Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry. This journal is © The Royal Society of Chemistry 2021

# **Electronic Supplementary Information**

## Rhodium(III)-Catalyzed Cascade C-H Functionalization/Annulation

## of Sulfoximines with Iodonium ylides for the Synthesis of

## Cyclohexanone-1,2-benzothiazines

Lu Chen,‡<sup>a</sup> Zhichao Wang,‡<sup>a</sup> Yangyang Wang,<sup>a</sup> Liqiang Hao,<sup>a</sup> Xiaobo Xu,<sup>a</sup> Gaorong Wu,\*<sup>a</sup> and Yafei Ji\*<sup>a</sup>

<sup>a</sup> Engineering Research Centre of Pharmaceutical Process Chemistry, Ministry of Education;

School of Pharmacy,

East China University of Science and Technology,

130 Meilong Road, Shanghai 200237, P. R. China

E-mail: jyf@ecust.edu.cn (Ji).

E-mail: gaorongwu09@163.com (Wu).

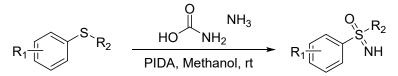
<sup>‡</sup>These authors have equally contributed to this article.

### **Table of Contents**

1.General Procedure for Preparation of the Substrates1
2.1.0 mmol scale experiment and derivatization reaction1
3.Control experiments
4. Single Crystal Structure and Crystallographic Data for <b>3d</b> 9
5. <sup>1</sup> H and <sup>13</sup> C NMR Spectrum for All isolated Products11
6.References

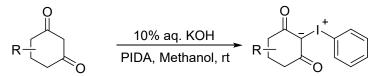
#### **1** General Procedure for Preparation of the Substrates

#### 1.1 Synthesis of substrates 1<sup>1</sup>



To a flask containing a stirrer bar was added successively, sulfide (1 equiv), ammonium carbamate (1.5 equiv) and then MeOH (0.5 M). PIDA (2.1 equiv) was added in one portion and the reaction was stirred at 20 °C for 30 min (open flask to the atmosphere). The solvent was removed under reduced pressure and the crude product was purified by flash chromatography on silica gel.

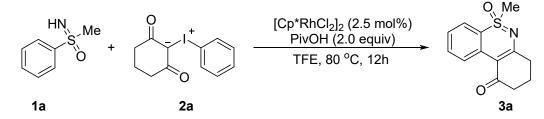
#### **1.2 Synthesis of substrates 2**<sup>2</sup>



Taken in a 100 mL round bottom flask fitted with a magnetic stirrer, added solution of cyclic 1,3dione (1.0 equiv) in 30 mL methanol at room temperature added 20 mL of 10% aqueous solution of KOH followed by addition of diacetoxy iodobenzene (PIDA) (1.2 equiv) in 40 mL methanol. The reaction mixture was stirred for 2 hours at room temperature and then quenched with ice cold water. The resulting white precipitate was filtered and mother solution was extracted with DCM, then washed with water three times, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. The resultant white solid was mixed with the first crop and the mixture recrystallized from DCM/Hexane. The precipitate compound was obtained and can be directly used further without purification.

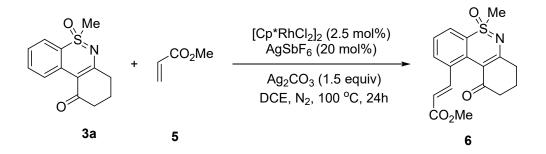
### 2 1.0 mmol scale experiment and derivatization reaction

#### 2.1 1.0 mmol scale experiment

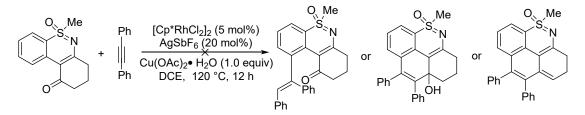


A mixture of substrates sulfoximine **1a** (155.0 mg, 1 mmol), iodonium ylide **2a** (376.8 mg, 1.2 mmol),  $[Cp*RhCl_2]_2$  (15.5 mg, 2.5 mol%), PivOH (204.3 mg, 2.0 mmol, 2.0 equiv) in TFE (5.0 mL) was charged in a glass sealed-tube and stirred at 80 °C (oil bath) for 12 h. Upon completion of the reaction, water (20 mL) and DCM (10 mL) were added to the mixture, then the aqueous layer was extracted with DCM (10 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 3:1) to supply the desired products **3a** as a white solid (214.9 mg, 87%).

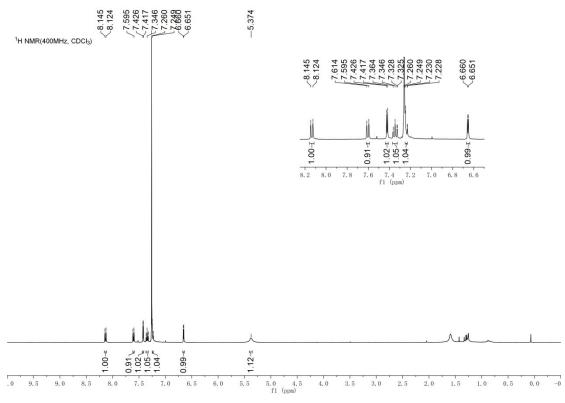
#### 2.2 Derivatization reaction



A mixture of **3a** (24.7 mg, 0.1 mmol), methyl acrylate **5** (17.2 mg, 0.2 mmol),  $[Cp*RhCl_2]_2$  (3.1 mg, 5 mol%), AgSbF<sub>6</sub> (7.1 mg, 20 mol%) and Ag<sub>2</sub>CO<sub>3</sub> (41.4 mg, 0.15 mmol, 1.5 equiv) in DCE (1.0 mL) was charged in a schlenk tube and stirred at 100 °C (oil bath) under N<sub>2</sub> for 24 h. Upon completion of the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 1:2) to supply the desired products **6** (28.1 mg, 85%).



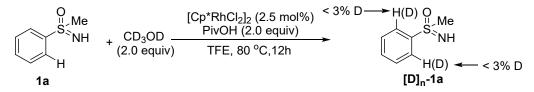
A mixture of **3a** (24.7 mg, 0.1 mmol), diphenylacetylene (21.4 mg, 0.2 mmol),  $[Cp*RhCl_2]_2$  (3.1 mg, 5 mol%), AgSbF<sub>6</sub> (7.1 mg, 20 mol%) and Cu(OAc)<sub>2</sub>·H<sub>2</sub>O (19.9 mg, 0.1 mmol, 1.0 equiv) in DCE (1.0 mL) was charged in a glass sealed-tube and stirred at 120 °C (oil bath) for 12 h. Upon completion of the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel to supply the major products. The following <sup>1</sup>H NMR analysis showed that it was not the desired product.



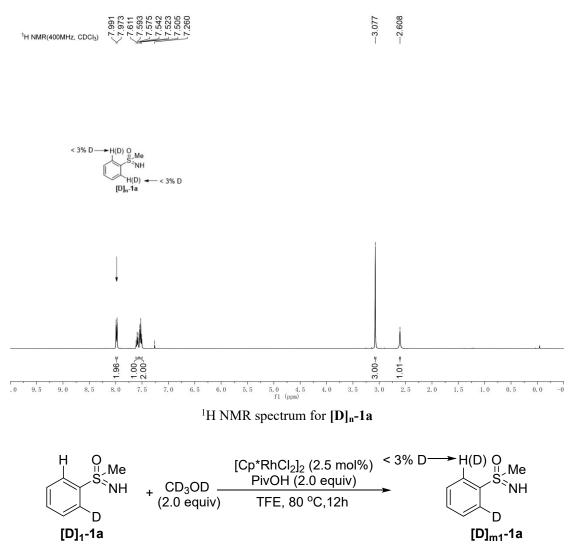
<sup>1</sup>H NMR spectrum for the major product

# **3** Control experiments

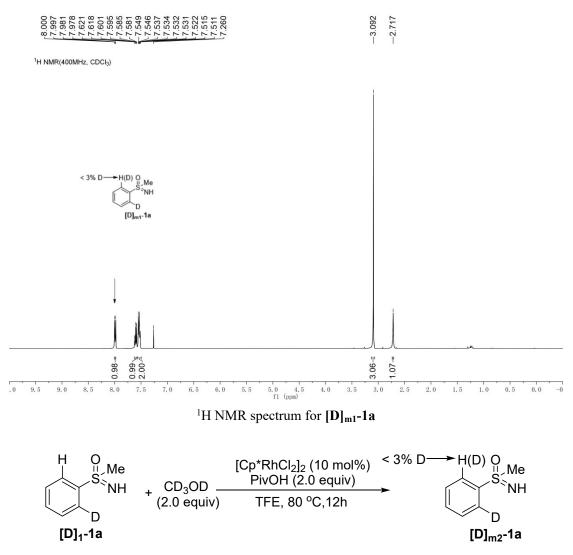
### **3.1 H/D Exchange reaction**



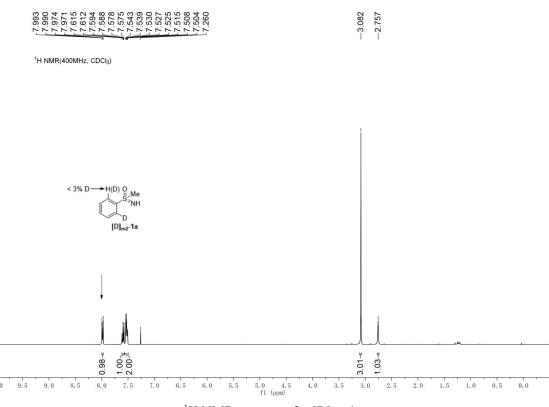
A mixture of sulfoximine **1a** (31.0 mg, 0.2 mmol),  $[Cp*RhCl_2]_2$  (3.1 mg, 2.5 mol%) and PivOH (40.9 mg, 0.4 mmol, 2.0 equiv), CD<sub>3</sub>OD (14.4 mg, 0.4 mmol, 2.0 equiv) in TFE (2.0 mL) was charged in a glass sealed-tube and the reaction mixture was allowed to stir at 80 °C for 12 h. Upon completion of the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 1:1) to give product [**D**]<sub>n</sub>-1**a**. The <sup>1</sup>H NMR analysis showed that < 3% hydrogen of the each *ortho*-position of 1**a** was deuterated.



A mixture of sulfoximine  $[D]_{1}$ -1a (31.2 mg, 0.2 mmol),  $[Cp*RhCl_{2}]_{2}$  (3.1 mg, 2.5 mol%) and PivOH (40.9 mg, 0.4 mmol, 2.0 equiv), CD<sub>3</sub>OD (14.4 mg, 0.4 mmol, 2.0 equiv) in TFE (2.0 mL) was charged in a glass sealed-tube and the reaction mixture was allowed to stir at 80 °C for 12 h. Upon completion of the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 1:1) to give product  $[D]_{m1}$ -1a. The <sup>1</sup>H NMR analysis showed that < 3% hydrogen of the  $[D]_{1}$ -1a was deuterated.

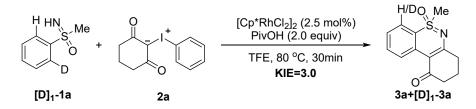


A mixture of sulfoximine  $[D]_{1}$ -1a (31.2 mg, 0.2 mmol),  $[Cp*RhCl_2]_2$  (12.4 mg, 10 mol%) and PivOH (40.9 mg, 0.4 mmol, 2.0 equiv), CD<sub>3</sub>OD (14.4 mg, 0.4 mmol, 2.0 equiv) in TFE (2.0 mL) was charged in a glass sealed-tube and the reaction mixture was allowed to stir at 80 °C for 12 h. Upon completion of the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL ×3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 1:1) to give product  $[D]_{m2}$ -1a. The <sup>1</sup>H NMR analysis showed that < 3% hydrogen of the  $[D]_1$ -1a was deuterated.



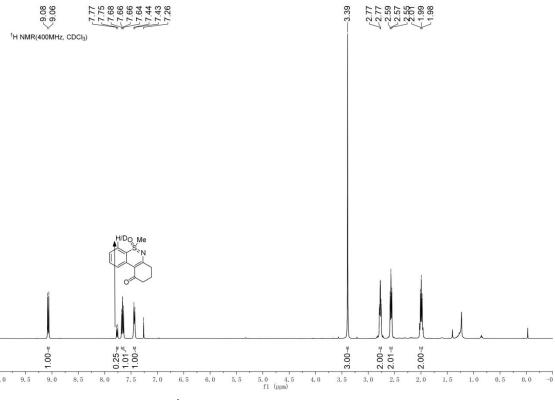
<sup>1</sup>H NMR spectrum for **[D]**<sub>m2</sub>-1a

#### **3.2 Intramolecular KIE Experiment**



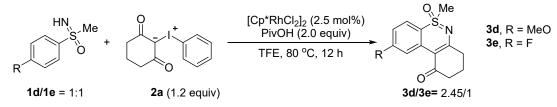
A mixture of substrates  $[D]_{1}$ -1a (31.2 mg, 0.2 mmol), iodonium ylide 2a (75.6 mg, 0.24 mmol),  $[Cp*RhCl_2]_2$  (3.1 mg, 2.5 mol%), PivOH (40.9 mg, 0.4 mmol, 2.0 equiv) in TFE (2.0 mL) was charged in a glass sealed-tube and stirred at 80 °C (oil bath) for 30 min. After the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL × 3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 3:1) to supply the mixed products **3a** and  $[D]_1$ -**3a** (13.8 mg, 28%). The product distribution  $k_H/k_D = 3.0$  (0.75/0.25) was analyzed by <sup>1</sup>H NMR.

 $(3a+[D]_1-3a)$ : <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, ppm):  $\delta$  9.07 (d, J = 8.4 Hz, 1H), 7.76 (d, J = 8.0 Hz, 0.25H), 7.66 (dd,  $J_1$  = 8.3,  $J_2$  = 7.5 Hz, 1H), 7.44 (d, J = 7.0 Hz, 1H), 3.39 (s, 3H), 2.81–2.74 (m, 2H), 2.60–2.53 (m, 2H), 2.03–1.96 (m, 2H).

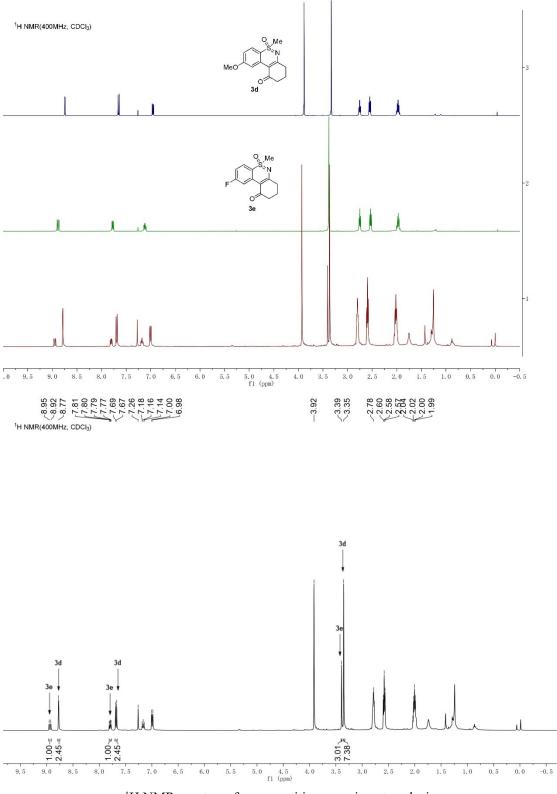


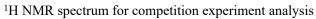
<sup>1</sup>H NMR spectrum for KIE analysis

#### 3.3 Competition experiment



A mixture of substrates 1d/1e (1d, 37.1 mg, 0.2 mmol; 1e, 34.6mg, 0.2 mmol), iodonium ylide 2a (75.6 mg, 0.24 mmol), [Cp\*RhCl<sub>2</sub>]<sub>2</sub> (3.1 mg, 2.5 mol%), PivOH (40.9 mg, 0.4 mmol, 2.0 equiv) in TFE (4.0 mL) was charged in a glass sealed-tube and stirred at 80 °C (oil bath) for 30 min. After the reaction, water (10 mL) and DCM (5 mL) were added to the mixture, then the aqueous layer was extracted with DCM (5 mL × 3). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Finally, the solution was concentrated in vacuo to provide a crude product, which was further purified via a column chromatography on silica gel (eluents: petroleum ether/ethyl acetate = 10:1 to 2:1) to supply the mixed products 3d and 3e. <sup>1</sup>H NMR of mixture was presented as red color; <sup>1</sup>H NMR of 3d was presented as blue color; <sup>1</sup>H NMR of 3e was presented as green color.





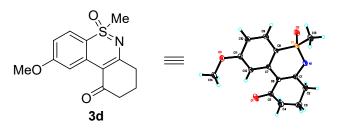
## 4 Single Crystal Structure of 3d

#### **Sample preparation**

20 mg of **3d** was added to a 10 mL glass vial and dissolved in 1 mL DCM and l mL methanol. The glass vial was capped loosely and kept for slow evaporation. After 7 days, single crystal was obtained and then subjected to X-ray diffraction.

#### **Crystal measurement**

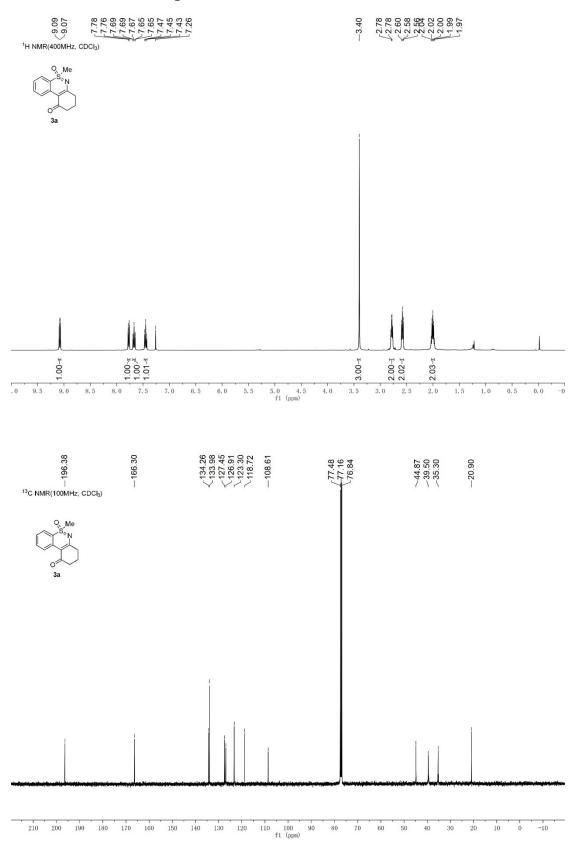
X-ray diffraction data were recorded on a Bruker D8 Venture single-crystal X-ray diffractometer. Absorption was corrected by semi-empirical from equivalents. The structure was solved by direct methods and refined by full-matrix least squares on  $F^2$ . All non-hydrogen atoms were refined anisotropically. All hydrogen atoms were placed in calculated positions. The X-ray crystallographic files, in CIF format, are available from the Cambridge Crystallographic Data Centre on quoting the deposition numbers CCDC 2114921 for compound **3d**.



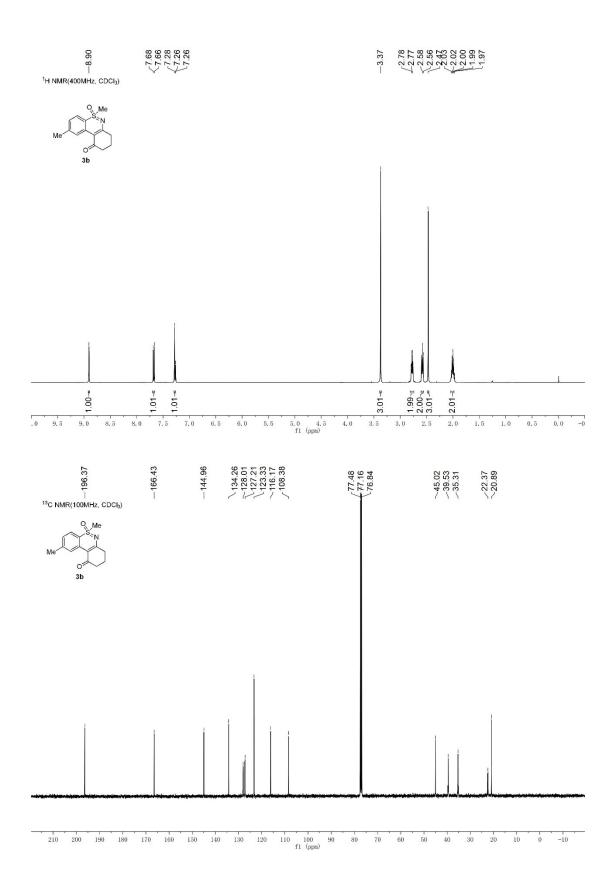
X-ray structure of **3d** CCDC 2114921 Ellipsoids are drawn at 50% probability level.

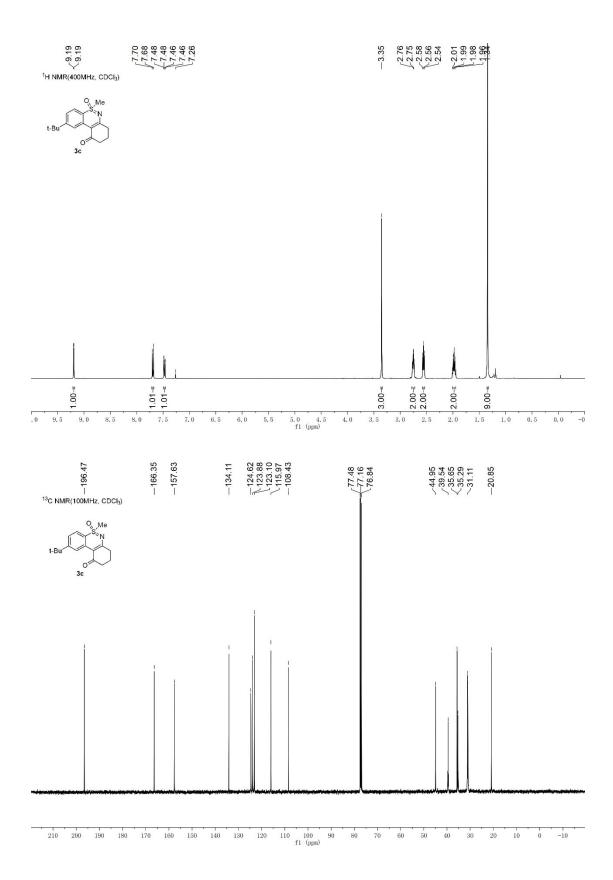
Identification code	d8v21755	
Empirical formula	C14 H15 N O3 S	
Formula weight	277.33	
Temperature	213(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P 21/n	
Unit cell dimensions	a = 10.3951(5) Å	a= 90°.
	b = 8.7516(4)  Å	b=95.6710(10)°.
	c = 14.2044(6)  Å	$g = 90^{\circ}$ .
Volume	1285.90(10) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.433 Mg/m <sup>3</sup>	
Absorption coefficient	0.255 mm <sup>-1</sup>	
F(000)	584	
Crystal size	0.200 x 0.160 x 0.120 mm <sup>3</sup>	
Theta range for data collection	3.049 to 25.995°.	
Index ranges	-12<=h<=12, -10<=k<=10, -17<=l<=17	

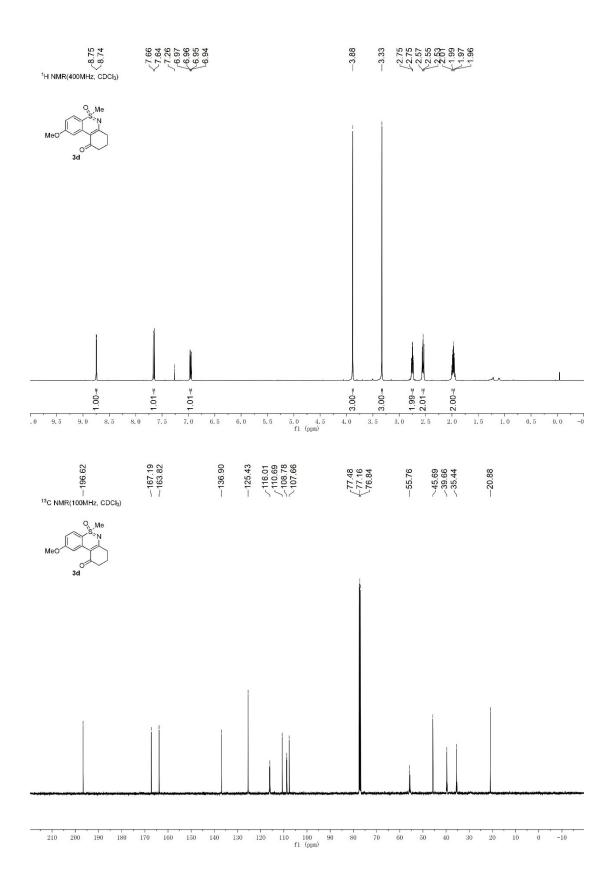
Reflections collected Independent reflections Completeness to theta =  $25.242^{\circ}$ Absorption correction Max. and min. transmission Refinement method Data / restraints / parameters Goodness-of-fit on F<sup>2</sup> Final R indices [I>2sigma(I)] R indices (all data) Extinction coefficient Largest diff. peak and hole 12404 2501 [R(int) = 0.0295] 98.7 % Semi-empirical from equivalents 0.7456 and 0.6497 Full-matrix least-squares on F<sup>2</sup> 2501 / 0 / 174 1.071 R1 = 0.0336, wR2 = 0.0879 R1 = 0.0365, wR2 = 0.0900 n/a 0.316 and -0.320 e.Å<sup>-3</sup>

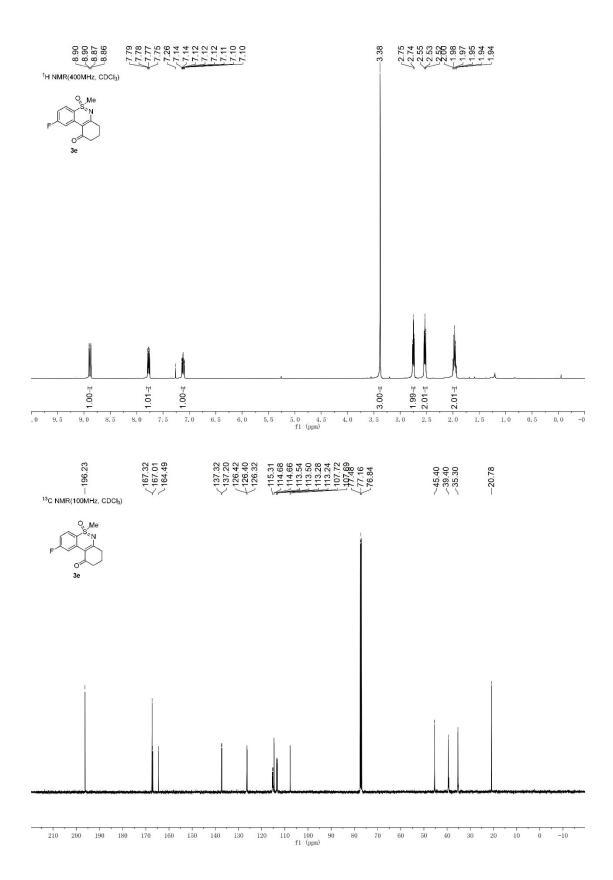


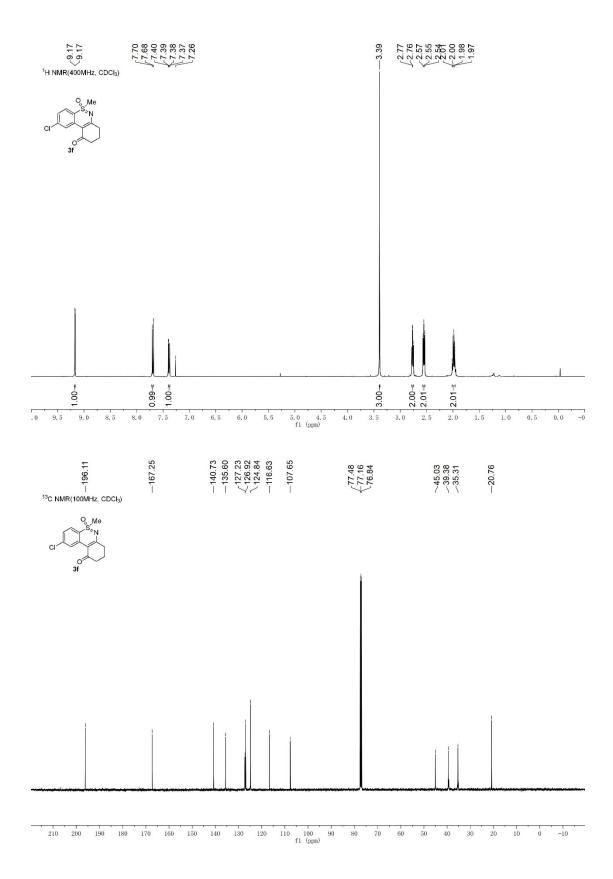
# 5<sup>1</sup>H and <sup>13</sup>C NMR Spectrum for All isolated Products

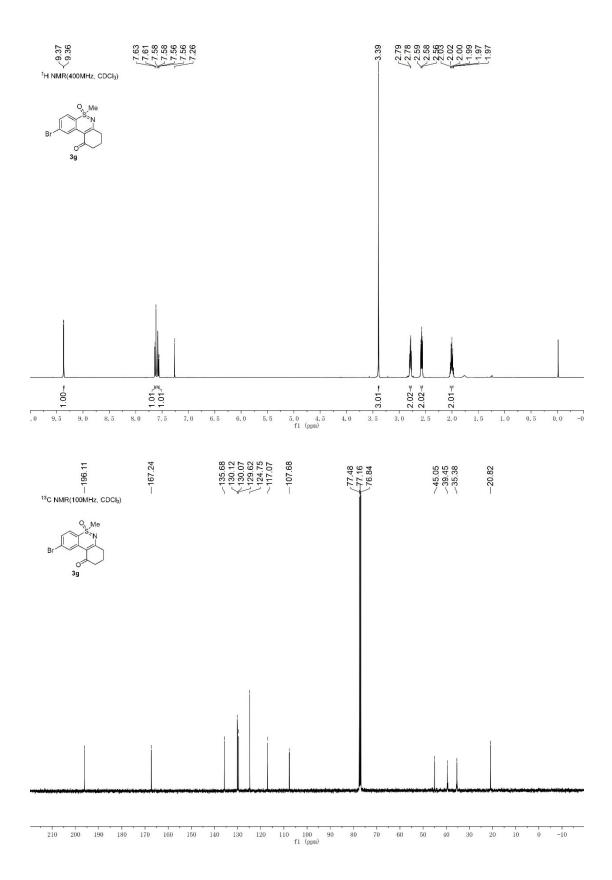


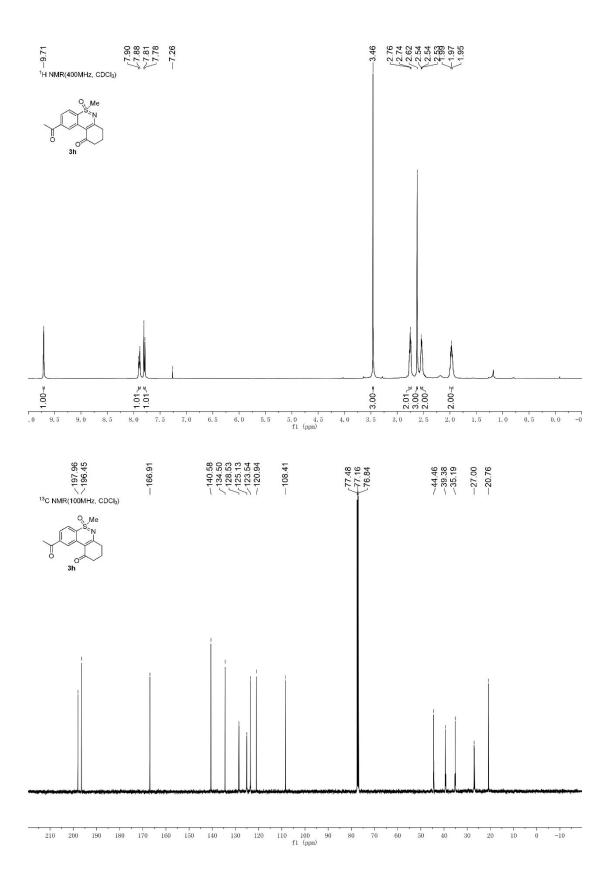


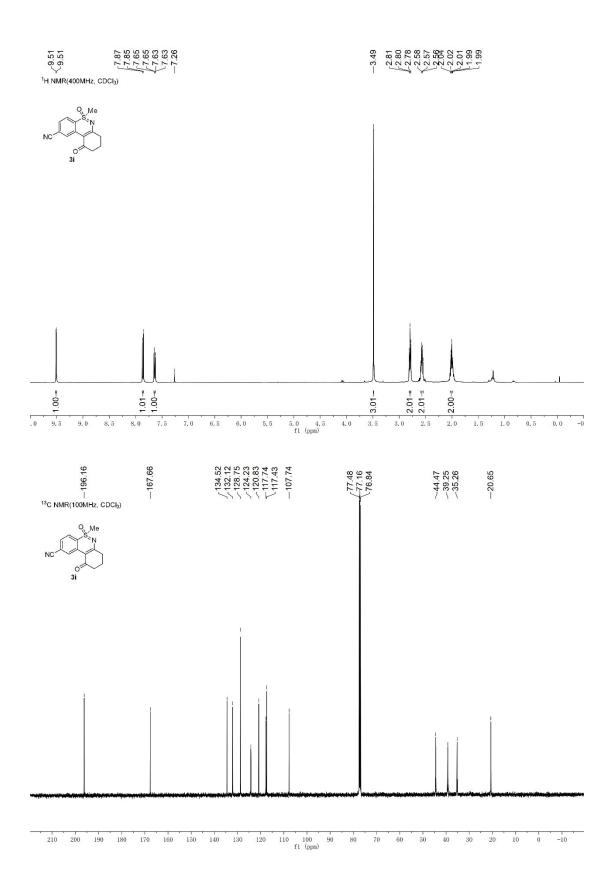


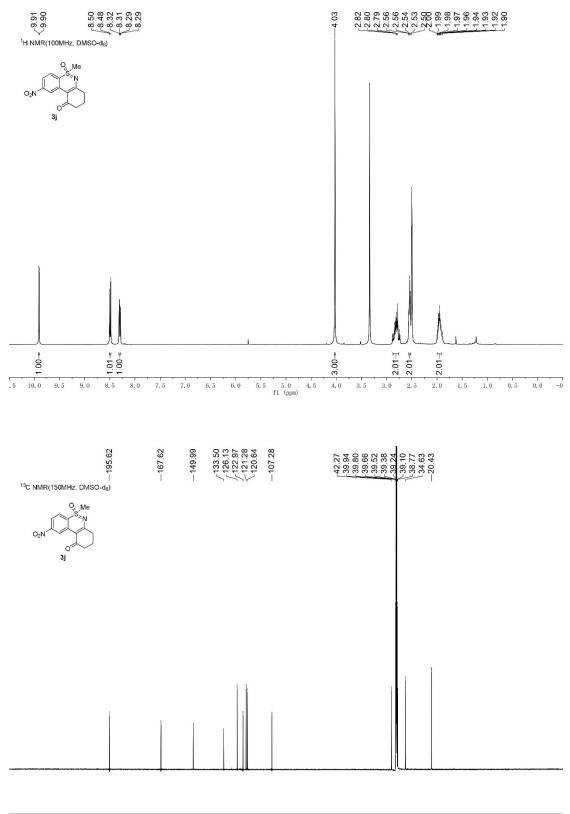




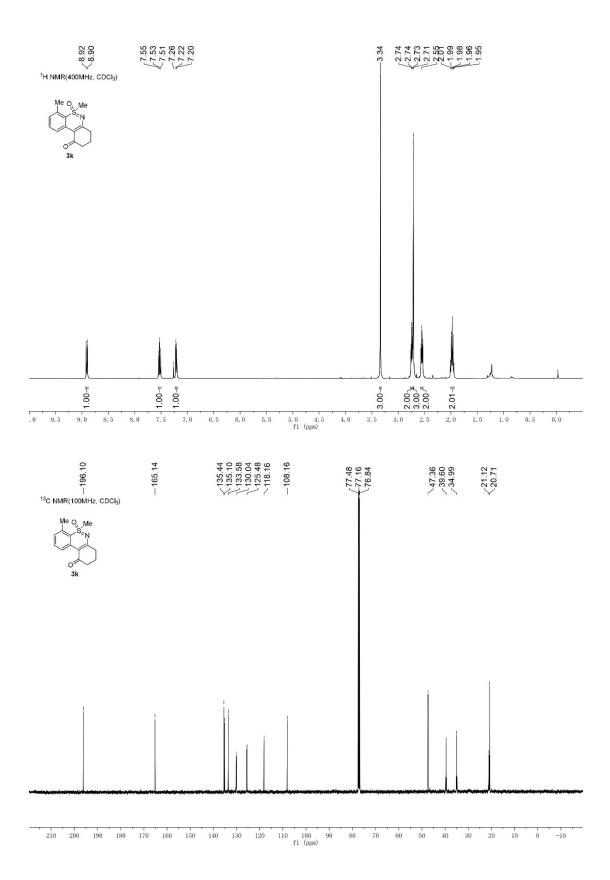


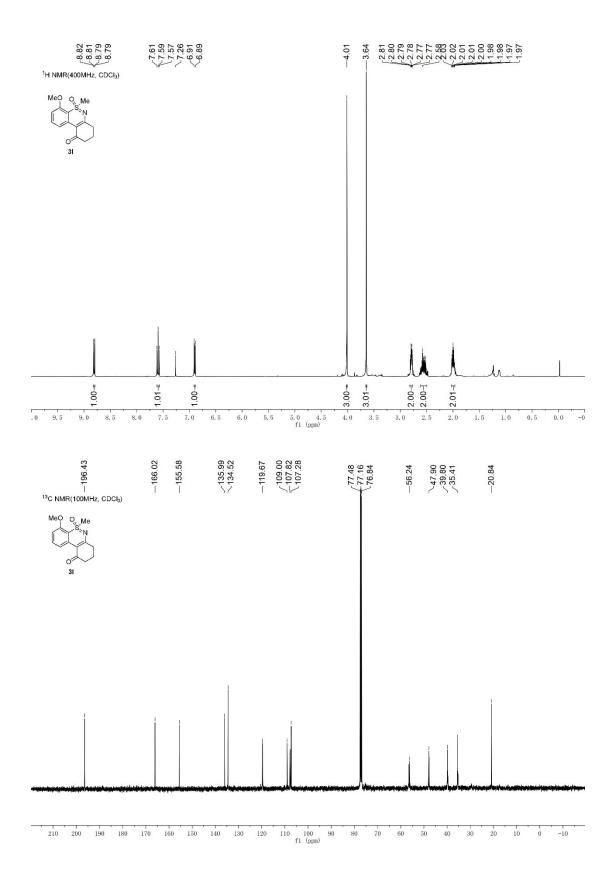


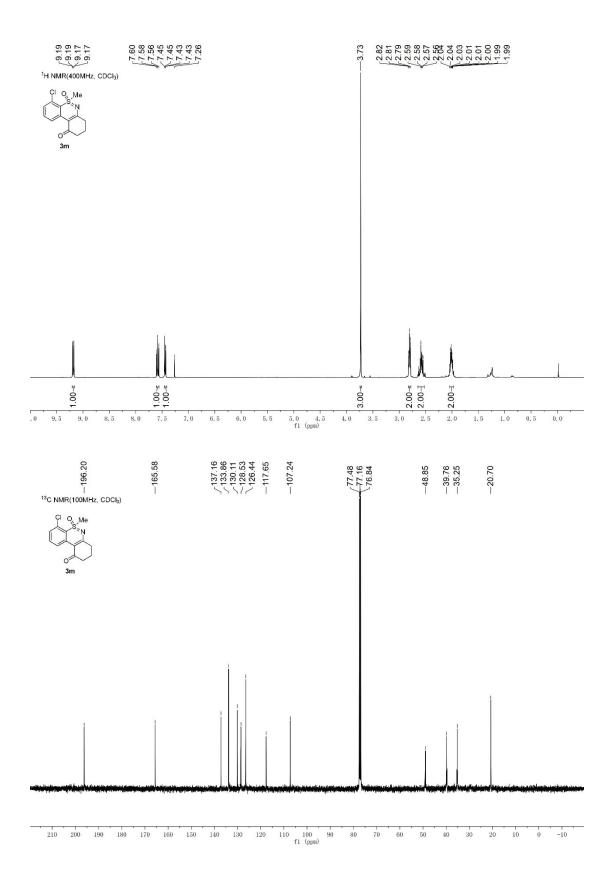


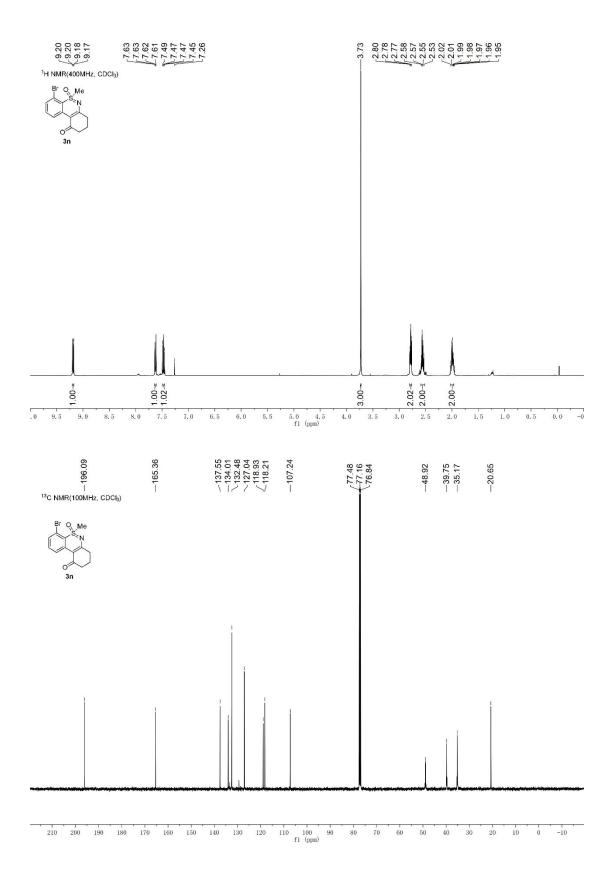


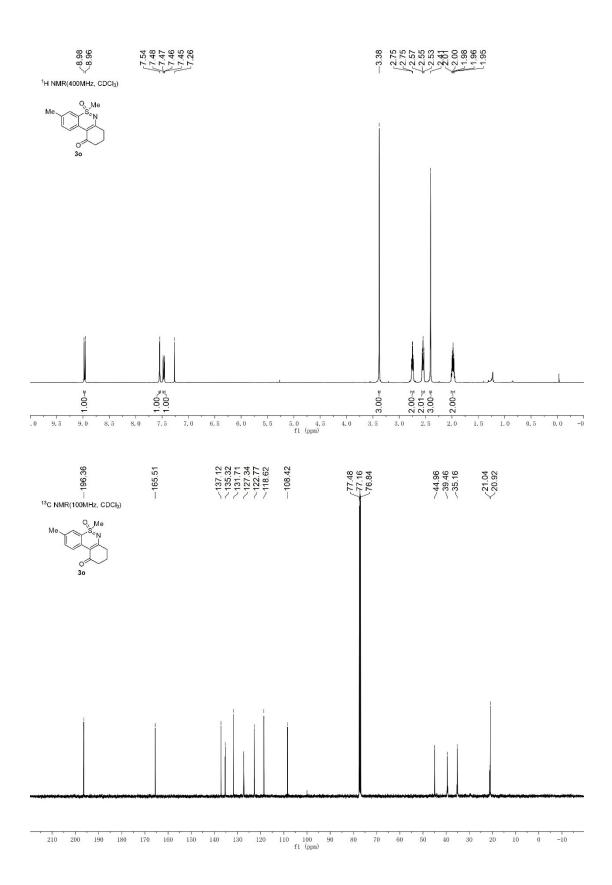
50 240 230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -5 f1 (ppm)

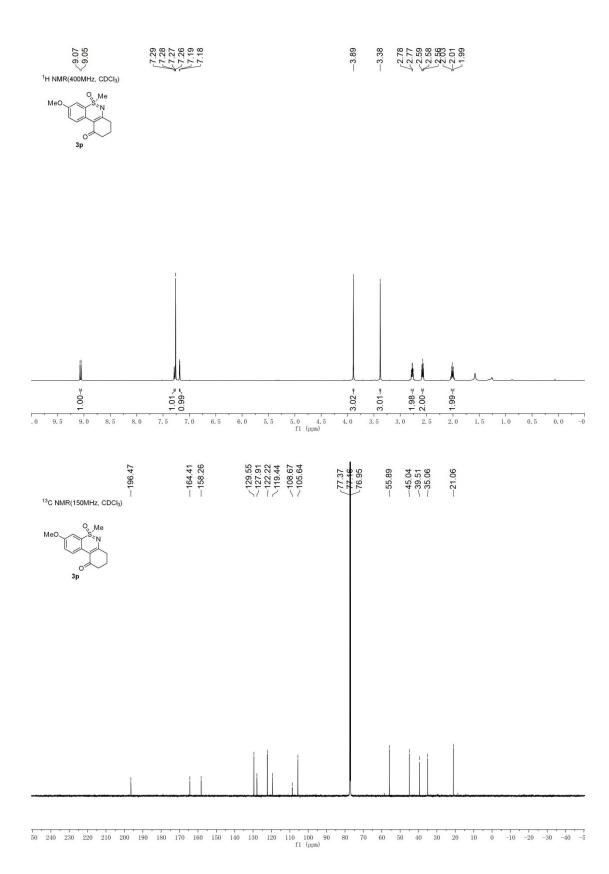


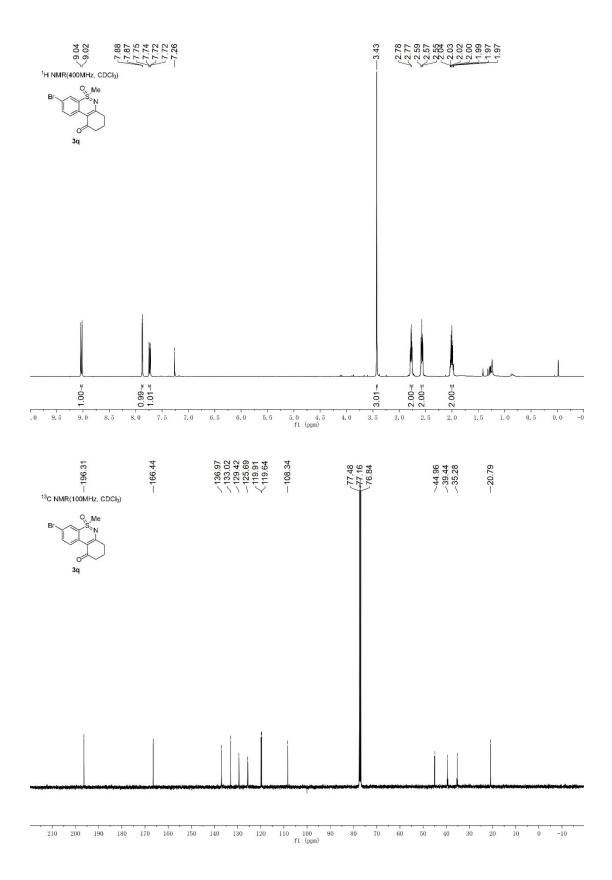


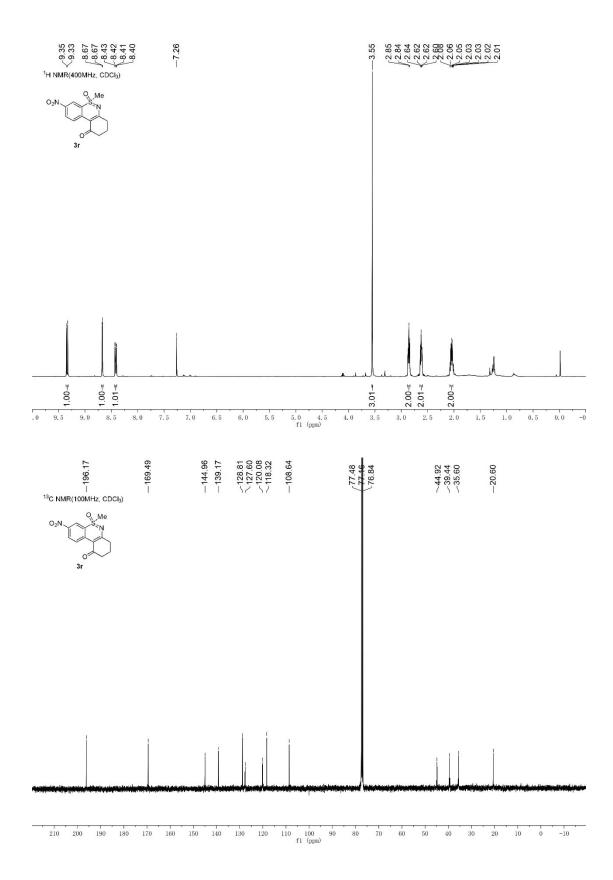


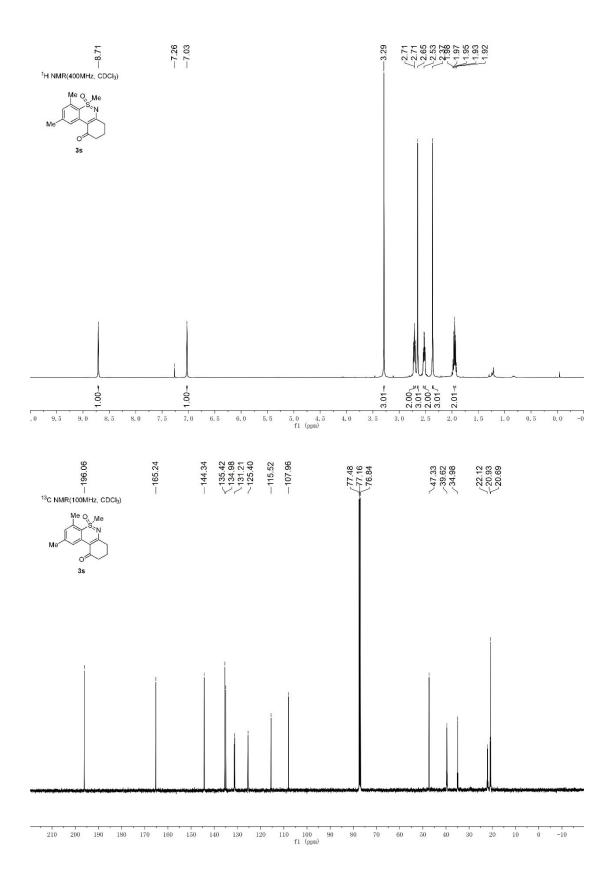


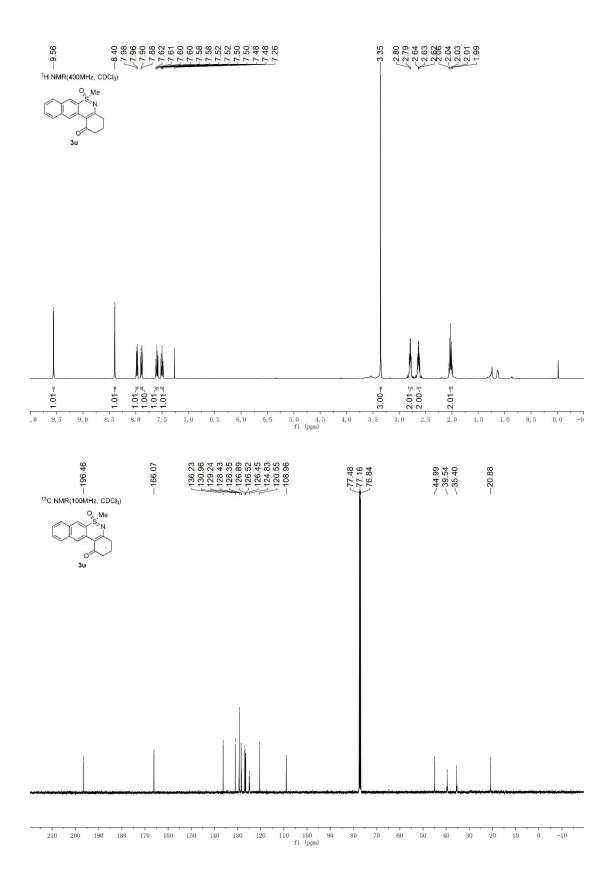


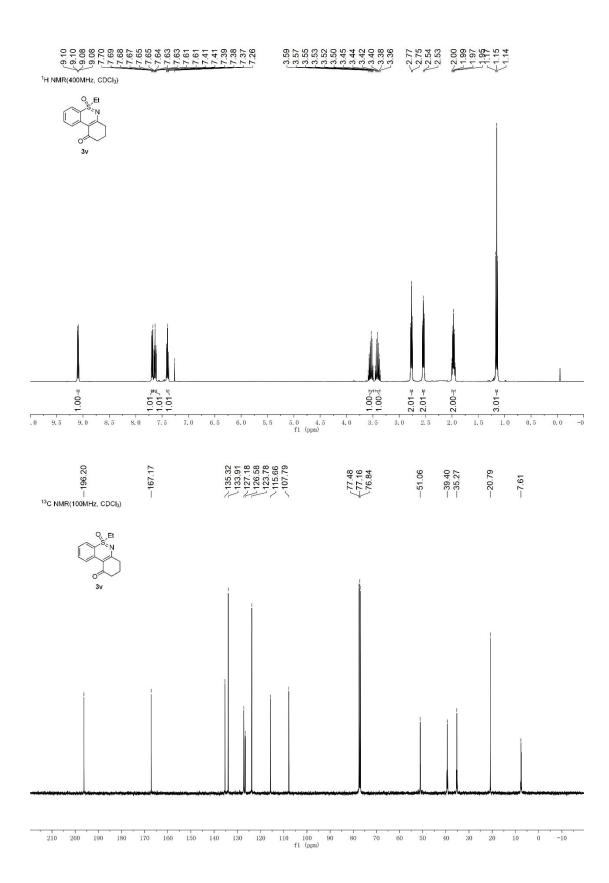




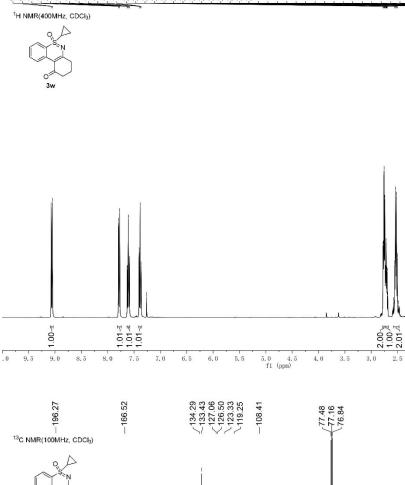


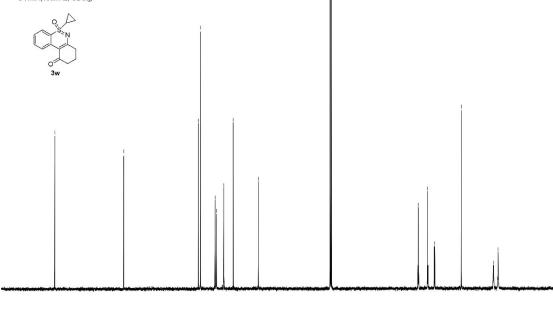






2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100
2.100





2.00 2.00 1.01 1.01

-20.82

1.0 0.5

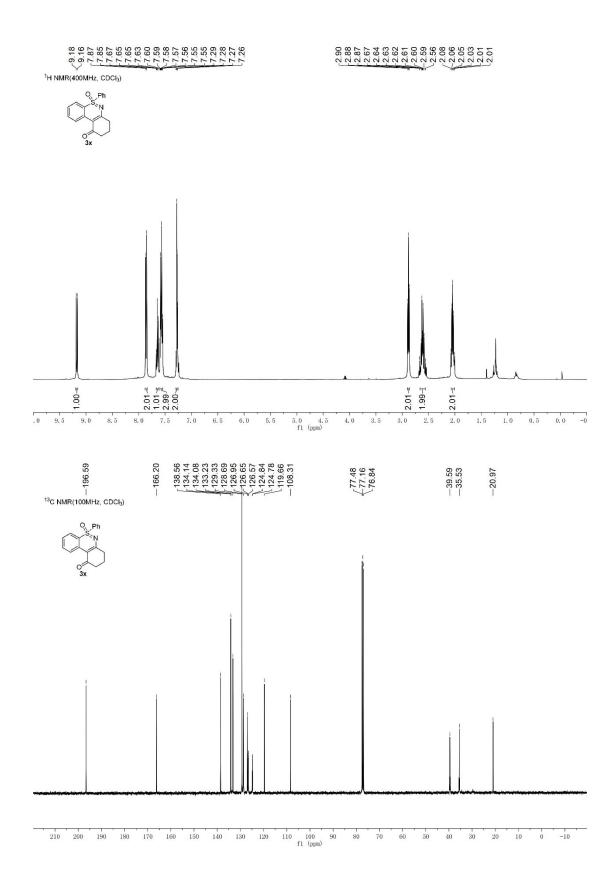
~6.93

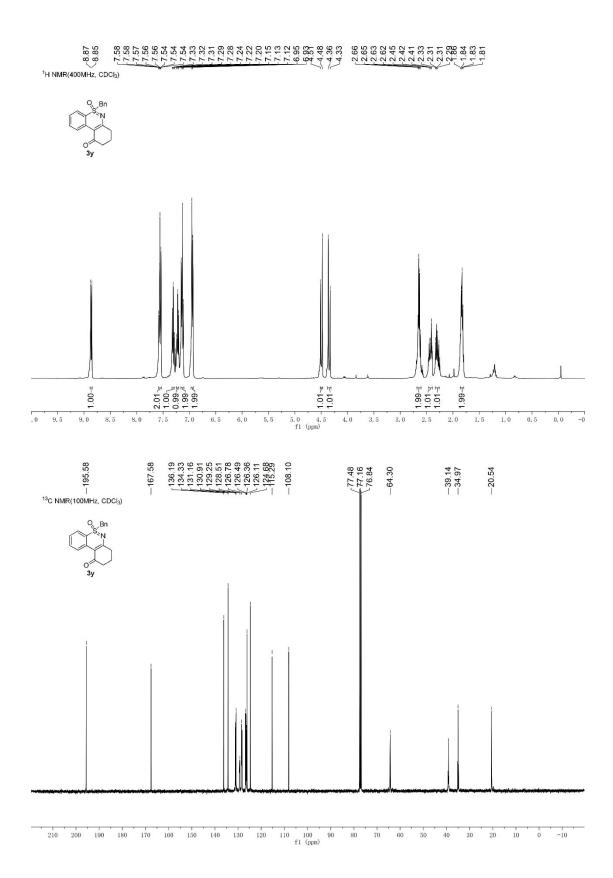
2.0

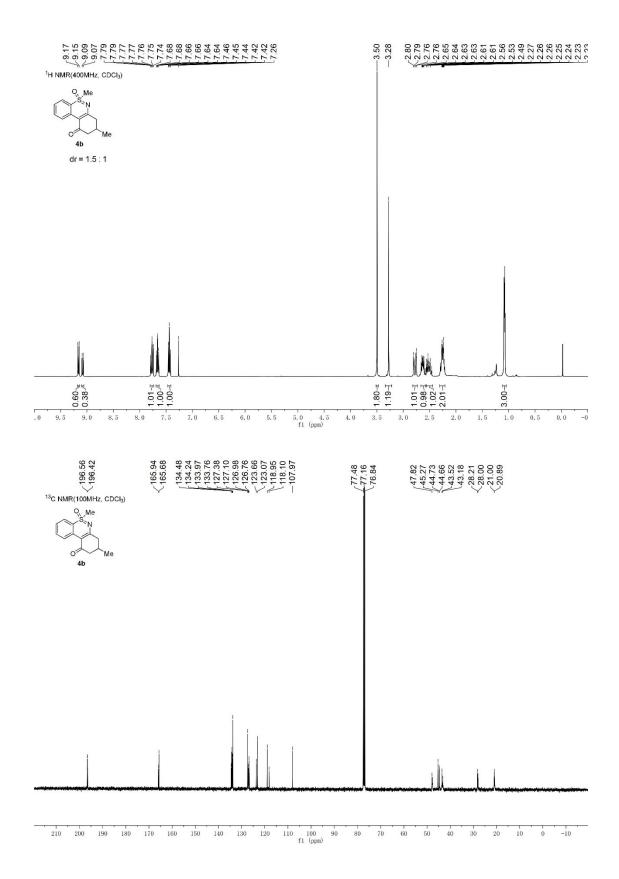
--39.41 --35.34 --32.28

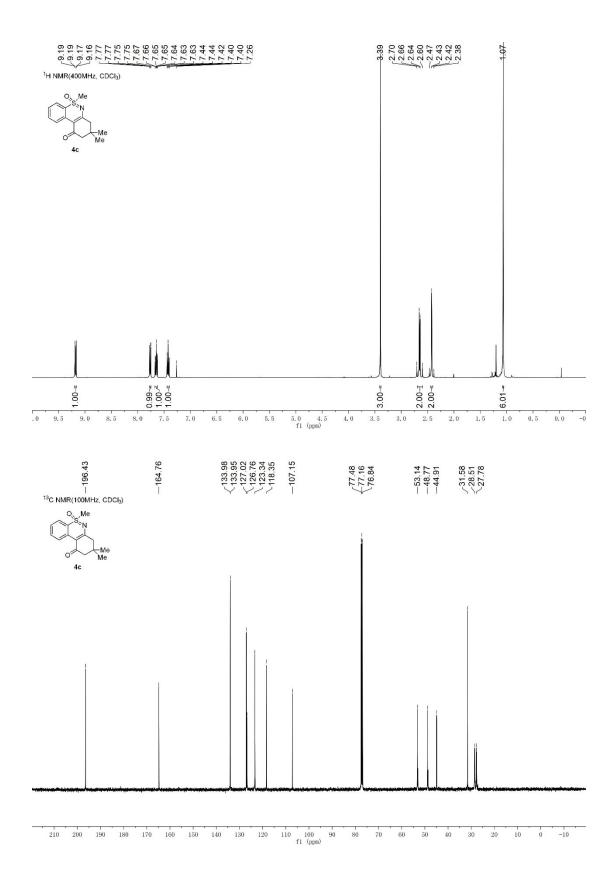
0.0 -0

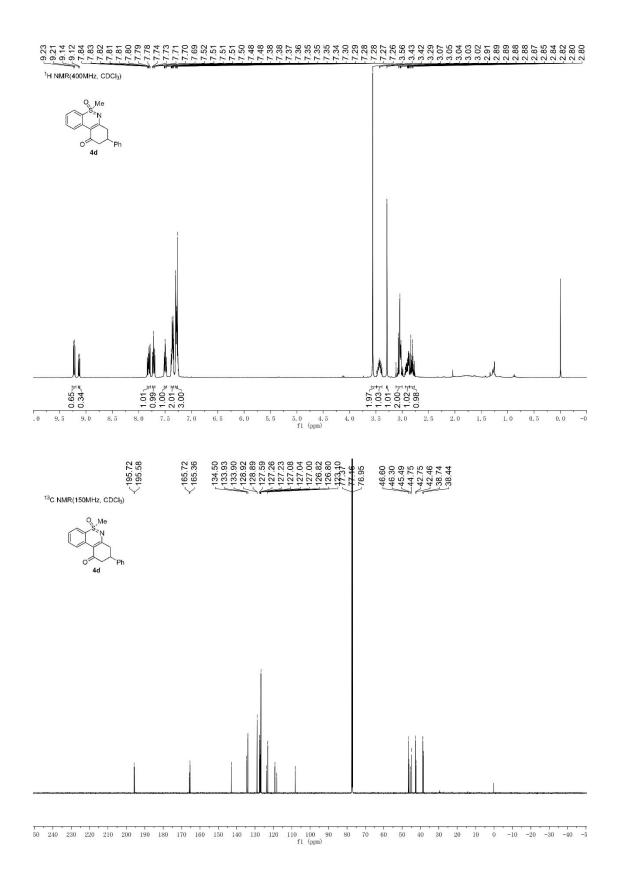
130 120 110 100 90 80 fl (ppm) 210 200 190 180 170 160 150 140 70 60 50 40 30 20 10 0 -10

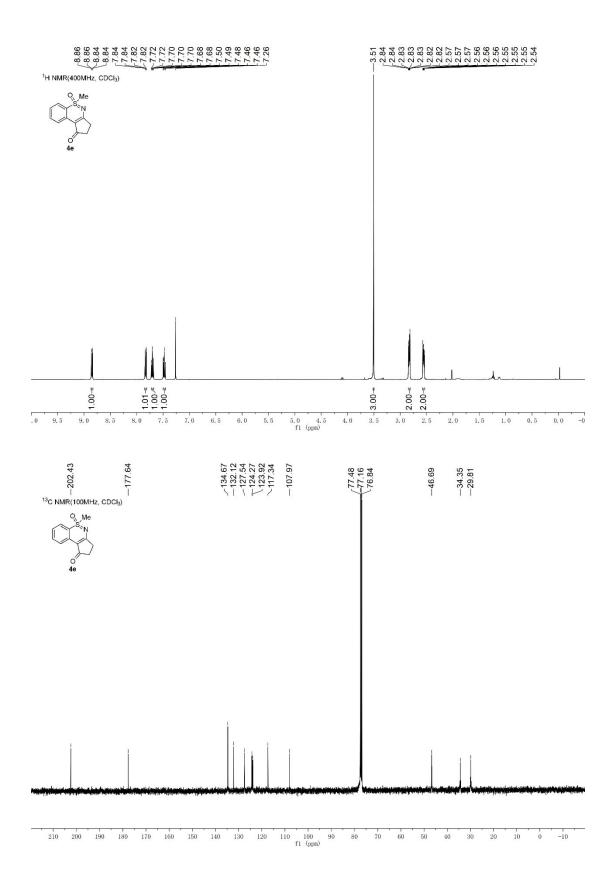


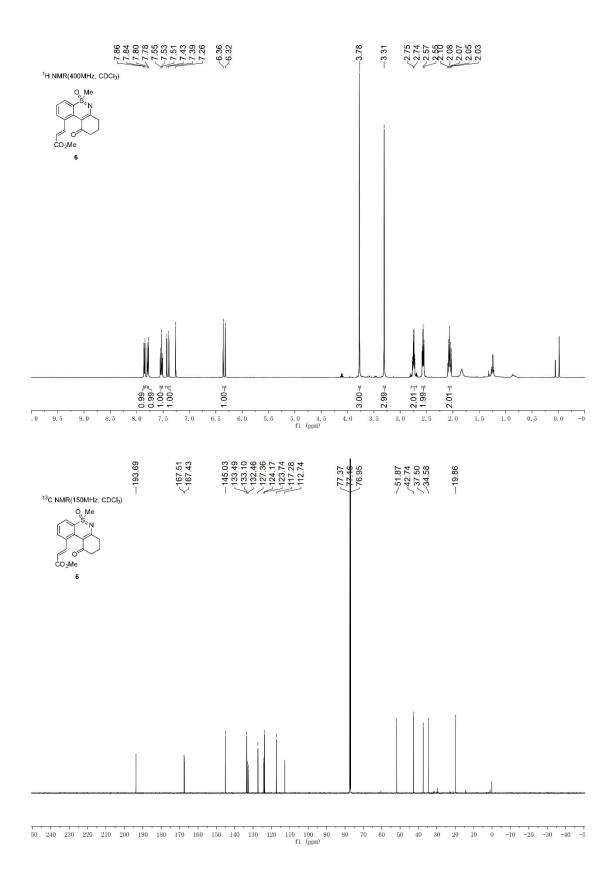












# **6** References

[1] J.-F. Lohier, T. Glachet, H. Marzag, A.-C. Gaumont and V. Reboul, *Chem. Commun.*, 2017, **53**, 2064.

[2] R. M. Moriarty, S. Tyagi, D. Ivanov and M. Constantinescu, J. Am. Chem. Soc., 2008, 130, 7564.