

Supplementary Information

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

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Experimental section

General Information. Reagent grade chemicals were purchased from Sigma Aldrich and Acros Organics and used without further purification. Yields reported are for isolated, spectroscopically pure compounds. (-)-Pinocarvone,¹ (1-(2-(2-bromophenyl)-2-oxoethyl)pyridin-1-ium iodide) and **1** (5S,7S)-2-(2-bromophenyl)-6,6-dimethyl-5,6,7,8-tetrahydro-5,7-methanoquinoline) were obtained as described in the literature.² When inert conditions were required, the reactions were performed under Ar atmosphere using glassware dried overnight at 120°C. Analytical thin layer chromatography (TLC) was performed on SiO₂ plates GF254 (0.25 mm layer thickness). Flash chromatography purifications were performed on a CombiFlash EZ Prep from Teledyne®. Optical rotations were measured on an Anton Paar Modular Circular Polarimeter MCP 100. The measurements were carried out in a quartz vessel ($\lambda = 100$ mm) with the sodium D line of a sodium lamp (589 nm) using spectroscopic-grade solvents. $[\alpha]_D^{20}$ values are given in 10⁻¹ deg cm² g⁻¹. IR spectra were recorded as solids between 4000 – 400 cm⁻¹ on a Bruker ALPHA FTIR. UV-Vis spectra were recorded at 20°C on an Evolution 220 spectrometer equipped with a thermostat. Fluorescence spectra were recorded at 20°C on a Fluoromax 4 spectrometer from HORIBA equipped with a thermostat. NMR spectra were recorded on a Bruker Advance DPX 300 spectrometer using TMS or the residual solvent proton as internal standard. HRMS spectra were recorded on FTMS 4.7T BioAPEX II and Waters SynaptG2-Si or at the Spectrometry Service of University of Bern. Crystallographic data were collected on a STOE IPDS-II diffractometer.

Synthesis of 2-((5S,7S)-6,6-dimethyl-5,6,7,8-tetrahydro-5,7-methanoquinolin-2-yl)benzoic acid (2). Under inert atmosphere, a solution of **1** (4.07 g, 12.4 mmol, 1 eq) in anhydrous THF (40 mL) was cooled at -78°C. n-BuLi (5.7 mL, 12.4 mmol, 2.2M in hexane, 1 eq) was added dropwise. The solution turned dark red. The reaction mixture was stirred at -78°C for 2h. Then, gaseous CO₂ was bubbled into the solution until complete discoloration. The reaction mixture was left to warm to RT and quenched with H₂O (0.9 mL). The solvents were removed under reduced pressure, and 20 mL of HCl 1M were added. Afterwards the solution was extracted with CH₂Cl₂ (3x 40 mL). The organic phase was dried over MgSO₄ and the solvent was removed under reduced pressure. Trituration of the crude compound with pentane gave the desired

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids compound **2** as a white solid (2.37 g, yield: 65%). ¹H NMR (300 MHz, CDCl₃) δ_H 8.34 – 8.21 (m, 1H, H3), 7.70 – 7.39 (m, 5H, H4, H5, H6, H9, H10), 3.23 (d, J = 2.9 Hz, 2H, H15), 2.91 (t, J = 5.6 Hz, 1H, H12), 2.79 (dt, J = 9.8, 5.8 Hz, 1H, H13a), 2.46 (tt, J = 5.9, 2.9 Hz, 1H, H14), 1.47 (s, 3H, H18), 1.35 (d, J = 9.9 Hz, 1H, H13b), 0.71 (s, 3H, H19). ¹³C NMR (75 MHz, CDCl₃) δ_C 170.2 (C1), 154.3 (C16), 154.1 (C11), 142.8 (C2), 136.4 (C_{arom}), 136.2 (C_{arom}), 136.2 (C_{arom}), 134.3 (C3), 133.3 (C_{arom}), 131.7 (C_{arom}), 130.8 (C_{arom}), 129.3 (C_{arom}), 121.8 (C_{arom}), 46.3 (C17) 46.2 (C12), 39.9 (C14), 34.9 (C15), 32.0 (C13), 26.0 (C18), 21.5 (C19). HRMS (ESI) *m/z* calcd. for C₁₉H₂₀NO₂⁺ [M+H]⁺ 294.1489, found 294.1478. [α]_D²⁰ -90° (c 1 in CH₂Cl₂).

Synthesis of (2S,4S)-3,3-dimethyl-3,4-dihydro-2,4-methanoisoindolo[2,1-a]quinolin-11(2H)-one (3). Under inert atmosphere, **2** (50 mg, 0.186 mmol, 1 eq.) was added into a 3-neck 10 mL flask. K₂CO₃ (327 mg, 2.366 mmol, 12.8 eq) and anhydrous MeCN (5 mL) were added successively. The suspension was stirred for 5 minutes at RT. Afterwards, p-TsCl (36 mg, 0.186 mmol, 1 eq.) was added in one portion. The suspension turned orange. The reaction mixture was stirred for 2 h at RT. Then, the solvent was removed under high vacuum. CH₂Cl₂ was added and the suspension was filtered under Ar. The filtrate was concentrated under vacuum. After drying the solid under high vacuum, the desired isoindolone **3** was obtained as an orange solid (51 mg, yield: 95%). ¹H NMR (300 MHz, CDCl₃) δ_H 7.93 (dt, J = 7.5, 1.0 Hz, 1H, H3), 7.69 (dt, J = 7.7, 1.0 Hz, 1H, H6), 7.58 (td, J = 7.5, 1.2 Hz, 1H, H5), 7.49 (td, J = 7.4, 1.1 Hz, 1H, H4), 7.41 (dd, J = 6.9, 1.7 Hz, 1H, H15), 6.37 (d, J = 6.4 Hz, 1H, H9), 5.99 (dd, J = 6.5, 1.7 Hz, 1H, H10), 2.84 (t, J = 5.9 Hz, 1H, H12), 2.68 – 2.61 (m, 1H, H13b), 2.58 (dd, J = 6.7, 5.5 Hz, 1H, H14), 1.69 (d, J = 8.3 Hz, 1H, H13a), 1.42 (s, 3H, H18), 0.93 (s, 3H, H19). ¹³C NMR (75 MHz, CDCl₃) δ_C 165.5 (C1), 145.1 (C16), 134.6 (C11), 132.7 (C8), 131.5 (C5), 131.4 (C2), 128.9 (C4), 128.1 (C7), 123.4 (C3), 119.6 (C6), 116.5 (C15), 113.8 (C10), 104.5 (C9), 50.8 (C12), 44.4 (C17), 42.8 (C14), 35.3 (C13), 26.4 (C18), 22.6 (C19). HRMS (ESI) *m/z* calcd. for C₁₉H₁₈NO⁺ [M+H]⁺ 276.1383, found 276.1383. [α]_D²⁰ -359 (c 0.276 in CH₂Cl₂). IR (neat, cm⁻¹): 2924 (C-H), 1661 (C=O), 1429, 1203, 1126, 800, 758, 722, 696. UV-Vis (THF): λ_{max}(ε): 288 nm (7200 M⁻¹cm⁻¹), 470 nm (2600 M⁻¹cm⁻¹).

Synthesis of ethyl 2-(5,6,7,8-tetrahydroquinolin-2-yl)benzoate (P4). Under inert atmosphere, a solution of 2-ethoxycarbonylphenyl boronic acid (1.44 g, 7.44 mmol, 1.05 eq) was mixed with degassed toluene (50 mL) and EtOH (15 mL) at RT. A solution of K₂CO₃ (6.60 g, 47.73 mmol, 6.74 eq) in water (22 mL), a degassed solution of Pd(PPh₃)₄ (0.21 g, 0.177 mmol, 0.025 eq) in EtOH (1 mL) and 2-bromo-5,6,7,8-tetrahydroquinoline (1.50 g, 7.08 mmol, 1 eq) were added successively. The reaction mixture was refluxed under inert atmosphere for 16 h and then cooled to RT and extracted with CH₂Cl₂ (3x 50 mL). The organic phase was dried over MgSO₄. The purification of the crude compound by column chromatography on SiO₂ (eluent: heptane/EtOAc = 4/1 (R_f = 0.3)) gave the compound **P4** as a translucent oil (0.14 g, yield: 21%). ¹H NMR (300 MHz, CDCl₃) δ_H 7.82 – 7.73 (m, 1H, H5), 7.57 – 7.46 (m, 2H, H1, H8), 7.45 – 7.37 (m, 2H, H6, H9), 7.20 (dt, J = 7.9, 0.8 Hz, 1H, H2), 4.16 (q, J = 7.1 Hz, 2H, H20), 2.93 (t, J = 6.3 Hz, 2H, H16), 2.81 (t, J = 6.3 Hz, 2H, H13), 1.98 – 1.76 (m, 4H, H14, H15), 1.10 (t, J = 7.1 Hz, 3H, H21). ¹³C NMR (75 MHz, CDCl₃) δ_C 169.2 (C17), 156.7 (C7), 155.7 (C10), 141.3 (C4), 137.1 (C6), 132.1 (C3), 131.0 (C1), 130.7 (C11), 129.8 (C5), 129.7 (C8), 128.0 (C9), 120.0 (C2), 60.9 (C20), 32.8 (C16), 28.7 (C13), 23.3 (C15), 22.9 (C14), 14.1 (C21). HRMS (ESI) *m/z* calcd. for C₁₈H₂₀NO₂⁺ [M+H]⁺, 282.1494, found 282.1504.

Synthesis of 2-(5,6,7,8-tetrahydroquinolin-2-yl)benzoic acid (4). A solution of **P4** (100 mg, 0.355 mmol, 1 eq) in MeOH (10 mL) was mixed with a solution of NaOH (250 mg, 6.25 mmol, 17.6 eq) in H₂O (5 mL). The reaction mixture was stirred at RT for 16 h. The MeOH was removed under reduced pressure. After pH adjustment to 2 with HCl 10%, water was removed under reduced pressure. The purification of the residue by column chromatography (eluent: EtOAc) gave the desired acid **4** as a white solid (48 mg, yield: 53%). ¹H NMR (300 MHz, DMSO-d₆) δ_H 7.65 (d, J = 7.4 Hz, 1H, H5), 7.59 – 7.53 (m, 2H, H1, H2), 7.53 – 7.42 (m,

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 2H, H6, H8), 7.29 (d, $J = 7.9$ Hz, 1H, H9), 2.79 (q, $J = 6.6$ Hz, 4H, H13, H16), 1.89 – 1.70 (m, 4H, H14, H15).
 ^{13}C NMR (75 MHz, DMSO-d6) δ_{C} 170.0 (C17), 155.7 (C4), 154.6 (C_{arom}), 139.8 (C3), 137.0 (C6), 133.1 (C_{arom}),
 130.4 (C_{arom}), 130.3 (C2), 129.6 (C1), 128.8 (C5), 127.9 (C8), 112.0 (C9), 31.9 (C16), 27.8 (C13), 22.6 (C15),
 22.3 (C14). HRMS (ESI) m/z calcd. for C₁₆H₁₆NO₂⁺ [M+H]⁺, 254.1181, found 254.1195.

Synthesis of 3,4-dihydroisoindolo[2,1-a]quinolin-11(2H)-one (5). Under inert atmosphere, 4 (9 mg, 0.0335 mmol, 1 eq.) was added into a 3-neck 10 mL flask. K₂CO₃ (59 mg, 0.4288 mmol, 12.8 eq), anhydrous MeCN (3 mL) and anhydrous acetone (3 mL) were added successively. The suspension was stirred for 5 minutes at RT. Then, p-TsCl (7 mg, 0.0335 mmol, 1 eq.) was added in one portion. The reaction mixture was stirred for 2h at RT, and the solvent was removed under high vacuum. After the addition of CH₂Cl₂ the suspension was filtered under Ar. The filtrate was concentrated under vacuum. Drying the resulting solid under high vacuum gave the desired isoindolone 5 as an orange solid (10 mg, yield: 95%). ^1H NMR (300 MHz, CDCl₃) δ_{H} 7.94 – 7.88 (m, 1H, H5), 7.67 (dt, $J = 7.7, 1.0$ Hz, 1H, H2), 7.57 (td, $J = 7.5, 1.3$ Hz, 1H, H1), 7.49 (td, $J = 7.4, 1.2$ Hz, 1H, H6), 7.40 (td, $J = 5.2, 1.8$ Hz, 1H, H16), 6.33 (d, $J = 6.4$ Hz, 1H, H7), 5.98 (dtt, $J = 7.3, 1.7, 0.9$ Hz, 1H, H8), 2.60 (ddd, $J = 12.7, 6.2, 1.7$ Hz, 3H, H13), 2.50 – 2.42 (m, 2H, H15), 1.84 – 1.74 (m, 2H, H14).
 ^{13}C NMR (75 MHz, CDCl₃) δ_{C} 165.6 (C17), 136.9 (C9), 134.8 (C4), 133.5 (C10), 132.1 (C12), 131.3 (C1), 128.8 (C6), 128.00 (C3), 123.2 (C5), 119.3 (C2), 116.4 (C8), 110.5 (C16), 104.3 (C7), 31.17 (C13), 25.4 (C15), 21.6 (C14). HRMS (ESI) m/z calcd. for C₁₆H₁₄NO⁺ [M+H]⁺ 236.1075, found 236.1077. IR (neat) ν_{max} (cm⁻¹) 2933 (C-H), 1681 (C=O), 1457, 1334, 1125, 754, 542. UV-Vis (THF): $\lambda_{\text{max}}(\epsilon)$: 284 nm (9700 M⁻¹cm⁻¹) 322 nm (3500 M⁻¹cm⁻¹), 470 nm (2500 M⁻¹cm⁻¹).

Synthesis of 2-(2-bromophenyl)-6-methylpyridine (P6). Under inert atmosphere, a solution of 2-bromo-6-methyl pyridine (0.53 mL, 4.7 mmol) in anhydrous THF (20 mL) was cooled to -78°C. A 2.2M n-BuLi solution in hexane (2.4 mL, 5.7 mmol) was added. The solution became red. The reaction mixture was stirred at -78°C for 30 minutes and afterwards a cold solution (0°C) of ZnCl₂ (0.769 g, 5.64 mmol) in anhydrous THF (6.5 mL) was added dropwise. The reaction mixture was brought to RT and stirred for 30 minutes. Separately, in a 100 mL 3-necked flask equipped with a condenser and a dropping funnel, under inert atmosphere, Pd(PPh₃)₄ (90 mg, 0.078 mmol), 2-bromo-iodobenzene (0.5 mL, 3.9 mmol) and anhydrous THF (20 mL) were mixed. To this, the aryl-zinc compound was added dropwise at RT with the dropping funnel. The reaction mixture was then refluxed under inert atmosphere overnight. After cooling to RT, 20 mL NH₄Cl sat. was added. The mixture was extracted with CH₂Cl₂ (3x50 mL) and washed twice with 15 mL of NH₄Cl (saturated aqueous solution) and 15 mL H₂O. The combined organic phases were dried over anhydrous MgSO₄ and the solvent was eliminated under reduced pressure. Hexane was added over the crude solid and the precipitate (NH₄Cl) was filtered off. The solvent was removed under reduced pressure. After purification by column chromatography on SiO₂ (eluent: hexane/EtOAc=4/0.5. R_f=0.5), the compound P6 was obtained as a yellow oil (0.37 g, yield: 38%). ^1H NMR (300 MHz, CDCl₃) δ_{H} 7.69 – 7.61 (m, 2H, H3, H10), 7.51 (dd, $J = 7.6, 1.8$ Hz, 1H, H7), 7.43 – 7.34 (m, 2H, H4, H9), 7.22 (dd, $J = 7.6, 1.8$ Hz, 1H, H8), 7.16 (d, $J = 7.8$ Hz, 1H, H2), 2.63 (s, 3H, H12). ^{13}C NMR (75 MHz, CDCl₃) δ_{C} 158.3 (C11), 157.9 (C1), 141.7 (C6), 136.2 (C10), 133.4 (C3), 131.5 (C7), 129.7 (C8), 127.7 (C9), 122.1 (C2), 122.0 (C5), 121.8 (C4), 24.8 (C12). HRMS (ESI) m/z calcd. for C₁₂H₁₁NBr⁺ [M+H]⁺ 248.0075 , found 248.0082.

Synthesis of 2-(6-methylpyridin-2-yl)benzoic acid (6). Under inert atmosphere, a 2.2M BuLi solution (3.6 mL, 7.98 mmol) in hexane was added dropwise to a solution of P6 (1.80 g, 7.25 mmol) in anhydrous THF (30mL) at-78°C. The solution turned dark red, and was stirred for 2h at -78°C. Into this was bubbled gaseous CO₂until the colour changed from dark red to yellow. The mixture was warmed up to RT and stirred for 1h. H₂O (0.3 mL) was added for quenching. The solvent was removed under reduced pressure

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General procedure for **P7** and **P8**

Under inert atmosphere, a solution of 2-ethoxycarbonylphenyl boronic acid (548 mg, 2.82 mmol, 1.05 eq) was mixed with degassed toluene (25 mL), H_2O (10 mL) and EtOH (7 mL) at RT. K_2CO_3 (2.506 g, 18.13 mmol, 6.74 eq) is afterwards added, followed by 6-bromo-2,3-dimethylpyridine (for **P7**) or 2-bromo-6-ethylpyridine (for **P8**) (2.69 mmol, 1 eq) and $[\text{Pd}(\text{PPh}_3)_4]$ (78 mg, 0.067 mmol, 0.025 eq). The reaction mixture was refluxed under inert atmosphere for 16h and then cooled to RT. The two phases were separated and the aqueous phase was extracted with EtOAc (3x25 mL). The combined organic phases were dried over MgSO_4 . The purification of the crude compound by column chromatography on SiO_2 (eluent: heptane/EtOAc = 4/1-> EtOAc 100% (R_f = 0.1)) gave the compound **P7** as a yellow oil (100 mg, yield: 15%), respectively **P8** as a light yellow oil (100 mg, yield: 15%). **P7**: ^1H NMR (300 MHz, CDCl_3) δ_{H} 7.77 (d, J = 7.6 Hz, 1H, H5), 7.57 – 7.50 (m, 2H, H1, H2), 7.50 – 7.38 (m, 2H, H6, H8), 7.23 (d, J = 7.7 Hz, 1H, H9), 4.16 (q, J = 7.2 Hz, 2H, H18), 2.52 (s, 3H, H13), 2.32 (s, 3H, H14), 1.10 (t, J = 7.1 Hz, 3H, H19). ^{13}C NMR (75 MHz, CDCl_3) δ_{C} 169.19, 156.27, 155.32, 140.98, 137.34, 132.06, 130.82, 129.73, 129.58, 129.48, 127.84, 119.99, 60.80, 22.62, 18.92, 13.90. HRMS (ESI) calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_2^+ [\text{M}+\text{H}]^+$, 256.1338, found 256.1357. **P8**: ^1H NMR (300 MHz, CDCl_3) δ_{H} 7.83 – 7.77 (m, 1H, H5), 7.66 (t, J = 7.7 Hz, 1H, H9), 7.57 – 7.51 (m, 2H, H1, H2), 7.44 (ddd, J = 7.6, 5.7, 3.2 Hz, 1H, H6), 7.28 (d, J = 7.9 Hz, 1H, H8), 7.13 (d, J = 7.7 Hz, 1H, H10), 4.12 (q, J = 7.2 Hz, 2H, H18), 2.85 (q, J = 7.6 Hz, 2H, H13), 1.36 – 1.28 (m, 3H, H14), 1.04 (t, J = 7.1 Hz, 3H, H19). ^{13}C NMR (75 MHz, CDCl_3) δ_{C} 162.82, 141.28, 136.51, 132.08, 130.90, 129.69, 129.66, 128.05, 120.20, 119.83, 60.80, 31.41, 29.70, 13.89, 13.81. HRMS (ESI) m/z calcd. for $\text{C}_{16}\text{H}_{18}\text{NO}_2^+ [\text{M}+\text{Na}]^+$, 278.0865, found 278.1012.

General procedure for **7** and **8**

A solution of **P7** or **P8** (0.355 mmol, 1 eq) in MeOH (10 mL) was mixed with a solution of NaOH (220 mg, 6.25 mmol, 17.6 eq) in H_2O (10 mL). The reaction mixture was refluxed for 16h. The MeOH was removed under reduced pressure. After pH adjustment to 2 with HCl 1 M, the water was removed under reduced pressure. 20 mL were added over the crude product and the suspension was filtered. The filtrate was concentrated under reduced pressure and the obtained solid was further dried under high vacuum to give **7** as an off-white solid (70 mg, yield: 88%), respectively **8** and a white solid (49 mg, yield: 61%). **7**: ^1H NMR (300 MHz, CD_3OD) δ_{H} 7.98 – 7.93 (m, 1H, H5), 7.78 (d, J = 7.9 Hz, 1H, H8), 7.69 – 7.53 (m, 3H, H1, H2, H6), 7.42 (d, J = 7.9 Hz, 1H, H9), 2.59 (s, 3H, H13), 2.43 (s, 3H, H14). ^{13}C NMR (75 MHz, CD_3OD) δ_{C} 154.79, 139.42, 138.53, 133.36, 131.38, 130.58, 129.85, 129.80, 128.41, 121.50, 19.49, 17.16. HRMS (ESI) calcd. for $\text{C}_{14}\text{H}_{14}\text{NO}_2^+ [\text{M}+\text{H}]^+$, 228.1015, found 228.1019. **8**: ^1H NMR (300 MHz, CD_3OD) δ_{H} 7.90 – 7.80 (m, 2H, H5, H9), 7.65 – 7.48 (m, 3H, H1, H2, H6), 7.40 (dd, J = 7.8, 1.0 Hz, 1H, H8), 7.31 (dd, J = 7.8, 0.9 Hz, 1H, H10), 2.85 (q, J = 7.6 Hz, 2H, H13), 1.36 – 1.27 (m, 3H, H14). ^{13}C NMR (75 MHz, CD_3OD) δ_{C} 162.05, 157.38,

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids 139.58, 138.02, 130.39, 129.78, 129.41, 128.24, 121.01, 120.77, 29.98, 13.11. HRMS (ESI) *m/z* calcd. for C₁₄H₁₄NO₂⁺ [M+H]⁺, 228.1015, found 228.1019.

Synthesis of 4-methylpyrido[2,1-*a*]isoindol-6(2*H*)-one **14**

Compound **14** was synthesised by following a reported procedure.³ Under inert atmosphere, 2-bromo-6-methylpyridine (860 mg, 5 mmol, 1 eq) and Pd(PPh₃)₄ (787 mg, 5.25 mmol, 1.05 eq) were dissolved in degassed toluene (20 mL). Over this, a solution of (2-formylphenyl)boronic acid (787 mg, 5.25 mmol, 1.05 eq) in degassed EtOH (10 mL) was added over, followed by a degassed Na₂CO₃ (1.060 g, 5 mmol, 1 eq) aqueous solution (5 mL). The reaction mixture was refluxed under inert atmosphere for 16h and then cooled to RT. 10 mL H₂O were added. The two phases were separated and the aqueous phase was extracted with EtOAc (3x25 mL). The combined organic phases were dried over Na₂SO₄. The purification of the crude compound by column chromatography on SiO₂ (eluent: heptane/EtOAc = 95/5 (R_f = 0.3) gave **14** as an orange solid (337 mg, yield: 34%). ¹H NMR (300 MHz, CDCl₃) δ_H 7.81 (dt, *J* = 7.5, 1.1 Hz, 1H, H5), 7.62 (dt, *J* = 7.6, 1.1 Hz, 1H, H2), 7.55 (td, *J* = 7.4, 1.2 Hz, 1H, H1), 7.46 (td, *J* = 7.3, 1.3 Hz, 1H, H6), 5.89 (td, *J* = 4.1, 1.7 Hz, 1H, H8), 4.76 (tp, *J* = 3.0, 1.5 Hz, 1H, H10), 3.05 (tq, *J* = 3.8, 1.9 Hz, 2H, H9), 2.45 (q, *J* = 1.8 Hz, 3H, H13). ¹³C NMR (75 MHz, CDCl₃) δ_C 164.37 (C14), 134.67 (C7), 133.47 (C3), 133.27 (C4), 131.83 (C1), 130.06 (C11), 129.09 (C6), 123.25 (C5), 119.19 (C2), 104.94 (C8), 104.63 (C10), 24.99 (C9), 20.03 (C13). HRMS (ESI) *m/z* calcd. for C₁₃H₁₀NO⁺ [M+H]⁺, 196.0753, found 196.0757.

Procedures for the titrations of **6-8** with TsCl

0.025 mmol (1eq.) of the corresponding substrate was dissolved in anhydrous acetone-*d*6 and treated with anhydrous Et₃N (8 uL, 0.055 mmol, 2.2 eq). Additions of 1.2 mg solid TsCl were made, corresponding to 0.25 eq.

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids
Structures, Emission, and UV-VIS spectra and $^1\text{H-NMR}$ spectra of compounds 2, 3 and 5

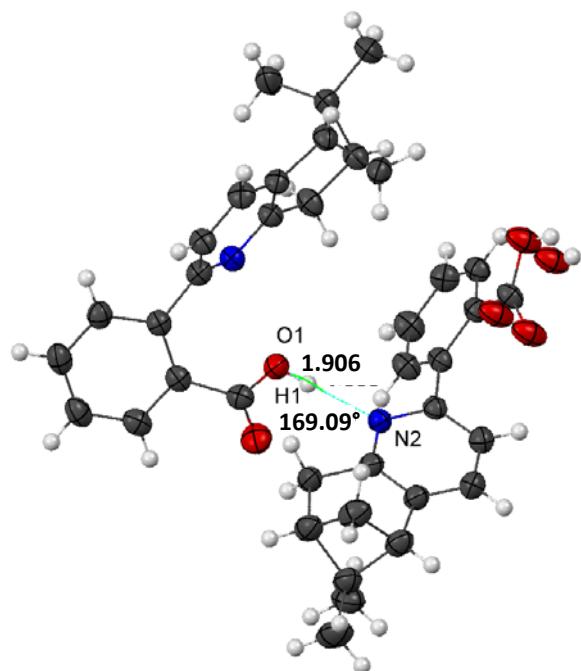


Figure S1. Asymmetric unit of the structure of 2, with depiction of the intermolecular H bonding and O1-H1-N2 angle with ellipsoids at 30% probability

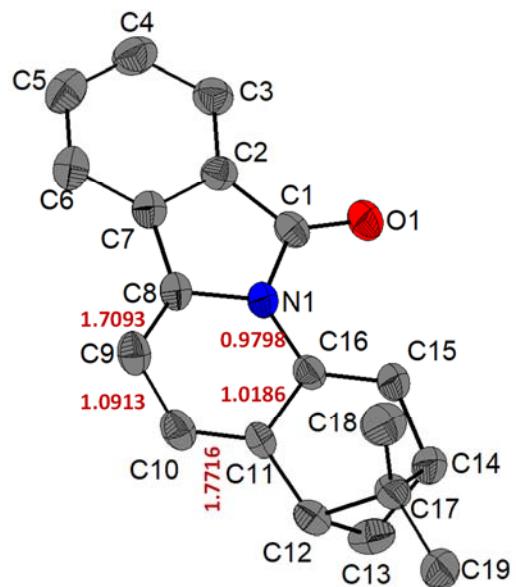


Figure S2. Molecular structure of 2 determined by single-crystal X-Ray diffraction with ellipsoids at 30% probability with atom numbering scheme and bond order values in red. The H atoms were omitted for clarity.

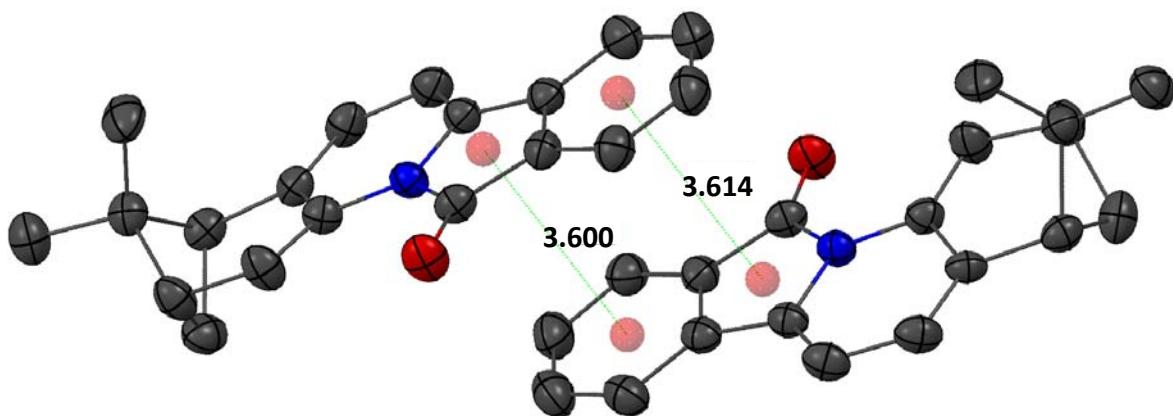


Figure S3. Depiction of the π - π stacking interactions (with the centrodi-centroid distances in black) in the crystal of **3**, with ellipsoids at 30% probability. The H atoms were omitted for clarity.

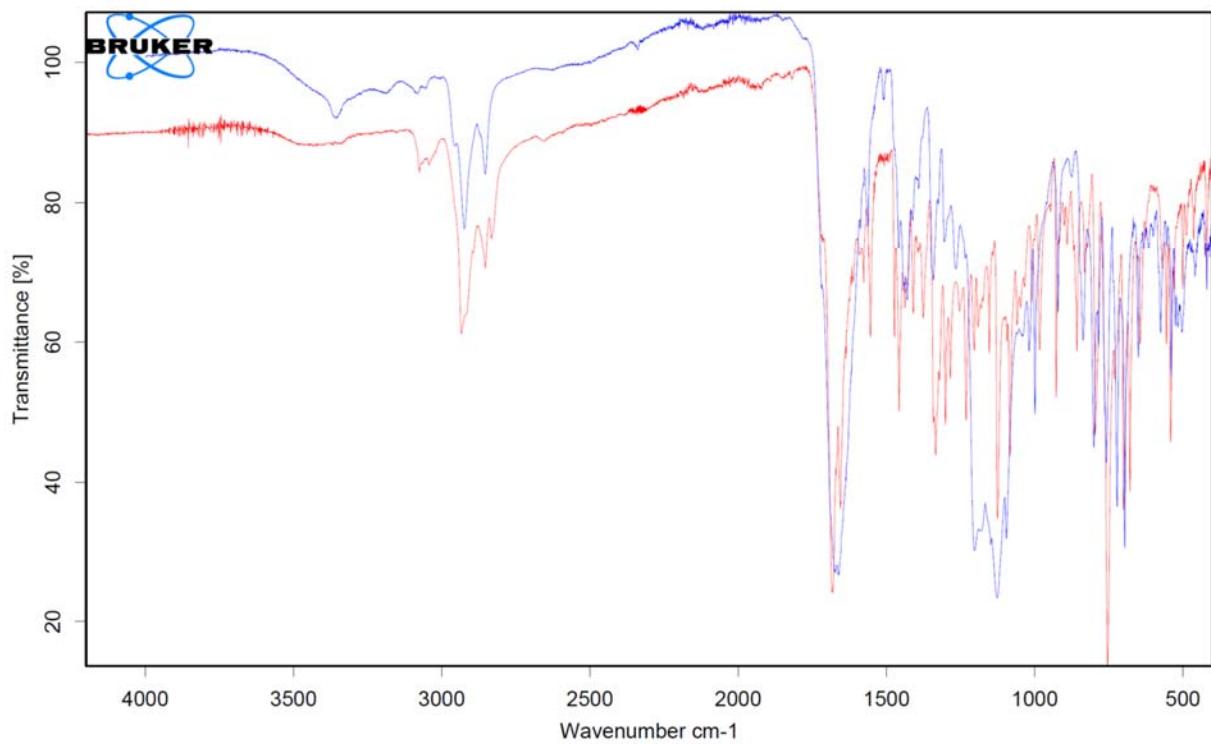
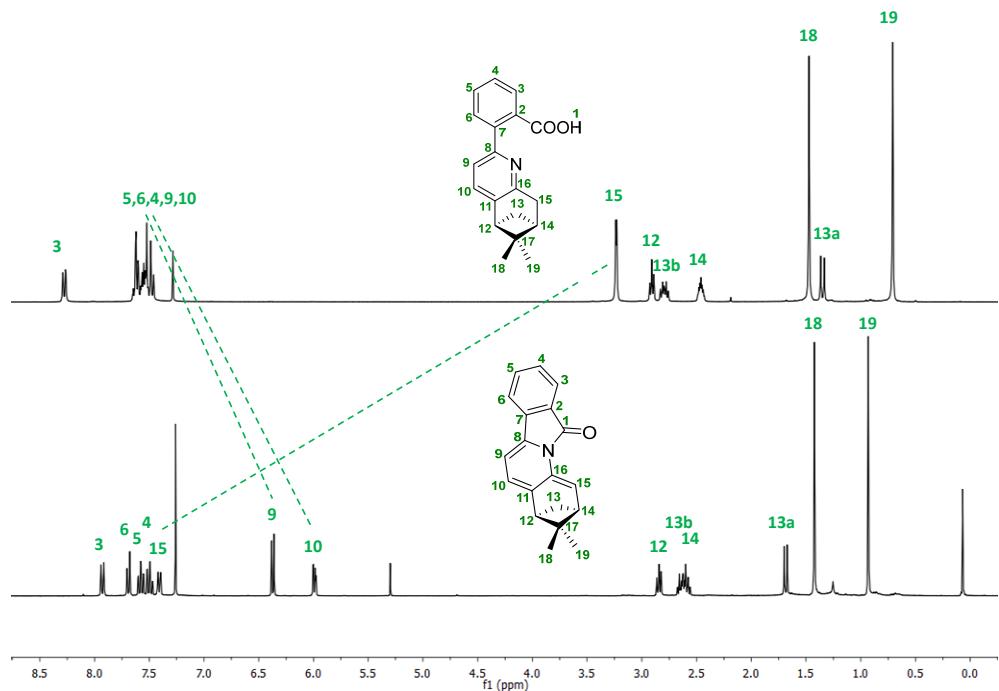
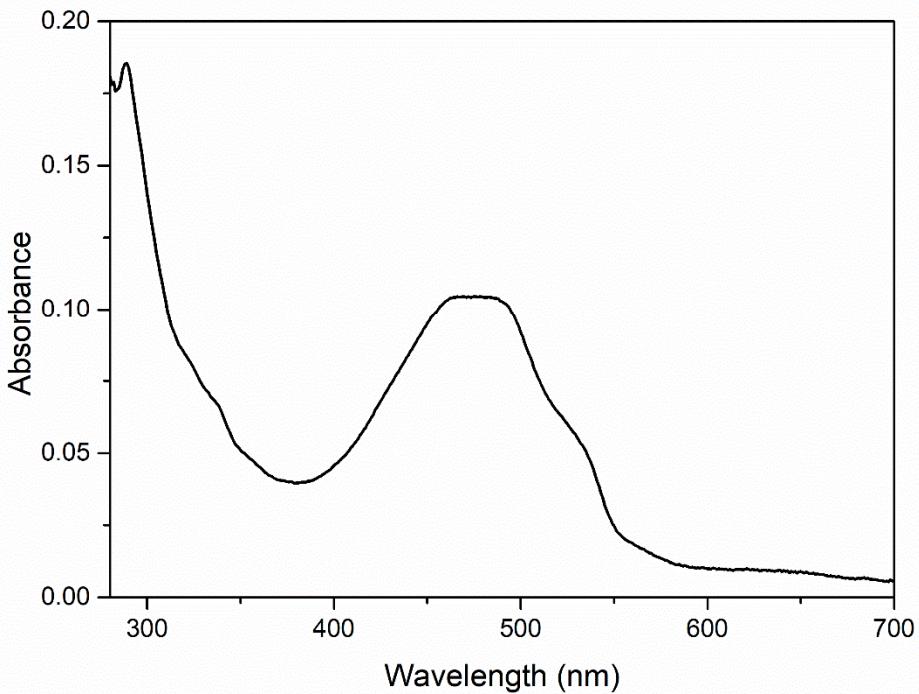


Figure S4. IR spectra of **3** (blue) and **5** (red)

Figure S5. Stacked ^1H -NMR spectra of **2** (top) and **3** (bottom)Figure S6. UV-Vis spectrum of **3** in THF (0.04 mM in THF)

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

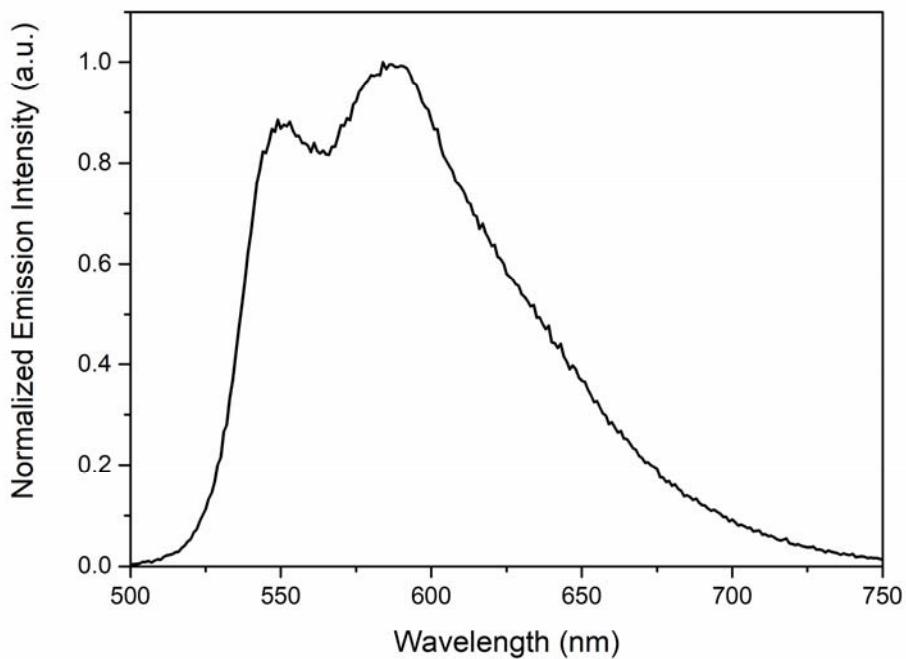


Figure S7. Emission spectrum of **3** in THF (0.04 mM, $\lambda_{\text{ex}} = 470 \text{ nm}$)

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0.12
0.11
0.11
0.10
0.10
0.09
0.09
0.08
0.08
0.07
0.07
0.06
0.06
0.05
0.05
0.04
0.04
0.03
0.03
0.02
0.02
0.01
0.01
0.00
0.00

Figure S8. $^1\text{H-NMR}$ spectrum of **5** in CDCl_3

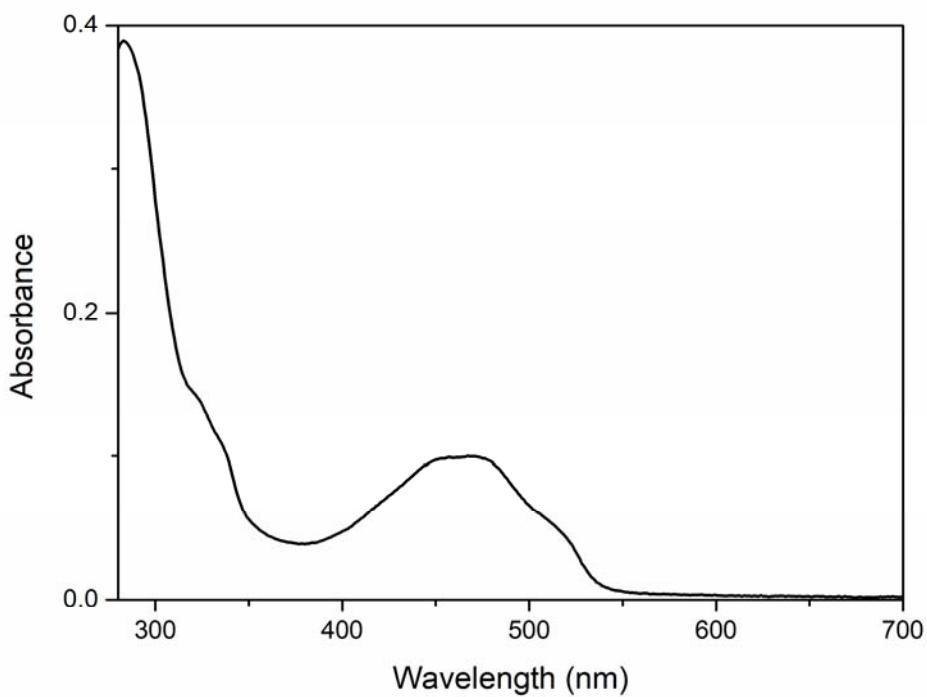


Figure S9. UV-Vis spectrum of **5** in THF (0.04 mM in THF)

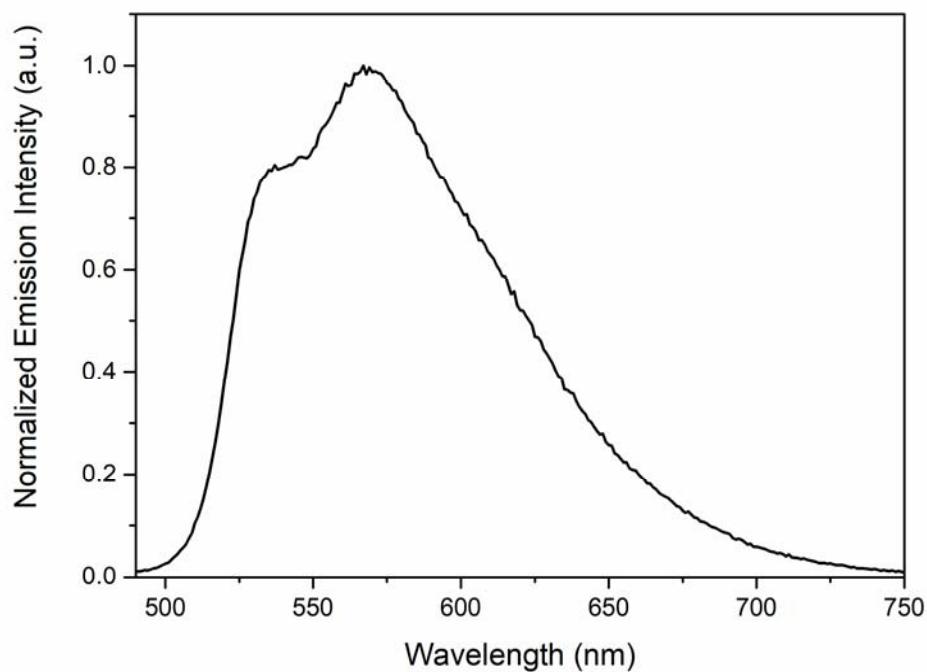


Figure S10. Emission spectrum of **5** in THF (0.04 mM, $\lambda_{ex} = 470$ nm)

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

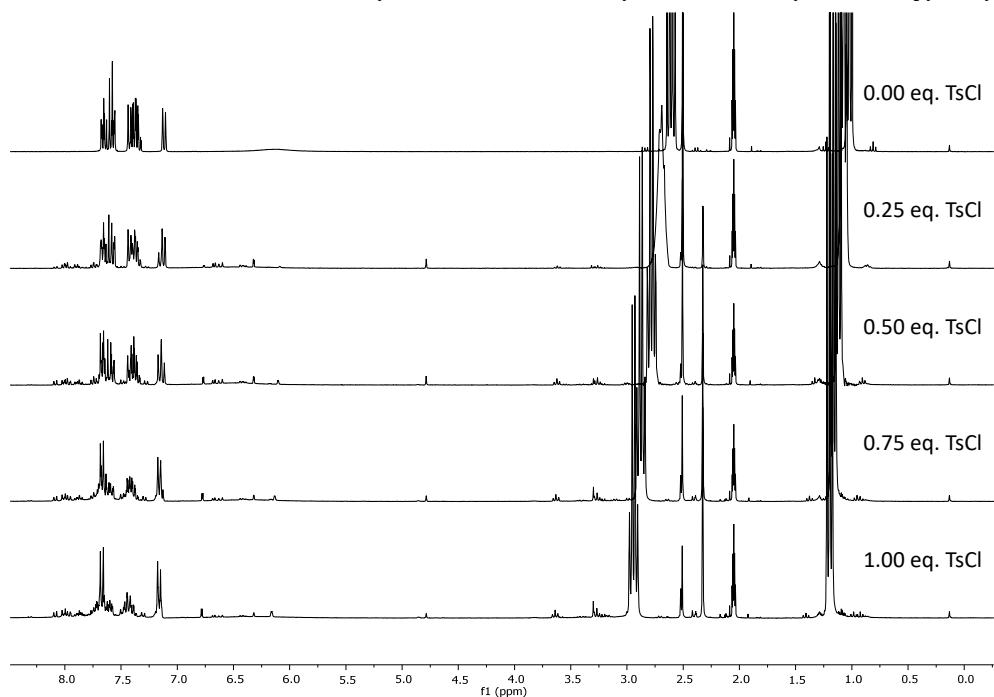


Figure S11. ¹H-NMR spectrum of the titration of **6** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d₆

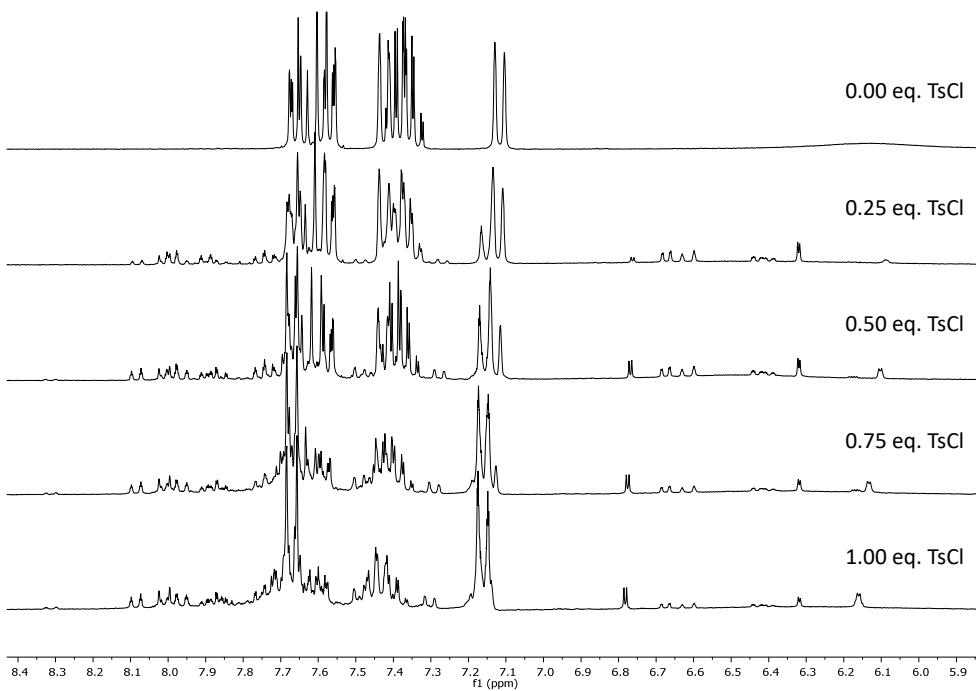


Figure S12. Aromatic region of the ¹H-NMR spectrum of the titration of **6** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d₆

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

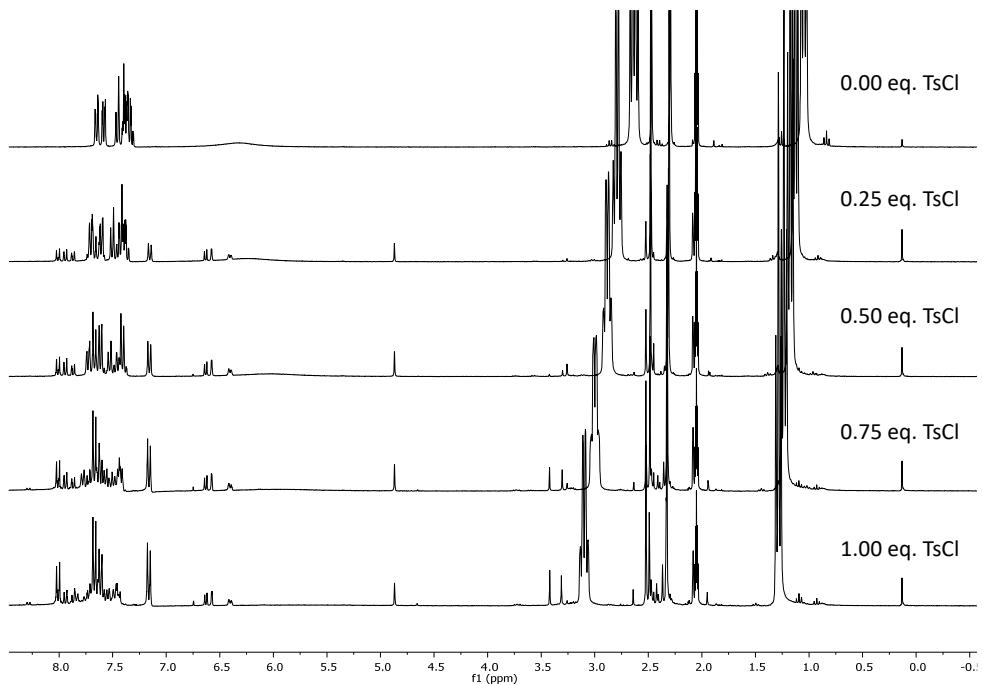


Figure S13. ¹H-NMR spectrum of the titration of **7** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d₆

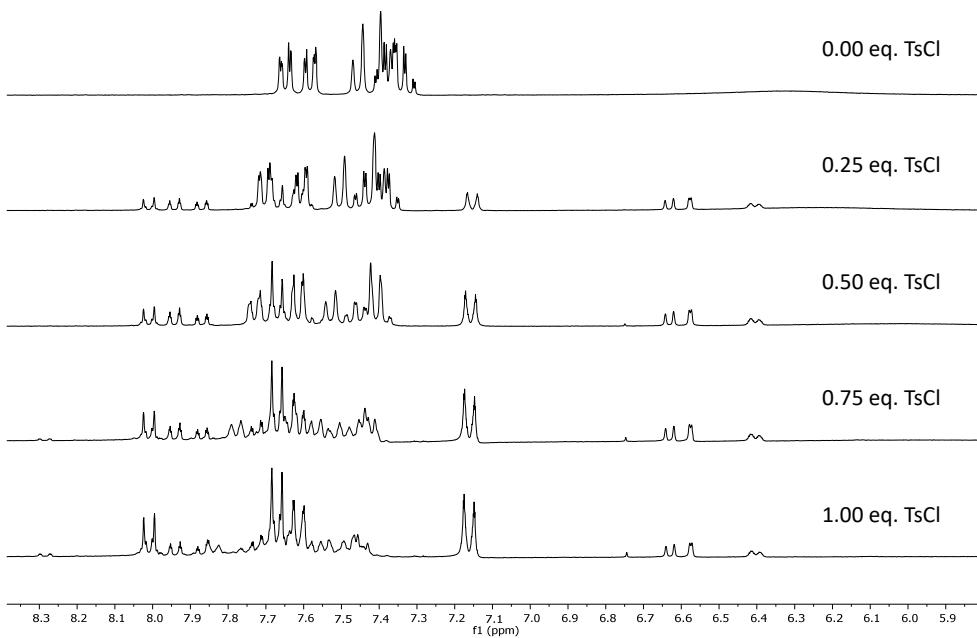


Figure S14. Aromatic region of the ¹H-NMR spectrum of the titration of **7** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d₆

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

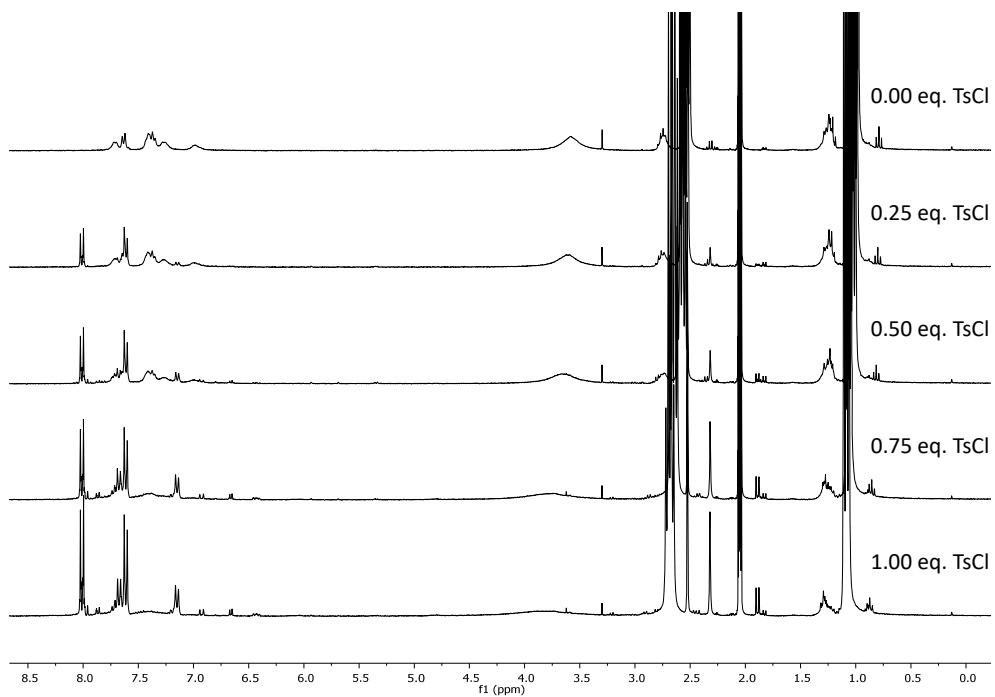


Figure S15. ¹H-NMR spectrum of the titration of **8** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d6

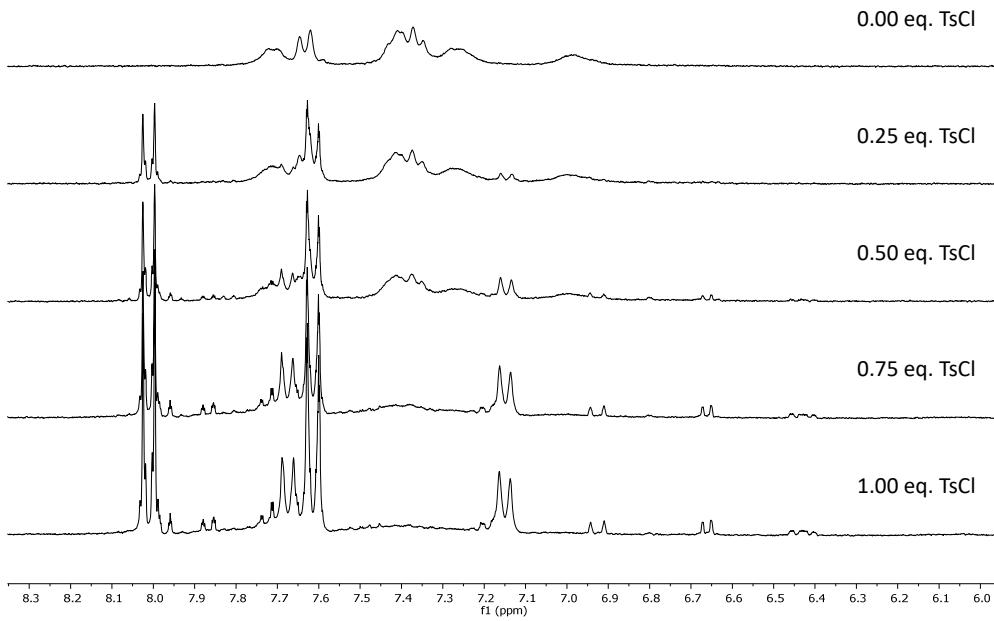
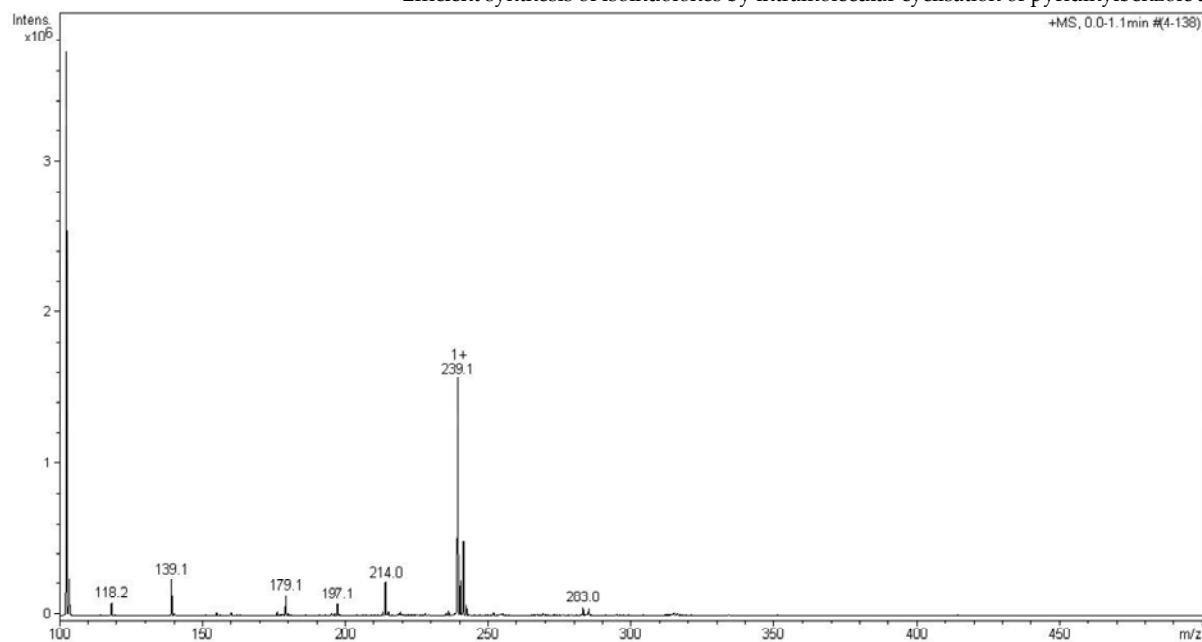
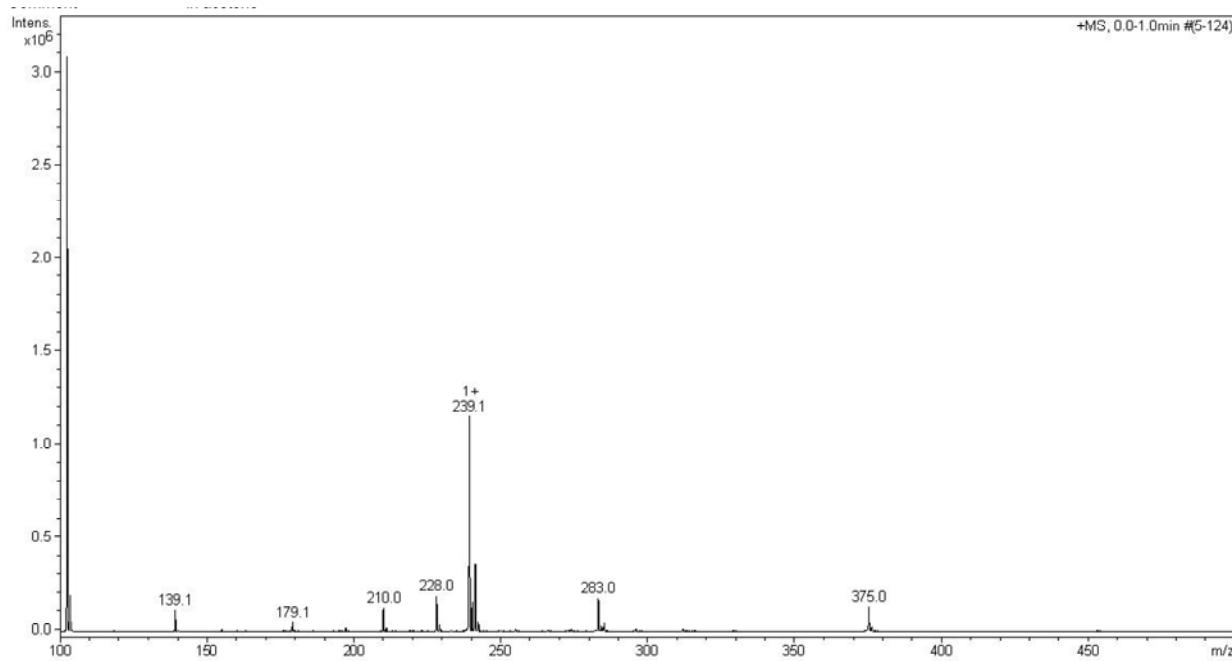
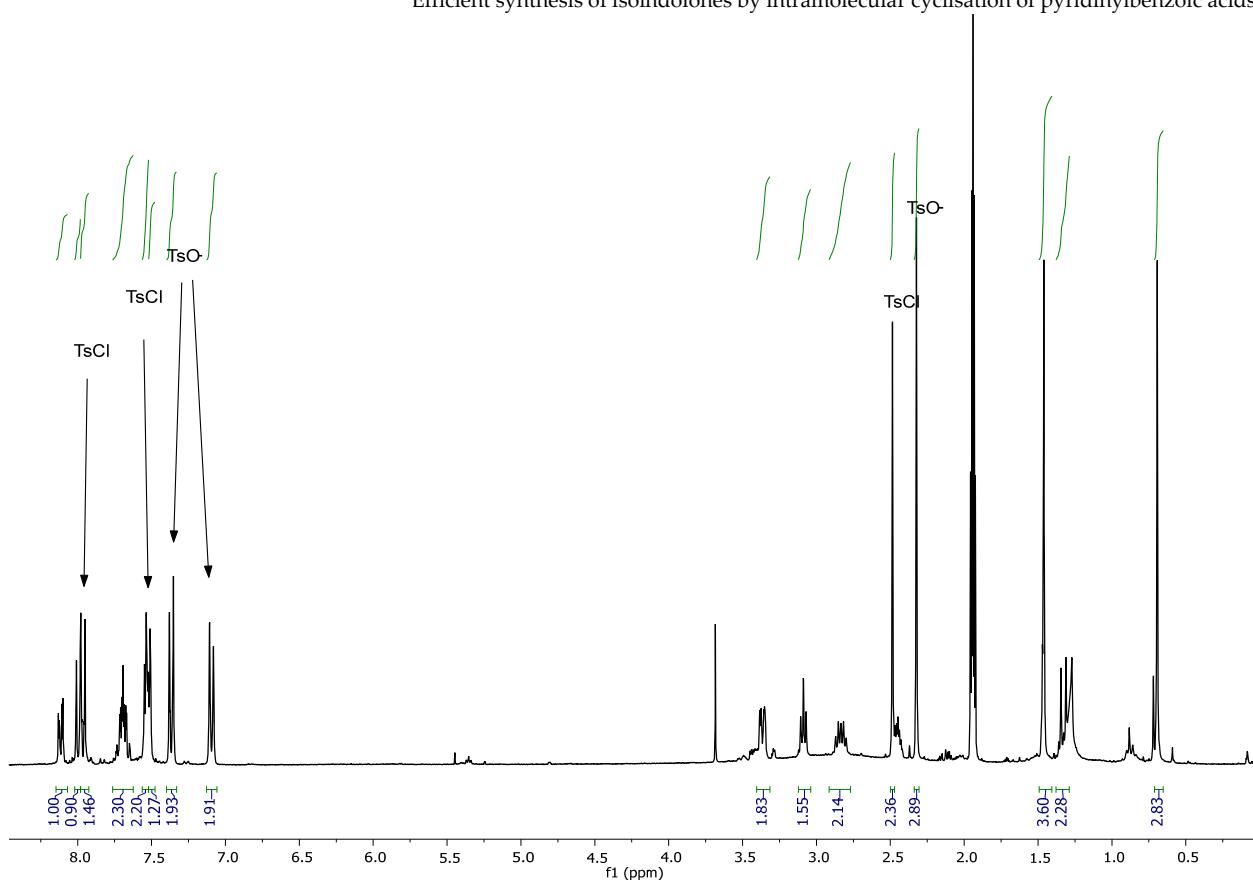


Figure S16. Aromatic region of the ¹H-NMR spectrum of the titration of **8** with TsCl in the presence of 2.2 eq. Et₃N, in acetone-d6

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

Figure S17. (+)-ESI-MS spectrum of **9** ($[M+H]^+$ at 196.1)Figure S18. (+)-ESI-MS spectrum of **10** ($[M+H]^+$ at 210.0)

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

Figure S19. ^1H -NMR spectrum of Complex-1 in CD_3CN

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

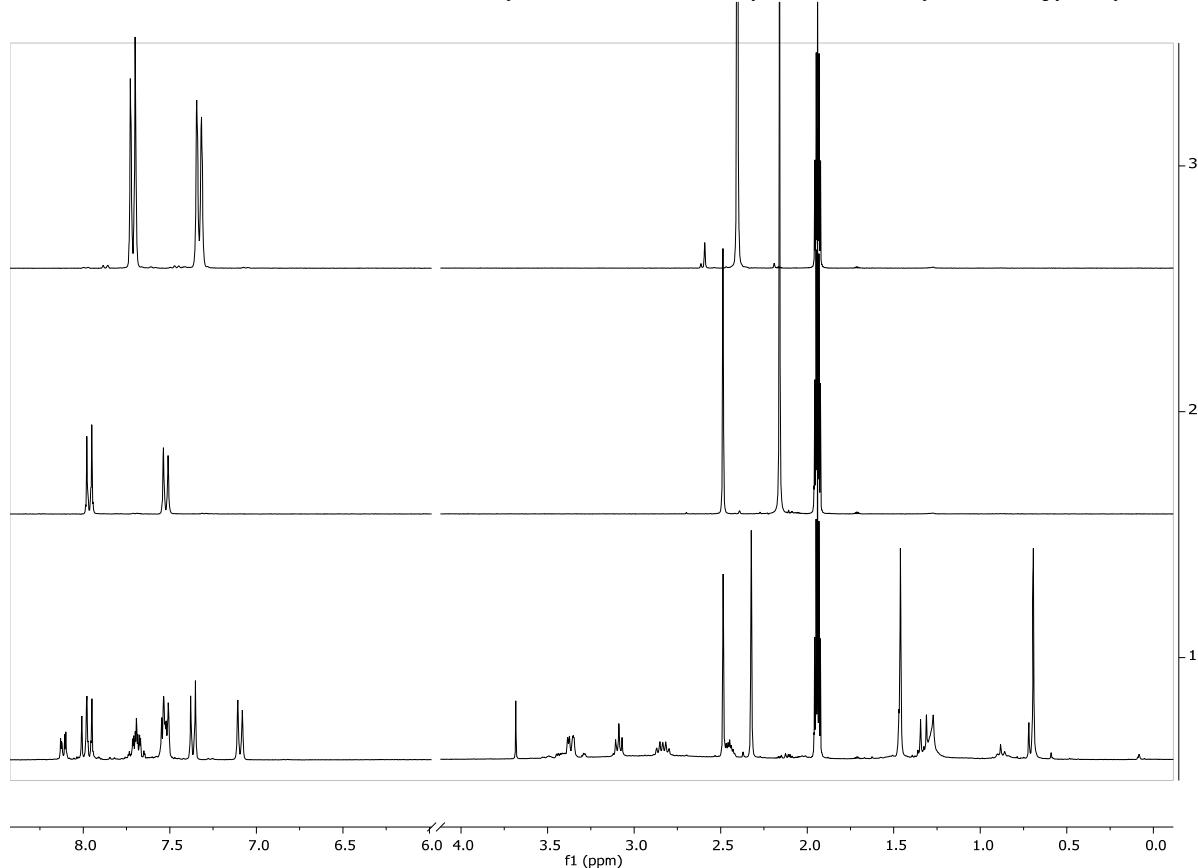


Figure S20. Stacked ¹H-NMR spectra of TsOK (top), TsCl (middle) and Complex-1 in CD₃CN

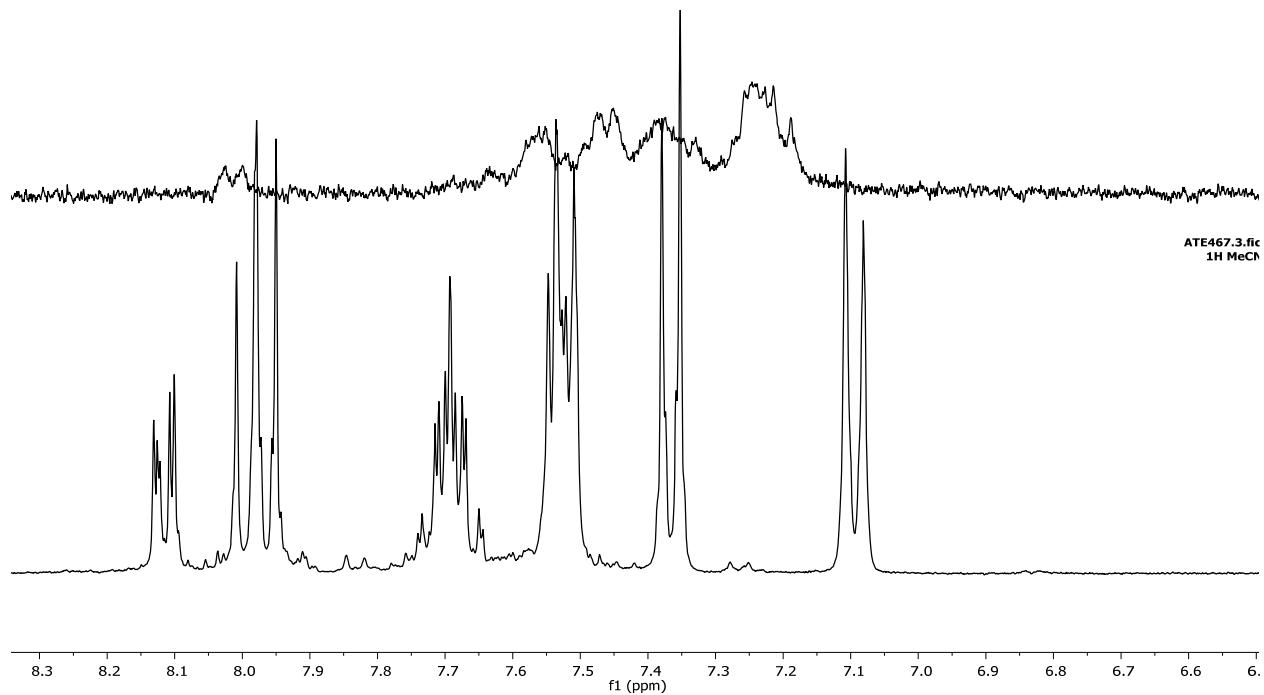
ATE467.1.fid
1H MeCN depro

Figure 21. Aromatic region of the ¹H-NMR spectra of R-COOK (top) and Complex-1 in CD₃CN

NMR spectra

NMR spectra of 2

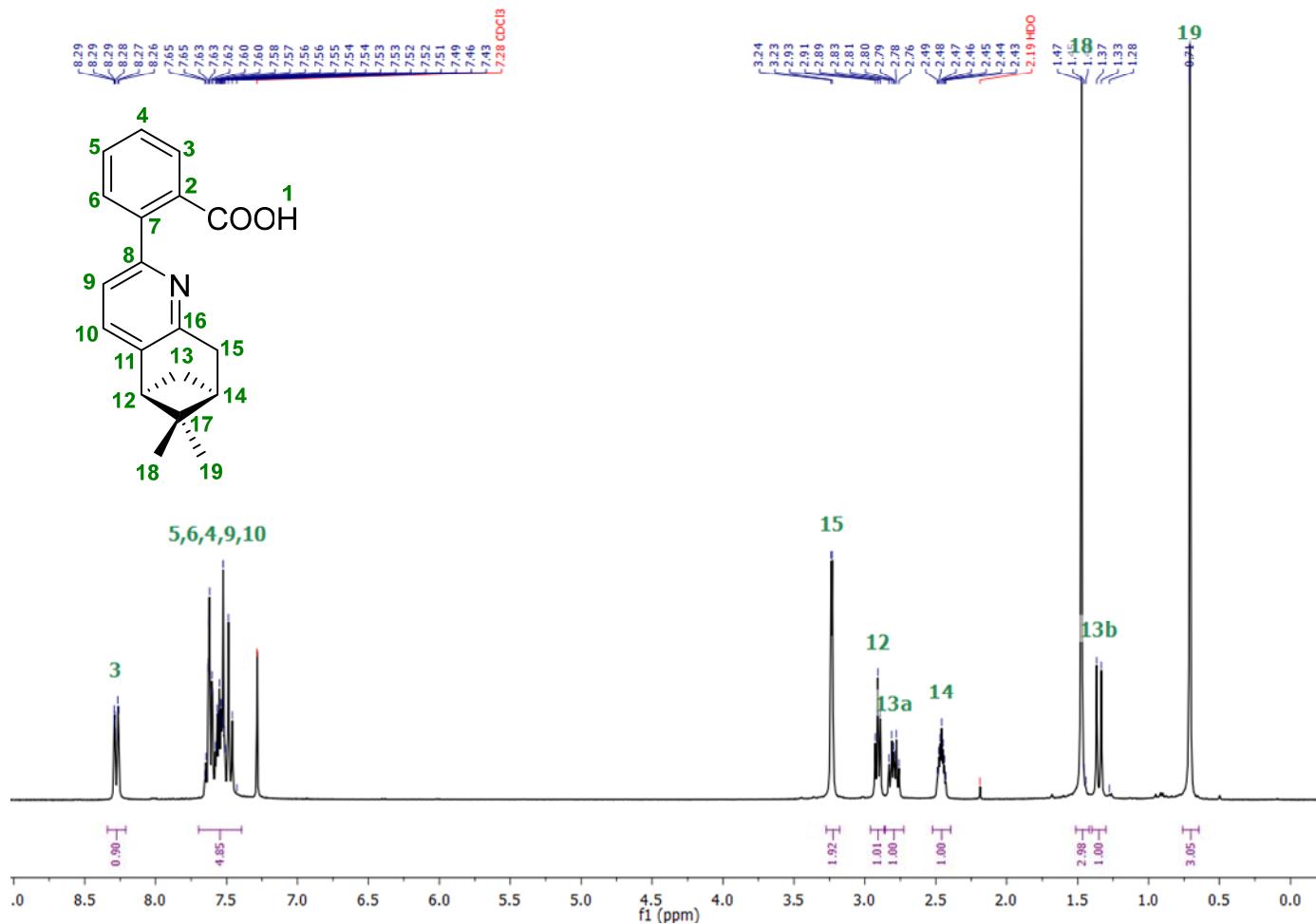
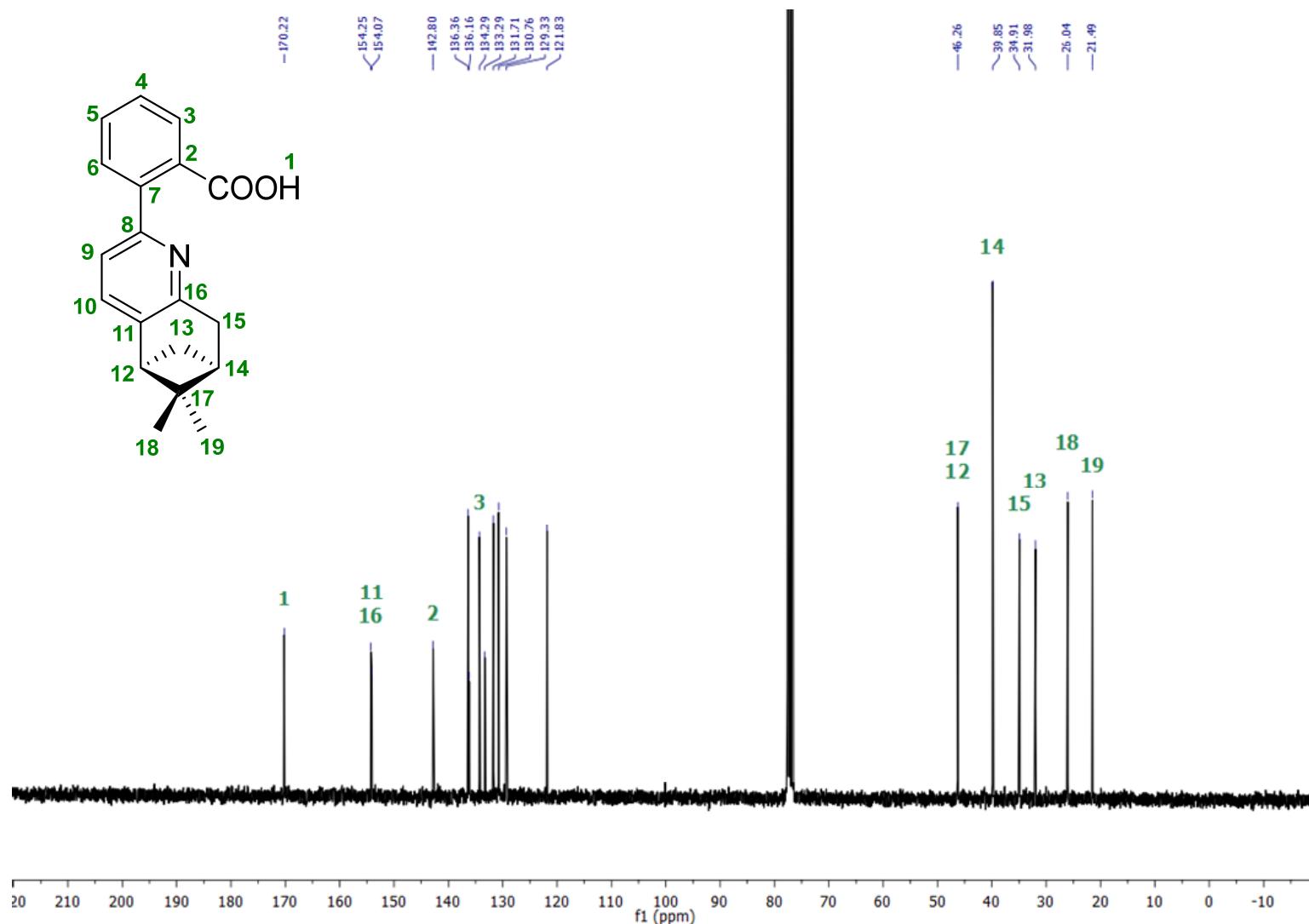


Figure S22. $^1\text{H-NMR}$ spectrum of **2** in CDCl_3

Figure S23. ^{13}C -NMR spectrum of **2** in CDCl_3

NMR spectra of 3

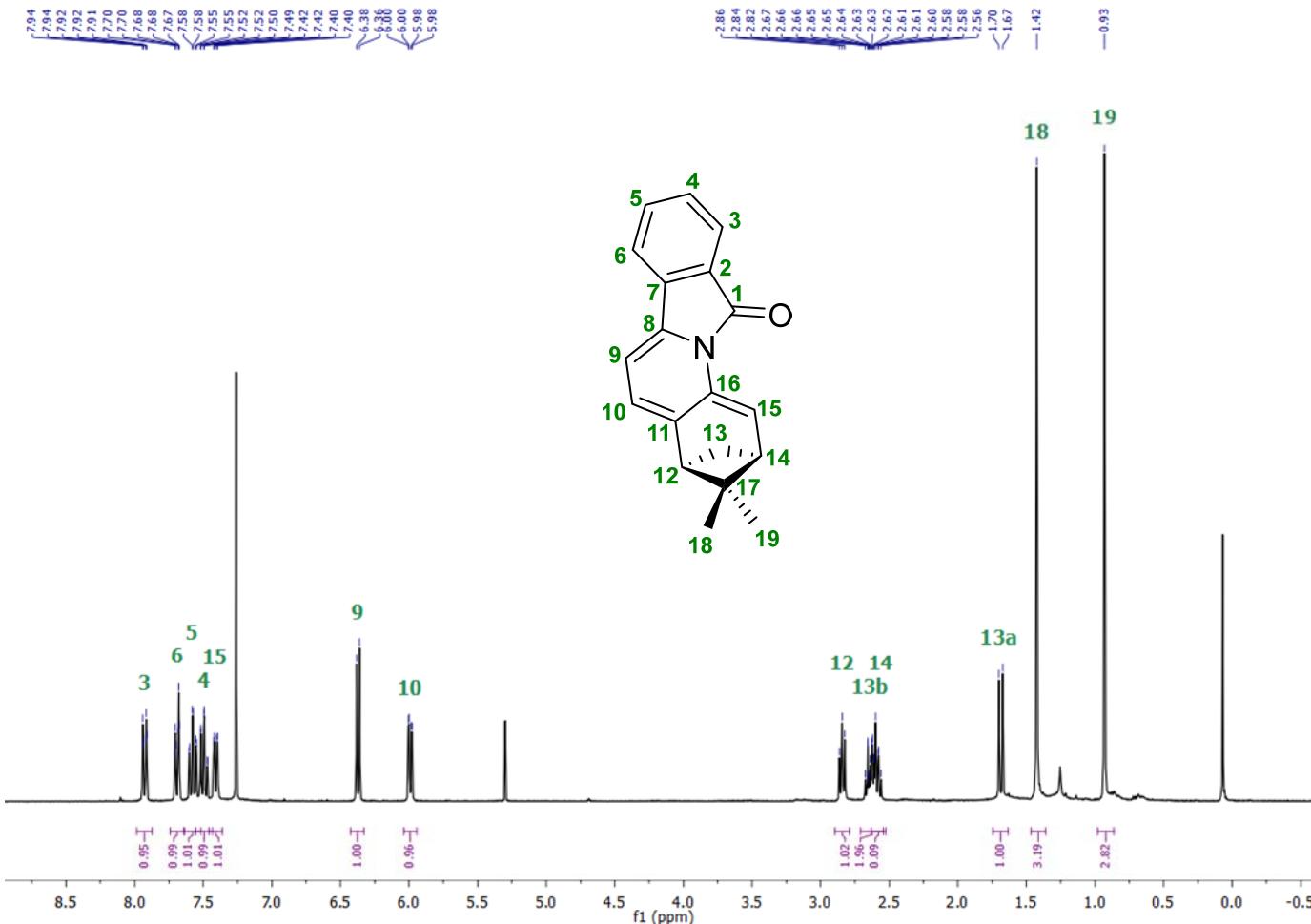
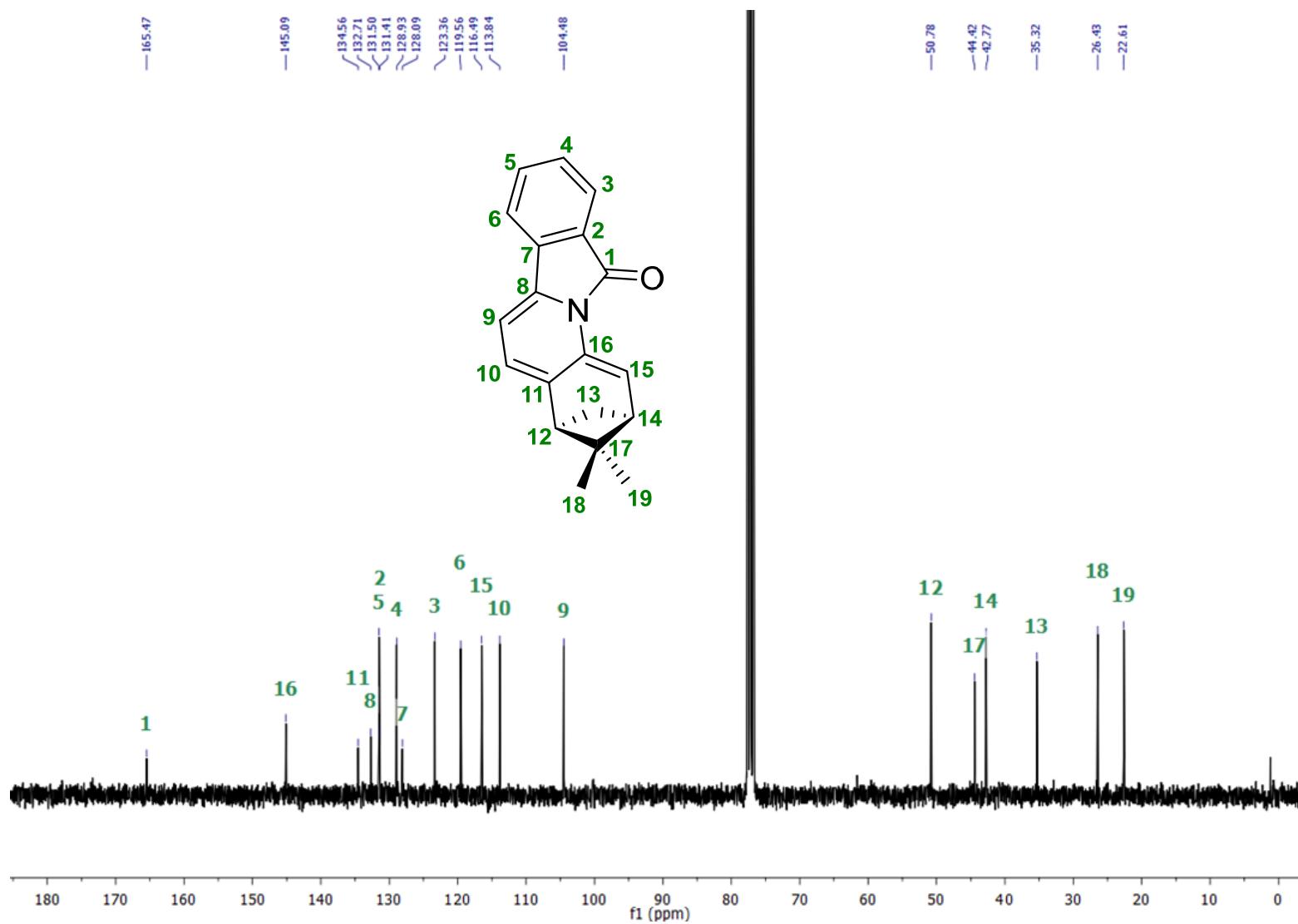
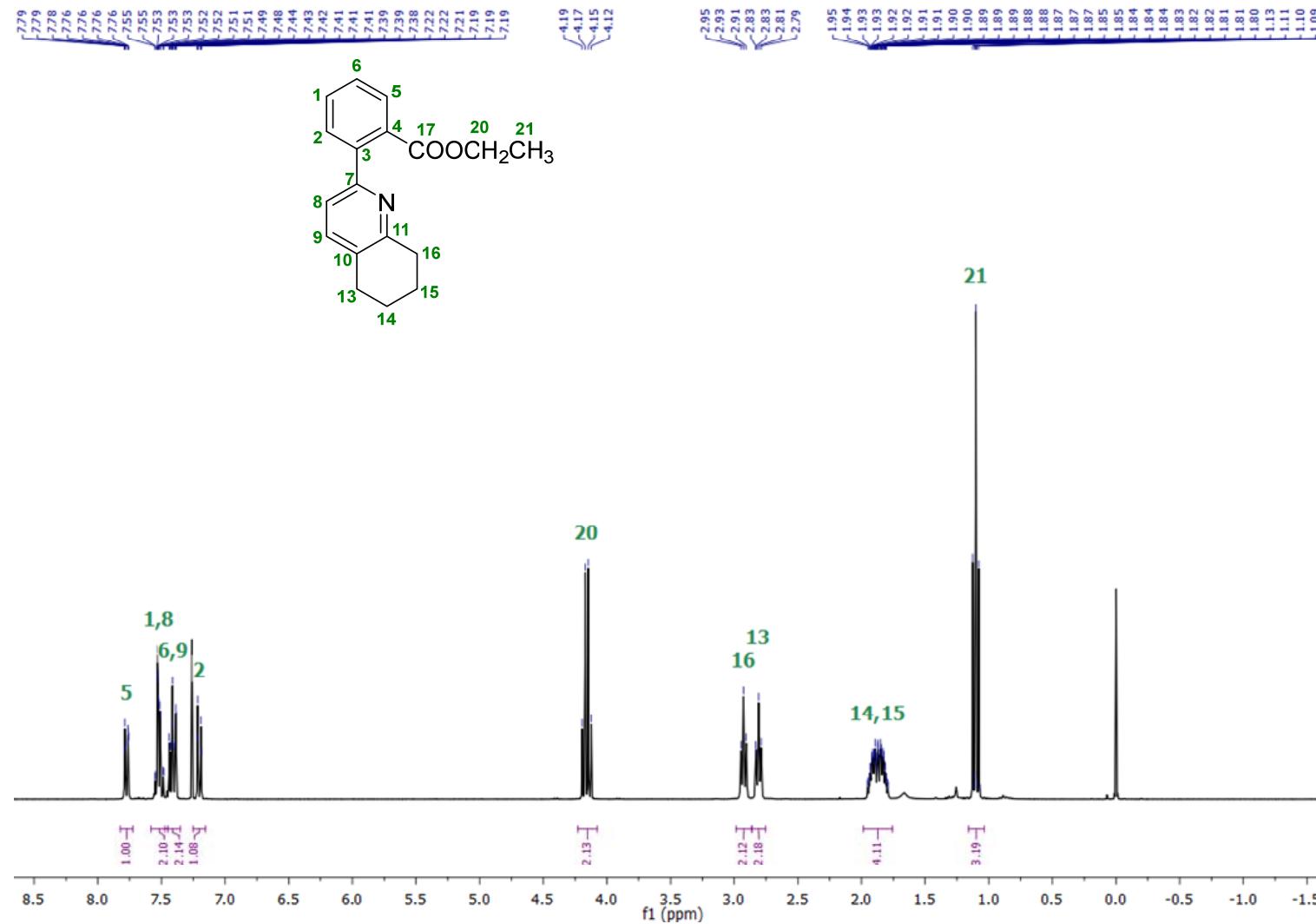


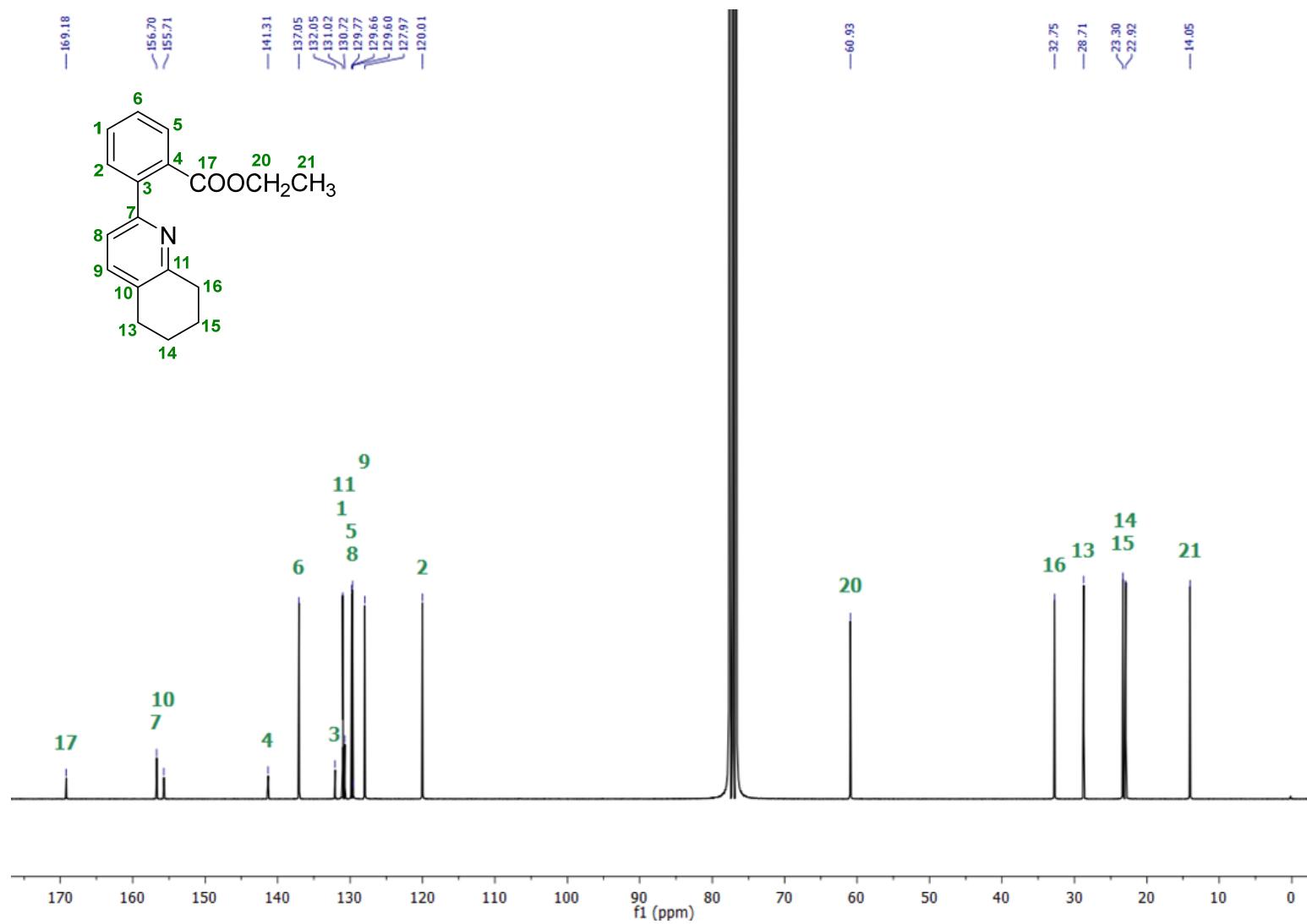
Figure S24. ^1H -NMR spectrum of **3** in CDCl_3

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

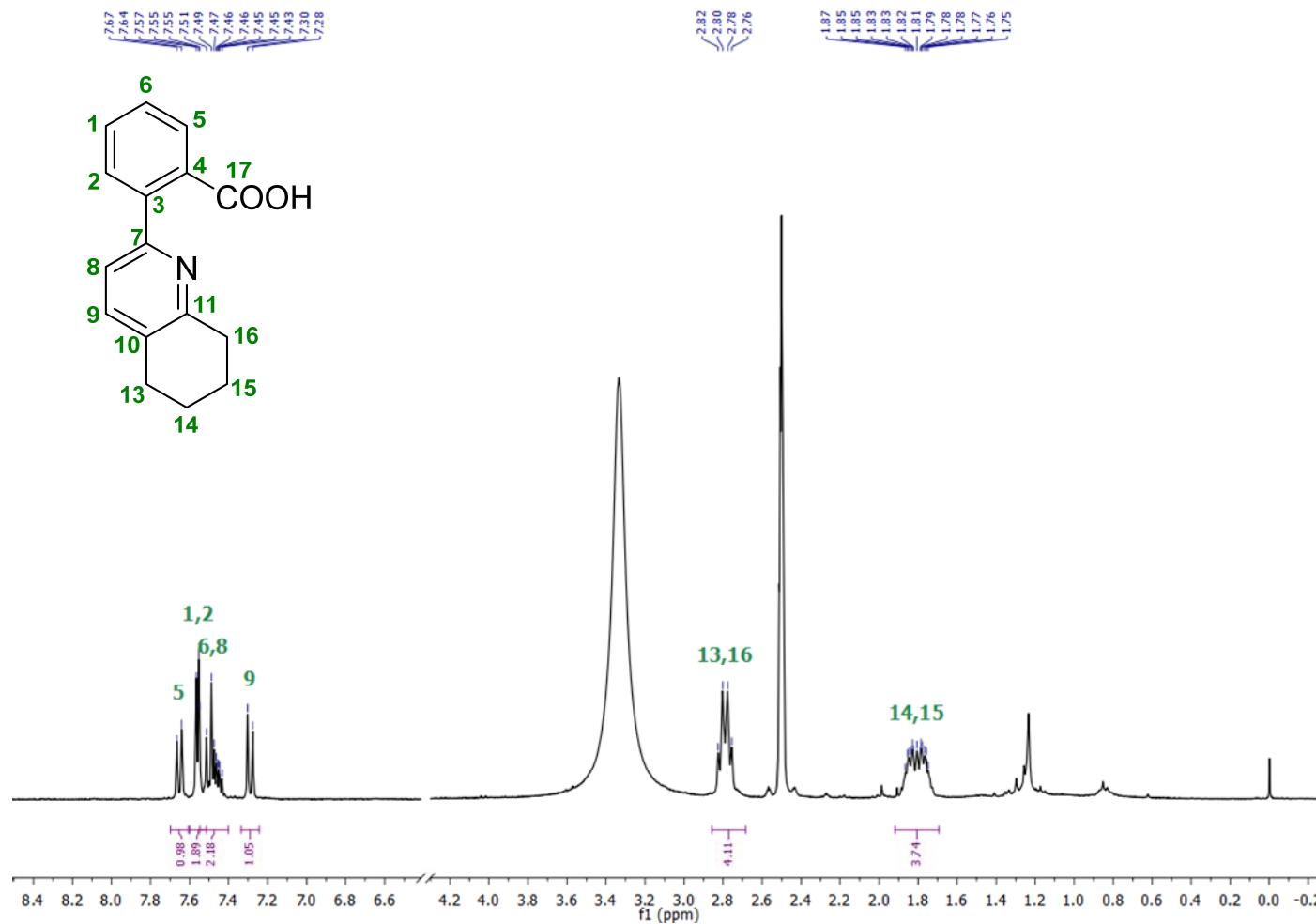
Figure S25. ^{13}C -NMR spectrum of **3** in CDCl_3

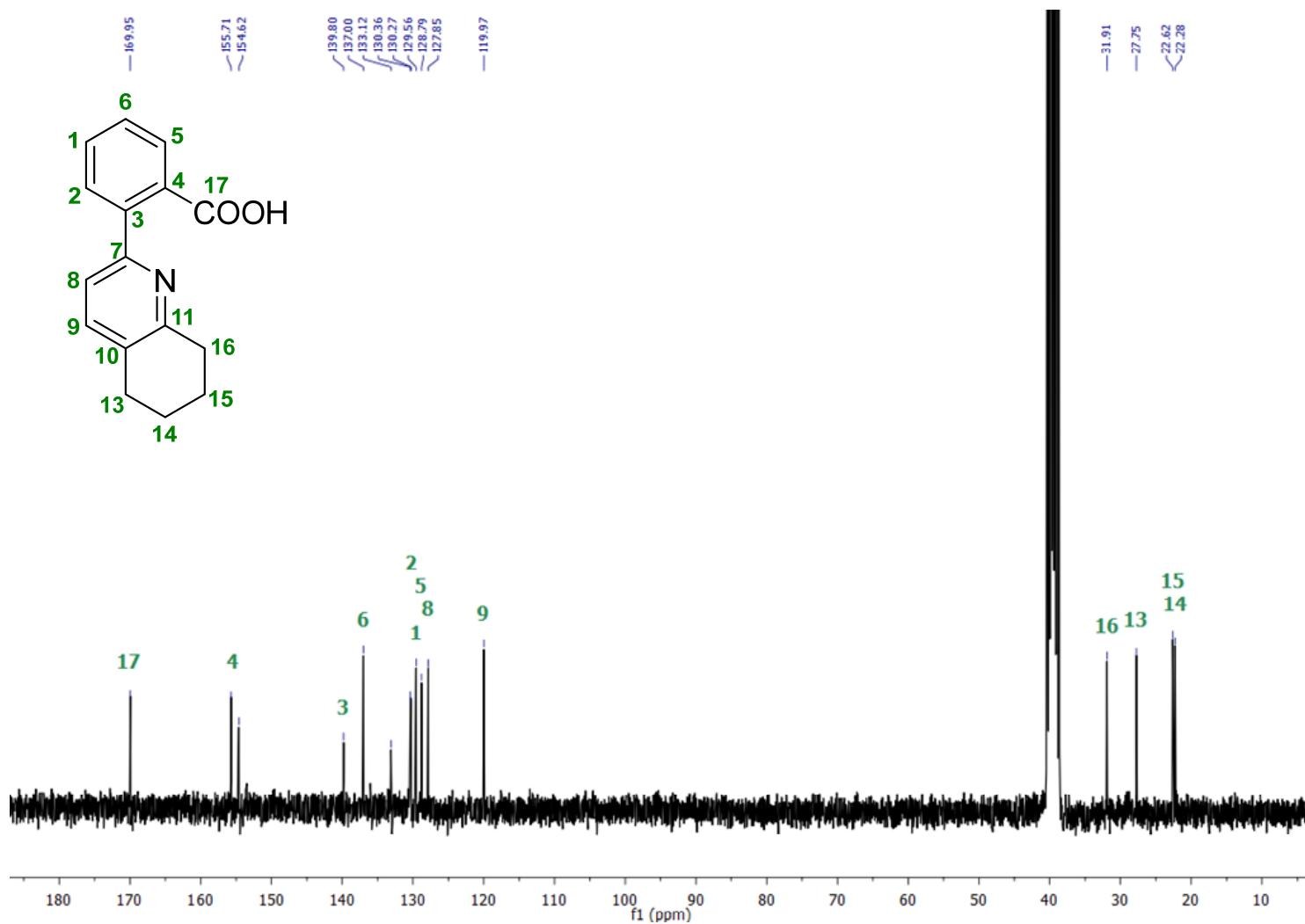
NMR spectra of P4



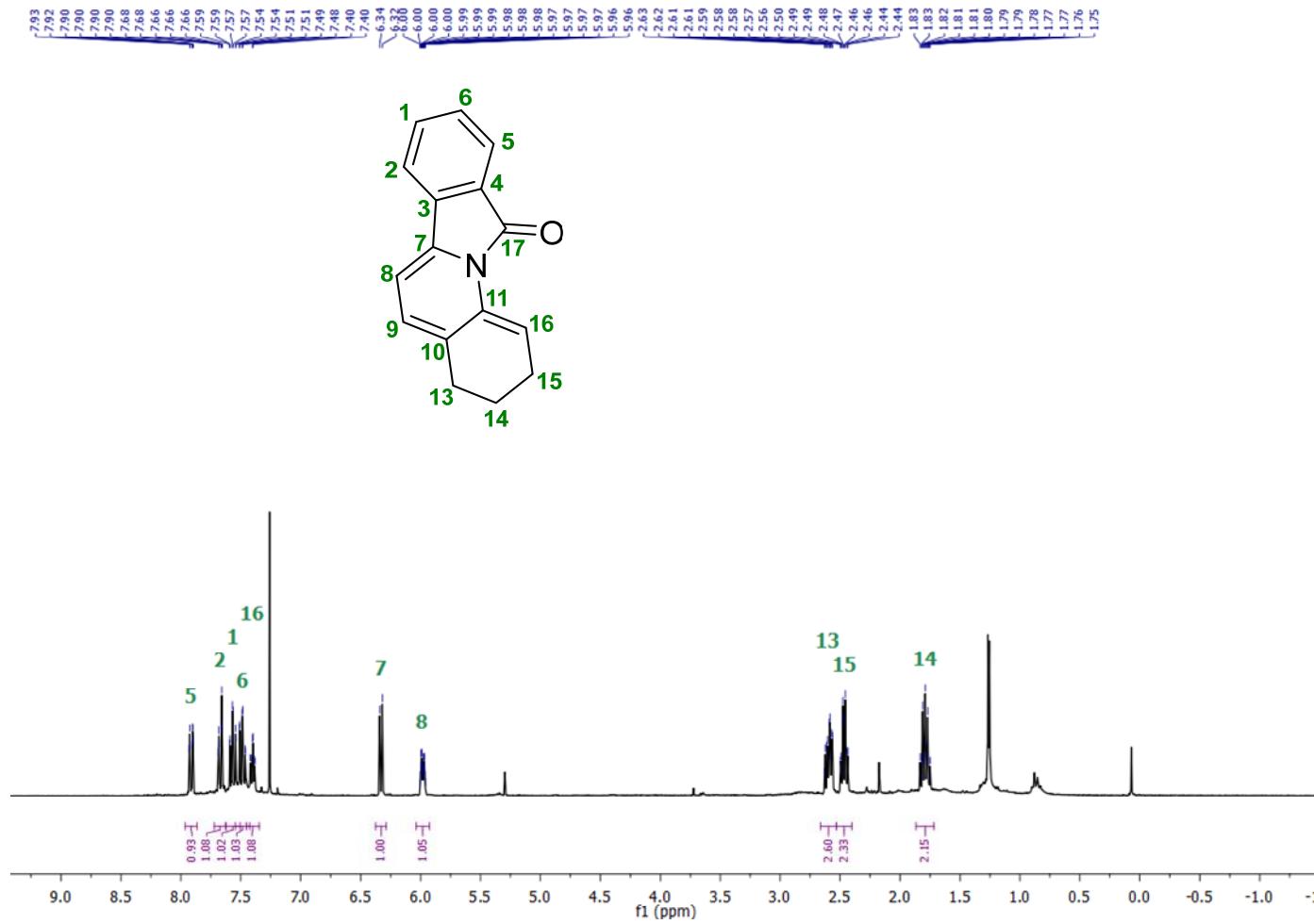
Figure S27. ^{13}C -NMR spectrum of **P4** in CDCl_3

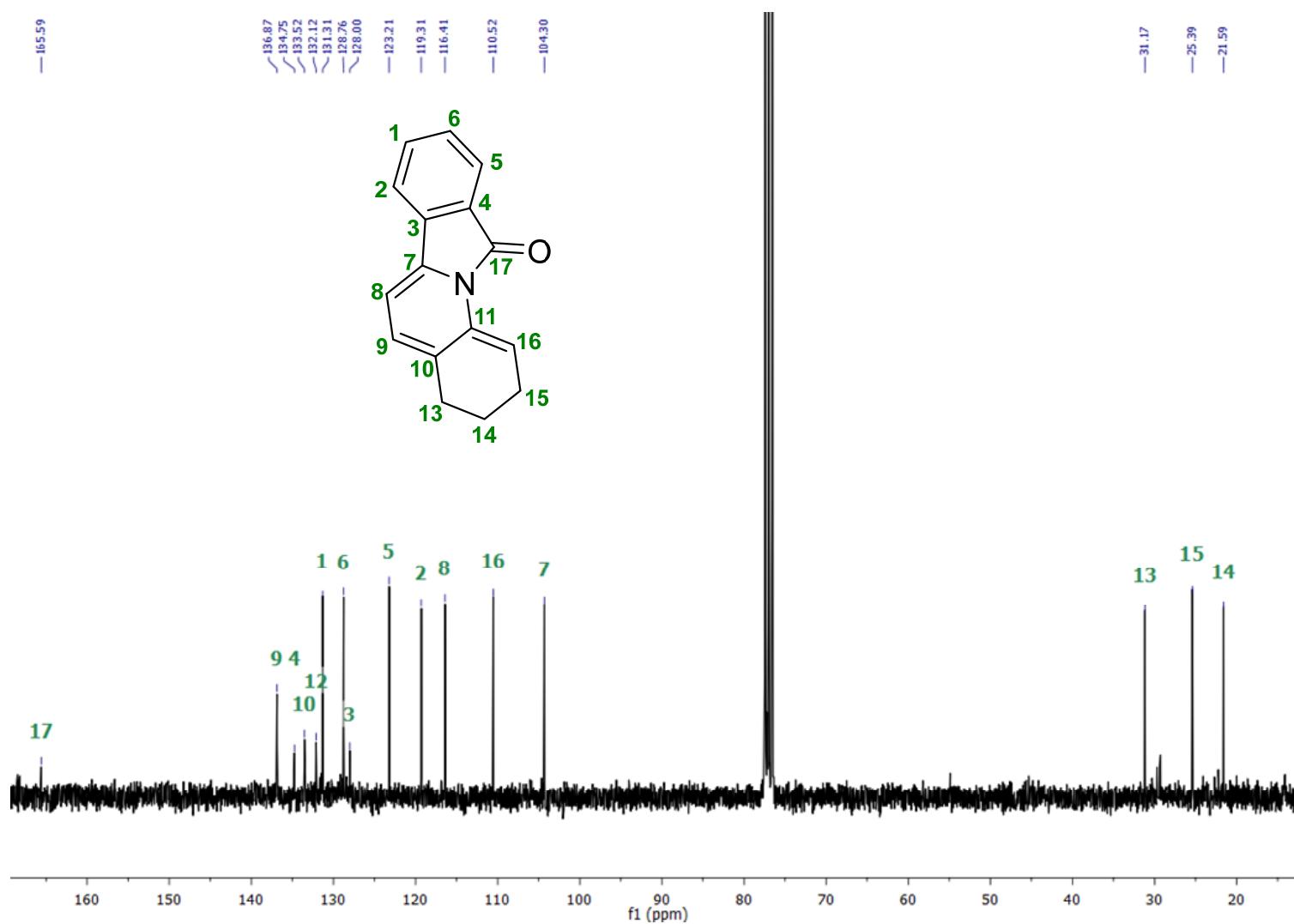
NMR spectra of 4

Figure S28. ^1H -NMR spectrum of 4 in $\text{DMSO}-d_6$

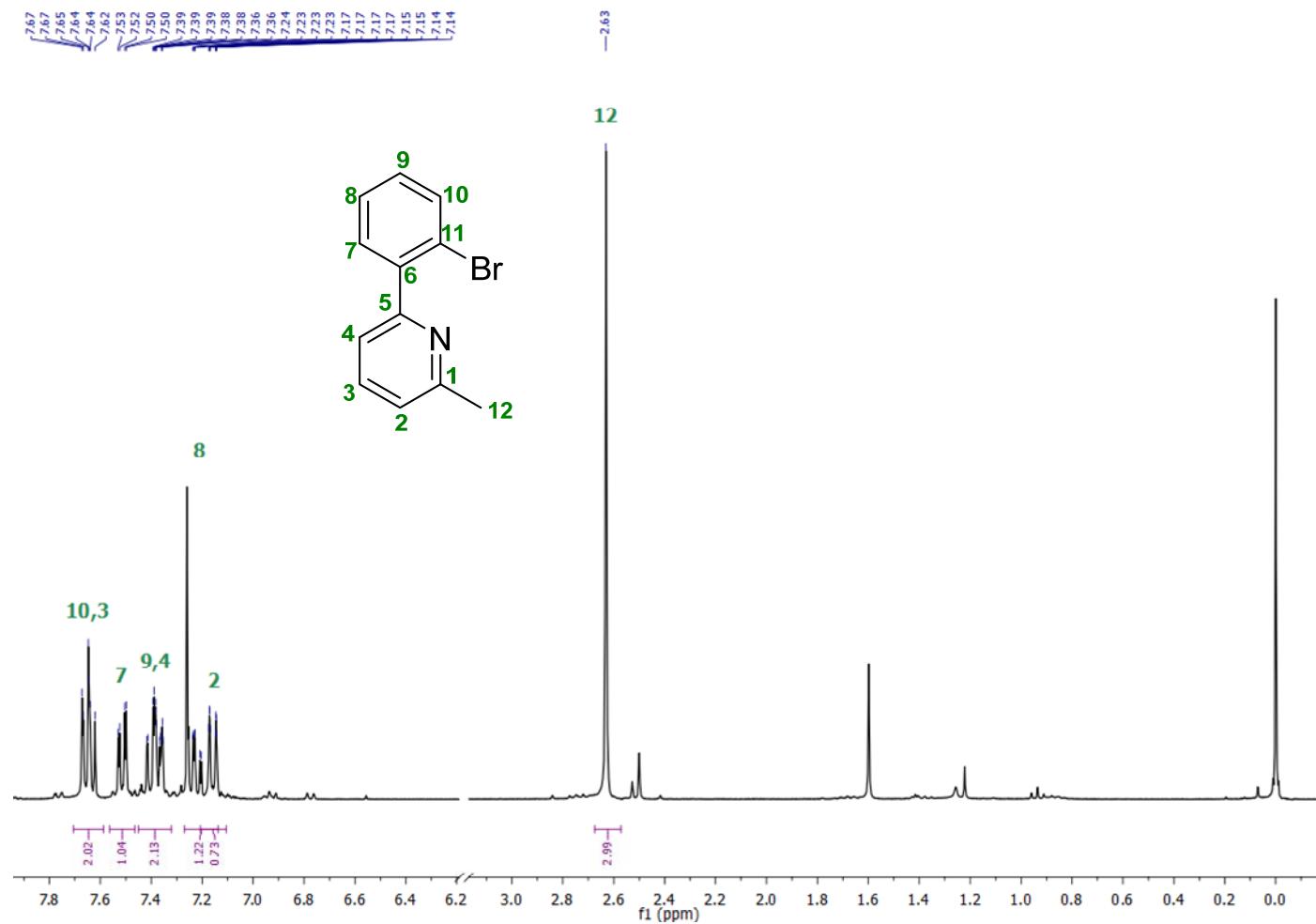
Figure S29. ^{13}C -NMR spectrum of **4** in DMSO-*d*6

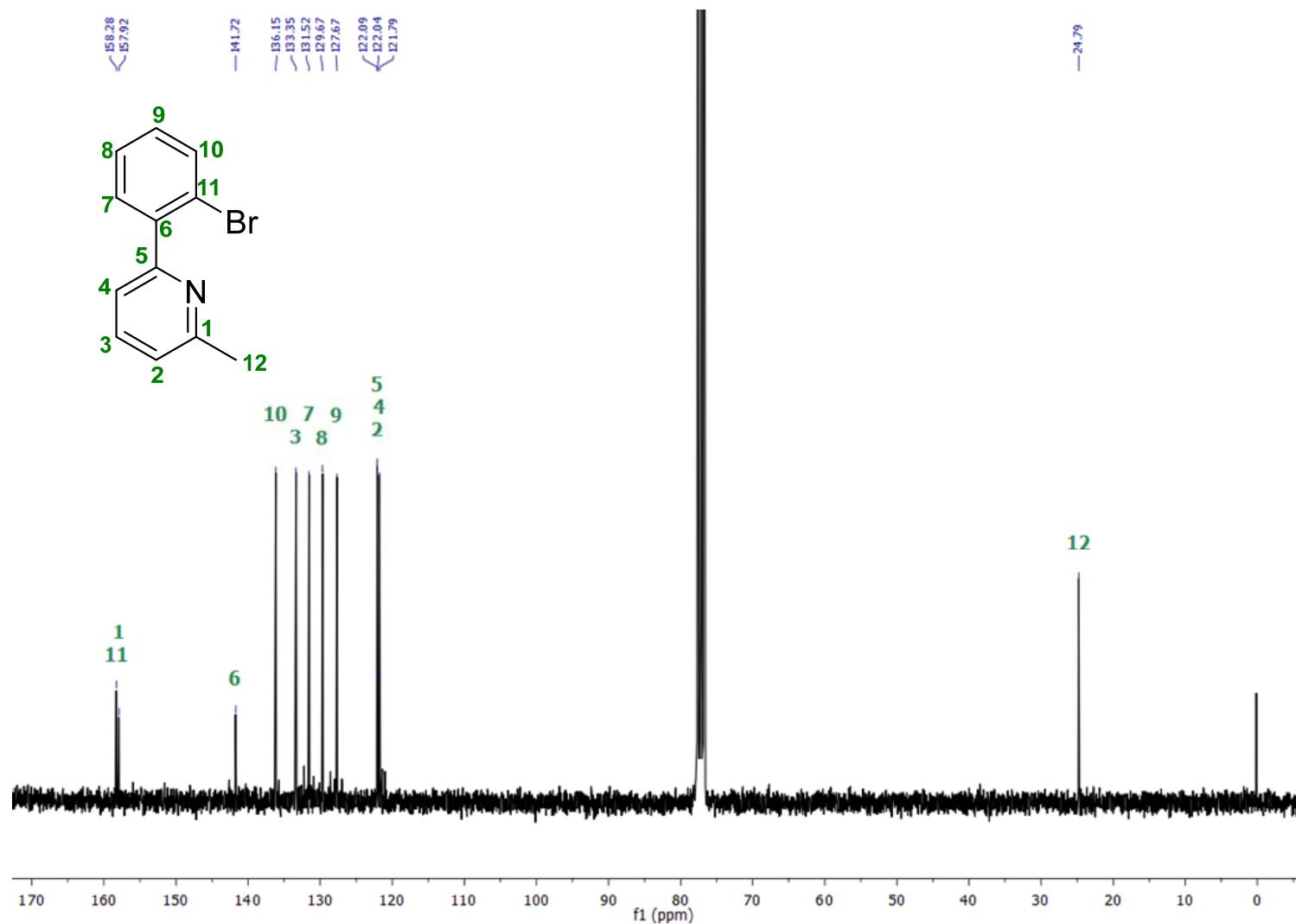
NMR spectra of 5



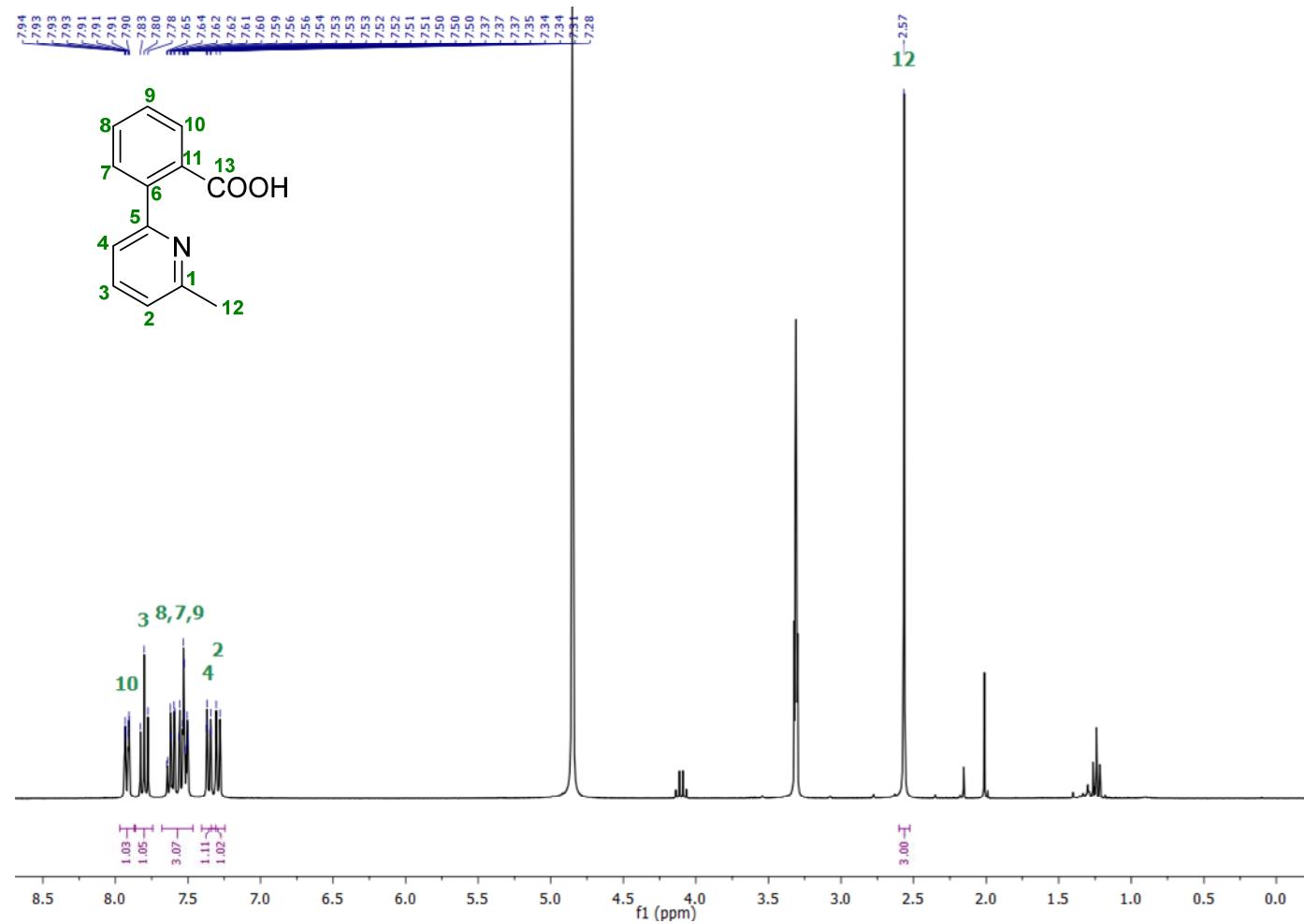
Figure S31. ^{13}C -NMR spectrum of 5 in CDCl_3

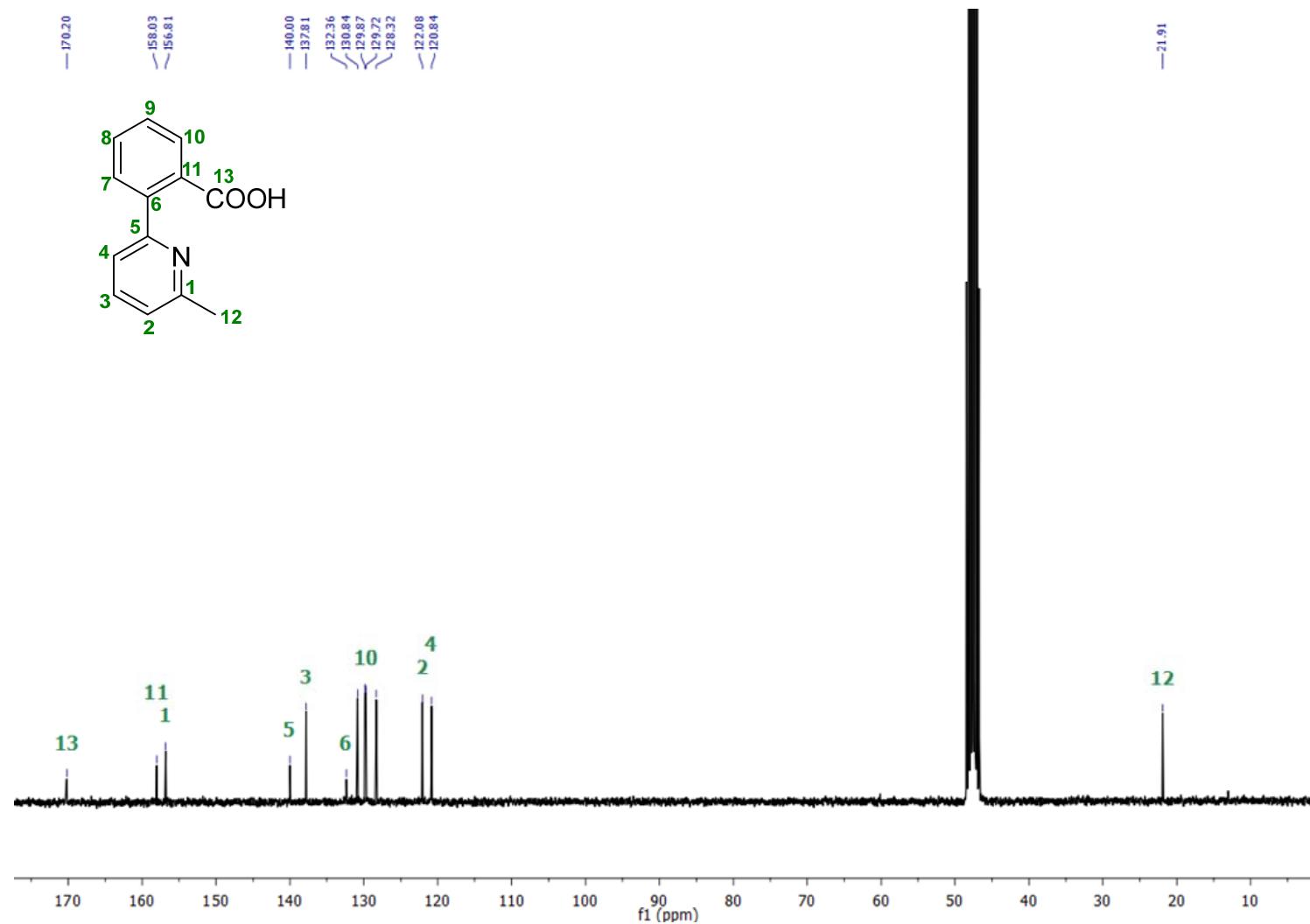
NMR spectra of P6

Figure S32. ^1H -NMR spectrum of P6 in CDCl_3

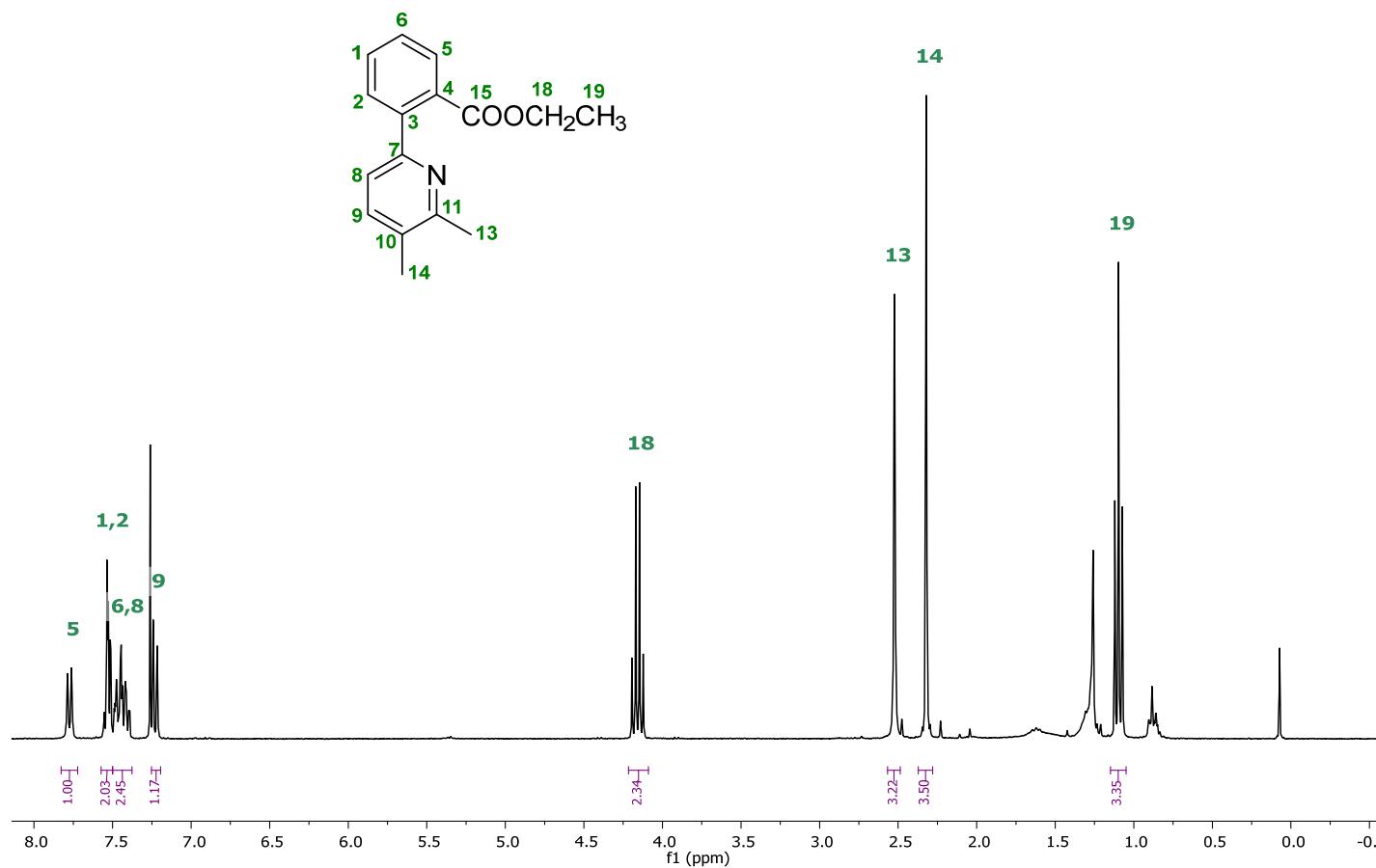
Figure S33. ^{13}C -NMR spectrum of **P6** in CDCl_3

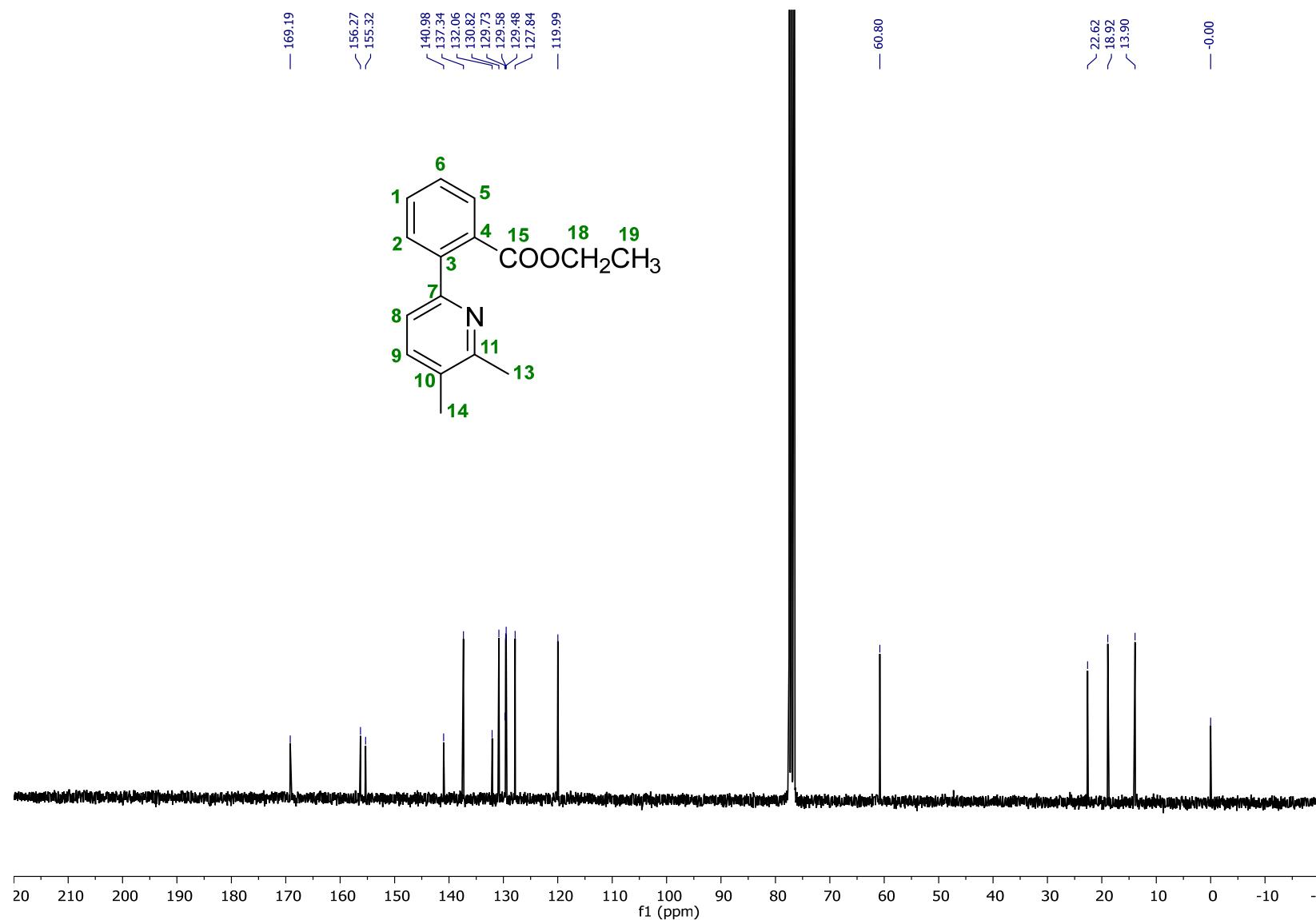
NMR spectra of 6

Figure S34. ^1H -NMR spectrum of 6 in CD_3OD

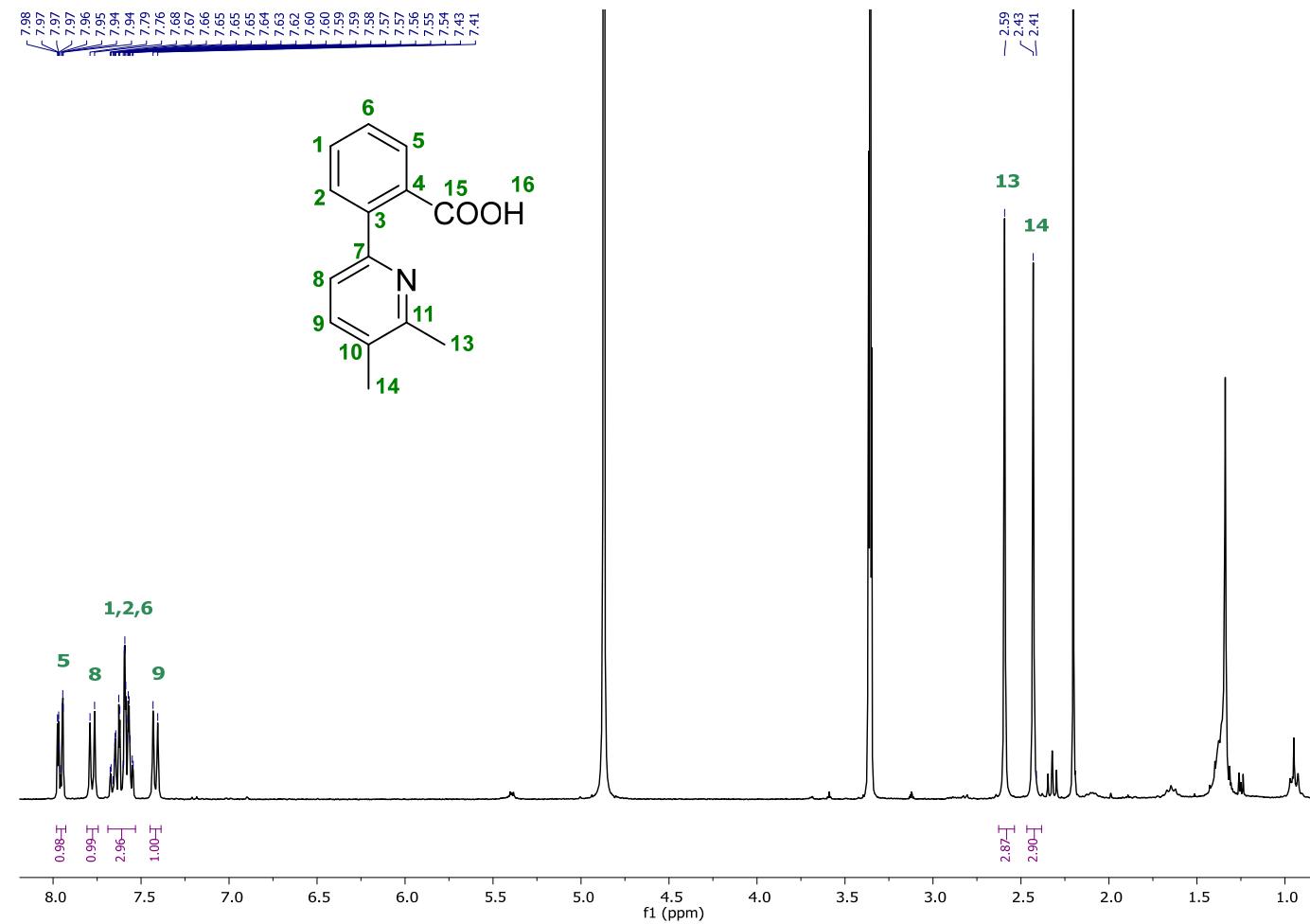
Figure S35. ^{13}C -NMR spectrum of **6** in CD_3OD

NMR spectra of P7

Figure S36. ^1H -NMR spectrum of **P7** in CDCl_3

Figure S37. ^{13}C -NMR spectrum of **P7** in CDCl_3

NMR spectra of 7

Figure S38. ^1H -NMR spectrum of 7 in CD_3OD

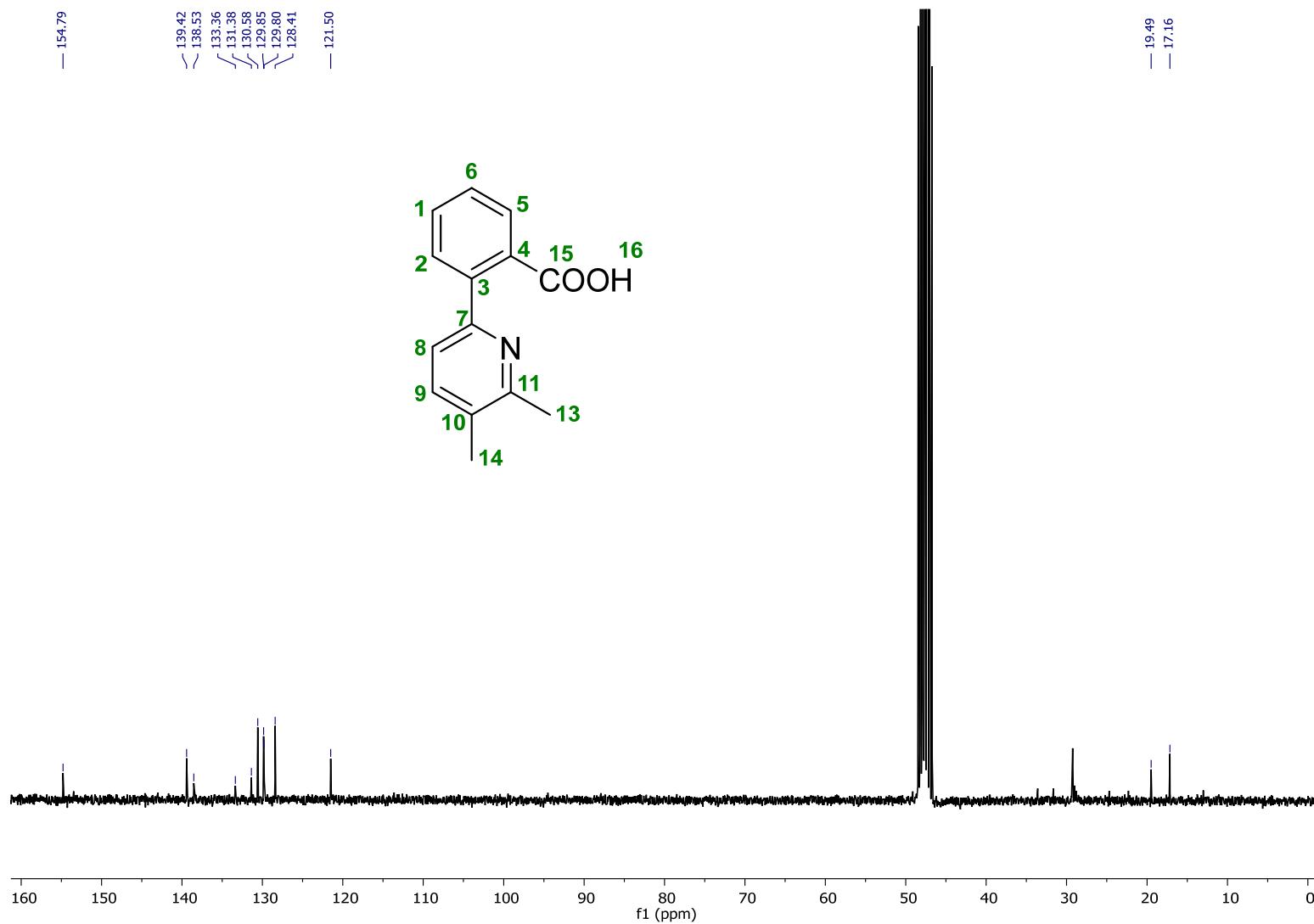
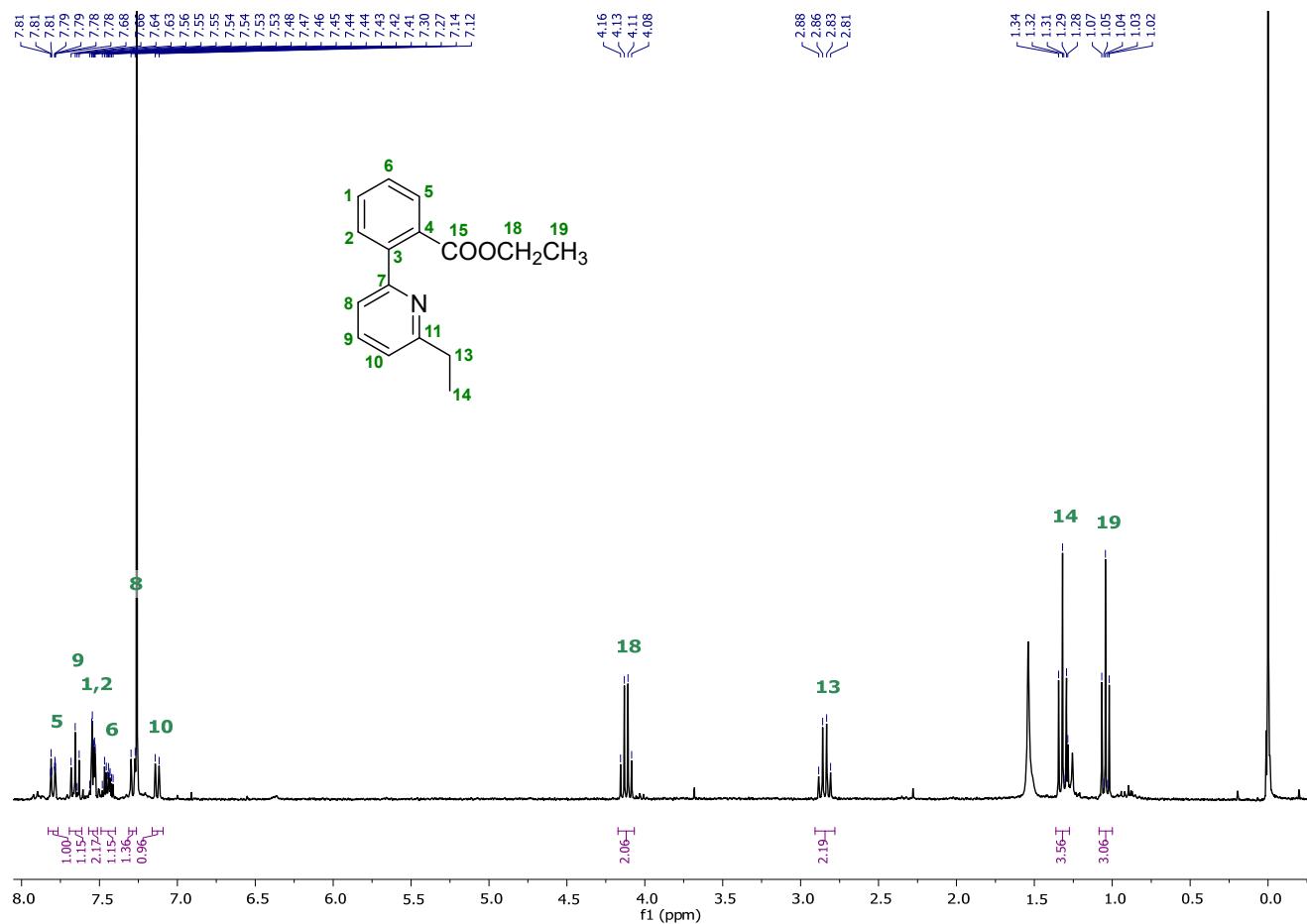


Figure S39. ^{13}C -NMR spectrum of **7** in CD_3OD

NMR spectra of P8

Figure S40. ^1H -NMR spectrum of P8 in CDCl_3

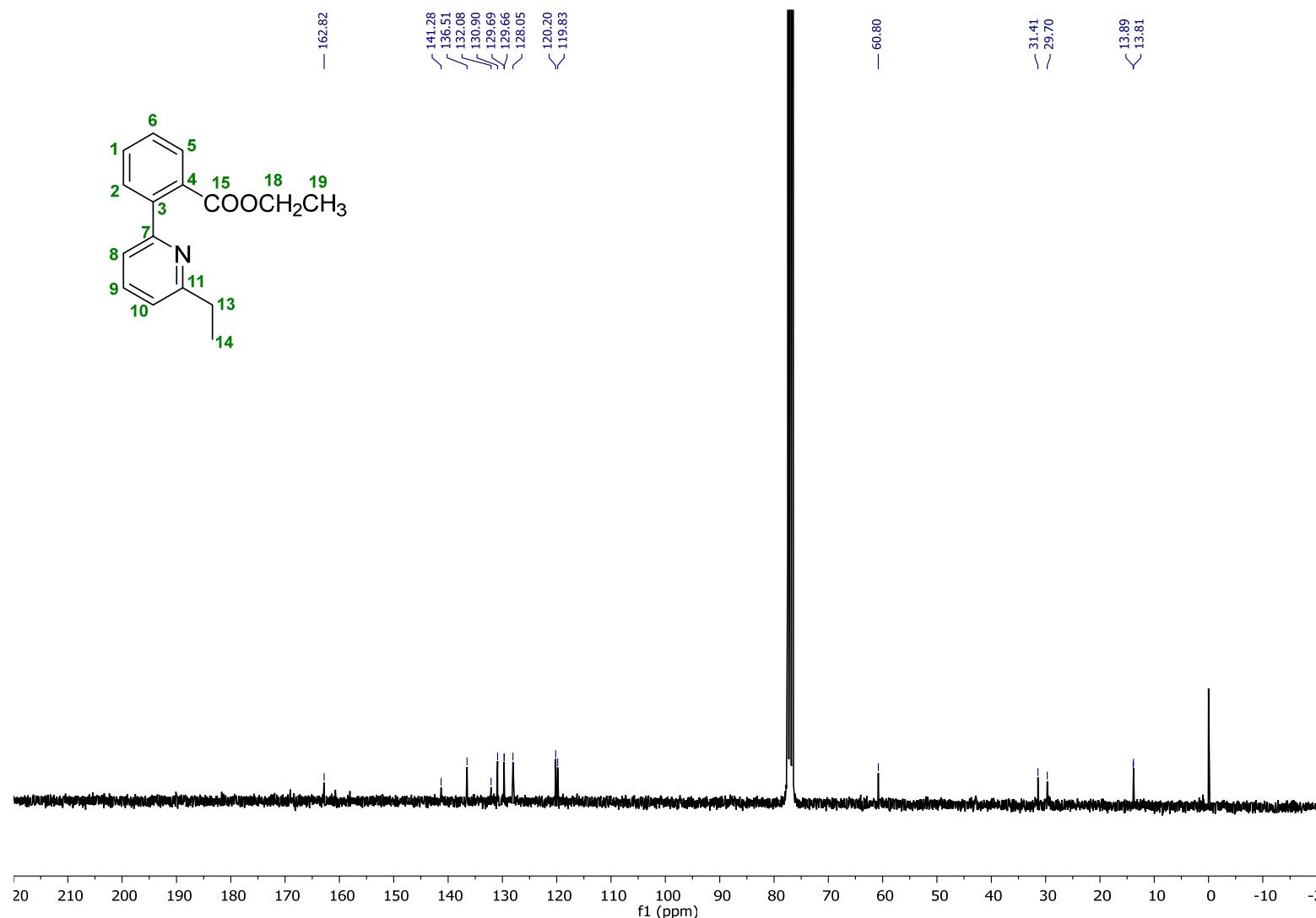
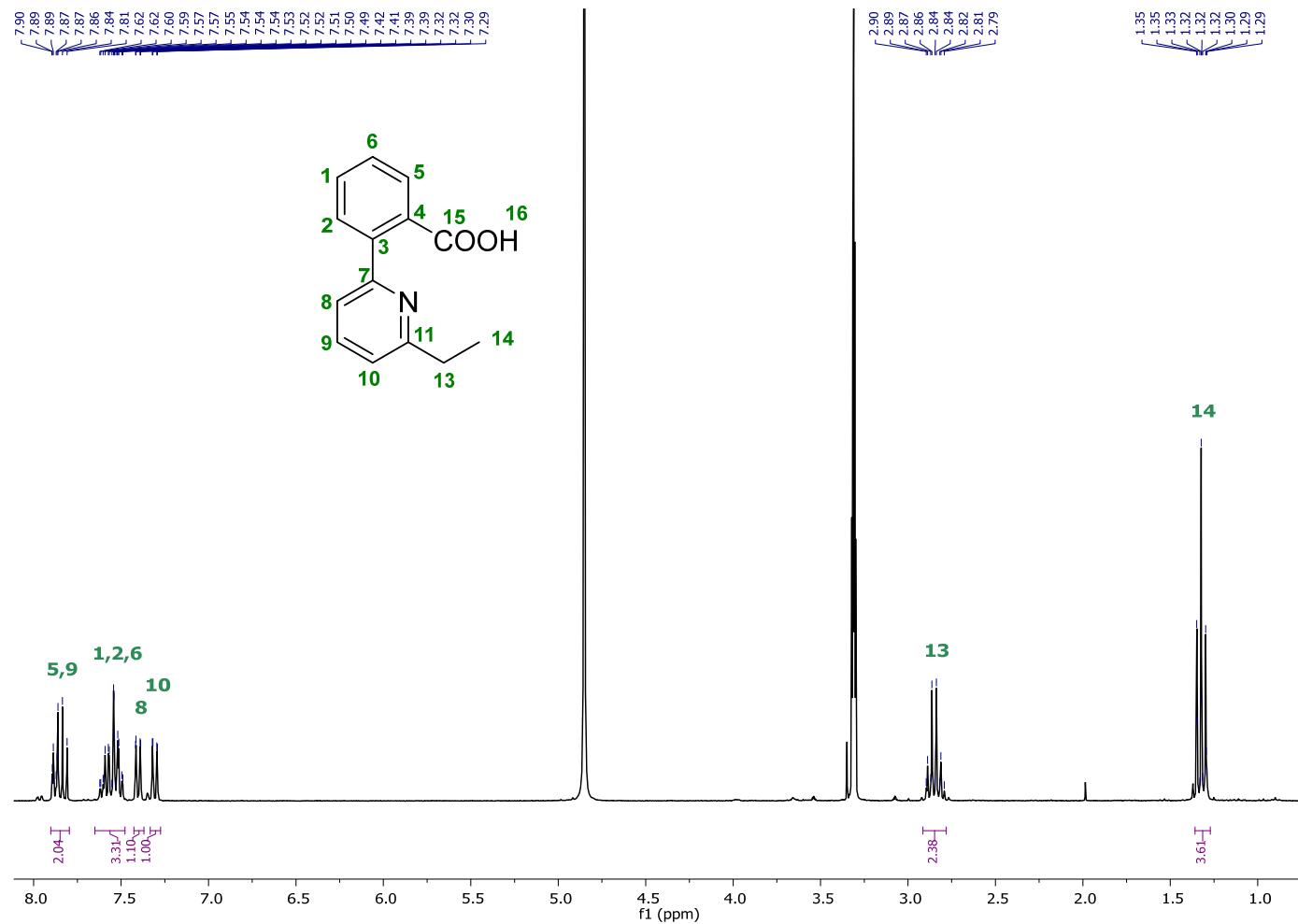
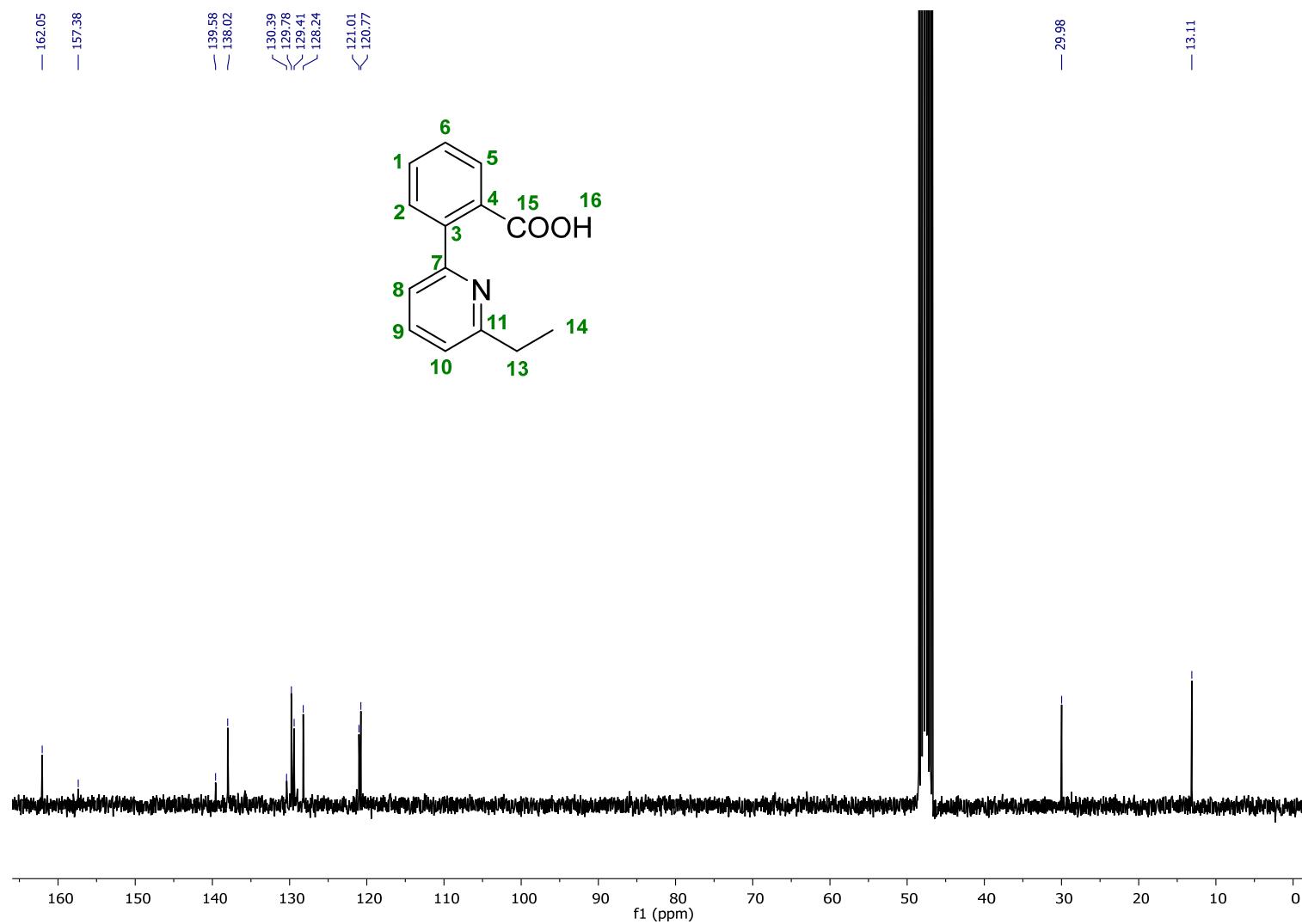


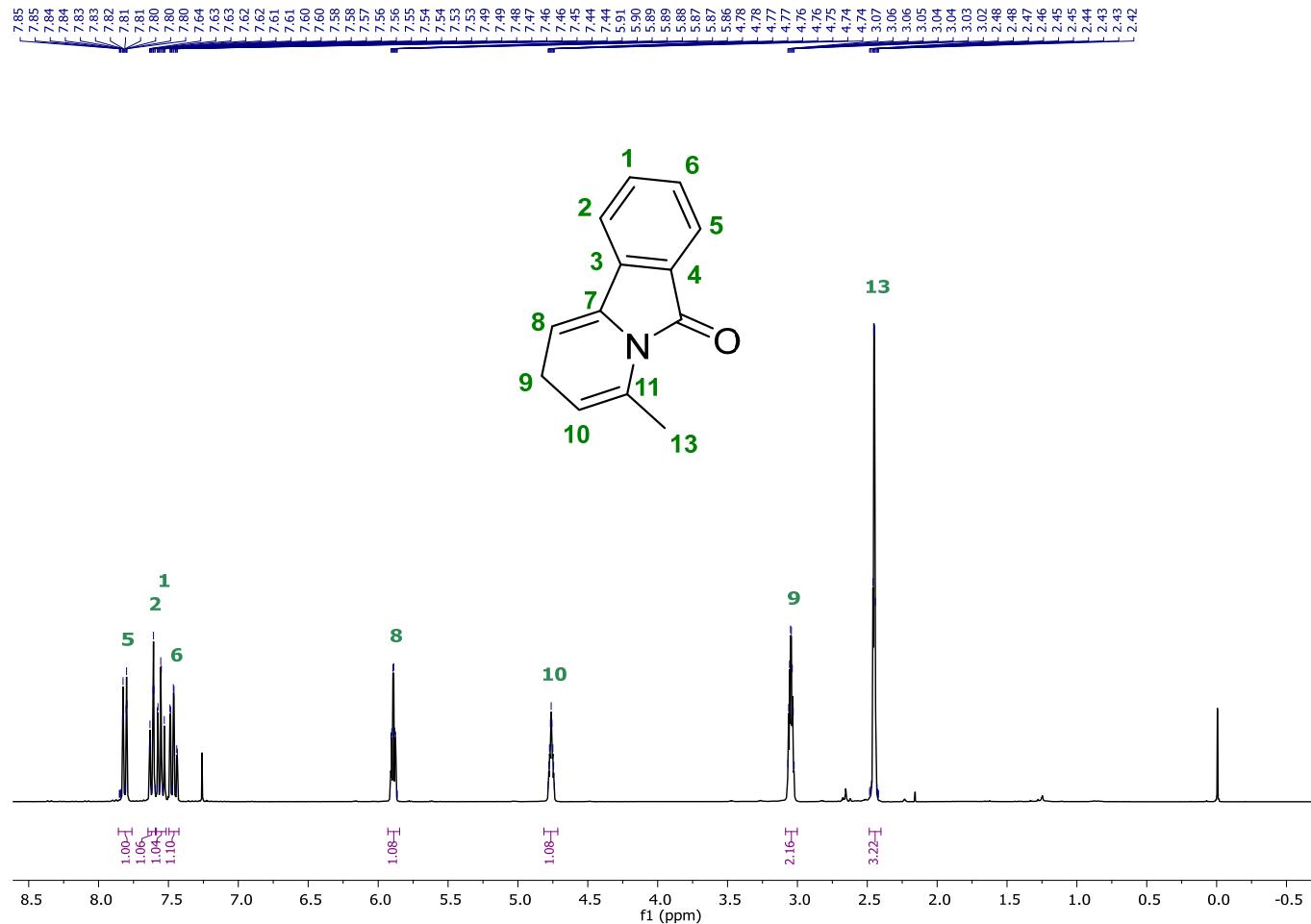
Figure S41. ^{13}C -NMR spectrum of **P8** in CDCl_3

NMR spectra of 8

Figure S42. ^1H -NMR spectrum of **8** in CD_3OD

Figure S43. ^{13}C -NMR spectrum of **8** in CD_3OD

NMR spectra of 14

Figure S44. ¹H-NMR spectrum of 14 in CDCl_3

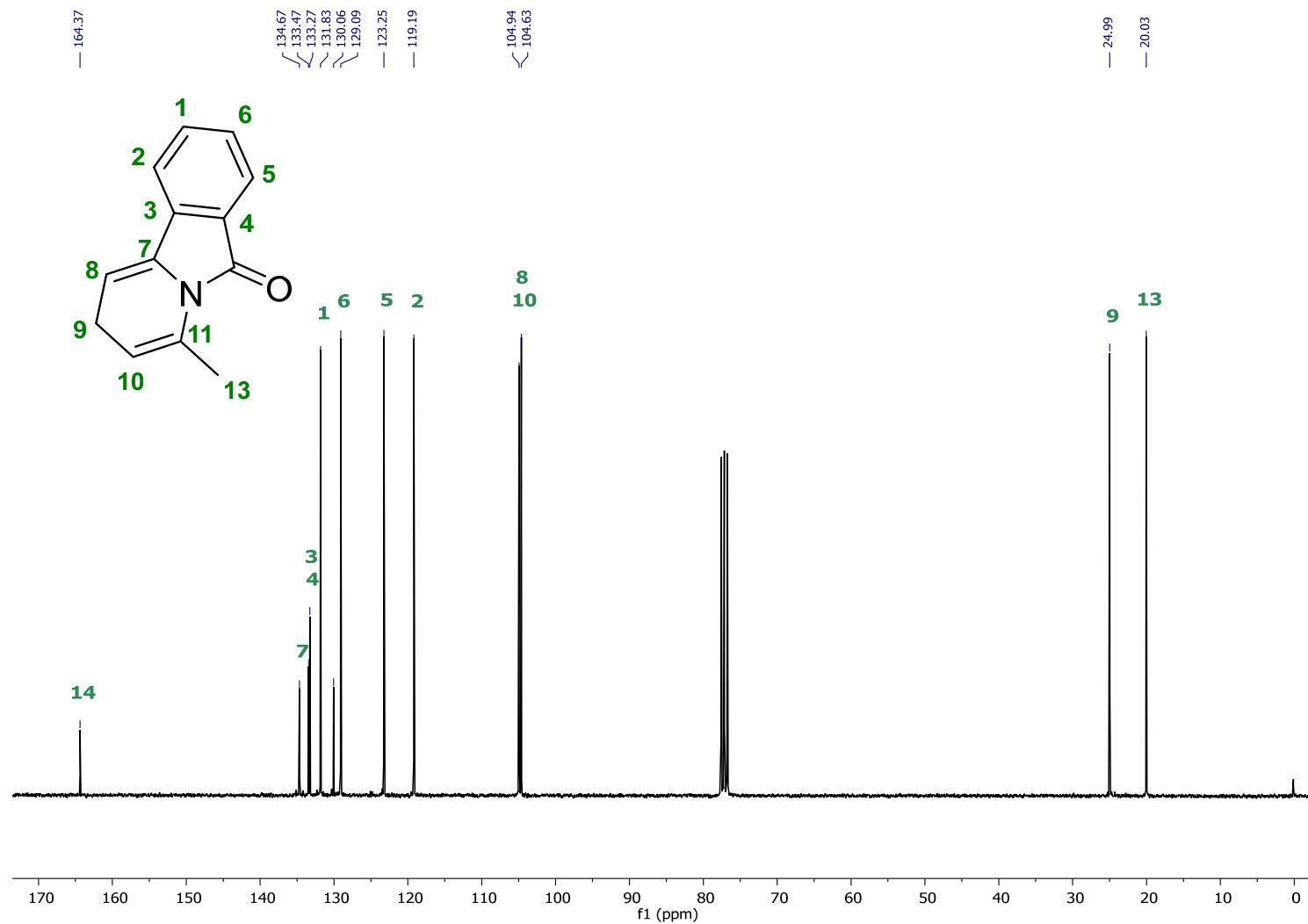
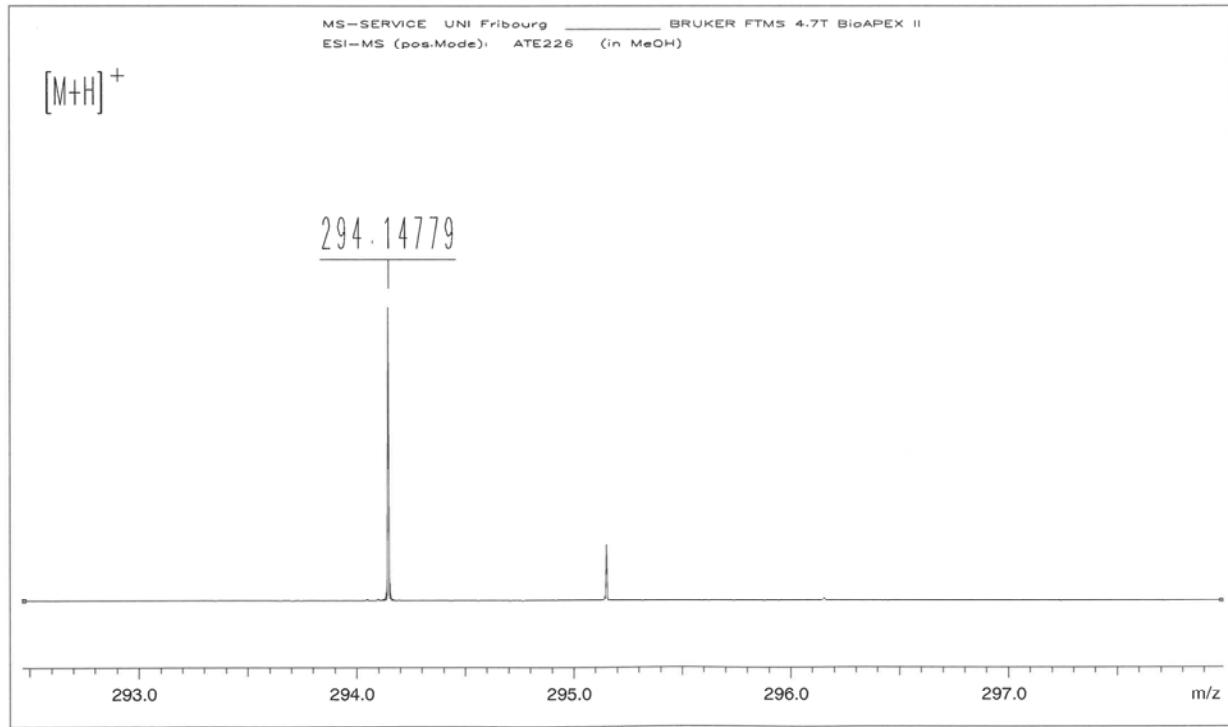


Figure S45. ^{13}C -NMR spectrum of **14** in CDCl_3

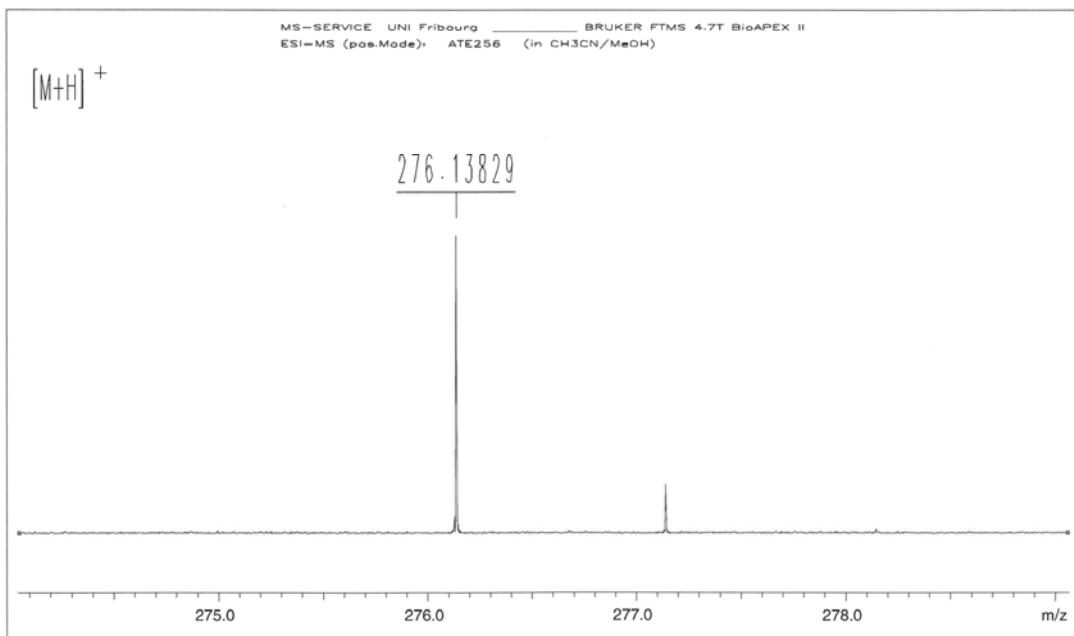
HRMS spectra**HRMS spectrum of 2****Ion mass = 294.1477860****Charge = +1**

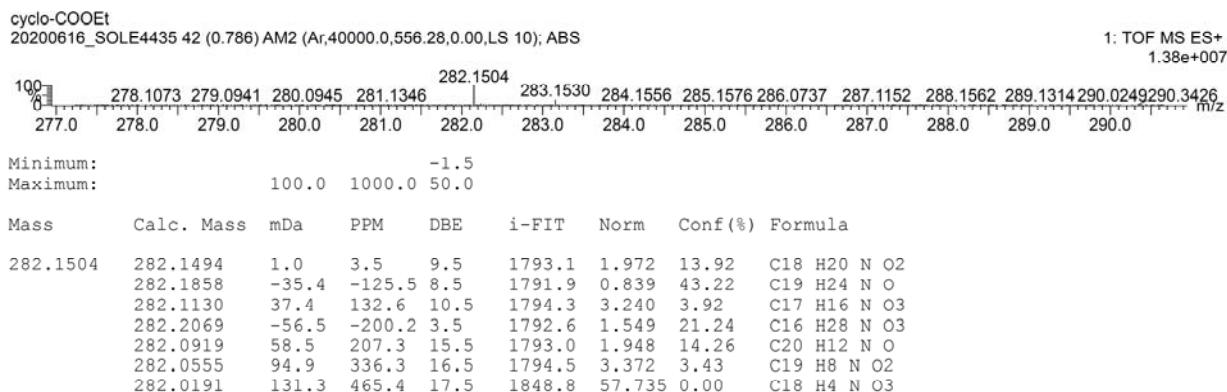
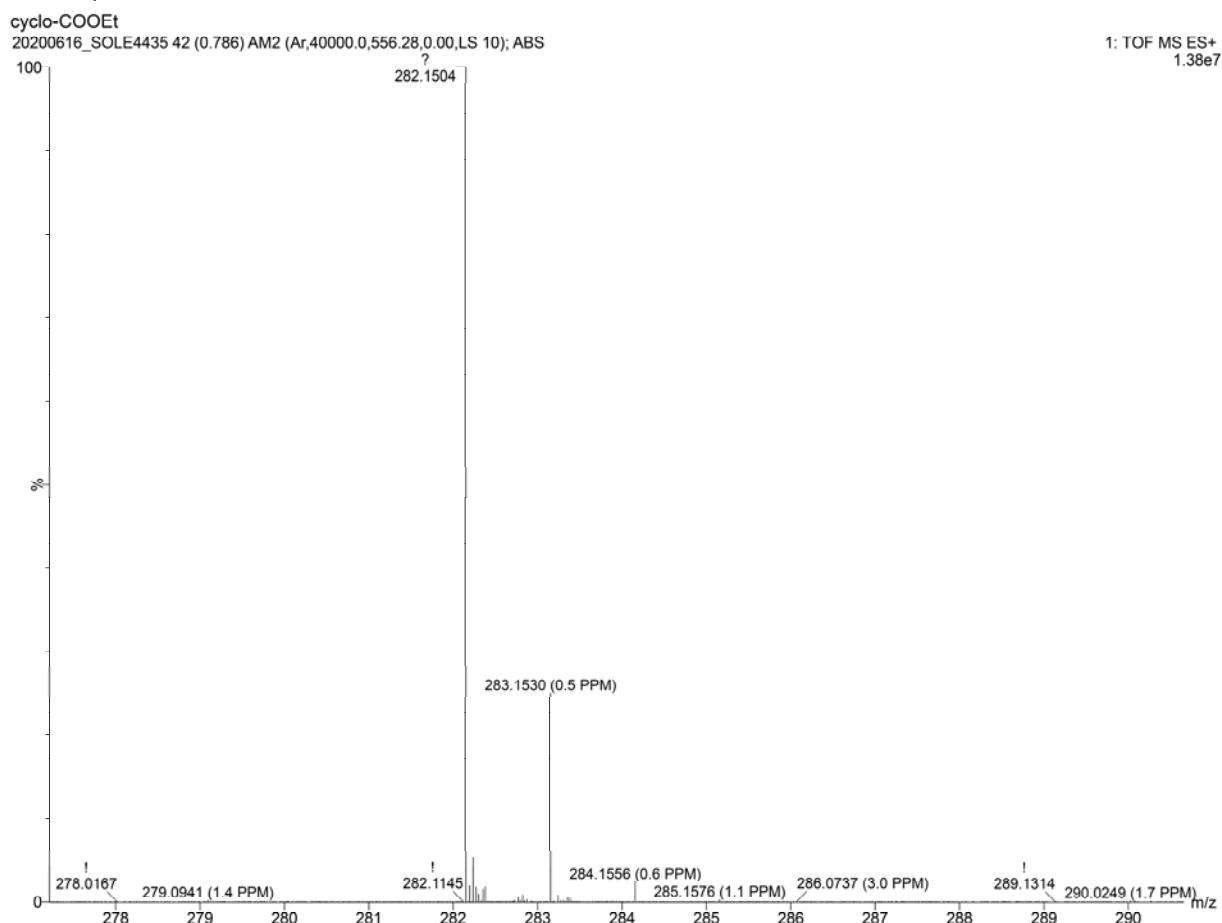
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3	14	20	3	4	294.1448326	6.5	2.953e-03
4	11	22	2	7	294.1421525	2.0	5.634e-03
5	12	24	1	7	294.1547286	1.5	6.943e-03
6	8	24	1	10	294.1394724	-2.5	8.314e-03
7	15	22	2	4	294.1574086	6.0	9.623e-03
8	18	18	2	2	294.1362793	11.0	1.151e-02
9	18	20	3	1	294.1600887	10.5	1.230e-02
10	15	20	1	5	294.1335992	6.5	1.419e-02

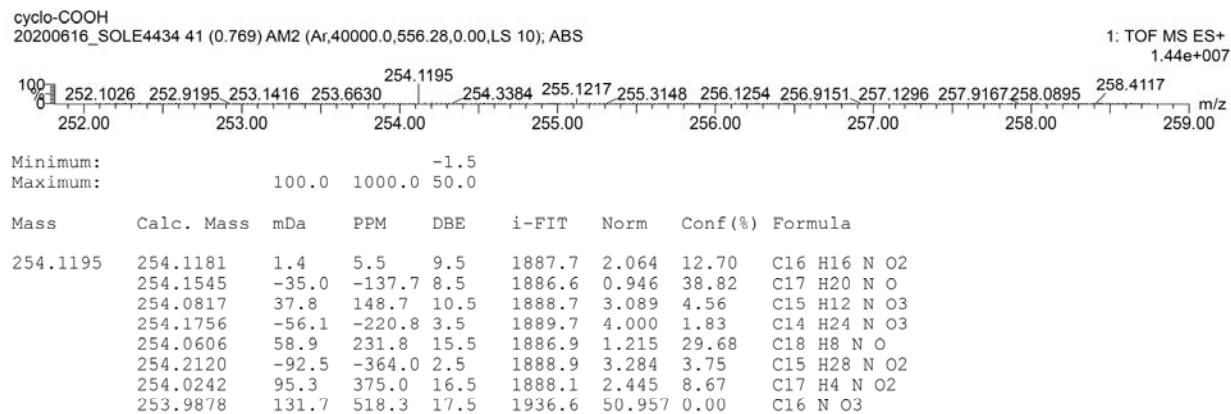
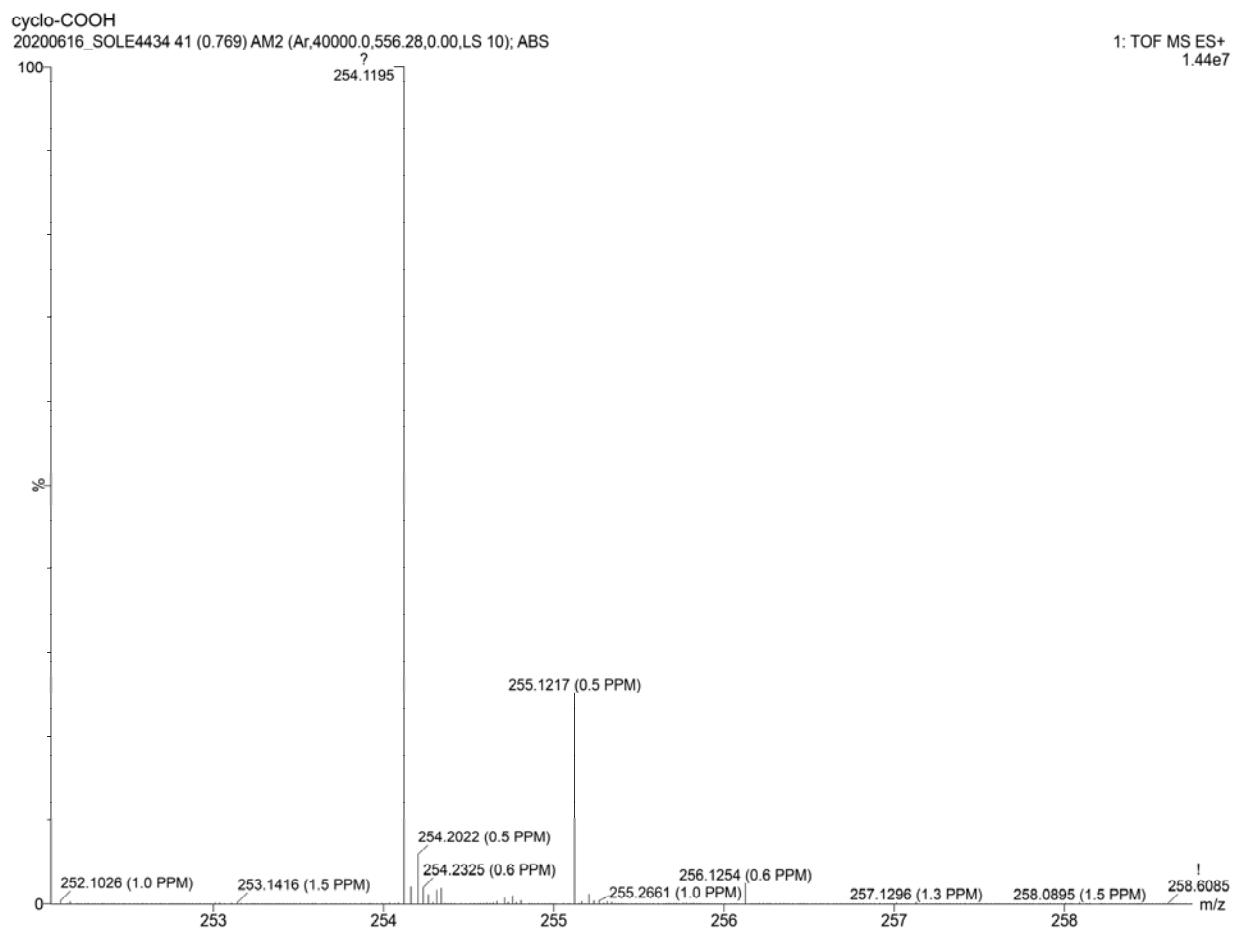


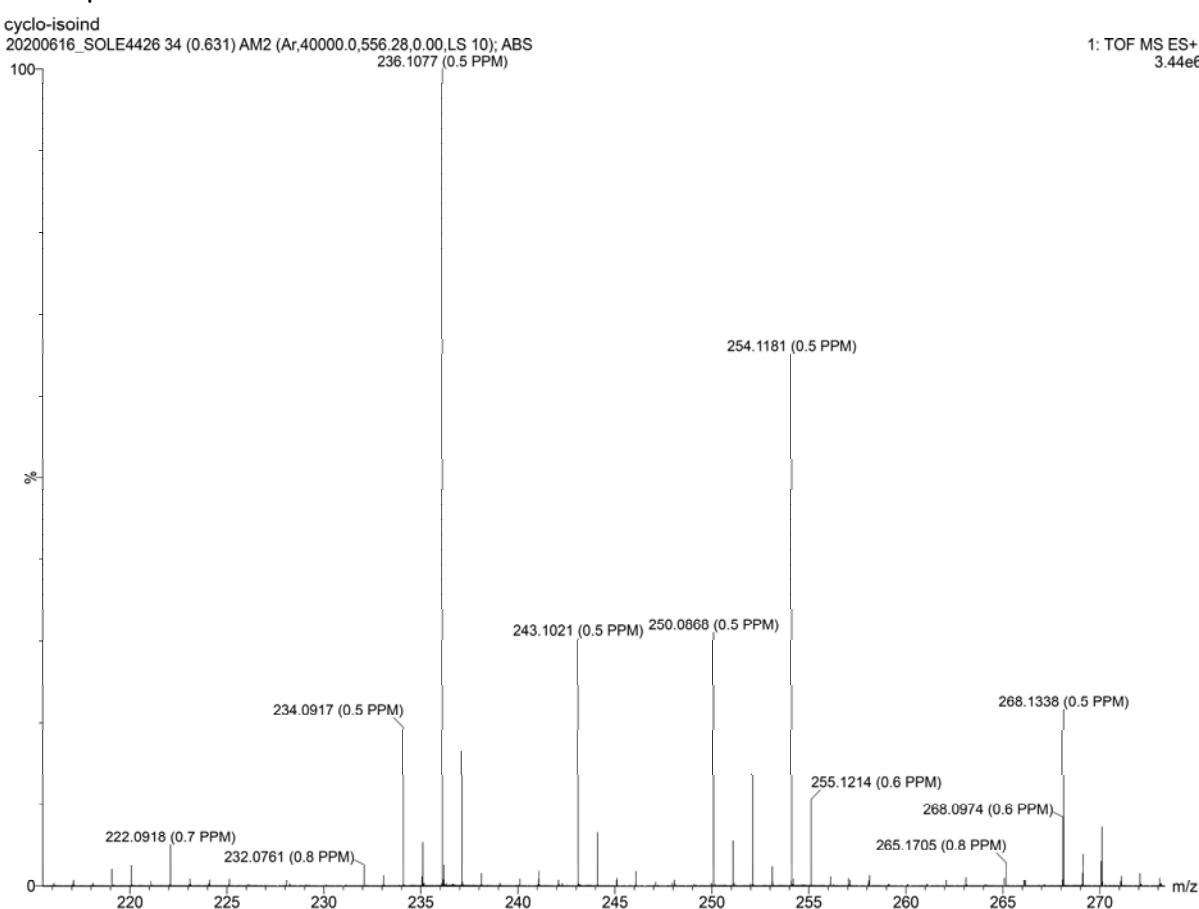
HRMS spectrum of 3**Ion mass = 276.1382870****Charge = +1**

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3	14	18	3	3	276.1342679	7.5	4.019e-03
4	12	22	1	6	276.1441639	2.5	5.877e-03
5	11	20	2	6	276.1315878	3.0	6.699e-03
6	15	20	2	3	276.1468439	7.0	8.557e-03
7	8	22	1	9	276.1289077	-1.5	9.379e-03
8	18	16	2	1	276.1257146	12.0	1.257e-02
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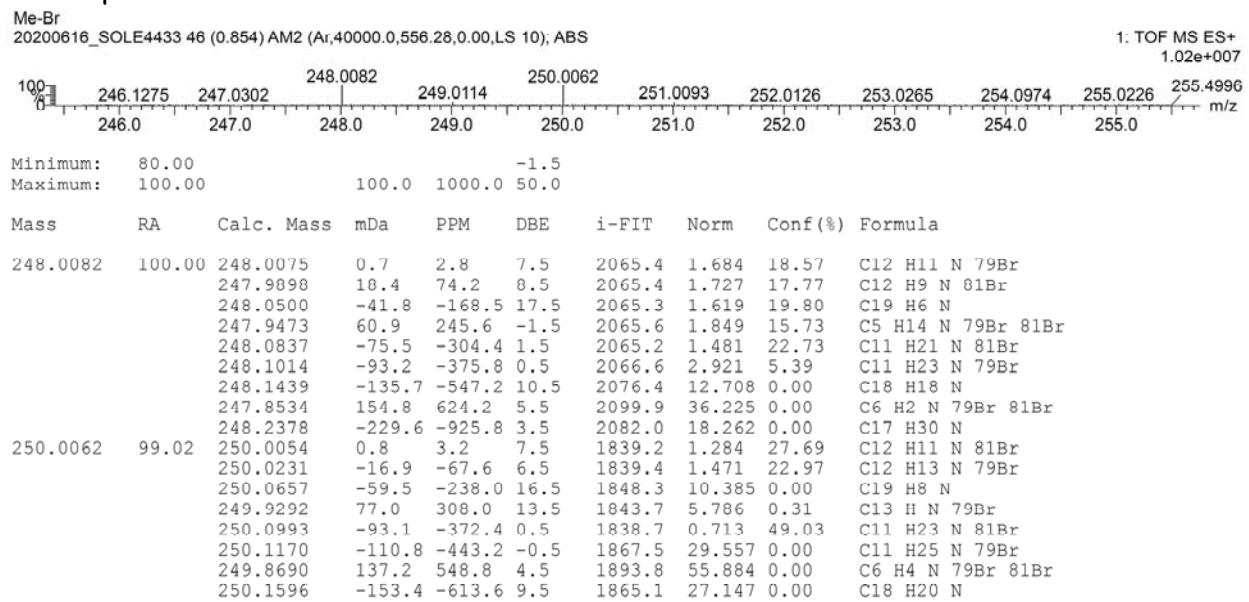


HRMS spectrum of P4

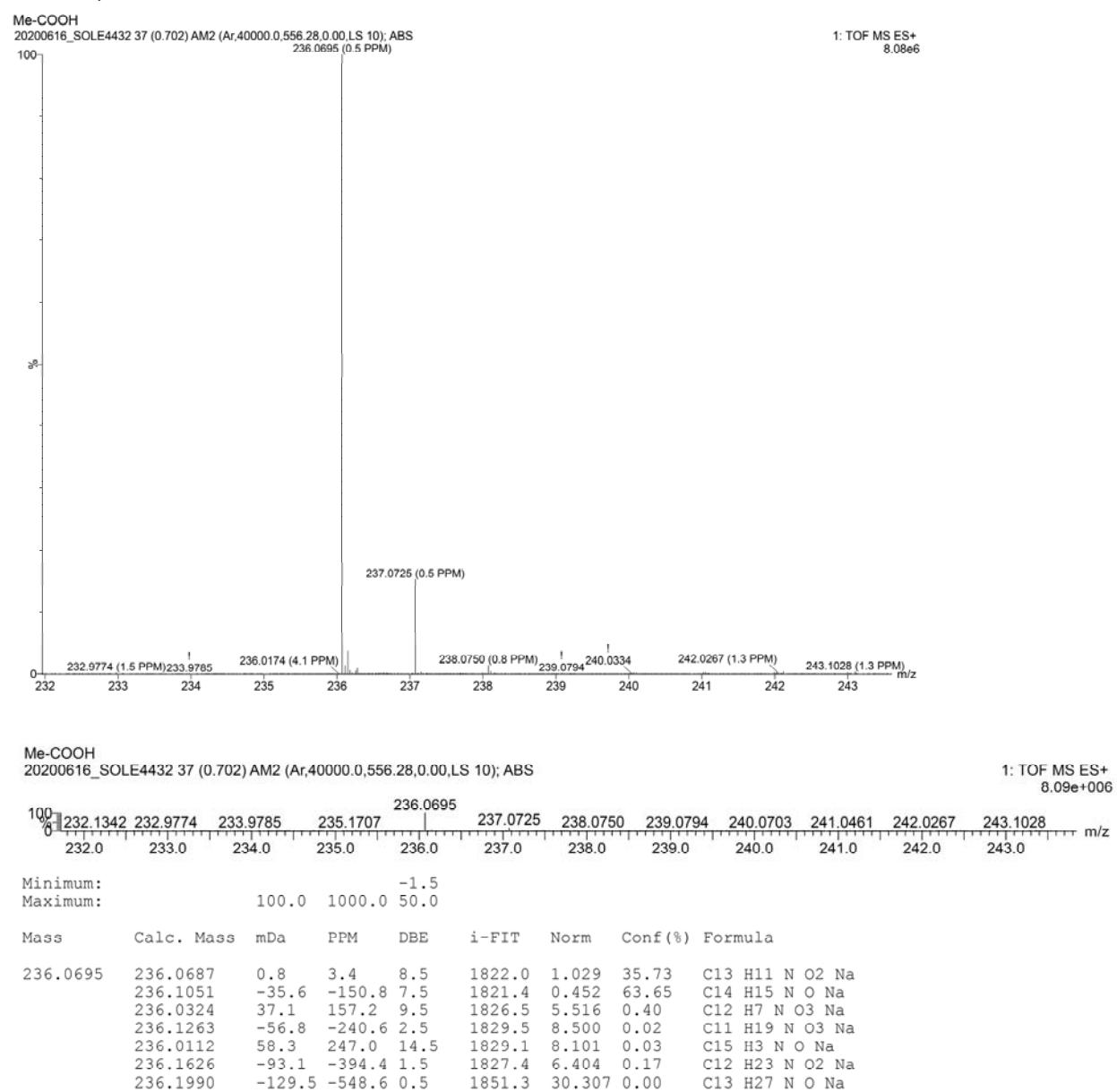
HRMS spectrum of 4

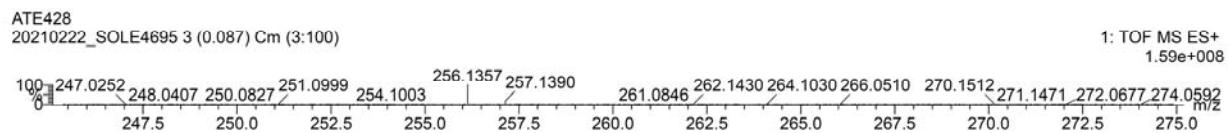
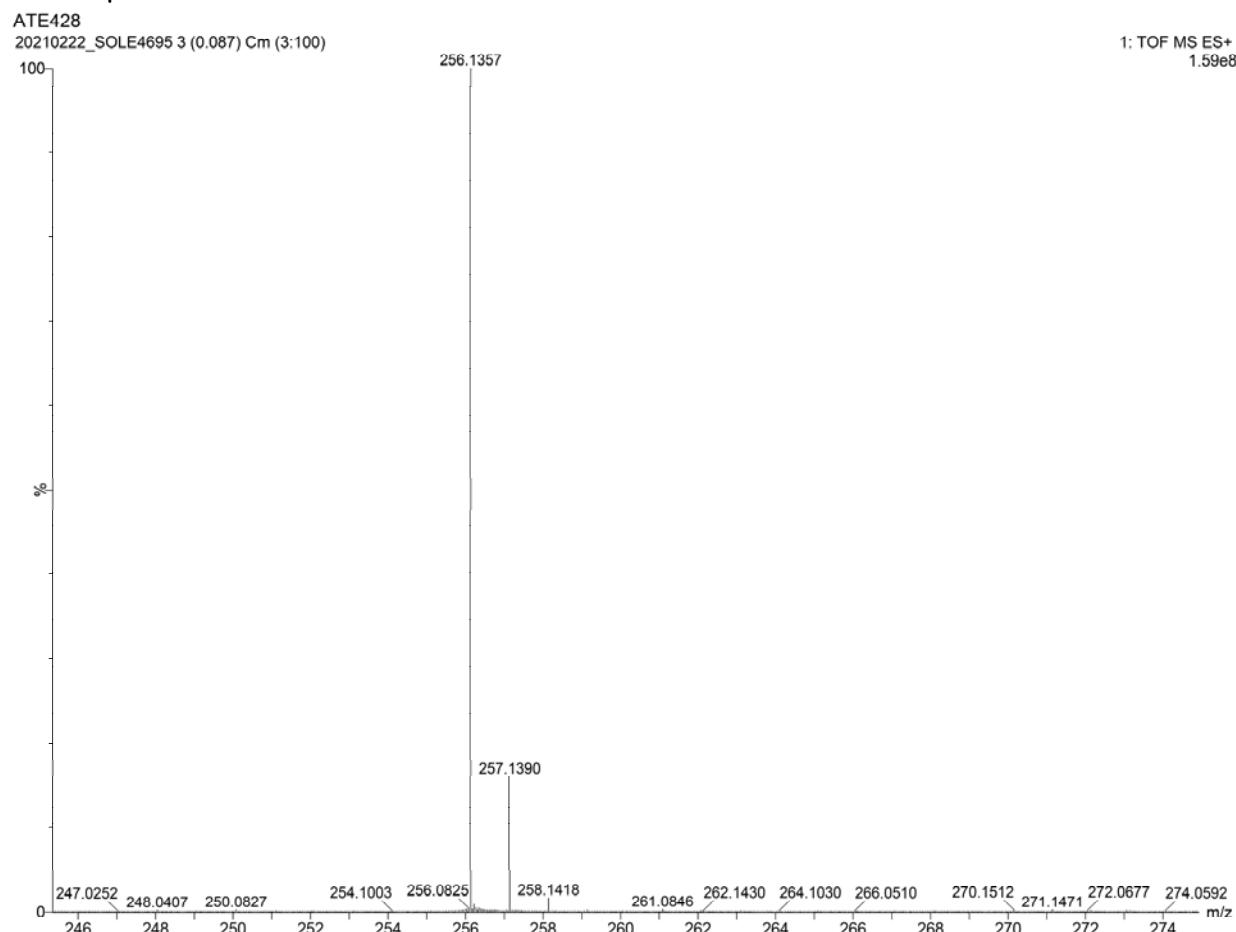
HRMS spectrum of 5

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

HRMS spectrum of P6

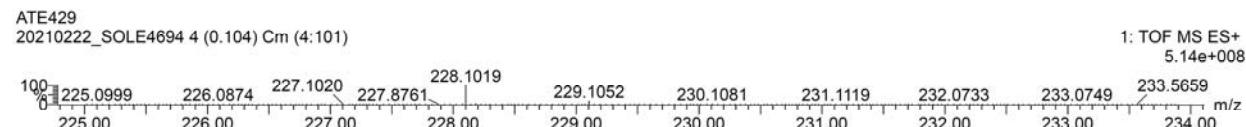
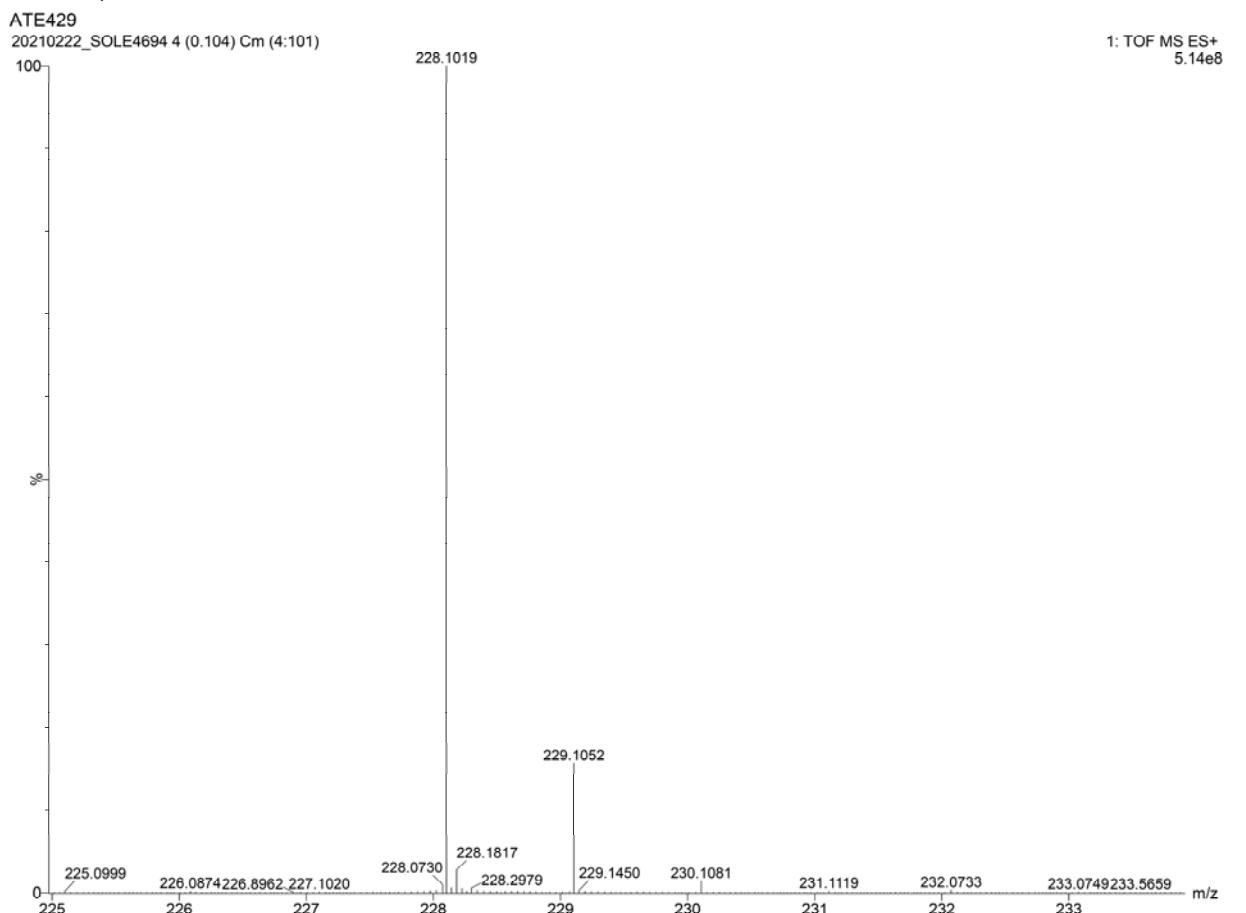
Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

HRMS spectrum of 6

HRMS spectrum of P7

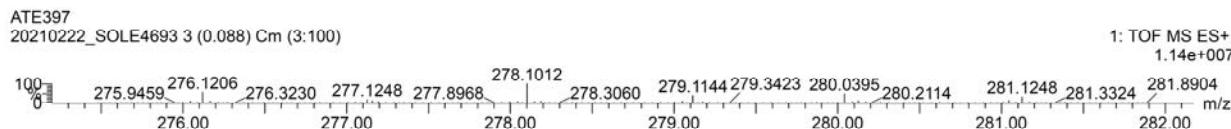
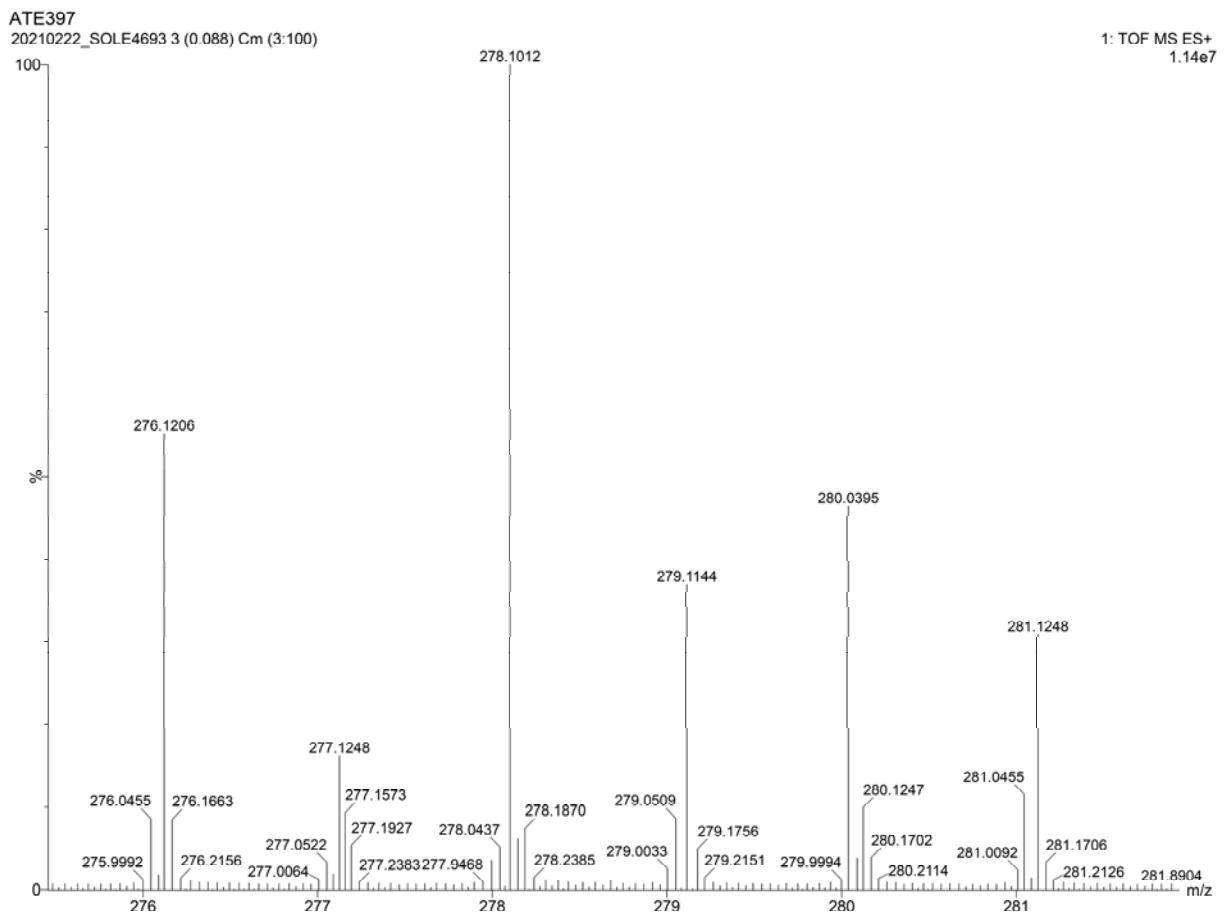
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	256.1396	-3.9	-15.2	-0.5	1516.6	10.711	0.00	C9 H22 N O7
	256.1410	-5.3	-20.7	4.5	1514.3	8.457	0.02	C10 H18 N5 O3
	256.1297	6.0	23.4	4.5	1512.2	6.392	0.17	C11 H18 N3 O4
	256.1450	-9.3	-36.3	8.5	1507.3	1.493	22.48	C15 H18 N3 O
	256.1257	10.0	39.0	0.5	1518.6	12.740	0.00	C6 H18 N5 O6
	256.1509	-15.2	-59.3	-0.5	1517.2	11.340	0.00	C8 H22 N3 O6
	256.1198	15.9	62.1	9.5	1506.6	0.754	47.05	C13 H14 N5 O
	256.1185	17.2	67.2	4.5	1510.8	4.922	0.73	C12 H18 N O5
	256.1549	-19.2	-75.0	3.5	1508.5	2.620	7.28	C13 H22 N O4

HRMS spectrum of 7

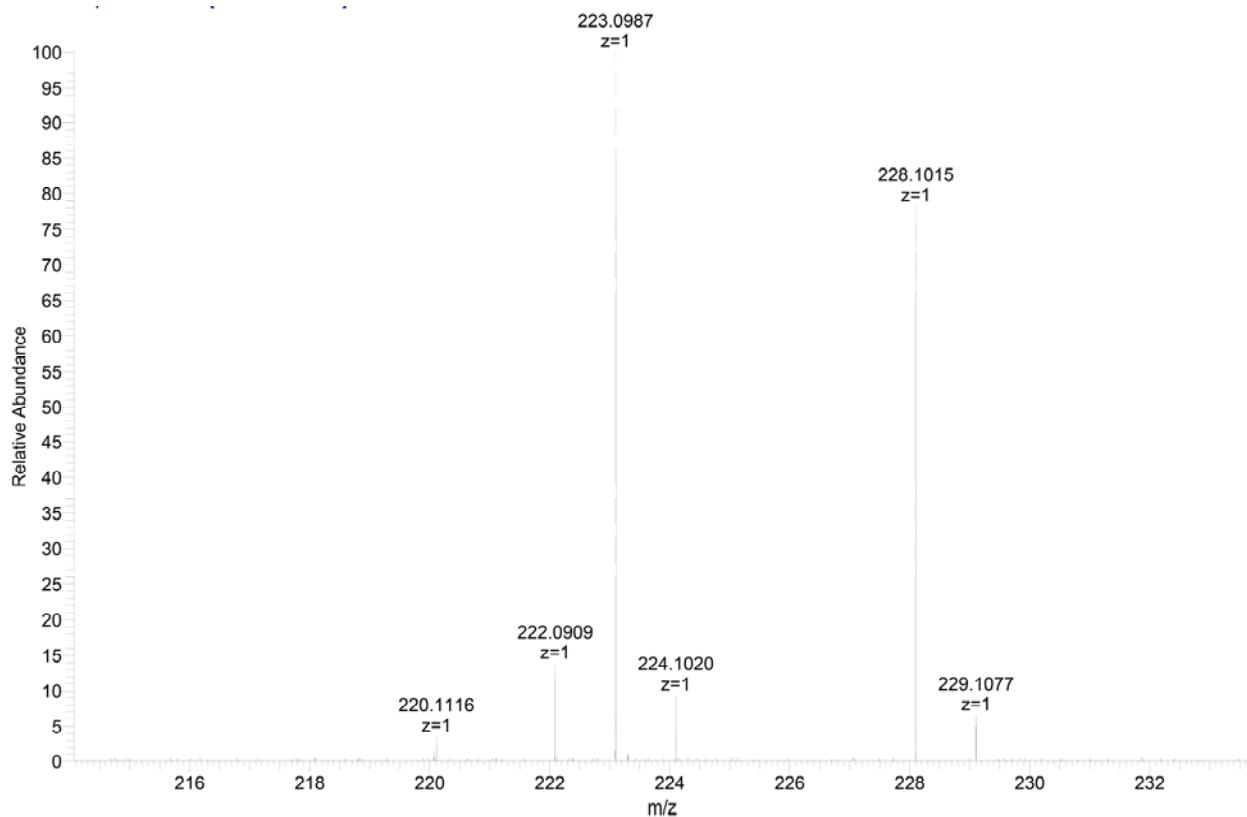
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Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
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	228.0984	3.5	15.3	4.5	1494.1	13.531	0.00	C ₉ H ₁₄ N ₃ O ₄
	228.1083	-6.4	-28.1	-0.5	1497.2	16.558	0.00	C ₇ H ₁₈ N ₀ 7
	228.0944	7.5	32.9	0.5	1498.7	18.093	0.00	C ₄ H ₁₄ N ₅ O ₆
	228.1097	-7.8	-34.2	4.5	1495.6	15.007	0.00	C ₈ H ₁₄ N ₅ O ₃
	228.1137	-11.8	-51.7	8.5	1480.9	0.327	72.10	C ₁₃ H ₁₄ N ₃ O
	228.0885	13.4	58.7	9.5	1490.1	9.461	0.01	C ₁₁ H ₁₀ N ₅ O
	228.0872	14.7	64.4	4.5	1493.1	12.522	0.00	C ₁₀ H ₁₄ N ₀ 5
	228.1196	-17.7	-77.6	-0.5	1497.7	17.049	0.00	C ₆ H ₁₈ N ₃ O ₆
	228.0832	18.7	82.0	0.5	1497.7	17.085	0.00	C ₅ H ₁₄ N ₃ O ₇

HRMS of P8

Minimum: -1.5
Maximum: 100.0 1000.0 50.0

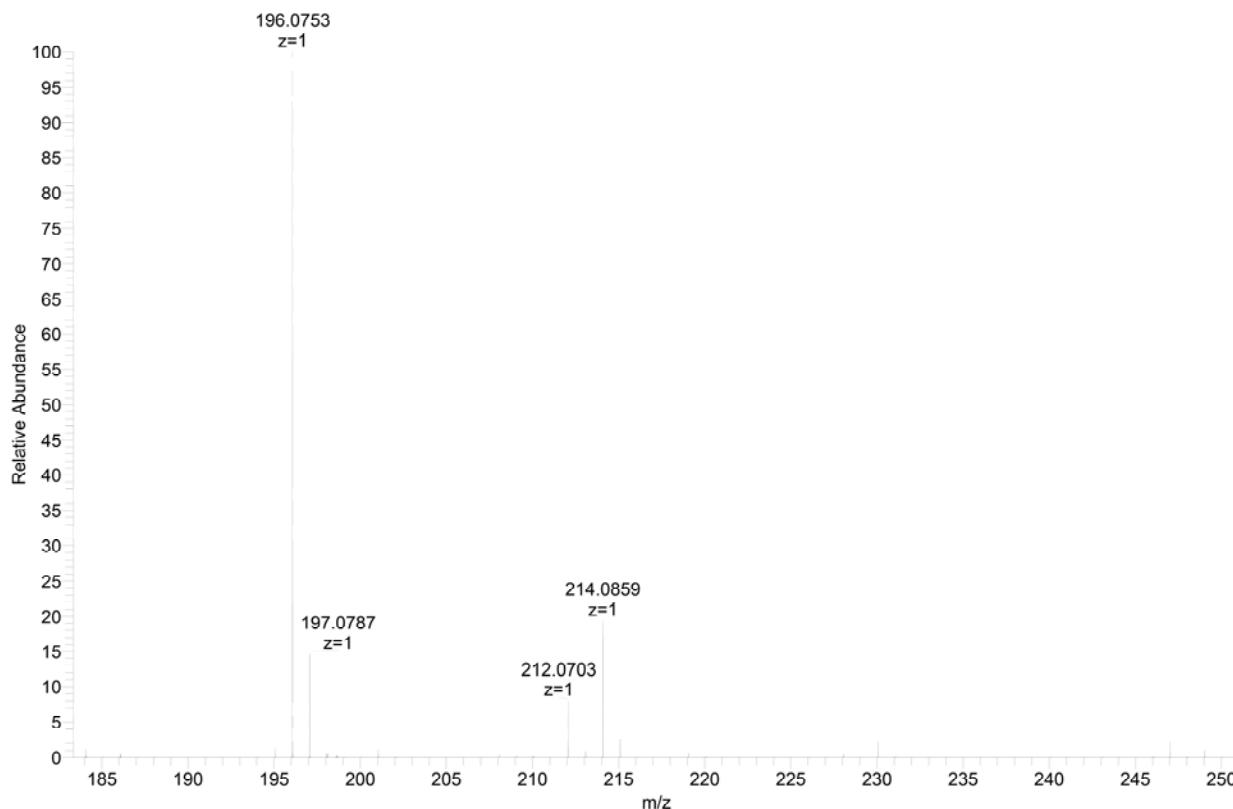
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
278.1012	278.1018	-0.6	-2.2	9.5	1459.0	2.039	13.01	C13 H13 N5 O Na
	278.1004	0.8	2.9	4.5	1459.1	2.213	10.93	C12 H17 N O5 Na
	278.0964	4.8	17.3	0.5	1459.9	2.991	5.03	C7 H17 N3 O7 Na
	278.1077	-6.5	-23.4	0.5	1460.1	3.153	4.27	C6 H17 N5 O6 Na
	278.1117	-10.5	-37.8	4.5	1459.3	2.391	9.16	C11 H17 N3 O4 Na
	278.0905	10.7	38.5	9.5	1458.8	1.831	16.03	C14 H13 N3 O2 Na
	278.1157	-14.5	-52.1	8.5	1458.4	1.490	22.54	C16 H17 N O2 Na
	278.0865	14.7	52.9	5.5	1459.6	2.687	6.81	C9 H13 N5 O4 Na
	278.0852	16.0	57.5	0.5	1459.8	2.848	5.80	C8 H17 N O8 Na
	278.1216	-20.4	-73.4	-0.5	1459.7	2.747	6.41	C9 H21 N O7 Na

HRMS of 8

Elemental composition search on mass 228.1015

m/z = 223.1015-233.1015

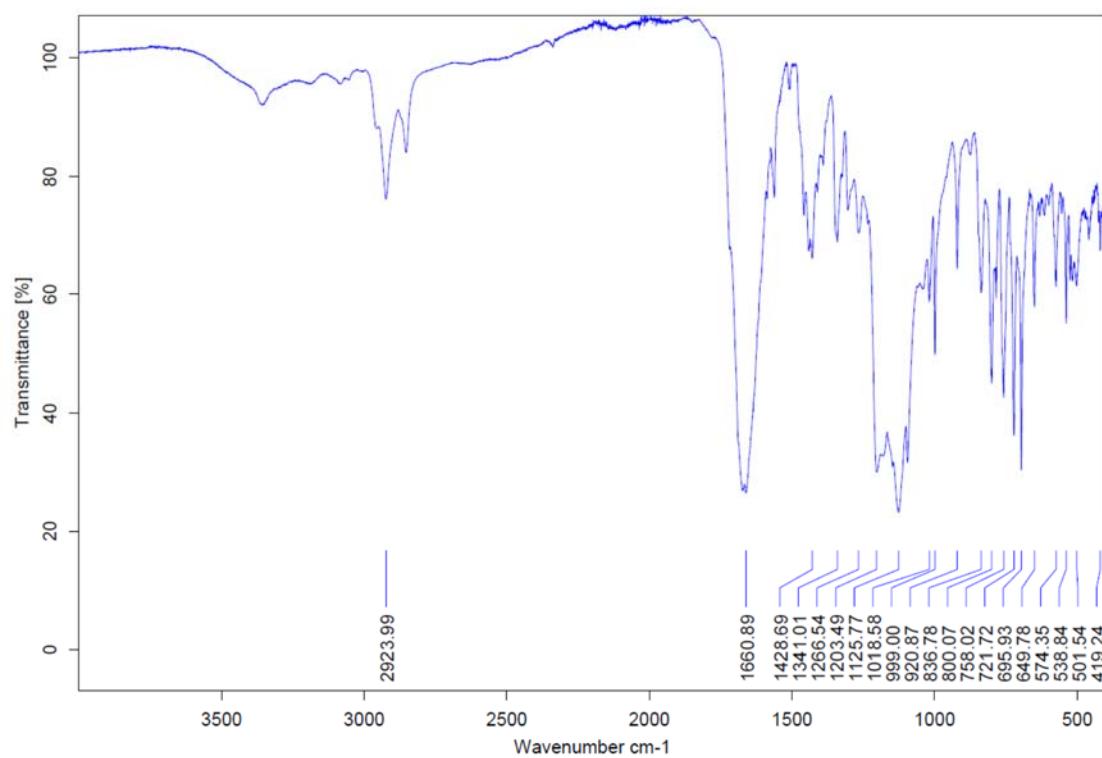
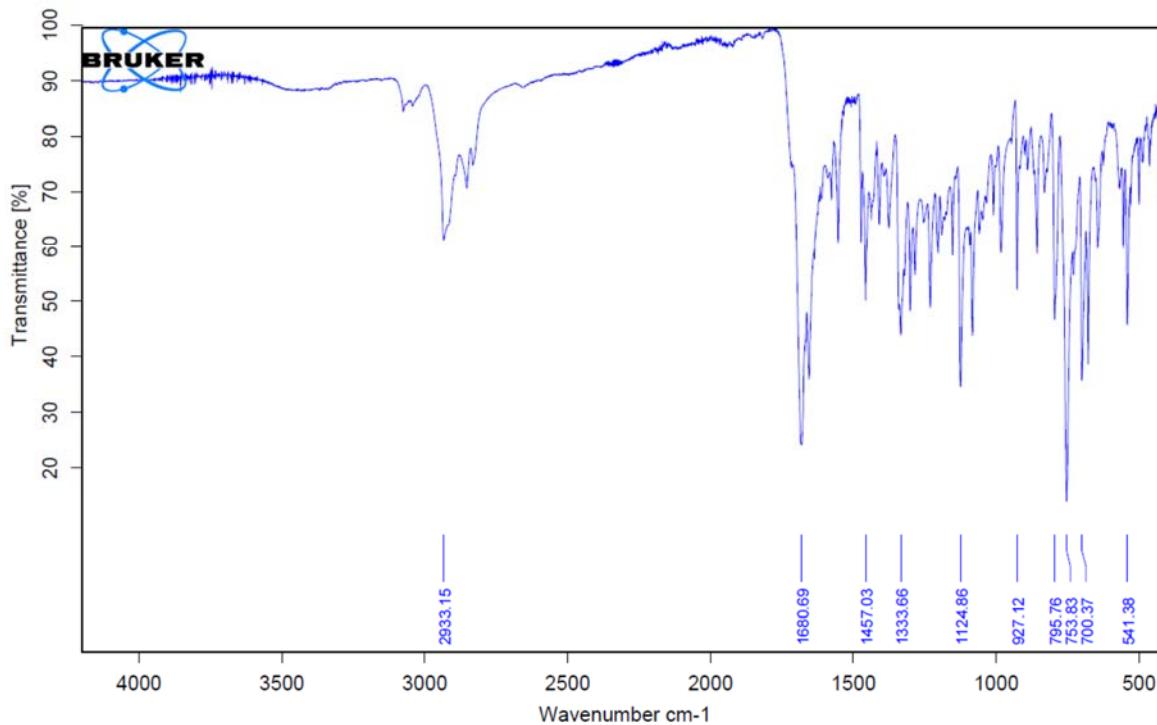
m/z	Theo. Mass	Delta (ppm)	Composition
228.1015	228.1019	-1.86	C ₁₄ H ₁₄ O ₂ N
	228.0995	8.68	C ₁₂ H ₁₅ O ₂ NNa
	228.0992	9.89	C ₁₁ H ₁₆ O ₅

HRMS of 14

Elemental composition search on mass 196.0753

m/z= 191.0753-201.0753			
m/z	Theo. Mass	Delta (ppm)	Composition
196.0753	196.0757	-1.84	C ₁₃ H ₁₀ ON
	196.0733	10.43	C ₁₁ H ₁₁ ONNa
	196.0730	11.83	C ₁₀ H ₁₂ O ₄

IR spectra

Figure S46. IR spectrum of **3**Figure S47. IR spectrum of **5**

Computational study

Computational methodologies

The bond orders for component **3** have been calculated with MOPAC2016⁴ with the AM1 Hamiltonian and a Molecular Mechanics correction. The interface Mercury⁵ has been used. Calculations have been performed on the x-ray crystal structure without structure optimization. All density functional theory (DFT) calculations were performed using Gaussian 16, Revision C.01.⁶ Geometry optimizations, as well as frequency and single-point energy calculations, were performed using the B3LYP⁷ functional augmented with Grimme's D3⁸ empirical dispersion term and the 6-31G(d) basis set. Solvation effects were included in the optimization and single-point calculations using the SMD⁹ solvation model with acetonitrile as the solvent. Gibbs free energies were obtained by summing the thermal correction and single point energies obtained from DFT. No scaling was applied for the ΔG values. Normal vibrational mode analysis was used to verify the structures of minima (zero imaginary frequencies) and transition states (single imaginary frequency). Intrinsic reaction coordinate (IRC) calculations were also used to verify transition state structures, and that they indeed connected the reactants and products. Mixed torsional/Low-Mode conformational sampling in gas phase was performed for the reactant, key intermediates, and transition states, where applicable, using the Merck molecular force field (MMFF94) implemented in MacroModel,¹⁰ followed by DFT optimization. Conformers obtained from each run that fell within 10 kcal/mol of the lowest energy conformer were selected and reoptimized using DFT. GaussView 6.0.16 was used to render optimized structures as well as construct the initial structures used in our calculations.¹¹ Conformational search calculations were set up and rendered in the Maestro GUI.¹² 3D renderings of minima and transition state structures were generated with CYLview 1.0.¹³

Supplemental Figures

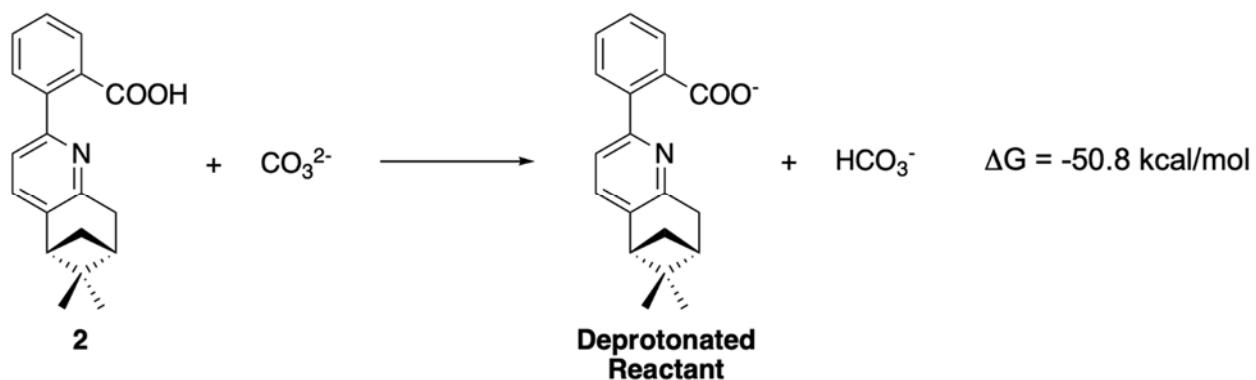
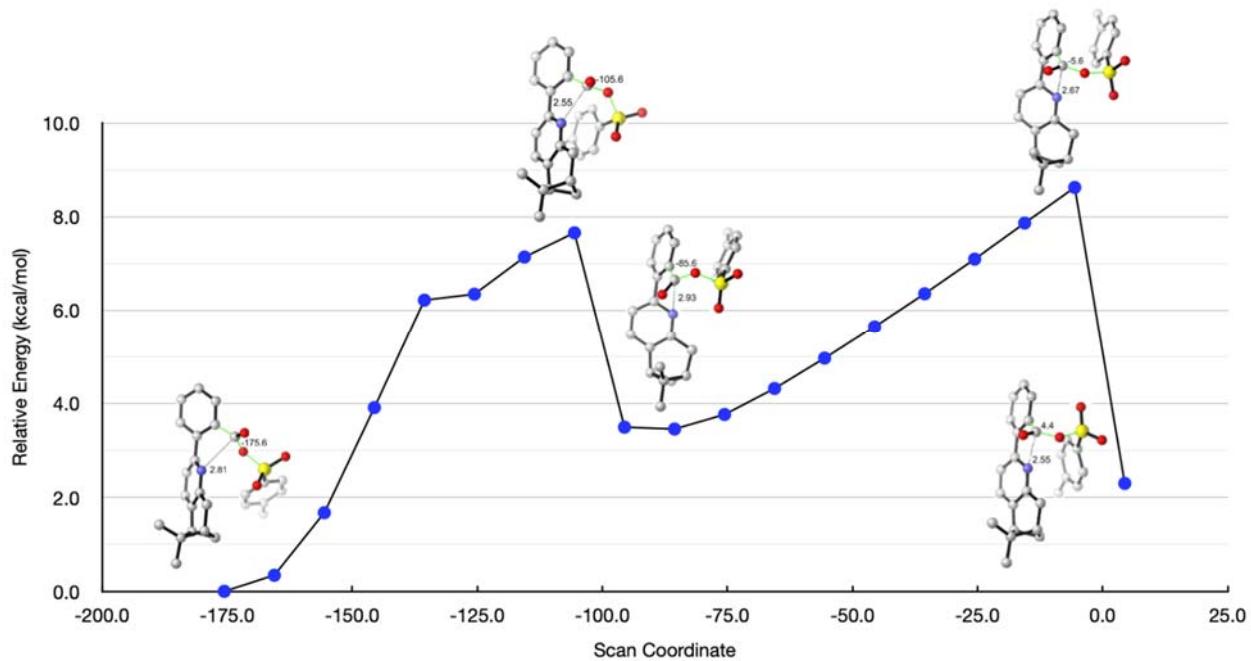
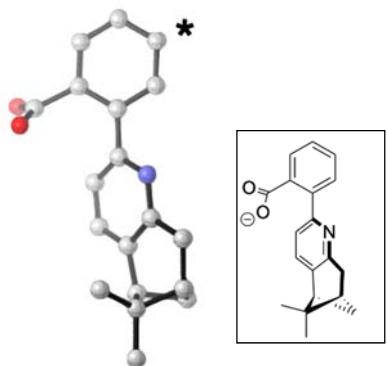
Figure S48. Schematic and energetics of the deprotonation of **2**.

Figure S49. Scan of rotation about the tosyl ester bond in int-1a. Energies shown are single point energies; scan coordinate refers to dihedral angles.

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids
B3LYP-D3/6-31G(d), SMD(CH₃CN) Calculated Cartesian Coordinates and Thermal Corrections

Reactant (deprotonated)

* For all 3D renderings, hydrogens are omitted for clarity.

C	4.69131300	-0.08823100	0.24013900
C	3.36663400	0.35636400	0.11599700
C	2.34401200	-0.60398700	-0.06415300
C	2.68028300	-1.96996400	-0.07632700
C	3.99868100	-2.39515300	0.07974000
C	5.01350400	-1.44579700	0.23578200
C	3.09893200	1.86509700	0.24756000
O	3.57749800	2.58411200	-0.66867800
O	2.44916000	2.20842100	1.26457500
C	0.92066800	-0.22278300	-0.28560700
C	0.56895400	0.91304900	-1.03113600
N	-0.01793000	-1.05684400	0.21513400
C	-0.77843900	1.19150900	-1.25481400
C	-1.30698600	-0.78306300	-0.00429900
C	-1.74865500	0.33503400	-0.74076300
C	-2.35912800	-1.70491000	0.59360500
C	-3.23783100	0.48705000	-0.91827700
C	-3.77489300	-1.23864500	0.20814900
C	-3.96731700	0.29967300	0.47092800
C	-3.35039700	0.92610100	1.72273000
C	-5.44322900	0.71713800	0.40255700
C	-3.79644000	-0.91650200	-1.31247100
H	5.47823700	0.65381000	0.35717500
H	1.88856900	-2.69994300	-0.22018300
H	4.23313700	-3.45668700	0.06665500
H	6.04799100	-1.76194000	0.35008900
H	1.34288700	1.55844400	-1.43185600
H	-1.07150200	2.06427100	-1.83440200
H	-2.23689500	-1.73319900	1.68456900
H	-2.19065200	-2.72955600	0.23556600
H	-3.50209600	1.35994700	-1.52416300
H	-4.53989300	-1.91261200	0.60952100
H	-3.87276000	0.57107600	2.62058000

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H	-2.28794900	0.70117300	1.84547500
H	-3.45289800	2.01868600	1.69035500
H	-5.52852100	1.81061500	0.36216400
H	-5.96990600	0.31206100	-0.46663600
H	-5.97427400	0.37433700	1.30004400
H	-3.16508900	-1.53077300	-1.96394600
H	-4.80413100	-0.87395000	-1.73016400

Zero-point correction= 0.326923 (Hartree/Particle)

Thermal correction to Energy= 0.345038

Thermal correction to Enthalpy= 0.345982

Thermal correction to Gibbs Free Energy= 0.280336

Sum of electronic and zero-point Energies= -939.915887

Sum of electronic and thermal Energies= -939.897772

Sum of electronic and thermal Enthalpies= -939.896828

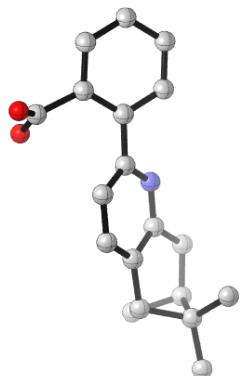
Sum of electronic and thermal Free Energies= -939.962475

Single point energy: -940.242810398

Total free energy: -939.962474398

Number of imaginary frequencies: 0

Reactant (deprotonated) alternative conformations

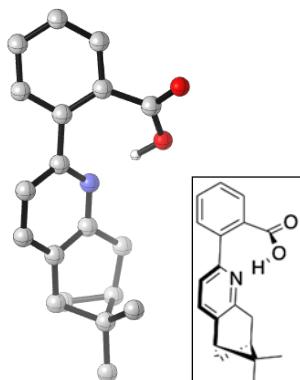


Thermal correction to Gibbs Free Energy= 0.281174 (Hartree/Particle)

Single point energy: -940.242823439

Total free energy: -939.961649439

ΔG (relative to lowest energy conformer): 0.5 kcal/mol

Reactant (protonated)

C	4.54987400	0.52007200	0.07190900
C	3.14793800	0.56197100	-0.01966100
C	2.42422200	-0.66236100	-0.00470000
C	3.15369500	-1.86103300	0.11637700
C	4.54089800	-1.87777900	0.22799500
C	5.24753800	-0.67566700	0.20658300
H	5.08987600	1.45960500	0.03267800
H	2.61239300	-2.80131800	0.14550000
H	5.06181900	-2.82498300	0.33598100
H	6.33036100	-0.66672600	0.29211500
C	2.60291000	1.97048000	-0.18131400
O	1.42668600	2.28547400	0.35362600
O	3.26874100	2.82792700	-0.74183600
C	0.94382800	-0.80452900	-0.12561700
C	0.37111500	-1.82343700	-0.89608800
N	0.16467900	0.05359200	0.56811300
C	-1.01788400	-1.95836400	-0.92749900
H	0.99961600	-2.48946800	-1.47661700
C	-1.16489800	-0.06809300	0.53793600
C	-1.81258600	-1.08289200	-0.19403300
H	-1.47788300	-2.73874100	-1.52840200
C	-2.01176700	0.94464200	1.28466100
H	-1.73531500	0.93618500	2.34701700
H	-1.78168700	1.95033400	0.90861200
C	-3.316555600	-1.10267500	-0.12567400
H	-3.75128400	-1.93108500	-0.69305200
C	-3.50675700	0.61499900	1.12285100
H	-4.13075000	1.32377500	1.67716700
C	-3.87948700	0.35332000	-0.38197500
C	-3.23802600	1.21604400	-1.47057100
H	-3.62923400	2.24012700	-1.41885000
H	-2.14859400	1.27471000	-1.40089300
H	-3.48223500	0.81688600	-2.46346200
C	-5.39759600	0.34318700	-0.60872300
H	-5.79187900	1.36612200	-0.55713900
H	-5.63060700	-0.05480200	-1.60445300

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H	-5.94325500	-0.25826600	0.12426200
C	-3.70884200	-0.90571900	1.37252800
H	-4.74631400	-1.17437000	1.57771900
H	-3.06157200	-1.38197200	2.11706300
H	0.93466800	1.45863300	0.68372700

Zero-point correction= 0.340325 (Hartree/Particle)

Thermal correction to Energy= 0.358049

Thermal correction to Enthalpy= 0.358994

Thermal correction to Gibbs Free Energy= 0.294897

Sum of electronic and zero-point Energies= -940.398315

Sum of electronic and thermal Energies= -940.380590

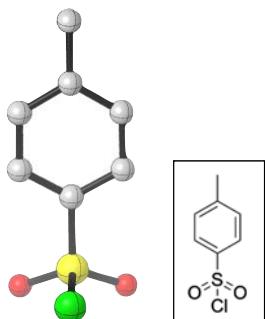
Sum of electronic and thermal Enthalpies= -940.379646

Sum of electronic and thermal Free Energies= -940.443742

Single point energy: -940.738639147

Total free energy: -940.443842147

Number of imaginary frequencies: 0

TsCl

C	2.16224000	-1.21115300	0.01675500
C	2.87086600	-0.00220100	0.10628700
C	2.16273700	1.20763500	0.00703900
C	0.78482700	1.22326900	-0.17967500
C	0.11915800	-0.00229100	-0.27450400
C	0.78320500	-1.22681900	-0.17015900
H	2.69675700	-2.15359000	0.09473100
H	2.69868800	2.15003100	0.07737000
H	0.24344900	2.16001800	-0.25481400
H	0.24129300	-2.16382300	-0.23790300
S	-1.63608200	-0.00234300	-0.51096800
O	-2.07741500	1.26842300	-1.08069500
O	-2.07940700	-1.27823800	-1.06748000
Cl	-2.36580000	0.00871300	1.52935200
C	4.36629400	0.00316600	0.28273600
H	4.86377300	0.13765200	-0.68716900
H	4.68740900	0.82700700	0.92897800
H	4.72315200	-0.93905000	0.70983900

Zero-point correction= 0.129064 (Hartree/Particle)

Thermal correction to Energy= 0.139872

Thermal correction to Enthalpy= 0.140816

Thermal correction to Gibbs Free Energy= 0.090888

Sum of electronic and zero-point Energies= -1279.638496

Sum of electronic and thermal Energies= -1279.627689

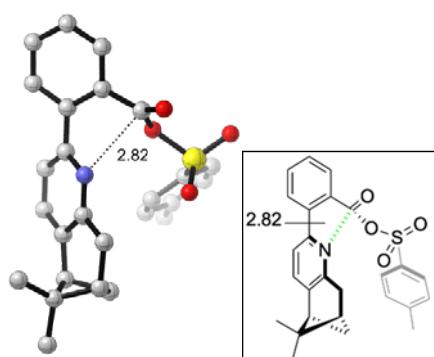
Sum of electronic and thermal Enthalpies= -1279.626745

Sum of electronic and thermal Free Energies= -1279.676672

Single point energy: -1279.76756048

Total free energy: -1279.67667248

Number of imaginary frequencies: 0

Int-1a

C	-0.32001500	3.11661100	1.39070000
C	0.38675600	4.32115600	1.55606600
C	1.46648800	4.59087900	0.70342300
C	1.84187800	3.69162500	-0.29505800
C	1.11251700	2.51179600	-0.43418600
C	0.03066600	2.20951100	0.39862500
S	1.54814500	1.36343900	-1.71455300
O	0.35768800	0.92600700	-2.44103500
O	2.72814800	1.82972100	-2.43914800
C	-0.02752500	5.30334000	2.62113800
O	2.01454900	0.11571500	-0.66958200
C	2.33175300	-1.14034200	-1.17641900
O	2.48960500	-1.36092200	-2.35077900
C	2.60218700	-2.07763800	-0.05077900
C	3.83093800	-2.74705200	-0.08757500
C	1.71681200	-2.25517200	1.03587300
C	4.21771200	-3.56695000	0.97178700
C	2.12499500	-3.08519000	2.08946000
C	3.36498500	-3.72533300	2.06598700
C	0.34943600	-1.67532500	1.01864000
C	-0.27712700	-1.21010000	2.17997700
N	-0.26167400	-1.64998200	-0.18557300
C	-1.57961700	-0.71387300	2.09107300
C	-1.50534700	-1.17514900	-0.26559600
C	-2.21761300	-0.68814600	0.85461200
C	-2.20176000	-1.17314600	-1.61499700
C	-3.60357900	-0.15834500	0.59433200
C	-3.60948000	-0.56136900	-1.49806300
C	-4.41178700	-1.18324100	-0.29754900
C	-4.28240900	-2.67999600	-0.00885300
C	-5.90534200	-0.83300200	-0.35572200
C	-3.51450500	0.75164900	-0.67191600
H	-1.15357900	2.88665300	2.04912300
H	2.02526900	5.51501200	0.82215000
H	2.68245700	3.90191500	-0.94740500
H	-0.51631400	1.28213000	0.28324200
H	0.74639400	6.05644100	2.79768200

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H -0.94519800 5.82792800 2.32392800
H -0.24079600 4.79302200 3.56723300
H 4.48941400 -2.59887600 -0.93820500
H 5.17697200 -4.07508000 0.94234000
H 1.44729900 -3.25954200 2.91984800
H 3.65387100 -4.36555800 2.89478000
H 0.24768100 -1.21947700 3.12961900
H -2.08736000 -0.33883800 2.97636200
H -2.24756300 -2.20117200 -1.99772100
H -1.59340100 -0.59770800 -2.32244100
H -4.08236200 0.22629900 1.50043100
H -4.11177800 -0.52984800 -2.47102100
H -4.76408300 -3.26108500 -0.80575100
H -3.24779000 -3.02339100 0.07191700
H -4.78594300 -2.92968500 0.93412400
H -6.39712400 -1.41371700 -1.14672000
H -6.39309300 -1.08414100 0.59478200
H -6.09673700 0.22529400 -0.55635700
H -2.60583800 1.35170500 -0.78577200
H -4.38501600 1.39921900 -0.78984400

Zero-point correction= 0.457872 (Hartree/Particle)

Thermal correction to Energy= 0.485954

Thermal correction to Enthalpy= 0.486898

Thermal correction to Gibbs Free Energy= 0.397716

Sum of electronic and zero-point Energies= -1759.245146

Sum of electronic and thermal Energies= -1759.217064

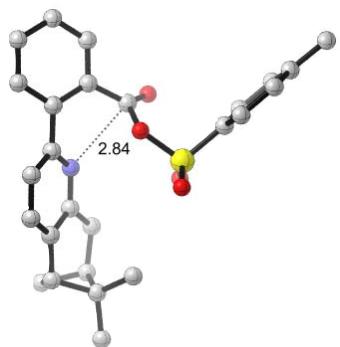
Sum of electronic and thermal Enthalpies= -1759.216119

Sum of electronic and thermal Free Energies= -1759.305301

Single point energy: -1759.70301746

Total free energy: -1759.30530146

Number of imaginary frequencies: 0

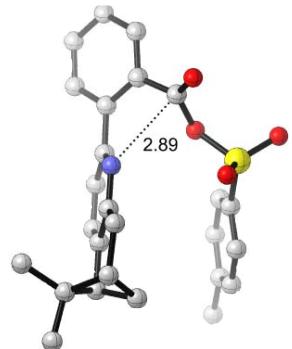
Int-1a alternative conformations

Thermal correction to Gibbs Free Energy= 0.397753 (Hartree/Particle)

Single point energy: -1759.70214757

Total free energy: -1759.30439457

ΔG (relative to lowest energy conformer): 0.6 kcal/mol

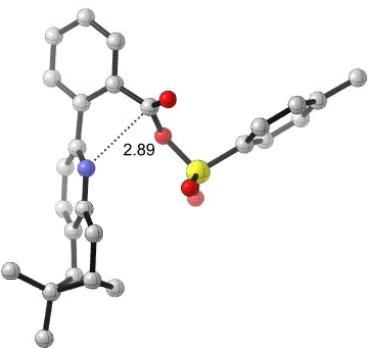


Thermal correction to Gibbs Free Energy= 0.397522 (Hartree/Particle)

Single point energy: -1759.70156833

Total free energy: -1759.30404633

ΔG : 0.8 kcal/mol



Thermal correction to Gibbs Free Energy= 0.397442 (Hartree/Particle)

Single point energy: -1759.70136351

Total free energy: -1759.30392151

ΔG : 0.9 kcal/mol

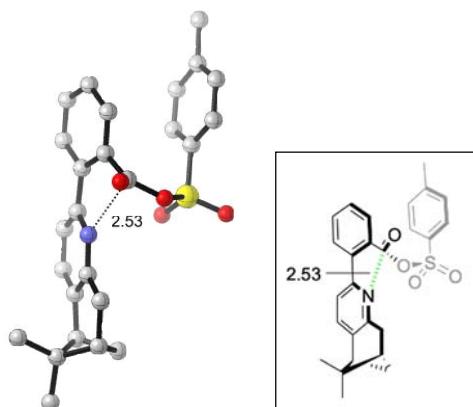
Cl-

Cl 0.00000000 0.00000000 0.00000000

Zero-point correction= 0.000000 (Hartree/Particle)
Thermal correction to Energy= 0.001416
Thermal correction to Enthalpy= 0.002360
Thermal correction to Gibbs Free Energy= -0.015023
Sum of electronic and zero-point Energies= -460.360860
Sum of electronic and thermal Energies= -460.359444
Sum of electronic and thermal Enthalpies= -460.358500
Sum of electronic and thermal Free Energies= -460.375883

Single point energy: -460.360860002
Total free energy: -460.375883002

Number of imaginary frequencies: 0

Int-1b

C	0.66742100	-0.18468200	1.80184600
O	0.39614900	-0.24089700	2.97190100
C	1.28431000	1.00270700	1.13095600
C	2.55599400	1.35657400	1.59865100
C	0.64853900	1.76417500	0.12721300
C	3.23760700	2.43564700	1.04089100
C	1.34889400	2.85543000	-0.41174400
C	2.62997000	3.18182600	0.02829300
C	-0.72910500	1.42115000	-0.29428900
C	-1.31928600	1.87953200	-1.47738800
N	-1.37898600	0.58687900	0.53888500
C	-2.60394800	1.44027000	-1.80143400
C	-2.59667700	0.15546300	0.22412400
C	-3.26359900	0.55430600	-0.95435900
C	-3.30878900	-0.79214500	1.17312500
C	-4.62355000	-0.04692800	-1.19341100
C	-4.67795500	-1.19926100	0.59962600
C	-5.49736700	0.05405600	0.12001000
C	-5.44607500	1.34016700	0.94663900
C	-6.96991500	-0.28678300	-0.14855100
C	-4.49770000	-1.57573400	-0.89783400
H	3.02034200	0.76139600	2.37879200
H	4.23248600	2.69003800	1.39395900
H	0.87815600	3.47439900	-1.16830800
H	3.14756200	4.02997700	-0.41076600
H	-0.78482200	2.53795200	-2.15312500
H	-3.08097000	1.77291800	-2.72014900
H	-3.41501800	-0.31106800	2.15415600
H	-2.68281100	-1.68097800	1.32768400
H	-5.07021000	0.28416000	-2.13623300
H	-5.19066000	-1.90875000	1.25802400
H	-5.96802500	1.19691500	1.90152200
H	-4.43035500	1.67587300	1.17127600
H	-5.95074200	2.15528100	0.41186300
H	-7.50260800	-0.42721100	0.80081000

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H -7.45898300 0.53478700 -0.68728300
H -7.10472400 -1.19718400 -0.74011000
H -3.55457100 -2.05434500 -1.18400600
H -5.32955400 -2.15686900 -1.29961800
C 4.50330300 0.14353600 -1.44119300
C 5.44817600 -0.36178800 -0.53555400
C 5.04174400 -1.33780700 0.38961700
C 3.72730000 -1.79291000 0.42475900
C 2.80912100 -1.25282100 -0.48010900
C 3.18069800 -0.29149300 -1.42067500
S 1.12373800 -1.80563400 -0.43063400
O 1.06519400 -3.26548900 -0.39767900
O 0.62474200 -1.42688100 1.12333900
C 6.86081300 0.16234200 -0.53001600
O 0.33168300 -1.06660900 -1.40987600
H 4.80328300 0.89627400 -2.16495100
H 5.76445500 -1.74427100 1.09175600
H 3.42270200 -2.54420000 1.14640100
H 2.44837600 0.11869300 -2.10641500
H 7.18245700 0.45617800 -1.53459100
H 6.93199800 1.05242700 0.10982500
H 7.56365200 -0.58038400 -0.13918500

Zero-point correction= 0.457440 (Hartree/Particle)

Thermal correction to Energy= 0.485454

Thermal correction to Enthalpy= 0.486399

Thermal correction to Gibbs Free Energy= 0.397883

Sum of electronic and zero-point Energies= -1759.238527

Sum of electronic and thermal Energies= -1759.210513

Sum of electronic and thermal Enthalpies= -1759.209568

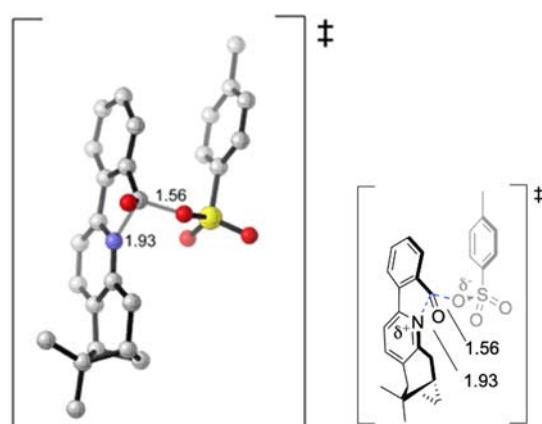
Sum of electronic and thermal Free Energies= -1759.298084

Single point energy: -1759.69596697

Total free energy: -1759.29808397

Number of imaginary frequencies: 0

TS1



C	0.16842500	0.13962600	1.63006300
O	-0.26534700	0.10227800	2.75710500
C	1.16953500	1.13113800	1.10372200
C	2.42613000	1.28380900	1.68277800
C	0.76832800	1.93558300	0.02792700
C	3.30359400	2.23568100	1.15673300
C	1.64605400	2.88999200	-0.49356800
C	2.91448100	3.03083200	0.07210900
C	-0.60104900	1.65085700	-0.42439400
C	-1.31393600	2.22487400	-1.47177400
N	-1.15055800	0.67489300	0.32051500
C	-2.60128600	1.74454000	-1.73674500
C	-2.36435200	0.19425500	0.08948500
C	-3.14193100	0.71694400	-0.96688200
C	-2.92808500	-0.91380400	0.95271400
C	-4.48528900	0.06928000	-1.17479200
C	-4.31919600	-1.32025600	0.43562400
C	-5.24040500	-0.06687400	0.20751800
C	-5.18112400	1.09410200	1.20194600
C	-6.71121400	-0.46050300	0.01358700
C	-4.25117700	-1.47458500	-1.10966800
H	2.72143400	0.65670900	2.51822400
H	4.29284700	2.35406300	1.58907700
H	1.34438000	3.51541500	-1.32843700
H	3.60380700	3.76606500	-0.33304400
H	-0.87476500	3.01382700	-2.07260000
H	-3.18012500	2.16619100	-2.55413200
H	-2.96455000	-0.57721100	1.99534600
H	-2.24192900	-1.76868600	0.92716500
H	-5.02965700	0.49757800	-2.02151400
H	-4.73263900	-2.14441900	1.02585700
H	-5.61035200	0.78817900	2.16452000
H	-4.16825100	1.45856600	1.39316900
H	-5.76999500	1.94225400	0.82971900

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H -7.28745400 0.39411800 -0.36278400
H -6.85087300 -1.28908200 -0.68693000
H -7.14848500 -0.76183200 0.97400700
H -3.31219200 -1.84653800 -1.53426400
H -5.08277600 -2.04695200 -1.52360800
C 4.40880000 -0.05064400 -1.41600800
C 5.31868900 -0.54652300 -0.47193400
C 4.85524900 -1.45222400 0.49695100
C 3.52048800 -1.84424500 0.53475900
C 2.63490000 -1.31843400 -0.41032000
C 3.06696700 -0.42841800 -1.39244500
S 0.91906500 -1.80066400 -0.35359100
O 0.83809800 -3.26419800 -0.27492200
O 0.41144900 -1.30607700 1.09856800
C 6.75476900 -0.08805000 -0.47004600
O 0.20514200 -1.11212700 -1.43557600
H 4.75114800 0.64797100 -2.17481400
H 5.54972100 -1.85219000 1.23115300
H 3.17293100 -2.54102800 1.29100600
H 2.36375900 -0.02892600 -2.11415900
H 7.42582600 -0.87228500 -0.10403400
H 7.08134000 0.21262300 -1.47092700
H 6.87679500 0.78125900 0.19043600

Zero-point correction= 0.456855 (Hartree/Particle)

Thermal correction to Energy= 0.484192

Thermal correction to Enthalpy= 0.485137

Thermal correction to Gibbs Free Energy= 0.397928

Sum of electronic and zero-point Energies= -1759.234696

Sum of electronic and thermal Energies= -1759.207359

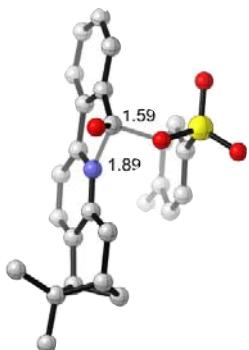
Sum of electronic and thermal Enthalpies= -1759.206414

Sum of electronic and thermal Free Energies= -1759.293623

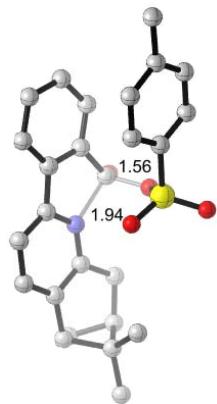
Single point energy: -1759.69155109

Total free energy: -1759.29362309

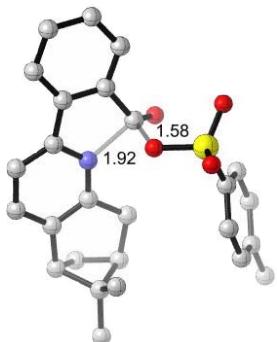
Number of imaginary frequencies: 1
Frequencies -- -169.8427
Red. masses -- 10.5269
Frc consts -- 0.1789
IR Inten -- 289.0098

TS1 alternative conformations

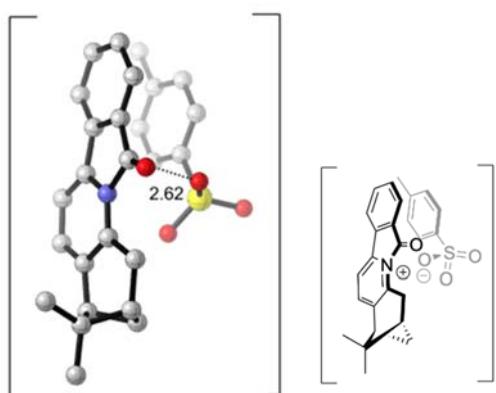
Thermal correction to Gibbs Free Energy= 0.399527 (Hartree/Particle)
Single point energy: -1759.69309030
Total free energy: -1759.2935633
 ΔG (relative to lowest energy conformer): 0.04 kcal/mol



Thermal correction to Gibbs Free Energy= 0.400004 (Hartree/Particle)
Single point energy: -1759.69313692
Total free energy: -1759.29313292
 ΔG : 0.3 kcal/mol



Thermal correction to Gibbs Free Energy= 0.399603 (Hartree/Particle)
Single point energy: -1759.69271515
Total free energy: -1759.29311215
 ΔG : 0.3 kcal/mol

Complex-1

C	0.20748800	-1.73477700	1.23474000
O	0.81515100	-2.02125700	2.22924600
C	-1.15107000	-2.06225400	0.78435700
C	-2.17488400	-2.68285300	1.48878700
C	-1.29957100	-1.68206800	-0.55741200
C	-3.37811500	-2.91282100	0.81230700
C	-2.48944700	-1.91598500	-1.23541800
C	-3.52898400	-2.53501900	-0.52899500
C	-0.05522200	-1.05664900	-1.00238600
C	0.32626500	-0.52557800	-2.21306300
N	0.82568300	-1.05055800	0.05336200
C	1.61105100	0.02363400	-2.32653000
C	2.04471300	-0.48314800	-0.00410600
C	2.46719000	0.06499500	-1.23235700
C	2.94595900	-0.40166500	1.19861600
C	3.82416000	0.71686800	-1.22298400
C	4.23440700	0.35504500	0.83261500
C	4.87192400	-0.20845800	-0.48828200
C	4.84524400	-1.71563800	-0.75069600
C	6.30826600	0.29037400	-0.69524600
C	3.86044700	1.63757800	0.03991400
H	-2.04321900	-2.97140500	2.52678300
H	-4.20471200	-3.38781500	1.33144900
H	-2.61488300	-1.62157100	-2.27184100
H	-4.47270900	-2.72284200	-1.03236400
H	-0.36489500	-0.53640900	-3.04777200
H	1.93497900	0.44291400	-3.27443200
H	3.16090000	-1.40856100	1.57057700
H	2.39232400	0.11149500	1.99402500
H	4.08749200	1.12625700	-2.20201300
H	4.88481900	0.44647400	1.70794300
H	5.51546600	-2.23094500	-0.05127700
H	3.85443000	-2.16740100	-0.64875000
H	5.19709900	-1.92829100	-1.76812500
H	6.98662600	-0.22776800	-0.00575700

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H 6.64152400 0.07488100 -1.71812500
H 6.42357600 1.36502900 -0.52743700
H 2.92239100 2.13395000 0.30212100
H 4.66957200 2.36808400 -0.00014700
C -3.43555600 1.58621400 -1.66369600
C -4.52134600 1.04202600 -0.96496400
C -4.36061900 0.76378700 0.40245500
C -3.15245700 1.01095600 1.04924700
C -2.07922200 1.55216000 0.33400000
C -2.22136400 1.84342900 -1.02276200
S -0.52861100 1.90172000 1.18815000
O -0.75426000 3.14682200 1.97126600
O -0.29893500 0.70042300 2.05470700
C -5.81656800 0.71330600 -1.66541300
O 0.49036900 2.04312300 0.10644500
H -3.53815600 1.80603500 -2.72401800
H -5.18924600 0.33327600 0.96004500
H -3.03270700 0.76609500 2.10017100
H -1.37864100 2.25055400 -1.57166000
H -5.90803300 1.24834800 -2.61639100
H -5.87735100 -0.36187900 -1.88329300
H -6.68337900 0.96424700 -1.04333200

Zero-point correction= 0.458872 (Hartree/Particle)

Thermal correction to Energy= 0.486720

Thermal correction to Enthalpy= 0.487664

Thermal correction to Gibbs Free Energy= 0.400634

Sum of electronic and zero-point Energies= -1759.261117

Sum of electronic and thermal Energies= -1759.233269

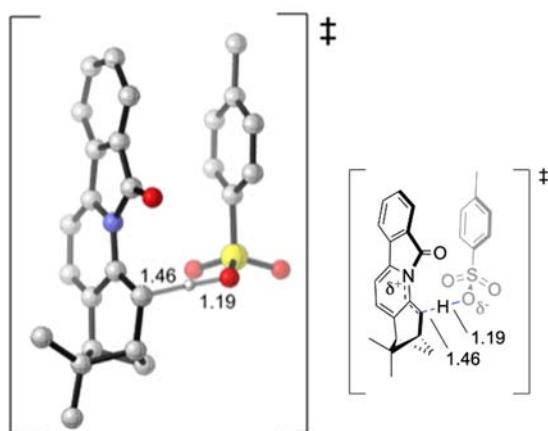
Sum of electronic and thermal Enthalpies= -1759.232325

Sum of electronic and thermal Free Energies= -1759.319355

Single point energy: -1759.7199889

Total free energy: -1759.3193549

Number of imaginary frequencies: 0

TS2a

C	-0.55741800	1.01665700	-1.26998900
O	-0.35546700	0.48412100	-2.34272200
C	-1.75350400	1.70013700	-0.75942000
C	-2.97840300	1.90358800	-1.38652400
C	-1.50444000	2.14546100	0.54833100
C	-3.97156400	2.57160500	-0.66631100
C	-2.49605800	2.80636300	1.27056900
C	-3.73073800	3.01099000	0.64585100
C	-0.13633700	1.78273400	0.89816900
C	0.62185600	2.02420500	2.00032800
N	0.40575600	1.09347600	-0.19701100
C	1.98053200	1.58796900	2.01131800
C	1.68889600	0.58772800	-0.20060700
C	2.50675700	0.89009200	0.95935700
C	2.22905400	-0.22263600	-1.21505000
C	3.92937500	0.40292200	0.86813800
C	3.70667900	-0.55661300	-1.01583600
C	4.52390800	0.71879200	-0.56524600
C	4.22253700	2.07201000	-1.20944700
C	6.04136300	0.49492700	-0.61233500
C	3.88711300	-1.09528100	0.43329100
H	-3.15147900	1.54660900	-2.39674100
H	-4.94209400	2.74653900	-1.12012000
H	-2.31934900	3.14918100	2.28531300
H	-4.52136300	3.52159000	1.18808600
H	0.19846500	2.56445600	2.83934200
H	2.60639500	1.82537000	2.86671500
H	1.84811000	-0.09853500	-2.22441400
H	1.46860200	-1.43689000	-0.93457400
H	4.51476000	0.69423100	1.74535100
H	4.10010000	-1.14578900	-1.84938100
H	4.60122400	2.09305600	-2.23928700
H	3.15714100	2.31398900	-1.24594600
H	4.72436800	2.87468600	-0.65380500

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H 6.38738900 0.49956800 -1.65384800
H 6.56187500 1.30484500 -0.08532800
H 6.35988200 -0.44999900 -0.16369900
H 3.07508800 -1.69971700 0.84188100
H 4.83889500 -1.60212800 0.59684200
C -3.51099400 -0.88609600 1.02956500
C -4.13616000 -1.07428200 -0.21087800
C -3.41101700 -1.69992100 -1.23766800
C -2.10636400 -2.14323100 -1.03333500
C -1.50958700 -1.94914900 0.21503800
C -2.20367800 -1.31757900 1.24920100
S 0.12200700 -2.60492900 0.55339600
O -0.00786800 -4.03821300 0.86905900
O 0.88537400 -2.46785100 -0.80738200
C -5.55550400 -0.61977800 -0.44036800
O 0.73081000 -1.73807000 1.58216400
H -4.04702500 -0.38125600 1.82870400
H -3.87310300 -1.83430600 -2.21252000
H -1.54991300 -2.61264500 -1.83757200
H -1.72015900 -1.15008200 2.20596400
H -6.25738800 -1.45190700 -0.29374300
H -5.83747900 0.17721500 0.25452700
H -5.69595400 -0.25312200 -1.46281200

Zero-point correction= 0.452782 (Hartree/Particle)

Thermal correction to Energy= 0.480177

Thermal correction to Enthalpy= 0.481121

Thermal correction to Gibbs Free Energy= 0.395635

Sum of electronic and zero-point Energies= -1759.231733

Sum of electronic and thermal Energies= -1759.204338

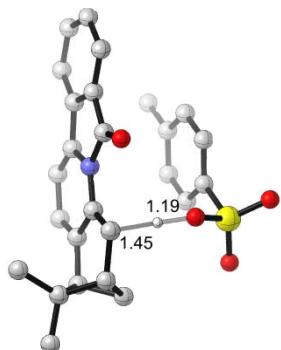
Sum of electronic and thermal Enthalpies= -1759.203394

Sum of electronic and thermal Free Energies= -1759.288879

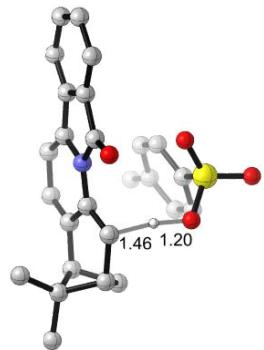
Single point energy: -1759.68451445

Total free energy: -1759.28887945

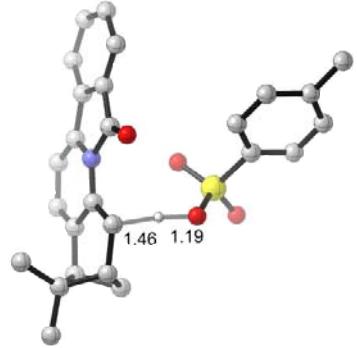
Number of imaginary frequencies: 1
Frequencies -- -1428.0717
Red. masses -- 1.2064
Frc consts -- 1.4496
IR Inten -- 10320.3909

TS2a alternative conformations

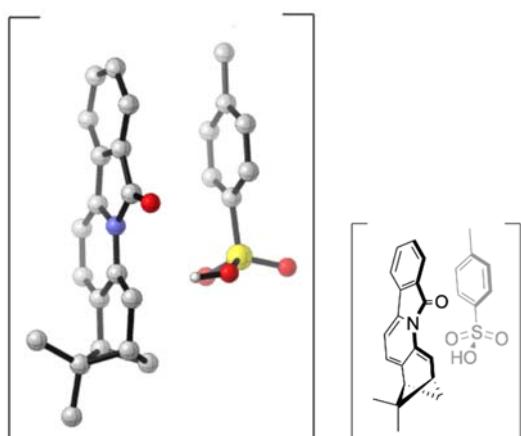
Thermal correction to Gibbs Free Energy= 0.394365 (Hartree/Particle)
Single point energy: -1759.6829728
Total free energy: -1759.2886078
 ΔG (relative to lowest energy conformer): 0.2 kcal/mol



Thermal correction to Gibbs Free Energy= 0.393034 (Hartree/Particle)
Single point energy: -1759.68058554
Total free energy: -1759.28755154
 ΔG : 0.8 kcal/mol



Thermal correction to Gibbs Free Energy= 0.391846 (Hartree/Particle)
Single point energy: -1759.67888435
Total free energy: -1759.28703835
 ΔG : 1.2 kcal/mol

Complex-2a

C	0.49346000	-1.02897700	-1.34253200
O	0.26929900	-0.50470000	-2.42694000
C	1.70770400	-1.70579600	-0.85633800
C	2.90862700	-1.94508000	-1.51888200
C	1.50202400	-2.11171300	0.47341600
C	3.91582400	-2.61197600	-0.81957600
C	2.51310900	-2.76899000	1.17670100
C	3.71775800	-3.01396000	0.51348000
C	0.15646700	-1.71491300	0.85820700
C	-0.54583600	-1.85332000	2.00866900
N	-0.42034800	-1.08428600	-0.26722000
C	-1.89742200	-1.36637700	2.07387600
C	-1.74095200	-0.60342300	-0.26048200
C	-2.47618000	-0.76328400	1.00077000
C	-2.39952300	-0.04740600	-1.32247100
C	-3.89204900	-0.25132000	0.95800000
C	-3.83398500	0.37093000	-1.07580100
C	-4.62503500	-0.78903900	-0.34002200
C	-4.40085700	-2.24075800	-0.76224800
C	-6.13710500	-0.53346200	-0.29385400
C	-3.88306000	1.15089100	0.27342500
H	3.05040100	-1.61778600	-2.54434900
H	4.86573700	-2.81676300	-1.30415900
H	2.36901900	-3.08096600	2.20676700
H	4.51969100	-3.52523900	1.03853200
H	-0.08696000	-2.33884100	2.86284000
H	-2.46214800	-1.49509600	2.99266800
H	-1.96917600	-0.02282800	-2.31452700
H	-1.18159300	1.54798500	-1.04025100
H	-4.39765200	-0.37539200	1.92055000
H	-4.29154200	0.81659600	-1.96325500
H	-4.86474900	-2.42687100	-1.73935200
H	-3.34598900	-2.51448700	-0.83810500
H	-4.86994000	-2.91866400	-0.03758600

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H -6.57687000 -0.71380700 -1.28309400
H -6.61564800 -1.22307300 0.41330200
H -6.40376300 0.48399500 0.00471900
H -3.03942700 1.80565000 0.49973100
H -4.81602000 1.69335800 0.42990900
C 3.40060200 0.78727900 1.24903700
C 4.15707500 0.89586800 0.07445900
C 3.57992400 1.52536600 -1.04127400
C 2.29552100 2.05693600 -0.98523700
C 1.57077100 1.94210000 0.20527800
C 2.10915300 1.30687100 1.32545800
S -0.01168500 2.74340700 0.36322000
O 0.16809700 4.19636100 0.38275600
O -0.73675900 2.44207200 -1.05403100
C 5.56481600 0.36194000 0.00882200
O -0.76047900 2.09216300 1.44398000
H 3.82225300 0.28278700 2.11370500
H 4.14528600 1.59618100 -1.96664000
H 1.85752400 2.53510100 -1.85493300
H 1.52589900 1.20981800 2.23462900
H 6.28779600 1.16149300 0.22006300
H 5.72568200 -0.43339100 0.74275900
H 5.79535000 -0.03175400 -0.98631700

Zero-point correction= 0.456821 (Hartree/Particle)

Thermal correction to Energy= 0.485038

Thermal correction to Enthalpy= 0.485982

Thermal correction to Gibbs Free Energy= 0.397438

Sum of electronic and zero-point Energies= -1759.234295

Sum of electronic and thermal Energies= -1759.206078

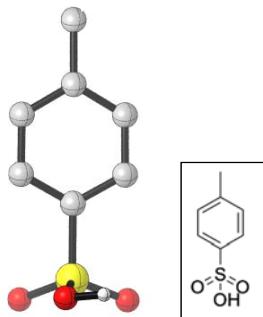
Sum of electronic and thermal Enthalpies= -1759.205134

Sum of electronic and thermal Free Energies= -1759.293678

Single point energy: -1759.6911159

Total energy: -1759.2936779

Number of imaginary frequencies: 0

TsOH

C	-1.94852400	1.20745700	0.00835600
C	-2.66550800	0.00040300	-0.02081300
C	-1.95198900	-1.20817500	-0.00219400
C	-0.56008600	-1.22227100	0.04643500
C	0.12135800	-0.00357000	0.07333400
C	-0.55726400	1.21756300	0.05699000
H	-2.48780700	2.15075600	-0.00585300
H	-2.49340700	-2.15002300	-0.02469100
H	-0.01408000	-2.15946300	0.06557300
H	-0.00902900	2.15331400	0.08450300
S	1.89474700	-0.00492000	0.12296000
O	2.38582300	-1.27914000	0.65570600
O	2.17879300	0.03023800	-1.49620400
C	-4.17255800	0.00485400	-0.03995000
H	-4.56822300	0.05136300	0.98361000
H	-4.56221700	0.87436000	-0.57966300
H	-4.57007800	-0.90317900	-0.50463800
O	2.39114700	1.24612900	0.70469800
H	3.15020700	0.10620400	-1.63275000

Zero-point correction= 0.141777 (Hartree/Particle)

Thermal correction to Energy= 0.152755

Thermal correction to Enthalpy= 0.153699

Thermal correction to Gibbs Free Energy= 0.102371

Sum of electronic and zero-point Energies= -895.250978

Sum of electronic and thermal Energies= -895.240001

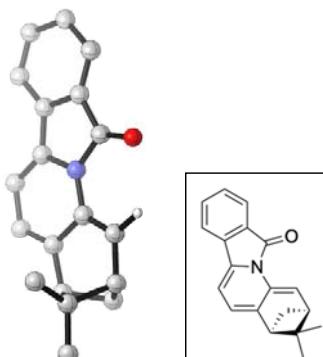
Sum of electronic and thermal Enthalpies= -895.239057

Sum of electronic and thermal Free Energies= -895.290385

Single point energy: -895.392755863

Total free energy: -895.290384863

Number of imaginary frequencies: 0

Product

C	1.46781800	-1.33075700	-0.10098800
O	1.14001300	-2.51351700	-0.13736600
C	2.81499000	-0.74096800	0.02075500
C	4.05144700	-1.37532000	0.11786900
C	2.70162500	0.66047000	0.02625200
C	5.18902400	-0.57334400	0.22340000
H	4.11999900	-2.45927600	0.11087200
C	3.83995500	1.46311200	0.13223000
C	5.08032400	0.82939700	0.23064600
H	6.16982800	-1.03367900	0.30105200
H	3.76466600	2.54658300	0.13837600
H	5.98087100	1.43149300	0.31446600
C	1.29085300	0.98968800	-0.09488500
C	0.63337100	2.17323200	-0.14137300
N	0.58857100	-0.23741300	-0.16984000
C	-0.80019200	2.18453000	-0.27613600
H	1.19373000	3.09991400	-0.08026900
C	-0.81971000	-0.28272400	-0.29049400
C	-1.49837300	1.02005000	-0.35533200
H	-1.32213600	3.13620000	-0.31764200
C	-1.58781500	-1.40338300	-0.31547600
H	-1.16136400	-2.39303900	-0.24034900
C	-2.99365800	0.92607600	-0.50919200
H	-3.45726100	1.91450700	-0.58857100
C	-3.07835500	-1.19842500	-0.46061100
H	-3.62761400	-2.14366100	-0.49576500
C	-3.59024000	-0.08954500	0.54877400
C	-3.30056600	-0.15138100	-1.59735900
H	-2.62013200	-0.20552100	-2.45234000
H	-4.32948000	-0.09930300	-1.95642700
C	-3.04897500	-0.06124800	1.97793300
H	-3.35922400	0.86500100	2.47850400
H	-3.45673700	-0.90157500	2.55442600
H	-1.95958000	-0.12046800	2.02982600
C	-5.12152200	-0.01753500	0.61824600
H	-5.61048800	-0.03061100	-0.35968900

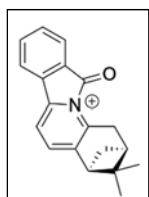
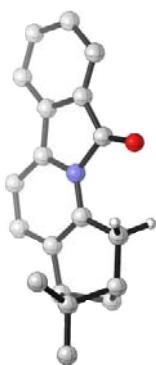
Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H -5.51065500 -0.86802400 1.19261400
H -5.43452400 0.89991900 1.13316700

Zero-point correction= 0.313078 (Hartree/Particle)
Thermal correction to Energy= 0.329128
Thermal correction to Enthalpy= 0.330072
Thermal correction to Gibbs Free Energy= 0.270566
Sum of electronic and zero-point Energies= -863.958802
Sum of electronic and thermal Energies= -863.942751
Sum of electronic and thermal Enthalpies= -863.941807
Sum of electronic and thermal Free Energies= -864.001313

Single point energy: -864.27187935
Total free energy: -864.00131335

Number of imaginary frequencies: 0

Int-2

C	1.47317700	-1.33764200	-0.09716600
O	1.08596600	-2.47391500	-0.14301300
C	2.80912300	-0.75179700	0.02128200
C	4.03679200	-1.39847800	0.11020400
C	2.70266500	0.64778500	0.03097500
C	5.17966400	-0.59773500	0.21166900
H	4.10087900	-2.48177600	0.10036800
C	3.83470600	1.44553400	0.13066000
C	5.07652400	0.79988800	0.22139200
H	6.15732400	-1.06358000	0.28315000
H	3.76432200	2.52819800	0.13876800
H	5.97897800	1.39845300	0.30063300
C	1.28917300	1.00567600	-0.08092600
C	0.64699900	2.22205500	-0.11245800
N	0.55352100	-0.15629300	-0.16036700
C	-0.75022800	2.23626900	-0.22902600
H	1.22209300	3.13822600	-0.04770300
C	-0.79319600	-0.18166900	-0.27962400
C	-1.47573800	1.05095000	-0.31703300
H	-1.27739200	3.18511600	-0.25304100
C	-1.57706000	-1.46766800	-0.35363500
H	-1.21870800	-2.06003000	-1.20338600
H	-1.37489800	-2.06617200	0.54190600
C	-2.97019000	0.96866800	-0.47039900
H	-3.43035600	1.95885800	-0.51323900
C	-3.07689000	-1.15624600	-0.50712800
H	-3.64965200	-2.08466500	-0.58709800
C	-3.55650100	-0.09345100	0.54440400
C	-3.00128600	-0.15002700	1.96846800
H	-3.32039000	0.73326400	2.53545300
H	-3.38793600	-1.03730600	2.48483500
H	-1.90947300	-0.19123500	2.01965900
C	-5.08669100	-0.008444800	0.62676000
H	-5.48029100	-0.88285800	1.15969000
H	-5.38672100	0.88675600	1.18524000
H	-5.57597200	0.03073200	-0.35061000

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

C -3.25263300 -0.06977100 -1.60167900
H -2.55310800 -0.08584800 -2.44402000
H -4.27151800 -0.00412900 -1.98436000

Zero-point correction= 0.326595 (Hartree/Particle)

Thermal correction to Energy= 0.342659

Thermal correction to Enthalpy= 0.343603

Thermal correction to Gibbs Free Energy= 0.284170

Sum of electronic and zero-point Energies= -864.423613

Sum of electronic and thermal Energies= -864.407549

Sum of electronic and thermal Enthalpies= -864.406605

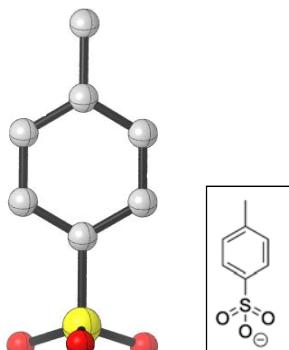
Sum of electronic and thermal Free Energies= -864.466039

Single point energy: -864.750208299

Total free energy: -864.466038299

Number of imaginary frequencies: 0

-OTs



C	1.93358500	-1.20057900	-0.00250900
C	2.65441400	0.00442200	0.01233000
C	1.93333700	1.20632400	-0.00290700
C	0.53699600	1.20978100	-0.03021800
C	-0.16443700	0.00266700	-0.04203800
C	0.53974800	-1.20566100	-0.03008500
H	2.47373100	-2.14524100	0.00383600
H	2.47158100	2.15177300	0.00328800
H	-0.01166900	2.14646200	-0.05485300
H	-0.00752600	-2.14326900	-0.05537400
S	-1.97373000	-0.00084900	0.00579600
O	-2.32350500	-0.04771200	1.45653000
O	-2.37890800	1.27061200	-0.66092500
C	4.16401700	-0.00324000	0.01722200
H	4.56040300	-0.28798800	-0.96674500
H	4.56982200	0.98334100	0.26432600
H	4.55812800	-0.72622200	0.74139900
O	-2.37518100	-1.22884400	-0.74052600

Zero-point correction= 0.131292 (Hartree/Particle)

Thermal correction to Energy= 0.141209

Thermal correction to Enthalpy= 0.142154

Thermal correction to Gibbs Free Energy= 0.094920

Sum of electronic and zero-point Energies= -894.808888

Sum of electronic and thermal Energies= -894.798970

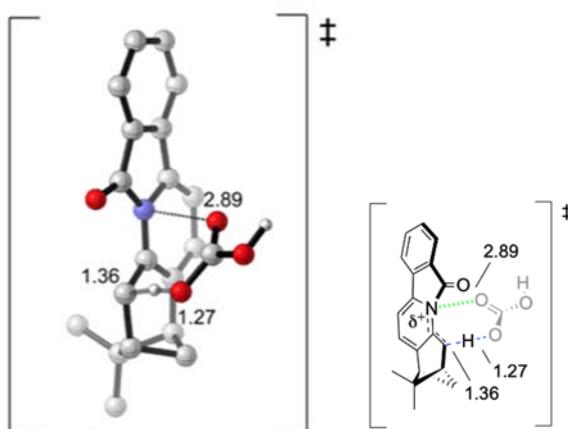
Sum of electronic and thermal Enthalpies= -894.798026

Sum of electronic and thermal Free Energies= -894.845259

Single point energy: -894.940179852

Total free energy: -894.845259852

Number of imaginary frequencies: 0

TS2b

C	-1.44187800	0.20405900	1.14797900
O	-1.07564000	0.93581100	2.04119300
C	-2.79640300	-0.21616900	0.75823700
C	-4.01949300	0.14191200	1.31601000
C	-2.70550800	-1.10877800	-0.32035500
C	-5.17313000	-0.41629200	0.75886900
C	-3.85370200	-1.667779100	-0.87651800
C	-5.08691700	-1.30829000	-0.32231900
C	-1.29599000	-1.26409900	-0.66223300
C	-0.66901500	-1.96840900	-1.64458500
N	-0.55660800	-0.48670100	0.23171100
C	0.74726700	-1.88058400	-1.74672400
C	0.80171100	-0.32353500	0.13642400
C	1.46879300	-1.07545600	-0.90245200
C	1.56867400	0.54760000	0.95582000
C	2.96609900	-0.92107500	-0.93225400
C	3.07317200	0.45305500	0.68479400
C	3.55338600	-1.04122900	0.53331000
C	3.28069700	0.60463900	-0.84932200
H	-4.06878100	0.83276200	2.15215400
H	-6.14707500	-0.15890700	1.16412700
H	-3.79595000	-2.35858500	-1.71198400
H	-5.99799200	-1.72921500	-0.73777300
H	-1.24787300	-2.57697500	-2.32966200
H	1.26328800	-2.46144900	-2.50579800
H	1.24405500	0.64790800	1.98914500
H	1.21952100	1.79429100	0.52620200
H	3.42203300	-1.52548900	-1.72195900
H	3.63914100	1.10433100	1.35779500
O	1.03739000	3.00105200	0.17049500
C	0.02212300	3.04926200	-0.62517500
O	-0.69594100	2.12263300	-1.00990400
O	-0.21616500	4.33013900	-1.05214800
H	-0.99143900	4.27012800	-1.64037700

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H	2.59718300	1.28198500	-1.37053100
H	4.30681600	0.83727000	-1.13694900
C	2.98755400	-2.10740000	1.47080600
H	3.25328000	-3.10963000	1.11034100
H	1.89939200	-2.07031400	1.56910200
H	3.41358500	-1.99123600	2.47553500
C	5.08215200	-1.17414800	0.52026900
H	5.37418700	-2.18139900	0.19640100
H	5.47953400	-1.02164800	1.53190600
H	5.57863800	-0.45761700	-0.14021800

Zero-point correction= 0.348519 (Hartree/Particle)

Thermal correction to Energy= 0.369391

Thermal correction to Enthalpy= 0.370335

Thermal correction to Gibbs Free Energy= 0.298212

Sum of electronic and zero-point Energies= -1128.928226

Sum of electronic and thermal Energies= -1128.907354

Sum of electronic and thermal Enthalpies= -1128.906410

Sum of electronic and thermal Free Energies= -1128.978533

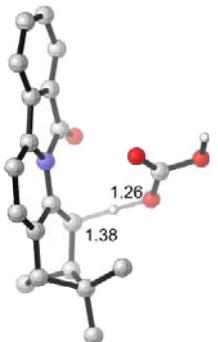
Single point energy: -1129.2767451

Total free energy: -1128.9785331

Number of imaginary frequencies: 1
Frequencies -- -1454.8392
Red. masses -- 1.1677
Frc consts -- 1.4562
IR Inten -- 11090.9759

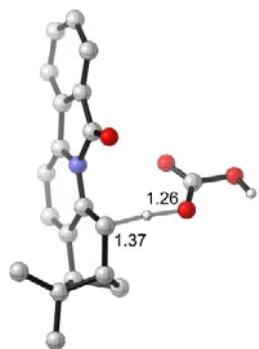
TS2b alternative conformations

Thermal correction to Gibbs Free Energy= 0.299894 (Hartree/Particle)



Single point energy: -1129.27787337

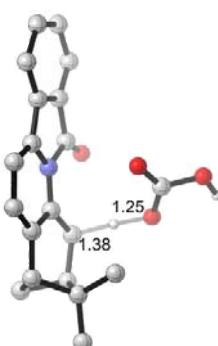
Total free energy: -1128.97797938

 ΔG (relative to lowest energy conformer): 0.3 kcal/mol

Thermal correction to Gibbs Free Energy= 0.298347 (Hartree/Particle)

Single point energy: -1129.2760669

Total free energy: -1128.9777199

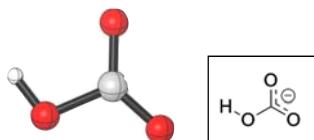
 ΔG : 0.5 kcal/mol

Thermal correction to Gibbs Free Energy= 0.299873 (Hartree/Particle)

Single point energy: -1129.27721464

Total free energy: -1128.97734164

 ΔG : 0.7 kcal/mol

HCO₃⁻

C	0.14740900	0.06222800	0.00004700
O	1.19921300	-0.60448400	-0.00004500
O	-0.04828700	1.30083300	0.00001100
O	-1.04176800	-0.73354300	0.00006700
H	-1.75771700	-0.07581600	-0.00054000

Zero-point correction= 0.026197 (Hartree/Particle)

Thermal correction to Energy= 0.029718

Thermal correction to Enthalpy= 0.030662

Thermal correction to Gibbs Free Energy= 0.000425

Sum of electronic and zero-point Energies= -264.494550

Sum of electronic and thermal Energies= -264.491029

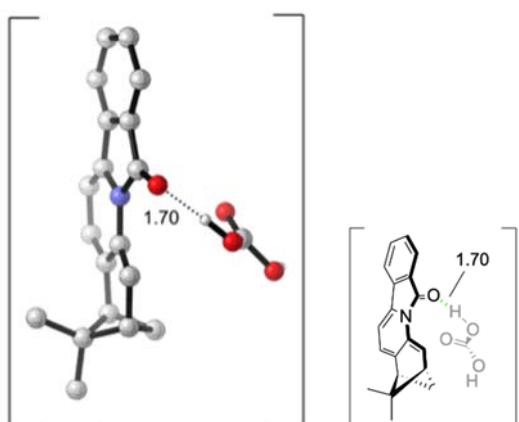
Sum of electronic and thermal Enthalpies= -264.490085

Sum of electronic and thermal Free Energies= -264.520322

Single point energy: -264.520746874

Total free energy: -264.520321874

Number of imaginary frequencies: 0

Complex-2b

C	-1.53407300	0.42830700	0.95883300
O	-1.19012200	1.35643800	1.71111000
C	-2.89212000	-0.02641800	0.64241500
C	-4.11757000	0.44270600	1.11340000
C	-2.80997300	-1.07594400	-0.29199300
C	-5.27538100	-0.16842500	0.63196500
C	-3.97061100	-1.68396800	-0.77617600
C	-5.19845700	-1.21965600	-0.30114900
C	-1.40448700	-1.29172800	-0.58244700
C	-0.76683100	-2.09301900	-1.46923900
N	-0.67567000	-0.38935900	0.23524700
C	0.66433400	-2.02110900	-1.59939300
C	0.73739300	-0.34383400	0.20458600
C	1.38685100	-1.17437100	-0.81751700
C	1.53444400	0.35131600	1.05635900
C	2.88480700	-1.03129700	-0.86240300
C	3.02100100	0.28921100	0.79928200
C	3.48794700	-1.21082800	0.59212400
C	3.23400700	0.48462600	-0.73498000
H	-4.16074200	1.25724000	1.83041400
H	-6.24835200	0.16821100	0.97802100
H	-3.92081400	-2.49341700	-1.49850300
H	-6.11546400	-1.67852100	-0.66023100
H	-1.34730900	-2.76708600	-2.09004500
H	1.16347900	-2.64868000	-2.33169100
H	1.13965500	0.92628500	1.88070600
H	-0.23990800	2.57617300	1.00375800
H	3.32583000	-1.61772200	-1.67432200
H	3.59019900	0.90778500	1.49842400
O	0.35334000	3.26607200	0.59480300
C	0.37526200	3.06033600	-0.71551600
O	-0.31293400	2.27360700	-1.33971600
O	1.29018900	3.87680200	-1.27406100
H	1.26476100	3.70790700	-2.23535000

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

H	2.56948400	1.19496200	-1.23514800
H	4.26716700	0.69920600	-1.01089000
C	2.91749600	-2.30643700	1.49151800
H	3.18720300	-3.29439400	1.09674300
H	1.82968800	-2.26809000	1.58233300
H	3.34052000	-2.22575700	2.50101900
C	5.01475000	-1.35597900	0.56931000
H	5.29359600	-2.35067600	0.19854900
H	5.41491400	-1.25623400	1.58634900
H	5.51928100	-0.61597700	-0.05782600

Zero-point correction= 0.354004 (Hartree/Particle)

Thermal correction to Energy= 0.375203

Thermal correction to Enthalpy= 0.376148

Thermal correction to Gibbs Free Energy= 0.303617

Sum of electronic and zero-point Energies= -1128.944450

Sum of electronic and thermal Energies= -1128.923251

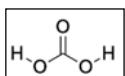
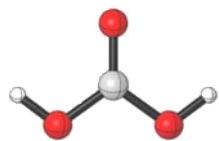
Sum of electronic and thermal Enthalpies= -1128.922307

Sum of electronic and thermal Free Energies= -1128.994837

Single point energy: -1129.29845416

Total free energy: -1128.99483716

Number of imaginary frequencies: 0

H_2CO_3 

O	1.08628900	-0.68312400	-0.00002900
H	1.86609700	-0.09493400	-0.00002900
C	0.00000000	0.10107500	-0.00009000
O	-1.08628800	-0.68312600	0.00006900
H	-1.86609500	-0.09493500	0.00015700
O	-0.00000100	1.31417700	0.00001200

Zero-point correction= 0.039122 (Hartree/Particle)

Thermal correction to Energy= 0.042862

Thermal correction to Enthalpy= 0.043806

Thermal correction to Gibbs Free Energy= 0.013206

Sum of electronic and zero-point Energies= -264.966649

Sum of electronic and thermal Energies= -264.962909

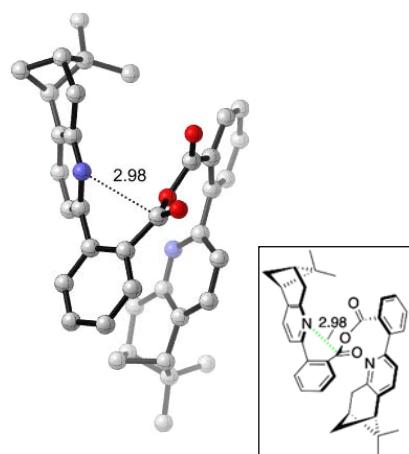
Sum of electronic and thermal Enthalpies= -264.961965

Sum of electronic and thermal Free Energies= -264.992565

Single point energy: -265.005771132

Total free energy: -264.992565132

Number of imaginary frequencies: 0

Anhydride intermediate

C	-0.13606400	-1.17922700	2.49518200
O	-0.33962700	-1.26989800	3.68138000
C	0.70670900	-2.11347600	1.69950800
C	1.79954800	-2.66740000	2.38115900
C	0.50495600	-2.39756800	0.32973000
C	2.71634800	-3.47856500	1.71672500
C	1.43602200	-3.22034000	-0.31965000
C	2.53361100	-3.75101200	0.35978700
C	-0.68918900	-1.91769000	-0.41290500
C	-0.57658500	-1.39691900	-1.70615200
N	-1.87524000	-2.01851000	0.22482800
C	-1.73473700	-0.98720100	-2.37035600
C	-2.97507500	-1.61493500	-0.41191200
C	-2.96373900	-1.10243600	-1.72913900
C	-4.30508100	-1.67958500	0.32004200
C	-4.29975800	-0.73875100	-2.32378100
C	-5.45311100	-1.22253800	-0.59839200
C	-5.12136500	0.14361300	-1.30255600
C	-4.38742900	1.22661000	-0.51385600
C	-6.35847900	0.77417300	-1.95642300
C	-5.28165500	-1.90487400	-1.98457400
O	-0.51577400	-0.08561700	1.73853100
C	-1.46277600	0.82494800	2.15617900
O	-2.36342400	0.53574900	2.90978900
C	-1.30439300	2.14523800	1.48415400
C	-2.41917000	2.99662700	1.59529600
C	-0.19485900	2.54398600	0.69727600
C	-2.47246600	4.20986800	0.92083100
C	-0.28412200	3.76043100	0.00144500
C	1.09733300	1.80358600	0.57043000
C	-1.40215800	4.58507400	0.10462600
C	1.86278400	1.46224700	1.69114000
N	1.52205100	1.57600400	-0.68779700
C	3.09327800	0.83468700	1.50238300

Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

C 2.70600300 0.97871700 -0.86161200
 C 3.53366400 0.57606600 0.20627900
 C 3.20483600 0.75592100 -2.28003700
 C 4.83505400 -0.08821400 -0.16117700
 C 4.58191200 0.06718200 -2.27122800
 C 5.57310600 0.76490400 -1.27019000
 C 4.52359500 -1.12874000 -1.28221400
 C 5.57099700 2.29061200 -1.15785800
 C 7.02199200 0.30071900 -1.47509100
 H 1.93447900 -2.43073800 3.43174100
 H 3.56628800 -3.89072300 2.25239400
 H 1.28092200 -3.46691200 -1.36587400
 H 3.23899000 -4.38408000 -0.17098000
 H 0.39737100 -1.30106100 -2.17523400
 H -1.67800200 -0.57786400 -3.37608800
 H -4.24171600 -1.05659900 1.22230000
 H -4.48225100 -2.70766700 0.66232300
 H -4.21734100 -0.38784700 -3.35738000
 H -6.42229500 -1.30148100 -0.09398600
 H -5.03874500 1.63665400 0.26839400
 H -3.47719300 0.86852300 -0.03058700
 H -4.10067200 2.05493000 -1.17430000
 H -7.01952200 1.19128100 -1.18574500
 H -6.06084400 1.59568000 -2.62052500
 H -6.94712000 0.06669000 -2.54799000
 H -4.86516700 -2.91815400 -1.99703800
 H -6.18961300 -1.88557700 -2.59004200
 H -3.25779600 2.66943700 2.19969900
 H -3.34559100 4.84831100 1.01677900
 H 0.55588400 4.06363700 -0.61577500
 H -1.43270300 5.52255800 -0.44370800
 H 1.49354500 1.67714700 2.68903200
 H 3.70590600 0.55309800 2.35535900
 H 3.24599400 1.72122800 -2.80173000
 H 2.47988800 0.13721800 -2.82659300
 H 5.41015300 -0.39964300 0.71657600
 H 4.94432100 -0.11220300 -3.28938200
 H 3.57267600 -1.66234800 -1.20272300
 H 5.32785100 -1.85256500 -1.42635100
 H 5.98212500 2.73729400 -2.07234500
 H 4.57700100 2.71662300 -0.99911300
 H 6.20345000 2.61057900 -0.31948600
 H 7.43571700 0.74864900 -2.38781100
 H 7.64769800 0.62205700 -0.63264500
 H 7.12202100 -0.78488000 -1.56640700

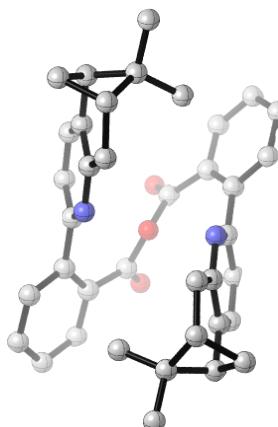
Zero-point correction= 0.654790 (Hartree/Particle)

Thermal correction to Energy= 0.690528

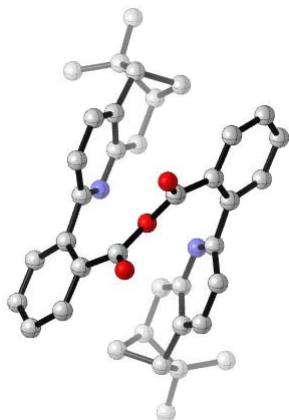
Efficient synthesis of isoindolones by intramolecular cyclisation of pyridinylbenzoic acids

Thermal correction to Enthalpy= 0.691472
 Thermal correction to Gibbs Free Energy= 0.585613
 Sum of electronic and zero-point Energies= -1804.386831
 Sum of electronic and thermal Energies= -1804.351093
 Sum of electronic and thermal Enthalpies= -1804.350149
 Sum of electronic and thermal Free Energies= -1804.456008

Single point energy: -1805.04162086
 Total free energy: -1804.45600786
 Number of imaginary frequencies: 0

Anhydride intermediate alternative conformations

Thermal correction to Gibbs Free Energy= 0.588524 (Hartree/Particle)
 Single point energy: -1805.04376414
 Total free energy: -1804.45524014
 ΔG (relative to lowest energy conformer): 0.5 kcal/mol

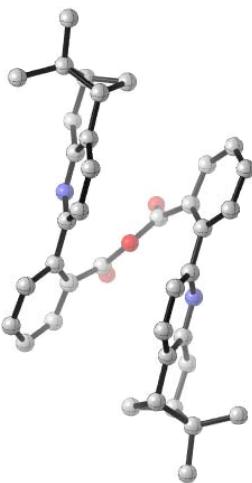


Thermal correction to Gibbs Free Energy= 0.587995 (Hartree/Particle)

Single point energy: -1805.04268603

Total free energy: -1804.45469103

ΔG : 0.8 kcal/mol



Thermal correction to Gibbs Free Energy= 0.587094 (Hartree/Particle)

Single point energy: -1805.04170734

Total free energy: -1804.45461334

ΔG : 0.9 kcal/mol

X-Ray data

Table 1 Crystal data and structure refinement for 2 (CCDC n° 2036677).

Empirical formula	C ₁₉ H ₁₉ NO ₂
Formula weight	293.35
Temperature/K	250(2)
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2
a/Å	25.3513(5)
b/Å	13.4380(7)
c/Å	9.4134(12)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	3206.9(4)
Z	8
ρ _{calcd} /g/cm ³	1.215
μ/mm ⁻¹	0.079
F(000)	1248.0
Crystal size/mm ³	0.35 × 0.35 × 0.13
Radiation	MoKα ($\lambda = 0.71073$)
2θ range for data collection/°	3.214 to 53.74
Index ranges	-32 ≤ h ≤ 32, -17 ≤ k ≤ 16, -11 ≤ l ≤ 11
Reflections collected	48181
Independent reflections	6844 [$R_{\text{int}} = 0.1844$, $R_{\text{sigma}} = 0.0780$]
Data/restraints/parameters	6844/0/405
Goodness-of-fit on F ²	1.026
Final R indexes [I>=2σ (I)]	$R_1 = 0.0720$, $wR_2 = 0.1638$
Final R indexes [all data]	$R_1 = 0.1163$, $wR_2 = 0.2001$
Largest diff. peak/hole / e Å ⁻³	0.29/-0.33
Flack parameter	2.2(10)

Table 2. Crystal data and structure refinement for 3 (CCDC n° 2036678).

Empirical formula	C ₁₉ H ₁₇ NO
Formula weight	275.33
Temperature/K	250(2)
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
a/Å	7.0872(8)
b/Å	12.3506(16)
c/Å	16.599(2)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	1452.9(3)
Z	4
ρ _{calc} g/cm ³	1.259
μ/mm ⁻¹	0.077
F(000)	584.0
Crystal size/mm ³	0.21 × 0.1 × 0.034
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/°	4.11 to 53.89
Index ranges	-9 ≤ h ≤ 8, -15 ≤ k ≤ 15, -21 ≤ l ≤ 20
Reflections collected	20659
Independent reflections	3105 [R _{int} = 0.1351, R _{sigma} = 0.0797]
Data/restraints/parameters	3105/0/193
Goodness-of-fit on F ²	0.964
Final R indexes [I>=2σ (I)]	R ₁ = 0.0471, wR ₂ = 0.0857
Final R indexes [all data]	R ₁ = 0.1384, wR ₂ = 0.1140
Largest diff. peak/hole / e Å ⁻³	0.15/-0.14
Flack parameter	-0.2(10)

Table 3. Crystal data and structure refinement for 6 (CCDC n° 2036679).

Empirical formula	C ₁₃ H ₁₁ NO ₂
Formula weight	213.23
Temperature/K	250(2)
Crystal system	orthorhombic
Space group	Pna ₂ 1
a/Å	13.0488(6)
b/Å	7.7427(5)
c/Å	10.6551(5)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	1076.52(10)
Z	4
ρ _{calc} g/cm ³	1.316
μ/mm ⁻¹	0.090
F(000)	448.0
Crystal size/mm ³	0.37 × 0.29 × 0.21
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/°	6.118 to 52.758
Index ranges	-15 ≤ h ≤ 16, -9 ≤ k ≤ 9, -13 ≤ l ≤ 13
Reflections collected	12251
Independent reflections	2163 [R _{int} = 0.0297, R _{sigma} = 0.0198]
Data/restraints/parameters	2163/1/150
Goodness-of-fit on F ²	1.032
Final R indexes [I>=2σ (I)]	R ₁ = 0.0295, wR ₂ = 0.0685
Final R indexes [all data]	R ₁ = 0.0430, wR ₂ = 0.0732
Largest diff. peak/hole / e Å ⁻³	0.10/-0.11
Flack parameter	0.4(3)

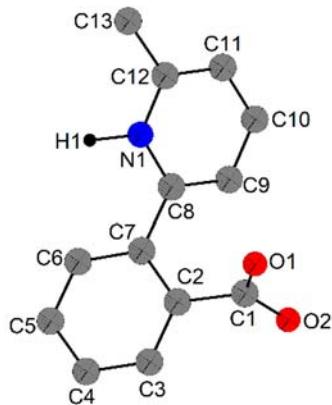


Figure S50. Molecular structure of **6** determined by single-crystal X-Ray diffraction with ellipsoids at 50% probability. The H atoms attached to C atoms were omitted for clarity.

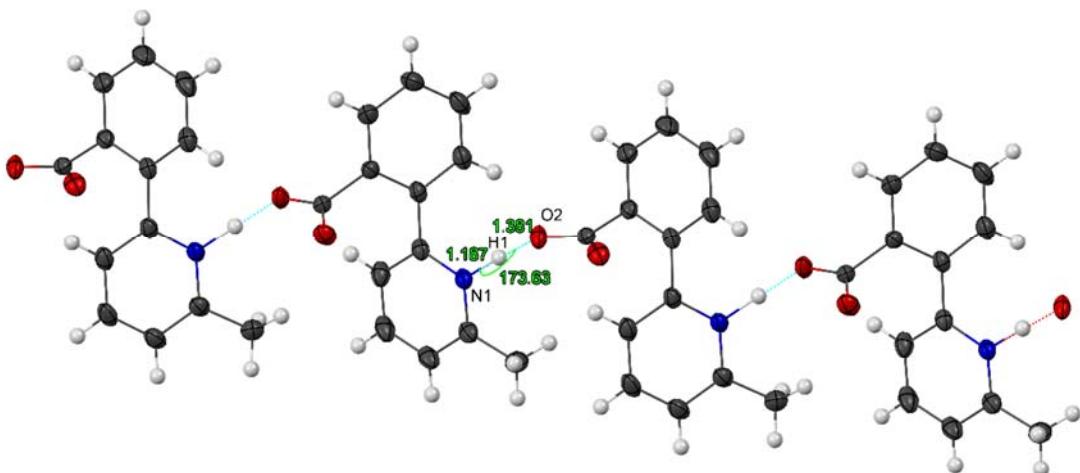


Figure S51. Depiction of the H-bond motif in the crystal of **6** with N1-H1-O1 angle, H1...O1 and N1-H1 distances shown in green.

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