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**Electronic Supplementary Information** 

# Propylene Homopolymerization and Copolymerization with Ethylene by Acenaphthene-Based α-Diimine Nickel Complexes to Access EPR-Like Elastomers

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## S1. Nickel-catalyzed polypropylenes

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Table S1 Calculated and observed carbon-13 chemical shifts and assignments for nickel catalyzed polypropylene.

Peak	Chemical shift					
no.	Exptl.	Calcd.	Sequence	car	bon	
1	11.26	11.36	E <u>Et</u> E	$1B_2$		
2	13.99	13.86	ELE	$1B_n$		
3	15.17	16.64	P <u>P</u> *P	$1B_1$	Ραγ	
4	16.49	17.13	P <u>P</u> P*	$1B_1$	$P_{\alpha\beta}$	
5	19.51-20.43	20.61	P <u>P</u> P <sub>rr</sub>	$1B_1$	$P_{\beta\beta}$	
	20.10	20.61	PPPPPrrrr	$1B_1$	$\mathbf{P}_{\beta\beta}$	
	19.96	20.61	PPPPPrrrm	$1B_1$	$\mathbf{P}_{\beta\beta}$	
	19.88	20.61	PP <u>P</u> PP <sub>mrrm</sub>	$1B_1$	$\mathbf{P}_{\beta\beta}$	
		19.63	E <u>P</u> E	$1B_1$	$P_{\delta\delta}$	
		19.63	E <u>P</u> *PE+EP* <u>P</u> E	1,4-1B <sub>1</sub>	$P_{\gamma\delta}P_{\gamma\gamma}$	
6	20.43-21.16					
	20.98	20.61	PPPPPmmrr	$1B_1$	$\mathbf{P}_{\beta\beta}$	
	20.78	20.61	PPPPPmrmm+rmrr	$1B_1$	$\mathbf{P}_{\beta\beta}$	
	20.64	20.61	PP <u>P</u> PP <sub>mrmr</sub>	$1B_1$	$\mathbf{P}_{\beta\beta}$	
		20.12	EPP+PPE	$1B_1$	$P_{\beta\delta}$	
			P* <u>P</u> P	1,4-1B <sub>1</sub>	$P_{\beta\gamma}$	
7	21.20-21.82	20.61	$P\underline{P}P_{mm}$	$1B_1$	$P_{\beta\beta}$	
	21.72	20.61	PP <u>P</u> PP <sub>mmmm</sub>	$1B_1$	$P_{\beta\beta}$	
	21.53	20.61	PP <u>P</u> PP <sub>mmmr</sub>	$1B_1$	$\mathbf{P}_{\beta\beta}$	
	21.35	20.61	PPPPPrmmr	$1B_1$	$P_{\beta\beta}$	
8	22.98-23.82					
	23.49	22.65	E <u>L</u> E	$2B_n$		
	23.26	22.62	E <u>iBu</u> E	$1B_{iBu}$		
	23.15	22.13	E <u>2MH</u> E	$1B_{2ME}$		
9	24.73	24.58	P <u>E</u> P	$\beta' B_1$	$S_{\beta\beta}$	
10	25.96	25.92	E <u>iBu</u> E	$2B_{iBu}$		
11	27.00-27.50					
	27.34	27.27	P <u>E</u> E+E <u>E</u> P PP <u>E</u> E+E <u>E</u> PP <sub>r</sub> PP <u>E</u> E+ <u>E</u> EPP <sub>m</sub> EP <u>E</u> E+E <u>E</u> PE	$\beta B_1$	$S_{\beta\delta}$	
			E <u>iBu</u> E	$\beta B_{iBu}$		
		27.52	ELE	$\beta B_n + (n-1)B_n$		
		27.77	E <u>Et</u> E	$\beta B_2$		
		27.16	E <u>Et</u> E	$2B_2$		
		27.27	E <u>2MH</u> E	$4B_{2ME}$		
12	27.50-27.98				_	
	27.72	27.52		1,6-β'B <sub>1</sub>	$\mathbf{S}_{\boldsymbol{\beta}\boldsymbol{\gamma}}$	
13	28.18	27.77	E <u>2MH</u> E	$5B_{2ME}$		
14	28.63	28.38	P <u>P</u> P	$brB_1$	$T_{\beta\gamma}$	
15	29.47	29.71	ELE	$4B_n$		
16	29.92	29.96	EEE		$S_{\delta\delta}$	
17	30.34	30.21	EE <u>E</u> P	$\gamma B_1$	$S_{\gamma\delta}$	
18	30.63-31.47				_	
	30.91	30.45	E <u>P</u> P+P <u>P</u> E	$brB_1$	$T_{\beta\delta}$	
		30.45	P* <u>P</u> P	_	$T_{\beta\gamma}$	
		30.21	ELE	$\gamma B_n + (n-2)B_n$		
		30.21	E <u>2MH</u> E	$\gamma B_{\rm 2MH}$		
		30.21	E <u>iBu</u> E	$\gamma B_{iBu}$		

$\begin{array}{cccccccccccccccccccccccccccccccccccc$			30.21	E <u>Et</u> E	$\gamma B_{Et}$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	19	33.22	32.52	E <u>P</u> E	$brB_1$	$T_{\delta\delta}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	33.66	32.52	EP*PE+EP*PE	$1,4$ -br $B_1$	$T_{\gamma\delta} T_{\gamma\gamma}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	21	34.66	34.47	$P*\underline{P}*PP+P*P*\underline{P}P$	1,4-α'B <sub>1</sub>	$S_{\gamma\alpha\beta\delta}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	22	35.10	34.47	ELE	$\alpha B_n + nB_n$	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			34.47	E <u>2MH</u> E	$\alpha B_{2MH}$ +6 $B_{2MH}$	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			34.47	E <u>iBu</u> E	$\alpha B_{iBu}$	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			34.47	E <u>Et</u> E	$\alpha B_{Et}$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			34.22	E <u>P</u> *PE+EP* <u>P</u> E		$S_{\delta\alpha\beta\delta}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	23	35.61	34.98	EiBuE	$brB_{iBu}$	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			34.99	P <u>P</u> P*	$1,2$ -br $B_1$	$T_{\alpha\beta}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	24	35.92	34.72	P <u>P</u> *PP	1,4-α'B1	$S_{\gamma\alpha\beta\delta}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	25	37.10-37.99				
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			36.91	E <u>P</u> E+EP <u>E</u> E	$\alpha B_1$	$S_{\alpha\delta}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			37.16	PP <u>E</u> E+EE <u>P</u> P+EP <u>E</u> P	$\alpha B_1$	$S_{\alpha\delta}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			37.41	PP <u>E</u> Pmm+PE <u>P</u> Pmm	$\alpha \gamma B_1$	$S_{\alpha\gamma}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			37.06	P <u>P</u> *P	$1,2$ -br $B_1$	$T_{\alpha\gamma}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			37.05	E <u>L</u> E+E <u>2MH</u> E	$brB_n + brB_{2MH}$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	38.36	37.16	P <u>E</u> PEr	α'B <sub>1</sub>	$S_{\alpha\gamma}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				EE <u>P</u> Pr	$\alpha' B_1$	$S_{\alpha\gamma}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	27	38.51	37.41	PEPP+PPEPmr+rm	$\alpha' B_1$	$S_{\alpha\gamma}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				PEPP+PPEPrr		
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				PEPPmr+rm		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				P <u>E</u> PP <i>rr</i>		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	28	39.43	39.35	E <u>2MH</u> E	$3B_{2ME}$ , $brB_2$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	29	44.01	43.86	EiBuE	$3B_{iBu}$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	30	45.32-48.13			$\alpha \alpha B_1$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			41.17	P*P <u>P</u> P*		$S_{\delta\alpha\alpha\beta}$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			41.42	PP <u>P</u> P*		$S_{\gamma\alpha\alpha\beta}$
44.11         PPPE           44.36         PPPrr			43.86	EP <u>P</u> E		$S_{\alpha\alpha}$
44.36 PPPrr			44.11	PP <u>P</u> E		
			44.36	P <u>P</u> Prr		

### S1.2

#### Equations for quantitatie analysis:

 $[PPP] = k (I_{14})$   $[PPE+EPP] = k (I_{18} - 3 I_2 - 2 I_{28} - 2 I_{29} - 2 I_1)$   $[EPE] = k (I_{19})$   $[EEE] = k (I_{16}/2)$   $[PEE+EEP] = k (I_{11} - I_{28} - 3I_1 - 3I_2 - 2I_{29})$   $[PEP] = k (I_9)$   $[PP*P] = k (I_9)$   $[PP*P] = k (I_4)$   $[PP*PP] = k (I_{24})$   $[P*P*PP] = k ((I_{21})/2)$   $[1,4-B_1] = k ((I_{20})/2)$ 

$$[1,6-\beta B_1] = k ((I_{12})/2)$$
  
[ELE] = k (I<sub>2</sub>)  
[EEtE] = k (I<sub>1</sub>)  
[EiBuE] = k ((I<sub>10</sub> + I<sub>29</sub>)/2)  
[E2MHE] = k (I<sub>28</sub>)/2

where k is the normalization constant.

### S2. Nickel catalyzed copolymerization of ethylene and propylene



S2.1 Kinetic studies of *rac*-Ni2 catalyzed copolymerization of ethylene and propylene.

**Figure S1**. Plots of  $M_n$  (**a**) and  $M_w/M_n$  (**b**) versus polymerization time for propylene and ethylene copolymerization

#### S2.1

Table S2.	Carbon-13	3 chemical	shifts ir	1 previous	and	present	researc	hes an	d their	assignm	ents f	or n	lick	el
catalyzed	ethylene/p	ropylene co	opolym	er.										

Peak	Chemical shift (ppm)		G		Carbon	
no.	Exptl.	Calcd.	Sequence	Carbon type	type	
1	11.27	11.36	E <u>Et</u> E	1B <sub>2</sub>		
2	13.99	13.86	ELE	$1B_n$		
3	14.46	14.35	E <u>Pr</u> E	1B <sub>3</sub>		
4	19.65	19.63	EPE	$1B_1$		
5	20.07	19.63	$E\underline{P}*PE+EP*\underline{P}E$	$1B_1$	$P_{\delta\delta}$	
	20.49	20.12	$\underline{EPP} + \underline{PPE}$	$1B_1$	$P_{\beta\delta}$	
		20.21	E <u>Pr</u> E	2B <sub>3</sub>		
6	22.51	22.65	ELE	$2B_n$		
7	23.12	23.13	E <u>2MH</u> E	$1B_{2MH}$		
8	23.24	22.62	E <u>iBu</u> E	$1\mathbf{B}_{i\mathbf{B}\mathbf{u}}$		
9	24.78	24.58	P <u>E</u> P	$\beta' \mathbf{B}_1$	$S_{\beta\beta}$	
10	26.00	25.92	E <u>iBu</u> E	$2B_{iBu}$		
11	27.06	27.52	ELE	$\beta B_n + (n-1)B_n$		

12	27.18-28.15				
	27.40	27.27	P <u>E</u> E+E <u>E</u> P	$\beta \mathbf{B}_1$	$S_{\beta\delta}$
			EPEE+EEPE		
			PPEE+EEPP		
			E <u>iBu</u> E	$\beta \mathbf{B}_{iBu}$	
		27.77	E <u>Et</u> E	$\beta B_2$	
		27.16	E <u>Et</u> E	$2B_2$	
		27.27	E <u>2MH</u> E	$4B_{2MH}$	
13	27.80	27.52		1,6-β'B <sub>1</sub>	$S_{\beta\gamma}$
14	28.20	27.77	E <u>2MH</u> E	$5B_{2MH}$	
15	29.50	29.71	ELE	$4B_n$	
16	29.95	29.96	E <u>E</u> E		$S_{\delta\delta}$
17	30.22-30.67				
		30.45	E <u>P</u> P+P <u>P</u> E	$brB_1$	$T_{\beta\delta}$
		30.21	E <u>L</u> E	$\gamma B_n + (n-2)B_n$	
		30.21	E <u>2MH</u> E	$\gamma B_{2MH}$	
		30.21	E <u>iBu</u> E	$\gamma B_{iBu}$	
		30.21	E <u>Et</u> E	$\gamma B_{\rm Et}$	
18	31.71-32.27				
	33.25	32.40	ELE	$3B_n$	
19	33.31	32.52	EPE	$brB_1$	$T_{\delta\delta}$
20	33.69	32.52	$\underline{EP}*PE+EP*\underline{P}E$	$1,4$ -br $B_1$	$T_{\gamma\delta},T_{\gamma\gamma}$
21	34.06-35.59				
		34.22	$\underline{EP}^{PE+EP*PE}$		$S_{\alpha\beta}$
		34.47	ELE	$\alpha B_n \!\!+\! n B_n$	
		34.47	E <u>2MH</u> E	$\alpha B_{2MH} + 6B_{2MH}$	
		34.47	E <u>iBu</u> E	$\alpha B_{iBu}$	
		34.47	E <u>Et</u> E	$\alpha B_{Et}$	
		34.98	E <u>iBu</u> E	$brB_{iBu}$	
22	37.00	36.91	E <u>Pr</u> E	$3B_3$	
23	37.41-38.14				
	37.55	36.91	EPE	$\alpha B_1$ , 1,4- $\alpha B_1$	$S_{\alpha\delta}$
		37.05	ELE+ E <u>2MH</u> E	$brB_n\!\!+\!\!brB_{2MH}$	
		37.05	E <u>Pr</u> E	$brB_3$	
24	38.25-38.82	37.16	P <u>E</u> PE	$\alpha' B_1$	$\mathbf{S}_{lpha\gamma}$
			EE <u>P</u> P	$\alpha' B_1$	$\mathbf{S}_{lpha\gamma}$
25	39.50	39.35	E <u>2MH</u> E	$3B_{2ME}$	
		39.12	E <u>Et</u> E	$brB_2$	
26	44.00	43.86	E <u>iBu</u> E	$3\mathbf{B}_{i\mathrm{Bu}}$	
27	45.30-46.44	43.86	EP <u>P</u> E		$S_{\alpha\alpha}$

### S2.3 Equations for quantitatie analysis:

[PPP] triad, not observed

$$[PPE+EPP] = k (I_{27})$$
  

$$[EPE] = k (I_{19})$$
  

$$[EEE] = k (I_{16}/2)$$
  

$$[PEE+EEP] = k (I_{12} - 3I_1 - 2I_8 - 2I_7)$$
  

$$[PEP] = k (I_9)$$
  

$$[EP*PE] = k (I_9)$$
  

$$[EP*PE] = k (I_{20}) / 2)$$
  

$$[1,6-\beta'B_1] = k ((I_{13})/2)$$
  

$$[ELE] = k (I_2)$$
  

$$[ELE] = k (I_2)$$
  

$$[EEEE] = k (I_1)$$
  

$$[EPrE] = k (I_3)$$
  

$$[EiBuE] = k (I_8)$$
  

$$[E2MHE] = k (I_7)$$

where k is the normalization constant.



Figure S2. <sup>13</sup>C NMR spectra of nickel-catalyzed polypropylenes. (carbon assignments can be found in

Table S1)



**Figure S3**. <sup>13</sup>C NMR spectra of *rac*-**Ni5**-catalysed ethylene/propylene copolymers. (carbon assignments can be found in Table S1)



**Figure S4**. <sup>13</sup>C NMR spectra of entry 1, entry 5 and entry 16 in Table 3. (carbon assignments can be found in Table S1)

Table S3. Quantitative analysis of nickel-catalyzed ethylene/propylene

copolymer.					
Sequence	Entry 1	Entry 5	Entry 16	Entry 17	Entry 18
0	0	0	0	0	0
PPE+EPP	0	0	0	0	0
EPE	6.3	9.9	7.4	4.0	5.0
EEE	50.2	54.6	60.2	59.4	53.9
PEE+EEP	27.3	21.8	23.3	29.8	32.2
PEP	2.8	3.3	1.0	0.8	0.7
EP*PE	2.0	1.4	2.0	1.4	2.2
1,6-β'B <sub>1</sub>	13.7	6.1	3.9	3.1	5.4
long branch	2.7	2.6	1.6	0.8	0.3
ethyl	0.3	0.1	0	0.2	0
propyl	0.1	0	0	0.1	0
iso-butyl	0.1	0	0	0.2	0.1
2-methyl hexyl	0.1	0.2	0	0.2	0.1



Ι

**Scheme S1.** Reported 8-(R)-Naphthyl  $\alpha$ -diimine nickel(II) catalysts.



Figure S5. Single crystal structure of rac-L4·ZnCl<sub>2</sub>.

### 1. Experimental sections

Procedure for the Synthesis of Ligand rac-L1-L5



Step 1: In a 100 ml round-bottom flask, ZnCl<sub>2</sub> (204 mg, 1.5 mmol) and acenaphthenequinone (274 mg, 1.5 mmol) was added. Then glacial acetic acid (5 mL) and 2,4-dimethyl-6-(phenyl(p-tolyl)methyl)benzenamine (904 mg, 3.0 mmol) was also added and the reaction mixture was refluxed under stirring overnight. After the reaction finished, the mixture was cooled to room temperature and a red solid precipitated. The solid was filtered and washed with diethyl ether (3 x 10 mL). Then drying under vacuum, a bright red solid was obtained. Step 2: The products of the previous step was dissolved in dichloromethane (40 mL) and a solution of potassium oxalate (0.553 g, 3 mmol) in water (3 mL) was added. The reaction mixture was stirred vigorously overnight. Until the reaction finished, the organic phase was separated and washed with water (3 x 10 mL), then the organic phase was dried with MgSO<sub>4</sub>. Then the solvent was removed under vacuum to afford the product as an orange powder (Yield: 643 mg, 54.6%).<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.57 (d, 2H), 7.23 – 6.93 (m, 14H), 6.88 – 6.79 (m, 2H), 6.70 (d, 3H), 6.31 (t, 4H), 6.08 (m, 3H), 5.65 (d, 2H), 2.32 (d, 6H), 2.29 (d, 6H), 2.25 (d, 6H).<sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  163.15, 162.92, 146.51, 143.67, 141.98, 140.64, 140.63, 139.78, 138.59, 135.39, 134.18, 133.17, 133.13, 132.67,

132.64, 129.84, 129.80, 129.64, 129.60, 129.52, 129.19, 129.15, 129.07, 129.04, 128.99, 128.88, 128.72, 128.03, 128.00, 127.91, 127.88, 127.57, 127.51, 127.18, 127.14, 127.01, 126.93, 125.89, 125.04, 125.02, 124.97, 124.75, 124.71, 122.55, 122.52, 122.39, 122.36, 52.20, 52.16, 21.28, 21.01, 20.09, 17.91, 17.87, 17.83.



The procedure was similar to the synthesis of *rac*-L1 except that 2-((4-chlorophenyl)(phenyl)methyl)-4,6-dimethylbenzenamine (966 mg, 3.0 mmol) was used. The product was obtained as a red powder (Yield: 770 mg, 76.1%).<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.60 (ddd, 2H), 7.20 – 6.89 (m, 14H), 6.75 (ddd, 2H), 6.70 – 6.63 (m, 2H), 6.57 (t, 2H), 6.23 (m, 3H), 6.13 (m, 2H), 6.03 (m, 1H), 5.58 (s, 2H), 2.26 (d, 6H), 2.19 (dd, 6H). <sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  163.36, 163.28, 163.09, 163.01, 146.40, 146.38, 146.33, 146.31, 142.90, 142.87, 142.15, 142.11, 141.35, 140.29, 140.28, 139.71, 133.01, 132.96, 132.92, 132.53, 132.51, 132.47, 131.83, 131.16, 130.96, 130.73, 130.00, 129.81, 129.68, 129.50, 129.44, 129.42, 128.96, 128.86, 128.67, 128.47, 128.44, 128.14, 127.94, 127.91, 127.86, 127.41, 127.36, 127.08, 126.99, 126.96, 126.88, 126.20, 125.22, 125.14, 125.10, 125.08, 125.05, 122.54, 122.50, 122.42, 122.37, 58.49, 51.93, 51.87, 21.25, 18.45, 17.87, 17.83.



The procedure was similar to the synthesis of *rac*-L1 except that 2-((4-methoxyphenyl)(phenyl)methyl)-4,6-dimethylbenzenamine (952 mg, 3.0 mmol) was used. The product was obtained as an orange powder (Yield: 756 mg, 61.7%).<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.93 – 7.53 (m, 2H), 7.34 – 6.87 (m, 14H), 6.81 (m, 4H), 6.74 (m, 2H), 6.56 – 6.39 (m, 3H), 6.23 (m, 2H), 5.93 (s, 1H), 5.82 – 5.59 (m, 2H), 3.77 (d, 6H), 3.00 (d, 6H), 2.55 – 2.16 (m, 12H).<sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  166.12, 165.83, 158.01, 156.56, 143.50, 142.31, 142.28, 142.14, 142.12, 142.09, 140.04, 139.90, 137.12, 136.75, 136.53, 134.38, 134.35, 133.86, 133.83, 133.28, 133.22, 131.46, 131.42, 131.29, 131.25, 130.84, 130.74, 130.55, 130.23, 130.17, 129.81, 129.74, 129.59, 129.40, 129.18, 129.10, 129.07, 129.03, 128.99, 128.89, 128.85, 128.75, 128.71, 128.15, 128.04, 127.98, 127.91, 127.45, 127.20, 127.05, 126.94, 126.23, 125.91, 125.79, 125.63, 125.07, 124.98, 124.93, 124.89, 124.78, 122.54, 122.39, 113.60, 113.41, 112.80, 112.61, 55.23, 54.42, 54.35, 51.68, 51.01, 50.88, 21.41, 21.39, 21.29, 21.27, 18.34, 18.27, 17.89.



The procedure was similar to the synthesis of *rac*-L1 except that 2-((4-tertbutylphenyl)(phenyl)methyl)-4,6-dimethylbenzenamine (1031 mg, 3.0 mmol) was used. The product was obtained as an orange powder (Yield: 513 mg, 39.3%).<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.53 – 7.42 (m, 2H), 7.17 (m, 14H), 6.83 – 6.71 (m, 4H), 6.65 (d, 2H), 6.32 (dd, 3H), 6.29 – 6.20 (m, 3H), 6.15 (d, 1H), 6.03 (d, 1H), 5.64 (d, 2H), 2.25 (d, 6H), 2.13 (d, 6H), 1.22 (s, 9H), 0.54 (s, 9H).<sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  162.82, 162.74, 162.63, 148.58, 147.62, 147.60, 146.55, 144.09, 142.00, 140.54, 139.72, 138.73, 133.53, 133.50, 133.18, 132.69, 132.65, 129.82, 129.73, 129.65, 129.44, 129.20, 129.04, 129.00, 128.98, 128.19, 128.14, 128.12, 128.02, 127.93, 127.19, 127.15, 126.94, 126.91, 125.84, 124.93, 124.88, 124.86, 124.81, 124.78, 124.03, 122.53, 122.48, 122.43, 52.07, 52.02, 34.34, 33.47, 31.40, 30.74, 21.29, 21.28, 18.03, 17.97, 17.89.



The procedure was similar to the synthesis of *rac*-L1 except that 2,4-dimethyl-6-(naphthalen-2-yl(phenyl)methyl)benzenamine (1012 mg, 3.0 mmol) was used. The product was obtained as an orange powder (Yield: 555 mg, 43.2%).<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.99 (dd, 1H), 7.89 (dd, 1H), 7.80 – 7.61 (m, 4H), 7.57 – 7.42 (m, 2H), 7.36 – 7.22 (m, 4H), 7.17 – 6.88 (m, 14H), 6.88 – 6.77 (m, 4H), 6.74 – 6.62 (m, 2H), 6.44 – 6.37 (m, 2H), 6.21 – 6.14 (m, 2H), 5.64 (d, 2H), 2.23 (d, 6H), 1.94 (d, 6H).<sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  163.30, 163.25, 163.19, 146.62, 143.51, 142.84, 142.04, 141.53, 141.49, 140.08, 139.32, 139.26, 133.40, 133.23, 132.97, 132.87, 132.82, 132.77, 132.73, 132.55, 132.46, 132.31, 132.09, 132.01, 131.87, 131.03, 130.93, 130.63, 130.42, 129.98, 129.90, 129.78, 129.73, 129.66, 129.43, 129.36, 129.34, 129.22, 129.13, 128.75, 128.62, 128.58, 128.54, 128.52, 128.48, 128.30, 128.24, 128.20, 128.14, 128.10, 128.05, 127.94, 127.91, 127.83, 127.74, 127.70, 127.60, 127.50, 127.45, 127.42, 127.36, 127.25, 127.20, 127.01, 126.73, 126.66, 126.64, 126.59, 126.57, 126.45, 126.05, 125.99, 125.79, 125.71, 125.45, 125.37, 125.25, 125.14, 125.10, 125.05, 125.01, 124.90, 124.74, 124.55, 124.44, 122.49, 122.35, 122.12, 121.75, 121.65, 121.46, 56.06, 52.71, 52.63, 52.51, 51.90, 51.29, 21.35, 21.32, 21.25, 18.24, 17.93, 17.55.



Figure S7. <sup>13</sup>C NMR of complex *rac*-L1.



Figure S8. <sup>1</sup>H NMR of complex *rac*-L2.



Figure S9. <sup>13</sup>C NMR of complex *rac*-L2.



Figure S11. <sup>13</sup>C NMR of complex *rac*-L3.



Figure S13. <sup>13</sup>C NMR of complex *rac*-L4.



Figure S14. <sup>1</sup>H NMR of complex *rac*-L5.



Figure S15. <sup>13</sup>C NMR of complex *rac*-L5.