# Synthesis of methanesulfones-containing tetrasubstituted carbon stereocenters

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#### 1. General experimental information

Reactions were monitored by thin layer chromatography using UV light to visualize the course of reaction. Purification of reaction products was carried out by flash chromatography on silica gel. Chemical yields refered to pure isolated substances. <sup>1</sup>H and <sup>13</sup>CNMR spectra were obtained using a Bruker DPX-400 spectrometer. <sup>1</sup>H NMR chemical shifts are reported in ppm ( $\delta$ ) relative to tetramethylsilane (TMS) with the solvent resonance employed as the internal standard. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, m = multiplet), coupling constants (Hz) and integration. <sup>13</sup>C NMR chemical shifts are reported in ppm ( $\delta$ ) from tetramethylsilane (TMS) with the solvent resonance as the internal standard.

#### 2. Procedure for the synthesis of compounds 1



A solution of ninhydrin (0.60 mmol), dihydrochromone (0.80 mmol) and  $Et_2NH$  (0.15 mmol) in MeOH (3.0 mL) was stirred at 25 °C for 24 h. After completion of the reaction, as indicated by TLC, the removal of solvent and purification by flash column chromatography (hexane/EtOAc =  $6:1\sim4:1$ ) was carried out to furnish the indanedione-chromanone synthon **1** as a white solid.

#### 3. Typical experimental procedures for synthesis of compounds 3



Compound 1 (0.20 mmoL) and  $Et_3N$  (0.60 mmoL) were dissolved in DCM (1.5 mL). MsCl (0.50 mmoL) was added to this solution at room temperature, and the mixture was stirred for 8 h. After the removal of solvent, purification by flash column chromatography (hexane/ethyl acetate =  $8:1\sim5:1$ ) was carried out to give product **3** as a light yellow solid.

#### 4. Procedure for the synthesis of compounds IB



A solution of substrate 1 (0.20 mmol) in  $Ac_2O$  (2.0 mL) was added HCl (6 N, 0.1 mL), and then stirred at 65 °C for 8 h. After completion of the reaction, as indicated by TLC, the removal of solvent and purification by flash column chromatography (hexane/EtOAc = 5:1~3:1) was carried out to furnish the indanedione-chromanone synthon **1B** as a red solid.

#### 5. Gram scale synthesis of the product 3q



Compound 1 (2.0 mmoL) and  $Et_3N$  (6.0 mmoL) were dissolved in DCM (15 mL). MsCl (5.0 mmoL) was added to this solution at room temperature, and the mixture was stirred for 8 h. After the removal of solvent, purification by flash column chromatography (hexane/ethyl acetate = 8:1~5:1) was carried out to give product **3q** as a light yellow solid (0.78 g, 88% yield).

#### 6. Scheme S1: control experiments



Compound **1B-1h** (0.20 mmoL) and  $Et_3N$  (0.44 mmoL) was dissolved in DCM (1.5 mL). compound **2** (0.40 mmoL) was added to this solution at room temperature, and the mixture was stirred for 10 h, however, the expected corresponding sulfone **3** was not observed with intractable product mixtures (0% yield).



Compound **1B-1h** (0.20 mmoL) and NaOH (0.44 mmoL) was dissolved in DCM (1.5 mL). compound **2** (0.40 mmoL) was added to this solution at room temperature, and the mixture was

stirred for 5 h. After the removal of solvent, purification by flash column chromatography (hexane/ethyl acetate =  $8:1\sim5:1$ ) was carried out to give product **3h** as a light yellow solid (27% yield).

#### 7. Scheme S2: control experiments



Compound 1 (0.20 mmoL) was dissolved in Ac<sub>2</sub>O (2 mL). HCl (6 N, 0.1 mL) was added to this solution at 65 °C, and the mixture was stirred for 8 h. After the removal of solvent, purification by flash column chromatography (hexane/ethyl acetate =  $8:1\sim5:1$ ) was carried out to give intermediate **IB** as a red solid.

Intermediate **IB** (0.2 mmoL) and Et<sub>3</sub>N (4.4 mmoL) were dissolved in DCM (1.5 mL). MsCl (0.40 mmoL) was added to this solution at room temperature, and the mixture was stirred for 3 h. After the removal of solvent, purification by flash column chromatography (hexane/ethyl acetate =  $6:1\sim4:1$ ) was carried out to give product **3** as a light yellow solid.

#### 8. Scheme S3: other substrates in this transformation



Compound 4 or 6 (0.20 mmoL) and  $Et_3N$  (0.44 mmoL) was dissolved in DCM (1.5 mL). compound 2 (0.40 mmoL) was added to this solution at room temperature, and the mixture was stirred for 10 h, however, the expected corresponding sulfone 5 or 7 was not observed with recovery of starting material 4 or 6 (0% yield).

#### 9. X-Ray Crystal Data for Compound 3a



#### Table S1 Crystal data and structure refinement for 3a

Identification code	3a
Empirical formula	$C_{19}H_{12}O_6S$
Formula weight	368.35
Temperature/K	100.00(10)
Crystal system	triclinic
Space group	P-1
a/Å, b/Å, c/Å	7.6249(5), 9.1227(5), 12.8047(9)
$\alpha/^{\circ}, \beta/^{\circ}, \gamma/^{\circ},$	85.841(5), 88.200(5), 81.018(5).
Volume/Å <sup>3</sup>	877.27(10)
Ζ	2
$\rho_{calc}g/cm^3$	1.394
µ/mm <sup>-1</sup>	0.217
F(000)	380.0
Radiation	Mo K $\alpha$ ( $\lambda = 0.71073$ )
Crystal size/mm <sup>3</sup>	$0.14 \times 0.13 \times 0.12$
$2\Theta$ range for data collection/°	4.532 to 49.994

Index ranges	$\textbf{-9} \le h \le 9,  \textbf{-10} \le k \le 10,  \textbf{-12} \le l \le 15$
Reflections collected	5655
Independent reflections	$3074 [R_{int} = 0.0247, R_{sigma} = 0.0446]$
Data/restraints/parameters	3074/0/236
Goodness-of-fit on F <sup>2</sup>	1.097
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0470, wR_2 = 0.1165$
Final R indexes [all data]	$R_1 = 0.0532, wR_2 = 0.1205$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.41/-0.49

Crystal data for **3a**: (M = 368.35 g/mol): triclinic, space group P-1 (no. 2), a = 7.6249(5) Å, b =9.1227(5) Å, c = 12.8047(9) Å,  $a = 85.841(5)^{\circ}$ ,  $\beta = 88.200(5)^{\circ}$ ,  $\gamma = 81.018(5)^{\circ}$ , V = 12.8047(9) Å,  $a = 85.841(5)^{\circ}$ ,  $\beta = 88.200(5)^{\circ}$ ,  $\gamma = 81.018(5)^{\circ}$ , V = 12.8047(9) Å,  $\gamma = 81.018(5)^{\circ}$ ,  $\gamma = 81.018(5)^{\circ}$ , V = 12.8047(9) Å,  $\gamma = 81.018(5)^{\circ}$ , V = 12.8047(9) Å,  $\gamma = 81.018(5)^{\circ}$ , V = 12.8047(9) Å,  $\gamma = 81.018(5)^{\circ}$ ,  $\gamma = 8$ 877.27(10) Å<sup>3</sup>, Z = 2, T = 100.00(10) K,  $\mu$ (Mo K $\alpha$ ) = 0.217 mm<sup>-1</sup>, Dcalc = 1.394 g/cm<sup>3</sup>, 5655 reflections measured ( $4.532^\circ \le 2\Theta \le 49.994^\circ$ ), 3074 unique ( $R_{int} = 0.0247$ ,  $R_{sigma} = 0.0446$ ) which were used in all calculations. The final  $R_1$  was 0.0470 (I > 2 $\sigma$ (I)) and  $wR_2$  was 0.1205 (all data).

7,7,290 (1,7,1,287) (1,2,27) ( -3.812 -3.792 -3.786 -3.786 -3.741 2.00-1.24J 0.974 0.994 1.084 1.084 1.464 0.174 0.174 0.314 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 fl (ppm) 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.

10. The Copies of <sup>1</sup>H NMR, <sup>13</sup>C NMR and <sup>19</sup>F NMR Spectra for Compounds 1 and 3 <sup>1</sup>H and <sup>13</sup>C NMR of 1a



<sup>1</sup>H and <sup>13</sup>C NMR of 1b





## <sup>19</sup>F NMR of 1b





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<sup>1</sup>H and <sup>13</sup>C NMR of 1B-1g





<sup>1</sup>H and <sup>13</sup>C NMR of 1B-1h

<sup>1</sup>H and <sup>13</sup>C NMR of 3a





<sup>1</sup>H and <sup>13</sup>C NMR of 3b









<sup>1</sup>H and <sup>13</sup>C NMR of 3c





<sup>1</sup>H and <sup>13</sup>C NMR of 3d







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<sup>1</sup>H and <sup>13</sup>C NMR of 3f





## <sup>19</sup>F NMR of 3f









<sup>1</sup>H and <sup>13</sup>C NMR of 3h









<sup>19</sup>F NMR of 3i



<sup>1</sup>H and <sup>13</sup>C NMR of 3j





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<sup>19</sup>F NMR of 3k







## <sup>19</sup>F NMR of 3l













<sup>1</sup>H and <sup>13</sup>C NMR of 3n

<sup>19</sup>F NMR of 3n







#### <sup>1</sup>H and <sup>13</sup>C NMR of 30



## <sup>1</sup>H and <sup>13</sup>C NMR of 3q



<sup>1</sup>H and <sup>13</sup>C NMR of 3r







<sup>1</sup>H and <sup>13</sup>C NMR of 3s





<sup>1</sup>H and <sup>13</sup>C NMR of 3u

