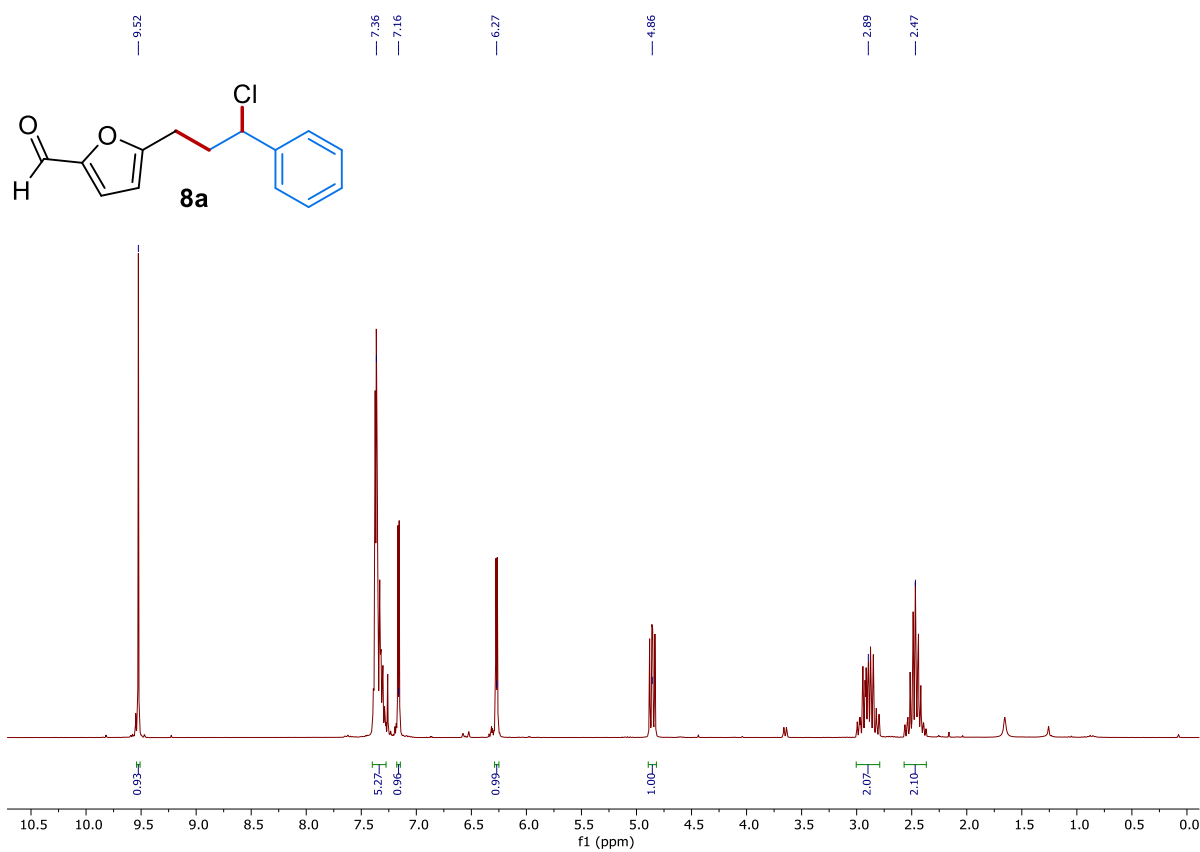
Zero-point correction= 0.229289 (Hartree/Particle)
 Thermal correction to Energy= 0.244965
 Thermal correction to Enthalpy= 0.245909
 Thermal correction to Gibbs Free Energy= 0.184573
 Sum of electronic and zero-point Energies= -783.561043
 Sum of electronic and thermal Energies= -783.545368
 Sum of electronic and thermal Enthalpies= -783.544423
 Sum of electronic and thermal Free Energies= -783.605760

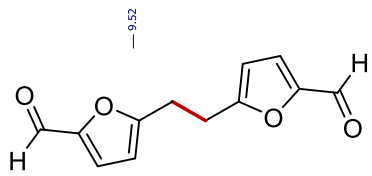
	E (Thermal)	CV	S
	KCal/Mol	Cal/Mol-Kelvin	Cal/Mol-Kelvin
Total	153.718	59.880	129.094

References

- [1] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox. *Gaussian09, Revision E.01*; Gaussian Inc.: Wallingford CT, 2013.
- [2] Wachters, A. J. H. *J. Chem. Phys.* **1970**, *52*, 1033–1036.
- [3] Raghavachari, K.; Trucks, G. W. *J. Chem. Phys.* **1989**, *91*, 1062–1065.
- [4] McLean, A. D.; Chandler, G. S. *J. Chem. Phys.* **1980**, *72*, 5639–5648.
- [5] McGrath, M. P.; Radom, L. *J. Chem. Phys.* **1991**, *94*, 511–516.
- [6] Krishnan, R.; Binkley, J. S.; Seeger, R.; Pople, J. A. *J. Chem. Phys.* **1980**, *72*, 650–654.
- [7] Hay, P. J. *J. Chem. Phys.* **1977**, *66*, 4377–4384.
- [8] Frisch, M. J.; Pople, J. A.; Binkley, J. S. *J. Chem. Phys.* **1984**, *80*, 3265–3269.
- [9] Curtiss, L. A.; McGrath, M. P.; Blaudeau, J.-P.; Davis, N. E.; Binning, R. C.; Radom, L. *J. Chem. Phys.* **1995**, *103*, 6104–6113.
- [10] Clark, T.; Chandrasekhar, J.; Spitznagel, G. W.; Schleyer, P. V. R. *J. Comput. Chem.* **1983**, *4*, 294–301.
- [11] Blaudeau, J.-P.; McGrath, M. P.; Curtiss, L. A.; Radom, L. *J. Chem. Phys.* **1997**, *107*, 5016–5021.
- [12] Binning, R. C.; Curtiss, L. A. *J. Comput. Chem.* **1990**, *11*, 1206–1216.
- [13] Sæbø, S.; Almlöf, J. *Chem. Phys. Lett* **1989**, *154*, 83–89.
- [14] Head-Gordon, M.; Pople, J. A.; Frisch, M. J. *Chem. Phys. Lett* **1988**, *153*, 503–506.
- [15] Head-Gordon, M.; Head-Gordon, T. *Chem. Phys. Lett* **1994**, *220*, 122–128.
- [16] Frisch, M. J.; Head-Gordon, M.; Pople, J. A. *Chem. Phys. Lett* **1990**, *166*, 281–289.
- [17] Frisch, M. J.; Head-Gordon, M.; Pople, J. A. *Chem. Phys. Lett* **1990**, *166*, 275–280.
- [18] Zhao, Y.; Truhlar, D. G. *Theor Chem Account* **2008**, *120*, 215–241.
- [19] E. D. Glendening, J. K. Badenhoop, A. E. Reed. *NBO 6.0*; Theoretical Chemistry Institute, University of Wisconsin, Madison, WI, 2013.

5. NMR spectra





9 (purity ~90%)

