

## Electronic Supplementary Information (ESI)

### Electrochemical fixation of CO<sub>2</sub> over Mo plate to prepare Mo<sub>2</sub>C film for electrocatalytic hydrogen evolution

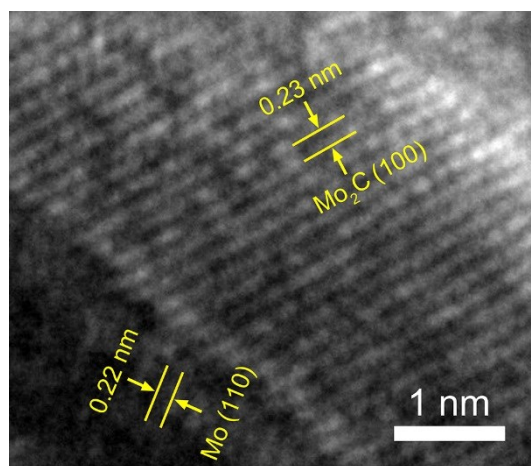
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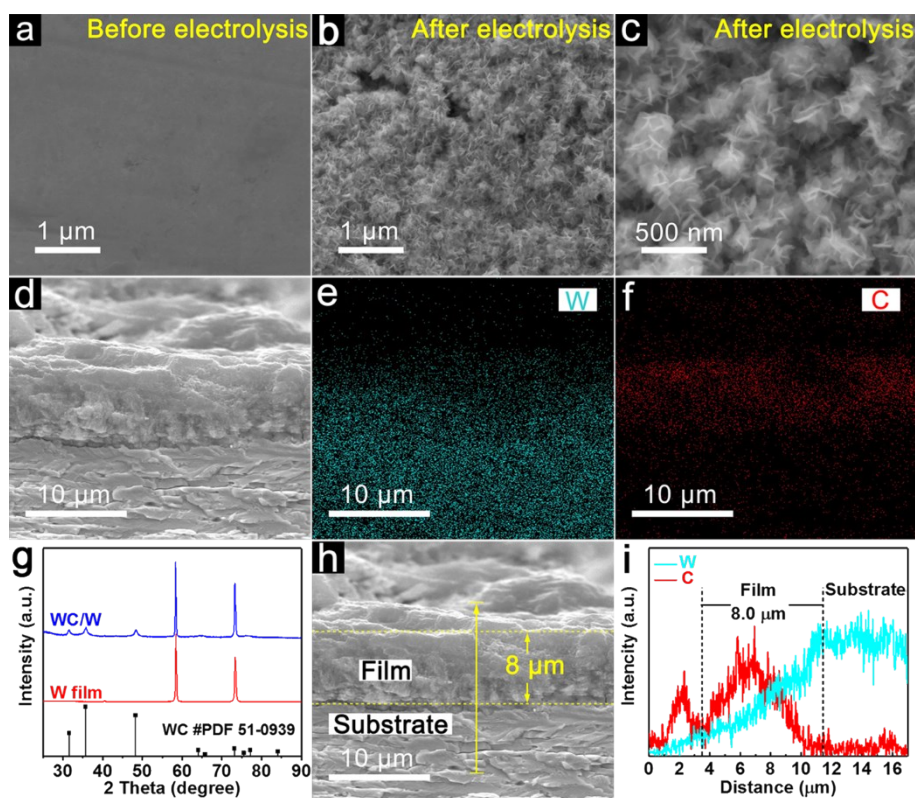
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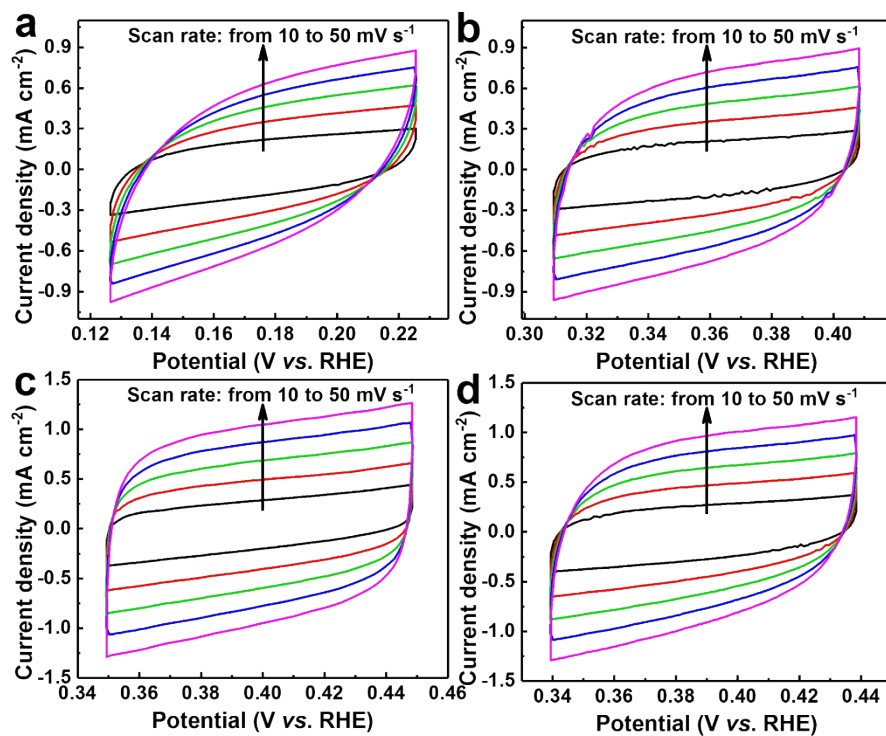
**Fig. S1** High-resolution TEM image of the Mo<sub>2</sub>C/Mo-60.

The film is extremely stable and can hardly peel off from the substrate even upon violent ultrasonication. For TEM test, powder is scratched from the substrate by a blade. The TEM image in Fig. S1 is based on the scratched powder.

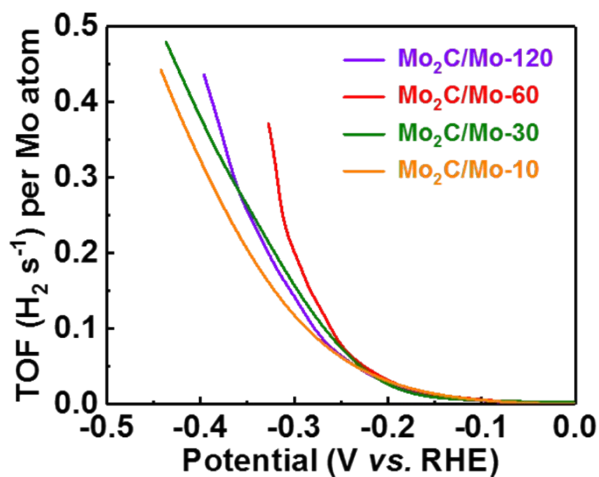


**Fig. S2** Typical SEM images of W plate (a) before electrolysis and (b-c) after electrolysis. Elemental distribution mappings (d-f) after electrolysis. (g) XRD patterns of WC/W and Mo film. Cross-sectional SEM images of W plates after electrolysis (h) and corresponding EDS results of linear sweep analysis (i). Electrolysis time: 120 min.; Cell voltage: 3.1 V.

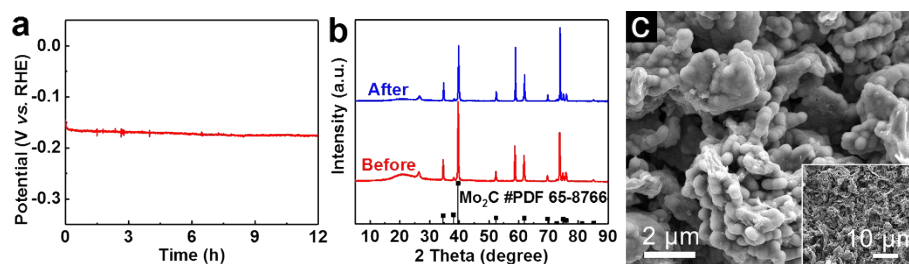
Similar with the Mo<sub>2</sub>C-coated Mo plates, the electrochemical treatment of tungsten substrate was conducted at 3.1 V for 2 h in 900 °C molten Li<sub>2</sub>CO<sub>3</sub>. Then the W plate cathode was lifted out from the molten salt and cooled to room temperature. Finally, the tungsten carbide-coated tungsten plate (denoted as WC/W) was obtained after rinsing and drying at 60 °C.



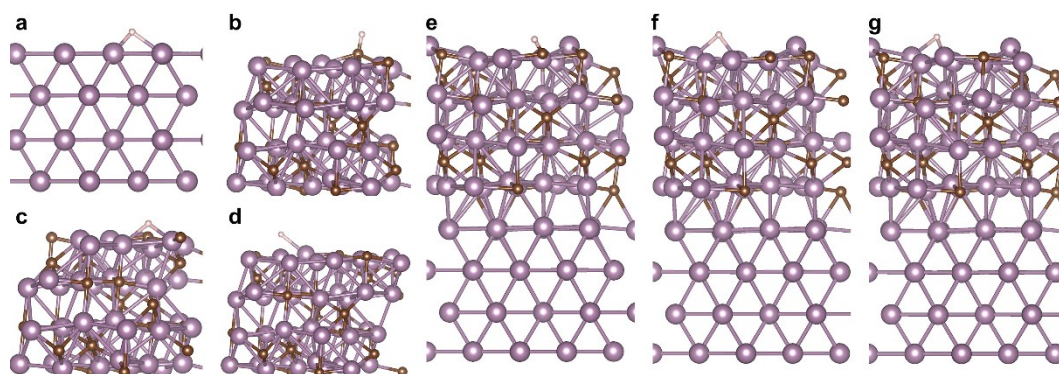
**Fig. S3** CV curves measured at different scan rates from 10 to 50  $\text{mV s}^{-1}$  in 0.5 M  $\text{H}_2\text{SO}_4$  for  $\text{Mo}_2\text{C}/\text{Mo}$  samples: (a)  $\text{Mo}_2\text{C}/\text{Mo}-10$ , (b)  $\text{Mo}_2\text{C}/\text{Mo}-30$ , (c)  $\text{Mo}_2\text{C}/\text{Mo}-60$ , and (d)  $\text{Mo}_2\text{C}/\text{Mo}-120$ .



**Fig. S4** TOFs of  $\text{Mo}_2\text{C}/\text{Mo}-10$ ,  $\text{Mo}_2\text{C}/\text{Mo}-30$ ,  $\text{Mo}_2\text{C}/\text{Mo}-60$  and  $\text{Mo}_2\text{C}/\text{Mo}-120$ .



**Fig. S5** (a) Long-term HER stability test of Mo<sub>2</sub>C/Mo-60 at 10 mA cm<sup>-2</sup>. (b) XRD patterns of Mo<sub>2</sub>C/Mo-60 before and after stability test. (c) SEM image after stability test.



**Fig. S6** Optimized geometry structures of hydrogen adsorption on (a) the surface Mo site of Mo (110), the surface (b) C, (c) Mo1 and (d) Mo2 sites of Mo<sub>2</sub>C (121) and (e) C, (f) Mo1 and (g) Mo2 sites of Mo (110)-Mo<sub>2</sub>C (121).

**Table S1** The electrochemical properties of all the tested samples

Samples	Overpotential (mV) @10 mA cm <sup>-2</sup>	Tafel slope (mV dec <sup>-1</sup> )	R <sub>s</sub> (Ω)	R <sub>ct</sub> (Ω)
Pt/C	25	45.3	2.20	9.9
Mo film	375	321.6	1.78	177.7
Mo <sub>2</sub> C/Mo	149	77.9	2.15	23.8

**Table S2** Comparison of Mo<sub>2</sub>C-based catalysts for HER performance in 0.5 M H<sub>2</sub>SO<sub>4</sub>

Catalyst	Overpotential (mV) @10 mA cm <sup>-2</sup>	Tafel slope (mV dec <sup>-1</sup> )	Preparation method	Reference
Mo <sub>2</sub> C/Mo	149	77.9	molten salt	This work
Ni-Mo <sub>2</sub> C <sub>CB</sub> /CFP	121.4	116.9	molten salt	1
Mo <sub>2</sub> C/RGO	130	57.3	hydrothermal	2
Mo <sub>2</sub> C/CC	140	124	hydrothermal	3
Mo <sub>2</sub> C/NC	140	116	hydrothermal	4
L-Mo <sub>2</sub> C	170	77	molten salt	5
Mo <sub>2</sub> C-G	150	55	hydrothermal	6
H-Mo <sub>2</sub> C/C	160	66	electrospinning	7
MoS <sub>2</sub> /CS <sub>2</sub>	208	43	Colloidal synthesis	8
MoS <sub>2</sub> @FePS <sub>3</sub>	168	127	hydrothermal	9

**Supplementary references**

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