

Supporting Information

Room temperature synthesis of UO_{2+x} nanocrystals and thin films via hydrolysis of uranium(IV) complexes

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Table of Contents

| | |
|---|-----|
| Table S1. X-ray Crystallography data | S3 |
| Figure S1. ^1H NMR spectrum of 1 in C_6D_6 | S4 |
| Figure S2. Images of 2 and $\text{UO}_2\text{-}\mathbf{2}$ thin films on Si | S5 |
| Figure S3. Profilometry of $\text{UO}_2\text{-}\mathbf{1}^{\text{film}}$ on glass substrate..... | S6 |
| Figure S4. Profilometry of $\text{UO}_2\text{-}\mathbf{1}^{\text{film}}$ on silicon substrate..... | S7 |
| Figure S5. Electronic absorption spectra of $\text{UO}_2\text{-}\mathbf{1}^{\text{film}}$ on glass | S8 |
| Figure S6. Room temperature electronic absorption spectra of 1 in toluene..... | S9 |
| Figure S7. FT-IR spectra of $\text{UO}_2\text{-}\mathbf{1}$, $\text{UO}_2\text{-}\mathbf{2}$, commercial UO_2 . and KBr background | S10 |
| Figure S8. Raman spectra of $\text{UO}_2\text{-}\mathbf{1}$ and $\text{UO}_2\text{-}\mathbf{2}$ powders..... | S11 |
| Figure S9. ImageJ generated histogram of NP size distribution in $\text{UO}_2\text{-}\mathbf{1}$ and $\text{UO}_2\text{-}\mathbf{2}$ | S12 |

Table S1. Crystallographic Data for **1**

| | |
|---|---|
| empirical formula | C ₃₆ H ₇₆ O ₄ U |
| crystal habit, color | block, light blue |
| crystal size (mm) | 0.30 × 0.41 × 0.33 |
| crystal system | Monoclinic |
| space group | P2 ₁ /n |
| volume (Å ³) | 4060.2(1) |
| <i>a</i> (Å) | 12.4277(3) |
| <i>b</i> (Å) | 17.9692(5) |
| <i>c</i> (Å) | 18.4162(5) |
| α (deg) | 90 |
| β (deg) | 99.156(1) |
| γ (deg) | 90 |
| <i>Z</i> | 4 |
| formula weight (g/mol) | 810.99 |
| density (calculated) (mg/m ³) | 1.327 |
| absorption coefficient (mm ⁻¹) | 4.028 |
| <i>F</i> ₀₀₀ | 1664.0 |
| total no. reflections | 78945 |
| unique reflections | 14383 |
| final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)] | <i>R</i> ₁ = 0.0238, <i>wR</i> ₂ = 0.0461 |
| largest diff. peak and hole (e ⁻ Å ⁻³) | -1.35 and 1.79 |
| GOF | 1.080 |

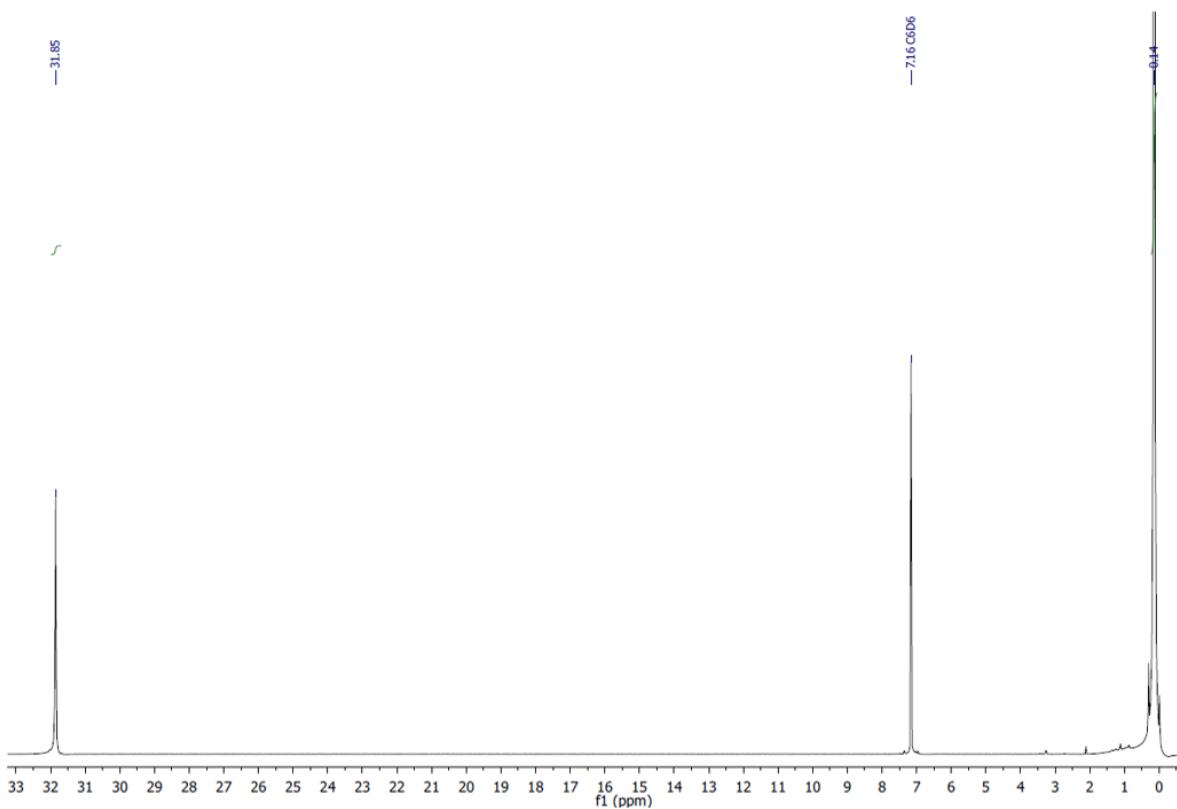


Figure S1. ^1H NMR spectrum of **1** in C_6D_6 (25°C).

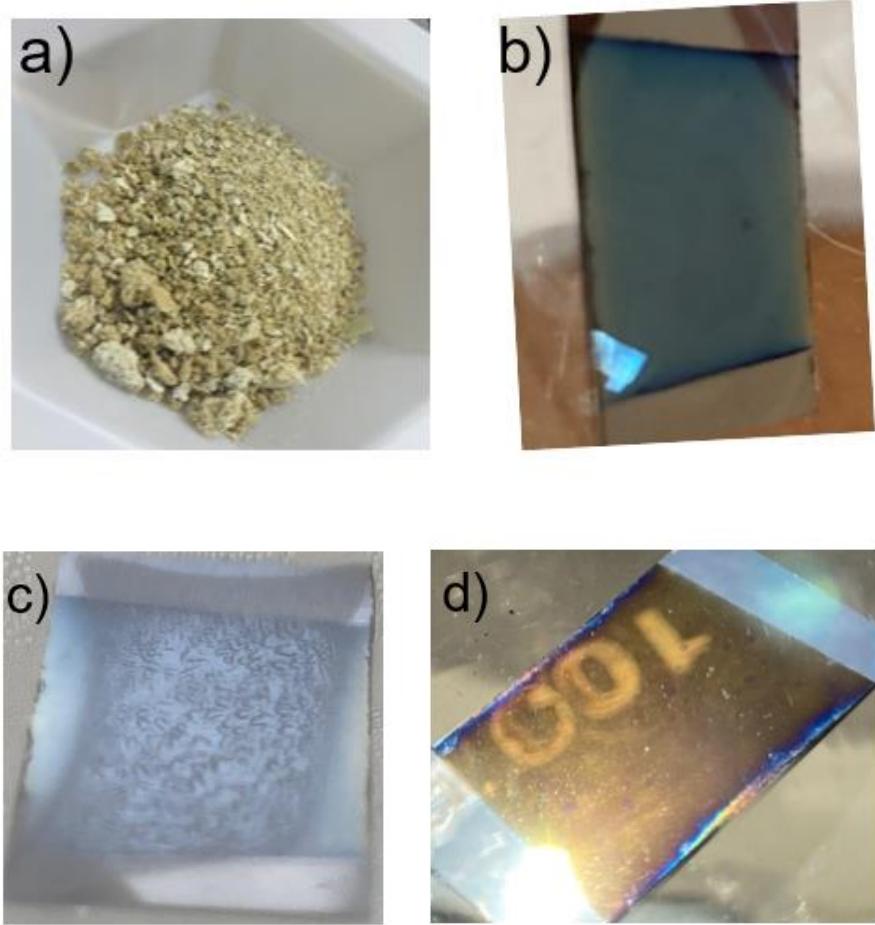


Figure S2. a) Compound **1** as a solid. b) Thin film of **1** deposited on silicon. c) Condensation of water on the film. d) Resulting $\text{UO}_2\text{-}\mathbf{1}^{\text{film}}$ from the hydrolysis of **1**.

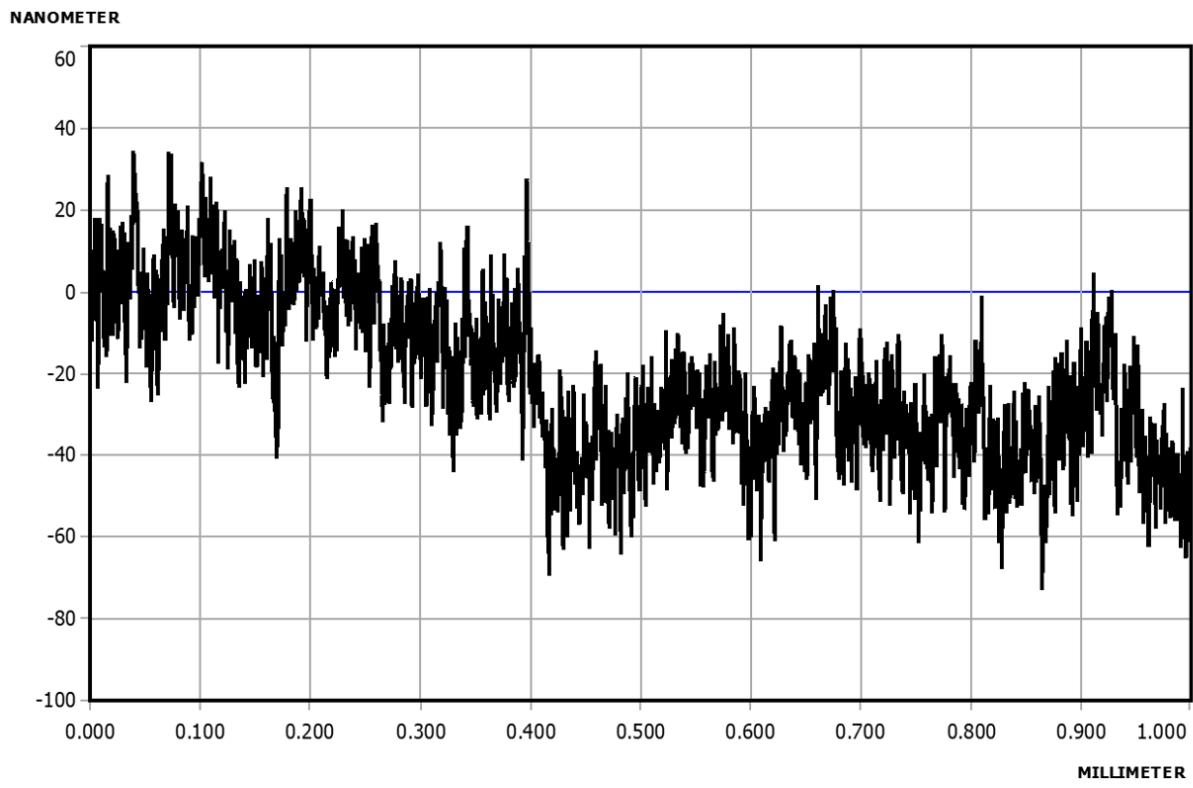


Figure S3. Step-down profile of thin film of $\text{UO}_2\text{-1}$ grown on glass substrate. Step down occurs at 0.4 mm into scan.

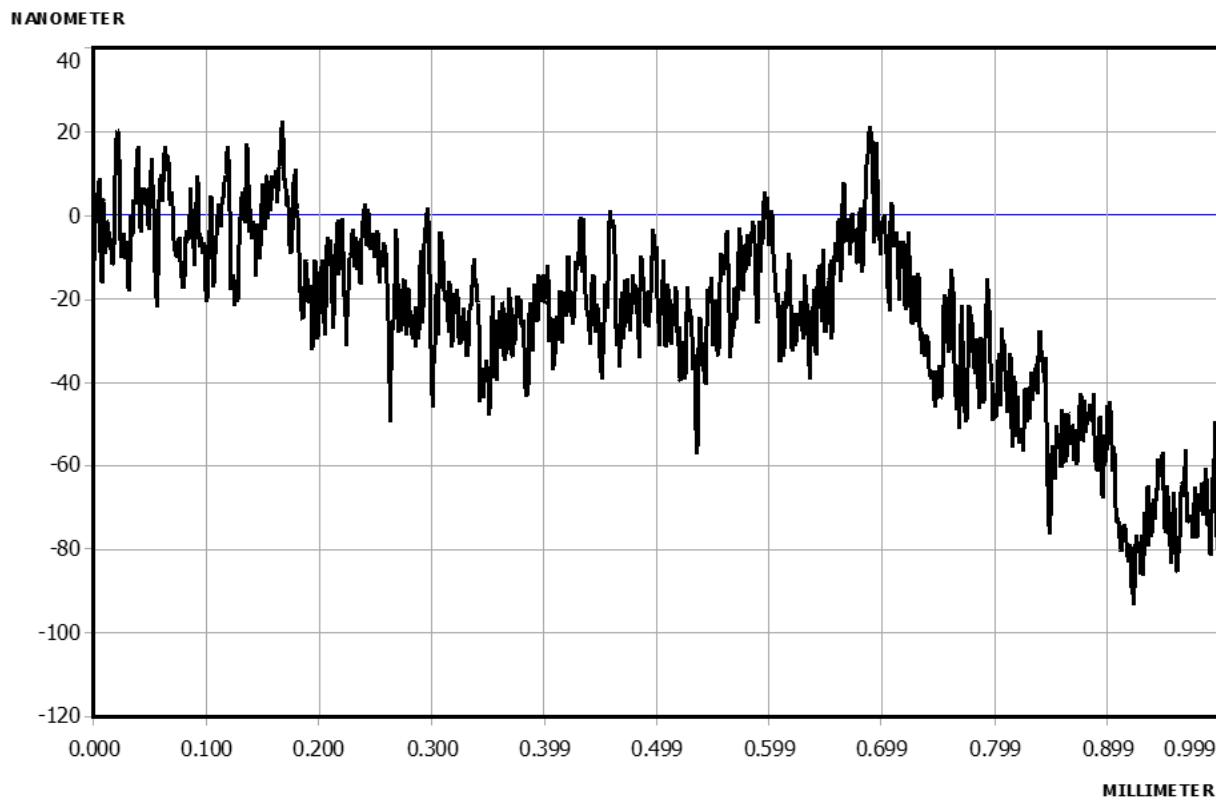


Figure S4. Step-down profile of thin film of $\text{UO}_2\text{-1}$ grown on silicon substrate. Step down occurs at 0.7 mm into scan.

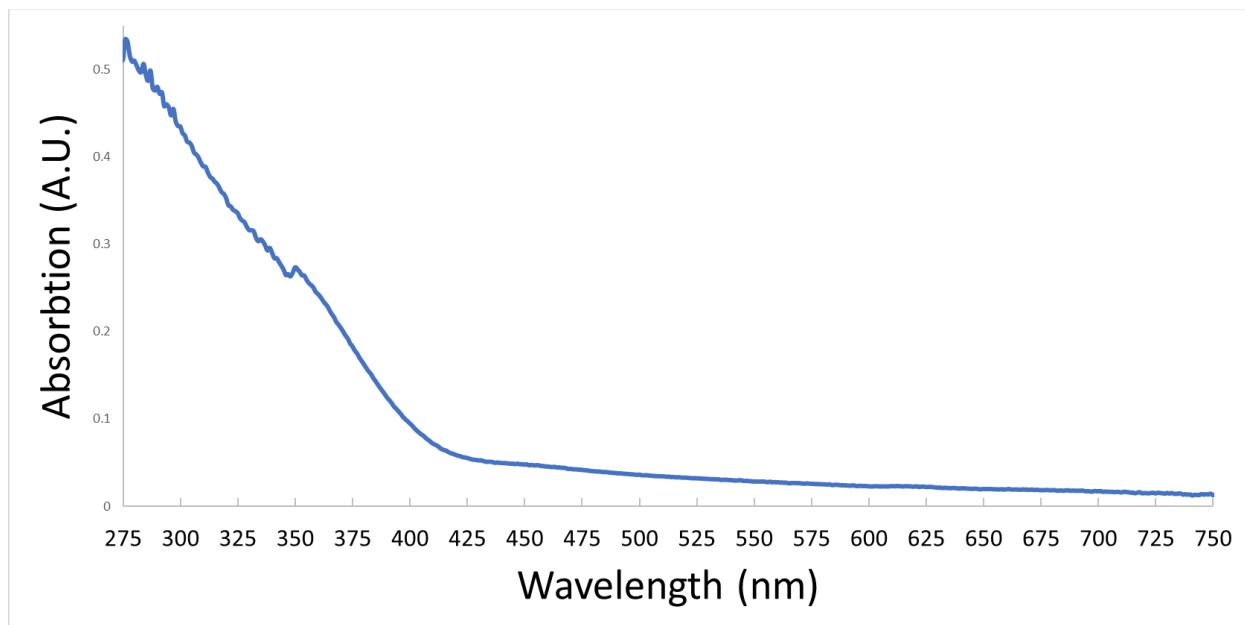


Figure S5. Room temperature electronic absorption spectra of $\text{UO}_2\text{-1}^{\text{film}}$ formed on glass substrate.

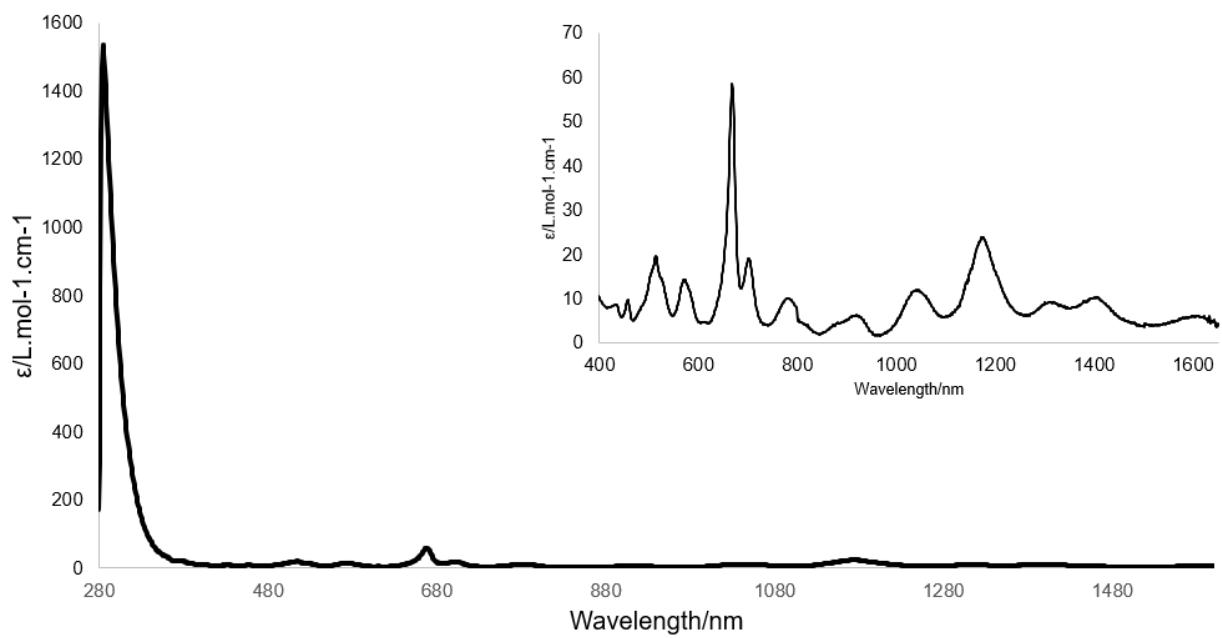


Figure S6. Room temperature electronic absorption spectra of **1**. Inset shows featured 400–1600nm in higher magnification. 1.60 mM solution in toluene.

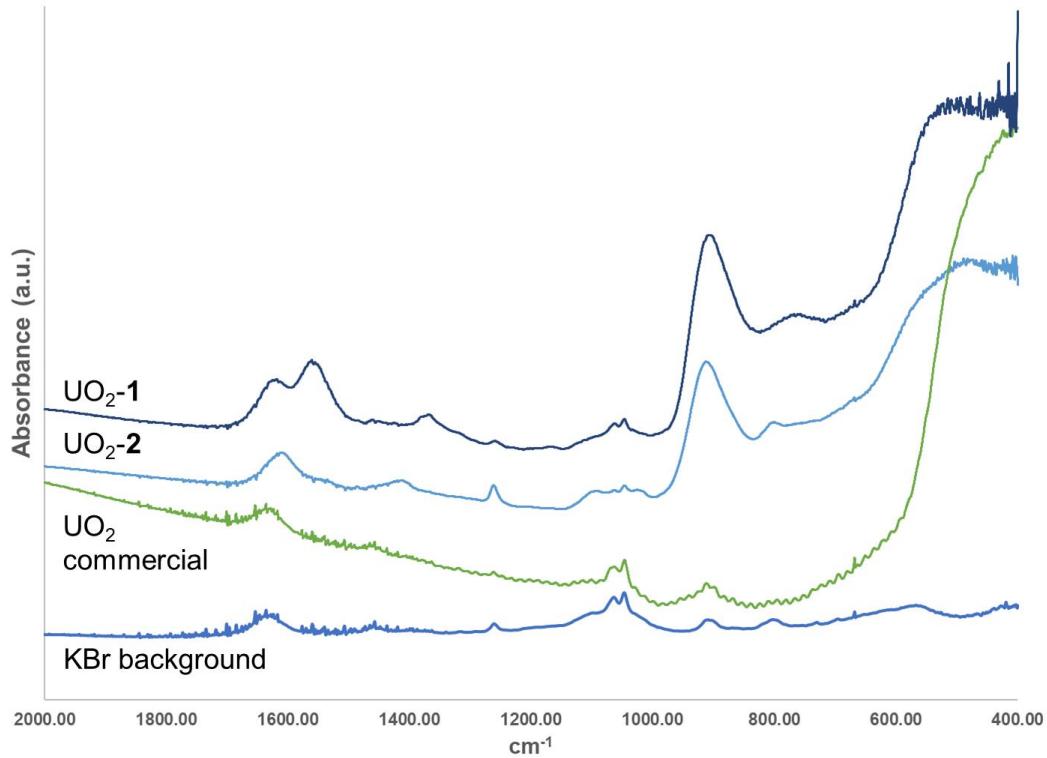


Figure S7. FT-IR spectra of UO₂-1, UO₂-2, commercial source UO₂ (KBr pellets) and the KBr background.

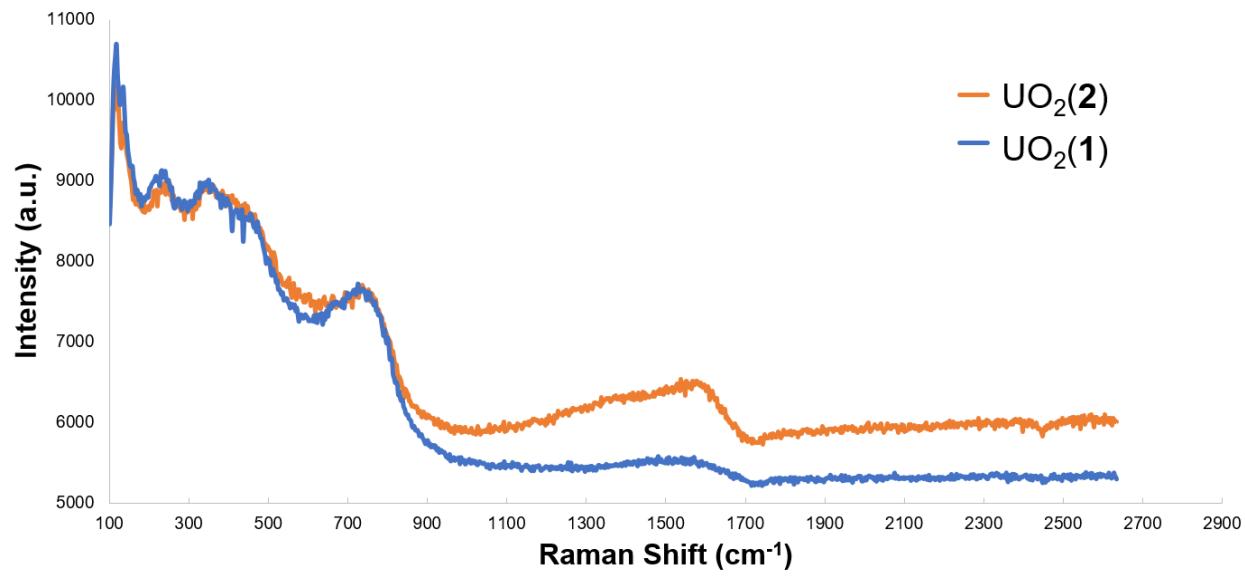


Figure S8. Raman spectra of $\text{UO}_2\text{-}\mathbf{1}$ and $\text{UO}_2\text{-}\mathbf{2}$ powder samples.

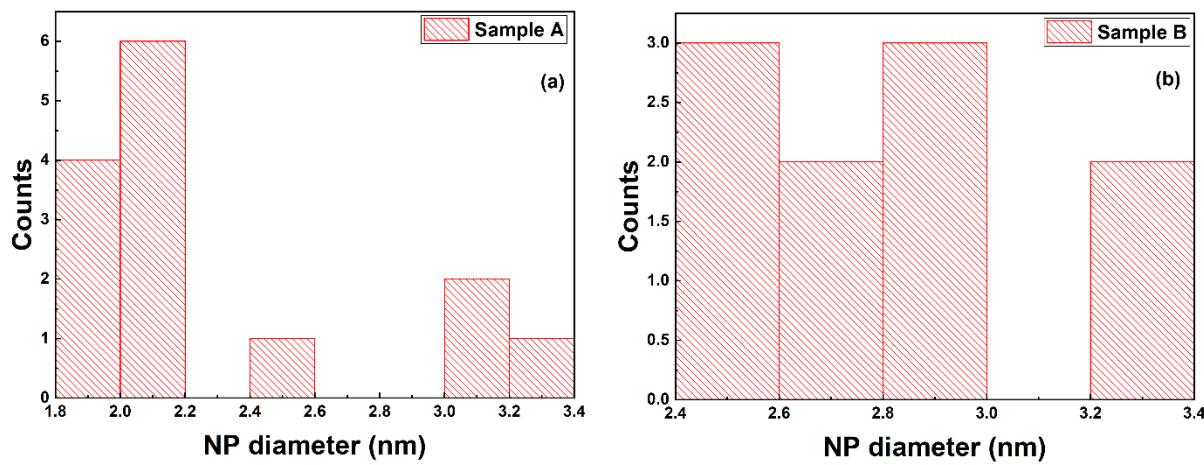


Figure S9. ImageJ generated histogram of NP size distribution in UO₂-1 (A) and UO₂-2 (B). It can be observed from the histograms that larger NPs are formed in case of sample UO₂-2, which is due to the coalescence of smaller NPs. However, discrete NPs are formed in case of UO₂-1, and the size of most of the NPs is below 2.2 nm.