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Supporting Information

Transition-metal-free Nucleophilic ²¹¹At-astatination of Spirocyclic Aryliodonium Ylides

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1. General information

Materials: Commercially available *N*,*N*-dimethylformamide (KANTO CHEMICAL CO., INC, Super Dehydrated grade) were used without further manipulation unless otherwise stated. Aryliodonium ylides **1** were synthesized according to the literature.^[1,2] Astatine-211 was produced from an elemental Bi target via the 209 Bi(α ,2n)²¹¹At nuclear reaction with a 34 MeV alpha beam for 3–4 h at a beam current of 12–13 µA using the NIRS-QST isochronous cyclotron AVF-930 in National Institutes for Quantum & Radiological Science & Technology (QST), Chiba, Japan. ²¹¹At was recovered from the irradiated target in chloroform by using a previously described dry distillation procedure.^[3] All other reagents were commercially available and used as received unless otherwise noted.

Analysis: Radioactivity was quantified using a dose curiemeter IGC-7R, Aloka. Reversed-phase highperformance liquid chromatography (HPLC) was performed on a JASCO LC-2000Plus series PU-2089i gradient pump equipped with a JASCO UV-2075 UV detector and a Universal Giken US-3000 radiation detector. Thin layer chromatography (TLC) was carried out on Merck KGaA F₂₅₄ plates and analyzed on a M&S Instruments Marita Star detector with GINA Star TLC Software.

2. Confirmation of the identity of ²¹¹At-labeled compounds



Synthesis of an authentic sample 2a via electrophilic astatodestannylation of arylstannane S1

To a V-shaped glass vial was added a solution of ²¹¹At (8.9 MBq) in CHCl₃, and then the solvent was evaporated with the gentle flow of N₂ gas at 100 °C. Subsequently, a solution of arylstannane S1^[4] (2 mg) in MeOH/AcOH = 1000:1 (50 μ L) and a solution of NCS (4 mg) in MeOH (10 μ L) were added to the residue. After the reaction at room temperature for 15 min, the reaction mixture was analyzed by radio-HPLC.

The reaction of aryliodonium ylide 1a with ²¹¹At-



To a V-shaped glass vial was added a solution of 211 At (52 MBq) in CHCl₃, and then the solvent was evaporated with the gentle flow of N₂ gas at 100 °C. The residue was dissolved in MeOH (30 µL), followed by the addition of a mixture of aryliodonium ylide **1a** (2 mg) and Et₄NHCO₃ (7 mg) in MeOH (70 µL), which is not fully optimized reaction conditions. After the reaction at 100 °C for 30 min, the reaction mixture was analyzed by radio-HPLC.

HPLC conditions and chromatograms

HPLC analysis was performed on InertSustain C18 (150×4.6 mm, 5 μm) column with flow rate at 1.0 mL/min using 30% H₂O (0.1% HCOOH)/70% MeCN eluent. The injected sample initially passed through the UV detector, followed by the radiation detector, which cause a slight delay between the corresponding UV and radiation peaks. For each analysis, a corresponding non-radioactive I-labeled compound was used as a UV reference. Due to the chemical similarities of iodine and astatine, the difference of retention time between the I-labeled compound and the ²¹¹At-labeled compound would be small. The resulting HPLC chromatograms were shown in below (Figure S1).



Figure S1. UV-HPLC chromatogram of I-labeled standard (top), radio-HPLC chromatogram of the electrophilic astatination of S1 (middle), radio-HPLC chromatogram of the nucleophilic astatination of 1a (bottom)

In the radio-HPLC chromatogram of each reaction, the same radiation peaks were observed at 13.0 to 15.0 min, respectively. In the adjacent to these radiation peaks, the UV peak of I-labeled standard was observed at 14.0 to 16.0 min. These results indicate that these radiation peaks at 13.0 min to 15.0 min would be ²¹¹At-labeled compound **2a**. Therefore, we have concluded that the reaction of aryliodonium ylide **1a** with ²¹¹At⁻ would afford ²¹¹At-labeled compound **2a**.

3. ²¹¹At-astatination of aryliodonium ylide with ²¹¹At-

General procedure for ²¹¹At-astatination of aryliodonium ylide



To a V-shaped glass vial was added a solution of ²¹¹At (16-43 MBq) in CHCl₃, followed by removal of the solvent at 100 °C with the gentle flow of N₂ gas. Then, a solution of aryliodonium ylide **1** (10 mg) in DMF (300 μ L), a solution of Et₄NHCO₃ (7 mg) in DMF (100 μ L), and a solution of PPh₃ (5 mg) in DMF (100 μ L) were successively added to the residue. After the reaction at 100 °C for 30 min under nitrogen atmosphere, the reaction mixture was directly analyzed by radio-HPLC and radio-thin-layer chromatography.

General methods for analysis of ²¹¹At-astatination reactions

The identity of ²¹¹At-labeled compound **2** was confirmed by radio-HPLC. Reversed-phase HPLC analysis was performed on InertSustain C18 ($150 \times 4.6 \text{ mm}$, 5 µm) column with flow rate at 1.0 mL/min using H₂O (0.1% HCOOH)/MeCN eluent. The eluent was changed in response to the product polarity. The injected sample initially passed through the UV detector, followed by the radiation detector, which cause a slight delay between the corresponding UV and radiation peaks.

In the HPLC analysis, the adsorption of unlabeled ²¹¹At⁻ on the column has often become a problem for the calculation of radiochemical yields (RCYs).^[5] Furthermore, insufficient sensitivity of our radiation detector led to an insufficient signal-noise ratio of the analysis. For these reasons, RCYs were calculated as peak area of compound 2/total×100% on radio-TLC analysis. The decay of radioactivity during the reaction has not been considered for the calculation of RCYs since the reaction time was only 30 minutes, which was enough short compared with the half-life of ²¹¹At (7.2 h). Appropriate solvent conditions were determined to separate the free ²¹¹At⁻ (Rf = 0.00-0.25) from the target radiolabeled compounds **2**. For each analysis, corresponding non-radioactive I-labeled compound was used as references due to the chemical similarities of iodine and astatine.

(8R,9S,13S,14S)-3-[²¹¹At]astato-13-methyl-6,7,8,9,11,12,13,14,15,16-decahydro-

17*H*-cyclopenta[a]phenanthren-17-one (2a)

According to the general procedure, the reaction of 8-(((8R,9S,13S,14S)-13-methyl-17-oxo-7,8,9,11,12,13,14,15,16,17-decahydro-6*H*-cyclopenta[a]phenanthren-3-yl)- λ^3 -iodanylidene)-6,10-dioxaspiro[4.5]decane-7,9-dione **1a** with ²¹¹At⁻ (43 MBq) afforded **2a** in 58% RCY.

Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 5:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.35 Retention factor of ²¹¹At-labeled **2a**: 0.31 The trace amount of the least polar material (Rf = 0.55 highlighted in blue) is an unidentified byproduct.

Radio-HPLC chromatography

Eluent: 30% H₂O (0.1% HCOOH)/70% MeCN

Retention time of the corresponding non-radioactive I-labeled compound: 14.0-16.0 min

Retention time of radioactive ²¹¹At-labeled **2a**: 13.0-15.0 min





Reg #2	0.308	57.49	DD	142106.0	58.04
Reg #3	0.550	1.15	DD	2855.0	1.17
Sum in ROI				244839.0	
Total area				247195.0	
Area RF				201859.0	

Figure S2. The result of radio-TLC analysis in the crude mixture of 2a

ethyl 2-[²¹¹At](4-astatophenoxy)-2-methylpropanoate (2b)

According to the general procedure, the reaction of ethyl 2-(4-((7,9-dioxo-6,10-dioxaspiro[4.5]decan-8-ylidene)- λ^3 -iodanyl)phenoxy)-2-methylpropanoate **1b** with ²¹¹At ²¹¹At⁻ (34 MBq) afforded **2b** in 63% RCY.



Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 8:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.45 Retention factor of 211 At-labeled **2b**: 0.50

Radio-HPLC chromatography Eluent: 40% H₂O (0.1% HCOOH)/60% MeCN Retention time of the corresponding non-radioactive I-labeled compound: 13.8-15.2 min Retention time of radioactive ²¹¹At-labeled **2b**: 12.2-13.6 min



Figure S3. The result of radio-TLC analysis in the crude mixture of 2b

methyl (S)-3-[²¹¹At](4-astatophenyl)-2-((*tert*-butoxycarbonyl)amino)propanoate (2c)

According to the general procedure, the reaction of methyl (*S*)-2-((*tert*-butoxycarbonyl)amino)-3-(4-((7,9-dioxo-6,10-dioxaspiro[4.5]decan-8-ylidene)- λ^3 -iodanyl)phenyl)propanoate **1c** with ²¹¹At⁻ (16 MBq) afforded **2c** in 69% RCY.



Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 3:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.33 Retention factor of ²¹¹At-labeled **2c**: 0.45 The trace amount of the least polar material (Rf = 0.73 highlighted in blue) is an unidentified byproduct. Radio-HPLC chromatography

Eluent: 45% H₂O (0.1% HCOOH)/55% MeCN

Retention time of the corresponding non-radioactive I-labeled compound: 14.8-16.2 min Retention time of radioactive ²¹¹At-labeled **2c**: 11.6-12.8 min



Figure S4. The result of radio-TLC analysis in the crude mixture of 2c

2-[²¹¹At](4-(4-astatophenyl)-4-oxobutyl)isoindoline-1,3-dione (2d)

According to the general procedure, the reaction of $2-(4-(4-((7,9-dioxo-6,10-dioxaspiro[4.5]decan-8-ylidene)-\lambda^3-iodanyl)$ phenyl)-4-oxobutyl)isoindoline-1,3-dione **1d** with ²¹¹At⁻ (16 MBq) afforded **2d** in over 99.5% RCY.



Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 3:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.25 Retention factor of 211 At-labeled **2d**: 0.32

Radio-HPLC chromatography Eluent: 45% H₂O (0.1% HCOOH)/55% MeCN Retention time of the corresponding non-radioactive I-labeled compound: 13.5-14.7 min Retention time of radioactive ²¹¹At-labeled **2d**: 7.6-13.0 min



Figure S5. The result of radio-TLC analysis in the crude mixture of 2d

5-[²¹¹At]astato-3-methylbenzo[d]isoxazole (2e)

According to the general procedure, the reaction of 8-((3-methylbenzo[d]isoxazol-5-yl)- λ^3 -iodanylidene)-6,10-dioxaspiro[4.5]decane-7,9-dione **1e** with ²¹¹At⁻ (28 MBq) afforded **2e** in 92% RCY.



Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 8:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.42 Retention factor of 211 At-labeled **2e**: 0.45

Radio-HPLC chromatography Eluent: 55% H₂O (0.1% HCOOH)/45% MeCN Retention time of the corresponding non-radioactive I-labeled compound: 16.4-17.8 min Retention time of radioactive ²¹¹At-labeled **2e**: 14.8-16.6 min



Figure S6. The result of radio-TLC analysis in the crude mixture of 2e

6-[²¹¹At]astatoquinoline (2f)

According to the general procedure, the reaction of 8-(quinolin-6-yl- λ^3 -iodanylidene)-6,10dioxaspiro[4.5]decane-7,9-dione **1f** with ²¹¹At⁻ (24 MBq) afforded **2f** in over 99.5% RCY.



Radio-TLC chromatography Development solvent: hexane/ethyl acetate = 1:1 Retention factor of the corresponding non-radioactive I-labeled compound: 0.38 Retention factor of ²¹¹At-labeled **2f**: 0.50

Radio-HPLC chromatography Eluent: 70% H₂O (0.1% HCOOH)/30% MeCN Retention time of the corresponding non-radioactive I-labeled compound: 11.0-13.0 min

Retention time of radioactive ²¹¹At-labeled **2f**: 6.8-7.8 min

Different basicity of nitrogen atom on iodinated or astatinated quinolines might affect their retention time in the acidic eluent.



Figure S7. The result of radio-TLC analysis in the crude mixture of 2f

5-[²¹¹At]astatobenzo[b]thiophene (2g)

According to the general procedure, the reaction of 8-(benzo[b]thiophen-5-yl- λ^3 - ²¹¹At iodanylidene)-6,10-dioxaspiro[4.5]decane-7,9-dione **1g** with ²¹¹At⁻ (31 MBq) afforded **2g** in 93% RCY.



Radio-TLC chromatography Development solvent: hexane Retention factor of the corresponding non-radioactive I-labeled compound: 0.51 Retention factor of ²¹¹At-labeled **2g**: 0.43

Radio-HPLC chromatography Eluent: 40% H₂O (0.1% HCOOH)/60% MeCN Retention time of the corresponding non-radioactive I-labeled compound: 13.4-14.6 min Retention time of radioactive ²¹¹At-labeled **2g**: 14.2-16.0 min



Figure S8. The result of radio-TLC analysis in the crude mixture of 2g

4. ²¹¹At-astatination of indolyliodonium ylide



According to the general procedure, $8-((1-\text{tosyl-}1H-\text{indol-}5-\text{yl})-\lambda^3-\text{iodanylidene})-6,10-$ dioxaspiro[4.5]decane-7,9-dione **5** was reacted with ²¹¹At⁻ (33 MBq), Et₄NHCO₃, and PPh₃ in DMF at 100 °C for 30 min. The reaction mixture was analysed by radio-TLC, and the result is shown in below (Figure S9).



Figure S9. The result of radio-TLC analysis in the reaction of **5** with ²¹¹At⁻ Development solvent: hexane/ethyl acetate = 8:1 (developed twice) Retention factor of the corresponding non-radioactive I-labeled compound: 0.42

On the TLC plate, different three peaks were observed as the major materials (Rf = 0.24, 0.47, 0.56). The most polar material (Rf = 0.24 highlighted in green) would be unlabeled ²¹¹At⁻. The others with similar polarity (Rf = 0.47 highlighted in red, and Rf = 0.56 highlighted in blue) were positioned near the corresponding non-

radioactive *N*-Ts-5-[¹²⁷I]iodoindole (Rf = 0.42). These results indicated that these two radioactive materials would be regioisomers of ²¹¹At-labeled indole **6**. We propose the following three pathways to afford the regioisomers **6**. First, nucleophilic aromatic substitution reaction of **5** would proceed to afford 5- $[^{211}At]$ astatoindole (**C5-6**) (Scheme S1a). As a second pathway, the aryne formation would initially occur, and then $^{211}At^-$ attacked the aryne to afford C-4 or C-6 labeled compounds (**C4-** or **C6-6**) (Scheme S1b). The formation of aryne from aryliodonium ylide has also been reported in a radiofluorination reaction.^[6] In this report, an electron-rich arene is more likely to form the aryne and converted to the mixture of ¹⁸F-labeled regioisomers. As the other pathway, an electrophilic aromatic substitution reaction at C-2 or C-3 position of indole **5** with ²¹¹At⁺ and decomposition of iodonium ylide might proceed to afford ²¹¹At-labeled compounds (**C2-** or **C3-6**) (Scheme S1c). Although the reaction was performed under reducing conditions, the possibility that ²¹¹At⁻ was oxidized to ²¹¹At⁺ cannot be fully excluded.

Scheme S1. Plausible pathways to afford regioisomers 6

(a) Nucleophilic aromatic substitution reaction



(b) Aryne formation/Nucleophilic aromatic substitution reaction



(c) Electrophilic aromatic substitution reaction



5. ²¹¹At-astatination of benzofuryliodonium ylide



According to the general procedure, 8-(benzofuran-5-yl- λ^3 -iodanylidene)-6,10-dioxaspiro[4.5]decane-7,9dione 7 was reacted with ²¹¹At⁻ (31 MBq), Et₄NHCO₃, and PPh₃ in DMF at 100 °C for 30 min. The reaction mixture was analyzed by radio-TLC and radio-HPLC. These results are shown in below (Figure S10, 11).



Figure S10. Radio-HPLC (top) and UV-HPLC (bottom) chromatogram of the reaction mixture of **7** with ²¹¹At⁻ Eluent: 45% H₂O (0.1% HCOOH)/55% MeCN

Retention time of the corresponding non-radioactive I-labeled compound: 13.8-15.6 min



Figure S11. The result of radio-TLC analysis in the reaction of 7 with ²¹¹At-

Development solvent: hexane

Area RF

Retention factor of corresponding non-radioactive I-labeled compound: 0.46

On the TLC plate, two peaks were observed as major materials (Rf = 0.19, 0.39). The more polar material (Rf = 0.19 highlighted in green) would be unlabeled ²¹¹At⁻. Although peak tailing occurred, the other (Rf = 0.39 highlighted in red) was close to the corresponding non-radioactive 5-[¹²⁷I]iodobenzofuran (Rf = 0.46). In the HPLC chromatogram, the radiation peak (retention time: 12.8-14.2 min) was also observed close to the UV

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peak of the I-labeled standard. These analyses ensure that the reaction of 7 with 211 At⁻ afforded 211 At-labeled benzofuran 8 in ca. 76% RCY.

To examine the identity of peak tailing on the radio-TLC analysis, we developed the same TLC plate again, and the resulting image of radio-TLC analysis is shown in below (Figure S12).



Figure S12. The result of radio-TLC analysis in the reaction of 7 with ²¹¹At⁻ (developed twice) Development solvent: hexane (developed twice)

In the first TLC, the peak tailing was observed (Rf = 0.39 highlighted in red), which suggested that the less polar peak was the major product and the more polar peak was the minor product (**Figure S11**). However, in the second TLC, in which several radiation peaks were observed, the less polar peaks were minor products (Rf= 0.47 highlighted in red, and Rf = 0.56 highlighted in dark green) and the more polar peak was the major product (Rf = 0.37 highlighted in red) (**Figure S12**). If the peak tailing in the first TLC had been due to only the overlapped peaks, the less polar peak would have become the major product and the more polar peak would have become the minor product in the second TLC. These results indicate that the behavior of the compounds were different in the first and second development on the TLC plate. We consider that the less polar ²¹¹Atlabeled benzofuran **8** would be unstable on the silica gel and decomposed into the more polar compounds on the TLC plate.

6. Supplementary references

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