

## Supporting Information

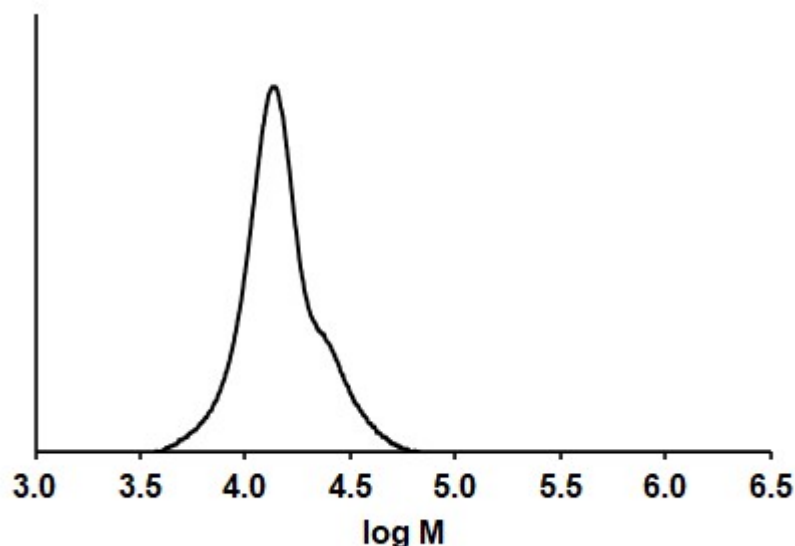
### Synthesis of low glass transition temperature worms comprising a poly(styrene-*stat*-*n*-butyl acrylate) core segment via polymerization-induced self-assembly in RAFT aqueous emulsion polymerization

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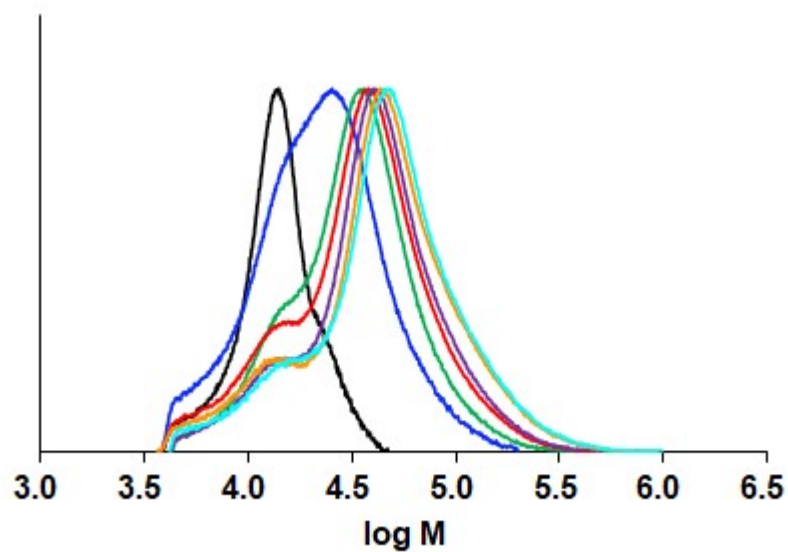
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**Fig. S1** Molecular weight distributions ( $w(\log M)$  vs.  $\log M$ ) of P(AA-*stat*-PEGA)-TTC macroRAFT agent ( $M_n = 13,400$  g/mol;  $D = 1.20$ ).



**Fig. S2** Molecular weight distributions ( $w(\log M)$  vs.  $\log M$ ) of RAFT aqueous emulsion polymerization of styrene and *n*BA in the presence of P(AA-*stat*-PEGA)-TTC macroRAFT agent with different  $[\text{hydrophobic monomer}]_0/[\text{macroRAFT}]_0$  (macroRAFT agent,  $[\text{hydrophobic monomer}]_0/[\text{macroRAFT}]_0 = 50$  (B-1), 100 (B-2), 130 (B-3), 150 (B-4), 170 (B-5), 200 (B-6) from left to right) (Table 2).

**Table S1** Experimental conditions and results of RAFT aqueous emulsion polymerization of styrene and *n*BA in the presence of P(AA-*stat*-PEGA) macroRAFT agent at 80 °C for 4 h; pH 5, molar ratio AA/PEGA = 50/50, styrene/*n*BA = 70/30, [macroRAFT]<sub>0</sub> = 6.2 mmol/L, [NaHCO<sub>3</sub>]<sub>0</sub>/[ACPA]<sub>0</sub> = 3.5.

#	Crosslinker	[M] <sub>0</sub> /[crosslinker] <sub>0</sub> /[RAFT] <sub>0</sub>	Conv. <sup>a</sup> (%)	<i>M</i> <sub>n,th</sub> <sup>b</sup> (g/mol)	Solids content <sup>c</sup> (%)
<i>Entries C-x and D-x: crosslinker added at t = 0 h</i>					
C-1	PEGDA	100/1/1	73 <sup>a</sup>	20,200 <sup>b</sup>	12.6
C-2	PEGDA	100/2.5/1	98 <sup>a</sup>	23,100 <sup>b</sup>	12.7
D-1	EGDA	100/0.5/1	92	22,000	12.7
D-2	EGDA	100/1/1	89	22,800	12.5
D-3	EGDA	100/10/1	89	23,300	13.4
<i>Entries E-x and F-x: crosslinker added at t = 2 h</i>					
E-1	PEGDA	100/1/1	69 <sup>a</sup>	19,800 <sup>b</sup>	12.6
E-2	PEGDA	100/10/1	100 <sup>a</sup>	25,500 <sup>b</sup>	13.7
F-1	EGDA	100/3/1	92	22,100	12.6
F-2	EGDA	100/5/1	93	23,100	12.9
F-3	EGDA	100/10/1	100	24,700	13.3

<sup>a</sup> Monomer conversion by gravimetry. Note that PEGDA was non-volatile and was excluded from calculation of monomer conversion determined by gravimetry.

<sup>b</sup> Theoretical *M*<sub>n,th</sub> calculated using monomer conversion obtained from gravimetry via eqn (S1).

<sup>c</sup> Solids content in wt% based on total weight.

The theoretical number-average molecular weight (*M*<sub>n,th</sub>) was determined by eqn (S1):

$$M_{n,th} = M_{\text{macroRAFT}} + \frac{(X_{\text{conv}} \cdot [\text{styrene}]_0 \cdot M_S) + (X_{\text{conv}} \cdot [n\text{BA}]_0 \cdot M_{n\text{BA}}) + (X_{\text{conv}} \cdot [\text{EGDA}]_0 \cdot M_{\text{EGDA}})}{[\text{macroRAFT}]_0} \quad (\text{S1})$$

where *M*<sub>macroRAFT</sub>, *M*<sub>S</sub>, *M*<sub>*n*BA</sub>, *M*<sub>EGDA</sub> are the molar masses of macroRAFT agent, styrene, *n*BA, and EGDA, respectively, [macroRAFT]<sub>0</sub>, [styrene]<sub>0</sub>, [*n*BA]<sub>0</sub>, [EGDA]<sub>0</sub> are the initial concentrations of macroRAFT agent, styrene, *n*BA, and EGDA respectively, and *X*<sub>conv</sub> denotes total monomer conversion determined by gravimetry.

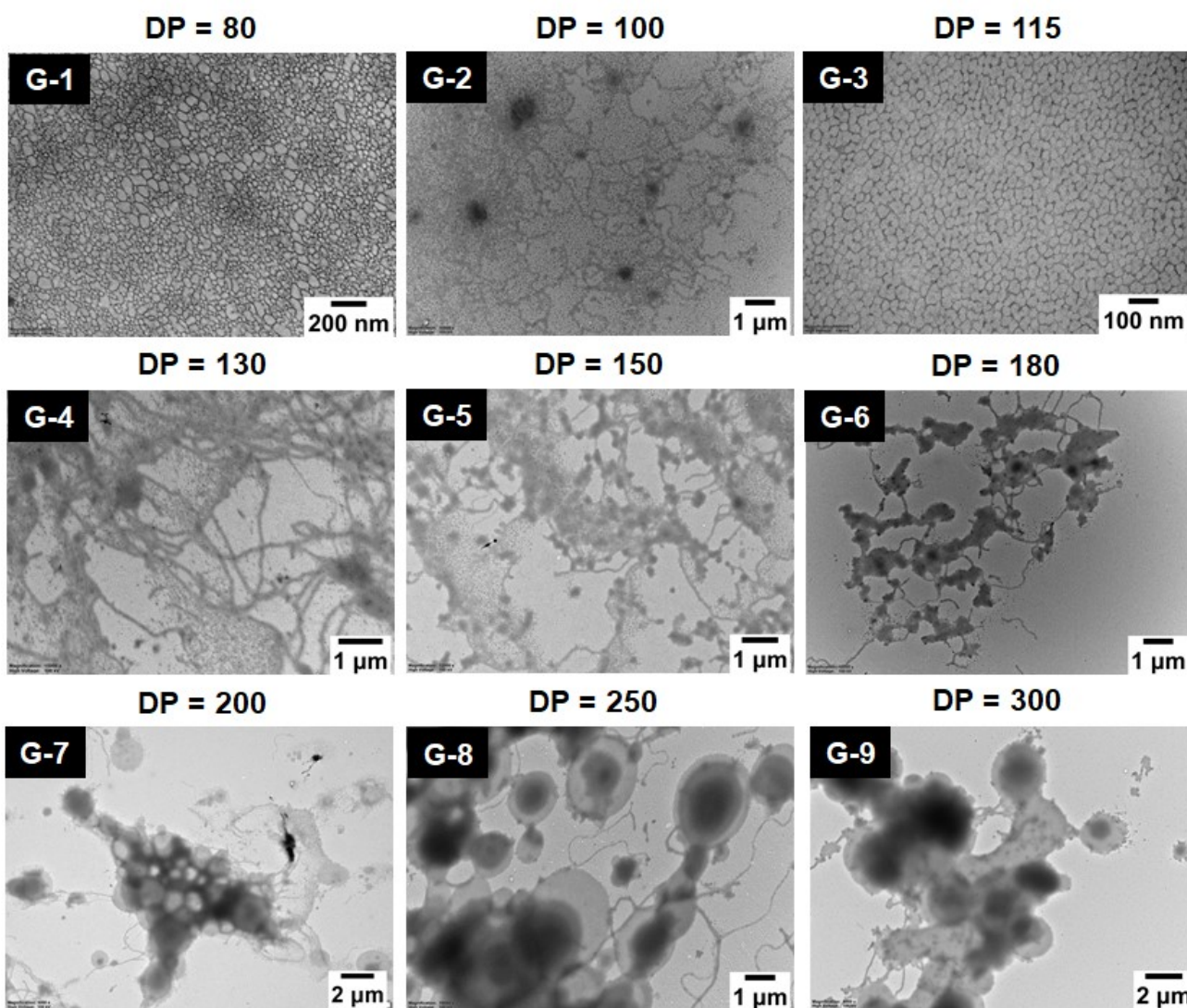
**Table S2** Experimental conditions and results of RAFT aqueous emulsion polymerization of styrene and *n*BA in the presence of P(AA-*stat*-PEGA) macroRAFT agent at 80 °C, EGDA added at  $t = 2$  h; pH 5, molar ratio AA/PEGA = 50/50, styrene/*n*BA = 20/80, EGDA/macroRAFT = 5/1, [macroRAFT]<sub>0</sub> = 6.2 mmol/L, [NaHCO<sub>3</sub>]<sub>0</sub>/[ACPA]<sub>0</sub> = 3.5.

#	DP	$t$ (h)	Conv. <sup>a</sup> (%)	$M_{n,th}^b$ (g/mol)	Solids content <sup>c</sup> (%)
G-1	80	4	100	22,300	12.2
G-2	100	4	100	24,600	13.4
G-3	115	4	98	26,300	14.2
G-4	130	4	100	28,800	15.2
G-5	150	4	100	30,800	16.1
G-6	180	4	100	34,500	17.9
G-7	200	4	96	35,900	18.7
G-8	250	4	94	41,200	21.1
G-9	300	4	93	46,600	23.5

<sup>a</sup> Monomer conversion by gravimetry.

<sup>b</sup> Theoretical  $M_{n,th}$  calculated using monomer conversion obtained from gravimetry via eqn (S1).

<sup>c</sup> Solids content in wt% based on total weight.



**Fig. S3** TEM images of nanoparticles synthesized via RAFT aqueous emulsion polymerization of styrene and *n*BA ( $[\text{styrene}]_0/[\text{nBA}]_0 = 20/80$ , solids content varied from 12.2% to 23.5%) in the presence of P(AA-*stat*-PEGA)-TTC macroRAFT agent at pH 5 with EGDA ( $[\text{EGDA}]_0/[\text{macroRAFT}]_0 = 5/1$ ) after 2 h of the polymerization with different  $[\text{hydrophobic monomer}]_0/[\text{macroRAFT}]_0$  ( $[\text{hydrophobic monomer}]_0/[\text{macroRAFT}]_0 = 80$  (G-1), 100 (G-2), 115 (G-3), 130 (G-4), 150 (G-5), 180 (G-6), 200 (G-7), 250 (G-8), 300 (G-9)). Scale bars: G-1 = 200 nm, G-2 = 1  $\mu\text{m}$ , G-3 = 100 nm, G-4 = 1  $\mu\text{m}$ , G-5 = 1  $\mu\text{m}$ , G-6 = 1  $\mu\text{m}$ , G-7 = 2  $\mu\text{m}$ , G-8 = 1  $\mu\text{m}$ , G-9 = 2  $\mu\text{m}$ .

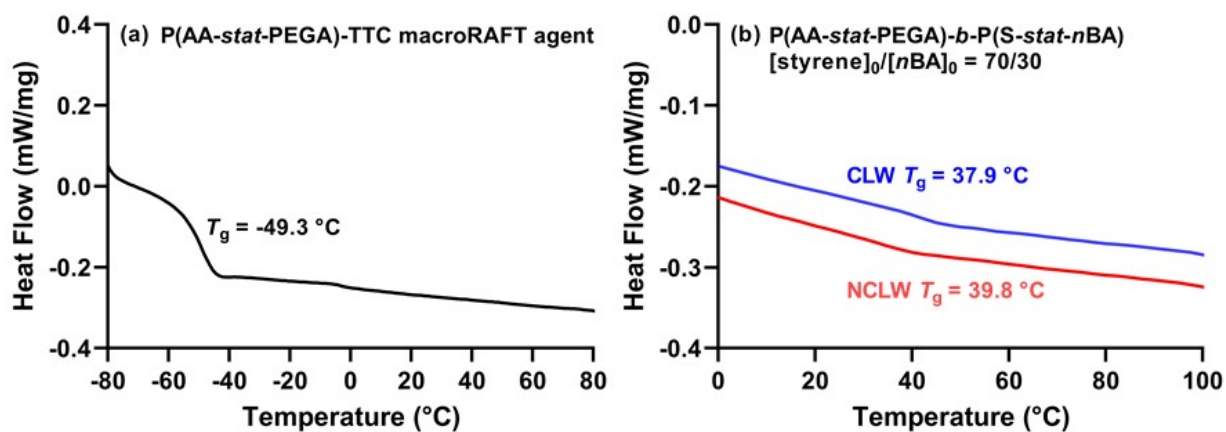
**Table S3** Experimental conditions and results of RAFT aqueous emulsion polymerization of styrene and *n*BA in the presence of P(AA-*stat*-PEGA) macroRAFT agent at 80 °C for 4 h, EGDA added at  $t = 2$  h; pH 5, molar ratio AA/PEGA = 50/50, styrene/*n*BA = 20/80, EGDA/macroRAFT = 5/1, [macroRAFT]<sub>0</sub> = 6.2 mmol/L, [NaHCO<sub>3</sub>]<sub>0</sub>/[ACPA]<sub>0</sub> = 3.5.

#	DP	$t$ (h)	Conv. <sup>a</sup> (%)	$M_{n,th}^b$ (g/mol)	Solids content <sup>c</sup> (%)
H-1	130	4	100	28,800	15.2
H-2	130	4	100	28,300	24.7
H-3	130	4	91	26,800	32.7

<sup>a</sup> Monomer conversion by gravimetry.

<sup>b</sup> Theoretical  $M_{n,th}$  calculated using monomer conversion obtained from gravimetry via eqn (S1).

<sup>c</sup> Solids content in wt% based on total weight.



**Fig. S4** DSC traces of (a) P(AA-*stat*-PEGA)-TTC macroRAFT agent, (b) P(AA-*stat*-PEGA)-*b*-P(S-*stat*-*n*BA) non-crosslinked worms (NCLW) and crosslinked worms (CLW) ([styrene]<sub>0</sub>/[nBA]<sub>0</sub> = 70/30).