SUPPORTING INFORMATION

1,2-Bis-perfluoroalkylations of alkenes and alkynes with perfluorocarboxylic anhydrides *via* the formation of perfluoroalkylcopper intermediates

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1. General Experimental

General: All reactions involving air- and/or moisture-sensitive compounds were conducted in a dry vessel under a positive pressure of nitrogen gas by using a balloon filled with it. Analytical thin-layer chromatography (TLC) was performed on glass plates coated with 0.25 mm 230–400 mesh silica gel (Merck, Silica gel 60 F_{254}) containing a fluorescent indicator. Visualization was accomplished by means of ultraviolet irradiation at 254 nm and/or by spraying an ethanolic solution of 12-molybdo(VI)phosphoric acid as a developing agent. Flash column chromatography was performed using Silica gel N-60 (spherical, neutral, 40–50 μ m, Kanto Chemical Co., Inc. (Kanto)).

Instrumentation:

NMR analysis

¹H, ¹³C, and ¹⁹F NMR spectra were recorded at room temperature on a JEOL JNM-ECS-400 NMR spectrometer at 400, 100, and 376 MHz, respectively. The proton chemical shift values are reported in parts per million (ppm, δ scale) downfield from tetramethylsilane and referenced to the proton resonance of CHCl₃ (δ 7.26). The carbon chemical shift values are reported in parts per million (ppm, δ scale) downfield from tetramethylsilane and referenced to the carbon resonance of CDCl₃ (δ 77.16). The fluorine chemical shift values are reported in parts per million (ppm, δ scale) with CFCl₃ (δ 0.00) as an external standard. Chemical shifts are reported in ppm and *J* values in hertz (Hz). The data are presented in the following order: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, m = multiplet and/or multiple resonances, and br = broad), signal area integration in natural numbers.

IR analysis

Infrared spectra were measured on a Thermo Nicolet iS5. Only diagnostic absorptions are listed.

HRMS analysis

ESI-MS spectra were measured on a Brucker micrOTOF-QII-RSL. The samples were diluted with MeOH for the measurement. EI-MS were taken on a JMS-T100GCV.

Materials: Chemical reagents were purchased from Wako Pure Chemical Industries, Ltd., Tokyo Chemical Industry Co., Ltd., Sigma-Aldrich Inc., and other commercial suppliers.

Preparation of known alkenes and alkynes (1):

The known alkenes (1a, 1c, 2e, 1e, 3e, 1f, 1g, 5) and alkynes $(4c, 6e, 4g^7)$ were prepared according to the literatures.

¹ H. Clavier, S. P. Nolan and M. Mauduit, Organometallics, 2008, 27, 2287.

² K. Mori, *Tetrahedron*, 2009, **65**, 2798.

³ J. D. Neukom, N. S. Perch, J. P. Wolfe, J. Am. Chem. Soc. 2010, 132, 6276.

⁴ P. R. Walker, C. D. Campbell, A. Suleman, G. Carr and E. A. Anderson, Angew. Chem., Int. Ed., 2013, 52, 9139.

⁵ J. Mo, S. H. Kim, P. H. Lee, *Org. Lett.* 2010, **12**, 424.

⁶ W.-K. Chan, C.-M. Ho, M.-K. Wong and C.-M. Che, J. Am. Chem. Soc. 2006, 128, 14796.

⁷ Y. R. Malpani, B. K. Biswas, H. S. Han, Y.-S. Jung and S. B. Han, Org. Lett. 2018, 20, 1693.

2. Additional Results

2-1. Optimization of the amount of bpy

The amount of bpy used in the 1,2-bis-trifluoromethylation of alkene **1a** was optimised (Table S1). The conversion of **1a** was very rapid without bpy present, and the desired product **2a** was obtained despite the low yield and problematic amount of allylic trifluoromethylation product **3a** that was also formed (entry 1). To sufficiently suppress the formation of by-product **3a**, the use of 4.0 equiv. of bpy with respect to **1a** (2.0 equiv. of Cu salt) is essential (entries 2 vs. 3), in which the (bpy)₂Cu(I) complex was presumably formed as a reactive Cu intermediate. Although the 4.0 equiv. of bpy slightly decreased the rate of conversion, the use of this amount remarkably improved the selectivity and yield of **2a**.

			[Cu(CH ₃ CN) ₄]PF ₆ (2.0 equiv) bpy (x equiv)			
		urea·H ₂ O ₂ (2.4 equiv)	$Ph \rightarrow 0$ γ_2 1a	Ph O		
TFAA (8	0 CF ₃ - 3.0 equiv)	CH ₂ Cl ₂ 0 °C, 1 h	CH ₂ Cl ₂ 25 °C, Time	O 2a CF3	ation allylic tr	3a ifluoromethylation
Entry	Amount of	Time	Conversion	NMF	R Yield	
Linuy	bpy (equiv)	(min)	of 1a (%)	2a	3a (<i>E/Z</i>)	2a:3a
1	0	5	>99%	32	28 (78/22)	53:47
2	2.0	5	>99%	48	25 (78/22)	66:34
3	4.0	5	78%	42	4 (91/19)	91:9
4	4.0	30	>99%	75	4 (91/19)	94:6

Table S1. Optimisation of the amount of 2,2'-bipyridine (bpy).

2-2. TEMPO trapping test

The 1,2-bis-trifluoromethylation reaction was performed in the presence of TEMPO to trap the radical intermediate of this reaction (Scheme S1). As a result, the 1,2-bis-trifluoromethylation reaction was suppressed and a TEMPO-CF₃ adduct was obtained. In addition, the ESI-MS analysis of the crude mixture (HRMS-ESI (m/z) [M+Na]⁺ calcd. for C₂₀H₂₇F₃N₂NaO₂, 407.1917; found, 407.1916) confirmed the formation of TEMPO-bearing oxy-trifluoromethylation product **6e**, suggesting the generation of a vinyl radical intermediate.



Scheme S1. TEMPO trapping test to detect the radical intermediate of the 1,2-bistrifluoromethylation.

2-3. Detection of (bpy)Cu(CF₃)₃ by ¹⁹F NMR analysis

¹⁹F NMR analysis (using acetone- d_6) revealed the presence of a (bpy)Cu(CF₃)₃ complex in the crude mixture obtained after the workup of the 1,2-bis-trifluoromethylation reaction of **4e**.⁸



Figure S1. ¹⁹F NMR analysis of the crude mixture obtained from the 1,2-bis-trifluoromethylation reaction of 4e.

2-4. Stability comparison of copper intermediates

The thermal stabilities of several Cu(I) intermediates in the presence and absence of the bpy ligand (2.0 equiv. with respect to Cu) were compared to that of the precursor ($[Cu(CH_3CN)_4]^+$) by means of DFT calculations (Scheme S2a). According to the literature⁹ and experimental results in Table S1, in the presence of bpy, cationic intermediate **A** bearing two bpy ligands is formed. Under the optimal conditions determined for the 1,2-bis-trifluoromethylation process of the present study, a relatively high concentration of trifluoroacetic acid was found to exist in the reaction mixture. Therefore, neutral trifluoroacetate intermediate **F** and intermediate **G** bearing one bpy unit were also proposed to have formed; however, the calculated free energy relative to that of $[Cu(CH_3CN)_4]^+$ indicated that

⁸ The ¹⁹F NMR shifts and coupling patterns in acetone-*d*₆ were in good agreement with those reported in the literature: A. M. Romine, N. Nebra, A. I. Konovalov, E. Martin, J. Benet-Buchholz and V. V. Grushin, *Angew. Chem., Int. Ed.*, 2015, **54**, 2745.

⁹ (a) X. Lin, C. Hou, H. Li and Z. Weng, *Chem. Eur. J.*, 2016, **22**, 2075; (b) X. Lin, Z. Li, X. Han and Z. Weng, *RSC Adv.*, 2016, **6**, 75465.

intermediate **A** is the most stable (-25.7 kcal/mol for **A** vs. -23.9 and -21.6 kcal/mol for **F** and **G**, respectively). Thus, **A** was concluded to be the reactive Cu(I) intermediate formed during 1,2-bis-trifluoromethylation. In the absence of bpy, the formation of a neutral species *via* the substitution of the precursor acetonitrile ligand with trifluoroacetate is favoured; this is in contrast to what was observed for the bpy complex. Furthermore, intermediates **J** and **K**, which are formed when one or two acetonitrile ligands, respectively, dissociate from **I**, are more stable than intermediate **I**.

We then examined the effect of the ligand on the free energy change for the formation of Cu(II)– CF₃ intermediates from Cu(I) intermediates and CF₃ radicals (Scheme S2b). As a result, the formation of [(bpy)₂Cu(II)(CF₃)]⁺ (**C**) from [(bpy)₂Cu]⁺ (**A**) and CF₃ radicals ($\Delta G_{Cu(II)} = -16.1$ kcal/mol) decreased the free energy to a greater extent than the acetonitrile–Cu(II)–CF₃ complexes did ($\Delta G_{Cu(II)} \ge -8.8$ kcal/mol).



Scheme S2. Ligand effect on the 1,2-bis-trifluoromethylation of alkenes: (a) Identification of stable Cu(I) intermediates (the free energies of each intermediate relative to that of $[Cu(CH_3CN)_4]^+$ are shown in parentheses). (b) A comparison of the free energy change upon the formation of Cu(II)–CF₃ intermediates.

2-5. Possible pathways to $(bpy)Cu(CF_3)_3$ (**Q**)

Here, we discuss possible pathways for the formation of (bpy)Cu(CF₃)₃ (**Q**), which was observed in the crude mixture by NMR analysis. First, Cu(II)–CF₃ intermediate **C** reacts with the CF₃ radical, affording Cu(III) intermediate **R**. Then, the ligand exchange of intermediate **R** replaces the bpy ligand with trifluoroacetate and generates intermediate **S**. Finally, **Q** can be formed from **S** by ligand exchange with C or the disproportionation of S. The free energy changes of these steps suggest that ligand exchange between S and C is more likely than the disproportionation of S.



Scheme S3. Possible pathways for the formation of (bpy)Cu(CF₃)₃ (Q). The free energy of each intermediate relative to that of C are shown in parentheses.

2-6. Ligand effect on the coupling reaction with Cu(II)– CF_3 intermediates

The effect of the ligand on the coupling step was investigated by comparing the free energy changes between the reactions of the vinyl radical with $[(bpy)_2CuCF_3]^+$ (**C**) and the corresponding acetonitrile complex $(CH_3CN)_2Cu(O_2CCF_3)(CF_3)$ (**O**). The results indicate that bpy does not stabilise the Cu(III) intermediate because the decrease in the free energy of the formation of **D** is much smaller than that of the free energy of the formation of acetonitrile–Cu(III) intermediate **U** ($\Delta G_{Cu(III)} = -2.5$. kcal/mol for the formation of **D** from **C** and -11.8 kcal/mol for the formation of **U** from **O**). This could be due to steric repulsion between the ligands on **D**. In addition, the activation energies for the reductive eliminations of **D** and **U** indicated that bpy did not accelerate either process ($\Delta G_{RE}^{\ddagger} = +7.3$ kcal/mol for the reductive elimination of **D** and +5.8 kcal/mol for that of **U**). Thus, we concluded that the crucial role of bpy is to stabilise Cu(II)–CF₃ intermediate **C**, as discussed in the manuscript.



Scheme S4. Ligand effect on the coupling step.

2-7. Discussion of ligand effect on the product selectivity

When the reaction of **4e** was carried out in the absence of a ligand, we found that significant amounts of oxy-trifluoromethylation products, vinyl trifluoroacetate **6e** and ketone **7e**, were obtained instead of the desired bis-trifluoromethylated product (**5e**) (Scheme S5a). Ketone **7e** was considered to be formed by the hydrolysis of **6e** during the workup; it was also observed in low yield in the crude mixture of the reaction in the presence of bpy. Interestingly, the ratio of **7e** to **5e** was found to increase with increasing reaction time under optimal conditions using bpy.



Scheme S5. Proposed pathways of the oxy-trifluoromethylation side-reaction.

The proposed mechanism for the formation of **6e** and **7e** is depicted in Scheme S5b. Cationic Cu(II) trifluoroacetate intermediate **B**, which is formed during the CF₃-radical formation reaction that occurs between diacyl peroxide and **A**, was postulated to give the oxy-trifluoromethylation products. We proposed two pathways involving **B**; the first involves a C–O coupling reaction between **B** and the vinyl radical (top arrow). DFT calculations suggested that Cu(III) intermediate **V**, bearing vinyl and

trifluoroacetate groups, could be formed ($\Delta G_{Cu(III)tfa} = +14.1$ kcal/mol), and subsequent reductive elimination would afford the oxy-trifluoromethylation products ($\Delta G^{\ddagger} = +10.0$ kcal/mol). The concentration of intermediate V should increase at a later stage of the reaction. Furthermore, the stability of V is lower than that of $(bpy)_2Cu(CF_3)(vinyl)$ (D) ($\Delta G_{Cu(III)tfa} = +14.1$ kcal/mol for the formation of V vs. $\Delta G_{Cu(III)} = -2.5$ kcal/mol for the formation of D), which suggests that the oxytrifluoromethylation is much slower than the desired 1,2-bis-trifluoromethylation. This is in good agreement with the experimental results. The second proposed route is a vinyl cation-forming pathway that proceeds *via* the single-electron transfer (SET) between **B** and the vinyl radical (lower arrow). This pathway may also occur, but it is likely much slower than the other pathway due to the large increase in the free energy of the latter ($\Delta G_{ox} = +22.7 \text{ kcal/mol}$). On the other hand, without bpy, the SET event can proceed smoothly; this is according to its small change in free energy. Specifically, the change in free energy for the SET of the vinyl radical with $(CH_3CN)Cu(O_2CCF_3)_2$ is $\Delta G = -2.7$ kcal/mol. In the absence of bpy, not only does the C-O coupling increase the yield of the oxytrifluoromethylation products (the change in the free energy change of the formation of (CH₃CN)Cu^{III}(O₂CCF₃)₂(vinyl), $\Delta G = -13.2$ kcal/mol; the activation energy of reductive elimination of (CH₃CN)Cu^{III}(O₂CCF₃)₂(vinyl), $\Delta G^{\ddagger} = +8.4$ kcal/mol), but the SET process does as well. Therefore, bpy was considered to play a role in inhibiting the undesired oxidation of the vinyl radical and in improving the product selectivity. This is also supported by the improvement in the selectivity of the 1,2-bis-trifluoromethylation of alkenes brought about by bpy (Table 1). Furthermore, the allylic trifluoromethylation side reaction, which is known to proceed via a carbocation generated by oxidation of the alkyl radical intermediate with a Cu(II) species (Scheme 1a), could be suppressed by bpy.

3. Experimental Procedures

General procedure

Trifluoroacetic anhydride (220 μ L, 1.6 mmol) was slowly added to a suspension of urea H₂O₂ (45 mg, 0.48 mmol) in CH₂Cl₂ (2.0 mL) at 0 °C. After stirring for 1 h, the alkene or alkyne (0.20 mmol), 2,2'bipyridine (125 mg, 0.80 mmol), and [Cu(CH₃CN)₄]PF₆ (149 mg, 0.40 mmol) were added at 0 °C. The mixture was immediately warmed to 25 °C and stirred for 30 min. After the addition of 5 mL of CH₂Cl₂, the reaction was quenched with an aqueous NaHCO₃ solution at 0 °C and then stirred at room temperature for 20 min. The aqueous layer was extracted three times with 5 mL of CH₂Cl₂. The combined organic phases were checked with XploSens PS[®] to confirm the absence of peroxide, and the water phase was treated with saturated Na₂S₂O₃ to decompose H₂O₂. The combined organic phases were dried over Na₂SO₄ and the solvent was evaporated. Then, α , α , α -trifluorotoluene (20 mg) was added as an internal standard, followed by a deuterated solvent (CDCl₃ or acetone-*d*₆). ¹H and ¹⁹F NMR analyses were conducted to estimate the NMR yield.^{10,11} After the evaporation of the deuterated solvent, flash column chromatography of the crude mixture on silica gel afforded the target compound.

6,6,6-Trifluoro-4-(trifluoromethyl)hexyl benzoate (2a)

 $Ph \xrightarrow{CF_3} CF_3$ The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (38 mg, 58% yield) after silica gel column chromatography (CH₂Cl₂/hexane = 50/50).

¹H NMR (400 MHz, CDCl₃)

1.73–2.03 (m, 4H), 2.17–2.32 (m, 1H), 2.44–2.65 (m, 2H), 4.30–4.55 (m, 2H), 7.45 (t, *J* = 7.4 Hz, 2H), 7.57 (t, *J* = 7.4 Hz, 1H), 8.03 (d, *J* = 7.4 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

25.1, 25.7, 32.7 (q, *J* = 30 Hz), 38.4 (q, *J* = 27 Hz), 64.2, 126.0 (q, *J* = 274 Hz), 127.2 (q, *J* = 277 Hz), 128.6 (2C), 129.7 (2C), 130.2, 133.2, 166.6

¹⁹F NMR (376 MHz, CDCl₃)

-71.0 (d, J = 8.6 Hz, 3F), -64.3 (t, J = 10.1 Hz, 3F)

IR (neat, cm^{-1})

2965, 1721, 1453, 1276, 1253, 1213, 1149, 1116, 1017, 1028, 712

HRMS-ESI (m/z)

[M+Na]⁺ calcd. for C₁₄H₁₄F₆O₂, 351.0790; found, 351.0792

¹⁰ The stereochemistry of 1,2-bis-perfluoroalkylated alkenes **5a**, **5d–5f** was assigned by comparison with literature data. The stereochemistry of the other alkenes (**5b**, **5c**, **5g–5k**) was determined by the characteristic coupling pattern of their ¹⁹F NMR spectra; only *Z*-isomers show F–F coupling. See, S. Guo, D. I. AbuSalim, S. P. Cook, *Angew. Chem., Int. Ed.*, 2019, **58**, 11704.

¹¹ The E/Z ratio of 1,2-bis-perfluoroalkylated alkenes **5a**-k was determined by ¹⁹F NMR analysis of the crude product.

Ethyl 12,12,12-trifluoro-10-(trifluoromethyl)dodecanoate (2b)

 c_{F_3} The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (44mg, 62% yield) after silica gel column chromatography (CH₂Cl₂/hexane = 50/50).

1 H NMR (400 MHz, CDCl₃)

1.25 (t, *J* = 7.2 Hz, 3H), 1.25–1.34 (m, 6H), 1.36–1.45 (m, 2H), 1.48–1.76 (m, 6H), 2.12–2.24 (m, 1H), 2.28 (t, *J* = 7.6 Hz, 2H), 2.35–2.53 (m, 2H), 4.12 (q, *J* = 7.2 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

14.4, 25.1, 26.2, 28.4, 29.2 (2C), 29.3, 29.5, 32.8 (q, *J* = 30 Hz), 34.5, 38.0 (q, *J* = 27 Hz), 60.3,

126.1 (q, J = 274 Hz), 127.4 (q, J = 277 Hz), 174.0

¹⁹F NMR (376 MHz, CDCl₃)

-71.1 (d, *J* = 8.6 Hz, 3F), -64.5 (t, *J* = 10.2 Hz, 3F)

IR (neat, cm^{-1})

2932, 2860, 1738, 1467, 1374, 1252, 1176, 1143, 1115, 1037, 801

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for C₁₅H₂₄F₆O₂, 373.1573; found, 373.1578

15,15,15-Trifluoro-13-(trifluoromethyl)pentadecan-2-one (2c)

 $\begin{array}{c} \overset{O}{\underset{10}{}} \overset{CF_3}{\underset{10}{}} \overset{CF_3}{\underset{10}{}} \end{array} \\ \text{The reaction was carried out on according to general procedure. The target compound was obtained as a coreless oil (37 mg, 55% yield) after silica gel column chromatography (CH_2Cl_2/hexane = 40/60). \end{array}$

¹H NMR (400 MHz, CDCl₃)

1.24-1.30 (m, 12H), 1.36-1.45 (m, 2H), 1.50-1.60 (m, 3H), 1.65-1.76 (m, 1H), 2.13 (s, 3H),

2.15–2.29 (m, 1H), 2.37–2.52 (m, 2H), 2.41 (t, *J* = 7.6 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

24.0, 26.3, 28.4, 29.3, 29.4, 29.5 (3C), 29.6, 30.0, 32.8 (q, J = 30 Hz), 37.9 (q, J = 27 Hz), 44.0,

126.0 (q, J = 274 Hz), 127.3 (q, J = 277 Hz), 209.6

¹⁹F NMR (376 MHz, CDCl₃)

-117.5 (m, 1F), -71.1 (d, J = 8.6 Hz, 3F), -64.5 (t, J = 10.2 Hz, 3F)

IR (neat, cm^{-1})

2929, 2858, 1717, 1409, 1362, 1251, 1142, 1118

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for $C_{16}H_{26}F_6O$, 371.1780; found, 371.1780.

2-(5,5,5-Trifluoro-3-(trifluoromethyl)pentyl)isoindoline-1,3-dione (2d)



The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (37 mg, 54% yield) after silica gel column chromatography (CH_2Cl_2 /hexane = 60/40). The compound was

identified by comparison of its spectra with the literature data.¹¹

¹H NMR (400 MHz, CDCl₃)

1.92–2.03 (m, 1H), 2.08–2.20 (m, 1H), 2.24–2.39 (m, 1H), 2.47–2.66 (m, 2H), 3.82 (t, *J* = 7.2 Hz, 2H), 7.74 (dd, *J* = 5.6, 3.2 Hz, 2H), 7.86 (dd, *J* = 5.6, 3.2 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

27.6, 32.9 (q, *J* = 30 Hz), 35.3, 36.3 (q, *J* = 28 Hz), 123.6 (2C), 125.8 (q, *J* = 274 Hz), 126.9 (q, *J* = 277 Hz), 132.0 (2C), 134.3 (2C), 168.2 (2C)

¹⁹F NMR (376 MHz, CDCl₃)

-71.4 (d, J = 8.6 Hz, 3F), -64.3 (t, J = 10.2 Hz, 3F)

IR (neat, cm^{-1})

2962, 2924, 2854, 1773, 1718, 1399, 1386, 1253, 1175, 1145, 1093, 1020, 870, 798 HRMS-ESI (m/z)

[M+Na]⁺ calcd. for C₁₄H₁₁F₆NO₂, 362.0586; found, 362.0578.

6,6,6-Trifluoro-N-(4-fluorophenyl)-4-(trifluoromethyl)hexanamide (2e)



The reaction was carried out on according to general procedure. The target compound was obtained as a coreless oil (30 mg, 46% yield) after silica gel column chromatography (CH₂Cl₂).

¹H NMR (400 MHz, CDCl₃)

2,04–2.33 (m, 3H), 2.44–2.69 (m, 4H), 7.02 (t, *J* = 8.7 Hz, 2H), 7.16 (br, 1H), 7.40–7.48 (m, 2H) ¹³C NMR (100 MHz, CDCl₃)

24.0, 33.0 (q, *J* = 30 Hz), 33.6, 37.3 (q, *J* = 27 Hz), 115.9 (d, *J* = 22 Hz, 2C), 122.0 (d, *J* = 8 Hz, 2C), 125.9 (q, *J* = 274 Hz), 127.1 (q, *J* = 278 Hz), 133.6 (d, *J* = 2 Hz), 159.5 (d, *J* = 243 Hz), 169.3 ¹⁹F NMR (376 MHz, CDCl₃)

-117.5 (m, 1F), -70.7 (d, J = 8.6 Hz, 3F), -62.3 (t, J = 11.3 Hz, 3F) IR (neat, cm⁻¹)

3291, 1662, 1617, 1558, 1539, 1509, 1409, 1250, 1214, 1144, 1109, 834 HRMS-ESI (m/z)

[M+Na]⁺ calcd. for C₁₃H₁F₇NO, 354.0699; found, 354.0695

4-Methyl-N-(6,6,6-trifluoro-4-(trifluoromethyl)hexyl)benzenesulfonamide (2f)



The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (49 mg, 66% yield) after silica gel column chromatography (CH_2Cl_2).

¹H NMR (400 MHz, CDCl₃)

1.52–1.78 (m, 4H), 2.00–2.28 (m, 1H), 2.31–2.51 (m, 2H), 2.43 (s, 3H), 2.92–3.01 (m, 2H), 4.37

(t, J = 6.0 Hz, 1H), 7.32 (d, J = 8.3 Hz, 2H), 7.74 (d, J = 8.3 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

21.6, 25.5, 26.6, 32.6 (q, *J* = 30 Hz), 37.7 (q, *J* = 28 Hz), 43.0, 125.9 (q, *J* = 274 Hz), 127.0 (q, *J* = 277 Hz), 127.2 (2C), 130.0 (2C), 136.9, 143.8

¹⁹F NMR (376 MHz, CDCl₃)

-71.0 (d, *J* = 8.6 Hz, 3F), -64.4 (t, *J* = 10.2 Hz, 3F)

IR (neat, cm^{-1})

3287, 2929, 1326, 1300, 1288, 1202, 1151, 1117, 1094, 816, 670, 662

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for $C_{14}H_{17}F_6NO_2S$, 400.0776; found, 400.0771.

2-Methyl-5,6-bis(trifluoromethyl)hexahydro-1*H*-isoindole-1,3(2H)-dione (2g)

 $F_{3}C$ $F_{3}C$ F

¹H NMR (400 MHz, CDCl₃)

1.54 (q, *J* = 13.9 Hz, 1H), 1.68–1.78 (m, 1H), 2.41 (dt, *J* = 13.9, 5.4 Hz, 1H), 2.49 (ddd, *J* = 15.3, 6.9, 2.5 Hz, 1H), 2.53–2.66 (m, 1H), 2.70–2.81 (m, 1H), 2.83–2.92 (m, 1H), 3.00 (s, 3H), 2.98–3.08 (m, 1H)

¹³C NMR (100 MHz, CDCl₃)

19.3, 21.1, 25.1, 35.7, 35.7 (q, *J* = 30 Hz), 36.8 (q, *J* = 28 Hz), 37.3, 125.3 (d, *J* = 41 Hz), 128.1 (d, *J* = 40 Hz), 177.9, 178.2

¹⁹F NMR (376 MHz, CDCl₃)

-73.2 (m, 3F), -70.0 (m, 3F)

IR (neat, cm^{-1})

1703, 1700, 1286, 1256, 1218, 1183, 1163, 1130, 1102, 980, 970, 608 HRMS-ESI (*m/z*)

¹² The stereochemistry was confirmed by NOESY experiments.

 $[M+Na]^+$ calcd. for C₁₁H₁₁F₆NO₂, 326.0587; found, 326.0581

4-Methyl-N-(4,4,4-trifluoro-2-(trifluoromethyl)butyl)benzenesulfonamide (2h)

Ts $R_{H} \sim CF_{3}$ The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (47 mg, 68% yield) after silica gel column chromatography (CH₂Cl₂).

¹H NMR (400 MHz, CDCl₃)

2.33–2.52 (m, 2H), 2.44 (s, 3H), 2.55–2.70 (m, 1H), 3.20–3.32 (m, 2H), 4.71 (br, 1H), 7.34 (d, J = 8.4 Hz, 2H), 7.74 (d, J = 8.4 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)

21.7, 30.1 (q, *J* = 31 Hz), 38.6 (q, *J* = 27 Hz), 40.9, 125.7 (q, *J* = 274 Hz), 126.3 (q, *J* = 277 Hz), 127.2 (2C), 130.2 (2C), 136.4, 144.4

¹⁹F NMR (376 MHz, CDCl₃)

-69.9 (d, J = 8.6 Hz, 3F), -64.5 (t, J = 10.2 Hz, 3F)

IR (neat, cm^{-1})

3282, 1599, 1496, 1436, 1390, 1332, 1307, 1291, 1274, 1251, 1155, 1119, 1092, 815, 667 HRMS-ESI (m/z) [M+Na]⁺ calcd. for C₁₂H₁₃F₆NO₂S, 372.0463; found, 372.0464

4-Methyl-N-(4,4,5,5,5-pentafluoro-2-(perfluoroethyl)pentyl)benzenesulfonamide (2i)

Ts $h \to CF_2CF_3$ The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (45 mg, 50% yield) after silica gel column chromatography (CH₂Cl₂).

¹H NMR (400 MHz, CDCl₃)

2.28–2.60 (m, 2H), 2.44 (s, 3H), 2.70–2.85 (m, 1H), 3.22–3.38 (m, 2H), 4.69 (t, J = 7.2 Hz, 1H),

7.34 (d, *J* = 8.4 Hz, 2H), 7.74 (d, *J* = 8.4 Hz, 2H)

¹³C NMR (100 MHz, CDCl₃)¹³

21.7, 25.9 (t, *J* = 21 Hz), 34.8 (t, *J* = 21 Hz), 41.3, 127.2 (2C), 130.2 (2C), 136.3, 144.4 ¹⁹F NMR (376 MHz, CDCl₃)

-118.0 (m, 2F), -116.5 (m, 2F), -85.5 (s, 3F), -82.3 (s, 3F) IR (neat, cm⁻¹)

3304, 1339, 1307, 1195, 1162, 1093, 1069, 1043, 814, 668, 663

HRMS-ESI (m/z)

[M+Na]⁺ calcd. for C₁₄H₁₃F₁₀NO₂S, 472.0400; found, 472.0400.

¹³ The carbons of perfluoroalkyl group could not be assigned because of low intensity of signals and their complex coupling.

4-Methyl-*N*-(4,4,5,5,6,6,6-heptafluoro-2-(perfluoropropyl)hexyl)benzenesulfon-amide (2j)

Ts $N \rightarrow CF_2CF_2CF_3$ The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (47 mg, 43% yield) after silica gel column chromatography (CH₂Cl₂).

1 H NMR (400 MHz, CDCl₃)

2.35–2.65 (m, 2H), 2.44 (s, 3H), 2.82–2.97 (m, 1H), 3.22–3.38 (m, 2H), 4.78 (t, *J* = 7.2 Hz, 1H),

7.33 (d, J = 8.4 Hz, 2H), 7.73 (d, J = 8.4 Hz, 2H)

13 C NMR (100 MHz, CDCl₃) 13

21.7, 25.8 (t, *J* = 21 Hz), 35.1 (t, *J* = 21 Hz), 41.3, 127.2 (2C), 130.1 (2C), 136.2, 144.4 ¹⁹F NMR (376 MHz, CDCl₃)

-127.5 (m, 2F), -125.0 (m, 2F), -114.9 (m, 2F), -113.2 (m, 2F), -80.4 (s, 3F), -80.3 (s, 3F) IR (neat, cm⁻¹)

3296, 1436, 1330, 1123, 1176, 1158, 1107, 1081, 922, 814, 746, 714

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for $C_{16}H_{13}F_{14}NO_2S$, 572.0336; found, 572.0336.

(E)-1-bromo-2-(1,1,1,4,4,4-hexafluorobuten-2-yl)benzene (5a)

Br CF_3 The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (28 mg, 42% yield) after silica gel column chromatography (pentane).

¹H NMR (400 MHz, CDCl₃)

6.58 (qq, *J* = 7.0, 1.4 Hz, 1H), 7.22 (dd, *J* = 7.5, 1.9 Hz, 1H), 7.31 (td, *J* = 7.7, 1.9 Hz, 1H), 7.38 (dt, *J* = 7.5, 1.3 Hz, 1H), 7.66 (dd, *J* = 7.7, 1.3 Hz, 1H)

¹³C NMR (100 MHz, CDCl₃)

121.3 (q, *J* = 271 Hz), 121.6 (q, *J* = 275), 123.3, 124.6 (qq, *J* = 35, 6 Hz), 127.3, 130.3, 130.6, 131.3, 133.2, 139.4 (qq, *J* = 32, 6 Hz)

¹⁹F NMR (376 MHz, CDCl₃)

```
-67.8 (d, J = 1.4 Hz, 3F), -60.6 (m, 3F)
```

IR (neat, cm^{-1})

1269, 1187, 1145, 756, 738, 663, 646, 613

HRMS-EI (m/z)

```
[M] calcd. for C<sub>10</sub>H<sub>5</sub>BrF<sub>6</sub>, 317.9479; found, 317.9476
```

(*E*)-1-bromo-2-(1,1,1,2,2,5,5,6,6,6-decafluorohexen-3-yl)benzene (5b)

 CF_2CF_3 The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (38 mg, 45% yield) after silica gel column chromatography (pentane).

1 H NMR (400 MHz, CDCl₃)

6.49 (t, J = 12.8 Hz, 1H), 7.15 (m, 1H), 7.30 (td, J = 7.7, 1.9 Hz, 1H), 7.35 (td, J = 7.5, 1.4 Hz,

1H), 7.67 (dd, *J* = 7.7, 1.4 Hz, 1H)

¹³C NMR (100 MHz, CDCl₃)¹³

123.8, 125.2 (m), 127.0, 130.5, 130.7, 131.3, 133.0, 141.9 (m)

¹⁹F NMR (376 MHz, CDCl₃)

-117.11 (d, *J* = 268 Hz, 1F), -114.88 (dd, *J* = 275, 14 Hz, 1F), -113.17 (dd, *J* = 275, 11 Hz, 1F),

-110.60 (d, *J* = 268 Hz, 1F), -82.28 (d, *J* = 6.2 Hz, 3F), -84.76 (d, *J* = 6.7 Hz, 3F)

IR (neat, cm^{-1})

1327, 1120, 1168, 1130, 1065, 1048, 1034, 753, 744, 721, 707, 666, 658 HRMS-EI (*m*/*z*)

[M] calcd. for C₁₂H₅BrF₁₀, 417.9415; found, 417.9403

(*E*)-1-bromo-2-(1,1,1,2,2,3,3,6,6,7,7,8,8,8-tetradecafluoroocten-4-yl)benzene (5c)

 $\begin{array}{c} & & \\ & &$

¹H NMR (400 MHz, CDCl₃)

6.49 (dd, *J* = 14.3, 11.8 Hz, 1H), 7.17 (m, 1H), 7.30 (td, *J* = 7.8, 1.8 Hz, 1H), 7.38 (dt, *J* = 7.5, 1.4 Hz, 1H), 7.66 (dd, *J* = 7.8, 1.4 Hz, 1H)

¹³C NMR (100 MHz, CDCl₃)¹³

124.0, 125.8 (m), 127.0, 130.4, 130.9, 131.3, 133.0, 142.0 (m)

¹⁹F NMR (376 MHz, CDCl3)

-127.46 (m, 2F), -124.41 to -125.64 (m, 1F), -123.31 to -124.41 (m, 1F), -114.41 (m, 1F), -111.49 to -113.78 (m, 1F), -108.53 to -111.01 (m, 1F), -106.51 (m, 1F), -80.37 (m, 3F), -79.84 to -80.21 (m, 3F)

IR (neat, cm^{-1})

1345, 1229, 1184, 1118, 988, 961, 745, 724

HRMS-EI (m/z)

[M] calcd. for C₁₄H₅BrF₁₄, 517.9351; found, 517.9336

(E)-4-(1,1,1,4,4,4-hexafluorobuten-2-yl)benzaldehyde (5d)



The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (21 mg, 40% yield) after silica gel column chromatography (CH_2Cl_2 /hexane = 20/80).

1 H NMR (400 MHz, CDCl₃)

6.57 (qq, *J* = 7.2, 1.5 Hz, 1H), 7.48 (d, *J* = 8.2 Hz, 2H), 7.96 (d, *J* = 8.2 Hz, 2H), 10.07 (s, 1H) ¹³C NMR (100 MHz, CDCl₃)

121.3 (q, J = 272 Hz), 121.7 (q, J = 275 Hz), 123.9 (qq, J = 35, 6 Hz), 129.7 (2C), 129.8 (2C),

134.8, 137.3, 140.2 (qq, *J* = 31, 5 Hz), 191.5

¹⁹F NMR (376 MHz, CDCl₃)

-67.9 (s, 3F), -58.1 (d, J = 7.2 Hz, 3F)

IR (neat, cm⁻¹)

1710, 1267, 1185, 1140, 825, 740, 649, 634, 623

HRMS-EI (m/z)

[M] calcd. for C₁₁H₆F₆O, 268.0323; found, 268.0310

(E)-N-(4-(1,1,1,4,4,4-hexafluorobuten-2-yl)phenyl)acetamide (5e)

 CF_3 The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (29 mg, 48% yield) after silica gel column chromatography (AcOEt/hexane = 30/70).

¹H NMR (400 MHz, acetone- d_6)

2.10 (s, 3H), 6.84 (qq, *J* = 7.2, 1.5 Hz, 1H), 7.29 (d, *J* = 8.6 Hz, 2H), 7.77 (d, *J* = 8.6 Hz, 2H), 9.37 (bs, 1H)

¹³C NMR (100 MHz, acetone- d_6)

24.3, 119.5 (2C), 121.5 (q, *J* = 271 Hz), 123.1 (q, *J* = 274 Hz), 123.9 (qq, *J* = 35, 5.5 Hz), 130.2 (2C), 141.6 (qq, *J* = 31, 5 Hz), 142.1, 169.3

¹⁹F NMR (376 MHz, acetone- d_6)

```
-68.7 (s, 3F), -58.6 (d, J = 7.2 Hz, 3F)
```

IR (neat, cm^{-1})

1671, 1606, 1543, 1538, 1515, 1403, 1321, 1269, 1199, 1136

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for C₁₂H₉F₆NO, 320.0481; found: 320.0479.

(E)-1-(1,1,1,4,4,4-hexafluorobuten-2-yl)-4-methoxybenzene (5f)



The reaction was carried out on according to general procedure. The target compound was obtained as a colorless oil (9 mg, 15% yield) after silica gel column chromatography (hexane).

1 H NMR (400 MHz, CDCl₃)

3.84 (s, 3H), 6.45 (qq, *J* = 7.2, 1.4 Hz, 1H), 6.94 (d, *J* = 8.8 Hz, 2H), 7.23 (d, *J* = 8.8 Hz, 2H) ¹³C NMR (100 MHz, CDCl₃)

55.4, 114.0 (2C), 121.0, 121.7 (q, *J* = 272 Hz), 122.1 (q, *J* = 276 Hz), 122.6 (qq, *J* = 35, 5 Hz), 130.3 (2C), 141.2 (qq, *J* = 31, 5 Hz), 160.9

¹⁹F NMR (376 MHz, CDCl₃)

-68.2 (s, 3F), -57.9 (d, J = 7.2 Hz, 3F)

IR (neat, cm⁻¹)

2956, 2923, 2853, 1734, 1730, 1654, 1632, 1468, 1287, 1263, 1222, 1183, 1163, 1135, 808, 722 HRMS-EI (*m*/*z*)

[M] calcd. for C₁₁H₈F₆O, 270.0479; found, 270.0475

(*E*)-4-(1,1,1,2,2,5,5,6,6,6-decafluorohexen-3-yl)benzyl 2,2,3,3,3-pentafluoropropa-noate (5g)



The reaction of (4-ethynylphenyl)methane **4e** was carried out on according to general procedure. The target compound was obtained as a colorless oil (56 mg, 54% yield) after silica gel column chromatography (CH_2Cl_2 /hexane = 5/95).

¹H NMR (400 MHz, CDCl₃)

5.42 (s, 2H), 6.47 (t, *J* = 13.0 Hz, 1H), 7.28 (d, *J* = 8.2 Hz, 2H), 7.41 (d, *J* = 8.2 Hz, 2H) ¹³C NMR (100 MHz, CDCl₃)¹³

69.0, 123.9 (m), 127.9 (2C), 129.7 (2C), 129.9, 135.1, 142.3 (m), 158.4 (t, *J* = 30 Hz) ¹⁹F NMR (376 MHz, CDCl₃)

-121.5 (s, 2F), -115.0 (s, 2F), -110.9 (dd, *J* = 13.0, 2.9 Hz, 2F), -85.0 (s, 3F), -82.8 (s, 3F), -81.7 (s, 3F)

IR (neat, cm^{-1})

1784, 1331, 1302, 1120, 1150, 1129, 1063, 1034, 746, 713, 664

HRMS-EI (m/z)

```
[M] calcd. for C_{16}H_7F_{15}O_2, 516.0207; found, 516.0173
```

(1,1,1,4,4,4-hexafluoro-3-methylbuten-2-yl)benzene (5h)¹⁴



The reaction was carried out on according to general procedure. After the extraction, the organic phase was very carefully evaporated because of high volatility of the product (200 mmHg, 20 $^{\circ}$ C). The target compound was identified by ¹H and ¹⁹F NMR

analyses, and the ¹⁹F NMR yield was estimated (Figure S6, 46% yield, E:Z = 67:33).

E-isomer

¹H NMR (400 MHz, CDCl₃)

2.18–2.22 (m, 3H), 7.13–7.20 (m, 2H), 7.30–7.42 (m, 3H) ¹⁹F NMR (376 MHz, CDCl₃)

-61.7 (s, 3F), -59.3 (s, 3F)

Z-isomer

¹H NMR (400 MHz, CDCl₃)

1.75 (q, J = 2.2 Hz, 3H), 7.16–7.19 (m, 2H), 7.41–7.46 (m, 3H)

¹⁹F NMR (376 MHz, CDCl₃)

-62.1 (q, J = 14.5 Hz, 3F), -58.7 (q, J = 14.5 Hz, 3F)



Figure S2. ¹⁹F NMR analysis of the crude product.

N-(4-(1,1,1,4,4,4-hexafluoro-3-methylbuten-2-yl)phenyl)-*N*-methylacetamide (5i)¹⁵



The reaction was carried out on according to general procedure. The target compound was obtained after silica gel column chromatography (AcOEt/hexane = 60/40).

E-isomer: white solid, 19 mg, 29% yield

¹⁴ The stereochemistry was analogously determined by comparing the chemical shifts and coupling patterns of the ¹⁹F NMR spectra of **5h** and **5i**; a unique F–F coupling was observed only in the case of the Z-isomer.

¹⁵ The stereochemistry was determined by NOESY experiments; only the *Z*-isomer showed a correlation between the aromatic proton and methyl proton on the double bond.

 1 H NMR (400 MHz, CDCl₃)

1.88 (s, 3H), 2.22 (q, J = 2.4 Hz, 3H), 3.29 (s, 3H), 7.19–7.25 (m, 4H)

 13 C NMR (100 MHz, CDCl₃)

14.5, 22.5, 37.2, 122.7 (q, *J* = 277 Hz), 122.9 (q, *J* = 277 Hz), 127.0 (2C), 130.5 (2C), 131.8, 135.3

(m), 135.5 (m), 145.2, 170.6

¹⁹F NMR (376 MHz, CDCl₃)

-61.2 (s, 3F), -58.6 (s, 3F)

IR (neat, cm^{-1})

1348, 1230, 1214, 1189, 1139, 1114, 1013, 941, 897, 832, 751, 723, 698

HRMS-ESI (m/z)

[M+Na]⁺ calcd. for C₁₄H₁₃F₆NO, 348.0794; found: 348.0791

Z-isomer: colorless oil, 8 mg, 12% yield

 1 H NMR (400 MHz, CDCl₃)

1.79 (q, *J* = 2.1 Hz, 3H), 1.92 (bs, 3H), 3.31 (s, 3H), 7.25 (d, *J* = 8.4 Hz, 2H), 7.29 (d, *J* = 8.4 Hz, 2H) 2H)

¹³C NMR (100 MHz, CDCl₃)

17.3, 22.6, 37.3, 121.3 (q, *J* = 275 Hz), 122.2 (q, *J* = 276 Hz), 127.8 (2C), 130.3 (2C), 132.9, 135.0

(m), 135.3 (m), 145.4, 170.4

¹⁹F NMR (376 MHz, CDCl₃)

-61.6 (q, J = 14.2 Hz, 3F), -58.0 (qd, J = 14.2, 2.1 Hz, 3F)

IR (neat, cm^{-1})

```
1662, 1606, 1511, 1381, 1321, 1292, 1227, 1143, 1104, 1086, 1020, 932, 712, 672
```

HRMS-ESI (m/z)

 $[M+Na]^+$ calcd. for $C_{14}H_{13}F_6NO$, 348.0794; found: 348.0794

(E)-N-methyl-N-(4-(1,1,1,2,2,3,3,6,6,7,7,8,8,8-tetradecafluoro-5-methylocten-4-

yl)phenyl)acetamide (5j)¹⁶



The reaction was carried out on according to general procedure. The target compound was obtained as a white solid (68 mg, 65% yield) after silica gel ³ column chromatography (AcOEt/hexane = 60/40).

¹H NMR (400 MHz, CDCl₃)

1.83 (s, 3H), 2.19 (s, 3H), 3.26 (s, 3H), 7.11–7.21 (m, 4H)

¹⁶ The stereochemistry was analogously determined by comparing the chemical shifts and coupling patterns of the ¹⁹F NMR spectra of **5j** and the CF₃ analog (**5i**); a unique F–F coupling was observed only in the case of the Z-isomer.

¹³C NMR (100 MHz, CDCl₃)¹³

15.7, 22.3, 37.1, 126.2 (2C), 131.3 (2C), 131.5, 136.5 (t, *J* = 24 Hz), 137.3 (t, *J* = 20 Hz), 145.1, 170.5

¹⁹F NMR (376 MHz, CDCl₃)

-122.9 (s, 2F), -122.7 (s, 2F), -105.5 (q, *J* = 10.1 Hz, 2F), -102.9 (q, *J* = 10.1 Hz, 2F), -80.8 (t, *J* = 10.1 Hz, 3F), -80.4 (t, *J* = 10.1 Hz, 3F)

IR (neat, cm^{-1})

CF₃CF₂CF₂

1167, 1508, 1343, 1224, 1184, 1151, 1112, 1030, 955, 906, 865, 843, 749, 730, 716, 701, 668, 648 HRMS-ESI (*m*/*z*)

[M+Na]⁺ calcd. for C₁₈H₁₃F₁₄NO, 548.0666; found: 548.0671

(1,1,1,2,2,3,3,6,6,7,7,8,8,8-tetradecafluorooct-4-en-4,5-diyl)dibenzene (5k)¹⁷

The reaction was carried out on according to general procedure. The target compound was obtained after silica gel column chromatography (hexane). $CF_2CF_2CF_3$

```
E-isomer: white solid, 44 mg, 43%
```

¹H NMR (400 MHz, CDCl₃)

7.31–7.44 (m, 10H)

¹³C NMR (100 MHz, CDCl₃)¹³

127.6 (4C), 129.2 (2C), 129.9 (4C), 131.1 (2C), 139.4 (t, *J* = 26.6 Hz, 2C)

¹⁹F NMR (376 MHz, CDCl₃)

-120.2 (s, 4F), -102.0 (q, J = 10.1 Hz, 4F), -80.7 (t, J = 10.1 Hz, 6F)

IR (neat, cm^{-1})

```
1342, 1220, 1188, 1147, 1112, 960, 720, 702
```

HRMS-EI (m/z)

[M] calcd. for C₂₀H₁₀F₁₄, 516.0559; found, 516.0549

Z-isomer: colorless oil, 17 mg, 16%

 1 H NMR (400 MHz, CDCl₃)

6.97–7.02 (m, 4H), 7.06–7.11 (m, 6H)

¹³C NMR (100 MHz, CDCl₃)¹³

127.7 (4C), 128.4 (2C), 130.5 (4C), 133.5 (2C), 140.8 (q, *J* = 13.5 Hz, 2C)

¹⁹F NMR (376 MHz, CDCl₃)

-119.7 (m, 4F), -102.6 (m, 4F), -80.6 (m, 6F)

¹⁷ The stereochemistry was determined by the chemical shifts and unique coupling patterns in the ¹⁹F NMR spectra, which depend on the geometrical configuration of the two perfluoroalkyl groups on the double bond.

IR (neat, cm⁻¹)

1668, 1608, 1511, 1378, 1260, 1226, 1180, 1138, 1112, 1030, 927, 659

HRMS-EI (m/z)

[M] calcd. for $C_{20}H_{10}F_{14},\,516.0559;\,found,\,516.0548$

4. ¹H and ¹³C NMR spectra









S25







S28







































S40







S43

Computational details:

DFT calculations were conducted using Gaussian 16 software. All structures except Cu (optimised by SDD) were optimised using UM06/6-311G+(d,p). The CPCM solvation model (dichloromethane) was used to determine the solvent effect. The free energies described in this work were obtained using vibrational analysis at the UM06 level of theory. No imaginary frequencies for were observed for the intermediates, but one imaginary frequency was noted for the transition state. The reaction pathway from the transition state was confirmed by IRC calculations and the vibration mode of the imaginary frequency.

Cartesian coordinates and energies

• CF₃

E(UM06) = -337.542753086						
Char	ge = 0, Multiplicit	y = 2				
С	0. 0.0000002727 0.3257056461					
F	0.0000000444	1.248725998	-0.0719813204			
F	1.0814281782	-0.6243626284	-0.0719813204			
F	-1.0814282226	-0.6243625514	-0.0719813204			
Dh	<u> </u>					
Pfi	п					
E(U	M06) = -308.22657	76804				
Char	ge = 0, Multiplicit	y = 1				
С	2.0134156043	-0.0000142392	0.0000224296			
С	3.2169221889	0.000030717	0.0000042684			
С	0.5873826611	-0.0000095105	0.0000111601			
Н	4.2821022694	0.000040596	-0.0000206594			
С	-0.1196539442	1.2071833155	0.0000053432			
С	-0.1196618307	-1.2071978444	0.0000073107			
С	-1.5054042703	1.2031871605	-0.0000037407			
Н	0.4292280914	2.1438315325	0.0000083879			
С	-1.5054120688	-1.2031927066	-0.000001781			
Н	0.4292142431	-2.1438495545	0.0000118725			
С	-2.2010195002	-0.0000004763	-0.0000074217			
Н	-2.0459252456	2.1444030741	-0.000008012			
Н	-2.0459392765	-2.144405046	-0.0000045198			
Н	-3.286460922	0.0000029819	-0.0000146379			

E(UM06) = -645.770641158

Charge = 0, Multiplicity = 2

С	-0.0620286276	0.000000012	-1.260332804
С	-1.1925555104	0.000000232	-1.7060180715
С	1.1975132339	-0.000000233	-0.6124802549
Н	-2.043625539	0.000000397	-2.3511325314
С	-2.6008142495	0.0000000505	0.2308469713
С	1.8262935401	1.210004589	-0.2872709063
С	1.8262934931	-1.2100046599	-0.2872709063
F	-3.358859682	-1.0807914593	0.2370139763
F	-1.8056724799	0.000000351	1.280230503
F	-3.35885964	1.0807915897	0.2370139763
С	3.0552923021	1.2043789834	0.3501398455
Н	1.3386671266	2.1462080143	-0.540367371
С	3.0552922553	-1.2043791021	0.3501398455
Н	1.3386670432	-2.1462080663	-0.540367371
С	3.6722701433	-0.0000000713	0.6702096359

Н	3.5357962731	2.1448673878	0.5999761158
Н	3.5357961898	-2.1448675251	0.5999761158
Н	4.6357027244	-0.00000009	1.1700246048

E(UM06) = -645.840068489

Ciia	Charge = 0, Multiplicity = 2				
С	0.2311991222	0.0000027677	-0.6173048082		
С	1.4833792429	-0.0000051005	-0.9504055841		
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С	1.8652228733	-2.4999933352	-0.6036618524		
Н	1.0145320059	-3.1751503634	-0.6530348906		
С	3.1436758448	-2.9390362556	-0.9033730214		
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Ν	0.9010614657	1.2698596554	0.2360199585
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С	3.1457546772	1.9870797636	0.5988039987
Н	4.2063147123	1.7656172656	0.6124199577
С	2.6940333026	3.2439504937	0.9655705249
Н	3.4012162336	4.0137307024	1.2550209053
С	1.3314219355	3.5000849405	0.9659841684
Н	0.9369179799	4.4693028171	1.247238414
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Н	-0.6047994991	2.6304796741	0.5831096207
Ν	-1.6526826818	-1.0851914679	-1.4011297377
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С	-4.3252490282	-0.4432878537	2.5029709923
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Ν	1.1566971056	-0.0583945948	1.3142382537
С	2.2426630317	-0.7354060649	0.9040909193
Ν	0.8626501839	-1.4900078645	-0.8998854115
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Н	-0.3611501195	-2.0980352932	-2.439807503
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Cu

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F	1.8084918903	0.7995144129	-2.0682402348
F	-0.1006621658	1.2088491582	-2.9419330794
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С	2.5150529761	0.0002376911	0.0911367908	
С	-1.8162782743	-1.2436856179	-0.1688036371	
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F	3.2803850682	1.0751091775	-0.0529805293	
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Н	-1.2675124757	-2.1695165496	-0.2991455851	
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Н	-1.2679119628	2.1695761893	-0.2982394958	
С	-3.8041241924	-0.0003288434	0.2981367499	
Н	-3.6945292111	-2.1536607626	0.272307269	
Н	-3.6949230471	2.1530338081	0.2732172504	
Н	-4.8613488251	-0.000477649	0.5450893458	



 $\begin{array}{ll} E(UM06) = -1525.34129933\\ Charge = 0, Multiplicity = 1\\ Cu & 0.3585084256 & -1.7334931769 & -1.7818190849\\ C & 0.9460555632 & -2.3836115213 & 1.2008715592\\ N & 1.4907605262 & -2.3783710887 & -0.0246187079\\ N & 0.2200226309 & -3.5705274571 & -2.7406303166 \end{array}$

Ν	-1.2655650536	-2.5420490856	0.2879848752
С	-0.5055313965	-2.1003198626	1.2945161903
С	1.0332311393	-3.8255082782	-3.779221749
Ν	2.1240601135	-1.7618011016	-3.2820105925
С	1.7023249381	-2.6545210719	2.3386613682
Н	1.2294072513	-2.6728005311	3.3147095015
С	1.0435493317	-5.0673338083	-4.4085993967
Н	1.7268198697	-5.2632514254	-5.2271395654
С	1.9260438483	-2.7207388903	-4.1951644195
С	-2.5682572918	-2.2807710328	0.324843765
Н	-3.1579410342	-2.6556762651	-0.5115270589
С	-1.0301483425	-1.3889214896	2.370645459
Н	-0.3810689633	-1.0251624818	3.1603467729
С	3.0521857659	-2.9310265859	2.203091622
Н	3.6564700955	-3.1541908557	3.0763288569
С	2.79072755	-2.6482630298	-0.1449405598
Н	3.1829162199	-2.6299395528	-1.1600987535
С	-2.3887842595	-1.1195485726	2.393555759
Н	-2.8219628426	-0.5544500392	3.2125657908
С	2.9164420833	-0.7379222368	-3.5868620986
Н	3.0461602577	0.0143118151	-2.8105748585
С	3.6120492496	-2.9304984262	0.9346916937
Н	4.663089893	-3.1455304105	0.7788781871
С	-0.593796919	-4.5352937986	-2.3065687301
Н	-1.2121343488	-4.2712324535	-1.4511418217
С	2.5125104105	-2.666323529	-5.4574404932
Н	2.3146328635	-3.4345101431	-6.1965877462
С	-0.647933848	-5.7911474961	-2.8869421187
Н	-1.3295144775	-6.539789998	-2.4997061044
С	-3.1801609153	-1.5727899879	1.3493754209
Н	-4.2475418453	-1.3835286198	1.3244935649
С	0.189625945	-6.0597919896	-3.9594455189
Н	0.1866869894	-7.0349927078	-4.4349577675
С	3.5516123579	-0.6117796525	-4.8125618938
Н	4.191852249	0.2396804759	-5.0133166924
С	3.3363186404	-1.5965401408	-5.765751642
Н	3.8000721874	-1.529198655	-6.7443730904
С	-0.4847215614	0.031979247	-1.6988542429
F	-0.5040040249	0.6088408352	-0.4346550277
F	-1.8181223968	0.1265635688	-2.0762822256
F	0.1028521254	1.0318255075	-2.463386109

[Cu(CH₃CN)₄]+

E(UM06) = -728.119844908 Charge = 1, Multiplicity = 1

		- J		
Cu	0.0070924327	0.0071104129	0.0004895804	
Ν	-0.3513199901	1.6532869359	-1.1075728905	
Ν	-1.6111974795	-0.3924776437	1.1367466687	
Ν	0.3696449849	-1.5701404961	-1.2033202233	
Ν	1.6132748586	0.3305285453	1.1752687038	
С	0.5707981159	-2.4792441236	-1.8768785445	
С	2.5371712925	0.5124097025	1.8340020651	
С	-2.5421963084	-0.6228667938	1.7698136006	
С	-0.5566285838	2.5992104901	-1.7269871231	
С	0.821736889	-3.6201645632	-2.7195705563	
Н	-0.0459288816	-3.8129569276	-3.3535888029	
Н	1.6901016886	-3.4320216859	-3.3539290019	
Н	1.0152799759	-4.5015852688	-2.1050031749	
С	3.6963627507	0.7395659237	2.6582313859	
Н	3.5286003791	1.5997542826	3.3092417256	
Н	3.8928302184	-0.1395477124	3.2751245426	
Н	4.5677226504	0.9343389517	2.0298603846	
С	-0.8137414792	3.7859079264	-2.5017849573	
Н	0.0774472979	4.0607781469	-3.0694085243	
Н	-1.6361880871	3.6067136623	-3.1970191914	

Н	-1.0817860852	4.6117987633	-1.8399437095
С	-3.7103284838	-0.9116990833	2.5615348566
Н	-3.5335477116	-1.7945351414	3.1789954734
Н	-3.9394120901	-0.0644990735	3.2109183972
Н	-4.5650573542	-1.10019323	1.9088893153

E(UM06) = -495.131539831 Charge = 0, Multiplicity = 1

2.6858923121

3.1554286258

3.4727572488

4.554332092

2.8339927848

3.4052185482

1.4491406126

0.920885447

0.7431617488

-1.3569007778

-0.7431617488

-1.4491406126

-0.920885447

-2.8339927848

-3.4052185482

-3.4727572488

-4.554332092

-2.6858923121

-3.1554286258

Ν С

Н С

Н

С

Н

С

Н

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С

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С

Н

С

Н

С Н 1.3569007778 -1.1351053668

-1.1499406138

-2.0824350718

-0.0620078331

-0.1353161721

1.1054052478

1.9792539601

1.1374236535

2.0302577434

-0.0054535471

-1.1351053668

-0.0054535471

1.1374236535

2.0302577434

1.1054052478

1.9792539601

-0.0620078331

-0.1353161721

-1.1499406138

-2.0824350718

0.365446587

0.3746593355

0.6841815834

0.0208838166

0.0459413946

-0.3652720096

-0.662049971

-0.3754703197

-0.6934030277

-0.0002161457

-0.365446587

0.0002161457

0.3754703197

0.3652720096

0.662049971

-0.0208838166

-0.0459413946

-0.3746593355

-0.6841815834

0.6934030277

Cu	0.3905512822	2 -2.4017181884	-1.4653360984
С	0.9751230936	-2.0495117784	1.3796175455
Ν	1.6003641048	-2.1631104541	0.1958666632
Ν	0.2845374177	-4.0168567819	-2.6722517713
Ν	-1.0546707238	-2.246174812	0.1314204484
С	-0.5071261298	-2.0099217383	1.3309882397
С	0.8948828684	-4.0381947592	-3.8678368465
Ν	2.457889558	-2.3356976318	-3.2063504251
С	1.6912738672	-1.9854260049	2.5712818285
Н	1.1789945377	-1.9101781591	3.5231251499
С	0.7186115694	-5.0934089906	-4.7574297417
Н	1.2432230014	-5.0944388306	-5.7067335384
С	1.754732069	-2.8786649515	-4.2058795567
С	-2.3784839936	-2.2153773729	-0.0027778616
Н	-2.7532681472	-2.3888454277	-1.0102175443
С	-1.2943879906	-1.7342127104	2.4455774108
Н	-0.8459461341	-1.5257187554	3.4098763044
С	3.0746633362	-2.0320939501	2.533140613
Н	3.6472319207	-1.9832084327	3.4533218418
С	2.9317136377	-2.2191618393	0.1661075068
Н	3.3732914243	-2.3348739599	-0.8213734646
С	-2.6725361182	-1.713940584	2.305117972
Н	-3.3031083949	-1.5001618882	3.1618090388
С	3.195535212	-1.2640979049	-3.4755968557
Н	3.7576565587	-0.8406093278	-2.6438018061
С	3.7125146097	-2.151429294	1.3078254425
Н	4.7928236434	-2.1973205657	1.233211937
С	-0.5185254201	-5.0328072737	-2.3434404522
Н	-1.0004707972	-4.9627692638	-1.3710383511
С	1.7784926522	-2.3717392007	-5.5016088803
Н	1.1771274585	-2.8268312035	-6.2816342893
С	-0.7460335082	-6.1156459297	-3.1744758195
Н	-1.4070141998	-6.9138282326	-2.8574729761
С	-3.2313751105	-1.9595829261	1.0601922262
Н	-4.3047886617	-1.9474521386	0.9095574051
С	-0.1100208965	-6.145557173	-4.4060975404
Н	-0.25337665	-6.9804226201	-5.0839420841
С	3.27492474	-0.6800777572	-4.7330273456
Н	3.8884485615	0.1999438536	-4.8903844698
С	2.5515103249	-1.2519442342	-5.7662981592
Н	2.5752520235	-0.8251726373	-6.7637452939
0	-0.1306491317	-0.6924856035	-2.6128189668
С	-1.2225934298	-0.7268997454	-3.2213161176
0	-2.1129032059	-1.5734352839	-3.2355683555
С	-1.430583997	0.5311929596	-4.1143186775
F	-2.6448795965	0.60016571	-4.6619872819
F	-0.5452308854	0.5438440623	-5.1257485041
F	-1.2492303503	1.6671357316	-3.4254044983

~

E(UM06) = -132.701263999 Charge = 0, Multiplicity = 1 -0.0000356171 0.0000140303 0.2735580482 С 0.0000298536 -0.0000563631 1.4258254902 Ν С 0.0000019173 0.0000375147 -1.1709132004 -1.5467072212 1.0250432957 Н -0.0000067133 0.8877350721 -0.512521884 -1.5465693812 Н Н -0.8877245127 -0.5125155937 -1.5465957356

CF₃CO₂⁻⁻

E(UM06) = -526.311698489

Charge = -	l, Mı	ultipli	icity = 1	1
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0	0.0440918997	-0.8800585852	0.3468334588
С	0.0651959577	-0.0632447926	1.2776931835
0	0.992269338	0.5670056663	1.7970703316
С	-1.3535544425	0.1904015969	1.8855266691
F	-1.880230436	-0.9444804427	2.3818671432
F	-2.2153747777	0.6341525531	0.9525388145
F	-1.3770045392	1.0850700043	2.8790413992



E(UM06) = -1713.92363168 Charge = 0, Multiplicity = 1

Cu^I O₂CCF₃

E(UM06) = -1218.76866197 Charge = 0. Multiplicity = 1

- Cincing	se o, maniphen.	, .	
Cu	-0.0681515222	-0.5703197628	0.0114103314
Ν	1.7900332103	-1.3016231534	0.0170557955
С	2.0913116824	-2.602012929	0.0123523488
Н	1.2520114655	-3.2902831351	0.0289937524
С	3.3943156328	-3.0680550662	-0.0117441702
Н	3.5898711762	-4.1335666761	-0.0142205529

0100000111102	111000000000	010112200020
4.4226038864	-2.1400949404	-0.0337003742
5.4582003693	-2.4619327562	-0.0555624536
4.115207689	-0.7897535938	-0.028657397
4.9107727345	-0.0550633463	-0.0493818559

С

Н

С

Н

С	2.7831329112	-0.3944294156	-0.0011571776
Ν	1.0440026848	1.2456935602	-0.020839181
С	2.3658846428	1.0299766787	0.0063823641
С	3.2705197043	2.0845872632	0.0416012812
Н	4.3388423171	1.907919905	0.0702083695
С	2.7847546084	3.3831233344	0.0442563692
Н	3.4742476248	4.2202287014	0.0719179837
С	1.4165208588	3.5967110142	0.0124549916
Н	0.9999039989	4.5971769001	0.0126692146
С	0.5780446756	2.4917597277	-0.0183276543
Н	-0.5054531749	2.5860397224	-0.0434751154
0	-1.9637068906	-0.8465116856	0.0337171176
С	-2.7300378107	0.1559951494	-0.0372923548
0	-2.4725076595	1.3446335595	-0.1273303282
С	-4.2197498873	-0.2739422045	0.0083241843
F	-4.5076719923	-1.12596042 -0	0.9837252924
F	-5.0558249289	0.7548997835	-0.098349404
F	-4.5013690065	-0.8982632147	1.1599652076

[Cu(CH₃CN)₃]+

E(UM06) = -932.970559034

Charge = 1, Multiplicity = 2

Cu	-0.6185107628	-0.1178388448	-0.2242299991
Ν	-0.5895487117	1.6376264193	-1.1686840563
Ν	-2.105308033	-0.4537645496	1.0596391491
Ν	0.1143903763	-1.6920420224	-1.2039489496
С	0.614947947	-2.5923009952	-1.7102587277
С	-2.9007178324	-0.641474139	1.8654963853
С	-0.5126076831	2.6651763453	-1.6741436329
С	1.2398347129	-3.7197068805	-2.345032645
Н	0.6130529528	-4.0767698349	-3.1644385904
Н	2.2144340361	-3.4270215362	-2.7404817954
Н	1.373919415	-4.5241092837	-1.6192737504
С	-0.4199838023	3.9492889472	-2.3120236108
Н	0.5436363477	4.042030535	-2.8166803204
Н	-1.2211347507	4.0543238006	-3.0463062614
Н	-0.5119708626	4.7405996041	-1.5656212678
С	-3.8994021675	-0.877639233	2.8712616002
Н	-3.5788227834	-1.6907092455	3.5256358234
Н	-4.045716219	0.0258055637	3.4663741548
Н	-4.8436143223	-1.1510179259	2.3962141764
С	0.877246166	0.184236626	1.07412064
F	1.0607077008	-0.8857231108	1.8527087581
F	0.6186527582	1.2285850373	1.8659009744
F	2.0245910879	0.4279899731	0.4345511255

CF₃CO₂ | |NCHCH₃ H₃CCN Cu^{-Cu}NCCH₃

E(U	E(UM06) = -1121.742477030				
Cha	Charge = 0, Multiplicity = 1				
Cu	-0.4322072821	-0.0941508838	-0.0469937392		
Ν	-0.5554623788	1.6010247818	-1.0930967674		
Ν	-2.0066978364	-0.425436495	1.1430823399		
Ν	0.1811013743	-1.6768318869	-1.0920046854		
С	0.6530063187	-2.5632578576	-1.6511525224		
С	-2.8701847374	-0.5883889179	1.8840168611		
С	-0.5218084089	2.6022450833	-1.656572516		
С	1.2480034874	-3.673093221	-2.3508983848		
Н	0.8717527362	-3.7145850575	-3.3749481472		
Н	2.333359872	-3.5566206652	-2.3766388599		

1.0038839277	-4.6090863703	-1.8445599438
-0.4776884015	3.8572904234	-2.362530069
0.53501641	4.0446073447	-2.7251397894
-1.1605114871	3.8317038302	-3.2140391013
-0.7714922291	4.6709443341	-1.696456407
-3.9512925403	-0.7895761203	2.8149762111
-3.712466629	-1.6116148407	3.4926020819
-4.1116760827	0.1174236036	3.40121076
-4.8695014447	-1.0301135136	2.2755154114
1.0780271311	0.2362130035	1.412022509
2.2422238548	0.5024890655	1.0372252286
2.721764703	0.618538677	-0.0854675328
3.2016846687	0.7094050158	2.245676349
3.2655261031	-0.3948852767	3.0068336783
2.7754724296	1.7043008379	3.0398455853
4.4515964414	1.0068161058	1.8886834502
	$\begin{array}{c} 1.0038839277\\ -0.4776884015\\ 0.53501641\\ -1.1605114871\\ -0.7714922291\\ -3.9512925403\\ -3.712466629\\ -4.1116760827\\ -4.8695014447\\ 1.0780271311\\ 2.2422238548\\ 2.721764703\\ 3.2016846687\\ 3.2655261031\\ 2.7754724296\\ 4.4515964414\end{array}$	1.0038839277 -4.6090863703 -0.4776884015 3.8572904234 0.53501641 4.0446073447 -1.1605114871 3.8317038302 -0.7714922291 4.6709443341 -3.9512925403 -0.7895761203 -3.712466629 -1.6116148407 -4.1116760827 0.1174236036 -4.8695014447 -1.0301135136 1.0780271311 0.2362130035 2.242238548 0.5024890655 2.721764703 0.618538677 3.2016846687 0.7094050158 3.2655261031 -0.3948852767 2.7754724296 1.7043008379 4.4515964414 1.0068161058

E(U	M06) = -989.03090	08474	
Cha	rge = 0, Multiplicit	y = 1	
Cu	1.2972584509	-0.1240429484	-0.1648940247
0	-0.7241175355	0.0489364396	-0.803363273
С	-1.4022482137	0.1316251428	0.2465253728
0	-1.0336210999	0.1103759825	1.4179195943
С	-2.9211495751	0.2797246781	-0.0350850695
F	-3.1690345678	1.3695730676	-0.7756210514
F	-3.6556315232	0.3824907203	1.0710402454
F	-3.3863253572	-0.7744344735	-0.7212801504
С	1.8678233918	-3.1077286912	0.0300864556
С	2.3108590942	2.739902174	0.0539273405
С	2.0822453481	-4.5272733342	0.1189950328
Н	1.3102870149	-4.9799289302	0.7445195011
Н	2.0371556707	-4.9709032073	-0.8776836177
Н	3.0616252684	-4.730593662	0.5564662115
С	2.7340605396	4.111054274	0.1540818011
Н	3.0729599754	4.4652301995	-0.8215862911
Н	1.9017402756	4.733374549	0.4888080294
Н	3.5546439611	4.196671596	0.8690899448
Ν	1.6971844972	-1.9742235045	-0.0412405787
Ν	1.9745123844	1.6447089278	-0.0264544727

CF₃CO₂ Cu H₃CCN

CF₃CO₂

H₃CCN

Cu^{___}NCCH₃

E(UM06) = -856.319746698 Charge = 0, Multiplicity = 1 0.8662935346 0.0547639739 0.7007321155 Cu 0 -0.9558370405 -0.2978775402 0.7841010548 С -1.6516881161 -0.1787240381 -0.2735701818 0 -1.315684868 0.1482185568 -1.3937825375 С -3.1370386781 -0.5237139699 -0.0004477922 F -3.2621481597 -1.7834912567 0.4329803631 F -3.8933818006 -0.3973228565 -1.0848500457 F -3.6489417046 0.9427635282 0.2749351166 С 0.6902664405 3.7968678648 0.5828069433 С 5.2103418134 0.8362747602 0.6751532578 Н 5.4567178429 1.4141732082 1.6012148391

Н	5.7525474677	-0.08088184	0.9136659492
Н	5.5119378418	1.1840434381	-0.3149693201
Ν	2.6676440024	0.3797938734	0.70122296

0.1933526124

[Cu(CH₃CN)₄(CF₃)]⁺

E(UM06) = -1065.69005012 Charge = 1. Multiplicity = 2

Chai	charge = 1, Multiplicity = 2				
Cu	0.0028300682	-0.0808633392	0.193352612		
С	0.0093939624	-2.0315136453	-0.2497427965		
F	0.0112289022	-2.266376234	-1.5810243555		
Б	1 0677210722	2 6050020662	0.2401000077		

F	-1.0677319732	-2.6850820663	0.2401009077
F	1.0898334194	-2.6782023773	0.2418500699
Ν	-1.8093246986	0.1467229367	-0.816570255
Ν	-0.0037965633	1.9429903322	0.6504737566
Ν	0.0014011335	-0.5341387107	2.2320588768
Ν	1.8151894038	0.1586696093	-0.8131701809
С	-0.0061814662	3.0618626588	0.9044668248
С	-2.8018899702	0.1738351001	-1.3926474357
С	-0.0036383766	-0.8786820396	3.3270809054
С	2.8087511134	0.1924457746	-1.3871720335
С	-4.0457830384	0.2115212302	-2.1148605876
Н	-4.7803812157	0.7970199068	-1.5587892867
Н	-3.8912750785	0.6688938583	-3.0939682294
Н	-4.4255978869	-0.802894405	-2.2507415744
С	-0.0088466855	4.4650352239	1.2229797836
Н	0.9362821112	4.7395699646	1.6952629031
Н	-0.136215517	5.051730846	0.3112906137
Н	-0.8286531169	4.6879935589	1.908448579
С	-0.0091771989	-1.3068873358	4.7007393255
Н	0.994711277	-1.2176465192	5.1202215411
Н	-0.6952535014	-0.6862493076	5.280150899
Н	-0.3303512125	-2.3484265386	4.7621689659
С	4.0537863775	0.2387622906	-2.1068984911
Н	3.8939028725	0.6751322582	-3.0946881762
Н	4.7763601522	0.847411515	-1.5600342572
Н	4.4518697066	-0.7710655457	-2.2230399047

[Cu(CH₃CN)₃(CF₃)]+

E(UM06) = -932.970304439

Charge = 1, Multiplicity = 2			
Cu	-0.0778857376	0.1230866127	0.3222159642
Ν	-0.6630402496	2.0336902856	0.2538418166
Ν	-1.6861168669	-0.5571284694	1.3459419419
Ν	0.1171082236	-1.46987546	-0.8697784873
С	0.2867037811	-2.359143754	-1.5744403892
С	-2.582314424	-0.9349599321	1.9547730701
С	-0.939440095	3.1459926908	0.203649132
С	0.4911009036	-3.4718996274	-2.4598676983
Н	-0.2318851926	-3.4293611584	-3.2768084169
Н	1.501290417	-3.4350143731	-2.8720850338
Н	0.3599743874	-4.4076915004	-1.9131442282
С	-1.292027865	4.5371520788	0.1390705205
Н	-0.4054718903	5.1322535899	-0.0881400916
Н	-2.038492144	4.6917809819	-0.6424685101
Н	-1.7043674187	4.857802809	1.0977068977
С	-3.7050272714	-1.4063307582	2.7178700092
Н	-3.4912514244	-2.4006574203	3.1145049551
Н	-3.9008731218	-0.7234523477	3.5467862478
Н	-4.5889722437	-1.4577690629	2.0791288954
С	1.8444520888	0.6083695895	0.1436363684
F	2.6642020211	-0.436540151	0.3474409343
F	2.2193622736	1.5669402191	1.0075390786
F	2.1081192689	1.0735767575	-1.0903529666

E(UM06) = -1459.31846019			
Cha	rge = 0, Multiplici	ty = 2	
Cu	0.4965864916	-1.4891009974	-1.7280225349
С	-0.8714578543	-0.6588347703	-2.8976494865
F	-0.9966327354	0.6735807072	-2.6815462771
F	-2.1177483661	-1.1721344122	-2.7237591338
F	-0.6153977917	-0.7940530036	-4.218078074
0	1.9770660774	-0.5318809616	-2.807106477
С	2.7985566716	0.3698157127	-2.5103671571
0	3.7853589319	0.747355898	-3.1269646728
С	2.5697221239	1.1067896907	-1.1571648382
F	1.3633436407	0.8919103277	-0.6192984258
F	3.4772177033	0.6943069913	-0.2547387315
F	2.7124968781	2.4259789807	-1.2828301297
С	-0.0333984563	-4.3847456348	-3.1929614691
С	2.8932865036	-2.4567445096	0.1610085797
С	-1.502167164	-0.9915338834	0.7820222772
С	-0.3011438243	-5.589204571	-3.9350812019
Н	0.5131435425	-6.3022310307	-3.7932895184
Н	-1.2336453669	-6.0391822563	-3.5893591686
Н	-0.3907358879	-5.3569168469	-4.9979596257
С	4.0390124948	-2.766923954	0.9740089256
Н	4.626787841	-1.8610518422	1.1378380267
Н	3.7161622744	-3.1634725313	1.938362332
Н	4.6613345801	-3.5093197457	0.4709130047
С	-2.4280634539	-0.6602863795	1.8332793567
Н	-2.531013889	-1.5037145404	2.5185594517
Н	-2.0640407675	0.2071511167	2.3871672107
Н	-3.4047832426	-0.4257722491	1.4057327645
Ν	0.182133485	-3.42521677 -2	.5996788022
Ν	1.9819957883	-2.2039647081	-0.4886508718
Ν	-0.7629172283	-1.2557558268	-0.0556213332

,CF₃

0₂CCF

(CH₃CN)₃Cu^{II}

CF₃

(CH₃CN)₂Cu^{II}

O₂CCF₃

E(UM06) = -1326.60704759 Charge = 0, Multiplicity = 20.2556429261 -0.4107601346 Cu -0.0018742164 С -0.1835185556 -2.3256247794 0.3430796081 F 0.2297338256 -3.1139130764 -0.6814312148 F 0.4506099571 -2.8049579088 1.4432338068 F -1.4860368835 -2.6179419847 0.5223654752 0 -1.5439289032 -0.0218714869 -0.7405068159 0 -2.1825789597 0.4642867232 1.3560553788 0.3293006679 С -2.3621182736 0.1586513915 С -3.7703325926 0.5998588515 -0.4284448458F -0.5032806102 -1.0024353963 -4.2630537723 F -3.7238862394 1.5519028701 -1.3681134826 F -4 6404978231 0.9932085762 0 4951905544 С 0.7648860108 2.7264352936 0.1594701743 0.0448272613 С 3.3240954333 -1.1758568318 С 0.9118908677 4.1541397147 0.2484744412 Н -0.07194729324.6273182126 0.240397289 Н 1.4278831467 4.415871983 1.1741786796 Н 1.4924074679 4.5200479044 -0.6005606361 4.6980344705 С -1.5970422274 0.0760395634 Н 5.131135073 -1.5147569032 -0.9227068224 Н 5.2628344941 -0.9659708056 0.7647377314

Н	4.75812499	-2.6351617395	0.4083649398
Ν	0.6461781961	1.5877472954	0.0877649872
Ν	2.2281794372	-0.837937604	0.0177721485

Cu 0.1962892064 -0.8643302502 -0.3625014321 C -0.307554591 -2.6173209809 0.495076227 0.3524360509 -3.6345429368 -0.0815218485 0.0421384379 -2.6055078313 1.7896429451

-0.1117780991

0.1981866379

0.2470512924

0.8090244215

-4.1970167805 -0.1064153953 -1.171454879

 $-3.3450942153 \qquad 1.8731973119 \qquad -1.2144200562$

1.1827114021

-1.1800760038

-1.3712277002

-1.3070389362

-0.5998337166

-2.353939758

-1.0234828124

0.4328879982

-0.8471037514

1.3039506678

0.1084198575

-0.4344839801

0.5401850195

-0.1403058415

-0.014603219

-0.9992670862

0.6347048966

0.4166919158

-0.2438394336

CF₃

`0₂C**CF**

-1.6094249969 -2.9035826451

E(UM06) = -1193.88639055 Charge = 0, Multiplicity = 2

-1.4589526442

-2.0321375402

-2.2251731238

-3.5615961583

-4.3802968666

3.2213602585

4.6375745569

5.1050471934

5.0561751442

4.8385285217

2.0899935469

(CH₃CN)Cu^{II}

F F F

0

0

С

С

F

F

F

С

С

Н

Н

Н

Ν

С	1.9740313467	3.0956017545	2.6319061519
С	2.7496605161	2.3498821996	1.7624998195
Н	-0.9430353014	1.4938403139	2.053174646
Н	-0.0198287391	3.3582378204	3.4241480409
Н	2.4191700073	3.9033614506	3.202416046
Н	3.8009823697	2.5772885884	1.6344823393
С	2.9491893957	0.4974445675	0.0828680583
С	4.32562458	0.3456712831	0.2065347747
С	4.9925938714	-0.4488430683	-0.7122289969
Н	4.8679735712	0.8165881702	1.0180561824
С	2.8995588997	-0.8651568121	-1.7646551004
С	4.2716977915	-1.0615934041	-1.7239159177
Н	6.0649147249	-0.5910203543	-0.6323271139
Н	2.2972727573	-1.3341648045	-2.5392320572
Н	4.7553094537	-1.686868276	-2.4650928606
Ν	0.8536982807	1.0546074473	1.1543465913
Ν	2.249292254	-0.1108121597	-0.8829043912
С	-0.4284422759	-1.7665201138	-1.2892411147
F	0.3592718292	-2.8415109184	-1.2594719114
F	-1.6897838617	-2.1757731147	-1.1740560609
F	-0.2859359125	-1.2426376397	-2.5145010075

CN_	F ₃ CF ₃ ⊕
	N

E(UM06) = -1862.77134244

	· · ·		
Charge = 1, Multiplicity = 1			
Cu	0.0000004468	-0.3466445289	0.0000009048
Ν	-0.8536983996	1.0546022053	-1.1543503257
С	-0.1048039345	1.7727505393	-1.9956559722
Н	0.9430346652	1.4938321564	-2.053180929
С	-0.6255409354	2.8019783094	-2.7556174093
Н	0.0198270146	3.3582243434	-3.4241607502
С	-1.9740326804	3.0955906177	-2.6319170892
Н	-2.4191718206	3.9033480111	-3.2024298714
С	-2.7496612753	2.3498740289	-1.7625076951
Н	-3.8009832323	2.5772803228	-1.6344907044
С	-2.1678360218	1.321025694	-1.0303925949
Ν	-2.2492926161	-0.1108091138	0.8829069141
С	-2.9491892742	0.4974428049	-0.082868996
С	-4.3256243272	0.3456687644	-0.2065357219
Н	-4.8679728541	0.8165821298	-1.018059513
С	-4.992593991	-0.4488415683	0.7122312743
Н	-6.0649147596	-0.5910194699	0.6323293851
С	-4.2716984545	-1.0615869507	1.7239215912
Н	-4.7553104535	-1.6868584421	2.4651011682
С	-2.89955969	-0.8651496693	1.7646608792
Н	-2.2972740231	-1.3341537893	2.5392405569
С	0.42844208	-1.7665144311	1.2892494973
F	0.2859360354	-1.2426262327	2.5145069642
F	-0.3592731835	-2.8415045643	1.2594851198
F	1.6897832938	-2.1757690944	1.1740664544
С	2.1678359015	1.3210308235	1.0303884399
С	0.1048032725	1.7727587094	1.9956492404
С	0.6255396581	2.8019894648	2.7556070649

E(UM06) =	-1893.94778556
Charge = 0	Multiplicity $= 1$

O₂CCF₃

CF₃

CF₃

Cha	inge = 0, Multiplier	ty = 1	
C	u -0.382689437	4 -0.705429764	5 -0.338826477
Ν	-0.2139898651	1.1090695777	-1.0796609395
С	0.4668475697	1.3226599318	-2.2074344765
Н	0.7514250276	0.4425130894	-2.7758734146
С	0.8011164189	2.5933502627	-2.6319291943
Н	1.3609470075	2.7274314852	-3.5490471589
С	0.4032634425	3.6706059184	-1.8547968296
Н	0.649530936	4.684284014	-2.1514094574
С	-0.3022814863	3.4437836411	-0.6866982444
Н	-0.5969475798	4.2760954972	-0.0599682717
С	-0.5954730823	2.1400214955	-0.3060595353
Ν	-1.2340733187	0.5165740473	1.3251006052
С	-1.2811223516	1.8059763039	0.9586769498
С	-1.9313316038	2.7608499426	1.7301877117
Н	-1.9822584807	3.7952950663	1.412635811
С	-2.5327401184	2.3678841635	2.9152553927
Н	-3.0468556542	3.096855811	3.532209104
С	-2.4715531657	1.0382215921	3.295127466
Н	-2.9257154125	0.6891805615	4.2147629295
С	-1.8148131904	0.145687848	2.4612488627
Н	-1.7501617226	-0.9073063698	2.7214147582
С	-0.5813549406	-2.4314520717	0.5057449913
F	0.0922734557	-2.3426107123	1.655239311
F	-1.831374812	-2.7656709365	0.8289656405
F	-0.0500134246	-3.4133909809	-0.208534889
0	1.5680687636	-0.8381959302	-0.0855242108
С	2.2777190203	0.0721752992	0.4404466172
0	1.9579969449	1.0705423383	1.05424416
С	3.7842733931	-0.1760033555	0.1607793476
F	4.5700639442	0.6331267367	0.8629207201
F	4.1501176216	-1.4300260089	0.4319761232
F	4.0393183944	0.043749119	-1.1384869552
С	-1.9409177842	-1.1348560681	-1.4698276871
F	-3.0287624087	-0.6009130557	-0.9085244226
F	-2.206918793	-2.4130553989	-1.7319742704

-CF₃

CF₃

	O ₂ CCF ₃		
E(U	M06) = -2082.5007	5338	
Cha	rge = 0, Multiplicity	y = 1	
Cu	0.0000433789	-0.0822957401	-0.5060192517
Ν	0.0000195755	0.5317742763	1.3390032175
С	0.0000429457	-0.2888193462	2.3860295971
Н	0.0001054984	-1.3527007768	2.1660702859
С	-0.0000126155	0.2001265652	3.6805705087
Н	0.0000084159	-0.4867902623	4.5175598729
С	-0.0000967557	1.5736756374	3.8670506463
Н	-0.0001434485	1.9903695252	4.8681521115
С	-0.0001227336	2.4184536171	2.768304356
Н	-0.0001892473	3.4926090773	2.9058057174
С	-0.0000634436	1.8645507052	1.4971356115
Ν	-0.0000185901	1.9023646929	-0.8948914196
С	-0.0000781678	2.6317181155	0.2446512271

н	-0.584/8514/2	3.000296/16/	-0.0000209112
С	-0.7026626351	-0.3068329595	-1.9863085471
С	-2.6127424401	0.1329083438	-0.0000129932
С	-0.7026795883	-0.306801931	1.9863110164
F	-1.2181429633	-1.4495197231	-2.4842399784
F	0.5498126637	-0.2332975035	-2.4940373811
F	-1.3664512059	0.7103979627	-2.5675186922
F	-2.7288269492	1.4766515484	-0.0000545533
F	-3.2997519234	-0.310533642	-1.0621397328
F	-3.2997583486	-0.3104663319	1.0621371435
F	-1.3664382793	0.7104654663	2.5674926481
F	-1.2182049579	-1.4494597794	2.4842612145
F	0.5497932859	-0.2332997793	2.4940509378
	O ₂ CCF ₃		

CF

E(U	M06) = -1705.385	87253	
Charge = 0, Multiplicity = 1			
Cu	-0.698621354	-0.2823872189	0.0000010895
Ν	1.0567139747	-1.2239801553	0.0000129349
С	1.1219625374	-2.5547534813	0.0000277847
Н	0.1749295748	-3.0873088347	0.0000280171
С	2.3299633744	-3.2276998592	0.0000423575
Н	2.3469702853	-4.3104815429	0.0000543967
С	3.4964065524	-2.4797344723	0.0000416001
Н	4.4642396975	-2.9690661949	0.0000532867
С	3.4205801132	-1.0969780373	0.0000261244
Н	4.327284327	-0.5052115857	0.0000262786
С	2.1754809393	-0.4814432009	0.0000117245
Ν	0.7258464598	1.416212636	-0.0000057411
С	1.9942546796	0.9893273035	-0.000005018
С	3.0576698065	1.8840058701	-0.0000206705
Н	4.0841955144	1.538317702	-0.0000210898
С	2.7855156278	3.2432721027	-0.0000373922
Н	3.6004321486	3.9591240015	-0.0000504629
С	1.469259088	3.6752407548	-0.0000376265
Н	1.2186061945	4.7293265955	-0.0000504431
С	0.4655029475	2.718329229	-0.0000213216
Н	-0.5847831472	3.0002967167	-0.0000209112
С	-0.7026626351	-0.3068329595	-1.9863085471
С	-2.6127424401	0.1329083438	-0.0000129932
С	-0.7026795883	-0.306801931	1.9863110164
F	-1.2181429633	-1.4495197231	-2.4842399784
F	0.5498126637	-0.2332975035	-2.4940373811
F	-1.3664512059	0.7103979627	-2.5675186922
F	-2.7288269492	1.4766515484	-0.0000545533
F	-3.2997519234	-0.310533642	-1.0621397328
F	-3.2997583486	-0.3104663319	1.0621371435
F	-1.3664382793	0.7104654663	2.5674926481
F	-1.2182049579	-1.4494597794	2.4842612145

Н	-0.0000637638	4.3720039069	-3.1545139101	
С	-0.0000134134	2.5308925132	-2.0702204613	
Н	0.000037549	1.9206971337	-2.9640565402	
С	0.0000763563	-0.6971915378	-2.3300180371	
F	-1.0795024643	-0.2221437013	-2.9284178238	
F	1.0796799484	-0.2221357249	-2.9283655104	
F	0.0000797478	-2.0011594604	-2.3905922656	
0	-1.5394334108	-1.3277211093	-0.2223436053	
С	-2.6338239438	-0.7142126921	-0.0181123131	
0	-2.8874022354	0.4694889981	-0.1174850268	
С	-3.7349337856	-1.6865802033	0.4779676494	
F	-4.931676491	-1.1141263606	0.5135732873	
F	-3.8259370413	-2.7731095767	-0.2885728266	
F	-3.4414753835	-2.0922209301	1.7210961907	
0	1.5394655351	-1.3277417239	-0.2222954521	
С	2.6338954418	-0.714249928	-0.0182121298	
0	2.8875087532	0.4694286458	-0.1177495516	
С	3.7350010244	-1.6866081342	0.4778941075	
F	3.8259521909	-2.7731825257	-0.2885894436	
F	3.4415793716	-2.0921712442	1.7210552402	
F	4.9317554538	-1.1141760794	0.5134223796	
		F ₃		
	Ph			
F				
	Çulli-	-CF3		
	H ₃ CCN			
	NCC	H ₃		
Εſ	JM06) = -1972.494	199366		
Cha	arge = 0, Multiplic	ity = 1		
Ν	-2.2426333682	0.5139209908	-1.1791790873	
С	-3.1966044019	0.8888856061	-1.6917521841	
С	-4.3931684719	1.3614877922	-2.3338651037	
Н	-5.1553159936	1.5755884966	-1.5822327856	

2.2743421597

0.6020036381

2.9238957722

2.8021263224

1.7556233189

4.0760112837

4.1691192351

5.1890658319

4.0522648967

0.0316648965

-0.2142809594

-0.9501227056

0.7178757962

-1.5142574648

-1 1067499199

1.8457774681

0.5461528184

-0.8559733148

-0.299104887

-2.3604678179

2.7753763639

1.9844500661

1.4793840664

-0.3195821825

-2.8916069084

-1.2768852292

-0.0710189442

0.6405212244

0.8153983204

1.518036348

0.1001701753

1.6927548347

-0.2820200573

0.1797043708

1.2485252512

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Ν	2.0166460091	0.9382696439	-1.3430405876
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Н

Н

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Н	-5.3171070057	0.6326810698	-1.3889210004	
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Н	-4.6805848084	0.4916093259	-3.0446478035	
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С	-0.231954509	2.9392613474	-0.1661214631	
0	-0.3031838184	1.8998428347	0.5751892389	
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F	-1.4125516903	4.2560428248	1.4287656332	
F	-0.3041874291	5.3201525342	-0.0760472758	
F	0.74091672	4.3092933781	1.5093239096	
Cu	-0.3196738203	0.2989567938	-0.4550021678	
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F ₃ CCO ₂	 _Cu ^{III} —O₂C CF₃ NCCH₃

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(C	(CH ₃ CN) ₂ Cu ^{II}			
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0	-1.8860114818	0.156299674	1.4498938312	
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Н	-0.0288770689	4.6133946762	-0.0651858078	
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