Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry. This journal is © The Royal Society of Chemistry 2021

SUPPORTING INFORMATION

Phosphine-catalysed one-pot domino sequence to access cyclopentene-fused coumarins

Yan-Shan Chen,^a Yu Zheng,^a Zhi-Jun Chen,^b Zhen-Zhen Xie,^a Xian-Chen He,^a Jun-An Xiao,^c Kai Chen,^{*,a,d} Hao-Yue Xiang,^a and Hua Yang^{*,a}

^aCollege of Chemistry and Chemical Engineering, Central South University, Changsha, 410083, P. R. China. E-mail: <u>hyangchem@csu.edu.cn; kaichen@csu.edu.cn</u>

^bGuangdong Zorun Pharmaceutical Research and Development Co. LTD, Guangzhou 510700, P. R. China

°College of Chemistry and Materials Science, Nanning Normal University, Nanning 530001, P. R. China

^dState Key Laboratory of Chemical Oncogenomics, Peking University Shenzhen Graduate School, Shenzhen 518055, P. R. China

Table of Contents

1. General Information	S1
2. General Procedure for the Preparation of 2-formylphenyl but-2-ynoate 1	S2
3. Synthesis of Cyclopenta[c]chromenone 3	S2
4. Mechanistic Studies	S5
5. NMR Spectra Copies for New Compounds of 3	S8
6. Assignment of the Relative Stereochemistry of 3t	S31
7. X-ray Crystallographic Data of Compound	S39
8. References	S42

1. General Information

Unless otherwise noted, all the starting reagents were obtained from commercial suppliers and used without further purification. And it also includes all the solvents that were used. Yields refer to isolated compounds through flash column chromatography performed using 300-400 mesh silica gel. ¹H NMR spectra were recorded at 400 MHz. The chemical shifts were recorded in *ppm* relative to tetramethylsilane and with the solvent resonance as the internal standard. Data were reported as follows: chemical shift (δ *ppm*), multiplicity (*s* = singlet, *d* = doublet, *t* = triplet, *q* = quartet, *dd* = doublet of doublets, *dt* = doublet of triplets, *td* = triplets of doublet, *ddd* = doublet doublet of doublets, *m* = multiplet), coupling constants (Hz), integration. Data for ¹³C NMR spectra are reported in terms of chemical shift and multiplicities, with coupling constants (Hz) in the case of *J*_{CF} coupling. High resolution mass spectroscopy (HRMS) was recorded on TOF MS ES+ Mass spectrometer and acetonitrile was used to dissolve the sample. Infrared spectra (IR) were measured by FT-IR apparatus.

2. General Procedure for the Preparation of 2-formylphenyl but-2ynoate 1

2-formylphenyl but-2-ynoates **1** are prepared according to literatures procedure with minor modification¹. To a stirred solution of but-2-ynoic acid (10 mmol, 1.0 equiv) in anhydrous CH₂Cl₂ (10 mL) is added DMAP (40 mg, 0.03 equiv) and *o*-hydroxyl arylaldehyde (10 mmol, 1.0 equiv). Dicyclohexylcarbodiimide (10 mmol) in 10 mL of anhydrous CH₂Cl₂ is added slowly (within 20 minutes) to the reaction mixture at 0 °C, which is then stirred for 10 minutes at 0 °C and 3 h at room temperature (monitored by TLC). When the reaction reaches a certain limit, precipitated urea was then filtered off and the filtrate evaporated down in vacuum. Column chromatography of the residue (petroleum ether/ethyl acetate = 19/1) gave the crude alkynoates **1** and the pure ones can be obtained by recrystallization from hexanes and CH₂Cl₂.

3. Synthesis of Cyclopenta[c]chromenone 3

3.1 General Procedure



The mixture of alkynoates **1** (0.15 mmol), active methylene compounds **2** (0.3 mmol), $P(p-OCH_3Ph)_3$ (30 mol%) were combined in 1,4-dioxane (1 mL) at room temperature for 8 - 24h. After the reaction, the solvent was evaporated under reduced pressure. The crude product was purified by silica gel chromatography (petroleum ether/ethyl acetate = 17/3) to give the desired products.

		+	Cat.(30 mol %), 24h additive, solvent, T(°C)		
	1a	2a		3a	
entry	catalyst	T(°C)	solvent	additive ^c	yield $(\%)^b$
1	PPh ₃	r.t.	toulene	-	50
2	$P(p-FPh_3)$	r.t.	toulene	-	trace
3	PBu ₃	r.t.	toulene	-	trace
4	PPh ₂ CH ₃	r.t.	toulene	-	trace
5	$P(p-CH_3Ph)_3$	r.t.	toulene	-	53
6	$P(p-OCH_3Ph)_3$	r.t.	toulene	-	55
7	P(p-OCH ₃ Ph) ₃	30	toulene	-	50
8	$P(p-OCH_3Ph)_3$	50	toulene	-	32
9	P(p-OCH ₃ Ph) ₃	80	toulene	-	29
10	$P(p-OCH_3Ph)_3$	r.t.	CH ₃ CN	-	44
11	$P(p-OCH_3Ph)_3$	r.t.	acetone	-	43
12	$P(p-OCH_3Ph)_3$	r.t.	DMSO	-	50
13	P(p-OCH ₃ Ph) ₃	r.t.	1,4-dioxane	-	72
14	$P(p-OCH_3Ph)_3$	r.t.	isopropanol	-	trace
15	$P(p-OCH_3Ph)_3$	r.t.	THF	-	49
16	$P(p-OCH_3Ph)_3$	r.t.	ether	-	51
17	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	NaHCO ₃	52
18	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	K_2CO_3	54
19	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	PhCO ₂ H	70
20	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	Na ₂ CO ₃	56
21	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	DIPEA	60
22	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	AcOH/AcONa	57
23^d	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	-	70
24^{e}	$P(p-OCH_3Ph)_3$	r.t.	1,4-dioxane	-	53

 Table S1. Optimization of Reaction Conditions^a

^{*a*}The reaction was carried out with compound **1a** (0.15 mmol), **2a** (0.3 mmol, 2equiv), PR₃ (0.3 equiv) at room temperature for 24h; ^{*b*}Isolated yield; ^{*c*}0.5 equiv of additive was added; ^{*d*}1.0 equiv of H₂O was added; ^{*e*}under anhydrous conditions.

3.2 Scale-up synthesis of 3a



The mixture of alkynoates **1a** (3.0 mmol, 564.54mg), active methylene compounds **2a** (6.0 mmol, 1.2 mL), P(p-OCH₃Ph)₃ (0.9 mmol, 318 mg, 30 mol%) were combined in 1,4-dioxane (20 mL) at room temperature for 24 h. After the reaction, the solvent was evaporated under reduced pressure. The crude product was purified by silica gel chromatography (petroleum ether/ethyl acetate = 17/3) to give the desired products **3a** (364.5 mg, 45%).

4. Mechanistic Studies



4.1 HRMS studies



4.2 Control Experiments



The mixture of alkynoates **1a** (0.15 mmol), active methylene compounds **2a** (0.3 mmol), base (Et₃N, K₂CO₃, or Na₂CO₃) (1.0 equiv), were combined in 1,4-dioxane (1 mL) at room temperature for 24h. The solvent was then removed under reduced pressure and the residue was purified by chromatography on silica gel (petroleum ether/ethyl acetate = 17/3) to afford a mixture of **4a** and **5a**.²



Figure S2 ESI-MS of the corresponding adducts

4.3 Deuterium Labeling Experiments

The mixture of alkynoates **1a** (0.15 mmol), active methylene compounds **2a** (0.3 mmol), $P(p-OCH_3Ph)_3$ (30mol%), D_2O (0.75 mmol, 5 equiv) were combined in 1,4-dioxane (1 mL) at room temperature for 24 h. The solvent was then removed under reduced pressure and the residue was purified by chromatography on silica gel (petroleum ether/ethyl acetate = 17/3) to afford product **3a'** in 68% yield with 40% deuterium. The ¹H NMR spectra copies of the products above are provided as follows.









5. NMR Spectra Copies for New Compounds of 3

¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3a**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3b**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3c**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3d**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3e**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3f**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3**g



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3h**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3i**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3**j



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product 3k



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3**l



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **30**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3p**





¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3q**

 $\begin{array}{c} 7.840\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.838\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.758\\ 7.738\\ 7.$





¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product 3r

 $\begin{array}{c} 8.003\\ 8.005\\ 8.009\\ 8.005\\ 8.$





¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3s**



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product 3t



The relative stereochemistry of 3u was assigned based on its 2D-NMR spectra and comparison with the spectra of 3w (*major diastereomer*) crystal structures. The NOESY and the 1D ¹H and ¹³C NMR spectra are provided below.

NOESY



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3u**

 $\begin{array}{c} 7.539\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.535)\\ (7.532)$



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product 3v



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of

product 3w (minor diastereomer).

 $\begin{array}{c} 7.428\\ 7.7407\\ 7.7407\\ 7.7407\\ 7.7407\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7286\\ 7.7110\\ 7.7110\\ 7.7286\\$



3w (minor diastereomer).



¹H NMR (400 MHz, CDCl₃), ¹³C NMR (100 MHz, CDCl₃) spectra of product **3w** (*major diastereomer*).







6. Assignment of the Relative Stereochemistry of 3t

The relative stereochemistry of **3t** was assigned based on its 2D-NMR spectra and DFT calculations.

NOESY

In the ¹H-¹H NOESY spectrum, H^a-H^{c'} and H^c-H^d cross signals could be observed. However, there is also a cross signal between H^a and H^{c'} in the COSY spectrum. Since it is difficult to completely remove unwanted COSY-like cross-peaks in NOE spectroscopy, the relative stereochemistry could not be unequivocally assigned by 2D NMR experiments.



COSY





To solve the challenge involved in the stereoassignment of 3t, GIAO NMR shift calculations were then conducted to provide helpful assistance in the structural elucidation.³ First, the most stable conformations of isomers 1 and 2 were obtained after a comprehensive conformational search. Actually, it was found that the distance between H^c and H^d in **3t-isomer-1** is 2.26 Å, even shorter than that in **3t-isomer-2**, 2.41 Å (Figure S4).



Figure S4. The most stable conformations of **3t-isomer-1** and **3t-isomer-2**. Structures are optimized at M062X/6-311+G(2d,p) theoretical level using Gaussian09 software.⁴ Distances were given in angstrom.

Comparing the experimental and calculated isotropic magnetic shielding values, it is suggested that **3t-isomer-1** is assigned to be the plausible configuration (Table S2). To be noted, the NMR chemical shifts for C¹, C⁵, H¹, and H⁸ of **3t-isomer-1** are obviously closer to the corresponding experimental values than those of 3t-isomer-2. What is more, the RMSDs with respect to experimental NMR chemical shifts for 3t-isomer-1 are 3.90 and 0.181 ppm for ¹³C and ¹H, respectively, lower than the corresponding values for **3t-isomer-2**.

Table S2. Comparison of the Boltzmann averaged GIAO isotropic magnetic shielding values and experimental results.5



3t-isomer-1

Wrong	
3t-isomer-2	

Nuclei	3t-isomer-1	3t-isomer-2	Exp (¹³ C)
¹³ C			
1	49.56	53.35	48.10
3	143.19	144.58	158.30
4	47.38	46.17	44.90
5	61.94	56.35	61.70
6	189.83	187.90	190.50

Table S2 continued.

	156.07	155.50	
	150.70	150.11	151.10
	135.32	133.96	139.30
	132.52	132.81	135.00
	132.14	129.72	133.30
	130.78	129.55	130.40
	129.97	131.89	
	128.65	128.72	129.80
	128.47	127.96	129.60
	128.19	126.67	129.20
	127.39	125.62	127.60
	123.58	123.52	125.00
	121.60	123.17	120.10
	120.88	121.30	118.70
	116.80	116.93	117.70
RMSD	3.90	4.31	

Nuclei	3t-isomer-1	3t-isomer-2	Exp (¹ H)
$^{1}\mathrm{H}$			
1	5.07	4.88	5.40
3	6.77	6.76	7.02
4	3.51	3.56	3.70
4	3.00	3.42	3.21
7	8.38	8.24	8.23
8	8.10	7.51	8.23
	7.57	7.50	7.72
	7.46	7.26	7.60
	7.39	7.33	7.60
	7.15	7.15	7.33
	7.14	7.07	7.15
	7.04	6.92	7.15
	6.99	6.76	7.05
RMSD	0.181	0.320	

^aThe GIAO isotropic magnetic shielding values were calculated at the mPW1PW91/6-311+G(2d,p) theoretical level in chloroform with IEFPCM solvent model using G09 software.⁶ The contributions of different conformations were taken into account according to the Boltzmann distribution principle.

Structure	Eele	Eo	Ε	Н	G
3t-isomer-1-conf1	-1050.252201	-1049.969899	-1049.951444	-1049.950500	-1050.017370
3t-isomer-1-conf2	-1050.247515	-1049.964792	-1049.946444	-1049.945500	-1050.012241
3t-isomer-1-conf3	-1050.242860	-1049.960956	-1049.942426	-1049.941482	-1050.008714
3t-isomer-2-conf1	-1050.249093	-1049.967053	-1049.948632	-1049.947688	-1050.014426
3t-isomer-2-conf2	-1050.248106	-1049.966421	-1049.948103	-1049.947159	-1050.013104
3t-isomer-2-conf3	-1050.244905	-1049.963098	-1049.944685	-1049.943741	-1050.009951
3t-isomer-2-conf4	-1050.243597	-1049.961704	-1049.943134	-1049.942190	-1050.009585
3t-isomer-2-conf4	-1050.243597	-1049.961704	-1049.943134	-1049.942190	-1050.009585

Table S3. Energies and thermodynamic parameters.

Notes: Eele, E0, E, H, and G were the electronic energies, sum of electronic and zero-point energies, sum of electronic and thermal energies, sum of electronic and thermal enthalpies, and sum of electronic and thermal free energies, respectively, which were given at the M062X/6-311+G(2d,p) theoretical level using Gaussian09 software

Coordinates of all stationary points.

3t-isomer-1-conf1

0 ima	ginary freque	ency		С	2
С	-4.201050	-1.390231	0.180528	С	3
С	-3.260090	-0.417514	-0.118990	С	3
С	-1.910866	-0.730795	-0.256534	С	5
С	-1.524993	-2.057965	-0.097272	Н	3
С	-2.453821	-3.040555	0.204049	С	4
С	-3.794541	-2.703313	0.343617	Н	2
Н	-5.237454	-1.095648	0.278705	С	5
Н	-0.480904	-2.320241	-0.219205	Н	5
Н	-2.132215	-4.065892	0.328475	Н	5
Н	-4.527093	-3.464648	0.578831	Н	6
С	-0.959333	0.376414	-0.611198		
Н	-0.770158	0.340145	-1.690529	3t-isoı	ner
С	-1.538860	1.714502	-0.243928	0 imag	gina
С	-2.991265	1.968522	-0.296224	С	3
0	-3.774230	0.857638	-0.304567	С	2
0	-3.499569	3.047930	-0.320453	С	1
С	0.437559	0.428168	0.080241	С	1
С	-0.622424	2.584580	0.155313	С	2
Н	-0.825042	3.609650	0.435530	С	3
С	0.746188	1.963226	0.156316	Н	4
Н	1.337228	2.226602	1.031949	Н	0
Н	1.317626	2.271584	-0.726402	Н	2
С	1.518845	-0.337518	-0.722233	Н	4
0	1.188698	-0.949789	-1.705669	С	0
С	0.340544	-0.106425	1.445322	Н	-0

N	0.278779	-0.503266	2.519129
С	2.954024	-0.240586	-0.317945
С	3.901568	-0.647300	-1.259701
С	3.378873	0.227087	0.925829
С	5.252179	-0.575020	-0.968520
Н	3.555086	-1.014043	-2.217267
С	4.734033	0.287369	1.218167
Н	2.668412	0.516445	1.687933
С	5.669873	-0.107032	0.272637
Н	5.981159	-0.885479	-1.706089
Н	5.057439	0.638538	2.189440
Н	6.726612	-0.054071	0.503405

:-1-conf2

) ima	ginary freque	ncy	
С	3.959035	0.349863	-1.324124
С	2.815880	-0.256418	-0.828330
С	1.703313	0.491883	-0.456089
С	1.753344	1.874277	-0.605287
С	2.890545	2.493282	-1.098567
С	3.993731	1.727894	-1.456671
Н	4.800620	-0.272448	-1.598355
Н	0.894122	2.470970	-0.318614
Н	2.916426	3.570039	-1.199082
Н	4.886464	2.204396	-1.840530
С	0.484773	-0.255298	0.012457
Н	-0.161897	-0.392033	-0.861934

С	0.872949	-1.596631	0.565827
С	1.977378	-2.370898	-0.036463
0	2.855654	-1.643130	-0.775925
0	2.142575	-3.545863	0.089324
С	-0.387142	0.350894	1.175575
С	0.198769	-1.942686	1.654325
Н	0.341625	-2.873913	2.186944
С	-0.762233	-0.873319	2.085817
Н	-0.638906	-0.617971	3.138495
Н	-1.802716	-1.176772	1.954649
С	-1.590251	1.137312	0.616558
0	-1.629458	2.332938	0.751088
С	0.423826	1.253600	2.003593
Ν	1.063419	1.896805	2.702871
С	-2.685307	0.415167	-0.116483
С	-3.792983	1.183182	-0.480373
С	-2.654254	-0.938085	-0.461649
С	-4.848947	0.613205	-1.170545
Η	-3.803724	2.231164	-0.210471
С	-3.710700	-1.505304	-1.159755
Η	-1.814505	-1.570011	-0.202035
С	-4.808661	-0.732993	-1.512929
Η	-5.703937	1.218229	-1.444019
Н	-3.673754	-2.553397	-1.427685
Н	-5.632318	-1.181007	-2.054684

С	0.381829	1.022505	0.874514
С	-0.705211	2.727669	-0.426918
Н	-1.079148	3.700361	-0.716747
С	0.545801	2.502357	0.377918
Н	0.679505	3.186134	1.214907
Н	1.429905	2.581021	-0.266517
С	1.801037	0.491843	1.178141
0	2.291487	0.762457	2.241914
С	-0.390433	1.051816	2.126207
Ν	-1.018591	1.087357	3.083257
С	2.620182	-0.177989	0.117130
С	2.195375	-1.292687	-0.603981
С	3.910098	0.316740	-0.074171
С	3.053150	-1.896015	-1.513334
Н	1.213034	-1.719179	-0.440100
С	4.752514	-0.270634	-1.004085
Н	4.240544	1.161004	0.518598
С	4.324467	-1.379059	-1.723977
Н	2.726452	-2.772828	-2.057639
Н	5.746339	0.129385	-1.159480
Н	4.984972	-1.844843	-2.444491

3t-isomer-1-conf3

0 imaginary frequency

С	-3.507071	-1.849445	-0.320493
С	-2.620164	-0.810770	-0.554661
С	-1.387024	-0.752776	0.088419
С	-1.071264	-1.755350	1.000423
С	-1.948089	-2.800492	1.245203
С	-3.165264	-2.847310	0.577199
Н	-4.453969	-1.850919	-0.843951
Н	-0.133825	-1.704272	1.545585
Н	-1.688018	-3.567126	1.962712
Н	-3.858807	-3.656636	0.765200
С	-0.474155	0.387951	-0.255102
Н	0.205562	0.082344	-1.058953
С	-1.262035	1.570424	-0.755842
С	-2.498730	1.384033	-1.536560
0	-3.039880	0.133043	-1.477860
0	-3.028167	2.220397	-2.201317

3t-isomer-2-conf1

0 imaginary frequency

С	-3.595360	1.823652	-0.570974
С	-2.927703	0.653957	-0.242601
С	-1.782097	0.674797	0.544286
С	-1.320371	1.900620	1.010626
С	-1.975116	3.078296	0.687412
С	-3.115386	3.036632	-0.106033
Н	-4.480878	1.760127	-1.189615
Н	-0.434832	1.929085	1.637035
Н	-1.599585	4.023943	1.055627
Н	-3.634818	3.950536	-0.363491
С	-1.141505	-0.634604	0.898557
Н	-1.449107	-0.895039	1.921093
С	-1.571950	-1.744490	-0.017268
С	-2.872522	-1.715308	-0.712988
0	-3.501416	-0.509683	-0.733023
0	-3.381894	-2.652104	-1.249059
С	0.408574	-0.774854	0.857103
С	-0.643154	-2.680223	-0.148090
Н	-0.748862	-3.580816	-0.737414
С	0.600153	-2.316477	0.621904

Н	1.518823	-2.537509	0.076868
Н	0.647639	-2.844889	1.579790
С	1.016260	-0.066189	-0.383310
0	0.286663	0.244375	-1.286637
С	0.993146	-0.307772	2.114835
Ν	1.402830	0.060219	3.120747
С	2.495373	0.142092	-0.476679
С	2.946422	0.967723	-1.507738
С	3.421274	-0.457419	0.376818
С	4.300275	1.200009	-1.675748
Н	2.215370	1.418725	-2.166320
С	4.778638	-0.232599	0.197977
Н	3.105856	-1.102449	1.186134
С	5.218648	0.597659	-0.823217
Н	4.641897	1.849007	-2.471822
Н	5.491789	-0.702830	0.862523
Н	6.278478	0.776847	-0.955111

3t-isomer-2-conf2

0 imaginary frequency -0.167381 С 0.278786 3.143026 С -0.9279721.979918 0.317615 С -0.441682 0.825931 0.921261 С 0.798287 0.883616 1.556050 С 1.564332 2.034100 1.527786 С 1.081387 3.163073 0.871838 Н -0.583292 4.015415 -0.207758 Н 1.169345 -0.000726 2.062050Η 2.532563 2.051971 2.010395 Н 1.673430 4.068871 0.839128 С -1.282856 -0.416474 0.911742 Н -1.603997 1.939148 -0.633006 С -2.493570 0.027544 -0.272726 С -3.043709 1.036034 -0.362909 0 -2.190866 2.092908 -0.234903 0 -4.129117 1.219420 -0.822041 С -0.680730 -1.770662 0.355938 С -2.887389 -1.434981 -0.476727 Н -3.729510 -1.565823 -1.142431 С -1.968991 -2.543970-0.041235 Н -1.775151 -3.271072 -0.827019Н -2.376048 -3.069198 0.828837 С 0.150778-1.483225 -0.932913 0 -0.387449 -1.636885 -1.997566

С	0.068966	-2.475800	1.393561
Ν	0.608325	-3.033877	2.238106
С	1.520975	-0.890458	-0.857984
С	1.752143	0.221411	-1.668836
С	2.543078	-1.388249	-0.053896
С	2.979836	0.859631	-1.639006
Н	0.951370	0.586885	-2.299843
С	3.782086	-0.760869	-0.051305
Н	2.391138	-2.264574	0.560161
С	3.996781	0.368940	-0.827545
Н	3.146008	1.738770	-2.248553
Н	4.579589	-1.157985	0.563658
Н	4.959524	0.864480	-0.807015

3t-isomer-2-conf3

0 imaginary frequency

С	2.061900	-2.433078	0.450085
С	1.114024	-1.558576	0.960256
С	-0.171456	-1.503908	0.431047
С	-0.504681	-2.374168	-0.603332
С	0.432515	-3.255154	-1.119642
С	1.719604	-3.276334	-0.594242
Н	3.052527	-2.436201	0.885340
Н	-1.507122	-2.345030	-1.017017
Н	0.160009	-3.919396	-1.928930
Н	2.459251	-3.958452	-0.993707
С	-1.145595	-0.517515	1.007500
Н	-1.853700	-1.055627	1.650517
С	-0.449974	0.538946	1.819259
С	0.835359	0.284083	2.495046
0	1.537202	-0.787174	2.029395
0	1.299884	0.952675	3.366636
С	-2.000902	0.348057	0.012288
С	-1.051410	1.718016	1.746750
Н	-0.721452	2.608561	2.265658
С	-2.238245	1.661904	0.822198
Н	-2.343597	2.534237	0.176827
Н	-3.169406	1.578683	1.391113
С	-1.235625	0.555831	-1.326352
0	-1.694708	0.097093	-2.337447
С	-3.274020	-0.315262	-0.288326
Ν	-4.279166	-0.840333	-0.450978
С	0.122129	1.198339	-1.341435
С	1.152263	0.445305	-1.903904

С	0.387976	2.485582	-0.880812
С	2.439189	0.957559	-1.964464
Н	0.937592	-0.550580	-2.273278
С	1.671424	3.005059	-0.967859
Н	-0.399363	3.104137	-0.473084
С	2.700943	2.237770	-1.495596
Н	3.236421	0.355476	-2.381381
Н	1.866729	4.010838	-0.618698
Н	3.704740	2.640208	-1.545987

Н	-0.207185	1.393390	-0.108774
С	-3.320660	2.648851	-0.478578
Н	-5.079213	1.498376	-0.926681
Н	-1.403491	3.526355	-0.076486
Н	-3.845640	3.595684	-0.463764

3t-isomer-2-conf4

0 imaginary frequency

С	2.983239	1.391732	-1.365809
С	2.349725	0.763373	-0.305415
С	1.605307	-0.398380	-0.493872
С	1.524633	-0.935047	-1.775378
С	2.152759	-0.314745	-2.844741
С	2.880060	0.850455	-2.637203
Н	3.546906	2.294876	-1.172038
Н	0.962524	-1.848710	-1.933432
Н	2.074344	-0.741657	-3.835637
Н	3.371926	1.341090	-3.467229
С	0.976656	-1.031926	0.714498
Н	1.651379	-1.823052	1.065375
С	0.794616	-0.033103	1.822775
С	1.750427	1.073113	2.015509
0	2.528830	1.362944	0.932313
0	1.867177	1.723061	3.008285
С	-0.449579	-1.678875	0.588414
С	-0.323491	-0.214658	2.510358
Η	-0.630783	0.391381	3.352574
С	-1.124600	-1.361759	1.967071
Η	-2.181831	-1.115301	1.850279
Н	-1.066631	-2.229602	2.629544
С	-1.341826	-1.147902	-0.574720
0	-1.639005	-1.895846	-1.467213
С	-0.305727	-3.127186	0.398239
Ν	-0.144824	-4.257886	0.308177
С	-1.967176	0.214929	-0.505587
С	-3.337946	0.263827	-0.766460
С	-1.276233	1.397484	-0.255992
С	-4.013635	1.472803	-0.737904
Н	-3.861000	-0.657045	-0.993027
С	-1.952142	2.610039	-0.252710

7. X-ray Crystallographic Data of Compound

X-Ray crystallographic analysis of **3a** (CCDC 2080977) showing the thermal ellipsoids at 30% probability level.



Bond precision: C-C = 0.0030 A Wavelength=0.71076			gth=0.71076	
Cell:	a=10.2334(13) alpha=115.446(4)	b=12.1426(15) beta=93.730(4)	c=12.2272(15) gamma=103.262(4)	
Temperature:	293 K			
	Calculated	Reporte	ed	
Volume	1312.0(3)	1312.0	(3)	
Space group	P -1	P -1		
Hall group	-P 1	-P 1		
Moiety formula	C16 H14 O4	2(C16 H	114 04)	
Sum formula	C16 H14 O4	C32 H28	3 08	
Mr	270.27	540.54		
Dx,g cm-3	1.368	1.368		
Z	4	2		
Mu (mm-1)	0.098	0.098		
F000	568.0	568.0		
F000'	568.32			
h,k,lmax	13,15,15	13,15,1	5	
Nref	6056	6038		
Tmin,Tmax	0.964,0.994	0.707,0	.746	
Tmin'	0.961			
Correction method= # Reported T Limits: Tmin=0.707 Tmax=0.746 AbsCorr = MULTI-SCAN				
Data completeness= 0.997 Theta(max)= 27.560				
R(reflections)=	= 0.0494(3735)	wR2(reflections	s)= 0.1323(6038)	
S = 1.017	Npar=	366		

X-Ray crystallographic analysis of **3v** (CCDC 2081221) showing the thermal ellipsoids at 30% probability level.



X-Ray crystallographic analysis of **3w** (*major diastereomer*) (CCDC 2080106) showing the thermal ellipsoids at 30% probability level.



Bond precision:	C-C = 0.0025 A	Wavelength	n=0.71073	
Cell:	a=9.9700(3) alpha=90	b=11.9190(3) beta=90	c=12.5827(4) gamma=90	
Temperature:	296 K		9	
	Calculated	Reported		
Volume	1495.23(8)	1495.23(8)	
Space group	P n a 21	Pna2(1)		
Hall group	P 2c -2n	?		
Moiety formula	C17 H16 O5	?		
Sum formula	C17 H16 O5	C1.55 H1	.45 00.45	
Mr	300.30	27.30		
Dx,g cm-3	1.334	1.334		
Z	4	44		
Mu (mm-1)	0.098	0.098		
F000	632.0	632.0		
F000'	632.37			
h,k,lmax	13,15,16	13,15,15		
Nref	3548[1854]	3498		
Tmin, Tmax	0.977,0.981			
Tmin'	0.971			
Correction method= Not given				
Data completeness= 1.89/0.99		Theta(max) = 27.830		
R(reflections)=	0.0407(3242)	wR2(reflections)	= 0.1239(3498)	
S = 1.018	Npar= 1	199		

8. References

- (a) X. Han and X. Lu, Org. Lett., 2010, 12, 108-111. (b) Z.-X. Deng, Y. Zheng, Z.-Z. Xie, Y.-H.
 Gao, J.-A. Xiao, S.-Q. Xie, H.-Y. Xiang, X.-Q. Chen and H. Yang, Org. Lett., 2020, 22, 488-492.
- 2. (a) E. Yoshioka, S. Kohtani and H. Miyabe, *Angew. Chem. Int. Ed.*, 2011, **50**, 6638-6642; (b) Y.
 Zhou, Y. Chen and Y. Huang, *Org. Lett*, 2020, **22**, 5941-5946.
- J. R. Cheeseman, G. W. Trucks, T. A. Keith and M. J. Frisch, J. Chem. Phys., 1996, 104, 5497-5509.
- (a) Y. Zhao and D. G. Truhlar, *Theor Chem Acc*, 2008, **120**, 215-241; (b) Gaussian 09, Revision A.02, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2016.
- (a) M. W. Lodewyk, C. Soldi, P. B. Jones, M. M. Olmstead, J. Rita, J. T. Shaw and D. J. Tantillo, *J. Am. Chem. Soc.*, 2012, **134**, 18550-18553; (b) M. W. Lodewyk, M. R. Siebert and D. J. Tantillo, *Chem. Rev.*, 2012, **112**, 1839-1862.
- 6. C. Adamo and V. Barone, J. Chem. Phys., 1998, 108, 664-675.