

Supporting Information

**Hydrogenation of Cage-Opened C₆₀ Derivatives
Mediated by Frustrated Lewis Pairs**

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1. General

The ^1H and ^{13}C NMR measurements were carried out at room temperature unless otherwise noted with a JEOL JNM ECA500 and Bruker Avance III 800US Plus instruments. The NMR chemical shifts were reported in ppm with reference to residual protons and carbons of CDCl_3 (δ 7.26 ppm in ^1H NMR, δ 77.00 ppm in ^{13}C NMR) and CD_2Cl_2 (δ 5.32 ppm in ^1H NMR, δ 53.80 ppm in ^{13}C NMR). APCI (atmospheric pressure chemical ionization) mass spectra were measured on a Bruker micrOTOF-Q II. The high-performance liquid chromatography (HPLC) was performed with the use of a Cosmosil Buckyprep column (250 mm in length, 4.6 mm in inner diameter) for analytical purpose. Thin layer chromatography (TLC) was performed on glass plates coated with 0.25 mm thick silica gel 60F-254 (Merck). Column chromatography was performed using PSQ 60B (Fuji Silysia).

Fullerene C_{60} was purchased from SES Research Co. Diethyl ether was purchased from Kanto Chemical Co., Inc. Carbon disulfide, trimethylamine, NaBH_4 , $\text{B}(\text{C}_6\text{F}_5)_3$, and acetone were purchased from FUJIFILM Wako Pure Chemical Corporation. *o*-Dichlorobenzene (ODCB) was purchased from Sigma-Aldrich Co. LLC. Triphenylsilane, pinacol borane (HBpin), and *N,N*-dimethylaniline were purchased from Tokyo Chemical Industry Co. Ltd.

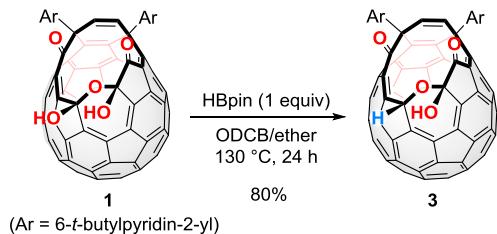
All reactions were carried out under Ar atmosphere. Unless otherwise noted, materials purchased from commercial suppliers were used without further purification. Compounds **1**,¹ **4**,² and **6**³ were synthesized according to literature procedures.

2. Computational Methods

All calculations were conducted using the Gaussian 09 program. All structures at the stationary states were optimized at the B3LYP-D3/6-31G(d) level of theory without any symmetry assumptions and confirmed by the frequency analyses at the same level of theory. Natural charges were obtained from the natural population analysis at the same level of theory.

3. Synthesis

3.1. Reduction of **1** by HBpin



Powdery **1** (10.0 mg, 8.92 μ mol) was placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL), ether (0.10 mL), and HBpin (1.30 μ L, ρ = 0.882 g/mL, 8.96 μ mol, 1.00 equiv) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel ($\text{CS}_2/\text{acetone}$ (40:1) to (20:1)) to give **3** (7.98 mg, 7.22 μ mol, 80%) followed by unreacted **1** (1.17 mg, 1.04 μ mol, 12%) as brown powders.

3: ^1H NMR (500 MHz, CDCl_3) δ 7.75 (s, 1H), 7.61 (t, 1H, J = 8.0 Hz), 7.55 (t, 1H, J = 8.0 Hz), 7.40 (d, 1H, J = 8.0 Hz), 7.21 (d, 1H, J = 8.0 Hz), 7.20 (d, 1H, J = 8.0 Hz), 7.15 (d, 1H, J = 8.0 Hz), 7.02 (d, 1H, J = 10.0 Hz), 6.86 (d, 1H, J = 10.0 Hz), 5.90 (br s, 1H), 1.24 (s, 9H), 1.16 (s, 9H); HRMS (APCI) m/z : [M] $^{+}$ Calcd for $\text{C}_{82}\text{H}_{28}\text{N}_2\text{O}_4$ (**3**) 1104.2055; Found 1104.2031. (These data were matched well with the reported one.⁴)

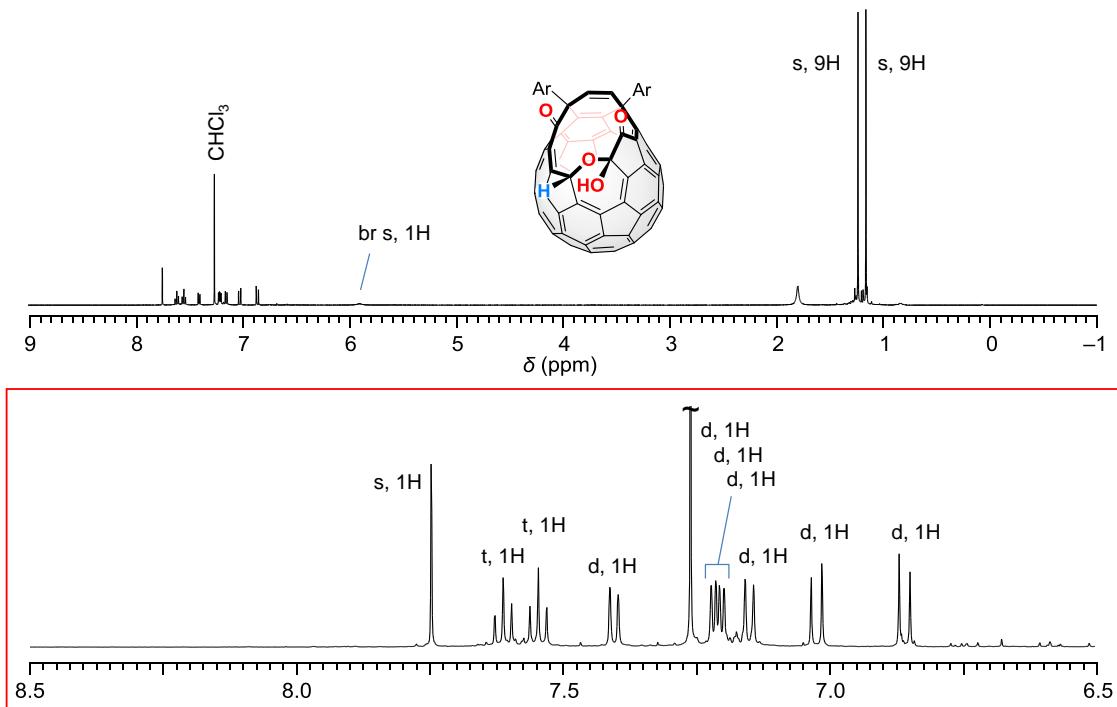


Figure S1. ^1H NMR spectra (500 MHz, CDCl_3) of **3**.

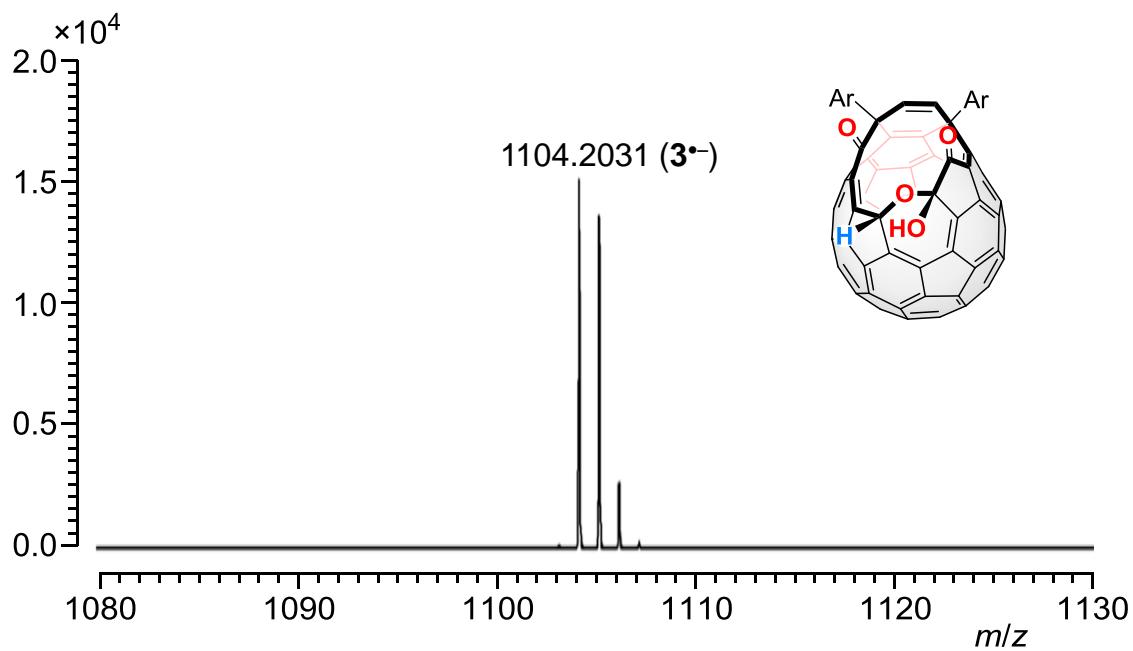
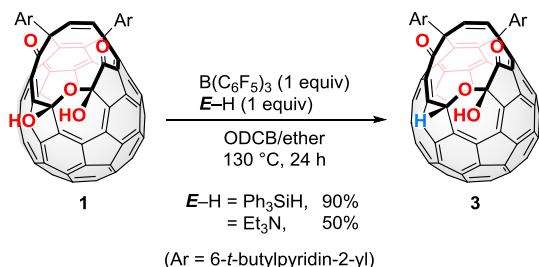


Figure 2. APCI mass spectrum (negative ion mode) of **3**.

3.2. Dehydroxyhydrogenation of **1**



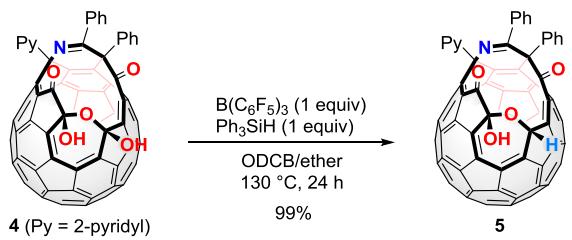
[The reaction using Ph₃SiH]

Powdery **1** (10.0 mg, 8.92 µmol) and B(C₆F₅)₃ (4.6 mg, 9.0 µmol, 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL) and ether (0.10 mL) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel (CS₂/acetone (40:1) to (20:1)) to give **3** (4.91 mg, 4.44 µmol, 50%) followed by unreacted **1** (1.03 mg, 0.919 µmol, 10%) as brown powders. **Note:** The use of crystalline B(C₆F₅)₃ is important.

[The reaction using Et₃N]

Powdery **1** (10.0 mg, 8.92 µmol) and B(C₆F₅)₃ (4.6 mg, 9.0 µmol, 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL), ether (0.10 mL), and Et₃N (1.25 µL, ρ = 0.726 g/mL, 8.97 µmol, 1.01 equiv) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel (CS₂/acetone (40:1)) to give **3** (4.91 mg, 4.44 µmol, 50%) as a brown powder. In this reaction, **1** was completely consumed. **Note:** The use of crystalline B(C₆F₅)₃ is important.

3.3. Dehydroxyhydrogenation of 4



Powdery **4** (10.3 mg, 9.49 μ mol) and B(C₆F₅)₃ (4.8 mg, 9.4 μ mol, 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL) and ether (0.10 mL) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel (CS₂/acetone (40:1)) to give **5** (10.0 mg, 9.36 μ mol, 99%) as a brown powder. **Note:** The use of crystalline B(C₆F₅)₃ is important.

5: ¹H NMR (500 MHz, CDCl₃) δ 8.56 (d, 1H, *J* = 5.2 Hz), 8.26 (d, 2H, *J* = 7.5 Hz), 7.89 (d, 1H, *J* = 8.0 Hz), 7.83–7.72 (m, 2H), 7.65 (s, 1H), 7.45–7.11 (m, 8H); HRMS (APCI) *m/z*: [M]⁺ Calcd for C₈₀H₁₆N₂O₄ (**5**) 1068.1116; Found 1068.1103. (These data were matched well with the reported one.^{2c})

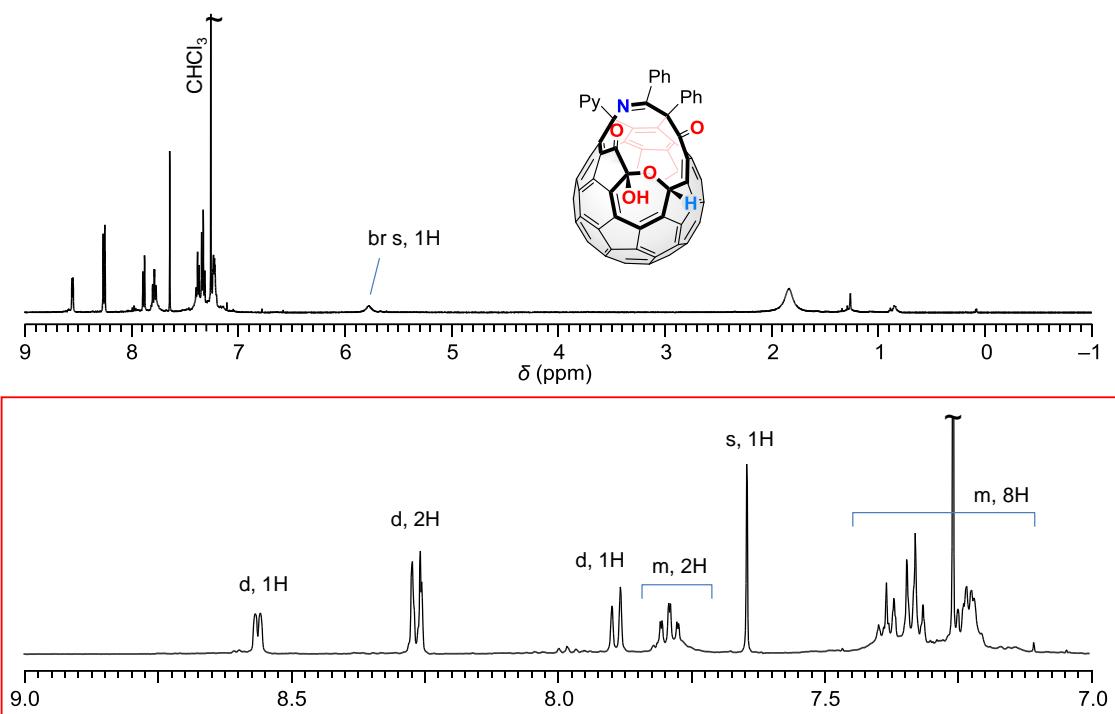


Figure S3. ^1H NMR spectra (500 MHz, CDCl_3) of **5**.

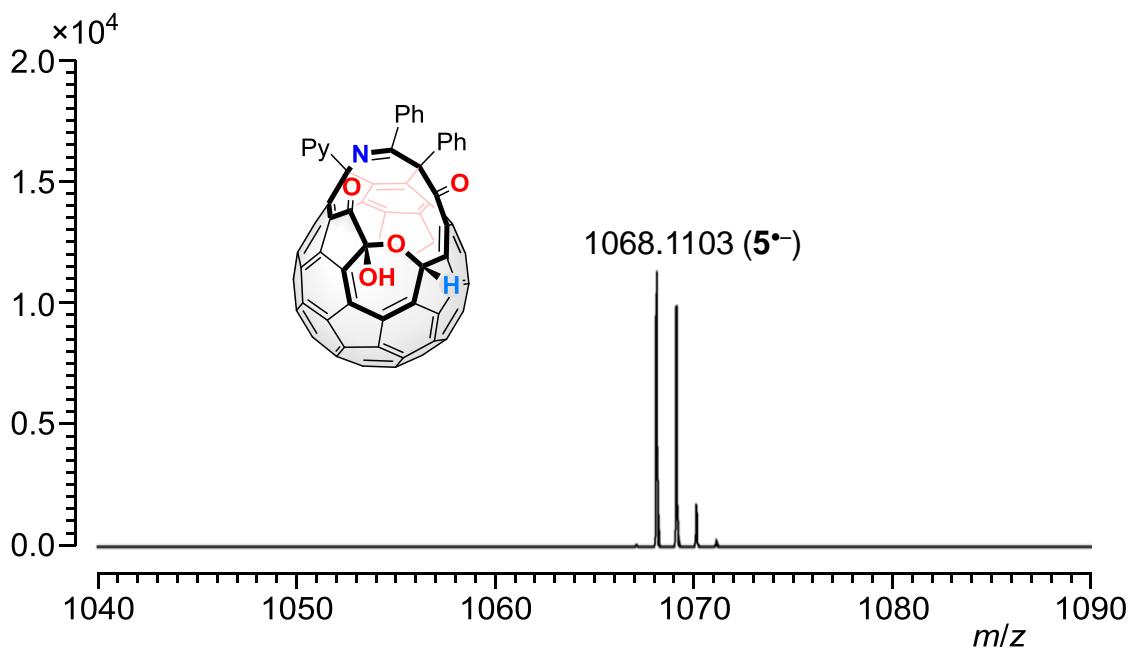
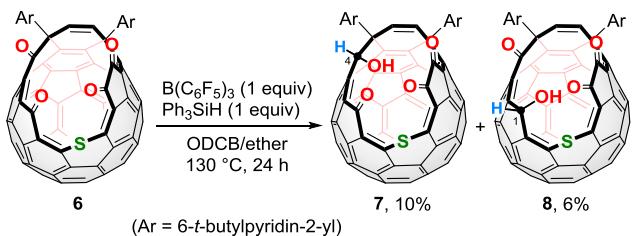


Figure 4. APCI mass spectrum (negative ion mode) of **5**.

3.4. Hydrogenation of **6**



Powdery **6** (10.0 mg, 8.81 μ mol), $B(C_6F_5)_3$ (4.6 mg, 8.9 μ mol, 1.0 equiv), and Ph_3SiH (2.3 mg, 8.8 μ mol, 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL) and ether (0.10 mL) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel ($CS_2/acetone$ (200:1) to (40:1)) to give **7** (0.97 mg, 0.85 μ mol, 10%) and **8** (0.62 mg, 0.55 μ mol, 6%), followed by unreacted **6** (7.99 mg, 7.04 μ mol, 80%) as reddish brown powders. *Note:* The use of crystalline $B(C_6F_5)_3$ is important.

7: 1H NMR (500 MHz, $CDCl_3$) δ 7.62 (t, 1H, J = 8.0 Hz), 7.39 (t, 1H, J = 8.0 Hz), 7.29 (d, 1H, J = 8.0 Hz), 7.21 (d, 1H, J = 8.0 Hz), 7.03 (d, 1H, J = 8.0 Hz), 6.88 (d, 1H, J = 10.3 Hz), 6.84 (d, 1H, J = 8.0 Hz), 6.65 (d, 1H, J = 6.9 Hz), 6.05 (d, 1H, J = 10.3 Hz), 3.50 (d, 1H, J = 6.9 Hz), 1.22 (s, 9H), 1.17 (s, 9H), -10.70 (s, 1.70H); ^{13}C NMR (201 MHz, $CDCl_3$) δ 193.55, 186.10, 184.09, 168.23, 167.81, 164.21, 161.20, 155.98, 152.57, 152.15, 152.01, 151.98, 151.01, 150.68, 150.61, 150.13, 149.77, 149.72, 149.62, 149.56, 149.44, 149.07, 147.96, 147.25, 147.04, 145.99, 145.60, 145.01, 144.06, 143.86, 143.78, 143.24, 143.20, 143.15, 143.11, 142.70, 142.30, 142.00, 141.74, 140.93, 140.20, 139.45, 139.27, 138.99, 138.02, 137.70, 137.57, 137.47, 137.44, 136.95, 136.92, 136.79, 136.60, 136.49, 136.46, 133.55, 132.28, 131.93, 131.06, 130.83, 128.74, 126.91, 125.15, 121.99, 120.91, 120.01, 119.82, 119.67, 117.48, 116.99, 116.82, 71.92, 62.19, 54.76, 37.58, 37.10, 29.90, 29.88 (The sum of carbon signals must be 78 in theory. Observed 78.); HRMS (APCI) m/z : [M] $^-$ Calcd for $C_{82}H_{28}N_2O_4S$ (**7**) 1136.1775; Found 1136.1733.

8: 1H NMR (500 MHz, $CDCl_3$) δ 7.59 (t, 1H, J = 8.0 Hz), 7.50 (t, 1H, J = 8.0 Hz), 7.40 (d, 1H, J = 4.6 Hz), 7.24 (d, 1H, J = 8.0 Hz), 7.22 (d, 1H, J = 8.0 Hz), 7.21 (d, 1H, J = 10.3 Hz), 7.20 (d, 1H, J = 8.0 Hz), 7.11 (d, 1H, J = 8.0 Hz), 6.65 (d, 1H, J = 10.3 Hz), 3.85 (d, 1H, J = 4.6 Hz), 1.20 (s, 9H), 1.10 (s, 9H), -11.06 (s, 1.70H); HRMS (APCI) m/z : [M] $^-$ Calcd for $C_{82}H_{28}N_2O_4S$ (**8**) 1136.1775; Found 1136.1777. (These data were matched well with the reported one.⁵)

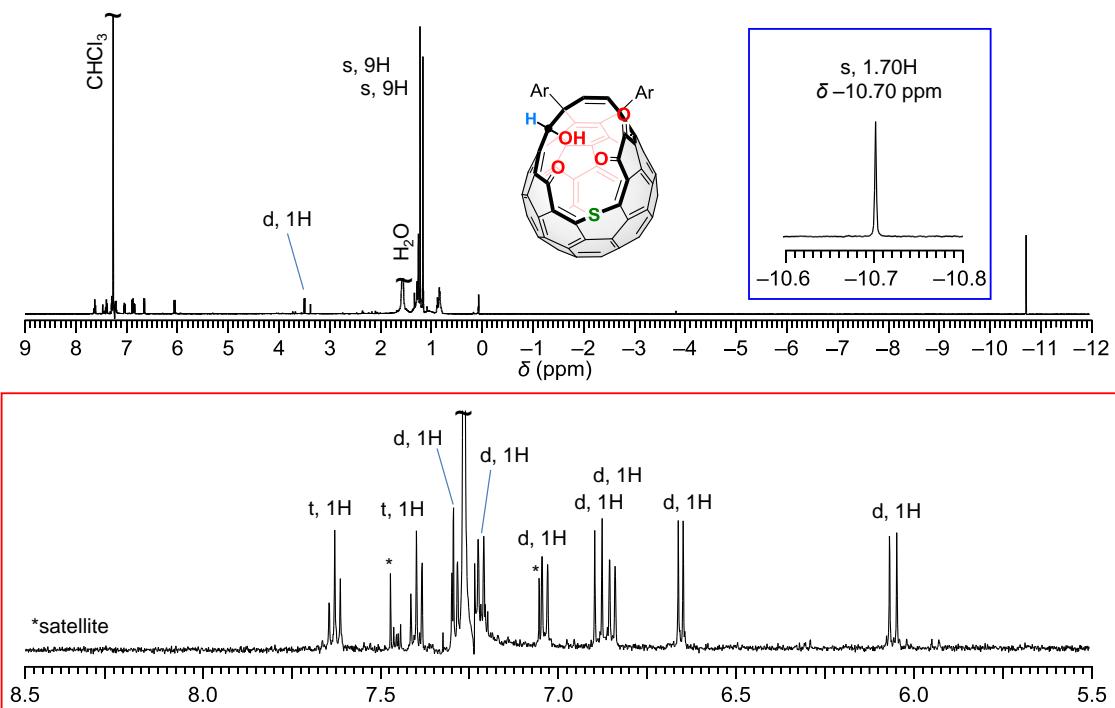


Figure S5. ^1H NMR spectra (500 MHz, CDCl_3) of **7**.

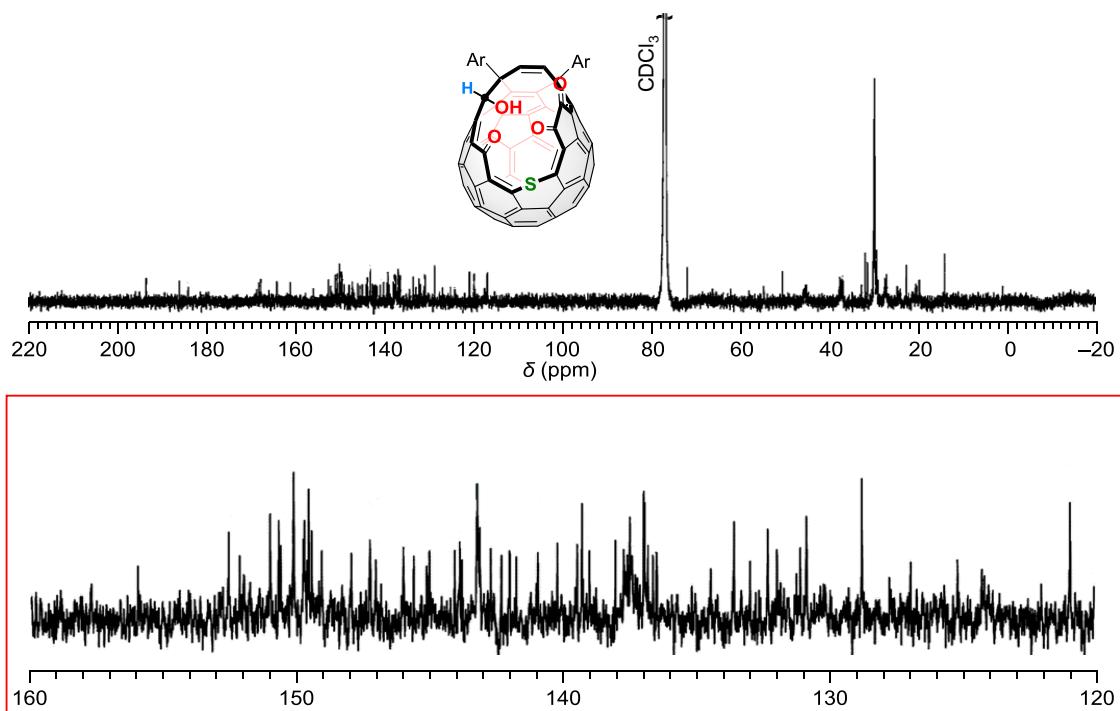


Figure S6. ^{13}C NMR spectra (201 MHz, CDCl_3) of **7**.

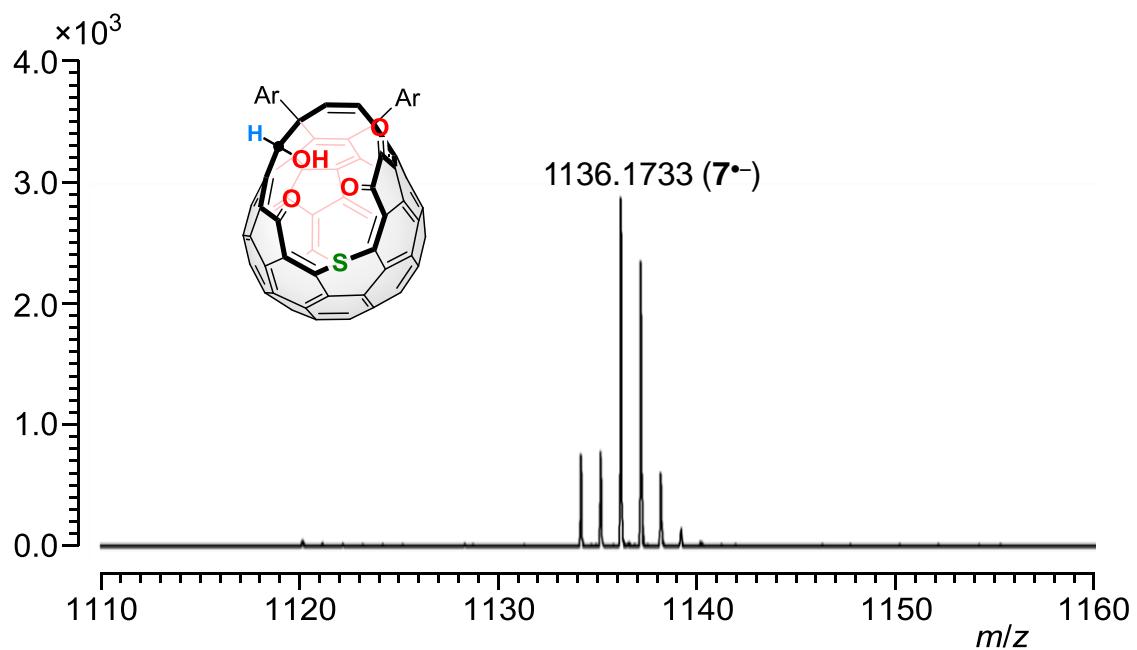


Figure 7. APCI mass spectrum (negative ion mode) of **7**.

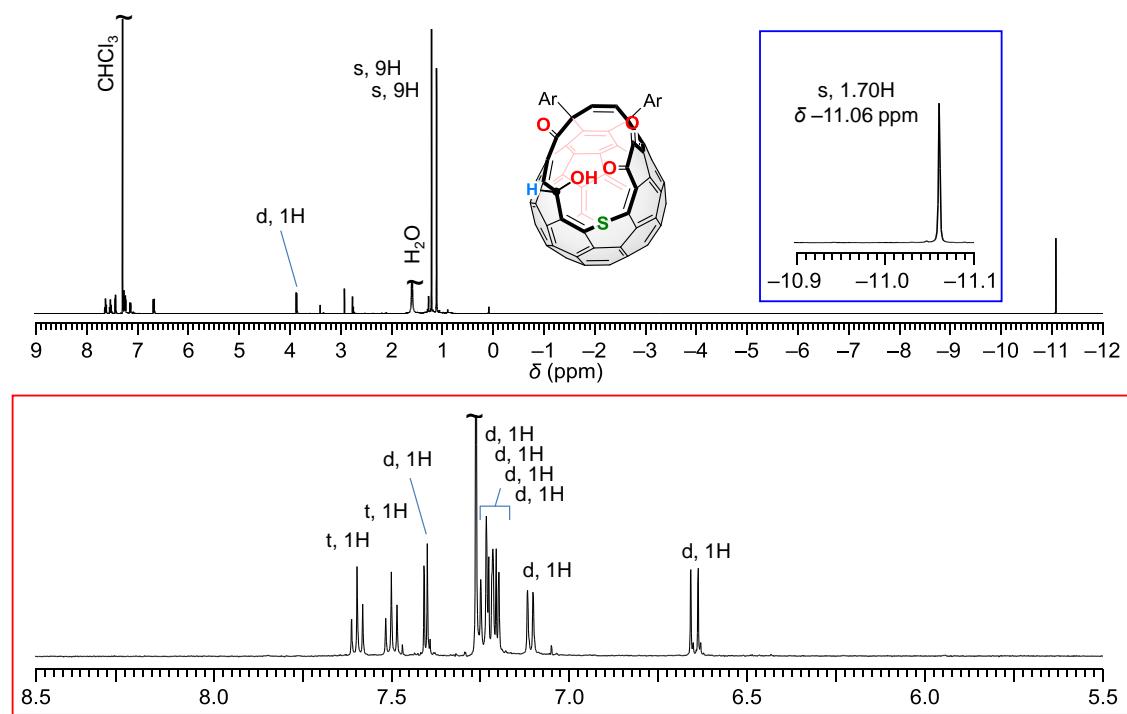


Figure S8. ^1H NMR spectra (500 MHz, CDCl_3) of **8**.

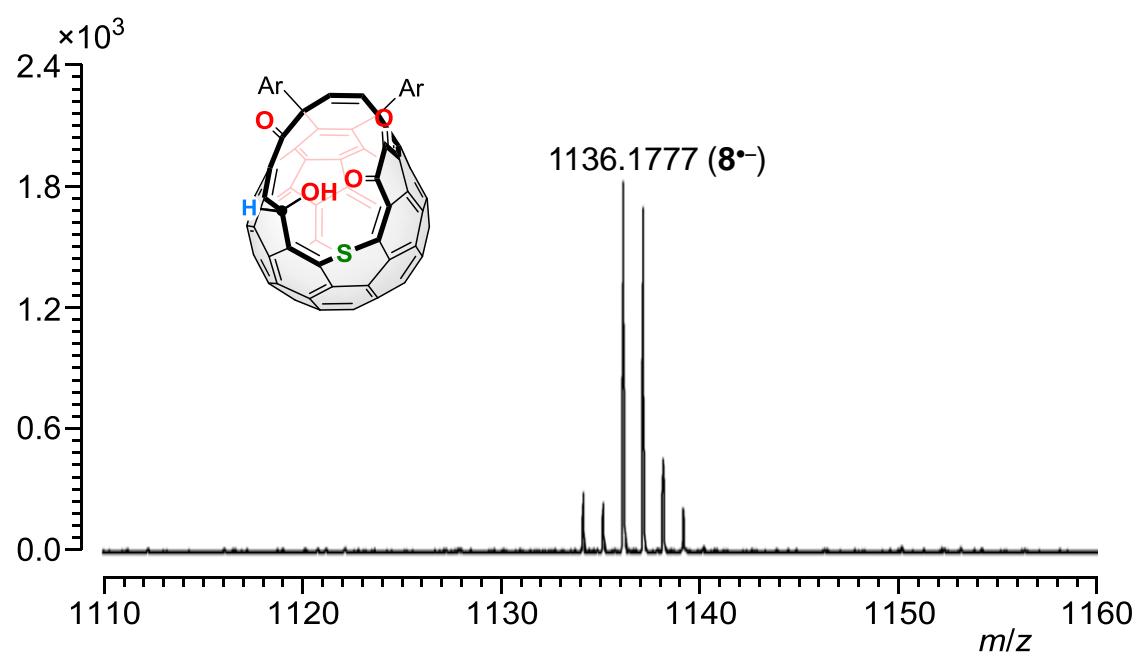
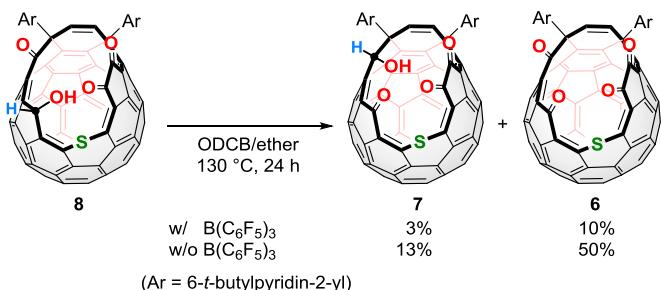


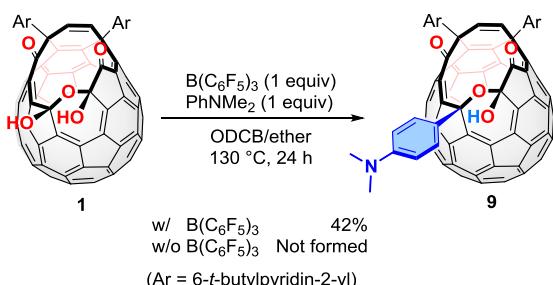
Figure 9. APCI mass spectrum (negative ion mode) of **8**.

3.5. Conversion of **8** into **7**



Powdery **8** (8.0 mg, 7.0 μmol) and $\text{B}(\text{C}_6\text{F}_5)_3$ (3.7 mg, 7.2 μmol , 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL) and ether (0.10 mL) were added and the resulting solution was heated at 130 °C for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel ($\text{CS}_2/\text{acetone}$ (200:1) to (100:1)) to give **7** (0.21 mg, 0.18 μmol , 3%), unreacted **8** (6.74 mg, 5.93 μmol , 84%), and **6** (0.80 mg, 0.70 μmol , 10%) as reddish brown powders.

3.6. Reductive Arylation of **1**



Powdery **1** (10.0 mg, 8.92 μmol) and $\text{B}(\text{C}_6\text{F}_5)_3$ (4.6 mg, 9.0 μmol , 1.0 equiv) were placed into a Schlenk tube and degassed through three vacuum-Ar cycles. ODCB (1.00 mL), ether (0.10 mL), and PhNMe_2 (1.14 μL , $\rho = 0.956 \text{ g/mL}$, 8.99 μmol , 1.01 equiv) were added and the resulting solution was heated at 130 $^\circ\text{C}$ for 24 h (Aluminum block heater). The crude mixture was purified by column chromatography using silica gel ($\text{CS}_2/\text{acetone}$ (40:1) to (20:1)) to give **9** (4.57 mg, 3.73 μmol , 42%) followed by unreacted **1** (5.89 mg, 5.25 μmol , 59%) as brown powders. **Note:** The use of crystalline $\text{B}(\text{C}_6\text{F}_5)_3$ is important.

9: ^1H NMR (500 MHz, CD_2Cl_2) δ 8.35 (d, 2H, $J = 8.5 \text{ Hz}$), 7.68 (t, 1H, $J = 8.0 \text{ Hz}$), 7.57 (t, 1H, $J = 8.0 \text{ Hz}$), 7.49 (d, 1H, 8.0 Hz), 7.26 (d, 1H, 8.0 Hz), 7.22 (d, 1H, 8.0 Hz), 7.18 (d, 1H, 8.0 Hz), 7.05 (d, 1H, $J = 10.3 \text{ Hz}$), 6.87 (d, 2H, 8.5 Hz), 6.86 (d, 1H, 10.3 Hz), 4.45 (s, 1H), 3.05 (s, 6H), 1.24 (s, 9H), 1.16 (s, 9H); ^{13}C NMR (201 MHz, CD_2Cl_2) δ 197.47, 194.49, 169.36, 168.73, 164.68, 163.19, 154.64, 153.18, 151.29, 150.26, 150.11, 149.85, 149.68, 149.56, 149.50, 149.08, 148.83, 148.76, 148.39, 148.18, 147.97, 147.91, 147.73, 147.53, 147.27, 147.05, 146.59, 146.09, 146.01, 145.80, 145.63, 144.44, 144.41, 143.91, 143.84, 143.76, 143.15, 142.88, 141.83, 141.70, 141.66, 141.25, 141.12, 140.79, 139.98, 139.16, 138.53, 137.69, 137.59, 137.54, 137.27, 137.21, 137.15, 136.71, 136.31, 135.80, 135.34, 134.14, 133.98, 133.37, 132.82, 132.15, 132.13, 130.71, 130.48, 128.52, 128.45, 127.31, 127.25, 120.39, 119.94, 117.80, 117.22, 111.66, 98.73, 96.90, 60.09, 55.16, 40.53, 38.11, 37.84, 30.13, 29.94 (The sum of carbon signals must be 83 in theory. Observed 83.); HRMS (APCI) m/z : [M] $^+$ Calcd for $\text{C}_{90}\text{H}_{37}\text{N}_3\text{O}_4$ (**9**) 1223.2790; Found 1223.2755.

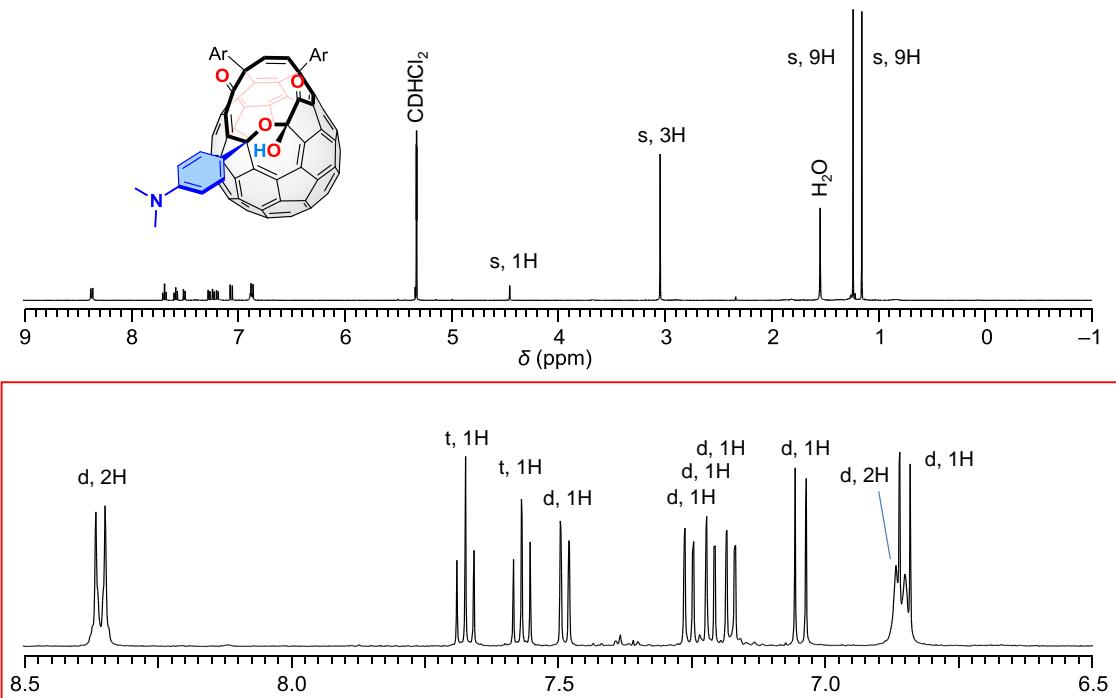


Figure S10. ^1H NMR spectra (500 MHz, CD₂Cl₂) of **9**.

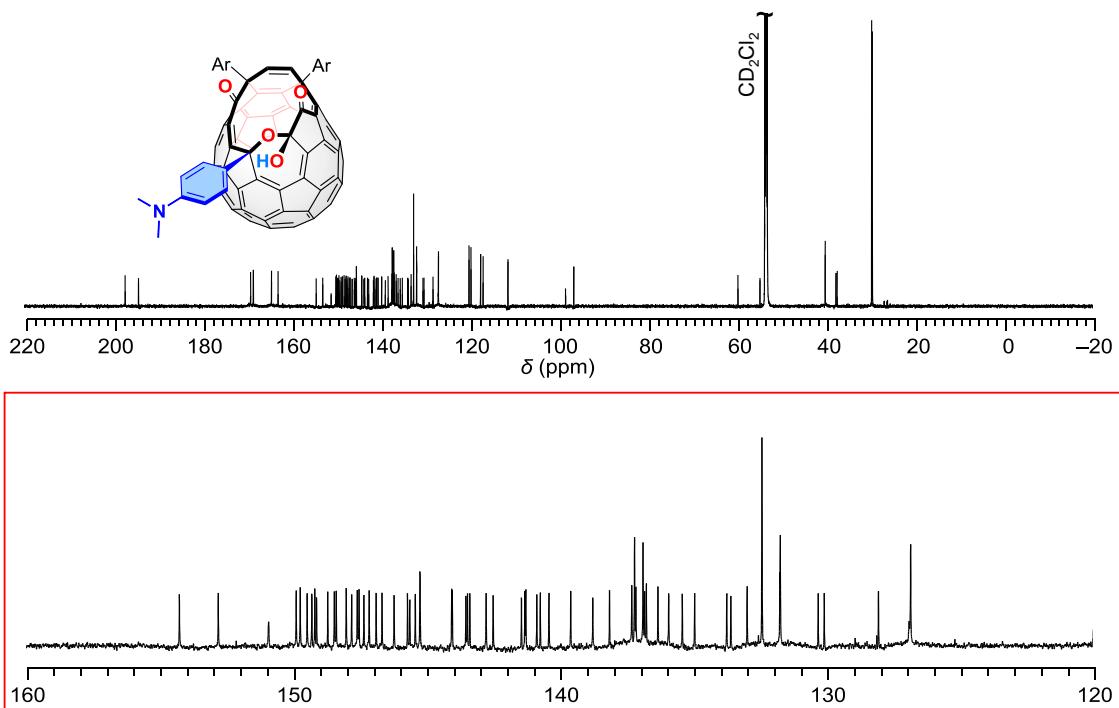


Figure S11. ^{13}C NMR spectra (201 MHz, CD₂Cl₂) of **9**.

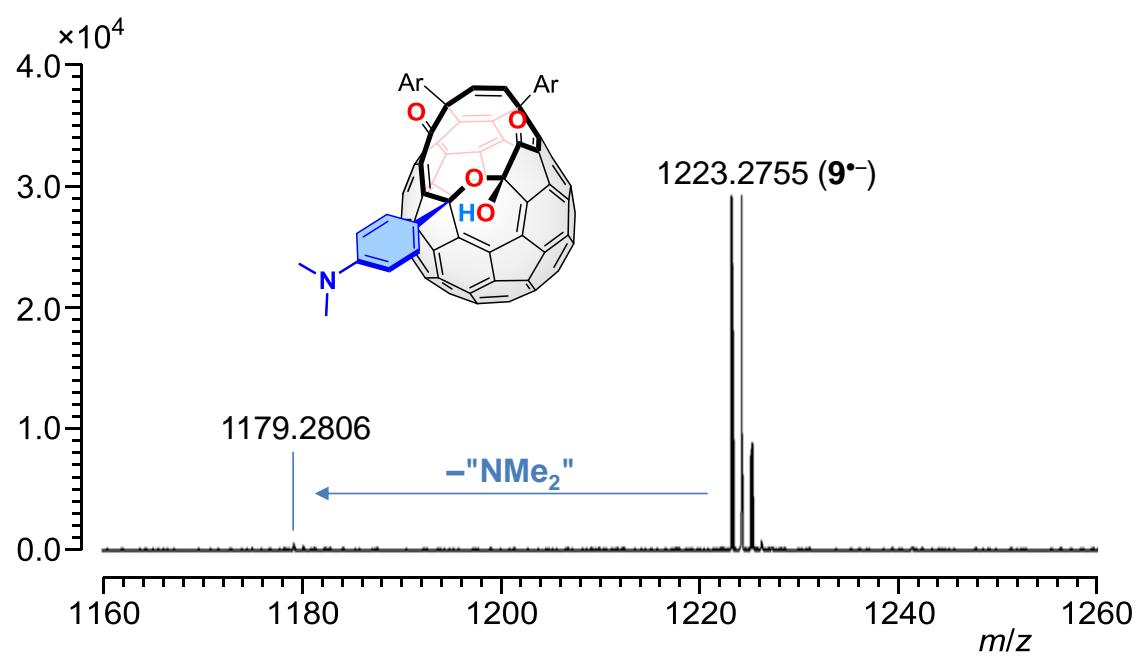


Figure 12. APCI mass spectrum (negative ion mode) of **9**.

4. Single Crystal X-Ray-Structures of $(\text{H}_2\text{O}@\mathbf{5})\cdot\text{CS}_2$

Single crystals of $\text{H}_2\text{O}@\mathbf{5}$ were obtained from a CS_2 solution by slow evaporation. Intensity data were collected at 100 K on a Bruker Single Crystal CCD X-ray Diffractometer (SMART APEX II) with Mo $\text{K}\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$) and graphite monochromater. A total of 5401 reflections were measured at the maximum 2θ angle of 50.05° , of which 4814 were independent reflections ($R_{\text{int}} = 0.0410$). The structure was solved by direct methods (SHELXT-2014/5⁶) and refined by the full-matrix least-squares on F^2 (SHELXL-2018/3⁶). All non-hydrogen atoms were refined anisotropically. All hydrogen atoms were placed using AFIX instructions except for the encapsulated H_2O molecule. A part of the fullerene skeleton was refined using SIMU and ISOR instructions. The crystal data are as follows: $\text{C}_{81}\text{H}_{18}\text{N}_2\text{O}_5\text{S}_2$; FW = 1163.09, crystal size $0.21 \times 0.08 \times 0.04 \text{ mm}^3$, monoclinic, $C2/c$, $a = 30.736(8) \text{ \AA}$, $b = 9.788(2) \text{ \AA}$, $c = 32.682(8) \text{ \AA}$, $\beta = 107.241(3)^\circ$, $V = 9391(4) \text{ \AA}^3$, $Z = 8$, $D_c = 1.645 \text{ g cm}^{-3}$. The refinement converged to $R_1 = 0.1154$, $wR_2 = 0.2766$ ($I > 2\sigma(I)$), GOF = 1.084. The data was deposited at the Cambridge Crystallographic Data Centre (CCDC 2088803).

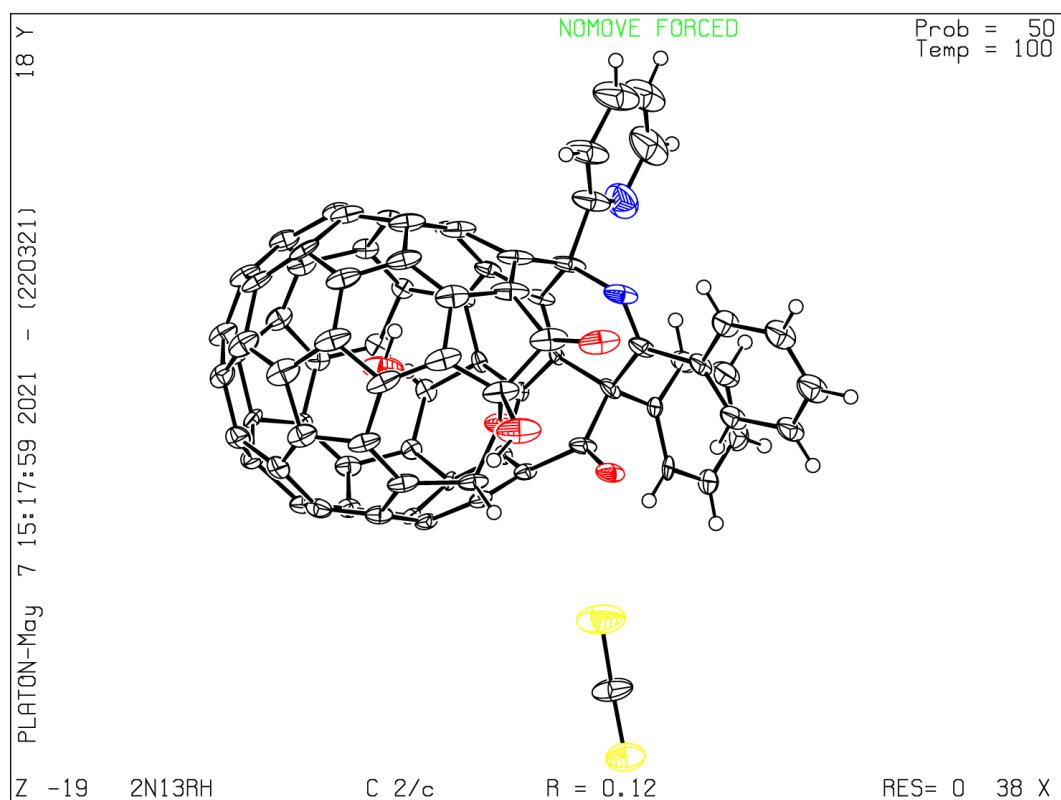


Figure S13. Single crystal X-ray structure of $(\text{H}_2\text{O}@\mathbf{5})\cdot\text{CS}_2$. Thermal ellipsoids are shown at 50% probability.

5. Natural Charges

For the calculations to be simplified, 6-*t*-butylpyridin-2-yl groups were replaced with 2-pyridyl groups for **2'** and **6'**. The calculations were performed at the B3LYP-D3/6-31G(d) level of theory using Gaussian 09.⁷

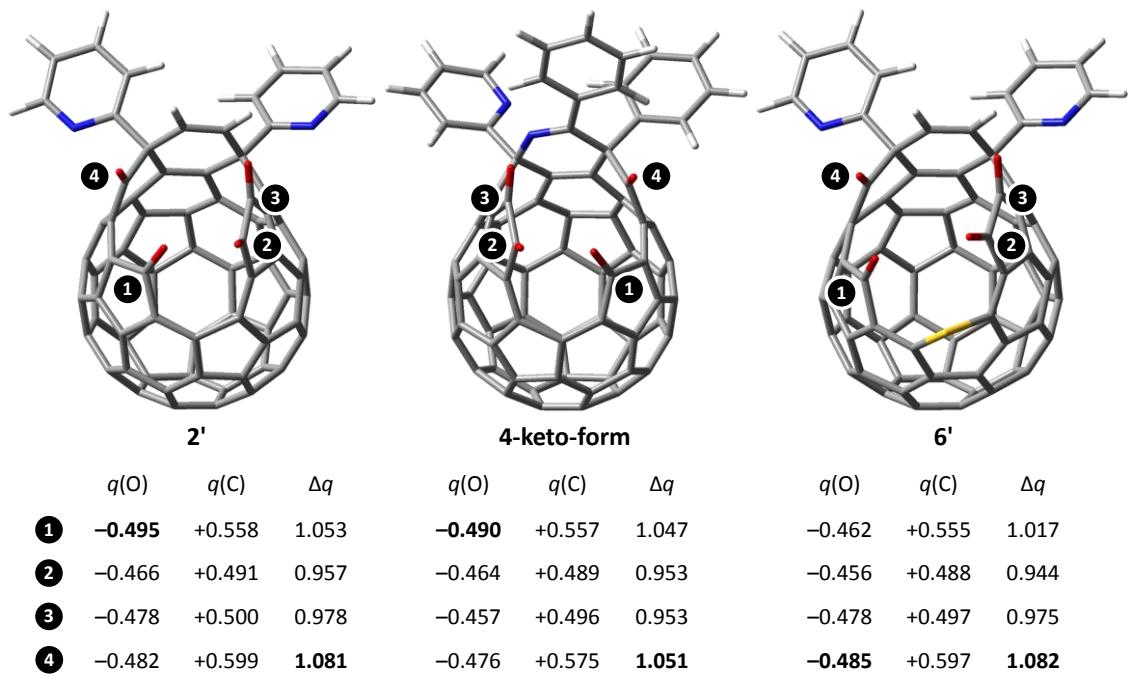
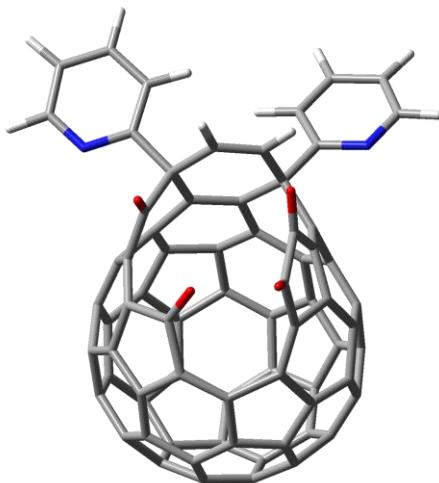


Figure S14. Natural charges q of **2'**, **4-keto-form**, and **6'** (B3LYP-D3/6-31G(d)).

Table S1. Optimized structure of **2'** (B3LYP-D3/6-31G(d))



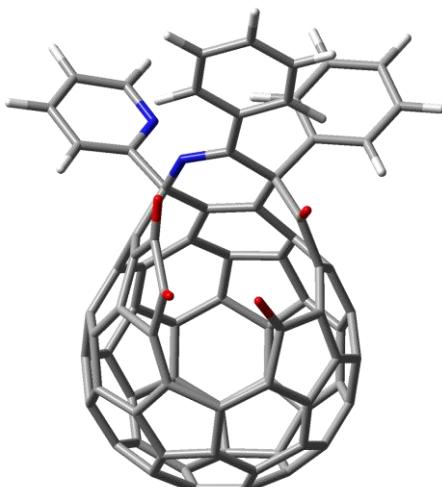
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			56	6	0	4.809509	-4.490236	-0.518572
			X	Y	Z						
1	6	0	3.956568	-0.193956	1.295735	60	6	0	-2.913872	3.274307	0.157804
2	6	0	3.775285	1.089298	0.998279	61	6	0	-1.409876	-3.649845	0.255355
3	6	0	2.824759	1.596474	-0.056318	62	6	0	-3.014347	2.209568	-2.474180
4	6	0	1.587075	2.297062	0.547863	63	6	0	-4.357184	-0.302915	-2.405150
5	6	0	1.244510	2.286320	1.900151	64	6	0	-3.963711	1.283978	1.874355
6	6	0	1.826150	1.784429	3.203374	65	6	0	5.160005	3.040154	-2.720186
7	6	0	0.583615	1.443965	4.121891	66	6	0	-1.277320	-3.148097	1.545567
8	6	0	-0.623065	1.694961	3.272939	67	6	0	-3.787394	-0.061190	2.362633
9	6	0	-1.939418	1.242376	3.127234	68	6	0	-4.757219	0.725158	-1.549502
10	6	0	-2.510279	-0.155252	3.083221	69	6	0	-4.090050	2.015427	-1.597356
11	6	0	-1.836070	-1.381007	3.001700	70	6	0	-2.374580	-2.438425	2.176938
12	6	0	-0.341857	-1.644147	3.110417	71	6	0	-2.510276	-1.332307	-3.427665
13	6	0	-0.031586	-2.666297	2.025856	72	6	0	-3.141415	-2.311309	-2.558949
14	6	0	1.077035	-2.619680	1.183732	73	6	0	-0.905458	-3.176052	-1.962148
15	6	0	2.403213	-2.107288	1.641709	74	6	0	-4.566861	-1.947393	-0.575089
16	6	0	3.265896	-1.329992	0.592214	75	6	0	-2.352171	-3.215624	-1.842058
17	7	0	3.954026	-3.607853	0.009941	76	6	0	-2.664033	-3.507882	-0.456918
18	8	0	2.963241	1.746009	3.605479	77	6	0	-3.590405	-2.348557	1.516071
19	8	0	2.812979	-2.295509	2.765638	78	6	0	-4.948500	-0.867990	0.323877
20	7	0	3.585168	3.865755	-0.563580	79	6	0	-3.739713	-2.876012	0.166771
21	6	0	2.381414	-0.864709	-0.543369	80	6	0	-4.274628	-1.669821	-1.915728
22	6	0	1.286575	0.842208	-1.913793	81	1	0	4.612844	-0.480571	2.110725
23	8	0	0.644016	1.281474	5.313963	82	1	0	4.283469	1.857022	1.569404
24	6	0	1.517821	-1.868315	-1.129425	83	1	0	4.258898	5.777894	-0.935688
25	6	0	0.592735	2.727724	-0.436682	84	1	0	5.743737	-0.794445	-0.501569
26	8	0	0.453573	-1.030712	3.785307	85	1	0	7.337064	-2.455230	-1.496479
27	6	0	-0.650722	2.226970	-2.531882	86	1	0	5.686071	5.138088	-2.884943
28	6	0	4.317492	4.741581	-1.263427	87	1	0	6.697384	-4.888733	-1.485907
29	6	0	-1.828117	2.907291	-2.019512	88	1	0	4.411113	1.064249	-2.254204
30	6	0	3.623549	2.580777	-0.932395	89	1	0	4.493212	-5.531314	-0.495828
31	6	0	0.520745	2.075419	-1.754809	90	1	0	5.770748	2.714230	-3.557818
32	6	0	0.792893	-0.169582	-2.747202						
33	6	0	2.233453	0.460148	-0.891453						
34	6	0	-1.790023	3.456062	-0.750474						

The total electronic energy was calculated to be -3236.211551 Hartree.

The total electronic energy was calculated to be -3236.211551 Hartree.

Table S2. Optimized structure of **4-keto-form** (B3LYP-D3/6-31G(d))



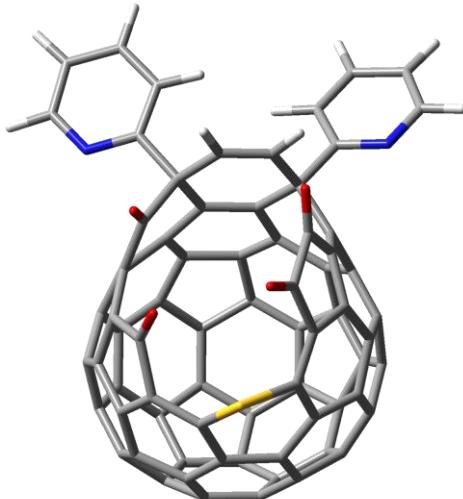
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			57	6	0	-1.392025	-3.162921	1.574429
			X	Y	Z						
1	6	0	-3.690557	1.119115	-2.925423	60	6	0	-1.731795	-1.439408	3.144421
2	6	0	-2.276658	1.196985	-3.252778	61	6	0	2.211860	1.658935	-0.660057
3	6	0	-1.567681	0.032425	-3.499608	62	6	0	2.859015	-1.192438	-0.176347
4	6	0	-2.221232	-1.261912	-3.415266	63	6	0	3.668809	0.019516	0.358490
5	6	0	-3.567665	-1.343156	-3.061541	64	7	0	3.312337	1.233099	0.165895
6	6	0	-4.317912	-0.124627	-2.810829	65	6	0	3.753201	-2.123540	-1.028514
7	6	0	-1.668617	2.256938	-2.469247	66	6	0	2.866323	2.651810	-1.647827
8	6	0	-0.226945	-0.093333	-2.995553	67	8	0	2.756401	-2.121638	2.076607
9	6	0	-1.260387	-2.182883	-2.849669	68	8	0	3.023175	1.767984	2.901224
10	6	0	-4.001972	-2.348565	-2.105820	69	6	0	3.011507	4.006437	-1.339309
11	6	0	-5.220038	-0.377233	-1.700207	70	6	0	3.682147	4.826002	-2.244381
12	6	0	-3.978619	2.172634	-1.975186	71	6	0	4.185943	4.263858	-3.417564
13	6	0	-3.069557	-3.235331	-1.559656	72	6	0	4.000594	2.896976	-3.625224
14	6	0	-1.671460	-3.154222	-1.945151	73	7	0	3.357333	2.100782	-2.763115
15	6	0	-3.113136	-3.548378	-0.145026	74	6	0	3.828194	-3.504932	-0.826726
16	6	0	-0.850871	-3.411069	-0.781102	75	6	0	4.615098	-4.297279	-1.665135
17	6	0	-0.370355	2.125812	-1.924856	76	6	0	5.329085	-3.717502	-2.713717
18	6	0	-0.020099	-1.475552	-2.617934	77	6	0	5.247341	-2.338833	-2.924183
19	6	0	-1.745502	-3.655231	0.322669	78	6	0	4.460039	-1.545770	-2.091934
20	6	0	0.827507	-1.780655	-1.555895	79	8	0	0.643631	-1.046288	3.542055
21	6	0	0.382445	0.921931	-2.249928	80	8	0	1.063621	1.266287	5.007259
22	6	0	-5.013365	-1.746099	-1.255351	81	6	0	6.726078	-1.508968	2.096537
23	6	0	-5.482238	0.629893	-0.769494	82	6	0	5.509121	-1.418339	1.423016
24	6	0	-0.072129	2.753653	-0.625260	83	6	0	4.947789	-0.169373	1.113278
25	6	0	-4.868743	1.939092	-0.918734	84	6	0	5.648730	0.986365	1.510907
26	6	0	-2.746641	2.896762	-1.734509	85	6	0	6.859497	0.894498	2.186562
27	6	0	-5.518513	0.316042	0.648526	86	6	0	7.408811	-0.356026	2.480777
28	6	0	-4.566234	2.448233	0.402473	87	1	0	2.612496	4.401557	-0.411286
29	6	0	-2.487702	3.430237	-0.484630	88	1	0	3.810497	5.884628	-2.036088
30	6	0	0.381083	-2.762117	-0.570492	89	1	0	4.713111	4.865064	-4.151897
31	6	0	-5.049009	-2.046357	0.111710	90	1	0	4.384655	2.415192	-4.522465
32	6	0	1.501739	0.553385	-1.421386	91	1	0	3.293835	-3.971276	-0.004466
33	6	0	-4.932046	1.426534	1.372347	92	1	0	4.669389	-5.368724	-1.492657
34	6	0	-3.420958	3.209440	0.611486	93	1	0	5.942995	-4.334819	-3.363690
35	6	0	-1.144670	3.378030	0.038296	94	1	0	5.795763	-1.877795	-3.741152

96	1	0	7.135685	-2.489481	2.323444	100	1	0	8.356706	-0.429767	3.007724
97	1	0	5.003732	-2.332078	1.154687						
98	1	0	5.213299	1.951922	1.291058						
99	1	0	7.374812	1.802475	2.488260						

The total electronic energy was calculated to be -3467.2768334 Hartree.

Table S3. Optimized structure of **6'** (B3LYP-D3/6-31G(d))



Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			57	58	59	60	61	62
			X	Y	Z						
1	6	0	3.799468	0.317295	1.475004						
2	6	0	3.808213	-0.936309	1.027266						
3	6	0	3.002590	-1.439803	-0.145266						
4	6	0	1.832075	-2.350711	0.295039						
5	6	0	1.477941	-2.582589	1.623565						
6	6	0	1.999513	-2.220246	2.996760						
7	6	0	0.720943	-2.025103	3.911556						
8	6	0	-0.432105	-2.446046	3.077558						
9	6	0	-1.767712	-2.058500	3.061953						
10	6	0	-2.864303	0.317652	3.058606						
11	6	0	-2.262776	1.569781	2.975552						
12	6	0	-0.787476	1.889869	3.267096						
13	6	0	-0.370057	2.755722	2.053475						
14	6	0	0.791312	2.660712	1.282036						
15	6	0	2.073413	2.111168	1.826277						
16	6	0	3.068102	1.449402	0.806778						
17	16	0	-2.290848	-0.855025	4.267879						
18	7	0	3.668871	3.807551	0.482432						
19	8	0	3.120302	-2.177514	3.443653						
20	8	0	2.364033	2.183083	2.999981						
21	7	0	4.058551	-3.536182	-0.831256						
22	6	0	2.326947	1.003745	-0.433312						
23	6	0	1.509498	-0.662091	-2.028458						
24	8	0	0.760313	-1.634993	5.053389						
25	6	0	1.456040	1.989863	-1.028701						
26	6	0	0.898538	-2.727830	-0.771661						
27	8	0	-0.100117	1.436903	4.145976						
28	6	0	-0.282009	-2.114786	-2.880520						
29	6	0	4.938071	-4.238925	-1.556474						
30	6	0	-1.444467	-2.885221	-2.478582						
31	6	0	3.972409	-2.220130	-1.053546						
32	6	0	0.835826	-1.957993	-2.023102						
33	6	0	1.002893	0.373616	-2.821469						
34	6	0	2.337311	-0.292584	-0.903820						

The total electronic energy was calculated to be -3634.4345584 Hartree.

6. Differences in Energies between **7'** and **8'**

For the calculations to be simplified, 6-*t*-butylpyridin-2-yl groups were replaced with 2-pyridyl groups for **7'** and **8'**.

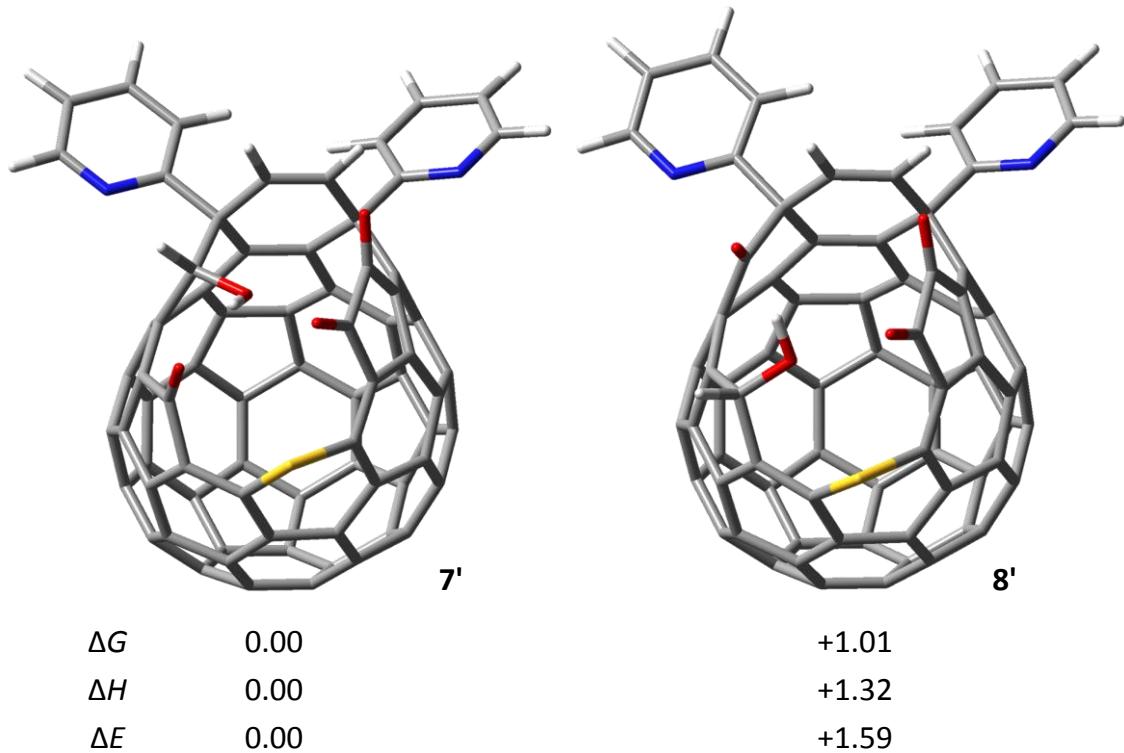
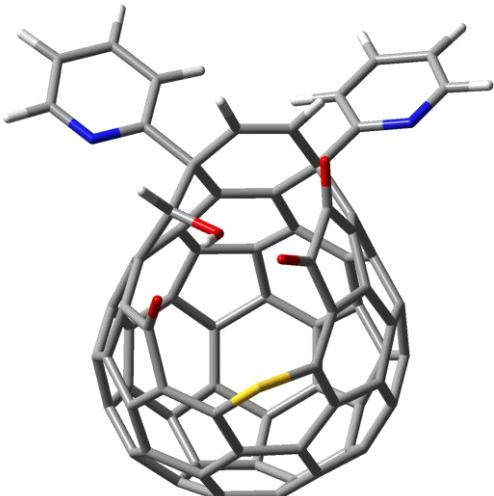


Figure S15. Difference in energies (units in kcal/mol) between two structural isomers, **7'** and **8'**, calculated at the B3LYP-D3/6-31G(d) level of theory (298 K).

Table S4. Optimized structure of **7'** (B3LYP-D3/6-31G(d))

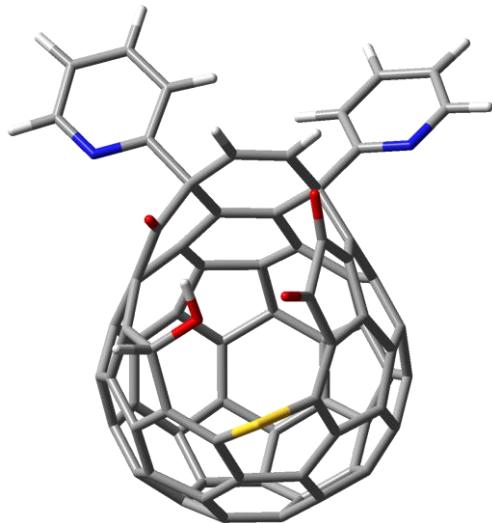


Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93
			X	Y	Z																																				
1	6	0	3.897887	0.898650	1.132940																																				
2	6	0	3.986539	-0.428995	1.100326																																				
3	6	0	3.140481	-1.322153	0.222361																																				
4	6	0	2.014840	-2.067690	0.978134																																				
5	6	0	1.682631	-1.841679	2.308539																																				
6	6	0	2.207477	-0.986650	3.430792																																				
7	6	0	0.929348	-0.493901	4.224312																																				
8	6	0	-0.215254	-1.248766	3.659618																																				
9	6	0	-1.570368	-0.943393	3.550031																																				
10	6	0	-2.853312	1.239569	2.814301																																				
11	6	0	-2.367488	2.449695	2.325995																																				
12	6	0	-0.935120	2.948080	2.467950																																				
13	6	0	-0.525631	3.271217	1.030800																																				
14	6	0	0.634467	2.875766	0.367229																																				
15	6	0	1.835786	2.343463	1.166954																																				
16	6	0	2.961387	1.716154	0.277191																																				
17	16	0	-2.150926	0.563864	4.303095																																				
18	7	0	3.387398	4.083166	-0.283695																																				
19	8	0	3.332306	-0.762177	3.807049																																				
20	8	0	1.406701	1.327211	2.067939																																				
21	7	0	4.304269	-3.466552	0.275301																																				
22	6	0	2.291291	0.861959	-0.777451																																				
23	6	0	1.576542	-1.259904	-1.756124																																				
24	8	0	0.959811	0.355523	5.086442																																				
25	6	0	1.356280	1.559234	-1.625236																																				
26	6	0	1.080258	-2.812242	0.125639																																				
27	8	0	-0.247606	2.975369	3.468739																																				
28	6	0	-0.156624	-2.980227	-2.034841																																				
29	6	0	5.182642	-4.354203	-0.207186																																				
30	6	0	-1.279132	-3.613141	-1.367168																																				
31	6	0	4.101679	-2.341339	-0.420002																																				
32	6	0	0.966289	-2.509033	-1.308985																																				
33	6	0	1.004871	-0.556540	-2.820667																																				
34	6	0	2.395139	-0.510191	-0.833017																																				
35	6	0	-1.226115	-3.785127	0.001271																																				

The total electronic energy was calculated to be – 3635.6737951 Hartree.

Table S5. Optimized structure of **8'** (B3LYP-D3/6-31G(d))



Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)			58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93
			X	Y	Z																																				
1	6	0	3.853846	0.184806	1.445100																																				
2	6	0	3.821972	-1.038519	0.923010																																				
3	6	0	2.985287	-1.448336	-0.263761																																				
4	6	0	1.792230	-2.353994	0.126191																																				
5	6	0	1.425298	-2.664652	1.436599																																				
6	6	0	1.936403	-2.379345	2.829018																																				
7	6	0	0.659203	-2.308456	3.765799																																				
8	6	0	-0.489505	-2.658382	2.899801																																				
9	6	0	-1.824119	-2.273347	2.912827																																				
10	6	0	-2.853101	0.109778	3.063822																																				
11	6	0	-2.186209	1.325638	3.043661																																				
12	6	0	-0.713428	1.651574	3.423672																																				
13	6	0	-0.313667	2.561543	2.218121																																				
14	6	0	0.845897	2.532645	1.445943																																				
15	6	0	2.099636	1.907035	1.940139																																				
16	6	0	3.112284	1.367003	0.879263																																				
17	16	0	-2.391301	-1.185665	4.204502																																				
18	7	0	3.754107	3.736755	0.906591																																				
19	8	0	3.055583	-2.299832	3.277699																																				
20	8	0	2.350748	1.790004	3.130486																																				
21	7	0	4.036732	-3.521743	-1.012691																																				
22	6	0	2.365706	1.021912	-0.392181																																				
23	6	0	1.509191	-0.517854	-2.088519																																				
24	8	0	0.711468	-2.045849	4.944147																																				
25	6	0	1.505150	2.058757	-0.911487																																				
26	6	0	0.861971	-2.650770	-0.967428																																				
27	8	0	0.033408	0.473583	3.516586																																				
28	6	0	-0.304306	-1.882523	-3.032721																																				
29	6	0	4.894096	-4.209958	-1.776689																																				
30	6	0	-1.476625	-2.665004	-2.686673																																				
31	6	0	3.927379	-2.204895	-1.221740																																				
32	6	0	0.814216	-1.799611	-2.166638																																				
33	6	0	1.017067	0.577383	-2.807442																																				
34	6	0	2.348649	-0.239993	-0.946038																																				
35	6	0	-1.475915	-3.377471	-1.505713																																				

The total electronic energy was calculated to be – 3635.6712615 Hartree.

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