

Highly Tunable Metal-Free Ring Opening Polymerization of Glycidol into Various Controlled Topologies Catalyzed by Frustrated Lewis Pairs

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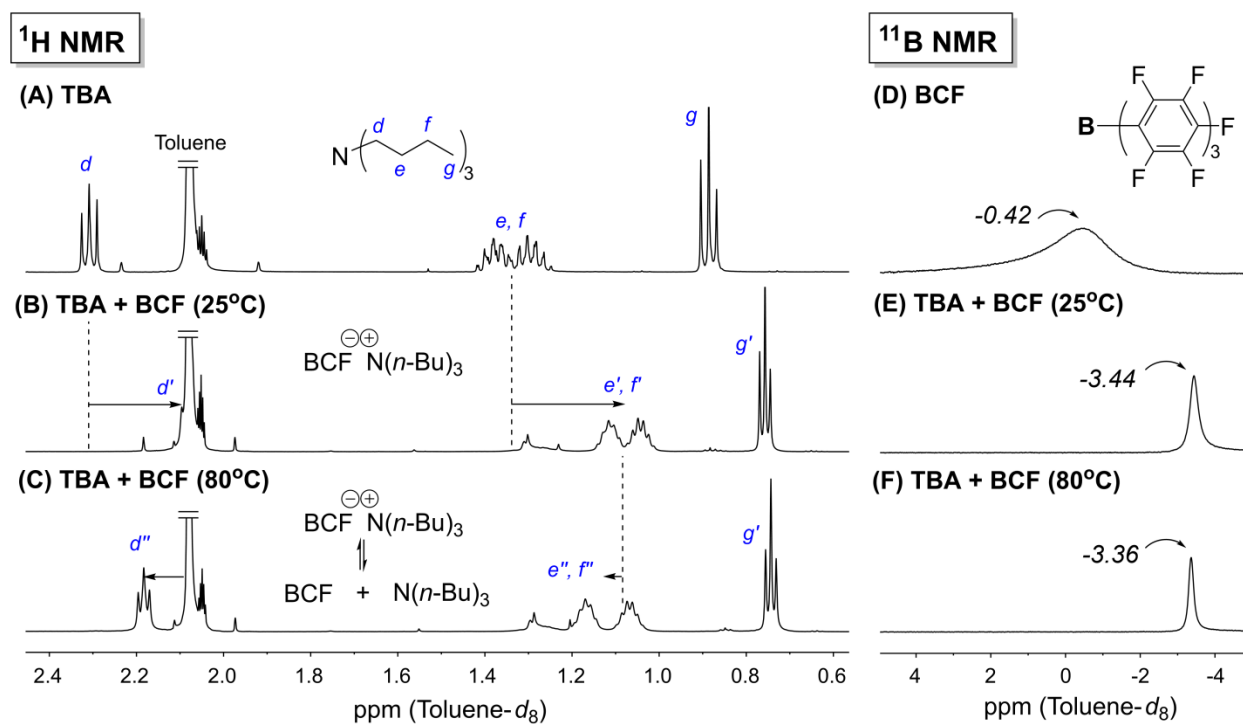


Figure S1. ¹H NMR spectra of (A) TBA ([TBA]₀ = 40 mM in toluene) at 25 °C, (B) a mixture prepared with a 2:1 ratio ([TBA]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, and (C) a mixture prepared with a 2:1 ratio ([TBA]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C; ¹¹B NMR spectra of (D) BCF ([BCF]₀ = 20 mM) at 25 °C, (E) a mixture prepared with a 2:1 ratio ([TBA]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, and (F) a mixture prepared with a 2:1 ratio ([TBA]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C.

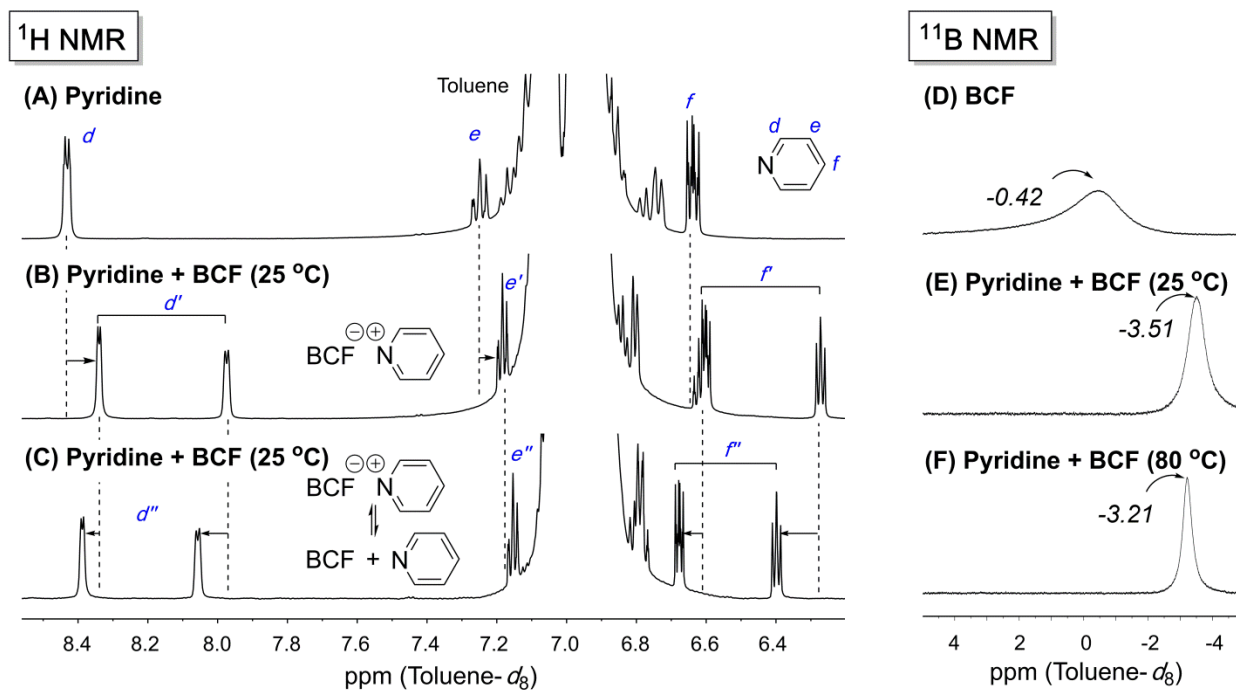


Figure S2. ¹H NMR spectra of (A) Py ([Py]₀ = 40 mM in toluene) at 25 °C, (B) a mixture of Py and BCF with a 2:1 ratio ([Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, (C) a mixture of Py and BCF with a 2:1 ratio ([Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C, ¹¹B NMR spectra of (D) BCF ([BCF]₀ = 20 mM) at 25 °C, (E) a mixture of Py and BCF with a 2:1 ratio ([Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, and (F) a mixture of Py and BCF with a 2:1 ratio ([Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C.

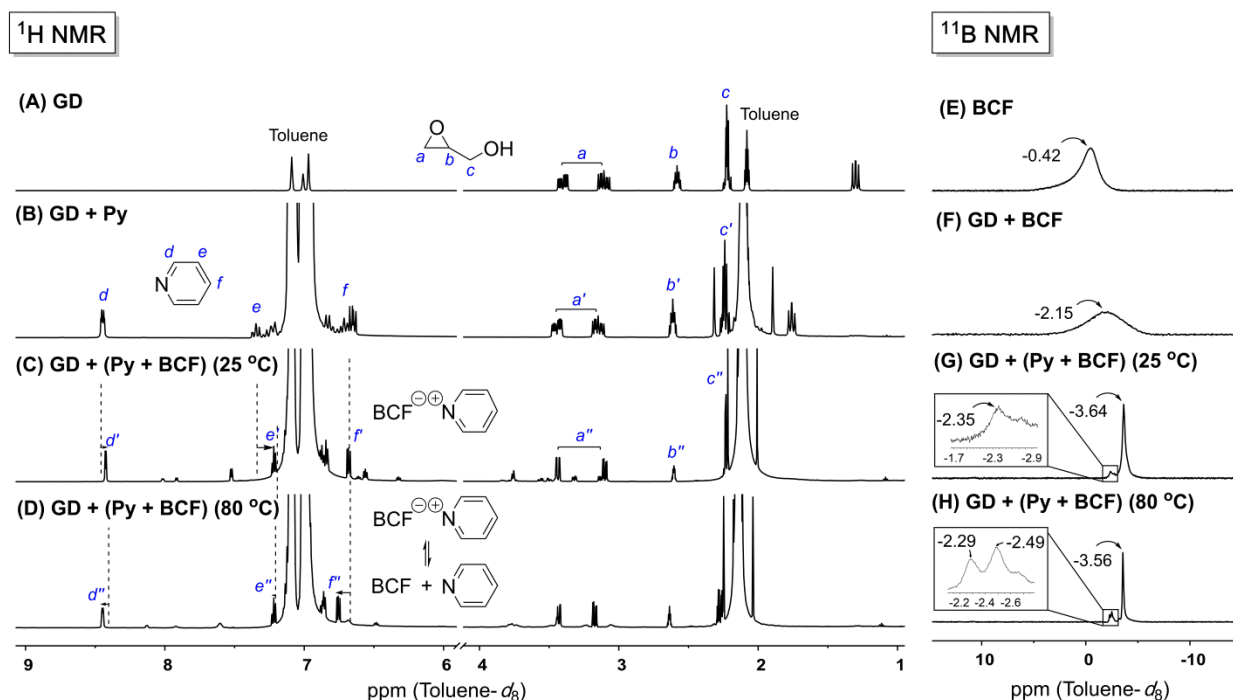
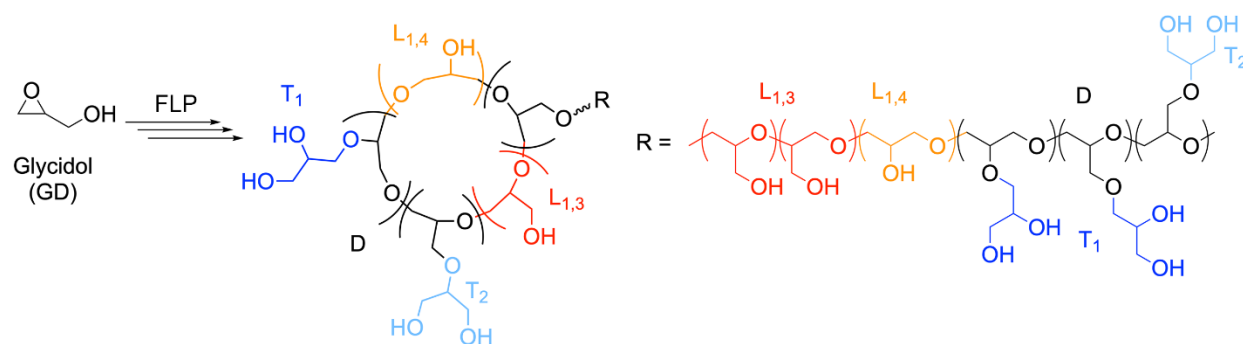


Figure S3. ¹H NMR spectra of (A) ([GD]₀ = 40 mM in toluene) at 25 °C, (B) a mixture of GD and Py with a 1:1 ratio ([GD]₀ = 40 mM; [Py]₀ = 40 mM) at 25 °C, (C) a mixture of GD and BCF-Py ([GD]₀ = 40 mM; [Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, (D) a mixture of GD and BCF-Py ([GD]₀ = 40 mM; [Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C; ¹¹B NMR spectra of (E) BCF ([BCF]₀ = 20 mM) at 25 °C, (F) a mixture of Py and BCF with a 2:1 ratio ([Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, (G) a mixture of GD and BCF-Py ([GD]₀ = 40 mM; [Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 25 °C, and (H) a mixture of GD and BCF-Py ([GD]₀ = 40 mM; [Py]₀ = 40 mM; [BCF]₀ = 20 mM) at 80 °C.



(A) TBA-P15

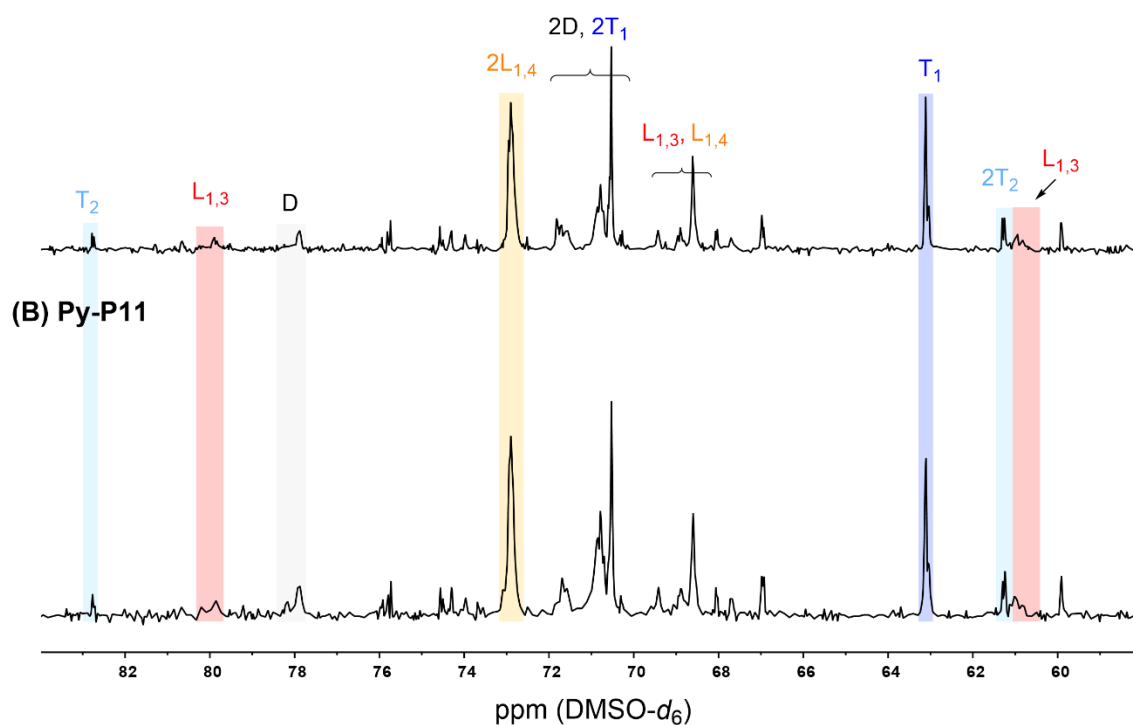


Figure S4. Inverse-gated ¹³C NMR spectra (DMSO-*d*₆, room temperature) of (A) TBA-P15 ([GD]₀ = 2500 mM; [B(C₆F₅)₃]₀ = 3 mM; [TBA]₀ = 12 mM in toluene), and (B) Py-P11 ([GD]₀ = 2500 mM; [B(C₆F₅)₃]₀ = 3 mM; [Py]₀ = 6 mM in toluene).

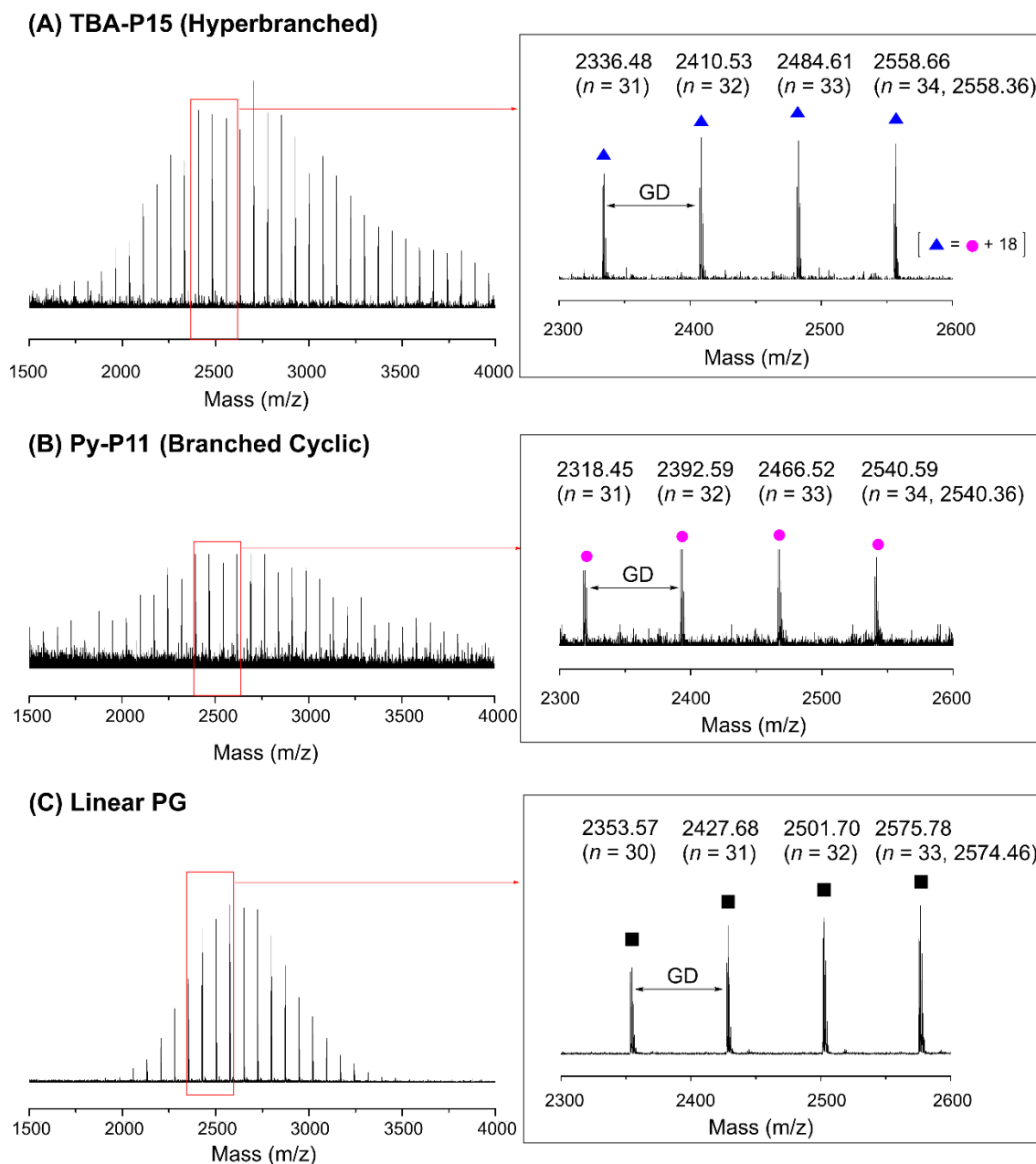


Figure S5. Representative MALDI-ToF spectra for (A) TBA-P15 (*hb*-PG), (B) Py-P11 (*bc*-PG) and (C) linear PG (*lin*-PG) from 1500 to 4500 Da.

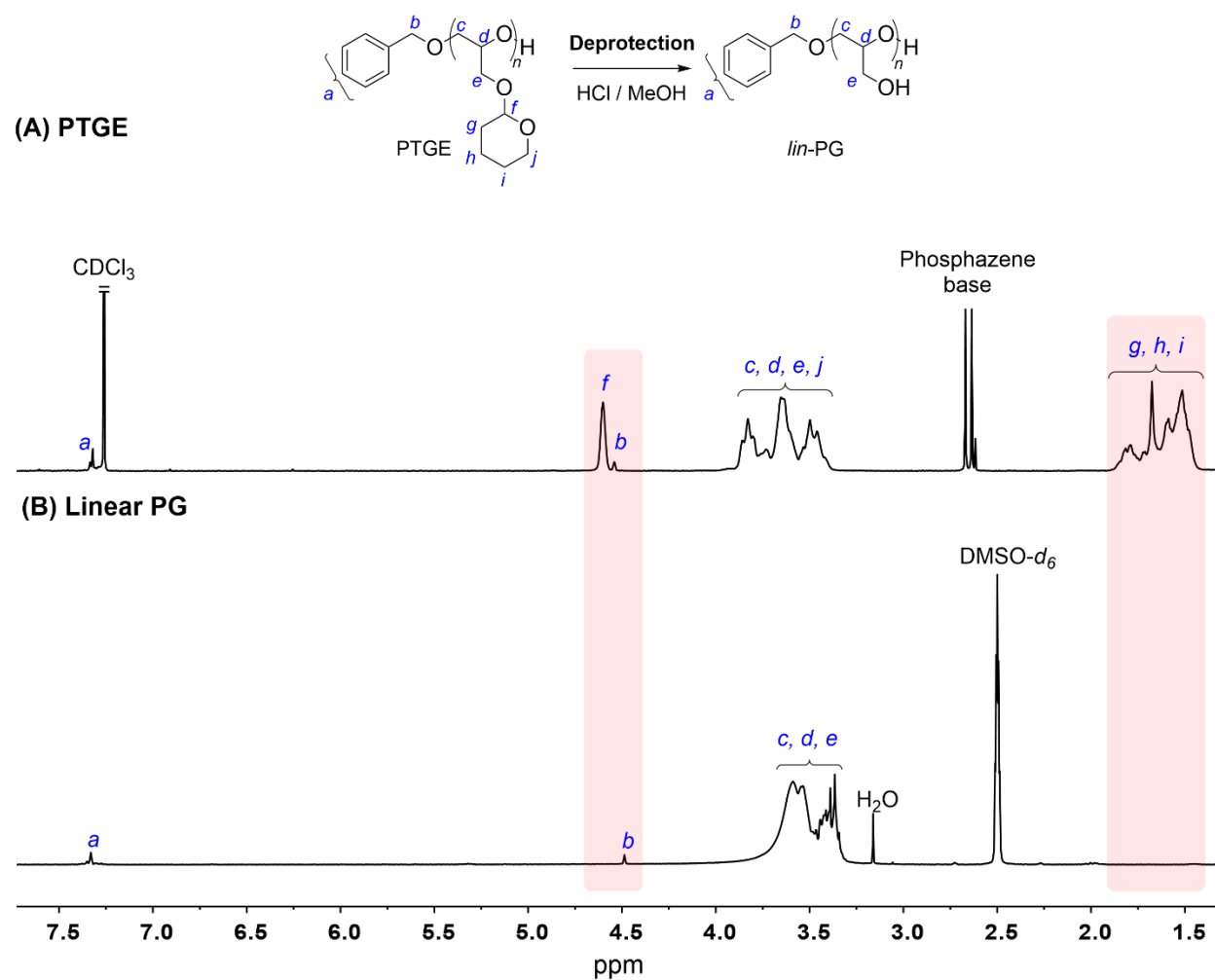


Figure S6. ¹H NMR spectra of (A) poly(TGE) ([benzyl alcohol]₀ = 65 mM; [*t*-BuP₄]₀ = 84.5 mM; [TGE]₀ = 2340 mM in toluene at 25 °C for 24 h) and (B) *lin*-PG (a solution of poly(TGE) (0.5 g $M_{n, NMR} = 5650$) in HCl/MeOH (1.25 M, 0.13 mL) at room temperature for 24 h).

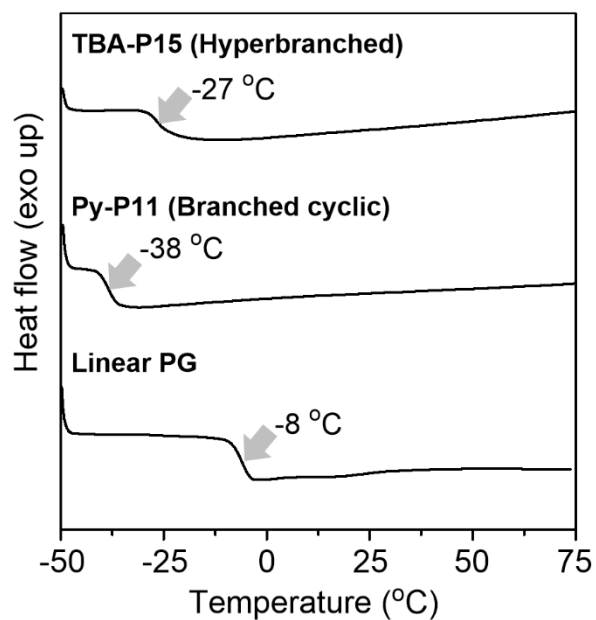


Figure S7. DSC profiles (heating, 10 °C/min) of topology-controlled PGs (TBA-P15 (hyperbranched), Py-P11 (branched cyclic), and linear PG).

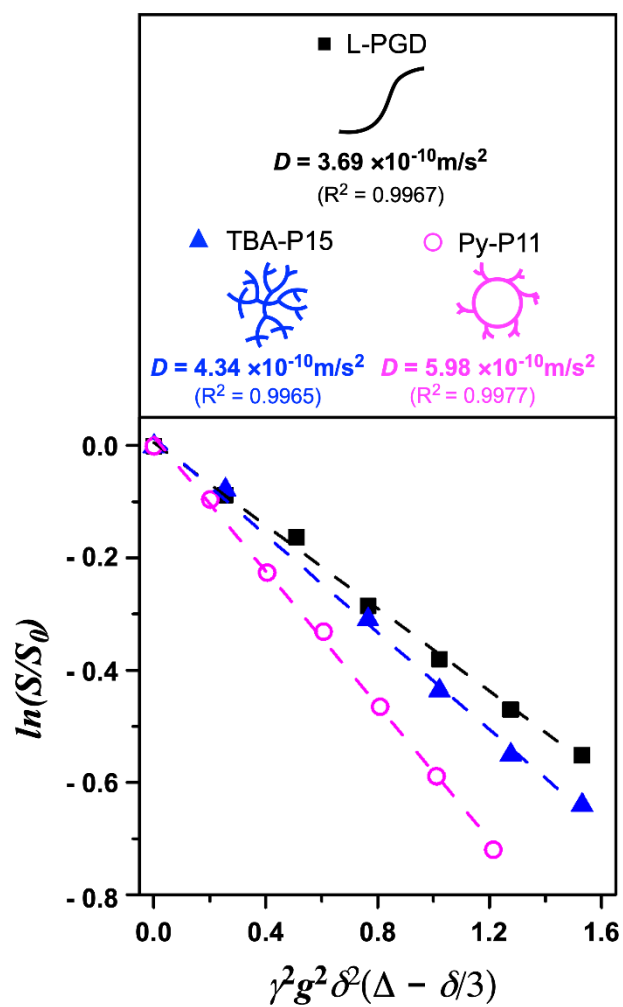


Figure S8. Stejskal–Tanner plot obtained from ^1H DOSY spectra and the determined diffusion coefficient (D) values.

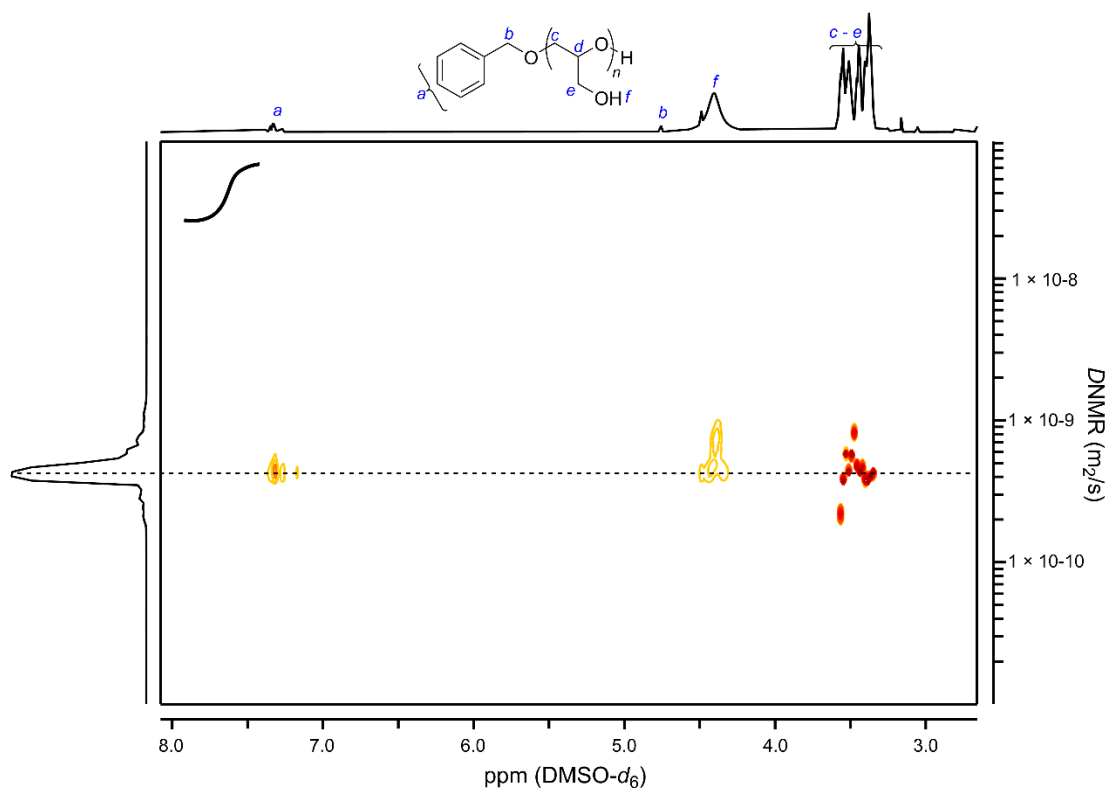


Figure S9. ^1H DOSY spectra ($\text{DMSO-}d_6$, room temperature) of *lin*-PG.

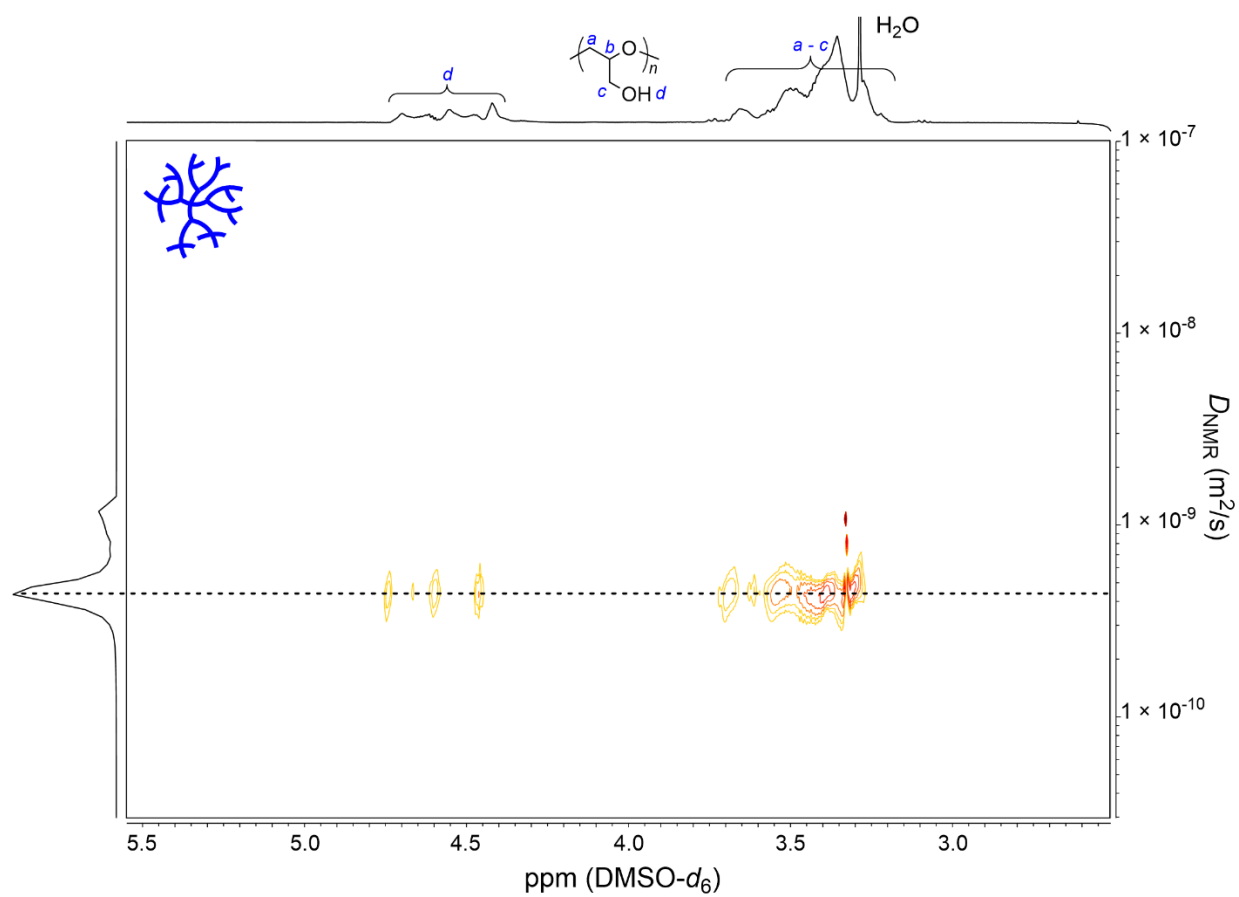


Figure S10. ^1H DOSY spectra ($\text{DMSO-}d_6$, room temperature) of TBA-P15 (hyperbranched).

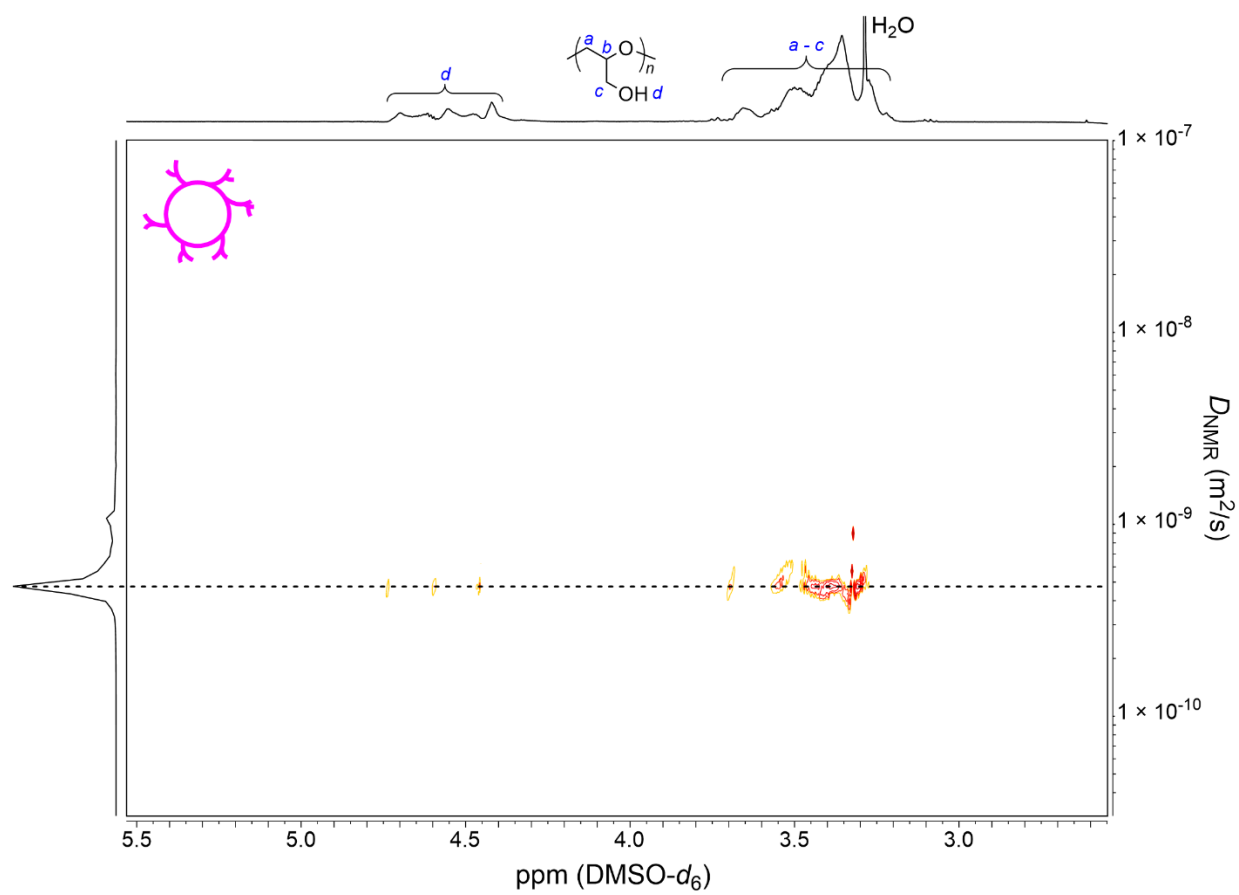


Figure S11. ^1H DOSY spectra ($\text{DMSO-}d_6$, room temperature) of Py-P11 (branched cyclic).

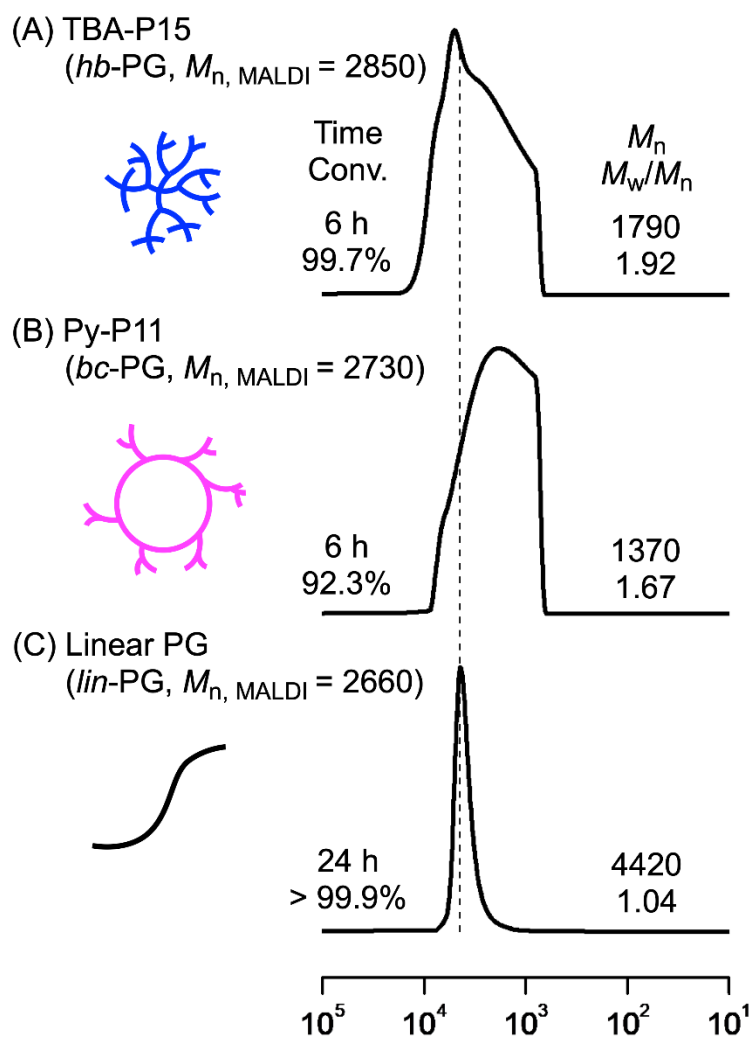


Figure S12. SEC curves of (A) TBA-P15 (*hb*-PG), (B) Py-P11 (*bc*-PG) and (C) linear PG (*lin*-PG).

Table S1. Structural control of PGs via BCF catalyst with $(n\text{-Bu})_3\text{N}$ as Lewis base ^a

Polymer code	$[\text{B}(\text{C}_6\text{F}_5)_3]_0$ (mM)	$[(n\text{-Bu})_3\text{N}]_0$ (mM)	Temp. (°C)	Time	Conv. (%) ^b	M_n^c	M_w/M_n^c	DB ^d
BCF-P1	3	0	25	15 min	94.7	2720	1.55	0.38
TBA-P2	0	12	25	72 h	55.0	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P3	0	12	100	6 h	99.7	1620	1.53	0.67
TBA-P4	3	3	0	24 h	< 5	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P5	3	3	25	24 h	< 5	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P6	3	3	80	24 h	99.8	1950	1.65	0.51
TBA-P7	3	3	100	24 h	99.8	1830	1.48	0.49
TBA-P8	3	6	0	24 h	< 5	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P9	3	6	25	72 h	44.7	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P10	3	6	80	6 h	93.3	1600	1.87	0.50
TBA-P11	3	6	100	3 h	96.4	1420	1.74	0.50
TBA-P12	3	12	0	72 h	49.6	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
TBA-P13	3	12	25	72 h	97.4	1500	1.60	0.46
TBA-P14	3	12	80	6 h	99.8	1640	1.73	0.49
TBA-P15	3	12	100	6 h	99.7	1790	1.92	0.51

^a $[\text{GD}]_0 = 2500$ mM in toluene. ^bConversion was determined by ¹H NMR with tetralin as an internal standard. ^cMeasured by size-exclusion chromatography calibrated with PEO standards in DMF (50 mM LiBr, 45 °C, flow rate 1.0 mL/min). ^dDegree of branching (DB) was calculated by inverse-gated ¹³C NMR using the following equation: $(D + T)/(D + T + L)$ (D = dendritic unit; T = terminal unit; L = linear unit). ^eNot determined.

Table S2. Structural control of PGs using BCF catalyst with pyridine as Lewis base^a

Polymer code	[B(C ₆ F ₅) ₃] ₀ (mM)	[Pyridine] ₀ (mM)	Temp. (°C)	Time	Conv. (%) ^b	<i>M_n</i> ^c	<i>M_w</i> / <i>M_n</i> ^c	DB ^d
BCF-P1	3	0	25	15 min	94.7	2720	1.55	0.38
Py-P2	0	6.0	0	24 h	<1.0	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
Py-P3	0	6.0	100	24 h	99.5	1670	1.50	0.50
Py-P4	3	3.0	0	12 d	49.03	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
Py-P5	3	3.0	25	12 d	50.41	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
Py-P6	3	3.0	80	12 d	>99.9	3210	1.10	0.44
Py-P7	3	3.0	100	6 d	>99.9	1480	1.99	0.45
Py-P8	3	6.0	0	18 h	<1.0	<i>n/d</i> ^e	<i>n/d</i> ^e	<i>n/d</i> ^e
Py-P9	3	6.0	25	6 d	91.6	1190	1.18	0.48
Py-P10	3	6.0	80	18 h	97.98	1430	1.60	0.49
Py-P11	3	6.0	100	18 h	92.3	1370	1.67	0.49

^a[GD]₀ = 2500 mM in toluene. ^bConversion was determined by ¹H NMR with tetralin as an internal standard. ^cMeasured by size-exclusion chromatography calibrated with PEO standards in DMF (50 mM LiBr, 45 °C, flow rate 1.0 mL/min). ^dDegree of branching (DB) was calculated by inverse-gated ¹³C NMR with following equation: $(D + T)/(D + T + L)$ (*D* = dendritic unit; *T* = terminal unit; *L* = linear unit). ^eNot determined.

Table S3. Structural control of PGs using BCF catalyst with (*n*-Bu)₃N as Lewis base at different polymerisation temperatures ^{a, b}

Region	Chemical shift (ppm)	BCF-P1	TBA-P3	TBA-P13	TBA-P14	TBA-P15
L _{1,3}	60.6–61.2	2.32	1.09	0.75	1.56	1.62
2T ₂	61.2–61.8	0.83	0.91	1.54	0.00	1.57
T ₁	63.0–63.3	1.41	2.62	3.89	4.82	6.13
L _{1,3} , L _{1,4}	68.5–69.7	2.60	4.36	6.54	7.57	8.35
2D, 2T ₁	70.4–72.0	5.47	9.54	12.91	16.15	21.01
2L _{1,4}	72.7–73.2	3.44	1.60	11.56	11.61	13.50
D	77.7–78.9	0.57	1.57	1.76	2.19	2.48
L _{1,3}	79.7–80.4	1.00	1.00	1.00	1.00	1.00
T ₂	81.5–82.2	0.42	0.00	0.00	0.00	0.00
Structure units (%) ^c						
<i>D units</i>		9	26	15	15	15
<i>L units</i>		63	31	53	51	49
(<i>L</i> _{1,3} units)		(36)	(18)	(6)	(11)	(9)
(<i>L</i> _{1,4} units)		(27)	(13)	(47)	(40)	(40)
<i>T units</i>		28	43	32	34	36
(<i>T</i> ₁ units)		(22)	(43)	(32)	(34)	(36)
(<i>T</i> ₂ units)		(6)	(0)	(0)	(0)	(0)
Degree of Branching ^d						
		0.37	0.69	0.47	0.49	0.51

^a[GD]₀ = 2500 mM, [B(C₆F₅)₃]₀ = 3 mM or/and [TBA]₀ = 12 mM in toluene at 25, 80, and 100 °C. ^bObtained by inverse gated ¹³C NMR. ^cThe structure units (%) were calculated from the following equation: integration ratio of each structure region / (*D* + *T* + *L*) × 100. ^dDegree of branching (DB) = (*D* + *T*)/(*D* + *T* + *L*).

Table S4. Structural control of PGs via BCF with different concentration of $(n\text{-Bu})_3\text{N}^{a, b}$

Region	Chemical shift (ppm)	BCF-P1	TBA-P3	TBA-P7	TBA-P11	TBA-P15
L _{1,3}	60.6–61.2	2.32	1.09	1.89	1.56	1.62
2T ₂	61.2–61.8	0.83	0.91	1.44	1.64	1.57
T ₁	63.0–63.3	1.41	2.62	9.49	6.88	6.13
L _{1,3} , L _{1,4}	68.5–69.7	2.60	4.36	11.52	10.11	8.35
2D, 2T ₁	70.4–72.0	5.47	9.54	23.34	23.30	21.01
2L _{1,4}	72.7–73.2	3.44	1.60	20.74	16.01	13.50
D	77.7–78.9	0.57	1.57	2.22	2.83	2.48
L _{1,3}	79.7–80.4	1.00	1.00	1.00	1.00	1.00
T ₂	81.5–82.2	0.42	0.00	0.05	0.00	0.00
Structure units (%) ^c						
<i>D units</i>		9	26	9	15	15
<i>L units</i>		63	31	51	49	49
(<i>L</i> _{1,3} units)		(36)	(18)	(8)	(8)	(9)
(<i>L</i> _{1,4} units)		(27)	(13)	(43)	(41)	(40)
<i>T units</i>		28	43	40	36	36
(<i>T</i> ₁ units)		(22)	(43)	(40)	(36)	(36)
(<i>T</i> ₂ units)		(6)	(0)	(0)	(0)	(0)
Degree of Branching ^d						
		0.37	0.69	0.49	0.50	0.51

^a[GD]₀ = 2500 mM, [B(C₆F₅)₃]₀ = 3 mM or/and [TBA]₀ = 3, 6, and 12 mM in toluene at 100 °C. ^bObtained by inverse-gated ¹³C NMR. ^cThe structure units (%) were calculated from the following equation: integration ratio of each structure region / (*D* + *T* + *L*) × 100. ^dDegree of branching (DB) = (*D* + *T*) / (*D* + *T* + *L*).

Table S5. Structural control of PGs using BCF catalyst with pyridine as Lewis base at different polymerisation temperatures ^{a, b}

Region	Chemical shift (ppm)	BCF-P1	Py-P3	Py-P9	Py-P10	Py-P11
L _{1,3}	60.6–61.2	2.32	0.70	0.88	1.37	1.17
2T ₂	61.2–61.8	0.83	0.58	0.00	0.86	2.82
T ₁	63.0–63.3	1.41	1.52	6.18	6.27	23.78
L _{1,3} , L _{1,4}	68.5–69.7	2.60	2.37	9.56	8.47	20.30
2D, 2T ₁	70.4–72.0	5.47	5.01	17.25	18.69	64.02
2L _{1,4}	72.7–73.2	3.44	3.27	19.02	15.55	55.84
D	77.7–78.9	0.57	0.85	1.90	1.47	3.57
L _{1,3}	79.7–80.4	1.00	1.00	1.00	1.00	1.00
T ₂	81.5–82.2	0.42	0.00	0.00	0.00	0.00
Structure units (%) ^c						
<i>D units</i>		9	18	10	9	6
<i>L units</i>		63	50	56	54	52
(<i>L_{1,3} units</i>)		(36)	(15)	(5)	(8)	(2)
(<i>L_{1,4} units</i>)		(27)	(35)	(51)	(46)	(50)
<i>T units</i>		28	32	34	37	42
(<i>T₁ units</i>)		(22)	(32)	(34)	(37)	(42)
(<i>T₂ units</i>)		(6)	(0)	(0)	(0)	(0)
Degree of Branching ^d						
		0.37	0.50	0.44	0.46	0.48

^a[GD]₀ = 2500 mM, [B(C₆F₅)₃]₀ = 3 mM or/and [pyridine]₀ = 6 mM in toluene at 25, 80, and 100 °C. ^bObtained by inverse-gated ¹³C NMR. ^cThe structure units (%) were calculated from the following equation: integration ratio of each structure region / (*D* + *T* + *L*) × 100. ^dDegree of branching (DB) = (*D* + *T*)/(*D* + *T* + *L*).

Table S6. Structural control of PGDs using BCF with different concentration of pyridine ^{a, b}

Region	Chemical shift (ppm)	BCF-P1	Py-P3	Py-P7	Py-P11
L _{1,3}	60.6–61.2	2.32	0.70	2.15	1.17
2T ₂	61.2–61.8	0.83	0.58	0.89	2.82
T ₁	63.0–63.3	1.41	1.52	4.08	23.78
L _{1,3} , L _{1,4}	68.5–69.7	2.60	2.37	4.05	20.30
2D, 2T ₁	70.4–72.0	5.47	5.01	9.03	64.02
2L _{1,4}	72.7–73.2	3.44	3.27	7.94	55.84
D	77.7–78.9	0.57	0.85	0.40	3.57
L _{1,3}	79.7–80.4	1.00	1.00	1.00	1.00
T ₂	81.5–82.2	0.42	0.00	0.60	0.00
Structure units (%) ^c					
<i>D units</i>		9	18	4	6
<i>L units</i>		63	50	54	52
(<i>L_{1,3} units</i>)		(36)	(15)	(19)	(2)
(<i>L_{1,4} Units</i>)		(27)	(35)	(35)	(50)
<i>T units</i>		28	32	42	42
(<i>T₁ units</i>)		(22)	(32)	(37)	(42)
(<i>T₂ units</i>)		(6)	(0)	(5)	(0)
Degree of Branching ^d		0.37	0.50	0.45	0.48

^a[GD]₀ = 2500 mM, [B(C₆F₅)₃]₀ = 3 mM or/and [pyridine]₀ = 3 and 6 mM in toluene at 100 °C. ^bObtained by inverse-gated ¹³C NMR. ^cThe structure units (%) were calculated from the following equation: integration ratio of each structure region / (*D* + *T* + *L*) × 100. ^dDegree of branching (DB) = (*D* + *T*) / (*D* + *T* + *L*).