# Preparation of humulene epoxides



To a solution of humulene **1** (124.0 mg, 0.61 mmol, 1 eq.) in  $CH_2Cl_2$  (6 mL) at -10 °C was added portionwise *m*CPBA (77%, 136.0 mg, 0.61 mmol, 1 eq.). After stirring at -10 °C for 24 h, the reaction mixture was diluted with H<sub>2</sub>O (20 mL) and extracted with diethyl ether (3 × 10 mL). The combined organic phases were dried over MgSO<sub>4</sub> and then concentrated under reduced pressure. The residue was purified by column chromatography on silica gel (petrol ether / diethyl ether = 6 / 1) to give **8** and a mixture of **6** and **7**. The mixture of **6** and **7** was further purified by preparative HPLC on a chiral stationary phase to give (+)-**6**, (-)-**6**,<sup>1</sup> (+)-**7** and (-)-**7**.<sup>1</sup>

Compound (+)-6. TLC (petrol ether / diethyl ether = 5 / 1):  $R_f$  = 0.38; yield: 4.0 mg, 0.02 mmol (3%); optical rotation: [ $\alpha$ ]<sub>D</sub><sup>25</sup> = +65.5 (*c* 0.41, benzene); GC (HP-5MS): *I* = 1629; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table S1.

Compound (–)-6. TLC (petrol ether / diethyl ether = 5 / 1):  $R_f = 0.38$ ; yield: 4.0 mg, 0.02 mmol (3%); optical rotation:  $[\alpha]_D^{2^5} = -45.9$  (*c* 0.47, benzene), lit:  $[\alpha]_D = -22.8$  (CHCl<sub>3</sub>);<sup>1</sup> GC (HP-5MS): I = 1629; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table S1.

Compound (+)-7. TLC (petrol ether / diethyl ether = 5 / 1):  $R_f = 0.38$ ; yield: 30 mg, 0.14 mmol (22%); optical rotation:  $[\alpha]_D^{25} = +95.9$  (*c* 0.15, benzene); GC (HP-5MS): *I* = 1641; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table S2.

Compound (–)-7. TLC (petrol ether / diethyl ether = 5 / 1):  $R_f = 0.38$ ; yield: 30 mg, 0.14 mmol (22%); optical rotation:  $[\alpha]_D^{25} = -106.3$  (*c* 0.14, benzene), lit:  $[\alpha]_D = -31.2$  (CHCl<sub>3</sub>);<sup>1</sup> GC (HP-5MS): I = 1641; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table S2.

Compound **8**. TLC (petrol ether / diethyl ether = 5 / 1):  $R_f = 0.71$ ; yield: 7.0 mg, 0.03 mmol (5%); GC (HP-5MS): I = 1665; <sup>1</sup>H-NMR and <sup>13</sup>C-NMR see Table S3.

#### HPLC

Analytical scale HPLC seperation was carried out using an Azura series HPLC system (Knauer, Berlin, Germany), equipped with DAD 6.1L photodiode array detector (190–1020 nm) and a Daicel (Tokyo, Japan) Chiralpak IC-U column (1.6  $\mu$ m; 3.0 mm × 100 mm) using an isocratic solvent mixture of [*n*-heptane / 2-propanol (98:2)] with 0.85 mL min<sup>-1</sup> (352 bar). The UV–Vis absorption was monitored at 210 nm.

Preparative scale HPLC purification was performed on an Azura series HPLC system (Knauer, Berlin, Germany) with a multi wavelength detector MWL 2.1L (190–700 nm) using a YMC ChiralART Cellulose-SC column (5  $\mu$ m, 250 × 20 mm). The solvent mixture [*n*-heptane / 2-propanol (98:2)] was used at 18.0 mL min<sup>-1</sup> (44 bar) and monitoring was conducted at 210 nm.

#### NMR spectroscopy

NMR spectra were recorded at 298 K on a Bruker (Billerica, MA, USA) Avance III HD Cryo (700 MHz) NMR spectrometer. Spectra were measured in C<sub>6</sub>D<sub>6</sub> and referenced against solvent signals (<sup>1</sup>H-NMR, residual proton signal:  $\delta$  = 7.16 ppm; <sup>13</sup>C-NMR:  $\delta$  = 128.06 ppm).<sup>2</sup>

#### GC/MS analyses

GC/MS analyses were carried out using a 7890B GC equipped with a HP5-MS fused silica capillary column (30 m, 0.25 mm i. d., 0.50  $\mu$ m film) and connected to a 5977A mass detector

(Agilent). GC parameters: 1) temperature program: 5 min at 50 °C increasing at 5 °C min<sup>-1</sup> to 320 °C, 2) injection volume: 2  $\mu$ L, 3) split ratio: 10:1, 60 s valve time, and 4) carrier gas: He at 1 mL min<sup>-1</sup>. Retention indices (*I*) were determined from a homologous series of *n*-alkanes (C<sub>7</sub>-C<sub>40</sub>). MS parameters: 1) inlet pressure: 77.1 kPa, He at 23.3 mL min<sup>-1</sup>, 2) transfer line: 250 °C, and 3) electron energy: 70 eV.

# **Optical rotations**

Optical rotations were recorded on a Modular Compact Polarimeter MCP 100 (Anton Paar, Graz, Austria). The temperature setting was 25 °C; the wavelength of the light used was 589 nm (the sodium D line); the path-length was 10 cm, the compound concentrations *c* are given in g 100 mL<sup>-1</sup>.



**Figure S1.** Structure elucidation of **6**. Bold lines: COSY correlations, single headed arrows: key HMBC correlations, double headed arrows: key NOESY correlations. The observed NOESY correlations point to the presence of several conformers.

C <sup>[a]</sup>	type	<sup>1</sup> H <sup>[b]</sup>	<sup>13</sup> C <sup>[b]</sup>
1	CH <sub>2</sub>	1.65 (d, <i>J</i> = 14.2, H <sub>β</sub> )	41.78
		1.36 (dd, <i>J</i> = 14.2, 9.8, H <sub>α</sub> )	
2	СН	2.53 (d, <i>J</i> = 9.6)	62.87
3	Cq	-	60.62
4	$CH_2$	2.02 (m, H <sub>α</sub> )	39.12
		1.15 (ddd, $J = 12.4$ , 12.4, 5.6, H <sub><math>\beta</math></sub> )	
5	$CH_2$	2.01 (m, H <sub>β</sub> )	23.75
		1.91 (m, H <sub>α</sub> )	
6	СН	5.06 (br dd, <i>J</i> = 8.9, 6.5)	126.66
7	Cq	-	139.06
8	$CH_2$	2.44 (d, <i>J</i> = 7.3, 2H)	39.90
9	СН	5.60 (ddd, <i>J</i> = 15.8, 7.2, 7.2)	128.25
10	СН	5.24 (d, <i>J</i> = 15.8)	141.66
11	Cq	-	35.11
12	$CH_3$	0.93 (s)	30.79
13	$CH_3$	1.04 (s)	23.87
14	$CH_3$	1.50 (br s)	18.10
15	$CH_3$	1.08 (s)	16.37

**Table S1.** NMR data of humulene epoxide I (6) in C<sub>6</sub>D<sub>6</sub> recorded at 298 K.

[a] Carbon numbering as shown in Figure S1. [b] Chemical shifts  $\delta$  in ppm, multiplicity: s = singlet, d = doublet, m = multiplet, br = broad, coupling constants *J* are given in Hertz.



Figure S2. <sup>1</sup>H-NMR spectrum of 6 (700 MHz,  $C_6D_6$ ).



Figure S3.  $^{13}\text{C}\text{-NMR}$  spectrum of 6 (176 MHz,  $C_6D_6).$ 



Figure S4. <sup>13</sup>C-DEPT spectrum of 6 (176 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S5. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 6 (700 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S6. HSQC spectrum of  $6 (C_6 D_6)$ .



Figure S7. HMBC spectrum of 6 (C<sub>6</sub>D<sub>6</sub>).



Figure S8. NOESY spectrum of 6 (C<sub>6</sub>D<sub>6</sub>).



**Figure S9.** Structure elucidation of **7**. Bold lines: COSY correlations, single headed arrows: key HMBC correlations, double headed arrows: key NOESY correlations.

C <sup>[a]</sup>	type	<sup>1</sup> H <sup>[b]</sup>	<sup>13</sup> C <sup>[b]</sup>
1	CH <sub>2</sub>	1.87 (dd, <i>J</i> = 13.6, 9.0, H <sub>β</sub> )	40.59
		1.77 (br dd, $J = 13.7, 5.9, H_{\alpha}$ )	
2	СН	4.88 (m)	125.76
3	Cq	-	132.01
4	$CH_2$	2.00 (m, H <sub>β</sub> )	36.99
		1.93 (m, H <sub>α</sub> )	
5	$CH_2$	2.01 (m, H <sub>β</sub> )	25.28
		1.26 (m, H <sub>α</sub> )	
6	СН	2.39 (dd, <i>J</i> = 10.1, 3.6)	61.23
7	Cq	-	62.51
8	$CH_2$	2.47 (dd, <i>J</i> = 12.5, 5.4, H <sub>α</sub> )	43.02
		1.69 (dd, <i>J</i> = 12.5, 10.5, H <sub>β</sub> )	
9	СН	5.12 (ddd, <i>J</i> = 15.9, 10.1, 5.3)	122.94
10	СН	4.96 (d, <i>J</i> = 16.0)	142.64
11	Cq	-	36.47
12	CH₃	0.98 (s)	29.06
13	CH₃	1.01 (s)	25.83
14	CH₃	1.13 (br s)	17.49
15	CH₃	1.34 (br s)	15.06

**Table S2.** NMR data of humulene epoxide II (7) in C<sub>6</sub>D<sub>6</sub> recorded at 298 K.

[a] Carbon numbering as shown in Figure S9. [b] Chemical shifts  $\delta$  in ppm, multiplicity: s = singlet, d = doublet, m = multiplet, br = broad, coupling constants *J* are given in Hertz.



Figure S10. <sup>1</sup>H-NMR spectrum of 7 (700 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S11.  $^{13}$ C-NMR spectrum of 7 (176 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S12. <sup>13</sup>C-DEPT spectrum of 7 (176 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S13. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 7 (700 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S14. HSQC spectrum of 7 ( $C_6D_6$ ).



Figure S15. HMBC spectrum of 7 ( $C_6D_6$ ).



Figure S16. NOESY spectrum of 7 ( $C_6D_6$ ).



**Figure S17.** Structure elucidation of **8**. Bold lines: COSY correlations, single headed arrows: key HMBC correlations, double headed arrows: key NOESY correlations.

C <sup>[a]</sup>	type	<sup>1</sup> H <sup>[b]</sup>	<sup>13</sup> C <sup>[b]</sup>
1	CH <sub>2</sub>	2.11 (dd, <i>J</i> = 14.6, 10.8, H <sub>β</sub> )	39.62
		1.79 (dm, <i>J</i> = 14.6, H <sub>α</sub> )	
2	СН	4.87 (br dd, <i>J</i> = 7.1, 7.1)	123.63
3	Cq	-	133.41
4	CH <sub>2</sub>	2.02 (m, H <sub>β</sub> )	39.51
		1.91 (m, H <sub>α</sub> )	
5	$CH_2$	1.99 (m, 2H)	23.72
6	СН	4.81 (dm, <i>J</i> = 10.4)	126.47
7	Cq	-	132.12
8	$CH_2$	2.64 (dd, <i>J</i> = 12.8, 4.2, H <sub>α</sub> )	42.49
		1.54 (dd, <i>J</i> = 12.8, 9.4, H <sub>β</sub> )	
9	СН	2.80 (ddd, <i>J</i> = 9.5, 4.3, 2.3)	56.53
10	СН	2.19 (d, <i>J</i> = 2.3)	64.79
11	Cq	-	34.59
12	CH₃	1.10 (s)	29.15
13	CH₃	0.74 (s)	19.28
14	CH₃	1.40 (br s)	17.84
15	CH₃	1.36 (dd, <i>J</i> = 1.4, 1.4)	15.36

**Table S3.** NMR data of humulene epoxide III (8) in C<sub>6</sub>D<sub>6</sub> recorded at 298 K.

[a] Carbon numbering as shown in Figure S17. [b] Chemical shifts  $\delta$  in ppm, multiplicity: s = singlet, d = doublet, m = multiplet, br = broad, coupling constants *J* are given in Hertz.



Figure S18. <sup>1</sup>H-NMR spectrum of 8 (700 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S19.  $^{13}$ C-NMR spectrum of 8 (176 MHz, C<sub>6</sub>D<sub>6</sub>).





Figure S21. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 8 (700 MHz, C<sub>6</sub>D<sub>6</sub>).



Figure S22. HSQC spectrum of 8 ( $C_6D_6$ ).



Figure S23. HMBC spectrum of 8 ( $C_6D_6$ ).



Figure S24. NOESY spectrum of 8 ( $C_6D_6$ ).

# Isotopic labelling experiments

Isotopic labelling experiments were performed with the substrates and enzymes as listed in Table S4.

For the reactions entries 1 - 4, the reaction mixtures contained substrates (5 mg each) in aqueous NH<sub>4</sub>HCO<sub>3</sub> solution (2 mL, 25 mM), enzyme elution fractions (5 mL each) purified as reported<sup>3</sup> and incubation buffer (13 mL, 50 mM Tris, 10 mM MgCl<sub>2</sub>, 20 vol-% glycerol, pH 8.2). After incubation with shaking at 28 °C overnight, the reaction mixtures were extracted with diethyl ether (50 mL).  $\alpha$ -Humulene (30 mg) was added and the solvent was evaporated, followed by conversion with mCPBA into epoxides and separation of stereoisomers by HPLC on a chiral stationary phase to obtain (1*S*,6*S*,7*S*)-[1-<sup>13</sup>C,1-<sup>2</sup>H]-7 and (1*S*,6*R*,7*R*)-[1-<sup>13</sup>C,1-<sup>2</sup>H]-7 from experiment 1, (1*R*,6*S*,7*S*)-[1-<sup>13</sup>C,1-<sup>2</sup>H]-7 and (1*R*,6*R*,7*R*)-[1-<sup>13</sup>C,1-<sup>2</sup>H]-7 from experiment 2, (6*S*,7*S*,11*R*)-[12-<sup>13</sup>C]-7 and (6*R*,7*R*,11*S*)-[12-<sup>13</sup>C]-7 from experiment 3, and (6*S*,7*S*,11*S*)-[12-<sup>13</sup>C]-7 from experiment 4. All products were analysed by NMR spectroscopy.

For the reactions entries 5 and 6, the reaction mixtures contained substrates (0.5 mg each) in aqueous  $NH_4HCO_3$  solution (0.1 mL, 25 mM), enzyme elution fractions (0.2 mL each) and incubation buffer (0.5 mL). After incubation with shaking at 28 °C for 5 h, the reaction mixtures were extracted with *n*-hexane (0.15 mL). The extracts were dried over MgSO<sub>4</sub> and analysed by GC-MS.

entry	substrate	enzymes	results shown in
1	( <i>R</i> )-(1- <sup>13</sup> C,1- <sup>2</sup> H)FPP <sup>4</sup>	AsR6⁵	Figure S25
2	(S)-(1- <sup>13</sup> C,1- <sup>2</sup> H)FPP <sup>4</sup>	AsR6	Figure S25
3	(12- <sup>13</sup> C)FPP <sup>6</sup>	AsR6	Figure S26
4	(13- <sup>13</sup> C)FPP <sup>6</sup>	AsR6	Figure S26
5	( <i>R</i> )-(1- <sup>2</sup> H)IPP <sup>6</sup>	IDI, <sup>8</sup> FPPS, <sup>9</sup> AsR6	Figure S27
6	(S)-(1- <sup>2</sup> H)IPP <sup>7</sup>	IDI, FPPS, AsR6	Figure S27

 Table S4. Labelling experiments with AsR6.





B) (R)-(1-<sup>13</sup>C,1-<sup>2</sup>H)FPP







**Figure S25.** Inversion of configuration at C1. Partial HSQC spectra of A) unlabelled **7** showing crosspeaks for the two diastereotopic hydrogens at C1 (signals down for CH<sub>2</sub> groups), B) (*S*,*S*)-**7** and (*R*,*R*)-**7** from (*R*)-(1-<sup>13</sup>C,1-<sup>2</sup>H)FPP mixed with unlabelled **7**, and C) (*S*,*S*)-**7** and (*R*,*R*)-**7** from (*S*)-(1-<sup>13</sup>C,1-<sup>2</sup>H)FPP mixed with unlabelled **7**. In B) and C) the signals pointing down are from unlabelled **7** (CH<sub>2</sub> groups) and the signals pointing up are from labelled **7** (CH<sup>2</sup>H groups), showing a slight upfield shift in both dimensions resulting from a deuterium isotope effect.



**Figure S26.** *Re* face attack at C11. Partial <sup>13</sup>C-NMR spectra of A) unlabelled **7**, B) (*S*,*S*)-**7** from (12-<sup>13</sup>C)FPP mixed with unlabelled **7** (incorporation into  $\delta = 25.83$  ppm), C) (*R*,*R*)-**7** from (12-<sup>13</sup>C)FPP mixed with unlabelled **7** (incorporation into  $\delta = 29.06$  ppm), D) (*S*,*S*)-**7** from (13-<sup>13</sup>C)FPP mixed with unlabelled **7** (incorporation into  $\delta = 29.06$  ppm), and E) (*R*,*R*)-**7** from (13-<sup>13</sup>C)FPP mixed with unlabelled **7** (incorporation into  $\delta = 29.06$  ppm), and E) (*R*,*R*)-**7** from (13-<sup>13</sup>C)FPP mixed with unlabelled **7** (incorporation into  $\delta = 29.06$  ppm).



**Figure S27.** Deprotonation from C9. EI mass spectra of A) unlabelled **1**, B)  $({}^{2}H_{2})$ -**1** obtained with IDI, FPPS and AsR6 from (*S*)-(1- ${}^{2}H$ )IPP, and C)  $({}^{2}H_{3})$ -**1** obtained with IDI, FPPS and AsR6 from (*R*)-(1- ${}^{2}H$ )IPP.

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