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Supplementary Information

for

n-Butyllithium as a Highly Efficient Precatalyst for Cyanosilylation

of Aldehydes and Ketones

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General Information

All manipulations of air-sensitive materials were carried out by using modified Schlenk line under an atmosphere of dry argon or glovebox techniques under nitrogen-atmosphere. Solvents of THF, DME, CHCl₃, CH₃CN and toluene were dried and freed of oxygen by refluxing over sodium/benzophenone ketyl and distilled prior to use. Trimethylsilyl cyanide (Me₃SiCN) was purchased from Macklin and used without further purification. All the liquid ketones and aldehydes were dried over CaH₂, freshly distilled, and degassed prior to use. All the solid ketones and aldehydes were degassed prior to use. Mesitylene was used for the clarification of product yield. CDCl₃ and THF- d_8 were purchased from TCI and stored over activated 4Å molecular sieves. ¹H and ¹³C{¹H} spectra were recorded on Bruker AV-400 MHz and referenced to the resonances of the solvent used. Chemical shifts of the cyanohydrin products were reported as parts per million in δ scale using residual solvent signal as internal standard (CDCl₃ referenced at δ 7.26 in ¹H NMR and δ 77.0 for central line of the triplet in ¹³C NMR) and compared to literature values. Data are represented as follows: chemical shift, integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad) and coupling constant (J, Hz). Fourier infrared spectroscopy was recorded on Bruker Vertex 70 spectrophotometer. Note: the catalytic cyanosilylation of aldehydes and ketones by n-BuLi is an exothermic reaction. Large-scale reactions should be carried out under cooling in an ice bath, and *n*-BuLi should be added in a slow dropwise manner.

General Catalytic Procedure for the Cyanosilylation of Aldehydes and Ketones

Aldehyde or ketones (1.0 mmol), Me₃SiCN (1.1 mmol) and *n*-BuLi (0.01-0.05 mol %, 0.0367 M) were charged in a Schlenk tube with a magnetic bead inside the glove box. The reaction mixture was allowed to stir at room temperature for 1 hour. Upon completion of reaction, the vial was moved out of the glove box and the reaction was exposed to the air. Subsequently, a drop of solution was added to the above J. Young tap NMR tube. The progress of the reaction was monitored by ¹H and ¹³C NMR spectroscopy.

Characterization Data (¹H and ¹³C NMR Spectra) for Corresponding Cyanosilylated Products of Aldehydes and Ketones:



2-phenyl-2-((trimethylsilyl)oxy)acetonitrile(3a)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.43 – 7.28 (m, 5H), 5.42 (s, 1H), 0.16 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 136.35, 129.37, 128.97, 126.39, 63.69, -0.22.



2-(4-fluorophenyl)-2-((trimethylsilyl)oxy)acetonitrile(3b)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.46 (dd, *J* = 8.6, 5.2 Hz, 2H), 7.10 (t, J = 8.6 Hz, 2H), 5.47 (s, 1H), 0.23 (s, 10H). ¹³C NMR (CDCl₃, 101 MHz): δ 164.67, 162.20, 132.56, 128.59, 119.26, 116.35, 63.28, 0.00, 0.00.



2-(4-chlorophenyl)-2-((trimethylsilyl)oxy)acetonitrile(3c)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.42 – 7.34 (m, 5H), 5.45 (s, 1H), 0.22 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 135.28, 134.93, 129.15, 127.71, 118.87, 62.97, -0.28.



2-(4-bromophenyl)-2-((trimethylsilyl)oxy)acetonitrile(3d)⁶: ¹H NMR (CDCl₃, 400 MHz): δ 7.55 (d, *J* = 8.5 Hz, 2H), 7.35 (d, J = 8.3 Hz, 2H), 5.45 (s, 1H), 0.24 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 132.13, 127.98, 63.04, -0.24.



2-(p-tolyl)-2-((trimethylsilyl)oxy)propanenitrile(3e)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.46 (d, *J* = 8.9 Hz, 2H), 6.90 (d, *J* = 8.9 Hz, 2H), 3.79 (s, 3H), 1.83 (s, 3H), 0.15 (s, 9H). ¹³C NMR (CDCl₃,

101 MHz): δ 139.57, 133.66, 129.81, 126.62, 119.53, 63.81, 21.42, 0.00.



2-(4-methoxyphenyl)-2-((trimethylsilyl)oxy)acetonitrile(3f)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.39 (d, *J* = 8.7 Hz, 2H), 6.93 (d, *J* = 8.7 Hz, 2H), 5.44 (s, 1H), 3.82 (s, 3H), 0.21 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 160.38, 128.52, 127.97, 119.40, 114.30, 63.38, 55.35, -0.19.



4-phenyl-2-((trimethylsilyl)oxy)but-3-enenitrile(3g)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.44 – 7.29 (m, 5H), 6.82 (d, *J* = 15.8 Hz, 1H), 6.20 (dd, *J* = 15.8, 6.0 Hz, 1H), 5.13 (d, *J* = 6.0 Hz, 1H), 0.27 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 135.10, 133.93, 128.81, 127.03, 123.67, 118.51, 62.27, -0.06.



2-(thiophen-2-yl)-2-((trimethylsilyl)oxy)acetonitrile(3h)⁴: ¹H NMR (CDCl₃, 400 MHz): δ 7.36 (d, *J* = 5.1 Hz, 1H), 7.19 (d, *J* = 3.6 Hz, 1H), 7.02 – 6.97 (m, 1H), 5.73 (s, 1H), 0.24 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 139.82, 127.49, 127.20, 126.60, 118.59, 59.87, -0.00.



2-(pyridin-2-yl)-2-((trimethylsilyl)oxy)acetonitrile(3i)⁹: ¹H NMR (CDCl₃, 400 MHz): δ 8.32 (s, 1H), 7.52 (t, *J* = 7.7 Hz, 1H), 7.33 (d, *J* = 7.8 Hz, 1H), 7.04 (d, *J* = 4.4 Hz, 1H), 5.34 (d, *J* = 2.0 Hz, 1H), 0.02 - 0.02 (m, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 155.78, 149.71, 137.84, 124.34, 120.82, 119.05, 65.47, 0.00.

2-(furan-2-yl)-2-((trimethylsilyl)oxy)acetonitrile(3j)¹: ¹H NMR (CDCl₃, 400 MHz): δ 7.50 (s, 1H), 6.60 (d, *J* = 3.3 Hz, 1H), 6.46 (s, 1H), 5.58 (s, 1H), 0.28 (s, 9H).

¹³C NMR (CDCl₃, 101 MHz): δ 148.25, 143.91, 117.18, 110.85, 109.78, 57.48, -0.35.

OSiMe₃ CN H NO₂

2-(2-nitrophenyl)-2-((trimethylsilyl)oxy)acetonitrile(3k)⁶: ¹H NMR (CDCl₃, 400 MHz): δ 8.09 (t, *J* = 10.5 Hz, 1H), 7.97 (t, *J* = 8.0 Hz, 1H), 7.73 (q, *J* = 7.7 Hz, 1H), 7.55 (q, *J* = 8.1 Hz, 1H), 6.17 (s, 1H), 0.24 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 146.91, 134.89, 132.60, 130.66, 129.00, 125.77, 118.31, 60.64, 0.00.



4-(cyano((trimethylsilyl)oxy)methyl)benzonitrile(31)²: ¹H NMR (CDCl₃, 400 MHz): δ 7.66 (d, *J* = 5.1 Hz, 2H), 7.56 (d, *J* = 4.8 Hz, 2H), 5.56 (s, 1H), 0.19 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 141.26, 132.70, 126.89, 118.11, 113.22, 67.82, 62.76, 25.56, -0.41.



2-(anthracen-9-yl)-2-((trimethylsilyl)oxy)acetonitrile(3m)⁵: ¹H NMR (CDCl₃, 400 MHz): δ 8.55 – 8.48 (m, 3H), 8.04 (d, *J* = 8.5 Hz, 2H), 7.67 – 7.59 (m, 2H), 7.55 – 7.48 (m, 2H), 6.94 (s, 1H), 0.11 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 131.66, 130.76, 129.59, 127.38, 126.16, 125.44, 119.95, 58.26, 0.00.



2-(3-((cyano((trimethylsilyl)oxy)methyl)phenyl)-2-((trimethylsilyl)oxy)acetonitrile(3n)⁵: ¹H NMR (CDCl₃, 400 MHz): δ 7.54 (s, 1H), 7.49 – 7.40 (m, 3H), 5.51 (s, 2H), 0.20 (s, 19H). ¹³C NMR (CDCl₃, 101 MHz): δ 137.66, 129.93, 127.44, 124.41, 119.16, 63.53, 0.00.



2-cyclohexyl-2-((trimethylsilyl)oxy)acetonitrile(3o)⁴: ¹H NMR (CDCl₃, 400 MHz): δ 4.15 (s, 1H), 1.90 – 1.75 (m, 4H), 1.72 – 1.59 (m, 2H), 1.25 – 1.05 (m, 4H), 0.36 (s, 1H), 0.19 (s, 9H). ¹³C

NMR (CDCl₃, 101 MHz): δ 119.89, 43.39, 28.59, 28.37, 26.49, 25.98, -0.00.

2-cyclopropyl-2-((trimethylsilyl)oxy)acetonitrile(3p): ¹H NMR (CDCl₃, 400 MHz): δ 4.07 (d, *J* = 6.8 Hz, 1H), 1.35 – 1.17 (m, 1H), 0.62 (d, *J* = 8.0 Hz, 2H), 0.44 (d, *J* = 4.8 Hz, 2H), 0.17 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 119.45, 64.94, 16.41, 3.88, 2.43, 0.00.

2-((trimethylsilyl)oxy)hexanenitrile(3q): ¹H NMR (CDCl₃, 400 MHz): δ 4.37 (t, *J* = 6.6 Hz, 1H), 1.84 – 1.71 (m, 2H), 1.49 – 1.28 (m, 4H), 0.91 (t, *J* = 7.2 Hz, 3H), 0.19 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 120.51, 36.36, 27.05, 22.48, 14.23, 0.00.



2-((trimethylsilyl)oxy)tridecanenitrile(3r): ¹H NMR (CDCl₃, 400 MHz): δ 4.37 (t, *J* = 6.6 Hz, 1H), 1.81 – 1.71 (m, 2H), 1.25 (s, 18H), 0.87 (t, *J* = 6.9 Hz, 3H), 0.19 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 120.49, 61.89, 36.65, 32.31, 30.00, 29.88, 29.78, 29.73, 24.96, 23.08, 14.50, 0.00.

OSiMe₃

2-phenyl-2-((trimethylsilyl)oxy)propanenitrile(4a)²: ¹H NMR (CDCl₃, 400 MHz): δ 7.56 (dd, *J* = 8.3, 1.4 Hz, 2H), 7.43 – 7.32 (m, 3H), 1.87 (s, 3H), 0.19 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ142.03, 128.70, 124.63, 121.64, 71.64, 33.60, 1.09.



2-(4-fluorophenyl)-2-((trimethylsilyl)oxy)propanenitrile(4b): ¹H NMR (CDCl₃, 400 MHz): δ 7.52 (dd, *J* = 8.9, 5.1 Hz, 2H), 7.07 (t, *J* = 8.7 Hz, 2H), 1.84 (s, 3H), 0.18 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 162.88, 160.42, 136.91, 125.52, 120.39, 114.39, 70.00, 32.53, -0.00.



2-(4-chlorophenyl)-2-((trimethylsilyl)oxy)propanenitrile(4c)⁹: ¹H NMR (CDCl₃, 400 MHz): δ 7.48 (d, *J* = 8.6 Hz, 2H), 7.36 (d, *J* = 8.6 Hz, 2H), 1.83 (s, 3H), 0.19 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 139.65, 133.54, 127.77, 125.02, 120.18, 70.01, 32.46, 0.00.



2-(4-nitrophenyl)-2-((trimethylsilyl)oxy)propanenitrile(4d)²: ¹H NMR (CDCl₃, 400 MHz): δ 8.24 (d, *J* = 8.8 Hz, 2H), 7.72 (d, *J* = 8.8 Hz, 2H), 1.85 (s, 3H), 0.21 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 148.94, 147.99, 125.76, 123.93, 120.64, 70.92, 33.31, 1.00.



2-(p-tolyl)-2-((trimethylsilyl)oxy)propanenitrile(4e)²: ¹H NMR (CDCl₃, 400 MHz): δ 7.46 (d, *J* = 8.9 Hz, 2H), 6.90 (d, *J* = 8.9 Hz, 2H), 3.79 (s, 3H), 1.83 (s, 3H), 0.15 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 138.01, 137.44, 128.18, 127.36, 123.51, 120.68, 70.44, 32.44, 19.98, -0.00.



2-(4-methoxyphenyl)-2-((trimethylsilyl)oxy)propanenitrile(4f)²: ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (d, *J* = 6.8 Hz, 2H), 6.80 (d, *J* = 8.6 Hz, 3H), 3.73 (s, 3H), 2.41 (s, 3H), 0.24 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 198.35, 165.43, 161.76, 132.44, 127.93, 115.58, 57.29, 28.13, -0.00.

2-(thiophen-2-yl)-2-((trimethylsilyl)oxy)propanenitrile(4g)⁷: ¹H NMR (CDCl₃, 400 MHz): δ 7.31 – 7.28 (m, 1H), 7.21 – 7.18 (m, 1H), 6.96 (dd, *J* = 5.1, 3.6 Hz, 1H), 1.97 (s, 3H). ¹³C NMR (CDCl₃, 101 MHz): δ 145.50, 125.81, 125.17, 123.90, 120.03, 67.46, 32.61, 0.00.



2,2-diphenyl-2-((trimethylsilyl)oxy)acetonitrile(4h)²: ¹H NMR (CDCl₃, 400 MHz): δ 7.60 (d, *J* = 6.9 Hz, 5H), 7.44 – 7.34 (m, 8H), 0.24 (s, 10H). ¹³C NMR (CDCl₃, 101 MHz): δ 142.04, 128.65, 125.97, 120.79, 1.03.



2-(naphthalen-2-yl)-2-((trimethylsilyl)oxy)propanenitrile(4i)⁸: ¹H NMR (CDCl₃, 400 MHz): δ 7.85 (s, 1H), 7.71 – 7.61 (m, 3H), 7.41 (dd, *J* = 8.7, 1.9 Hz, 1H), 7.35 – 7.28 (m, 2H), 1.73 (s, 3H), -0.00 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 138.14, 132.15, 131.76, 127.68, 127.27, 126.56, 125.63, 125.59, 122.60, 121.26, 120.52, 70.74, 32.37, 0.00.



5-bromo-1-((trimethylsilyl)oxy)-2,3-dihydro-1H-indene-1-carbonitrile(4j): ¹H NMR (CDCl₃, 400 MHz): δ 7.60 (s, 1H), 7.42 (d, *J* = 8.1 Hz, 1H), 7.12 (d, *J* = 8.1 Hz, 1H), 3.07 – 2.86 (m, 2H), 2.70 (ddd, *J* = 13.0, 7.6, 5.3 Hz, 1H), 2.43 – 2.33 (m, 1H), 0.21 (s, 10H). ¹³C NMR (CDCl₃, 101 MHz): δ 144.45, 141.28, 133.02, 127.27, 126.78, 120.91, 120.45, 42.97, 29.04, 1.18.



1-((trimethylsilyl)oxy)cyclohexane-1-carbonitrile(4k)⁴: ¹H NMR (CDCl₃, 400 MHz): δ 2.07 – 1.96 (m, 2H), 1.75 – 1.67 (m, 2H), 1.66 – 1.40 (m, 6H), 0.21 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 121.84, 67.82, 39.32, 24.48, 22.60, 1.35.



2-cyclohexyl-2-((trimethylsilyl)oxy)propanenitrile(4l)⁴: ¹H NMR (CDCl₃, 400 MHz): δ 2.24 (s, 1H), 2.02 (s, 3H), 1.73 (d, *J* = 26.3 Hz, 5H), 1.34 – 0.96 (m, 6H), 0.29 (s, 10H). ¹³C NMR (CDCl₃, 101 MHz): δ 213.84, 128.85, 53.22, 30.29, 29.68, 27.74, -0.00.



2-methyl-1-((trimethylsilyl)oxy)cyclohexane-1-carbonitrile(4m)¹⁰: ¹H NMR (CDCl₃, 400 MHz): δ 2.18 – 2.11 (m, 1H), 1.83 – 1.50 (m, 9H), 1.05 (d, *J* = 6.6 Hz, 3H), 0.20 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 118.71, 74.58, 41.72, 39.41, 38.29, 30.09, 23.46, 22.31, 18.75, 14.97, -0.00.



1-((trimethylsilyl)oxy)cyclobutane-1-carbonitrile(4n): ¹H NMR (CDCl₃, 400 MHz): δ 2.62 – 2.51 (m, 2H), 2.37 – 2.24 (m, 2H), 1.92 – 1.78 (m, 2H), 0.20 (d, J = 2.5 Hz, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 122.31, 66.96, 37.96, 12.22, 0.54.

2,3,3-trimethyl-2-((trimethylsilyl)oxy)butanenitrile(4o)¹¹: ¹H NMR (CDCl₃, 400 MHz): δ 1.47 (s, 3H), 0.99 (s, 9H), 0.20 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 121.77, 76.11, 38.77, 24.51, 23.71, 1.08.

2,3-dimethyl-2-((trimethylsilyl)oxy)butanenitrile(4p)³: ¹H NMR (CDCl₃, 400 MHz): δ 1.86 (p,

J = 6.8 Hz, 1H), 1.53 (s, 3H), 1.03 (dd, *J* = 6.7, 4.3 Hz, 6H), 0.24 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz): δ 121.59, 73.48, 39.10, 17.16, 16.95, 1.20.



2-methyl-2-((trimethylsilyl)oxy)nonanenitrile(4q): ¹H NMR (CDCl₃, 400 MHz) δ 6.84 (d, J = 1.3 Hz, 3H), 1.80 – 1.69 (m, 2H), 1.60 (s, 4H), 1.41 – 1.28 (m, 9H), 0.94 (t, J = 6.6 Hz, 3H), 0.28 (s, 9H). ¹³C NMR (CDCl₃, 101 MHz) δ 120.90, 66.64, 42.12, 30.47, 28.03, 27.84, 27.60, 23.02, 21.35, 19.89, 12.78, 0.00.

Copies of NMR Spectra of Cyanohydrin Products

¹H NMR spectrum of **3a** in CDCl₃ at 400 MHz



¹³C NMR spectrum of **3a** in CDCl₃ at 101 MHz



¹H NMR spectrum of **3b** in CDCl₃ at 400 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

 ^1H NMR spectrum of 3c in CDCl3 at 400 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

^1H NMR spectrum of 3d in CDCl3 at 400 MHz



$^{13}\mathrm{C}$ NMR spectrum of 3d in CDCl3 at 101 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

¹H NMR spectrum of **3e** in CDCl₃ at 400 MHz



 $^{13}\mathrm{C}$ NMR spectrum of 3e in CDCl3 at 101 MHz



¹H NMR spectrum of 3f in CDCl₃ at 400 MHz



 $^1\mathrm{H}$ NMR spectrum of $\mathbf{3g}$ in CDCl_3 at 400 MHz



$^1\mathrm{H}$ NMR spectrum of $\mathbf{3h}$ in CDCl_3 at 400 MHz



¹³C NMR spectrum of **3h** in CDCl₃ at 101 MHz



¹H NMR spectrum of **3i** in CDCl₃ at 400 MHz



 ^1H NMR spectrum of 3j in CDCl3 at 400 MHz



¹H NMR spectrum of **3k** in CDCl₃ at 400 MHz



¹³C NMR spectrum of **3k** in CDCl₃ at 101 MHz



¹H NMR spectrum of **31** in $CDCl_3$ at 400 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

¹H NMR spectrum of 3m in CDCl₃ at 400 MHz



¹H NMR spectrum of **3n** in CDCl₃ at 400 MHz



140 130 120 110 100 90 fl (ppm) -10 210 200 190

¹H NMR spectrum of **30** in CDCl₃ at 400 MHz



 $^{13}\mathrm{C}$ NMR spectrum of 3o in CDCl3 at 101 MHz



¹H NMR spectrum of **3p** in CDCl₃ at 400 MHz





¹H NMR spectrum of **3q** in CDCl₃ at 400 MHz



¹³C NMR spectrum of **3q** in CDCl₃ at 101 MHz



 ^1H NMR spectrum of 3r in CDCl3 at 400 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

¹H NMR spectrum of **4a** in CDCl₃ at 400 MHz



^{13}C NMR spectrum of 4a in CDCl₃ at 101 MHz



¹H NMR spectrum of **4b** in CDCl₃ at 400 MHz



¹H NMR spectrum of **4a** in CDCl₃ at 400 MHz



¹³C NMR spectrum of **4a** in CDCl₃ at 101 MHz



¹H NMR spectrum of **4d** in CDCl₃ at 400 MHz



¹³C NMR spectrum of **4d** in CDCl₃ at 101 MHz



¹H NMR spectrum of **4e** in CDCl₃ at 400 MHz



¹H NMR spectrum of **4f** in CDCl₃ at 400 MHz



¹H NMR spectrum of 4g in CDCl₃ at 400 MHz





$^1\mathrm{H}$ NMR spectrum of 4h in CDCl3 at 400 MHz



$^{13}\mathrm{C}$ NMR spectrum of 4h in CDCl3 at 101 MHz



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

¹H NMR spectrum of **4i** in CDCl₃ at 400 MHz



¹H NMR spectrum of **4j** in CDCl₃ at 400 MHz





¹H NMR spectrum of **4k** in CDCl₃ at 400 MHz



¹H NMR spectrum of **41** in CDCl₃ at 400 MHz





¹H NMR spectrum of 4m in CDCl₃ at 400 MHz



¹H NMR spectrum of **4n** in CDCl₃ at 400 MHz



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¹H NMR spectrum of **40** in CDCl₃ at 400 MHz





210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)

¹H NMR spectrum of **4p** in CDCl₃ at 400 MHz



 $^{13}\mathrm{C}$ NMR spectrum of 4p in CDCl3 at 101 MHz





S45

Competing Experiments for Selective Cyanosilylation of Aldehyde vs Ketone

Aldehyde (1 mmol), ketone (1 mmol), trimethylsilyl cyanide (1.1 mmol), and *n*-butyllithium (0.01-0.05 mol %) were taken in a screw capped vial equipped with a magnetic bar and the reaction mixture was stirred at room temperature for 1 h. Reaction progress was monitored by ¹H NMR, which clearly indicated the complete conversion of an aldehyde to its corresponding cyanohydrin product and predominant presence of the unreacted ketone; final spectra are provided as follows:

¹H NMR Spectra of Intermolecular Cyanosilylation Reactions (400 MHz, CDCl₃):





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¹H NMR Spectra of Intramolecular Cyanosilylation Reaction (400 MHz, CDCl₃):





Mechanistic Studies

General Procedure for Stoichiometric Reactions

In a glovebox, benzaldehyde (0.05 mmol) and *n*-BuLi (0.05 mmol) were mixed in the THF- d_8 solvent (0.5 mL) in a NMR tube for 1 h at room temperature. Reaction progress was monitored by ¹H NMR. The characteristic H signal of benzaldehyde was still there without change (**Fig. S1**; $\delta = 10.01$ ppm); nevertheless, the H signal of PhCHO was disappeared upon adding Me₃SiCN to the above mixture, subsequently accompanied by the appearance of the newly formed cyanohydrin peak (**Fig. S3**; $\delta = 4.68$ ppm). When stoichiometric reaction of *n*-BuLi (0.05 mmol) with Me₃SiCN (0.05 mmol) was carried out in THF- d_8 , however, the peaks were unattributed in the ¹H NMR spectra for unclear reasons yet (**Fig. S5**).

Moreover, we also utilized PhNHLi instead of *n*-BuLi to infer the reaction process of *n*-BuLi-catalyzed cyanosilylation. Firstly, The reaction of a 1:1.1 molar ratio of benzaldehyde and Me₃SiCN catalyzed by PhNHLi (0.01 mol %) was carryied out at room temperature under neat conditions. It was recorded that PhNHLi exhibited excellent catalytic activity similar to that of *n*-BuLi (**Fig. SA**). Next, stoichiometric reactions were executed. The procedure was operated in a glovebox by adding 0.05 mmol PhNHLi (4.9 mg in 0.5 mL THF-*d*₈) and 0.05 mmol PhCHO to a NMR tube, and no reaction occurred as demonstrated by the ¹H NMR spectrum as the same as the *n*-BuLi case (**Fig. S2**; $\delta = 10.01$ ppm); subsequently, one equivalent of Me₃SiCN was added to the above mixture, which did lead to the formation of cyanosilylated product (**Fig. 4**; top; $\delta = 5.52$ ppm). Moreover, when 0.05 mmol PhNHLi (4.9 mg in 0.5 mL THF-*d*₈) was treated with 0.05 mmol Me₃SiCN, it can been clearly seen that a new peak was recorded at $\delta = 4.41$ ppm (**Fig. S6**).



Fig. SA The copy of ¹H NMR spectra (400 MHz) with mesitylene as an internal standard of PhNHLi-catalyzed cyanosilylation. Reaction conditions: Benzaldehyde (1 mmol), Me₃SiCN (1.1 mmol) and PhNHLi (0.01 mol %) at 25 °C; 1 h.



Fig. S1 The copy of ¹H NMR spectra (400 MHz) with mesitylene as an internal standard. Reaction conditions: Benzaldehyde (0.05 mmol) and n-BuLi (0.05 mmol) in 0.5 mL THF- d_8 at 25 °C; 1 h.



Fig. S2 The copy of ¹H NMR spectra (400 MHz) with mesitylene as an internal standard. Reaction conditions: Benzaldehyde (0.05 mmol) and PhNHLi (0.05 mmol) in 0.5 mL THF- d_8 at 25 °C; 1h.



Fig. S3 The copy of ¹H NMR spectra of the 1:1:1 mixture of PhCHO, n-BuLi and Me₃SiCN. Reaction conditions: Benzaldehyde (0.05 mmol) and n-BuLi (0.05 mmol) in 0.5 mL THF- d_8 were first mixed at 25 °C for 1 h, then the Me₃SiCN (0.05 mmol) was added to the above mixture in 1 h.



Fig. S4 The copy of ¹H NMR spectra of compound PhNHLi (middle, green), the 1:1 mixture of PhCHO and PhNHLi (bottom, red), and the 1:1:1 mixture of PhCHO, PhNHLi and Me₃SiCN (top, blue).



Fig. S5 The copy of ¹H NMR spectra of the 1:1 mixture of Me₃SiCN and n-BuLi. Reaction conditions: Me₃SiCN (0.05 mmol) and n-BuLi (0.05 mmol) in 0.5 mL THF- d_8 at 25 °C; 1 h.



Fig. S6 The copy of ¹H NMR spectra (400 MHz) with mesitylene as an internal standard. Reaction conditions: TMSCN (0.05 mmol) and PhNHLi (0.05 mmol) in 0.5 mL THF- d_8 at 25 °C, 1 h; **Fig. S6** corresponding to the middle (green) of the figure below.





Fig. S7 The IR spectrum of the compound II.



Scheme S1 Another Possible Mechanistic Pathway for Cyanosilylation of Carbonyl Compounds Catalyzed by *n*-BuLi.

References

- 1 V. S. V. S. N. Swamy, M. K. Bisai, T. Dasb and S. S. Sen, Metal free mild and selective aldehyde cyanosilylation by a neutral penta-coordinate silicon compound, *Chem. Commun.*, 2017, **53**, 6910.
- 2 S. Yadav, R. Dixit, K. Vanka and S. S. Sen, Beyond Hydrofunctionalisation: A Well-Defined Calcium Compound Catalysed Mild and Efficient Carbonyl Cyanosilylation, *Chem.- Eur. J.*, 2018, 24, 1269.
- 3 M. K. Bisai, T. Das, K. Vanka and S. S. Sen, Easily accessible lithium compound catalyzed mild and facile hydroboration and cyanosilylation of aldehydes and ketones. *Chem. Commun.*, 2018, **54**, 6843.
- 4 W. Wang, M. Luo, J. Li, S. A. Pullarkat and M. Ma, Low-valent magnesium(I)-catalyzed cyanosilylation of ketones. *Chem. Commun.*, 2018, 54, 3042.
- 5 J. Li, T. Yu, M. Luo, Q. Xiao, W. Yao, L. Xu and M. Ma, Efficient and selective aldehyde cyanosilylation catalyzed by Mg-Li bimetallic complex, *J. Organomet. Chem.*, 2018, **874**, 83.
- 6 A. Harinath, J. Bhattacharjee, H. P. Nayek and T. K. Panda, Alkali metal complexes as efficient catalysts for hydroboration and cyanosilylation of carbonyl compounds. *Dalton Trans.*, 2018, 47, 12613.
- 7 S. Rawat, M. Bhandari, B. Prashanth and S. Singh, Three Coordinated Organoaluminum Cation for Rapid and Selective Cyanosilylation of Carbonyls under Solvent-Free Conditions, *ChemCatChem*, 2020, **12**, 2407.
- 8 M. K. Sharma, S. Sinhababu, G. Mukherjee, G. Rajaraman and S. Nagendran, A cationic aluminium complex: an efficient mononuclear main-group catalyst for the cyanosilylation of carbonyl compounds, *Dalton Trans.*, 2017, 46, 7672.
- 9 W. Wu, X. Zeng and J. Zhou, Carbonyl-Stabilized Phosphorus Ylide as an Organocatalyst for Cyanosilylation Reactions Using TMSCN, J. Org. Chem., 2020, 85, 14342.
- 10 M. Hatano, K. Yamakawa, T. Kawai, T. Horibe and K. Ishihara, Enantioselective Cyanosilylation of Ketones with Lithium(I) Dicyanotrimethylsilicate(IV) Catalyzed by a Chiral Lithium(I) Phosphoryl Phenoxide. *Angew. Chem. Int. Ed.*, 2016, **128**, 4089.
- 11 L. Bao, X. Kong and Y. Wang, Noncovalent Chalcogen-Bonding Catalysis Using ppm-Level Catalyst Loading to Achieve Cyanosilylation of Ketones, *Asian J. Org. Chem.*, 2020, 9, 757.