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## Supporting Information

for

## Preparation of nickel-bound porous carbon and its application in supercapacitors

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1. Preparation process of the working electrode

The working electrode was obtained by pressing the active material film on the nickel foam ( $\sim 1 \text{ cm}^2$ ). The film was created as a result of mixing active material (80 wt%), acetylene black (15 wt%), polytetrafluoroethylene (5 wt%), and a small amount of ethanol followed by grounding and flattening. After the process, the electrode was dried in an oven at 60°C for 2 h before testing





Figure S1. XRD patterns of Ni-PC-1, Ni-PC-2, Ni-PC-3 and Ni-PC-4.

Figure S2



Figure S2. XRD pattern of PC.





Figure S3. Raman spectra of Ni-PC-1, Ni-PC-2, Ni-PC-3 and Ni-PC-4.

Figure S4



Figure S4. Raman spectrum of PC. Figure S5



Figure S5. SEM images of Ni-PC-1 (a), Ni-PC-2 (b), Ni-PC-3 (c) and Ni-PC-4 (d).



Figure S6. The nitrogen adsorption-desorption curves (a) and the pore size distribution curves (b) of Ni-PC-1, Ni-PC-2, Ni-PC-3 and Ni-PC-4.



Figure S7. SEM (a, d), TEM (c), and HRTEM (d) images of Ni-PC-2 after the after the electrochemical experiment.

Figure S8



Figure S8. XPS spectra of Ni-PC-2 after the electrochemical experiment: (a) Survey XPS; (b–e) High-resolution XPS spectra of  $C_{1s}$ ,  $N_{1s}$ ,  $O_{1s}$ , and  $Ni_{2p}$ , respectively.



Figure S9. The SEM image and corresponding elemental mapping images of C, O, N, and Ni for Ni-PC-s-4 (a) and Ni-PC-p-16 (b).





Figure S10. (a) CV and (b) GCD curves of Ni-PC-s; (c) CV and (d) GCD curves of Ni-PC-p.



## and Ni-PC-4.

	Capacitance (F g <sup>-1</sup> )	Cycling capability (after 10000 cycles)
Ni-PC-1	522.5	111.5%
Ni-PC-2	565	105.8%
Ni-PC-3	427.5	69.3%
Ni-PC-4	285.0	95.4%
Ni-PC-s-x	163.5	98.8%
Ni-PC-p-4	178.9	83.3%
Ni-PC-p-8	198.4	106.7%
Ni-PC-p-16	295.6	120.7%
Ni-PC-p-32	282.4	150.4%