

## Supporting Information

**Table S1** Fraction of Q<sup>n</sup> Si population determined by quantitative data of solid-state <sup>29</sup>Si MAS NMR spectra

No.	Sample name	Chemical shift range (ppm)	(Q <sup>2</sup> +Q <sup>3</sup> )/Q <sup>n</sup> (%)	Q <sup>4</sup> /Q <sup>n</sup> (%)
1	<i>R0.5</i>	-80.0 ~ -127.4	64	36
2	<i>R0.5S5</i>	-81.5 ~ -126.0	55	45
3	<i>R0.5S10</i> (or <i>RS-160</i> )	-85.8 ~ -132.8	46	54
4	<i>RS-112</i>	-89.4 ~ -122.7	45	55
5	<i>RS-28</i>	-88.9 ~ -123.0	44	56
6	<i>R0.5S10-AT</i>	-80.0 ~ -132.4	55	45

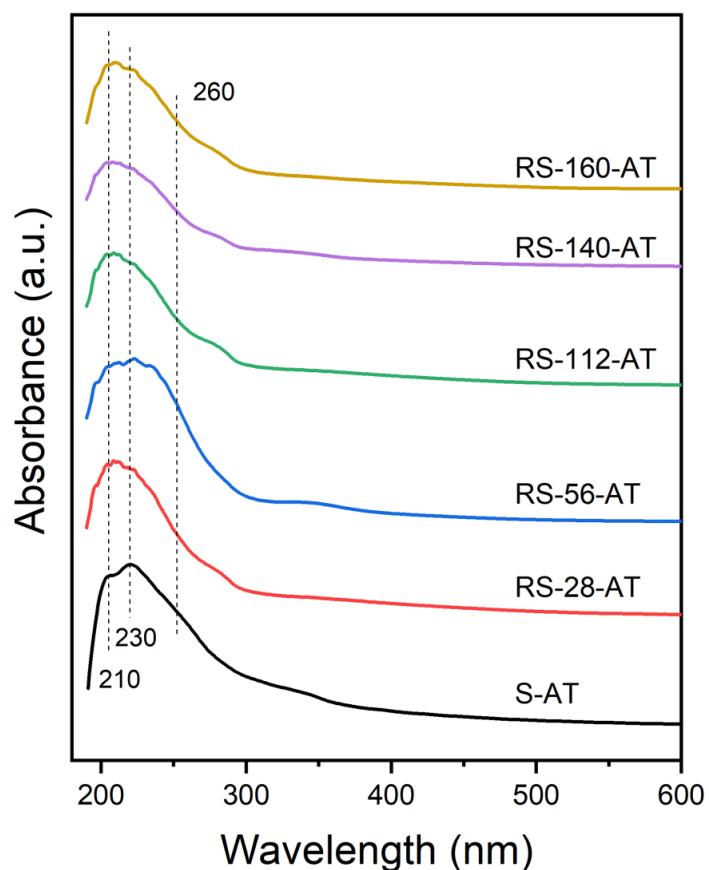
**Table S2** TG analysis results of zeolite precursors and catalysts

Sample	303~473 K	473~823 K	823~1073 K	Total weight loss	Removal rates of PI
	(%)	(%)	(%)	(303~1073 K)	(%)
<i>RS-28</i>	7.95	12.97	1.60	22.52	
<i>RS-28-AT</i>	9.56	5.34	1.27	16.17	46.42
<i>RS-56</i>	5.21	12.77	1.85	19.83	
<i>RS-56-AT</i>	8.79	6.86	1.64	17.29	58.82
<i>RS-112</i>	8.23	12.02	1.39	21.64	
<i>RS-112-AT</i>	9.59	4.89	1.08	15.56	59.35
<i>RS-140</i>	6.83	14.17	1.33	22.33	
<i>RS-140-AT</i>	8.87	5.48	1.24	15.59	61.32
<i>RS-160</i>	5.54	11.91	1.07	18.52	
<i>RS-160-AT</i>	9.22	4.52	1.05	14.79	62.05

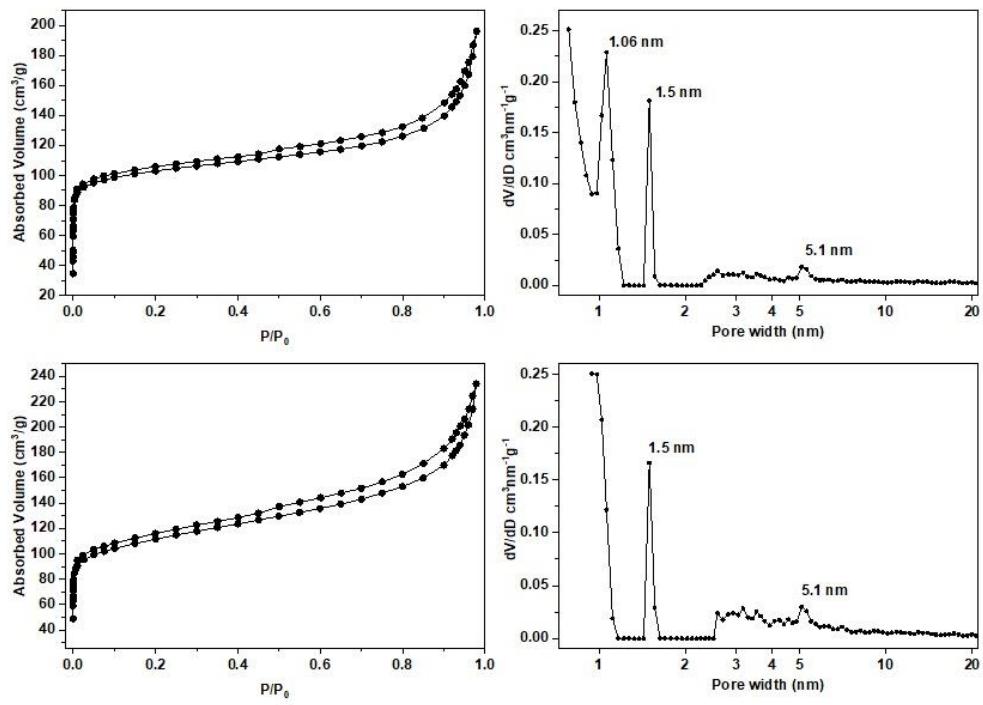
**Table S3** Textural properties of Fe<sub>2</sub>O<sub>3</sub> supported zeolite catalysts<sup>a</sup>

Catalysts	Total pore volume (cm <sup>3</sup> g <sup>-1</sup> )	BET surface area (m <sup>2</sup> g <sup>-1</sup> )	Exter. BET surface area (m <sup>2</sup> g <sup>-1</sup> )	Micro. BET surface area (m <sup>2</sup> g <sup>-1</sup> )	L/B	Acid amount (mmol/g)
Fe <sub>2</sub> O <sub>3</sub> @RS-28-C	0.38	422	90	332	77.9	0.71
Fe <sub>2</sub> O <sub>3</sub> @RS-56-C	0.37	445	99	346	66.5	0.75
Fe <sub>2</sub> O <sub>3</sub> @RS-112-C	0.41	490	111	379	17.8	1.19
Fe <sub>2</sub> O <sub>3</sub> @RS-140-C	0.40	502	112	390	4.2	1.49
Fe <sub>2</sub> O <sub>3</sub> @RS-160-C	0.41	498	110	388	12.7	0.35

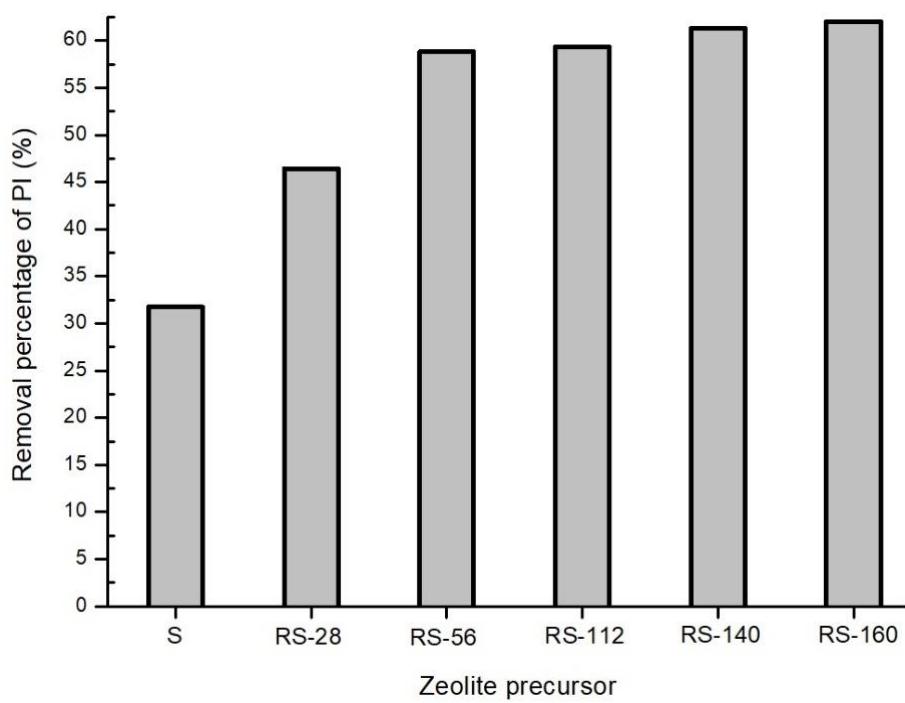
<sup>a</sup> The BET surface area and pore volume are calculated based on the N<sub>2</sub> adsorption-desorption analysis; the acid amount is calculated based on the NH<sub>3</sub>-TPD analysis results; the L/B ratio was calculated based on the pyridine-IR analysis results.



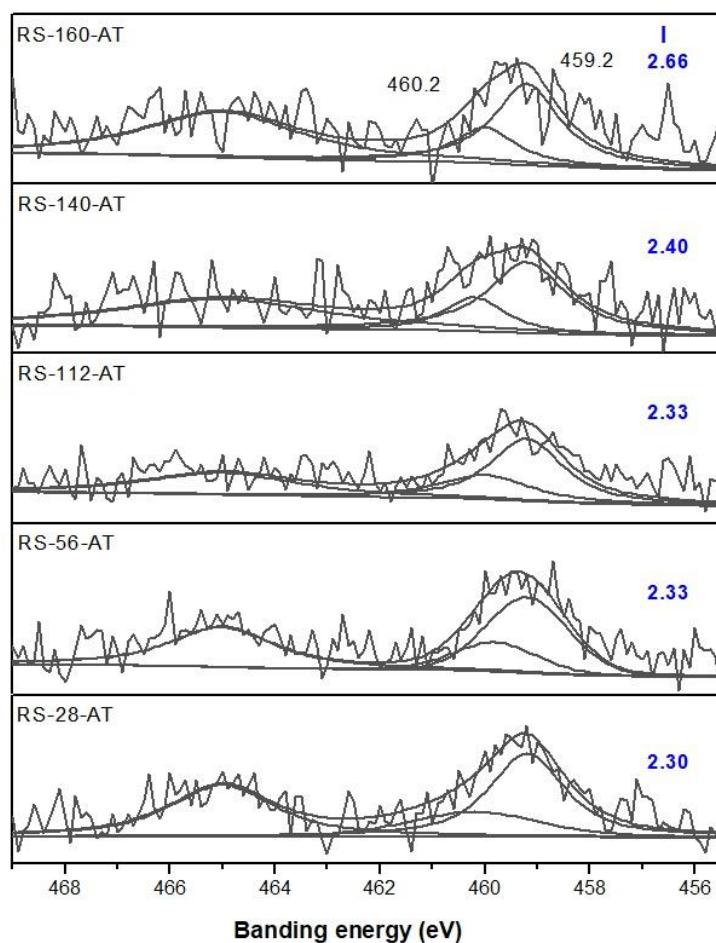
**Figure S1** UV-vis spectra of *RS*-series catalysts



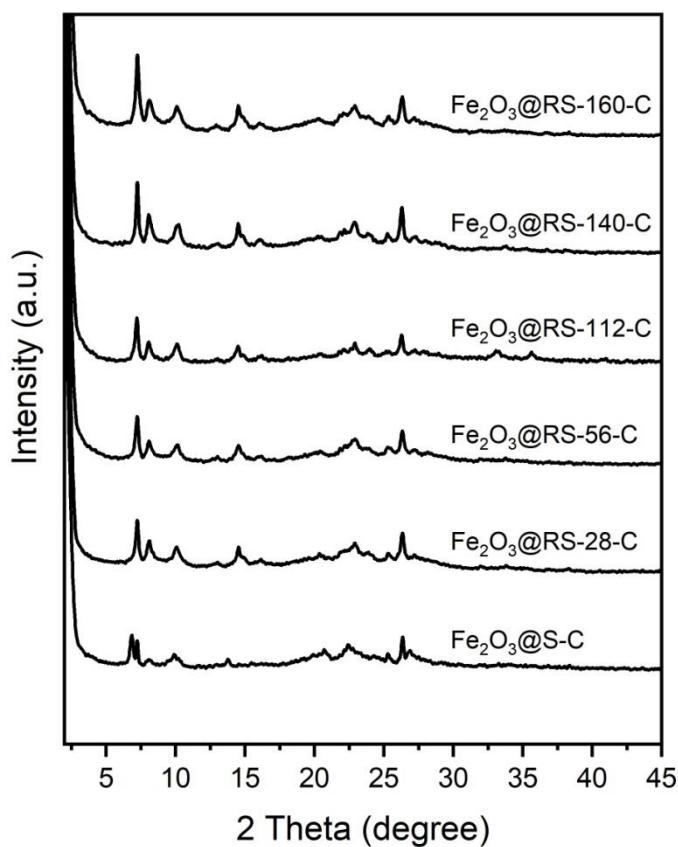
**Figure S2**  $N_2$  adsorption-desorption isotherms and pore size distributions of zeolite catalysts *RS-140-AT* (top) and *RS-160-AT* (bottom).



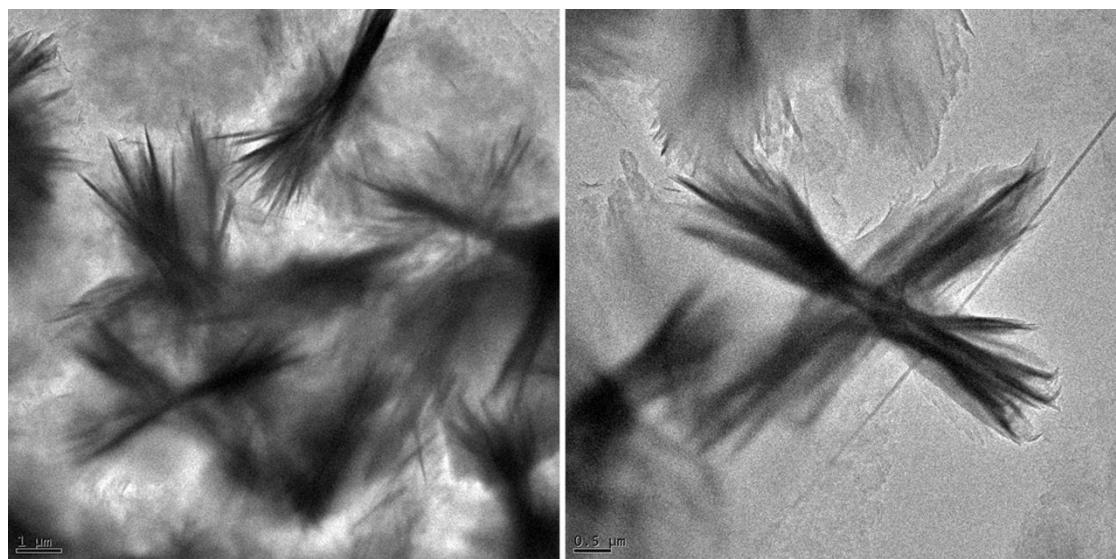
**Figure S3** Removal percentage of PI in zeolite precursor by acid treatment



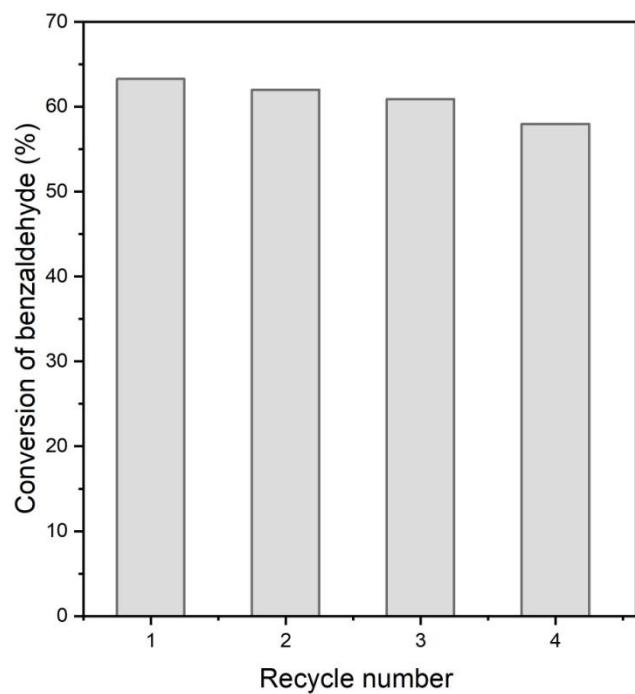
**Figure S4** XPS spectra of *RS*-series catalysts



**Figure S5** XRD patterns of 28% $\text{Fe}_2\text{O}_3$  supported zeolite catalysts



**Figure S6** HRTEM images of *RS-160* zeolite precursor.



**Figure S7** Recycling test in the condensation of benzaldehyde with ethylene glycol over catalyst *RS-160-AT* (Reaction conditions: catalyst, 25 mg; cyclohexane, 6 ml; benzaldehyde, 5 mmol; ethylene glycol, 7.5 mmol; Temp., 363 K; time, 5 h)