

Electronic Supporting information for

**Elemental analysis: an important purity control but prone to manipulations**

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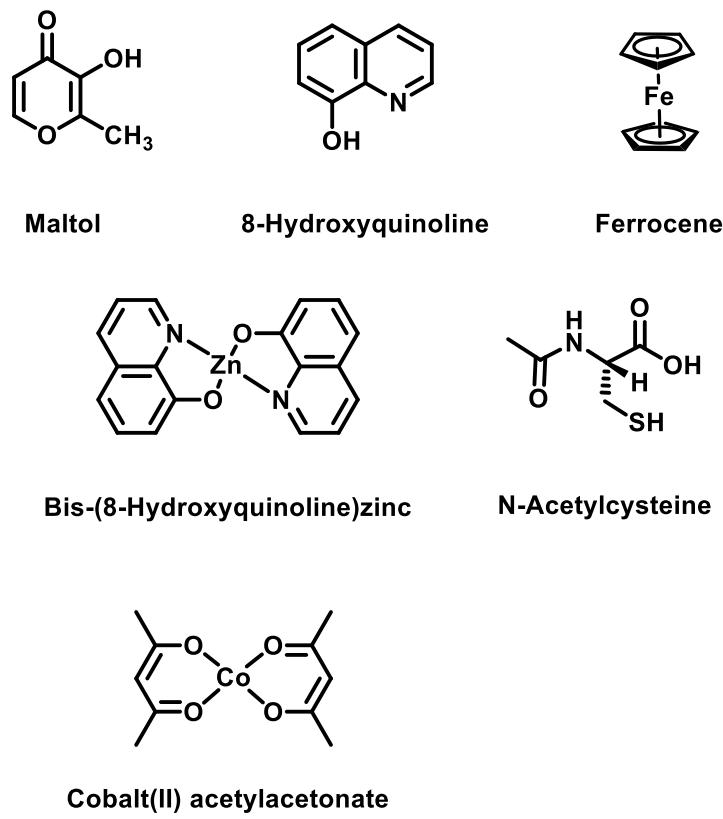
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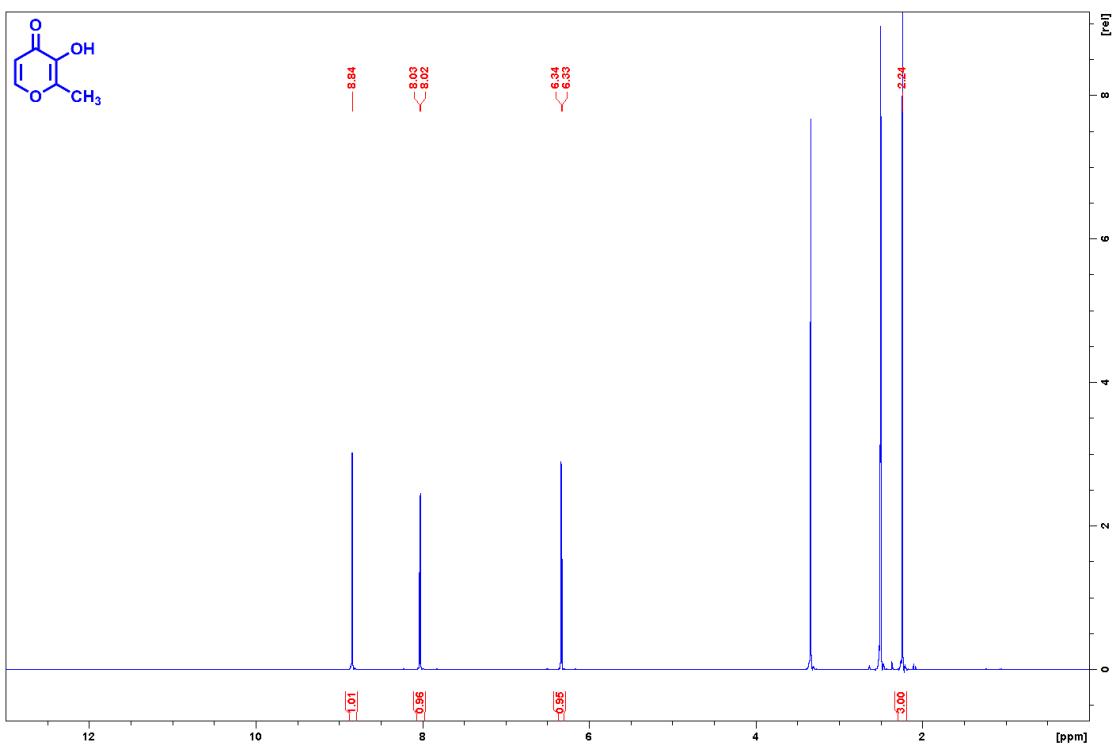
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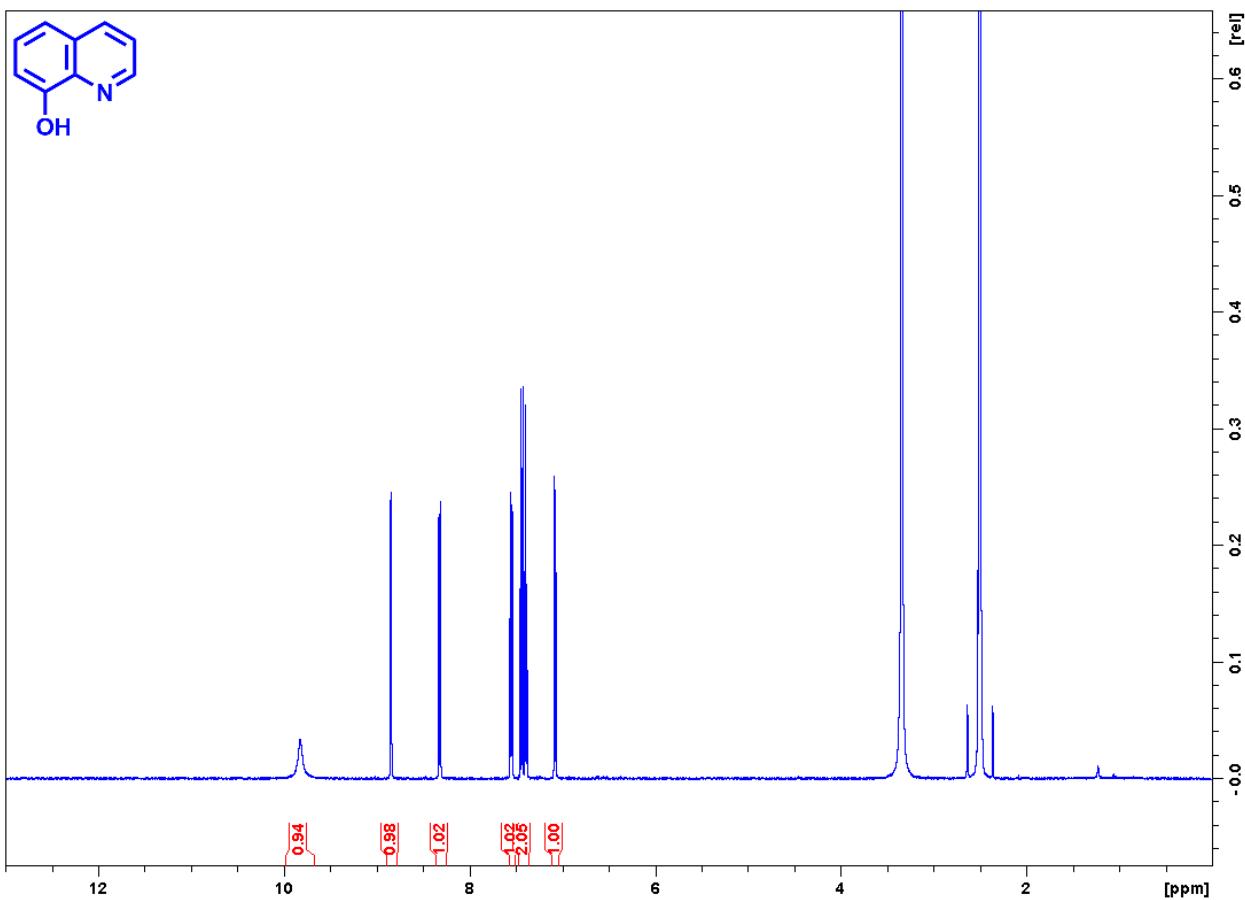
## Structure and NMR Spectra of the Investigated Compounds



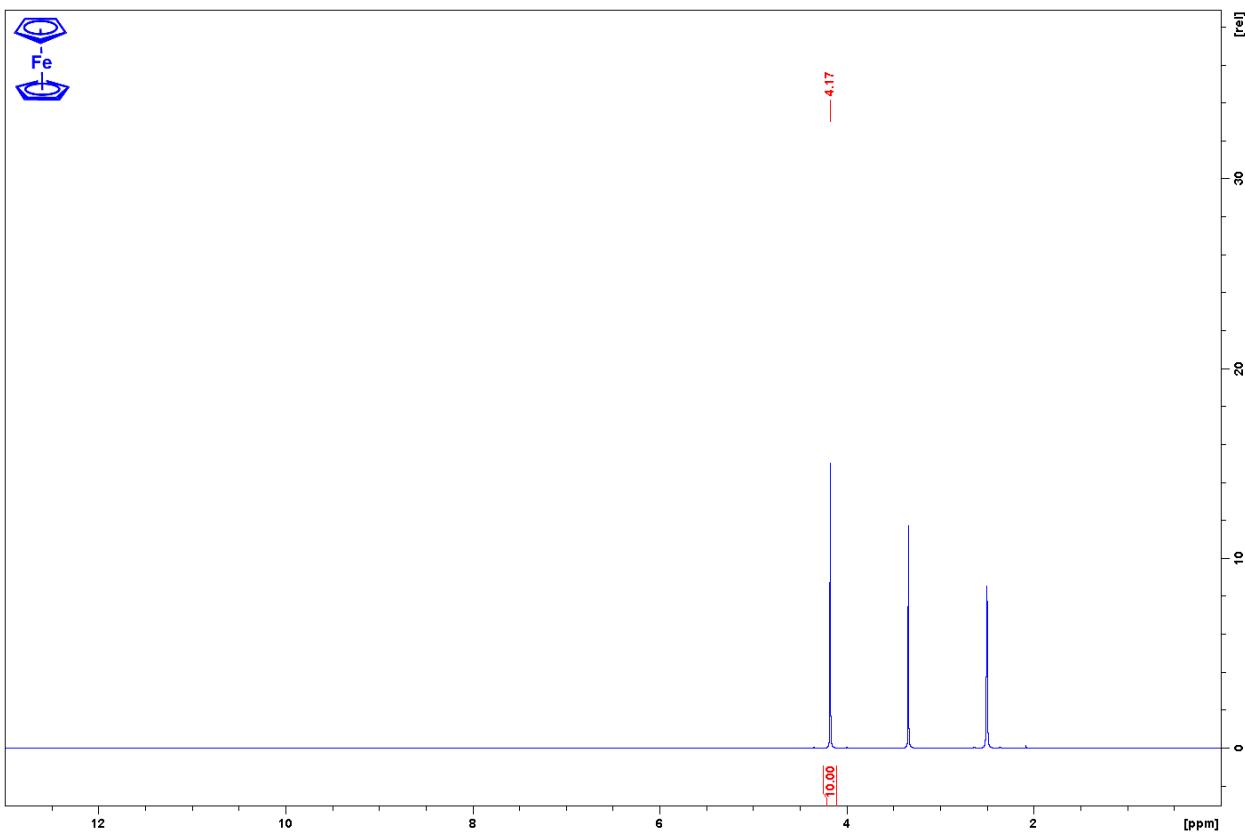
**Scheme S1.** Chemical structures of the investigated compounds.



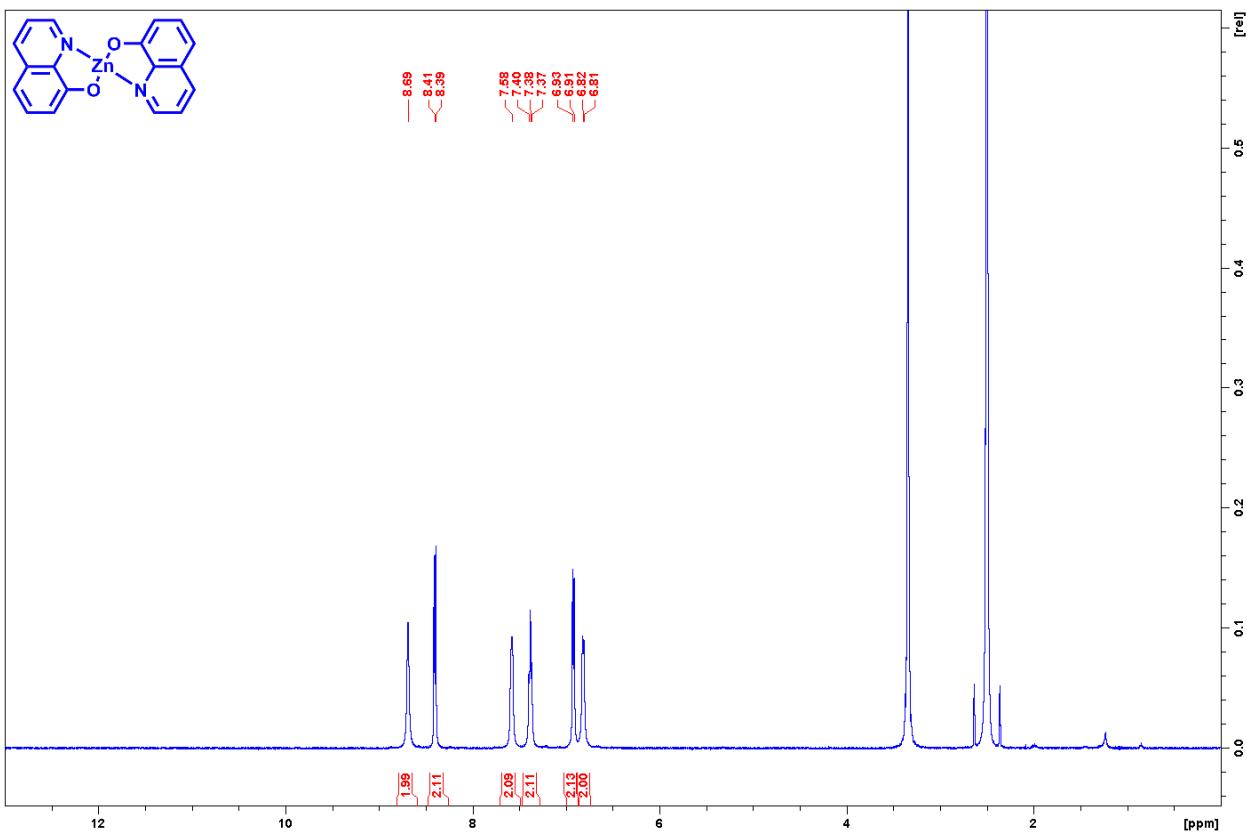
**Figure S2.**  $^1\text{H}$  NMR spectrum of 3-hydroxy-2-methyl-pyr-4-one (500.10 MHz,  $\text{d}_6$ -DMSO, 25 °C).



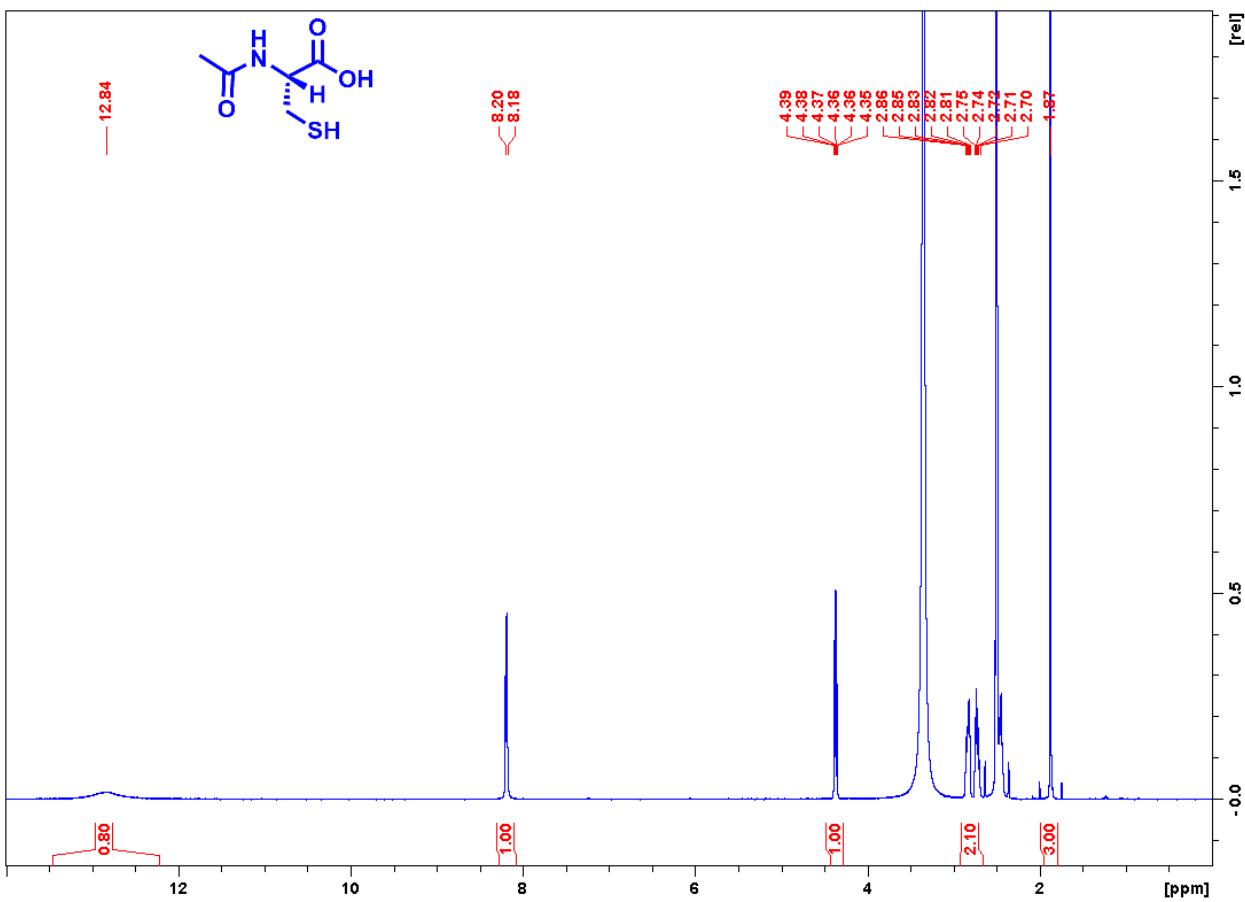
**Figure S3.** <sup>1</sup>H NMR spectrum of 8-hydroxychinoline (500.10 MHz, d<sub>6</sub>-DMSO, 25 °C).



**Figure S4.** <sup>1</sup>H NMR spectrum of ferrocene (500.10 MHz, d<sub>6</sub>-DMSO, 25 °C).



**Figure S5.** <sup>1</sup>H NMR spectrum of bis(8-hydroxyquinolinato)zinc (500.10 MHz, d<sub>6</sub>-DMSO, 25 °C).



**Figure S5.** <sup>1</sup>H NMR spectrum of *N*-acetyl-L-cysteine (500.10 MHz, d<sub>6</sub>-DMSO, 25 °C).

## Experimental Part

### Materials and Methods

All investigated compounds were bought from Sigma-Aldrich with a purity of >99%: 3-hydroxy-2-methyl-pyr-4-one; 8-hydroxyquinoline, N-acetyl-L-cysteine, ferrocene, cobalt(II) acetylacetone and bis(8-hydroxyquinolinato)zinc.  $^1\text{H}$  NMR spectra were recorded on a Bruker Avance III<sup>TM</sup> 500 MHz FT-NMR spectrometer.  $^1\text{H}$  NMR spectra were measured at 500.10 MHz from solutions in deuterated dimethyl sulfoxide.

### Elemental Analyses

Elemental analyses were performed by the Microanalytical Laboratory of the University of Vienna. Samples were weighed on a Sartorius SEC 2 ultra-micro balance with  $\pm 0.1 \mu\text{g}$  resolution. Sample weights from 1.0 and 3.0 mg were used. Calibration was done using NIST-certified standard reference material: sulfanilamide ( $\text{C}_6\text{H}_8\text{N}_2\text{O}_2\text{S}$ ) and BBOT (2, 5-bis-(5-tert-butyl-2-benzoxazol-2-yl)-thiophenone,  $\text{C}_{26}\text{H}_{26}\text{N}_2\text{O}_2\text{S}$ ) were used for C/H/N/S, acetanilide ( $\text{C}_8\text{H}_9\text{NO}$ ) was used for C/H/N-determination and L-cystine ( $\text{C}_{12}\text{H}_{21}\text{N}_2\text{O}_4\text{S}_2$ ), acetanilide and benzoic acid ( $\text{C}_7\text{H}_6\text{O}_2$ ) were used for oxygen determination. The limit of quantification (LOQ) is 0.05 w-% for C, H, N and O and 0.02 w-% for S. Using at least two different standard materials ensured to have relevant calibration for various sample types.

CHNS-analysis was performed on an EA3000 CHNSO analyzer manufactured by Eurovector. The instrument uses flash combustion and analyses the product gases by gas chromatography on line. Signals are detected using a Thermal Conductivity Detector (TCD) and recorded by the software "Callidus" supplied by the manufacturer. O-analysis used the HT 1500 high temperature unit coupled to the above instrument. Carbon monoxide is used as analytical species to quantify oxygen.

Data evaluation was done on a well-established laboratory developed software package. Calibration factors are applied as moving averages. Slightly significant changes in the gas flow during an analysis run cause a change in those factors.

CHN analyses were complemented using a 2400 CHNSO instrument from Perkin Elmer. This instrument uses an off-line process for digestion and separation. The GC-separation utilizes zone chromatography and the step height is used for quantification.

The same six compounds have also been measured by the companies: 1) Fa. HEKATech GmbH, Friedrich-List-Allee 26, D-41844 Wegberg, Germany ([www.hekatech.com](http://www.hekatech.com)) using an EA3000 CHNSO instrument; 2) Elementar Analysensysteme GmbH, Elementar-Straße 1, D-63505 Langenselbold, Germany ([www.elementar.com](http://www.elementar.com)) using a UNICUBE/rapid OXY cube instrument and 3) Solvias AG, Römerpark 2, CH-4303 Kaiseraugst, Switzerland ([www.solvias.com](http://www.solvias.com)) using a UNICUBE and EA3000 CHNSO instrument.

**3-Hydroxy-2-methyl-pyr-4-one**

Theoretical composition

C	H	N			
57.14	Δ	4.80	Δ	0.00	Δ
57.08	-0.06	4.63	-0.17	< 0,05	
57.10	-0.04	4.81	0.01	< 0,05	
57.14	0.00	4.76	-0.04	< 0,05	
<b>57.11</b>	<b>-0.03</b>	<b>4.73</b>	<b>-0.07</b>		
<i>σ</i>		0.09			

**8-Hydroxyquinoline**

Theoretical composition

C	H	N			
74.47	Δ	4.86	Δ	9.65	Δ
74.30	-0.17	4.79	-0.07	9.58	-0.07
74.22	-0.25	4.78	-0.08	9.63	-0.02
74.33	-0.14	4.80	-0.06	9.63	-0.02
<b>74.28</b>	<b>-0.19</b>	<b>4.79</b>	<b>-0.07</b>	<b>9.61</b>	<b>-0.04</b>
<i>σ</i>		0.01		0.03	

**Ferrocene**

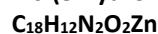
Theoretical composition

C	H	N			
64.56	Δ	5.42	Δ	0.00	Δ
64.57	0.01	5.36	-0.06	< 0,05	
64.51	-0.05	5.36	-0.06	< 0,05	
64.43	-0.13	5.39	-0.03	< 0,05	
<b>64.50</b>	<b>-0.06</b>	<b>5.37</b>	<b>-0.05</b>		
<i>σ</i>		0.02			

**Cobalt(II) acetylacetone**

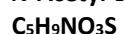
Theoretical composition

C	H	N			
46.70	Δ	5.49	Δ	0.00	Δ
45.95	-0.75	5.65	0.16	< 0,05	
46.00	-0.70	5.59	0.10	< 0,05	
45.93	-0.77	5.68	0.19	< 0,05	
<b>45.96</b>	<b>-0.74</b>	<b>5.64</b>	<b>0.15</b>		
<i>σ</i>		0.05			

**Bis-(8-hydroxyquinolinato)zinc**

Theoretical composition

C	H	N			
61.12	Δ	3.42	Δ	7.92	Δ
60.82	-0.30	3.38	-0.04	7.70	-0.22
60.71	-0.41	3.31	-0.11	7.77	-0.15
60.73	-0.39	3.32	-0.10	7.83	-0.09
<b>60.75</b>	<b>-0.37</b>	<b>3.34</b>	<b>-0.08</b>	<b>7.77</b>	<b>-0.15</b>
<i>σ</i>		0.04		0.07	

**N-Acetyl-L-cysteine**

Theoretical composition

C	H	N			
36.80	Δ	5.56	Δ	8.59	Δ
36.92	0.12	5.44	-0.12	8.45	-0.14
36.99	0.19	5.37	-0.19	8.42	-0.17
36.96	0.16	5.40	-0.16	8.46	-0.13
<b>36.96</b>	<b>0.16</b>	<b>5.40</b>	<b>-0.16</b>	<b>8.44</b>	<b>-0.15</b>
<i>σ</i>		0.04		0.02	

**Table S1:** CHN elemental analyses of the six test compounds measured by a 2400 CHNSO instrument at the Microanalytical Laboratory of the University of Vienna.

**3-Hydroxy-2-methyl-pyr-4-one**

C <sub>6</sub> H <sub>6</sub> O <sub>3</sub>	C	H	N	S				
Theoretical composition	<b>57.14</b>	<b>Δ</b>	<b>4.80</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	57.23	0.09	4.72	-0.08				
Elemental analysis 2	57.35	0.21	4.79	-0.01				
Elemental analysis 3	57.37	0.23	4.76	-0.04				
average	<b>57.32</b>	<b>0.18</b>	<b>4.76</b>	<b>-0.04</b>				
σ		<b>0.08</b>	<b>0.04</b>					

**8-Hydroxyquinoline**

C <sub>9</sub> H <sub>7</sub> NO	C	H	N	S				
Theoretical composition	<b>74.47</b>	<b>Δ</b>	<b>4.86</b>	<b>Δ</b>	<b>9.65</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	74.44	-0.03	4.83	-0.03	9.64	-0.01		
Elemental analysis 2	74.45	-0.02	4.83	-0.03	9.65	0.00		
Elemental analysis 3	74.49	0.02	4.86	0.00	9.65	0.00		
average	<b>74.46</b>	<b>-0.01</b>	<b>4.84</b>	<b>-0.02</b>	<b>9.65</b>	<b>0.00</b>		
σ		<b>0.03</b>	<b>0.02</b>		<b>0.01</b>			

**Ferrocene**

C <sub>10</sub> H <sub>10</sub> Fe	C	H	N	S				
Theoretical composition	<b>64.56</b>	<b>Δ</b>	<b>5.42</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	64.71	0.15	5.42	0.00				
Elemental analysis 2	64.76	0.20	5.28	-0.14				
Elemental analysis 3	64.78	0.22	5.43	0.01				
average	<b>64.75</b>	<b>0.19</b>	<b>5.38</b>	<b>-0.04</b>				
σ		<b>0.04</b>	<b>0.08</b>					

**Cobalt(II) acetylacetone**

C <sub>10</sub> H <sub>14</sub> CoO <sub>4</sub>	C	H	N	S				
Theoretical composition	<b>46.70</b>	<b>Δ</b>	<b>5.49</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	44.70	-2.00	5.63	0.14				
Elemental analysis 2	44.87	-1.83	5.63	0.14				
Elemental analysis 3	44.81	-1.89	5.63	0.14				
average	<b>44.79</b>	<b>-1.91</b>	<b>5.63</b>	<b>0.14</b>				
σ		<b>0.09</b>	<b>0.00</b>					

**Bis-(8-hydroxyquinolinato)zinc**

C <sub>18</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> Zn	C	H	N	S				
Theoretical composition	<b>61.12</b>	<b>Δ</b>	<b>3.42</b>	<b>Δ</b>	<b>7.92</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	60.72	-0.40	3.35	-0.07	7.88	-0.04		
Elemental analysis 2	60.63	-0.49	3.35	-0.07	7.89	-0.03		
Elemental analysis 3	60.51	-0.61	3.32	-0.10	7.90	-0.02		
average	<b>60.62</b>	<b>-0.50</b>	<b>3.34</b>	<b>-0.08</b>	<b>7.89</b>	<b>-0.03</b>		
σ		<b>0.11</b>	<b>0.02</b>		<b>0.01</b>			

**N-Acetyl-L-cysteine**

C <sub>5</sub> H <sub>9</sub> NO <sub>3</sub> S	C	H	N	S				
Theoretical composition	<b>36.80</b>	<b>Δ</b>	<b>5.56</b>	<b>Δ</b>	<b>8.59</b>	<b>Δ</b>	<b>19.65</b>	<b>Δ</b>
Elemental analysis 1	36.94	0.14	5.57	0.01	8.52	-0.07	19.65	0.00
Elemental analysis 2	36.64	-0.16	5.62	0.06	8.58	-0.01	19.69	0.04
Elemental analysis 3	36.91	0.11	5.59	0.03	8.59	0.00	19.65	0.00
average	<b>36.83</b>	<b>0.03</b>	<b>5.59</b>	<b>0.03</b>	<b>8.56</b>	<b>-0.03</b>	<b>19.66</b>	<b>0.01</b>
σ		<b>0.17</b>	<b>0.03</b>		<b>0.04</b>		<b>0.02</b>	

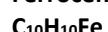
**Table S2:** CHNS elemental analyses of the six test compounds measured by HEKAttech GmbH on an EA3000 CHNSO instrument.

**3-Hydroxy-2-methyl-pyr-4-one**

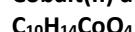
	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>57.14</b>	<b>Δ</b>	<b>4.80</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	57.10	-0.04	4.78	-0.02	< 0,05		< 0,05	
Elemental analysis 2	57.06	-0.08	4.79	-0.01	< 0,05		< 0,05	
Elemental analysis 3	57.12	-0.02	4.80	0.00	< 0,05		< 0,05	
average	<b>57.09</b>	<b>-0.05</b>	<b>4.79</b>	<b>-0.01</b>				
σ		<b>0.03</b>		<b>0.01</b>				

**8-Hydroxyquinoline**

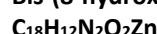
	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>74.47</b>	<b>Δ</b>	<b>4.86</b>	<b>Δ</b>	<b>9.65</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	74.38	-0.09	4.87	0.01	9.70	0.05	< 0,05	
Elemental analysis 2	74.34	-0.13	4.88	0.02	9.69	0.04	< 0,05	
Elemental analysis 3	74.40	-0.07	4.88	0.02	9.71	0.06	< 0,05	
average	<b>74.37</b>	<b>-0.10</b>	<b>4.88</b>	<b>0.02</b>	<b>9.70</b>	<b>0.05</b>		
σ		<b>0.03</b>		<b>0.01</b>		<b>0.01</b>		

**Ferrocene**

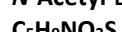
	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>64.56</b>	<b>Δ</b>	<b>5.42</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	64.46	-0.10	5.44	0.02	< 0,05		< 0,05	
Elemental analysis 2	64.55	-0.01	5.46	0.04	< 0,05		< 0,05	
Elemental analysis 3	64.51	-0.05	5.44	0.02	< 0,05		< 0,05	
average	<b>64.51</b>	<b>-0.05</b>	<b>5.45</b>	<b>0.03</b>				
σ		<b>0.05</b>		<b>0.01</b>				

**Cobalt(II) acetylacetone**

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>46.70</b>	<b>Δ</b>	<b>5.49</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	45.09	-1.61	5.74	0.25	< 0,05		< 0,05	
Elemental analysis 2	45.06	-1.64	5.67	0.18	< 0,05		< 0,05	
Elemental analysis 3	45.19	-1.51	5.75	0.26	< 0,05		< 0,05	
average	<b>45.11</b>	<b>-1.59</b>	<b>5.72</b>	<b>0.23</b>				
σ		<b>0.07</b>		<b>0.04</b>				

**Bis-(8-hydroxyquinolinato)zinc**

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>61.12</b>	<b>Δ</b>	<b>3.42</b>	<b>Δ</b>	<b>7.92</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
Elemental analysis 1	60.91	-0.21	3.43	0.01	7.93	0.01	< 0,05	
Elemental analysis 2	60.84	-0.28	3.44	0.02	7.93	0.01	< 0,05	
Elemental analysis 3	60.79	-0.33	3.43	0.01	7.94	0.02	< 0,05	
average	<b>60.85</b>	<b>-0.27</b>	<b>3.43</b>	<b>0.01</b>	<b>7.93</b>	<b>0.01</b>		
σ		<b>0.06</b>		<b>0.01</b>		<b>0.01</b>		

**N-Acetyl-L-cysteine**

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
Theoretical composition	<b>36.80</b>	<b>Δ</b>	<b>5.56</b>	<b>Δ</b>	<b>8.59</b>	<b>Δ</b>	<b>19.65</b>	<b>Δ</b>
Elemental analysis 1	36.96	0.16	5.60	0.04	8.65	0.06	19.73	0.08
Elemental analysis 2	36.89	0.09	5.61	0.05	8.62	0.03	19.65	0.00
Elemental analysis 3	36.89	0.09	5.62	0.06	8.62	0.03	19.80	0.15
average	<b>36.91</b>	<b>0.11</b>	<b>5.61</b>	<b>0.05</b>	<b>8.63</b>	<b>0.04</b>	<b>19.73</b>	<b>0.08</b>
σ		<b>0.04</b>		<b>0.01</b>		<b>0.02</b>		<b>0.08</b>

**Table S3:** CHNS elemental analyses of the six test compounds measured by Elementar GmbH on a UNICUBE instrument.

**3-Hydroxy-2-methyl-pyr-4-one**

<b>C<sub>6</sub>H<sub>6</sub>O<sub>3</sub></b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>57.14</b>	<b>Δ</b>	<b>4.80</b>	<b>Δ</b>
Elemental analysis 1	57.089	-0.05	4.807	0.01
Elemental analysis 2	57.101	-0.04	4.836	0.04
Elemental analysis 3	57.125	-0.02	4.820	0.02
average	<b>57.11</b>	<b>-0.04</b>	<b>4.82</b>	<b>0.02</b>
$\sigma$		<b>0.02</b>	<b>0.01</b>	

**8-Hydroxyquinoline**

<b>C<sub>9</sub>H<sub>7</sub>NO</b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>74.47</b>	<b>Δ</b>	<b>4.86</b>	<b>Δ</b>
Elemental analysis 1	74.715	0.25	4.871	0.01
Elemental analysis 2	74.585	0.11	4.775	-0.09
Elemental analysis 3	74.508	0.04	4.782	-0.08
average	<b>74.60</b>	<b>0.13</b>	<b>4.81</b>	<b>-0.05</b>
$\sigma$		<b>0.10</b>	<b>0.05</b>	<b>0.08</b>

**Ferrocene**

<b>C<sub>10</sub>H<sub>10</sub>Fe</b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>64.56</b>	<b>Δ</b>	<b>5.42</b>	<b>Δ</b>
Elemental analysis 1	64.470	-0.09	5.635	0.22
Elemental analysis 2	64.692	0.13	5.598	0.18
Elemental analysis 3	64.460	-0.10	5.585	0.17
average	<b>64.54</b>	<b>-0.02</b>	<b>5.61</b>	<b>0.19</b>
$\sigma$		<b>0.13</b>	<b>0.03</b>	

**Cobalt(II) acetylacetoneate**

<b>C<sub>10</sub>H<sub>14</sub>CoO<sub>4</sub></b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>46.70</b>	<b>Δ</b>	<b>5.49</b>	<b>Δ</b>
Elemental analysis 1	45.330	-1.37	5.865	0.38
Elemental analysis 2	45.164	-1.54	5.769	0.28
Elemental analysis 3	45.205	-1.50	5.798	0.31
average	<b>45.23</b>	<b>-1.47</b>	<b>5.81</b>	<b>0.32</b>
$\sigma$		<b>0.09</b>	<b>0.05</b>	

**Bis-(8-hydroxyquinolinato)zinc**

<b>C<sub>18</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>Zn</b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>61.12</b>	<b>Δ</b>	<b>3.42</b>	<b>Δ</b>
Elemental analysis 1	60.997	-0.12	3.621	0.20
Elemental analysis 2	60.769	-0.35	3.584	0.16
Elemental analysis 3	60.915	-0.20	3.649	0.23
average	<b>60.89</b>	<b>-0.23</b>	<b>3.62</b>	<b>0.20</b>
$\sigma$		<b>0.12</b>	<b>0.03</b>	<b>0.01</b>

**N-Acetyl-L-cysteine**

<b>C<sub>5</sub>H<sub>9</sub>NO<sub>3</sub>S</b>	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>
Theoretical composition	<b>36.80</b>	<b>Δ</b>	<b>5.56</b>	<b>Δ</b>
Elemental analysis 1	36.743	-0.06	5.616	0.06
Elemental analysis 2	36.732	-0.07	5.602	0.04
Elemental analysis 3	36.717	-0.08	5.607	0.05
average	<b>36.73</b>	<b>-0.07</b>	<b>5.61</b>	<b>0.05</b>
$\sigma$		<b>0.01</b>	<b>0.01</b>	<b>0.01</b>

**Table S4:** CHNS elemental analyses of the six test compounds measured by Solvias AG on a UNICUBE instrument.

**3-Hydroxy-2-methyl-pyr-4-one**

Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>57.14</b>	<b>Δ</b>	<b>4.80</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
57.073	-0.07	4.905	0.11					
57.231	0.09	4.853	0.05					
57.225	0.09	4.870	0.07					
	<b>57.18</b>	<b>0.04</b>	<b>4.88</b>	<b>0.08</b>				
		<b>0.09</b>	<b>0.03</b>					

**8-Hydroxyquinoline**

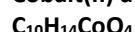
Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>74.47</b>	<b>Δ</b>	<b>4.86</b>	<b>Δ</b>	<b>9.65</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
74.163	-0.31	4.944	0.08	9.815	0.16			
74.275	-0.19	4.866	0.01	9.761	0.11			
74.385	-0.08	4.856	0.00	9.760	0.11			
	<b>74.27</b>	<b>-0.20</b>	<b>4.89</b>	<b>0.03</b>	<b>9.78</b>	<b>0.13</b>		
		<b>0.11</b>	<b>0.05</b>		<b>0.03</b>			

**Ferrocene**

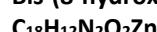
Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>64.56</b>	<b>Δ</b>	<b>5.42</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
64.739	0.18	5.455	0.04					
64.575	0.02	5.532	0.11					
64.666	0.11	5.437	0.02					
	<b>64.66</b>	<b>0.10</b>	<b>5.47</b>	<b>0.05</b>				
		<b>0.08</b>	<b>0.05</b>					

**Cobalt(II) acetylacetone**

Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>46.70</b>	<b>Δ</b>	<b>5.49</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
44.718	-1.98	5.894	0.40					
45.017	-1.68	5.831	0.34					
45.265	-1.44	5.784	0.29					
	<b>45.00</b>	<b>-1.70</b>	<b>5.84</b>	<b>0.35</b>				
		<b>0.27</b>	<b>0.06</b>					

**Bis-(8-hydroxyquinolinato)zinc**

Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>61.12</b>	<b>Δ</b>	<b>3.42</b>	<b>Δ</b>	<b>7.92</b>	<b>Δ</b>	<b>0.00</b>	<b>Δ</b>
60.829	-0.29	3.280	-0.14	7.932	0.01			
60.787	-0.33	3.459	0.04	7.977	0.06			
60.841	-0.28	3.451	0.03	7.943	0.02			
	<b>60.82</b>	<b>-0.30</b>	<b>3.40</b>	<b>-0.02</b>	<b>7.95</b>	<b>0.03</b>		
		<b>0.03</b>	<b>0.10</b>		<b>0.02</b>			

**N-Acetyl-L-cysteine**

Theoretical composition  
Elemental analysis 1  
Elemental analysis 2  
Elemental analysis 3  
average  
 $\sigma$

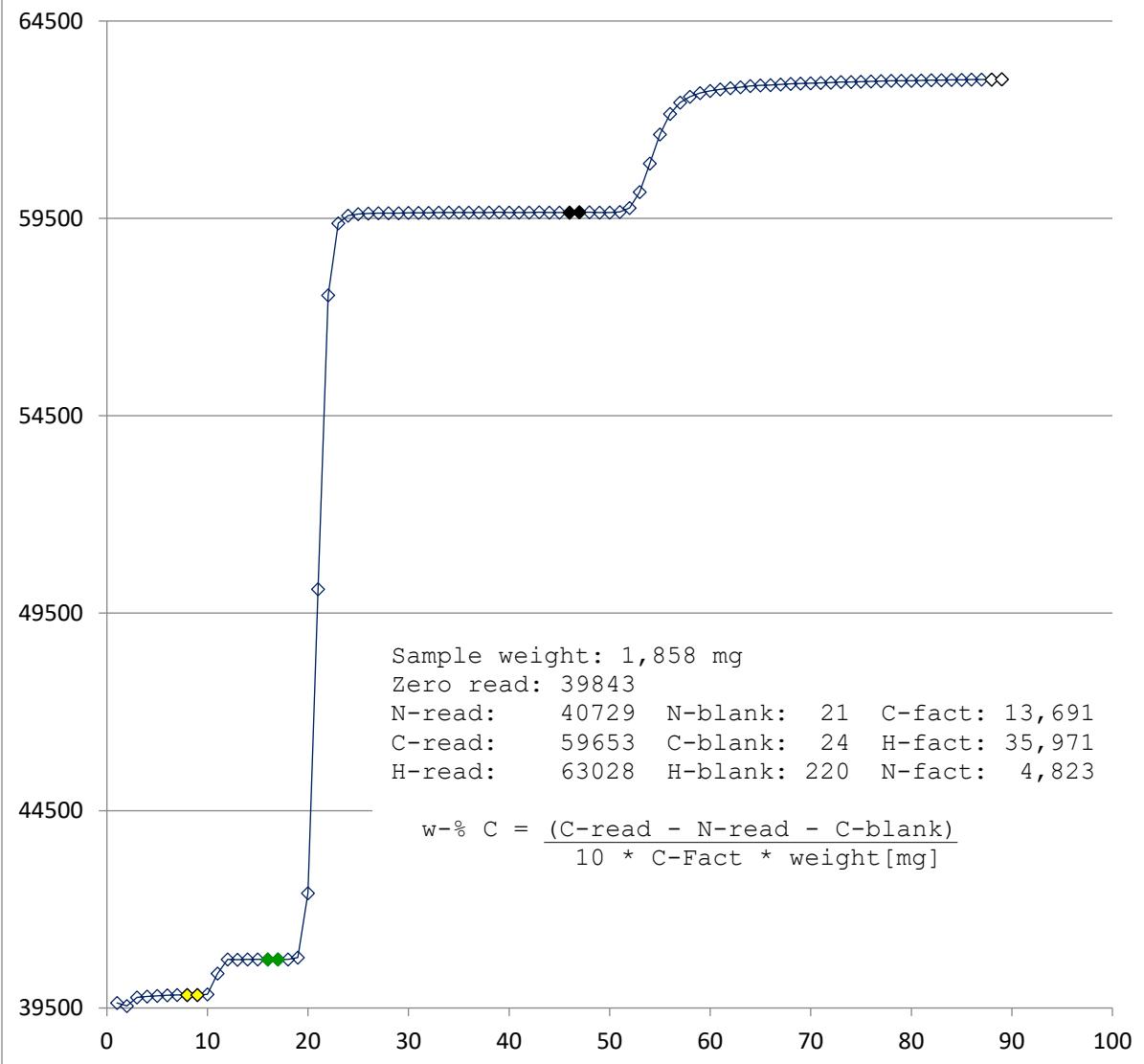
	<b>C</b>	<b>H</b>	<b>N</b>	<b>S</b>				
	<b>36.80</b>	<b>Δ</b>	<b>5.56</b>	<b>Δ</b>	<b>8.59</b>	<b>Δ</b>	<b>19.65</b>	<b>Δ</b>
36.877	0.08	5.647	0.09	8.637	0.05	20.349	0.70	
36.976	0.18	5.599	0.04	8.619	0.03	20.040	0.39	
36.994	0.19	5.574	0.01	8.631	0.04	20.371	0.72	
	<b>36.95</b>	<b>0.15</b>	<b>5.61</b>	<b>0.05</b>	<b>8.63</b>	<b>0.04</b>	<b>20.25</b>	<b>0.60</b>
		<b>0.06</b>	<b>0.04</b>		<b>0.01</b>		<b>0.19</b>	

**Table S5:** CHNS elemental analyses of the six test compounds measured by Solvias AG on an EA3000 CHNSO instrument.

<b>3-Hydroxy-2-methyl-pyr-4-one (<math>C_6H_6O_3</math>)</b>	<b>O</b>			
Theoretical composition	<b>38.06</b>	<b>Δ</b>		
Elemental analysis 1	37.98	-0.08		
Elemental analysis 2	38.03	-0.03		
Elemental analysis 3	38.04	-0.02		
average	<b>38.02</b>	<b>-0.04</b>		
$\sigma$	0.03			
<b>8-Hydroxyquinoline (<math>C_9H_7NO</math>)</b>	<b>O</b>			
Theoretical composition	<b>11.02</b>	<b>Δ</b>		
Elemental analysis 1	11.07	0.05		
Elemental analysis 2	11.06	0.04		
Elemental analysis 3	11.01	-0.01		
average	<b>11.05</b>	<b>0.03</b>		
$\sigma$	0.03			
<b>Cobalt(II) acetylacetone (<math>C_{10}H_{14}CoO_4</math>)</b>	<b>O                      <math>C_{10}H_{14}CoO_4 * 0.25H_2O</math></b>			
Theoretical composition	<b>24.89</b>	<b>Δ</b>	<b>25.99</b>	<b>Δ</b>
Elemental analysis 1	25.97	1.08	25.97	-0.02
Elemental analysis 2	26.07	1.18	26.07	0.08
Elemental analysis 3	25.84	0.95	25.84	-0.15
average	<b>25.96</b>	<b>1.07</b>	<b>25.96</b>	<b>-0.03</b>
$\sigma$	0.12			
<b>Bis-(8-hydroxyquinolinato)zinc (<math>C_{18}H_{12}N_2O_2Zn</math>)</b>	<b>O</b>			
Theoretical composition	<b>9.05</b>	<b>Δ</b>		
Elemental analysis 1	9.15	0.10		
Elemental analysis 2	8.76	-0.29		
Elemental analysis 3	8.69	-0.36		
average	<b>8.87</b>	<b>-0.18</b>		
$\sigma$	0.25			
<b>N-Acetyl-L-cysteine (<math>C_5H_9NO_3S</math>)</b>	<b>O</b>			
Theoretical composition	<b>29.41</b>	<b>Δ</b>		
Elemental analysis 1	29.35	-0.06		
Elemental analysis 2	29.33	-0.08		
Elemental analysis 3	29.37	-0.04		
average	<b>29.35</b>	<b>-0.06</b>		
$\sigma$	0.02			

**Table S6:** Measured oxygen values of the five oxygen containing compounds using an EA3000 CHNSO instrument at the Microanalytical Laboratory of the University of Vienna.

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**Figure S6.** Exemplary elemental analysis chromatogram with the respective peak areas and scale factors measured on a 2400 CHNSO Perkin Elmer instrument. The yellow diamonds indicate the blank, the green N, black C and white H values.