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Copper(I)-catalyzed selective oxidation of hydrazones through C(sp³)-H functionalization

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Supporting Information

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1. Experimental Section

1.1 General procedure for the synthesis of 1

A mixture of ketone (5.0 mmol, 1.0 equiv.), 4-aminomorpholine (5.5 mmol, 1.1 equiv.), acetic acid (5.5 mmol, 1.1 equiv.), and EtOH (15.0 mL) in a 25 mL flask was stirred at room temperature for 1 h. After completion of the reaction as indicated by TLC, purified water was added to the mixture to form the precipitate. The precipitate was washed with water and dried *in vacuo* to afford ketone hydrazone.



1.2 General procedure for oxidation of hydrazones

A mixture of 1 (0.2 mmol), $K_2S_2O_8$ (2.0 equiv.), CuI (5 mol %), dry MeCN (2.0 mL) in a 25 mL tube was stirred at 70 °C for 1 h. Then, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 3:1).



1.2.1 other protecting group screening



1.3 Gram-scale preparation

A mixture of 1 (10.0 mmol), K₂S₂O₈ (2.0 equiv.), CuI (5 mol %), dry MeCN (25

mL) in a 100 mL tube was stirred at 70 °C for 2 h. Then, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 3:1)



1.4 General procedure for the synthesis of 3a

A mixture of **2a** (0.2 mmol), NH₂OH-HCl (2.0 equiv.), dry MeCN (2.0 mL) in a 25 mL tube was stirred at 70 °C for 2 h. After completion of the reaction as indicated by TLC, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 4:1).



1.5 General procedure for the synthesis of 3b

A mixture of **2a** (0.2 mmol), K_2CO_3 (2.0 equiv.), MeCN/H₂O (1:1, 2.0 mL) in a 25 mL tube was stirred at room temperature for 2 h. After completion of the reaction as indicated by TLC, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 2:1).



1.6 Control Experiments

(a) A mixture of acetophenone (0.2 mmol), $K_2S_2O_8$ (2.0 equiv.), CuI (5 mol %), dry MeCN (2.0 mL) in a 25 mL tube was stirred at 70 °C for 1 h. Only a trace amount of the product was detected.



(b) A mixture of 1 (0.2 mmol), $K_2S_2O_8$ (2.0 equiv.), CuI (5 mol %), TEMPO (1.0 or 2.0 equiv.) dry MeCN (2.0 mL) in a 25 mL tube was stirred at 70 °C for 1 h. Then, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 3:1).



(c) A mixture of 1 (0.2 mmol), $K_2S_2O_8$ (2.0 equiv.), CuI (5 mol %), dry MeCN (2.0 mL) in a 25 mL tube was stirred at 70 °C for 1 h under ¹⁸O₂ atmosphere. Then, the mixture was diluted with EtOAc and filtered through a pad of Celite gradually. The solvent was removed under reduced pressure. The gathered residue was then purified by silica gel column chromatography (200–300 mesh silica gel, PE/EA = 3:1).



1. Copies of NMR Spectra

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4	44000000	
6		
1		





2a ¹H NMR (500 MHz, CDCl₃)







210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 Chemical shift (ppm)





2b ¹H NMR (500 MHz, CDCl₃)







$$\begin{array}{c} -9.45 \\ -9.45 \\ 7.26 \\ 7.23 \\ 7.10 \\ 7.10 \\ 7.10 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.09 \\ 7.00 \\$$



2c¹H NMR (500 MHz, CDCl₃)



























9.44 9.45





210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 Chemical shift (ppm)





2l¹H NMR (500 MHz, CDCl₃)









2m ¹H NMR (500 MHz, CDCl₃)







9.48 7.28 7.28 7.27 7.26 7.22 7.22 7.22 7.22 7.20 7.08 7.08 3.67 3.65 3.27 3.27 3.27 3.27 3.27 3.26 3.27 3.26 3.27 3.26 3.27 3.26 13.25 3.26 13.25 15







20 ¹H NMR (500 MHz, CDCl₃)















-1 -1











HRMS (ESI): $[M+Na]^+$ Calculated for $C_{12}H_{14}N_2O^{18}ONa$: 243.0995, Found 243.1005.