Supporting information

Cubane Cu₄I₄(phosphine)₄ Complexes as New Coinitiator for Free Radical Photopolymerization: Towards Aromatic Amine-Free Systems

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Figure S1. ¹H NMR (500 MHz, CDCl₃, 20 °C) recorded for compound Cu2.



Figure S2. ¹³C NMR (125 MHz, CDCl₃, 20 °C)) recorded for compound Cu2.



Figure S3. ³¹P NMR (202 MHz, CDCl₃, 20 °C) recorded for compound Cu2.



Figure S4. FT-IR spectrum of compound Cu2 in the solid state.



Figure S5. X-Ray powder diffraction analysis recorded for a solid-state sample of compound **Cu2**.



Figure S6. ATG analysis for a sample of compound Cu2.



Figure S7. DSC analysis recorded for a solid-state sample of compound Cu2.



Figure S8. The emission spectra of the LED centered at 405 nm and 455 nm.



Figure S9. (A) IR spectra recorded before and after polymerization when using ITX\Cu2 (1%\1% w\w) system in TMPTA for thick sample, upon exposure to LED light λ =405 nm. (B) IR spectra recorded before and after polymerization when using ITX\Cu2 (1%\1% w\w) system in TMPTA for thin sample, upon exposure to LED light (λ =405 nm).



Figure S10. Determination of the Stern Volmer quenching data.

Table S1. G' and G'' for TMPTA based polymer (from Figure 9).

	G' (MPa)	G'' (MPa)
ITX/EDB	395	7.9
ITX/Cu2	304	8.5