

Supporting Information

Homoleptic titanium and zirconium complexes exhibiting unusual O_{iminol}-metal coordination: Application in stereoselective ring-opening polymerization of lactide

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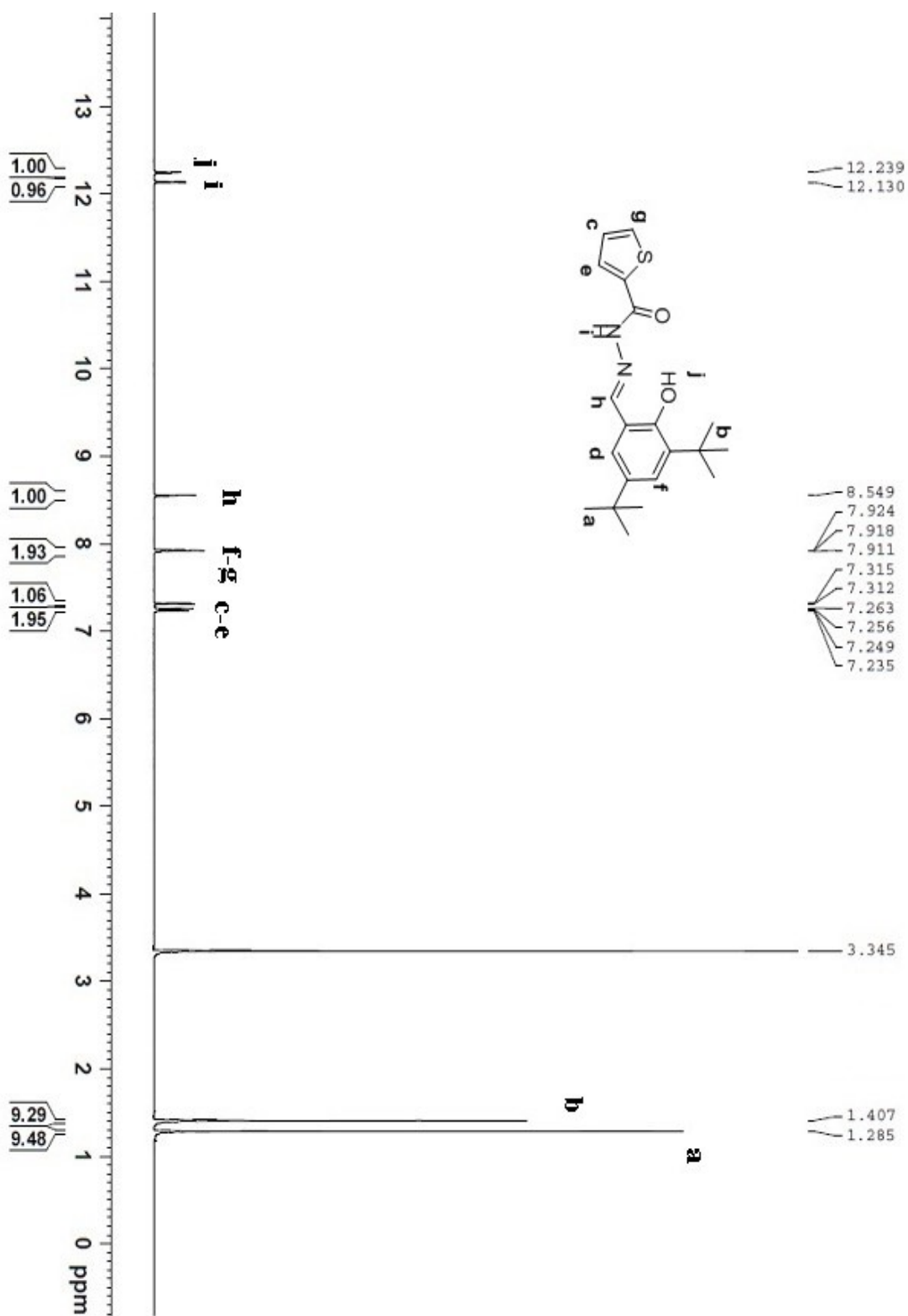


Fig. S1. ¹H NMR (400 MHz, DMSO-d₆) of Compound **L1H₂**

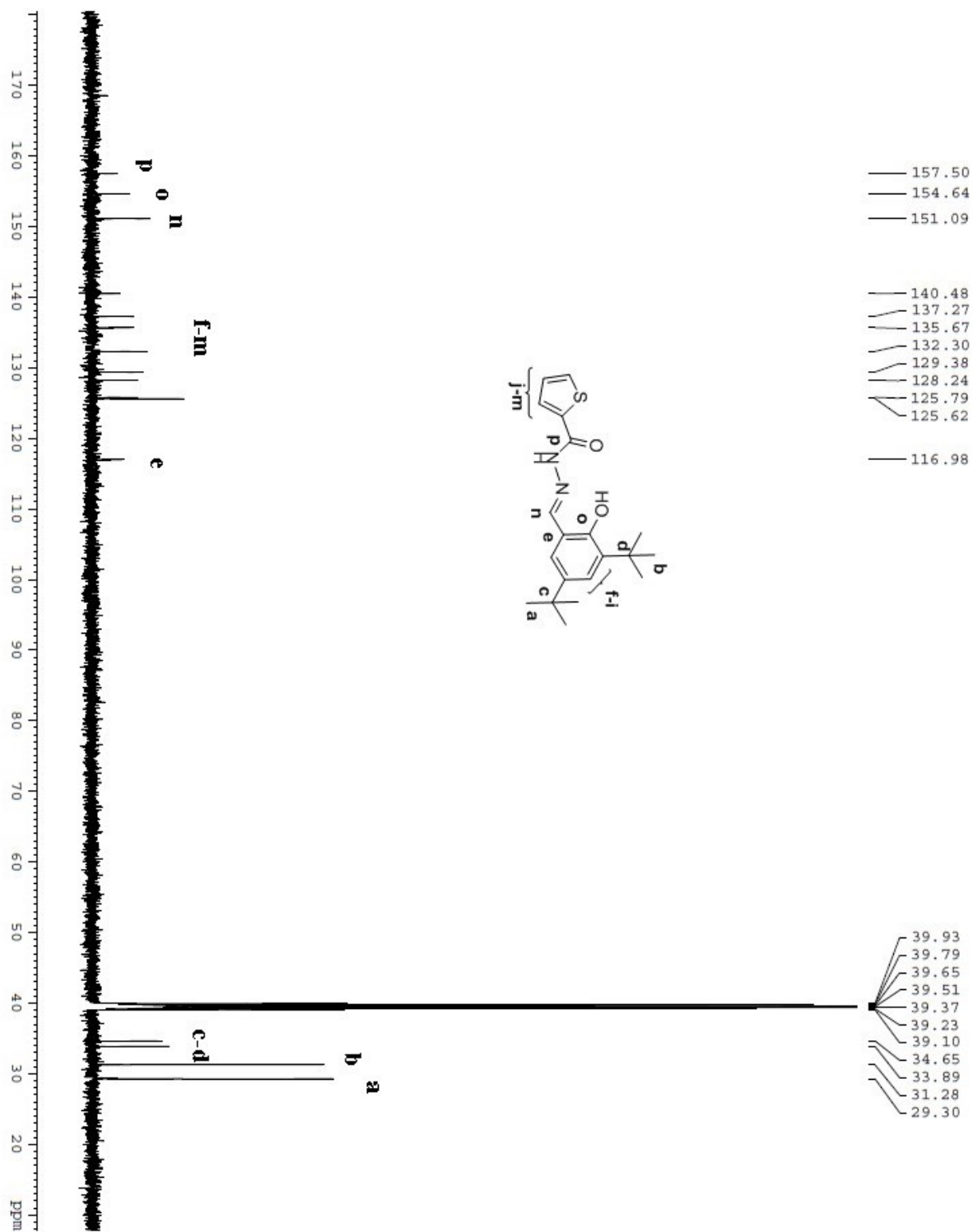
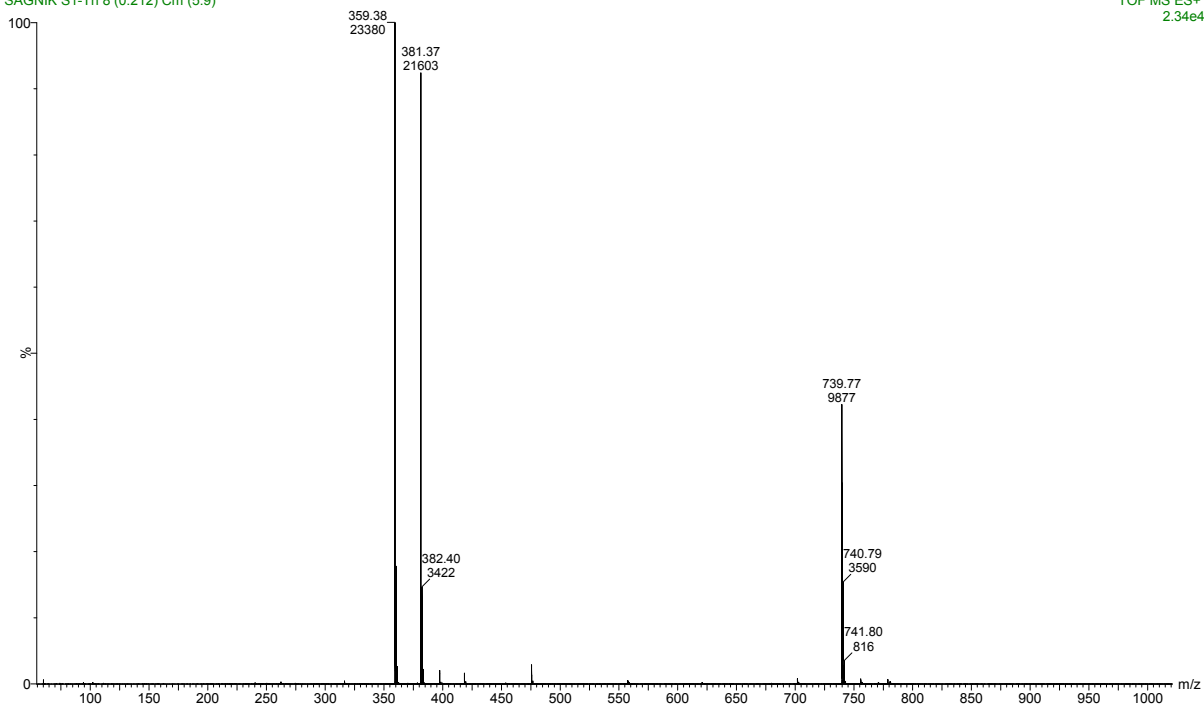


Fig. S2. ^{13}C NMR (100 MHz, DMSO- d_6) of Compound L1H₂

WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S1-Th 8 (0.212) Cm (5:9)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
2.34e4



WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S1-Th 8 (0.212) Cm (5:9)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
2.34e4

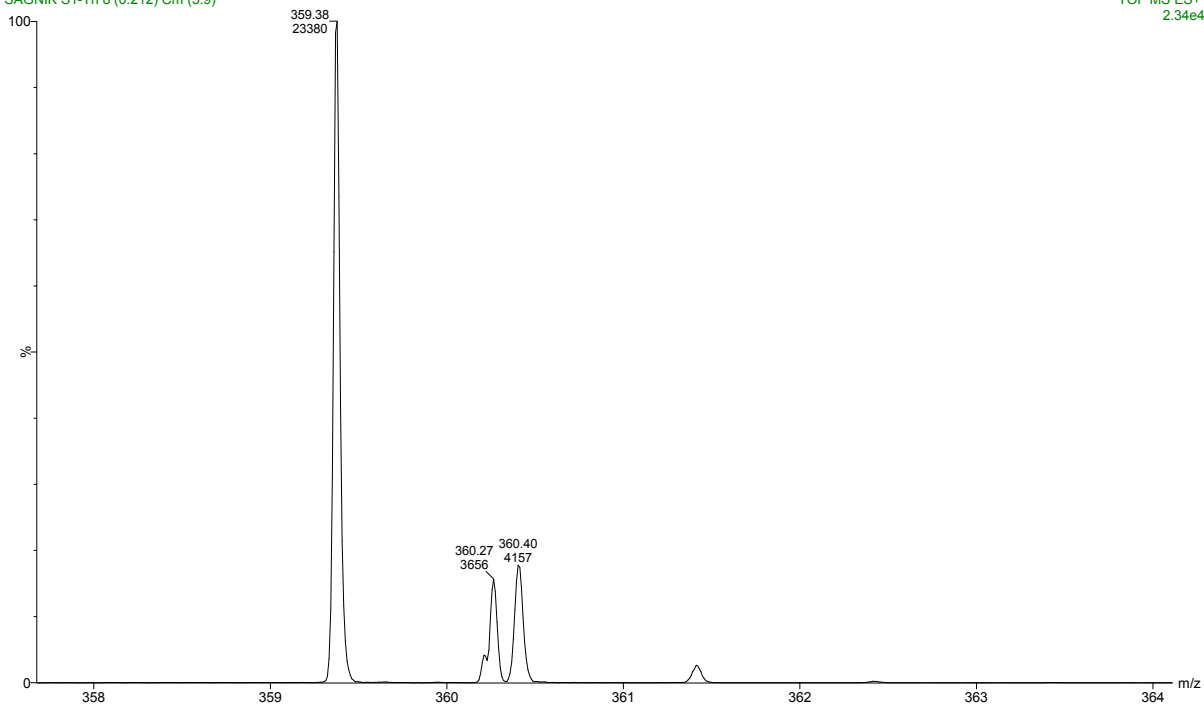


Fig. S3. ESI mass spectrum of Compound L1H₂

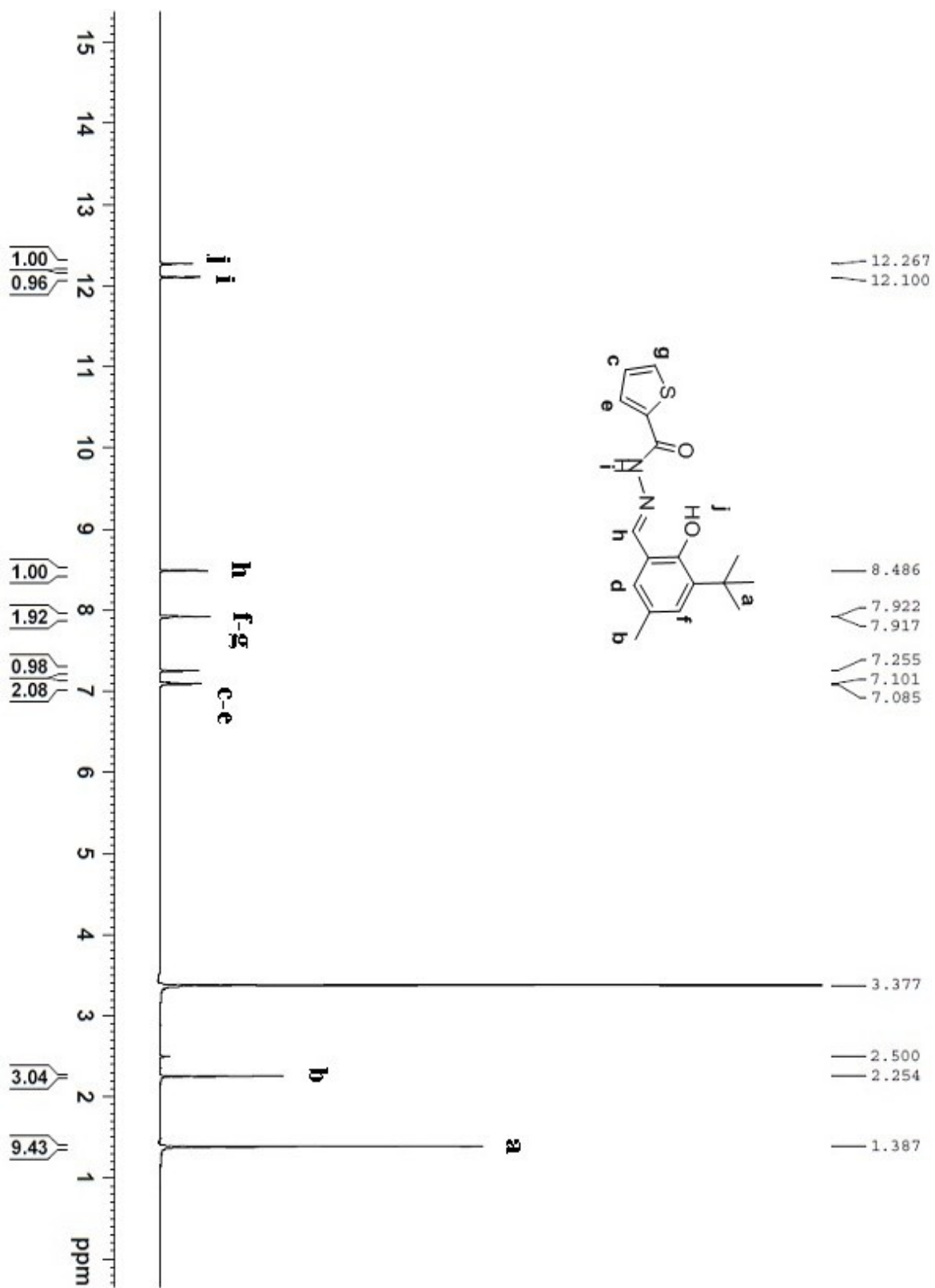


Fig. S4. ¹H NMR (400 MHz, DMSO-d₆) of Compound **L2H₂**

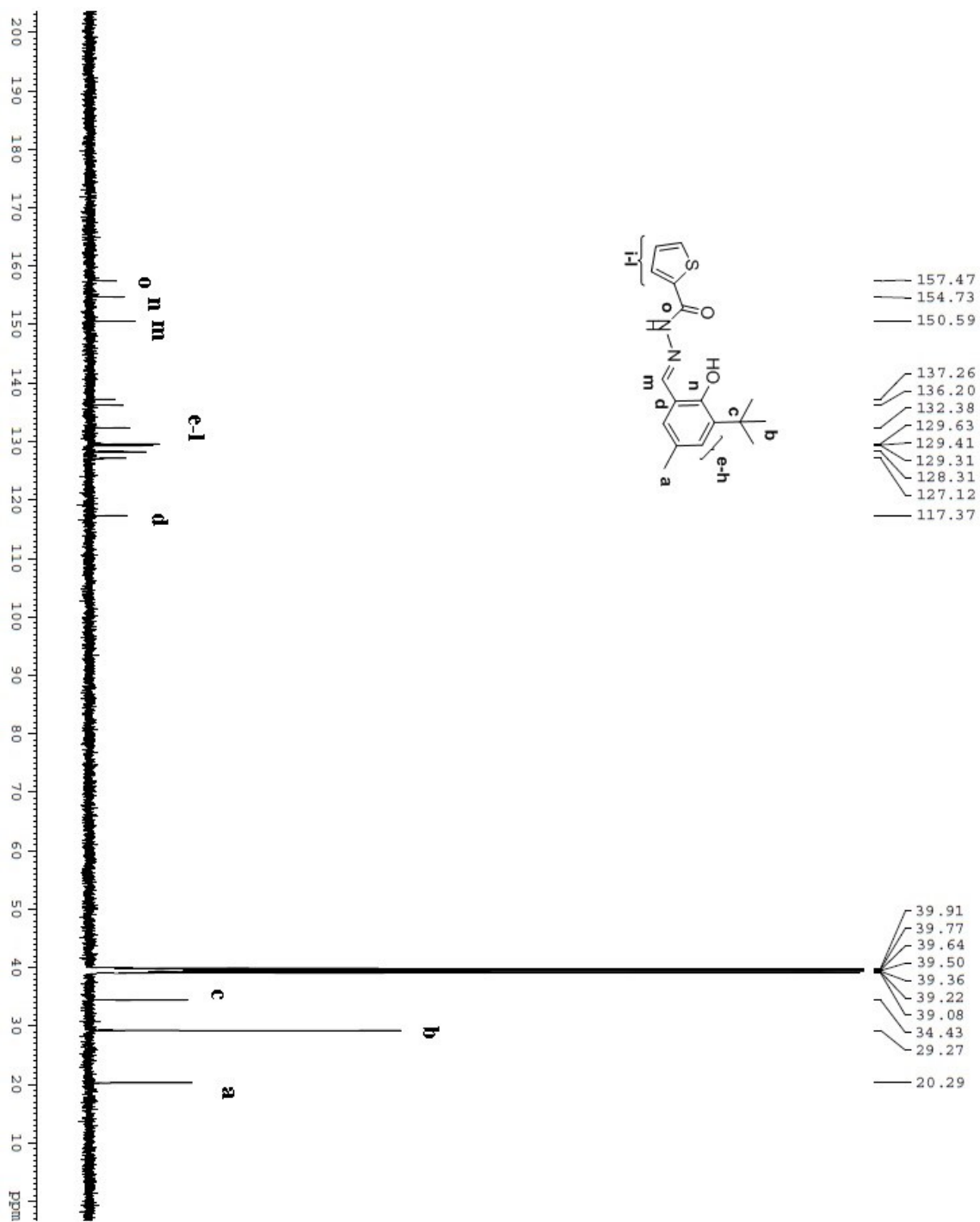
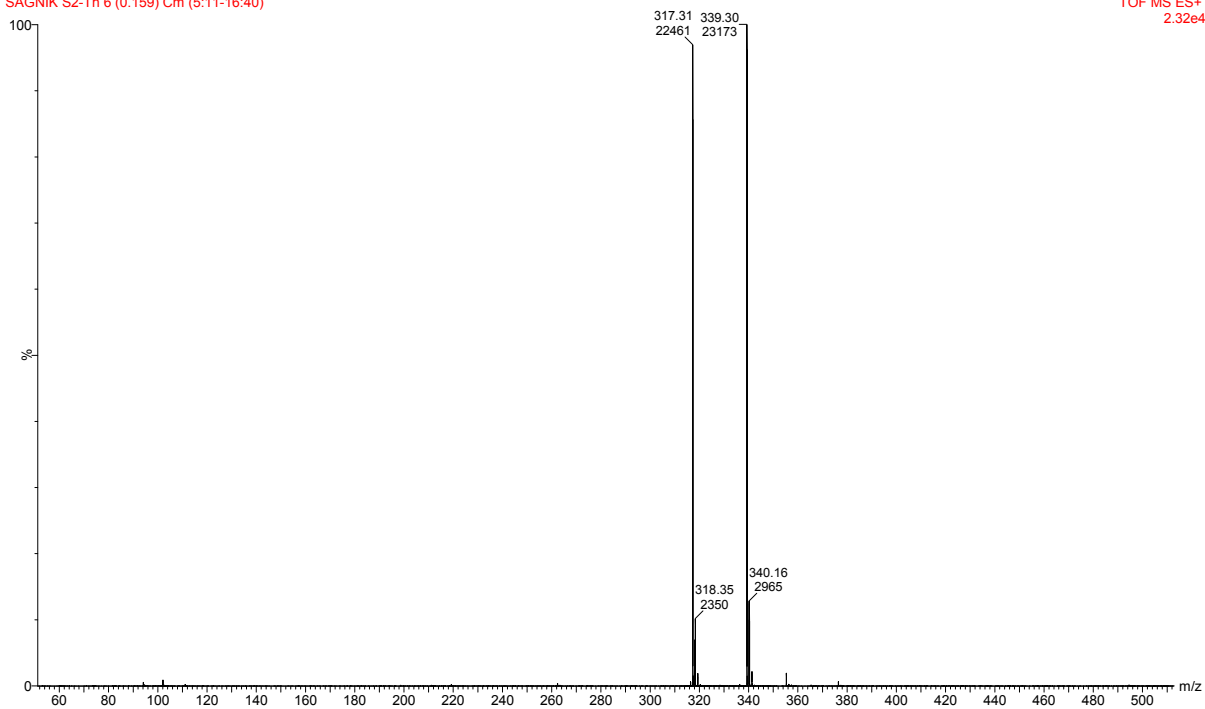


Fig. S5. ^{13}C NMR (100 MHz, DMSO- d_6) of Compound L2H₂

WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S2-Th 6 (0.159) Cm (5:11-16:40)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
2.32e4



WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S2-Th 6 (0.159) Cm (5:11-16:40)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
2.25e4

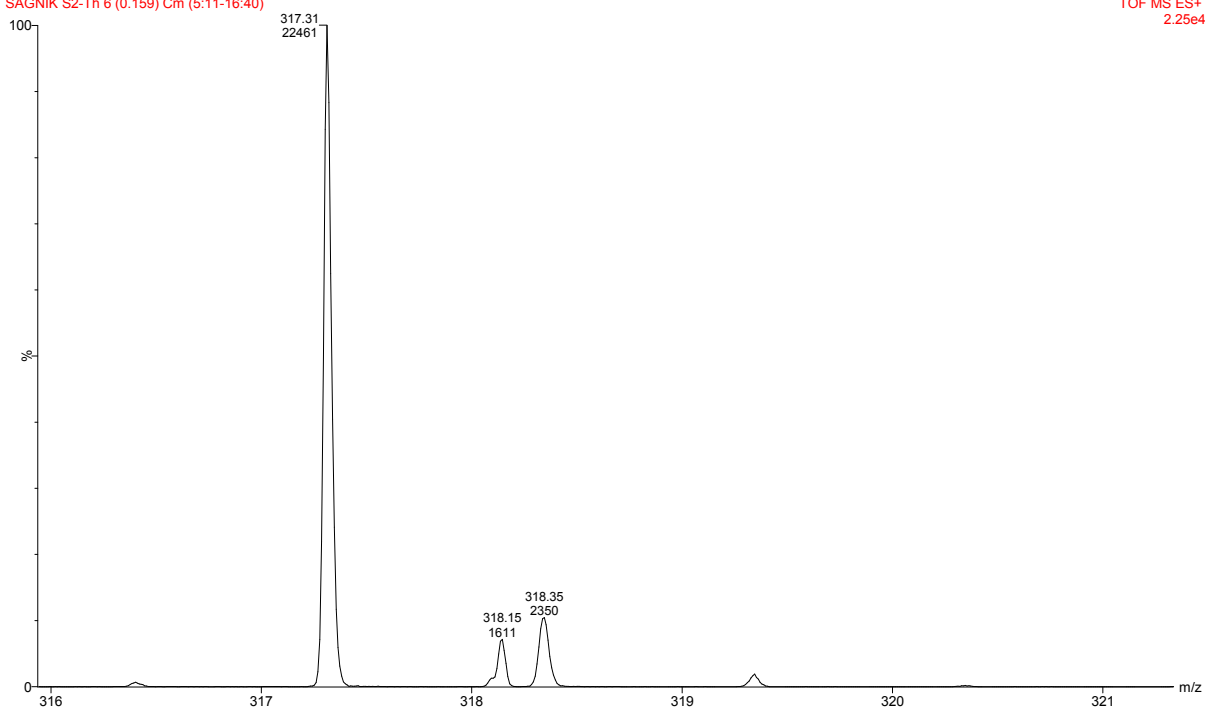


Fig. S6. ESI mass spectrum of Compound L2H₂

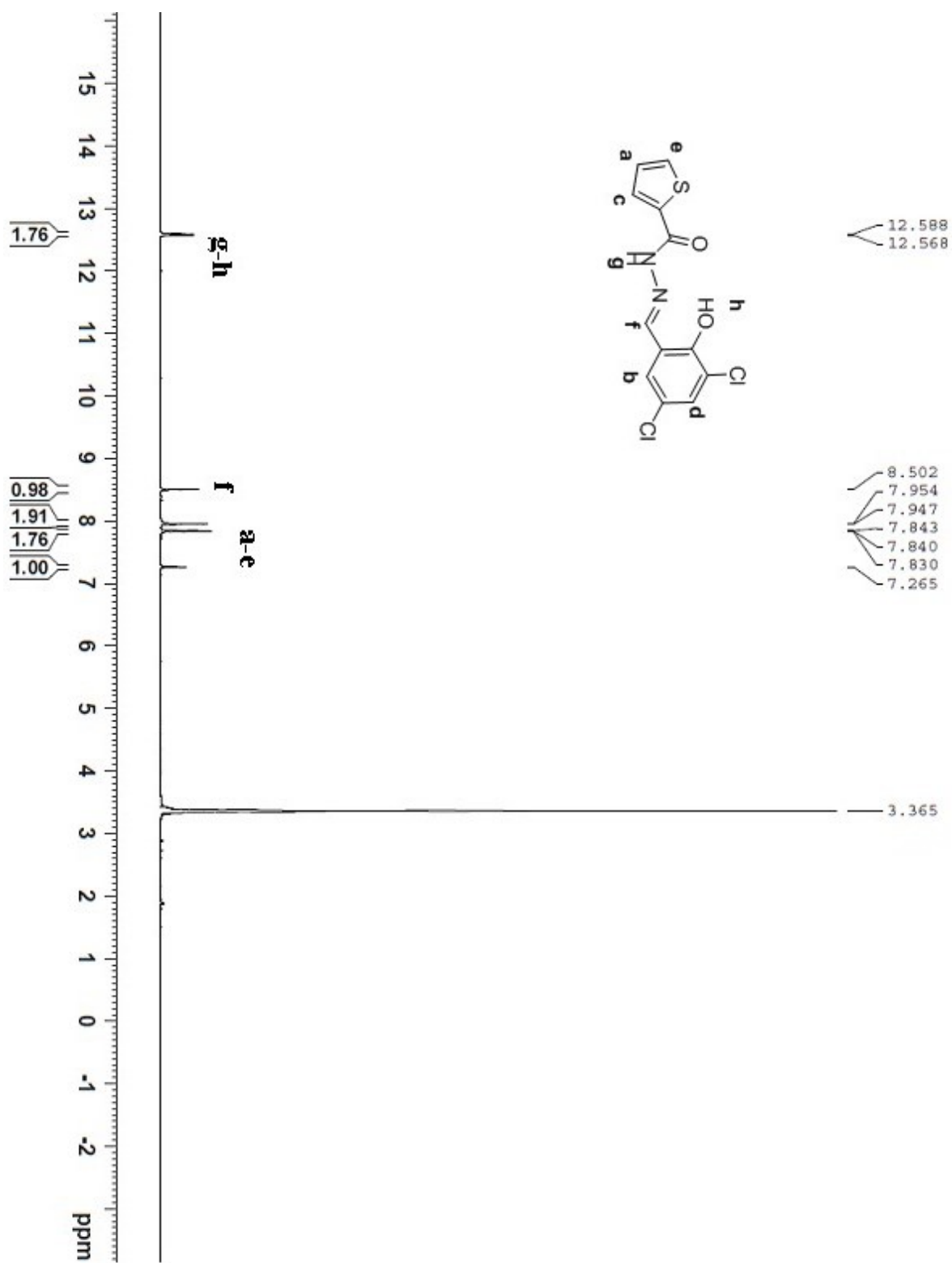


Fig. S7. ¹H NMR (400 MHz, DMSO-d₆) of Compound **L3H₂**

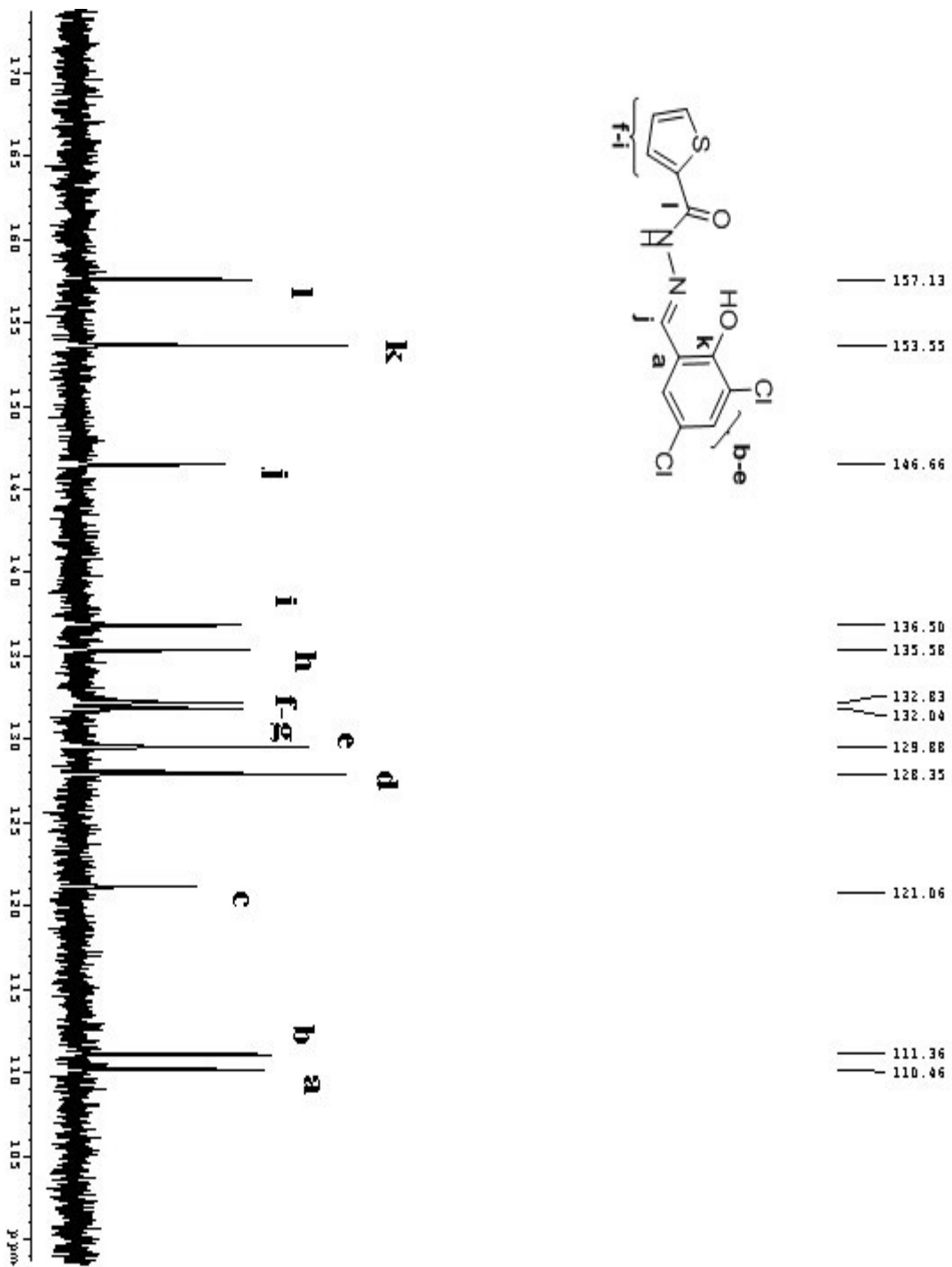
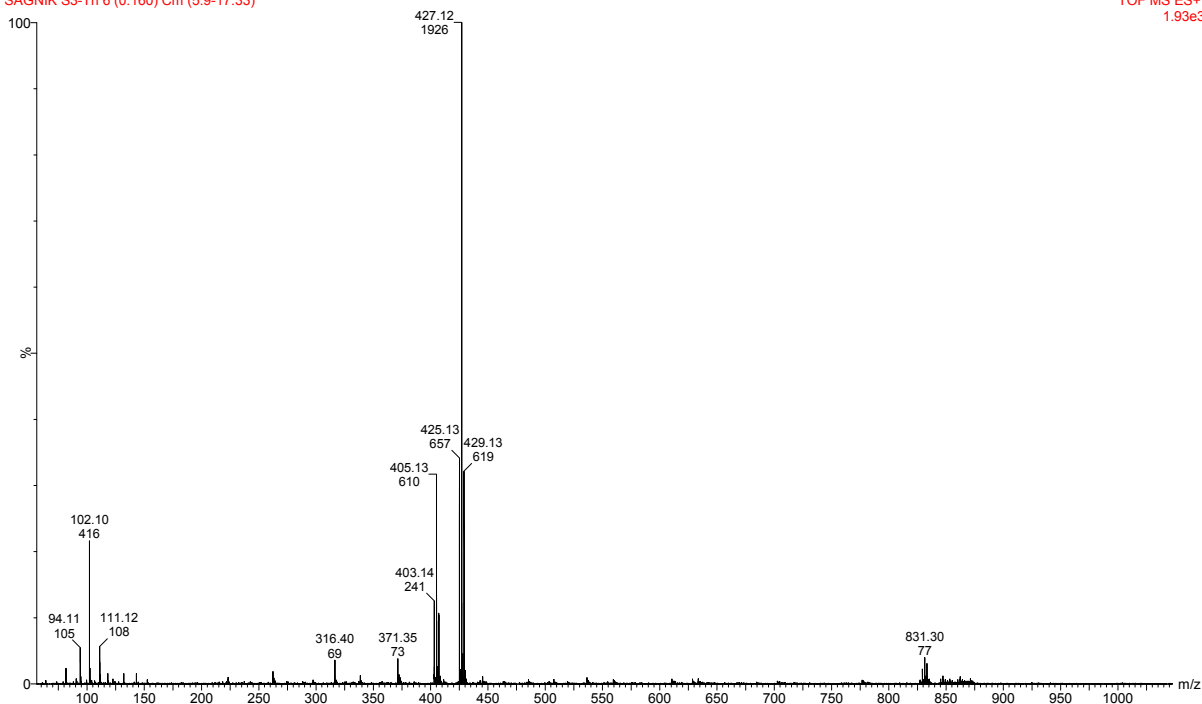


Fig. S8. ^{13}C NMR (100 MHz, DMSO- d_6) of Compound L3H₂

WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S3-Th 6 (0.160) Cm (5:9-17:33)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
1.93e3



WATERS, Q-TOF MICROMASS (LC-MS)
SAGNIK S3-Th 6 (0.160) Cm (5:9-17:33)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
1.93e3

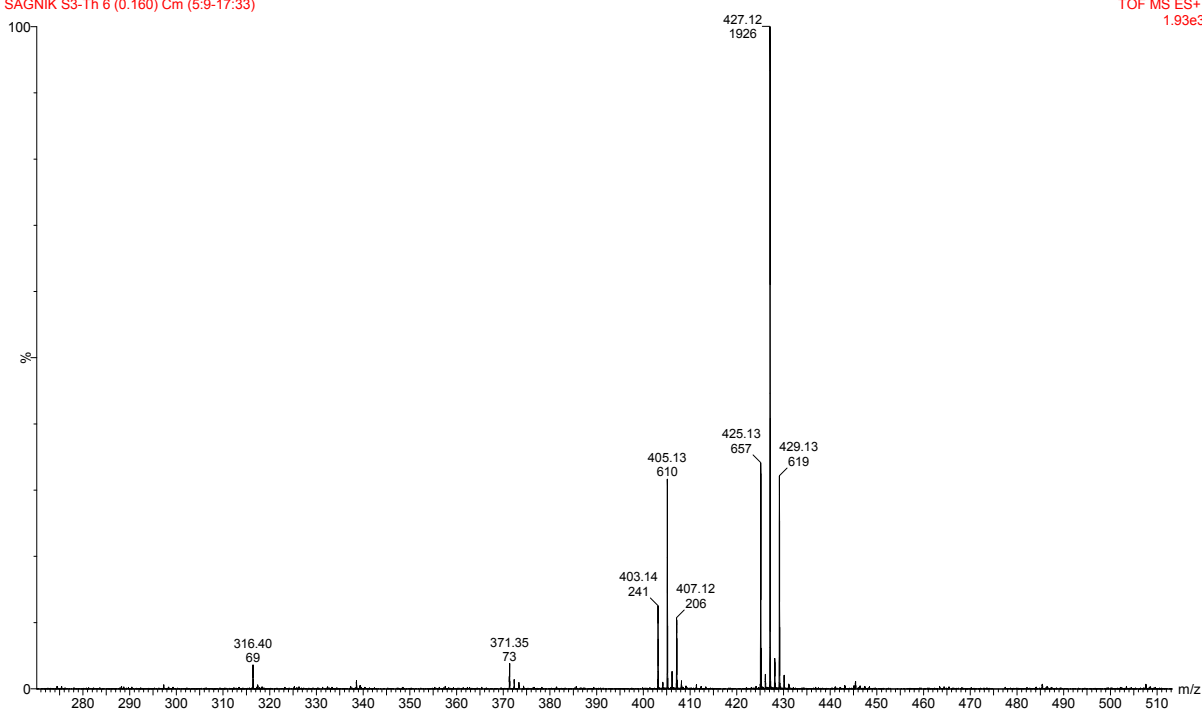


Fig. S9. ESI mass spectrum of Compound L3H₂

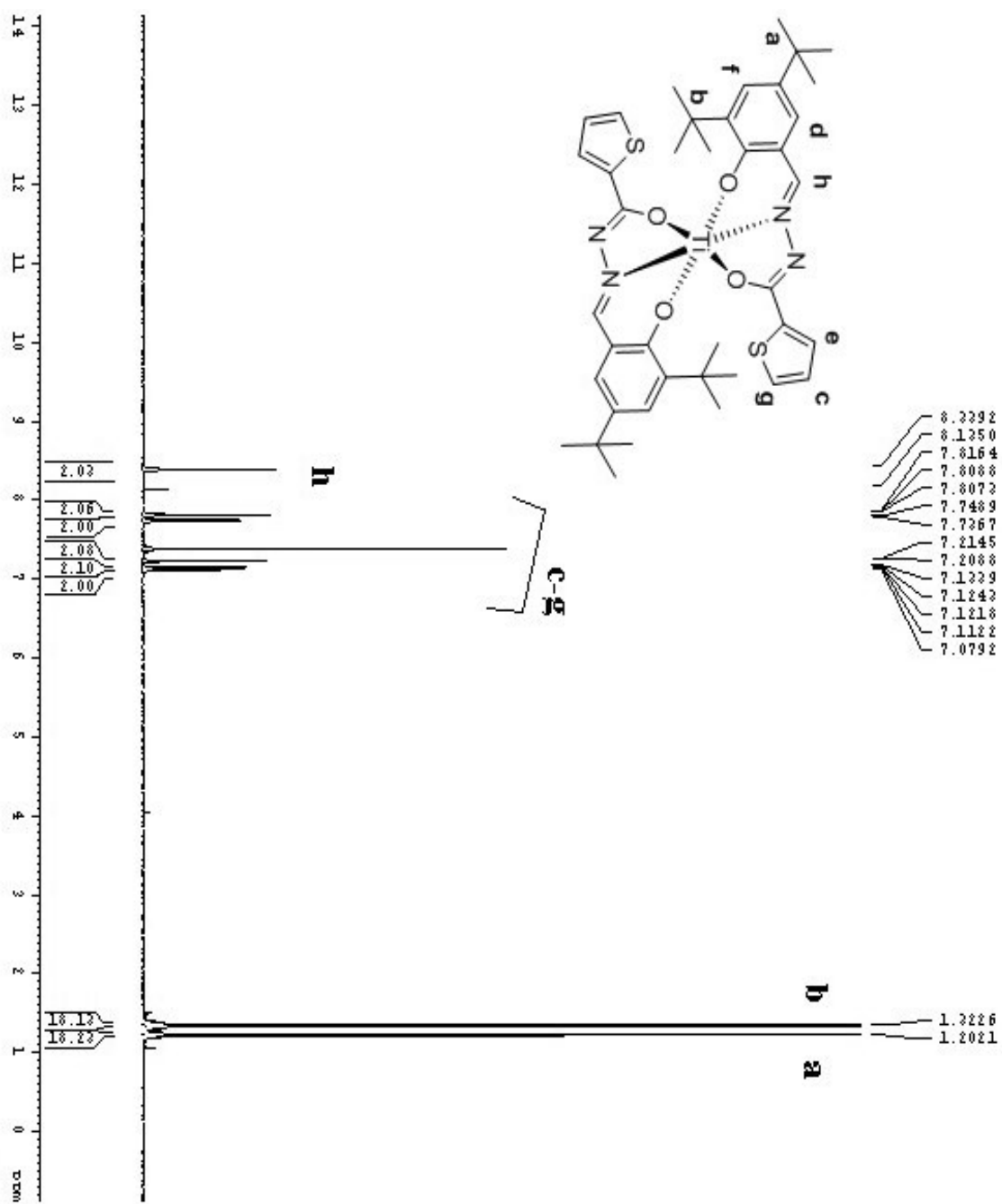


Fig. S10. ^1H NMR (400 MHz, CDCl_3) of Compound 1

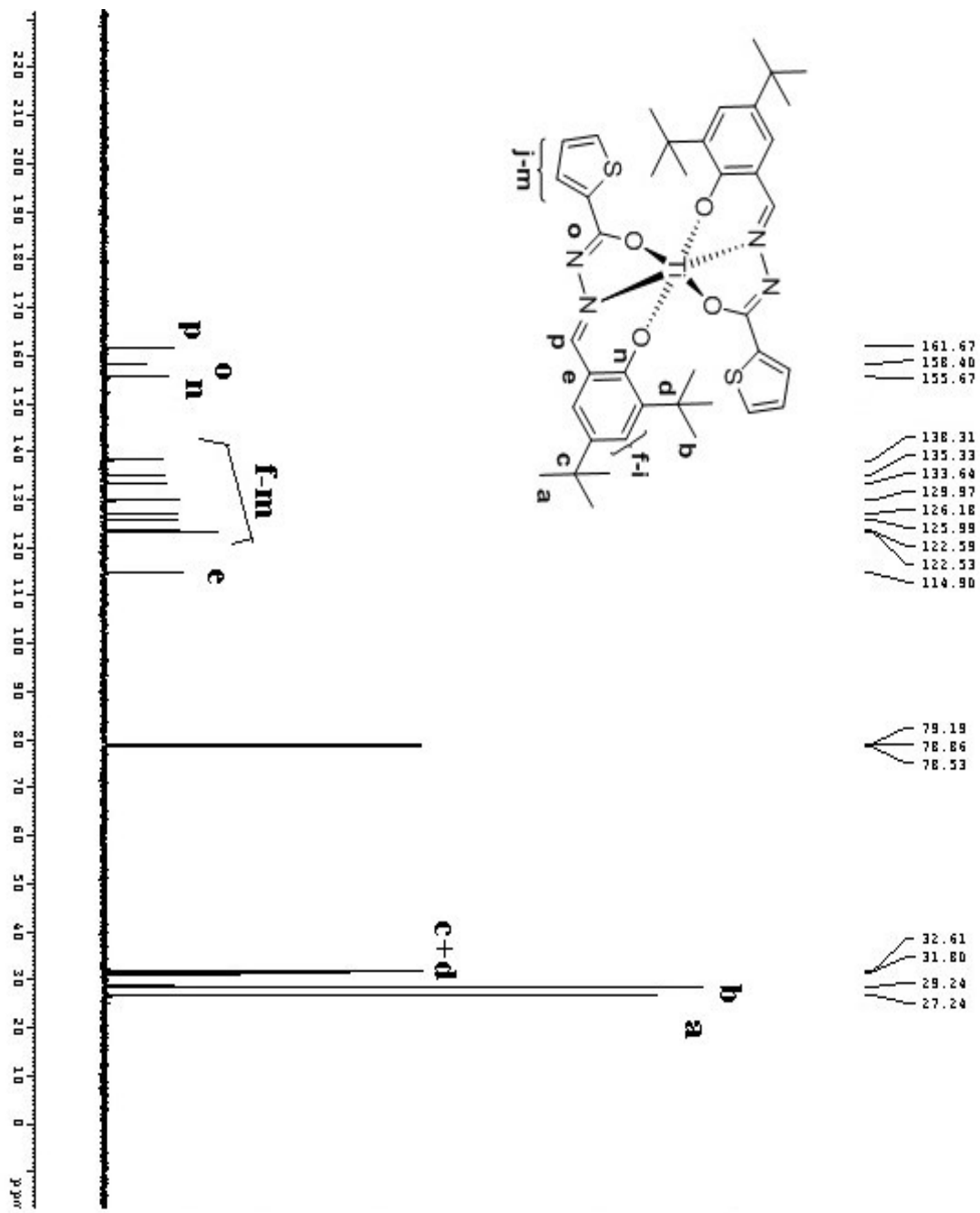


Fig. S11. ^{13}C NMR (100 MHz, CDCl_3) of Compound 1

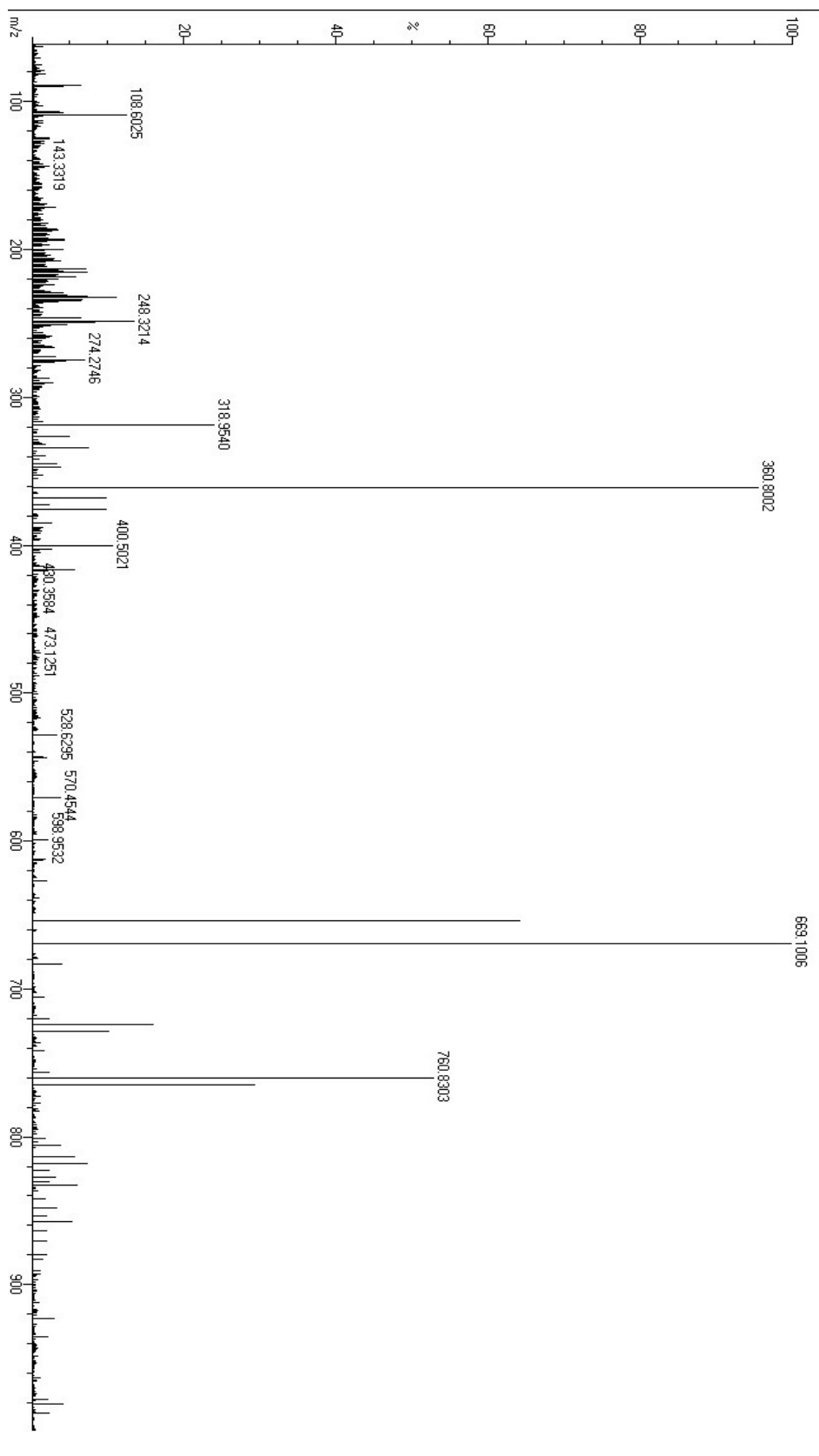


Fig. S12. ESI mass spectrum of Compound 1

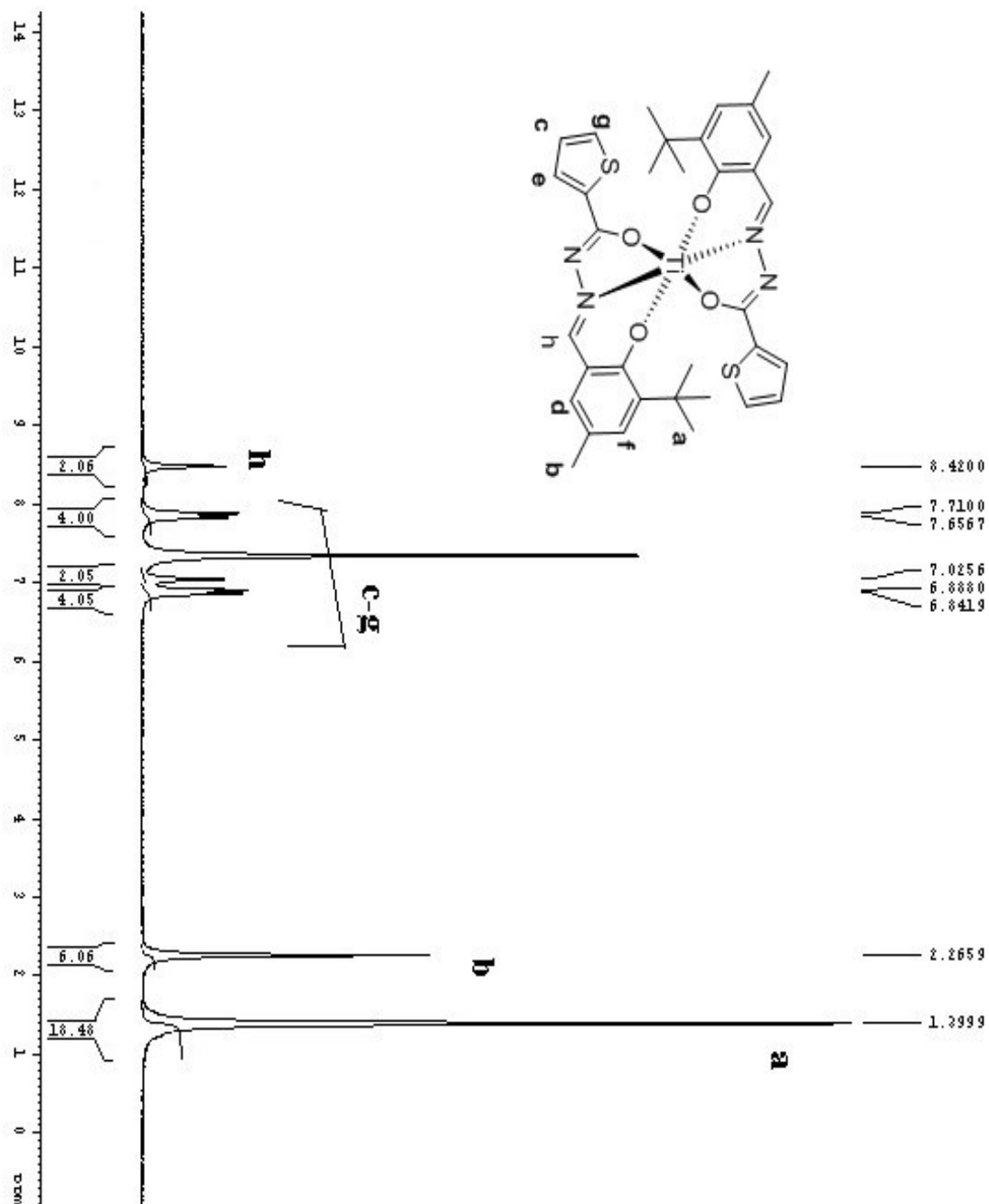


Fig. S13. ¹H NMR (400 MHz, CDCl₃) of Compound 2

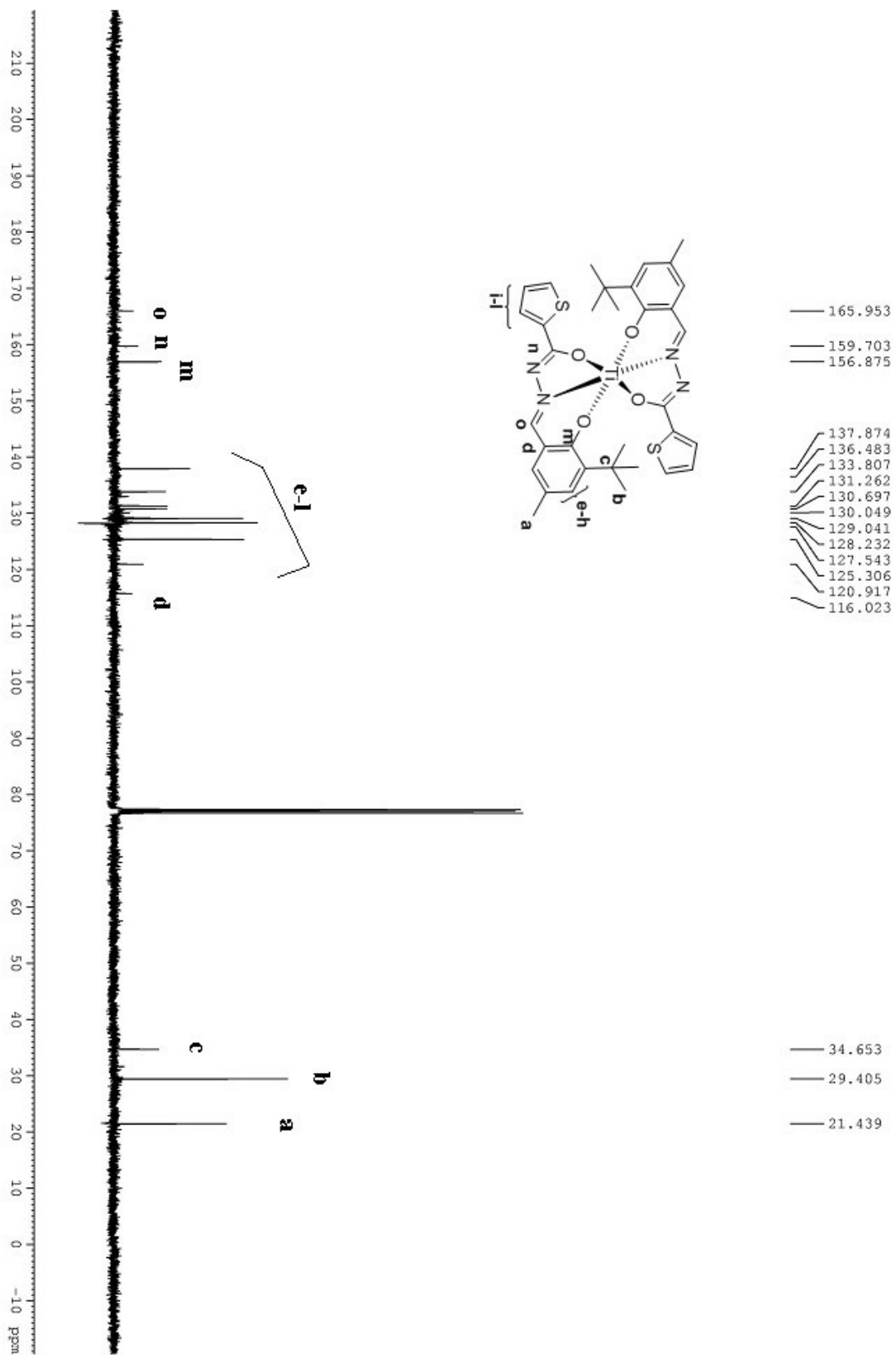


Fig. S14. ¹³C NMR (100 MHz, CDCl₃) of Compound 2

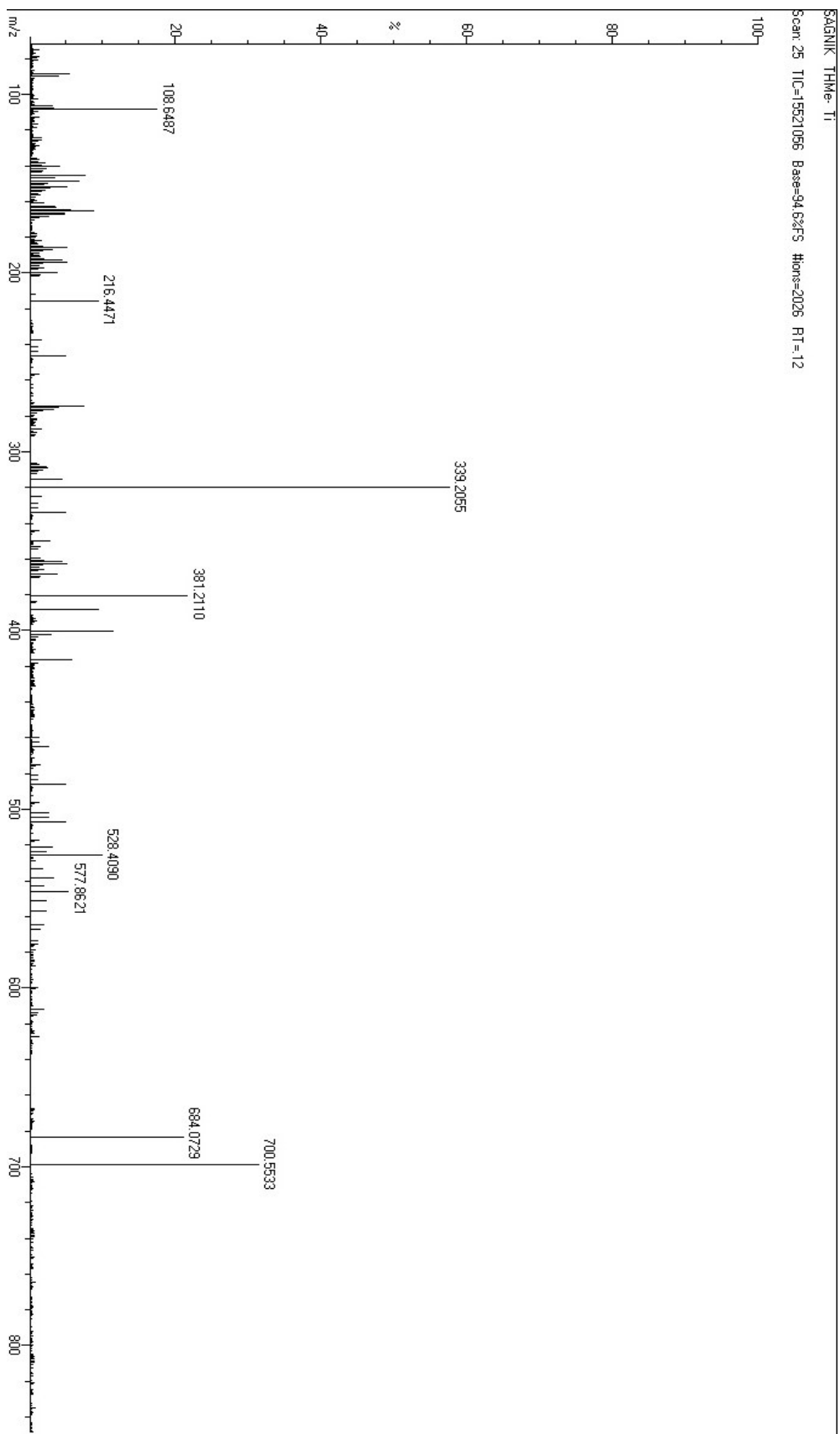


Fig. S15. ESI mass spectrum of Compound 2

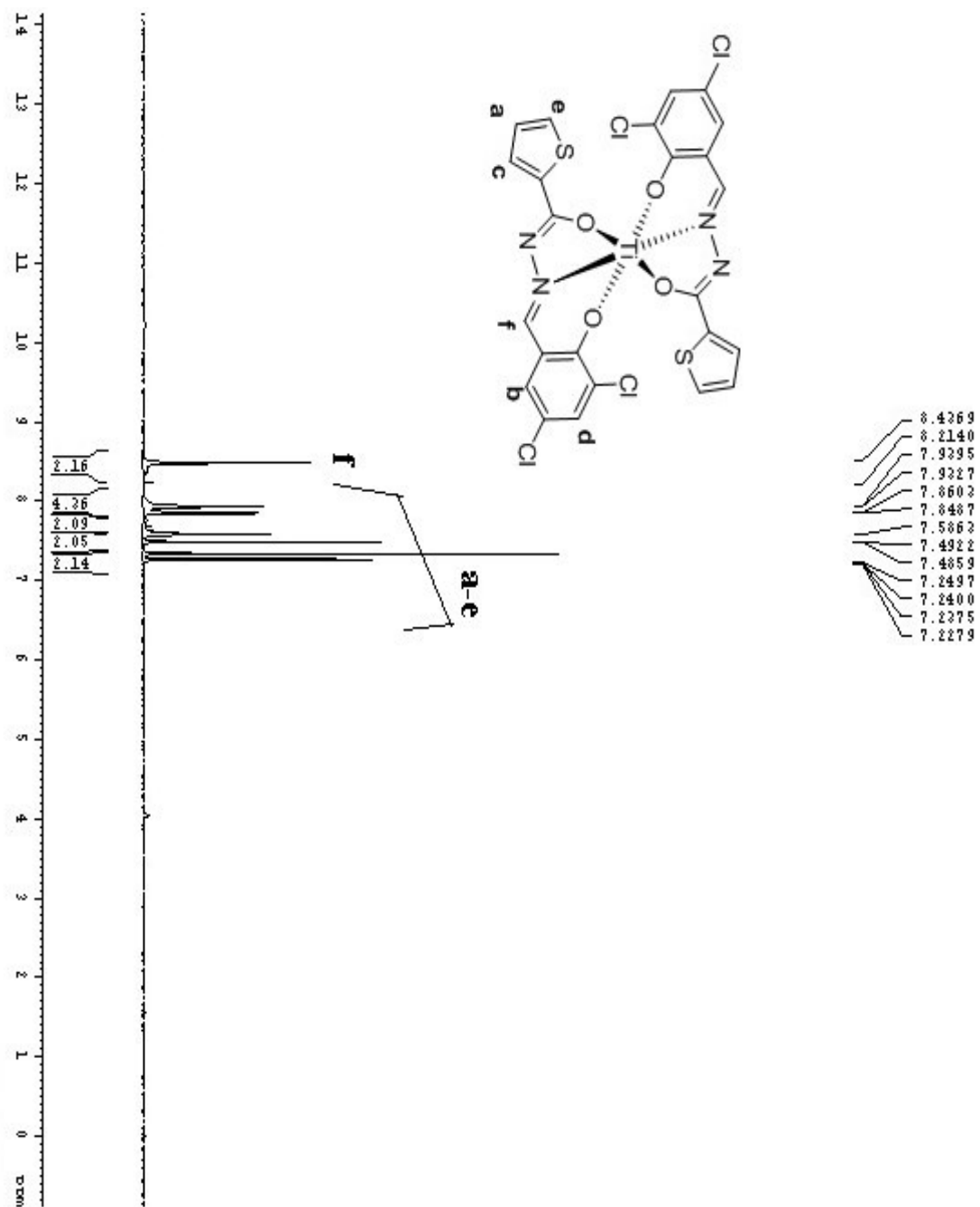


Fig. S16. ¹H NMR (400 MHz, CDCl₃) of Compound 3

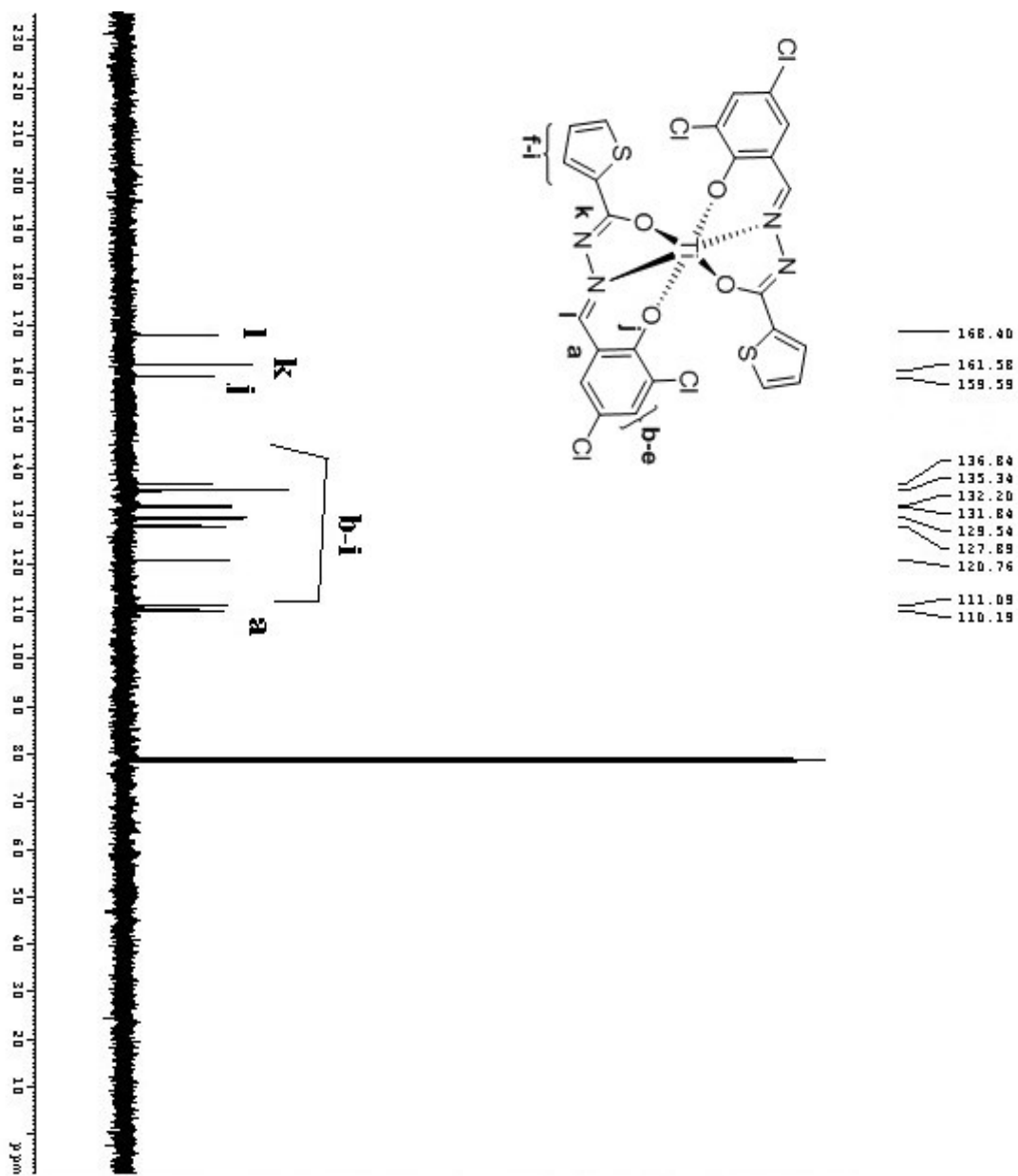


Fig. S17. ¹³C NMR (100 MHz, CDCl₃) of Compound 3

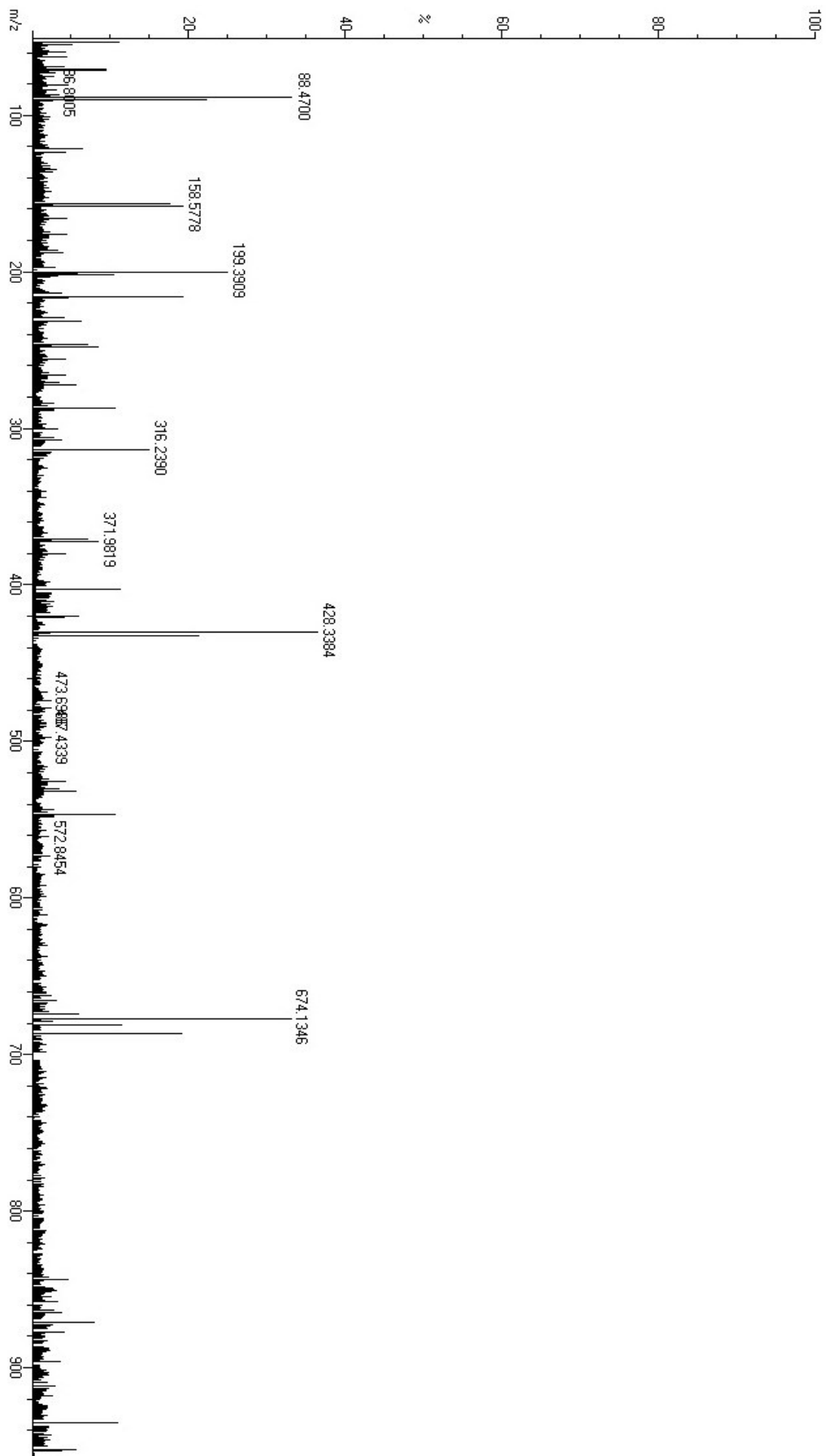


Fig. S18. ESI mass spectrum of Compound 3

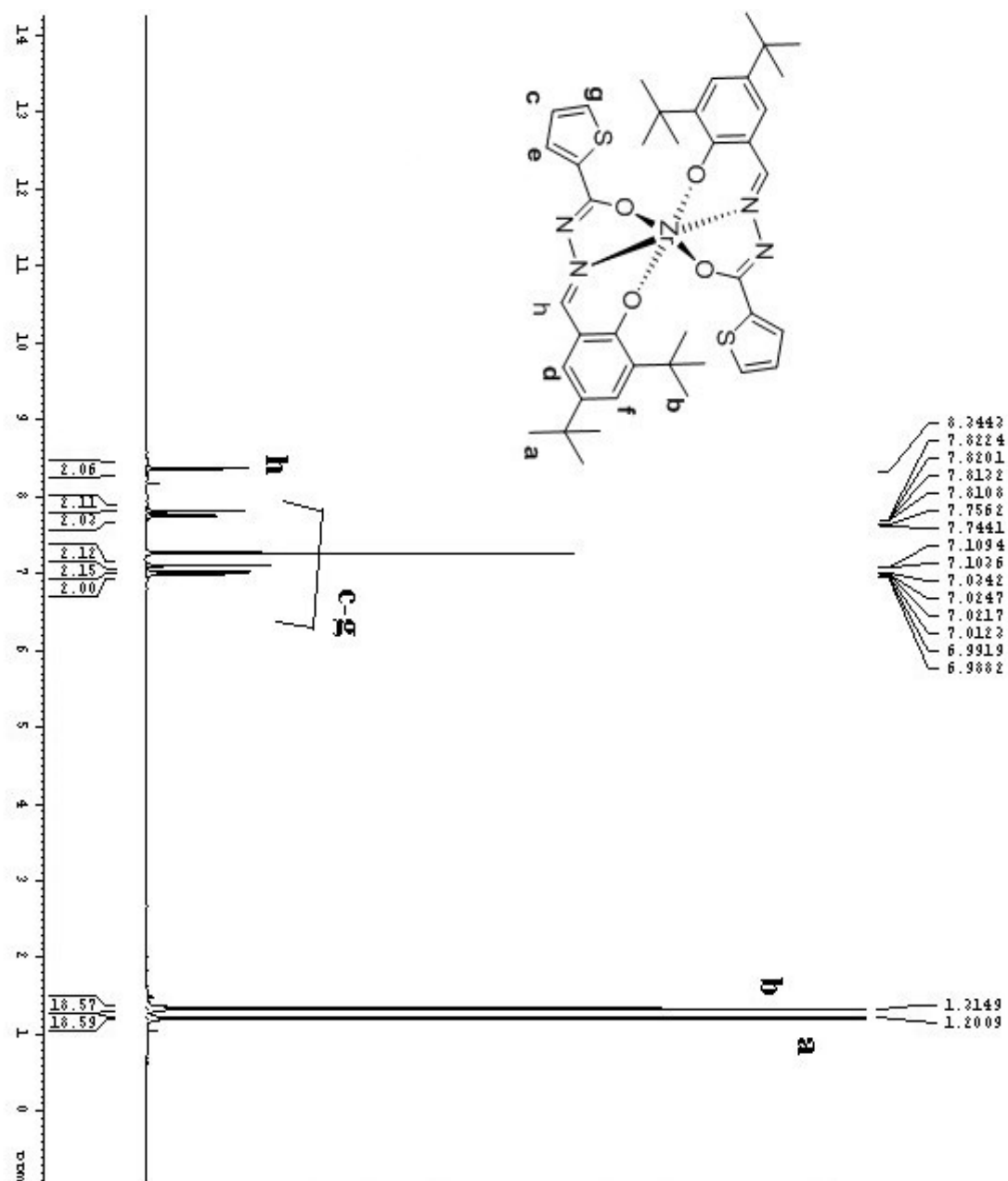


Fig. S19. ¹H NMR (400 MHz, CDCl₃) of Compound 4

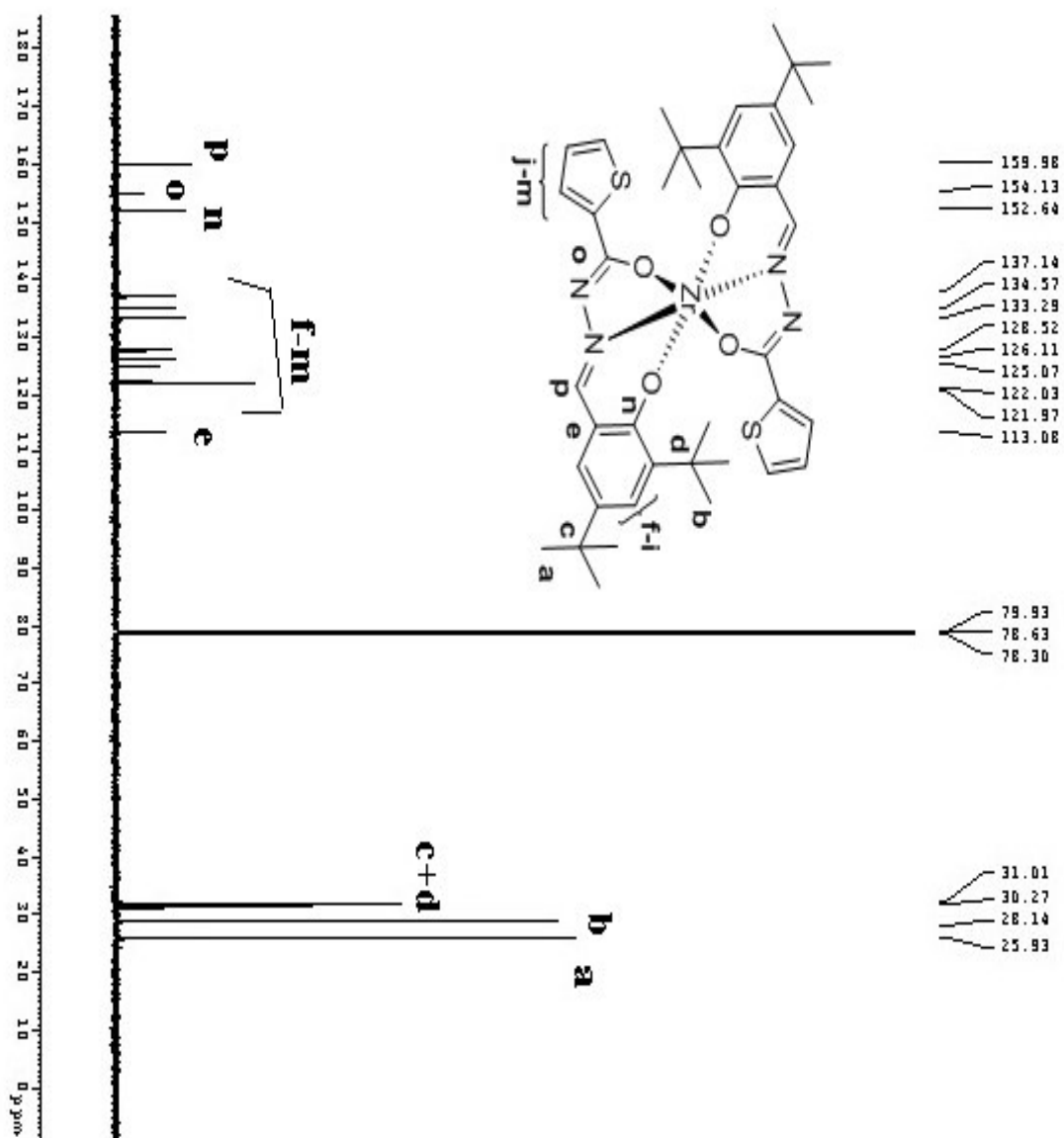


Fig. S20. ¹³C NMR (100 MHz, CDCl₃) of Compound 4

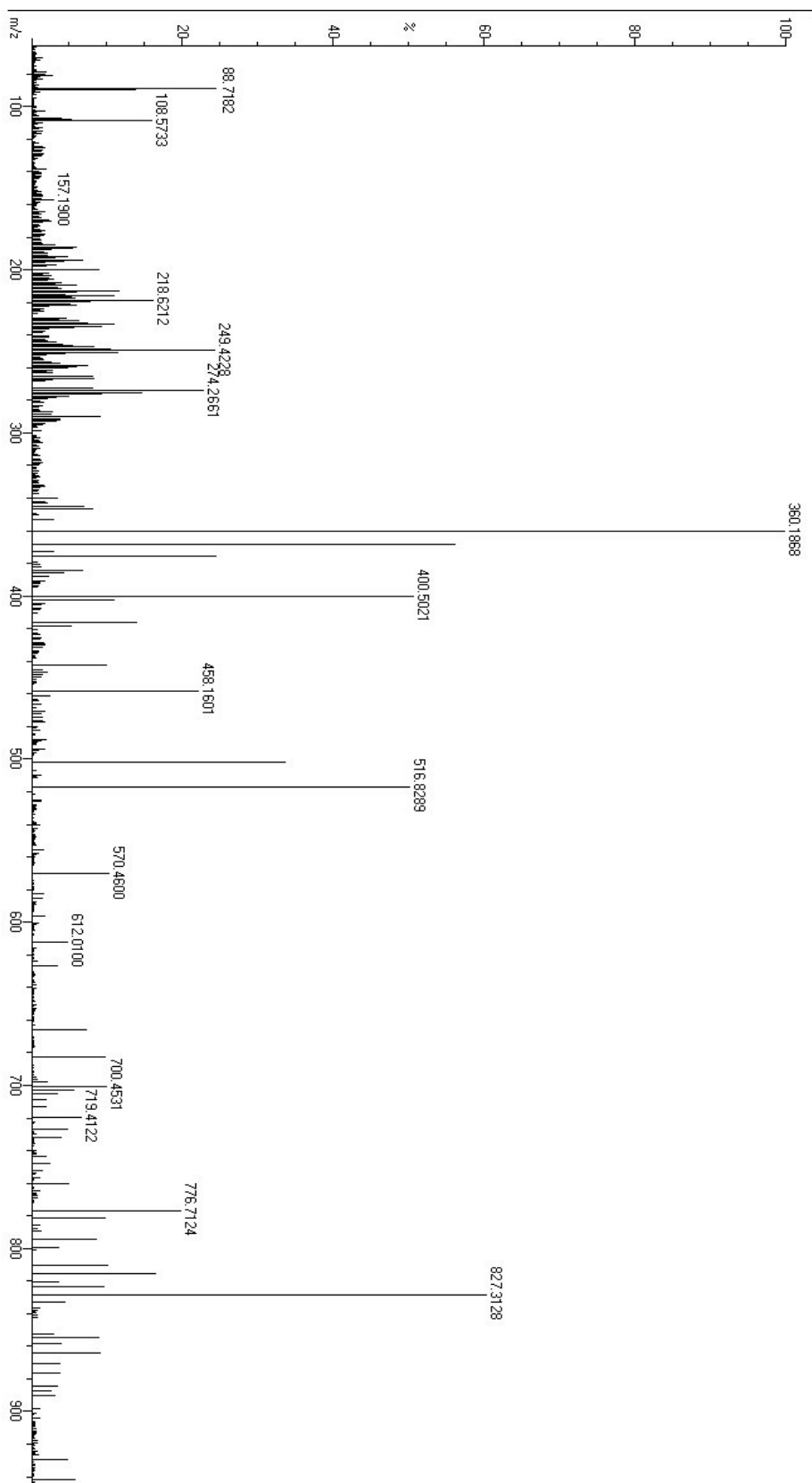


Fig. S21. ESI mass spectrum of Compound 4

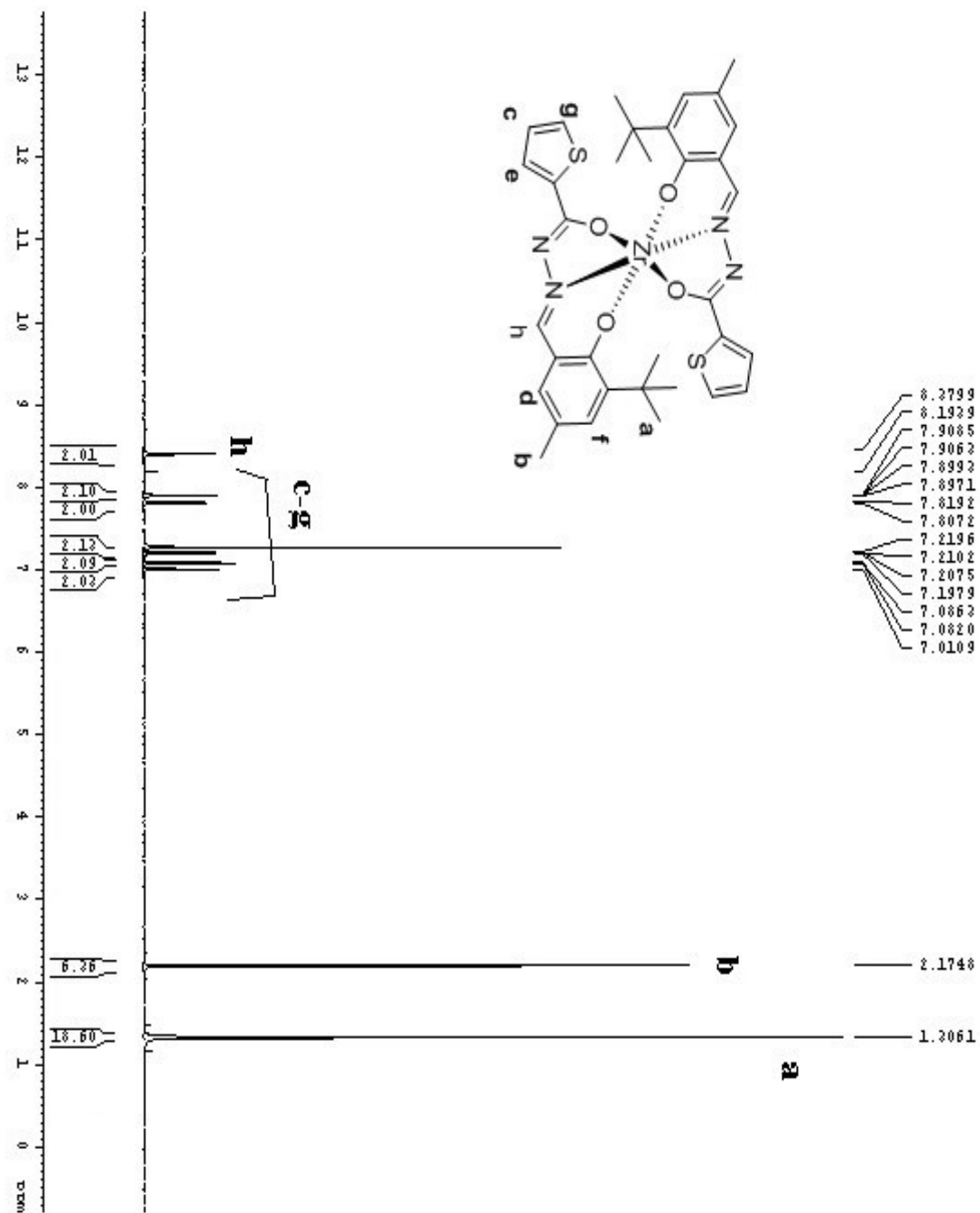


Fig. S22. ¹H NMR (400 MHz, CDCl₃) of Compound 5

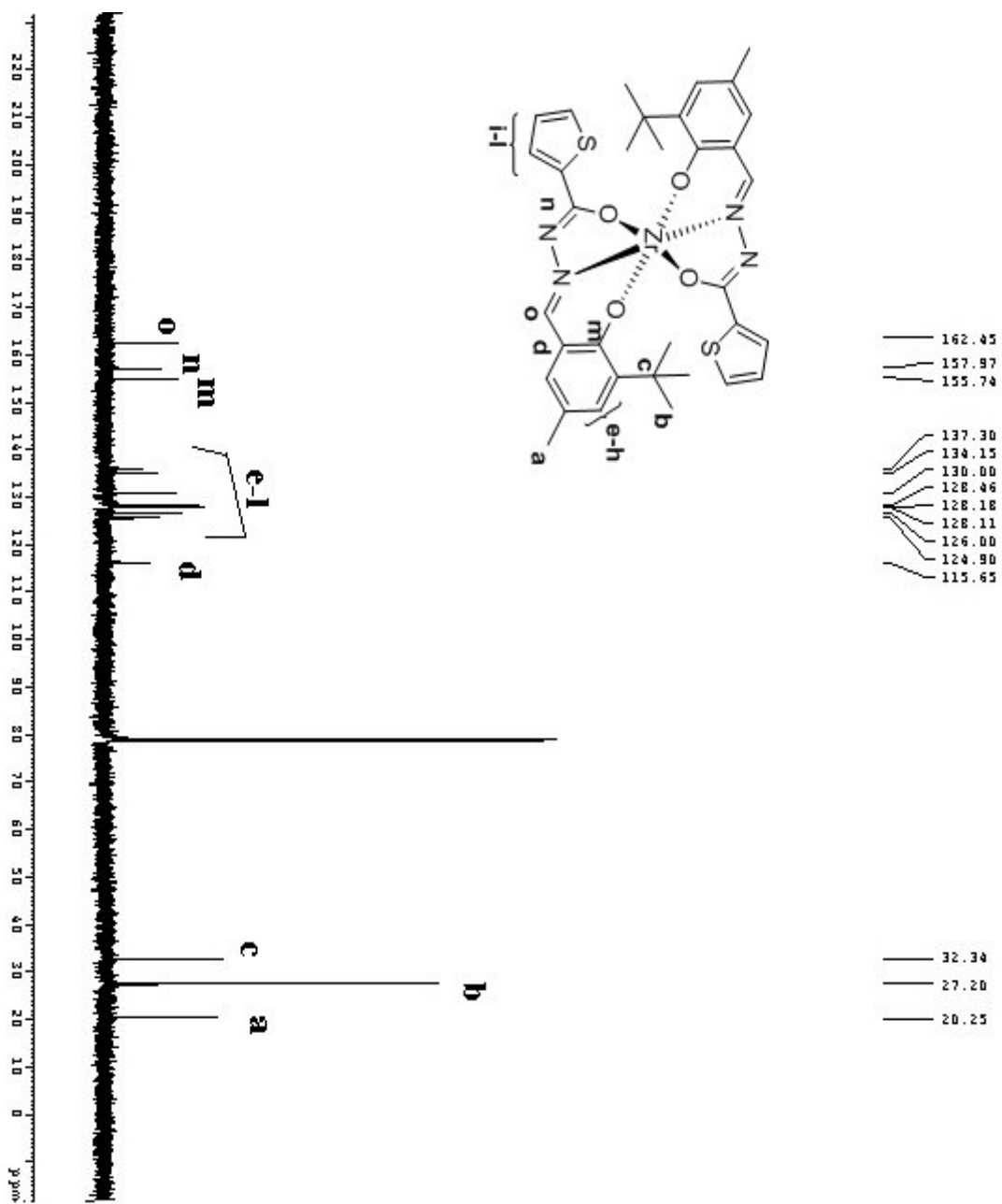


Fig. S23. ¹³C NMR (100 MHz, CDCl₃) of Compound 5

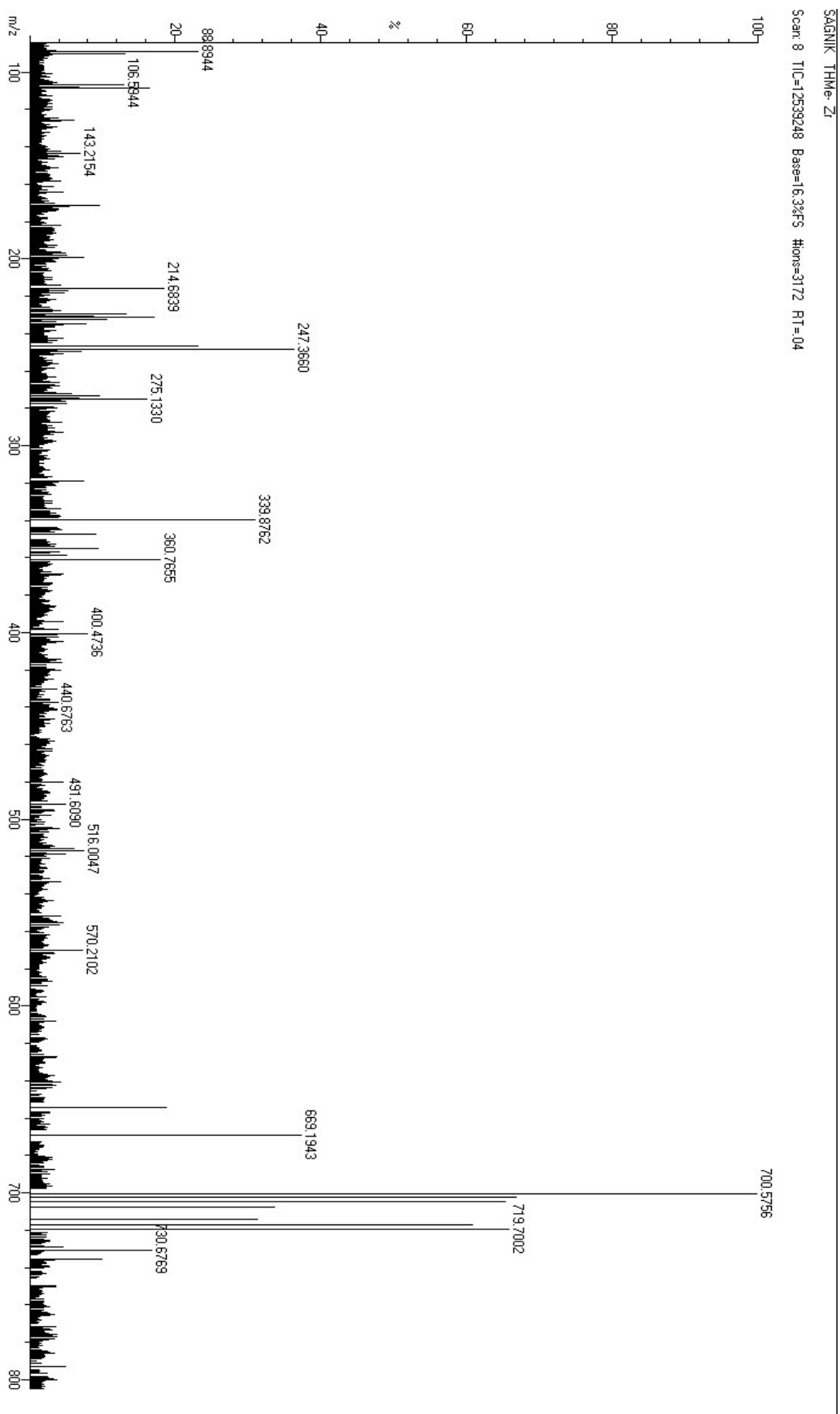


Fig. S24. ESI mass spectrum of Compound 5

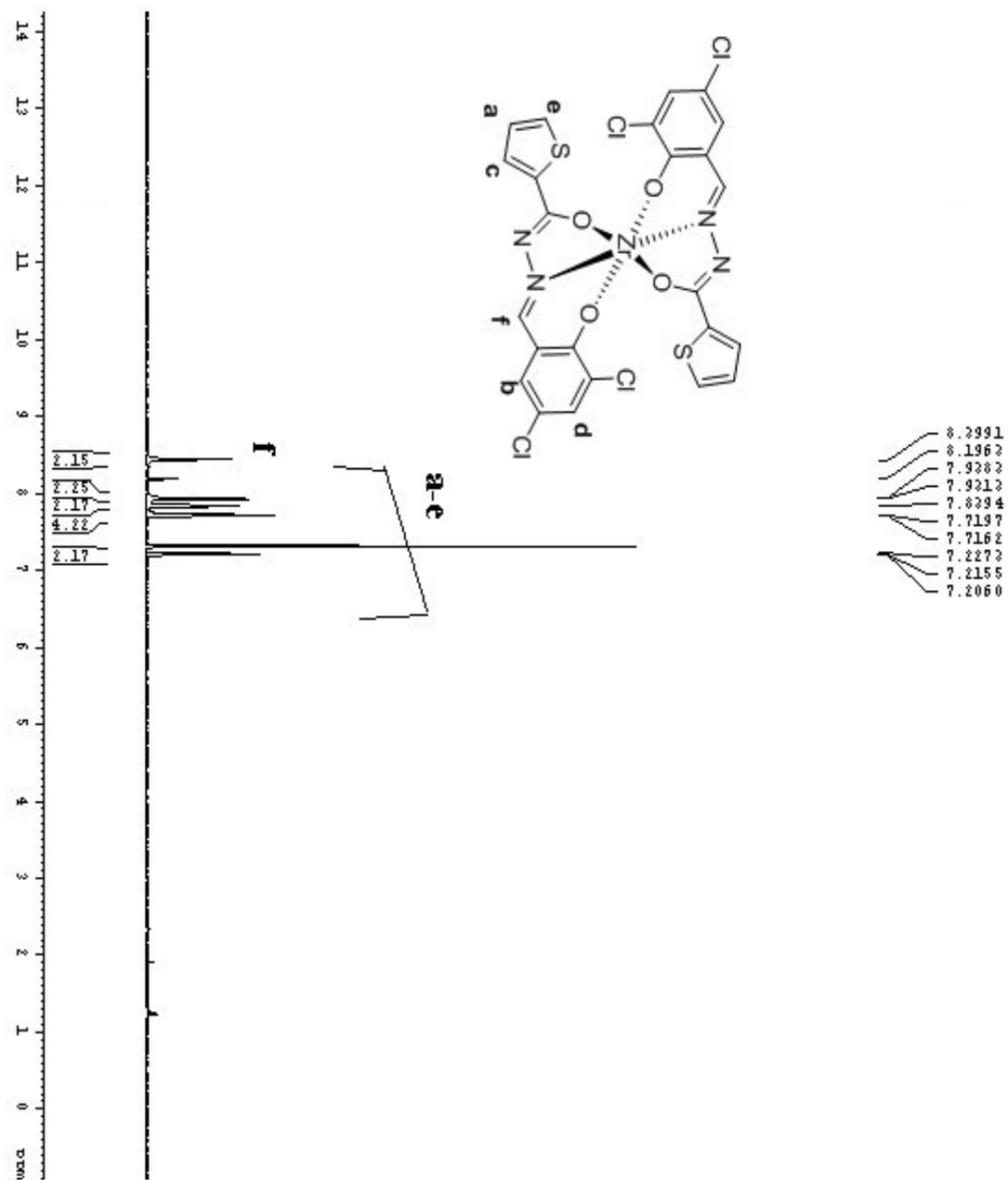


Fig. S25. ¹H NMR (400 MHz, CDCl₃) of Compound 6

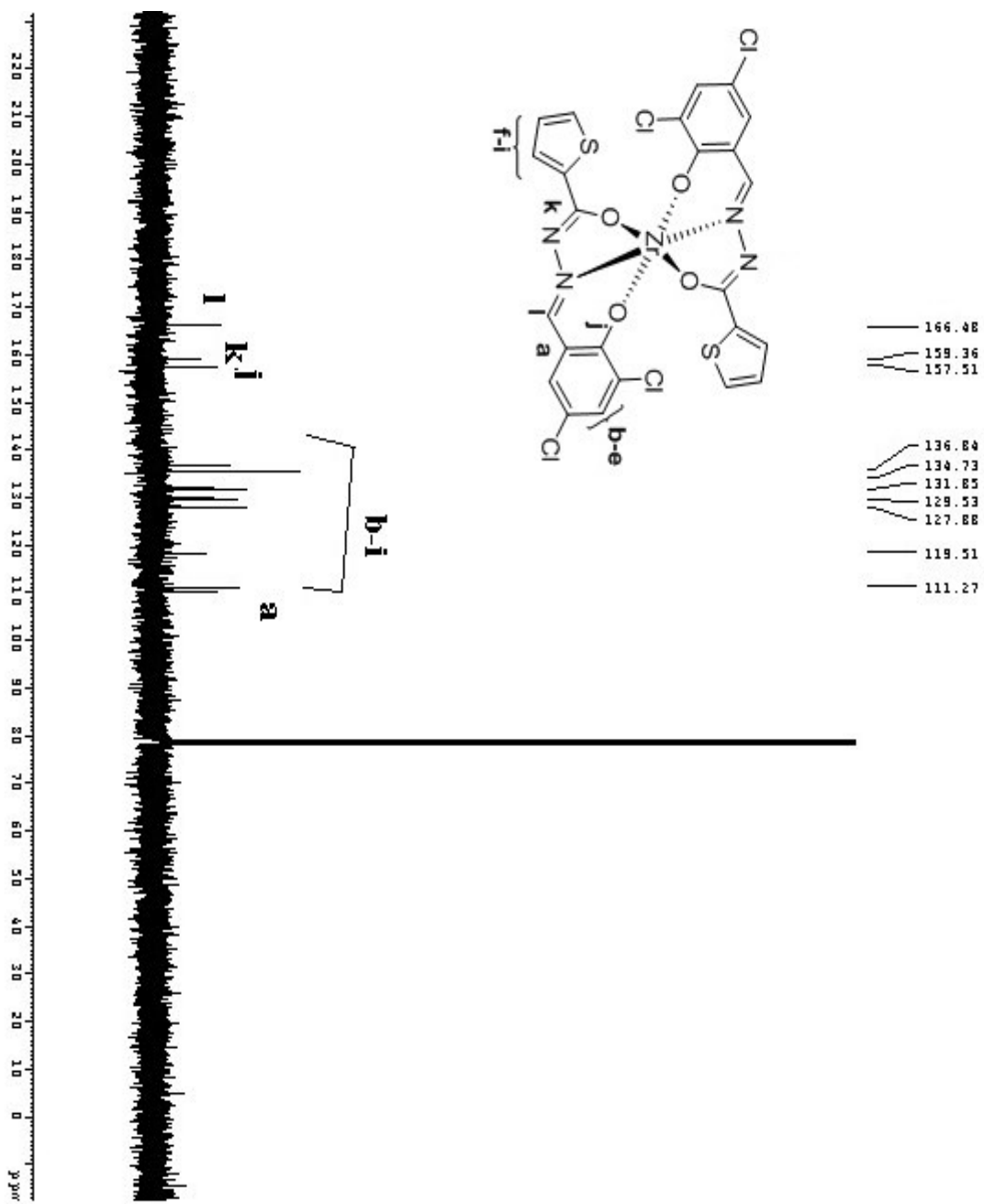


Fig. S26. ¹³C NMR (100 MHz, CDCl₃) of Compound 6

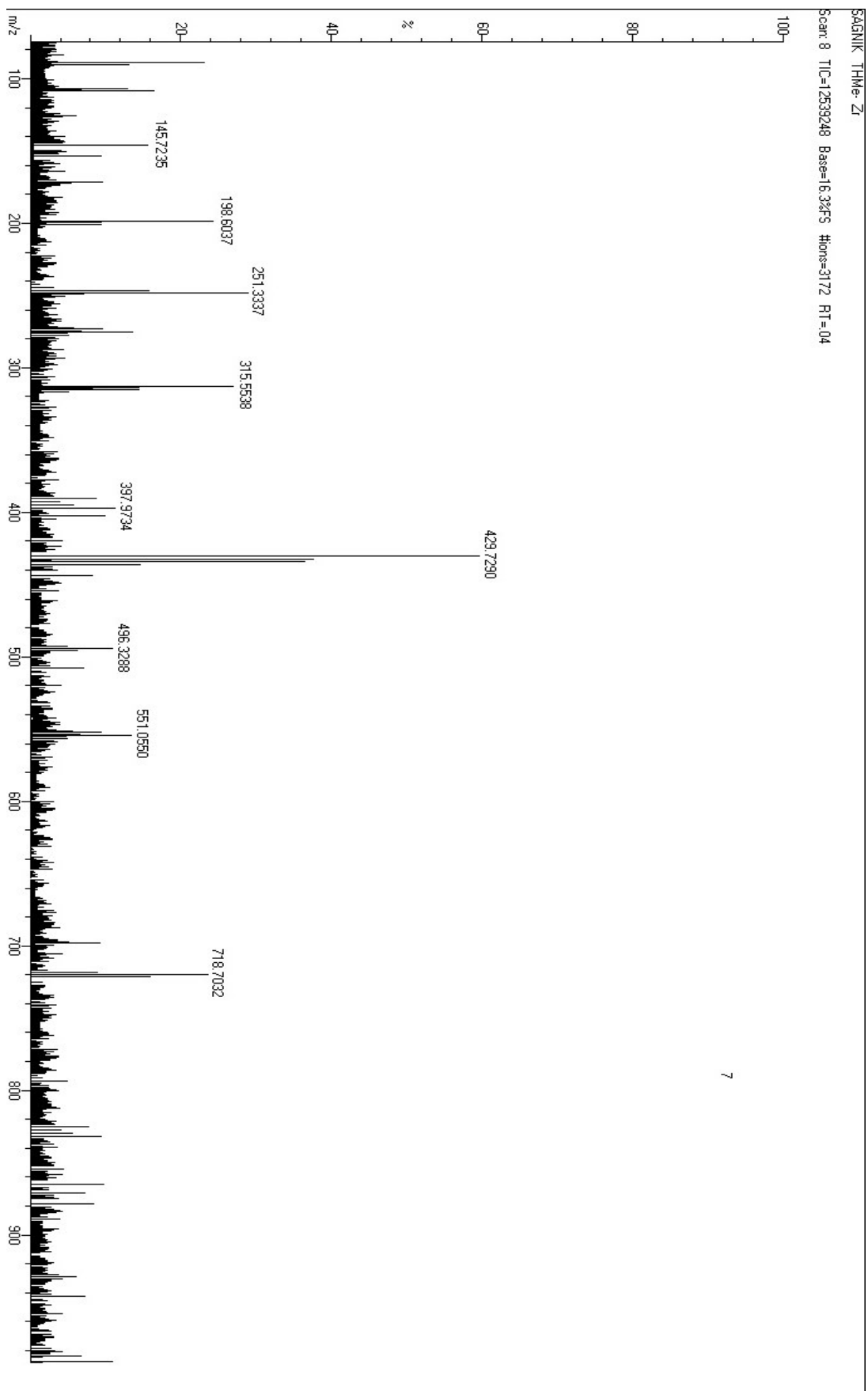


Fig. S27. ESI mass spectrum of Compound 6

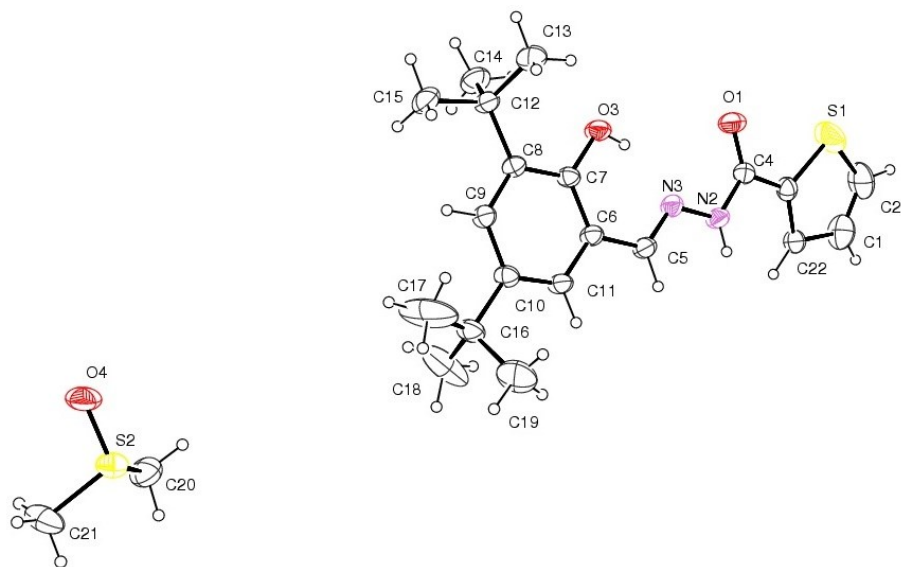


Fig. S28. Molecular structure of L1H₂

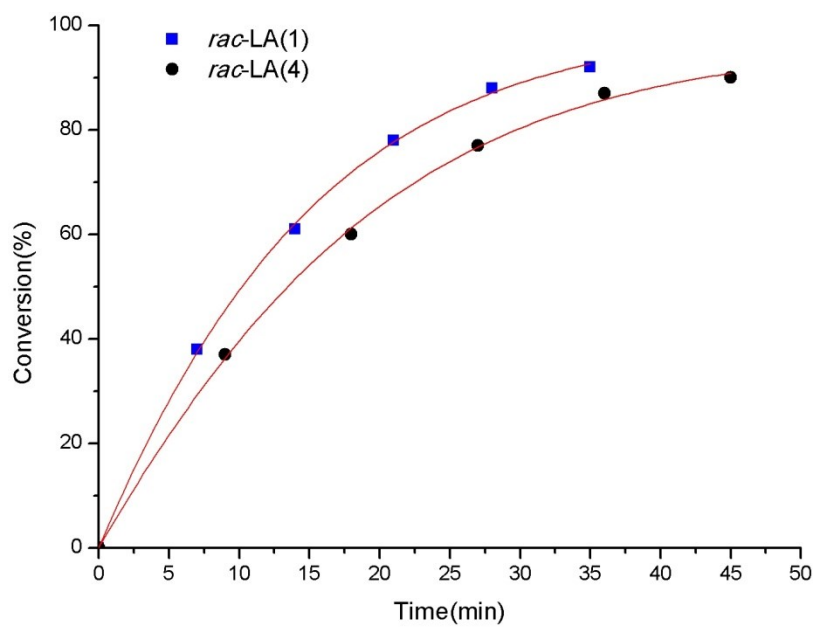


Fig. S29. *rac*-LA conversion vs. time plot using **1** and **4**: $[M]_0/[Cat]_0 = 200$ at 140 °C

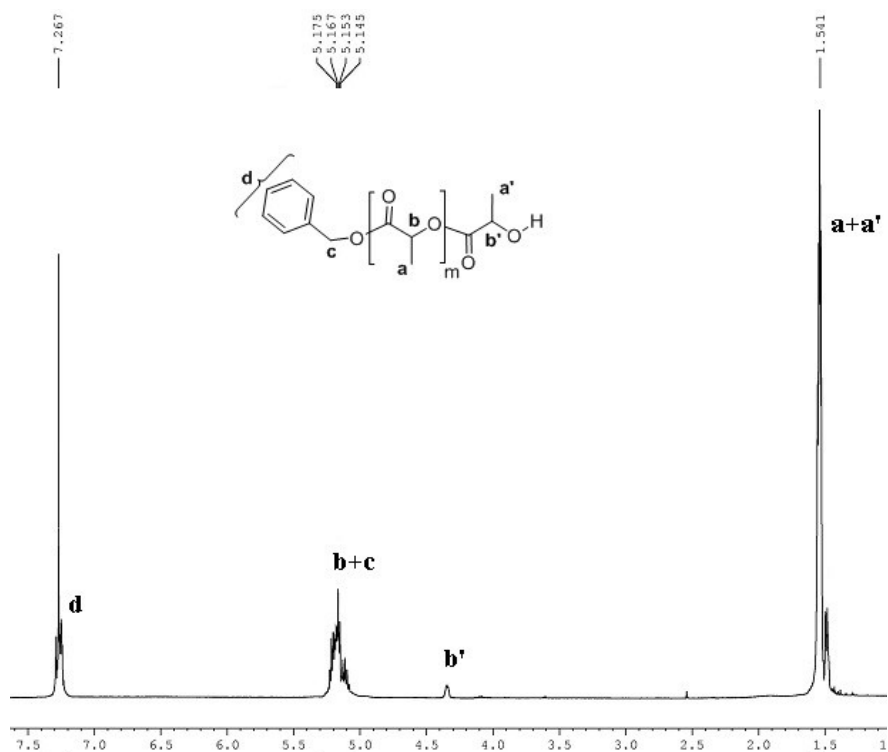


Fig. S30. ^1H NMR spectrum (400 MHz, CDCl_3) of the crude product obtained from a reaction between *rac*-LA and **1** and BnOH in 15: 1: 5 ratios at 140 °C.

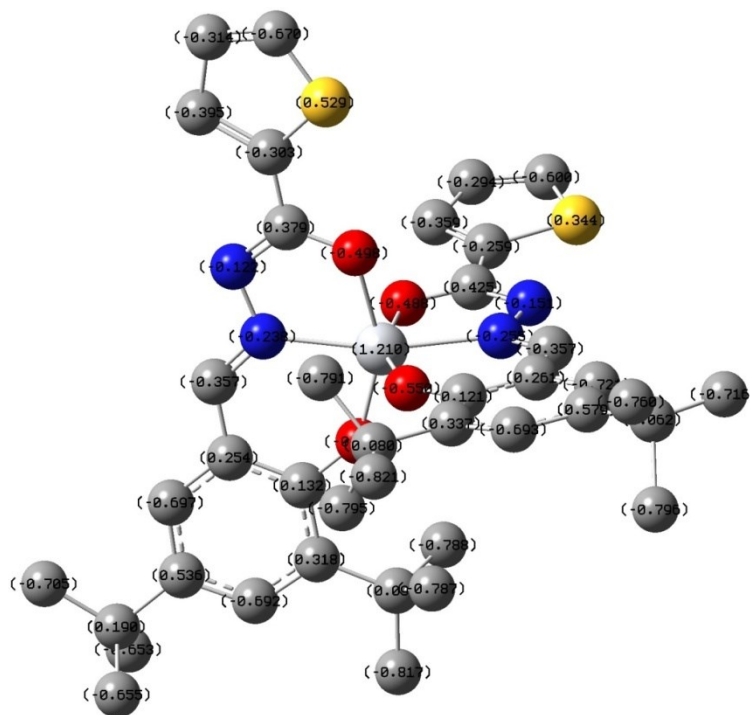


Fig. S31. Net Mulliken charge analysis of the optimized geometry of **1**.

Table S1. Crystal data for the structure **L1H₂, L2H₂, 1, 2 and 4**

Compound	L1H ₂ .DMSO	L2H ₂ .DMSO	1	2	4
Molecular formula	C ₂₂ H ₃₂ N ₂ O ₃ S ₂	C ₁₉ H ₂₆ N ₂ O ₃ S ₂	C _{40.03} H _{46.03} N ₄ O ₄ S ₂ Ti	C ₃₄ H ₃₆ N ₄ O ₄ S ₂ Ti	C ₄₀ H ₄₈ N ₄ O ₄ S ₂ Zr
Formula weight	436.61	394.54	758.83	676.69	804.16
T/K	296(2)	296(2)	296(2)	296(2)	296(2)
Wavelength (Å)	0.71073	0.71073	0.71073	0.71073	0.71073
Crystal system	Triclinic	Triclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P1</i>	<i>P1</i>	<i>P2₁/c</i>	<i>P2₁/n</i>	<i>P2₁/c</i>
a/Å	9.2586(2)	9.8710(3)	17.7190(9)	13.081(4)	17.8233(9)
b/Å	9.6167(2)	9.8841(3)	10.8186(5)	23.840(8)	10.9084(6)
c/Å	15.5862(4)	10.9592(3)	21.7958(11)	15.426(5)	21.6049(11)
α (°)	89.2238(12)	98.4485(11)	90	90	90
β (°)	75.0508(11)	94.8760(11)	93.293(2)	101.148(12)	94.0606(18)
γ (°)	65.7927(10)	94.1161(12)	90	90	90
V/Å ³	1216.17(5)	1049.95(5)	4171.2(4)	4720(3)	4190.0(4)
Z, Calculated density(Mg/ m ³)	2, 1.192	2, 1.248	4, 1.208	4, 0.952	4, 1.275
Absorption coefficient (mm ⁻¹)	0.242	0.273	0.347	0.301	0.404
Reflections collected/Independent reflections	16707 / 4271	15632 / 3699	14548 / 3837	13883 / 4625	31084 / 7376
Data/restraints/parameters	4271 / 0 / 278	3699 / 0 / 249	3837 / 68 / 535	4625 / 6 / 424	7376 / 50 / 514
Goodness of fit on F ²	1.037	1.042	1.057	1.111	1.077
Final R indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0737, <i>wR</i> ₂ = 0.2182	<i>R</i> ₁ = 0.0743, <i>wR</i> ₂ = 0.2076	<i>R</i> ₁ = 0.0471, <i>wR</i> ₂ = 0.1194	<i>R</i> ₁ = 0.0585, <i>wR</i> ₂ = 0.1552	<i>R</i> ₁ = 0.0374, <i>wR</i> ₂ = 0.0951
R indices (all data)	<i>R</i> ₁ = 0.0850, <i>wR</i> ₂ = 0.2338	<i>R</i> ₁ = 0.0832, <i>wR</i> ₂ = 0.2207	<i>R</i> ₁ = 0.0730, <i>wR</i> ₂ = 0.1493	<i>R</i> ₁ = 0.0769, <i>wR</i> ₂ = 0.1676	<i>R</i> ₁ = 0.0569, <i>wR</i> ₂ = 0.1113
CCDC	1562812	1562816	1562817	1562818	1562819

$$R_1 = \frac{\sum |F_o| - |F_c|}{\sum |F_o|}, wR_2 = \left[\frac{\sum (F_o^2 - F_c^2)^2}{\sum w(F_o^2)^2} \right]^{1/2}$$