

1 **Applying multi-scale silica-like three-dimensional networks**
2 **in PEO matrix via in-situ crosslinking for high-performance**
3 **solid composite electrolyte**

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19 **Calculation of crystallinity rate**

20 The corresponding crystallinity(χ_c) of the SPE was calculated according to the
21 following equation:

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$$\chi_c = \frac{S_{m-peak}}{S_{m-peak}^{pure\ PEO} f_{PEO}} \times 100\%$$

23 Where S_{m-peak} is the integral area of melting peak of different samples in the DSC
24 curve, f_{PEO} is the mass percentage of PEO in different samples.

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26 **Fitting of the Vogel-Tamman-Fulcher (VTF) quasi-thermodynamic model**

27 The conductivity-temperature curves of different samples were fitted to VTF
28 conducting model via the following equation:

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$$\sigma = AT^{-1/2} e^{\frac{-E_a}{T-T_0}}$$

30 Where A is the former factor, which relates to the concentration of carriers in
31 theory, E_a is the activation energy, T_0 is thermodynamic equilibrium temperature for
32 ideal glass transition.

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34 **Calculation of the distribution of the nano SiO₂ samples**

35 Series of distribution curves in the aspect of diameter and length were calculated
36 by classifying the products (over 150 times per sample) into different classes (discrete
37 but uniform) via observing the SEM images of different samples (such as **Fig. 1(a)-(d)**
38 in the manuscript). Secondly, an orthogonal multiply was conducted to acquire 2D
39 distribution cloud maps of different samples, which realizes the combination of length
40 and diameter.

41 In this work, the ratio of length to diameter was set as the standard: ≤ 2 was taken
42 as sphere (particle), with ≤ 10 taken as rod and > 10 taken as wire. In this way, the 2D

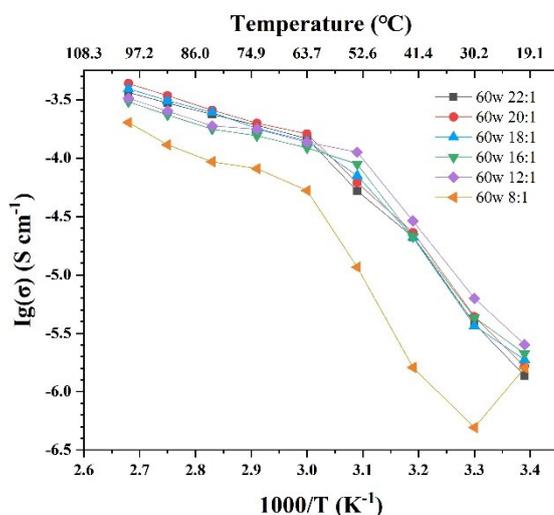
43 maps were divided into several regions (the slope of the top cut-off straight is 10, the
 44 second is 2 and the last is 1 in **Fig. S3**). By summing up the relative frequencies of
 45 different regions and making calibrations, the proportions of the three kinds could be
 46 calculated.

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Table S1 Detailed components of each sample

Sample	PEO/ g	EO ₁ / mol	LiTFSI /g	SiO ₂ / g	PEA- GPS/ g	EO ₂ / mol	G-mPEG /g	EO ₃ / mol	ΣEO/ mol	Σm _{GPS} / g	Σm/ g
PEO-SNW	0.160	0.00364	0.058	0.018					0.00364		0.23600
PEO- SNWAKP6	0.103	0.00235	0.058	0.018	0.0731	0.00081	0.0388	0.00048	0.00364	0.0391	0.29136
0											
PEO- SNWAKP4	0.105	0.00238	0.058	0.018	0.0487	0.00054	0.0582	0.00072	0.00364	0.0391	0.28775
0											
PEO- SNWAKP3	0.106	0.00240	0.058	0.018	0.0365	0.00041	0.0679	0.00084	0.00364	0.0391	0.28595
0											
PEO- SNWAKP2	0.106	0.00241	0.058	0.018	0.0243	0.00027	0.0776	0.00096	0.00364	0.0391	0.28415
0											
PEO- SNWAKP1	0.107	0.00243	0.058	0.018	0.0122	0.00014	0.0873	0.00108	0.00364	0.0391	0.28235
0											

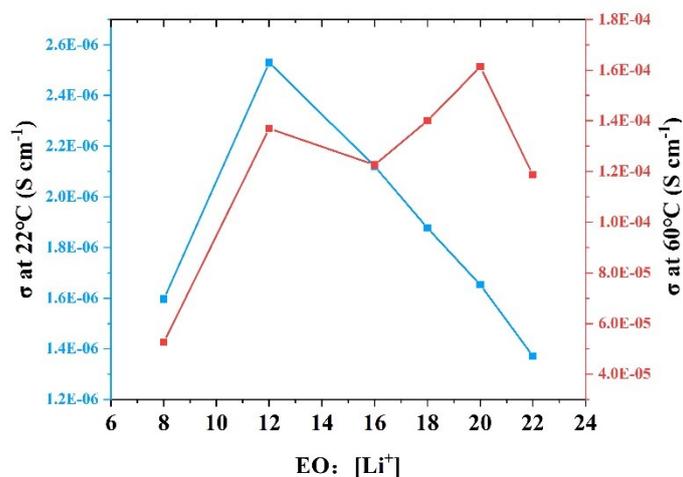
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51 **Figure S1** Conductivity-temperature curves of PEO matrix with a different amount of LiTFSI

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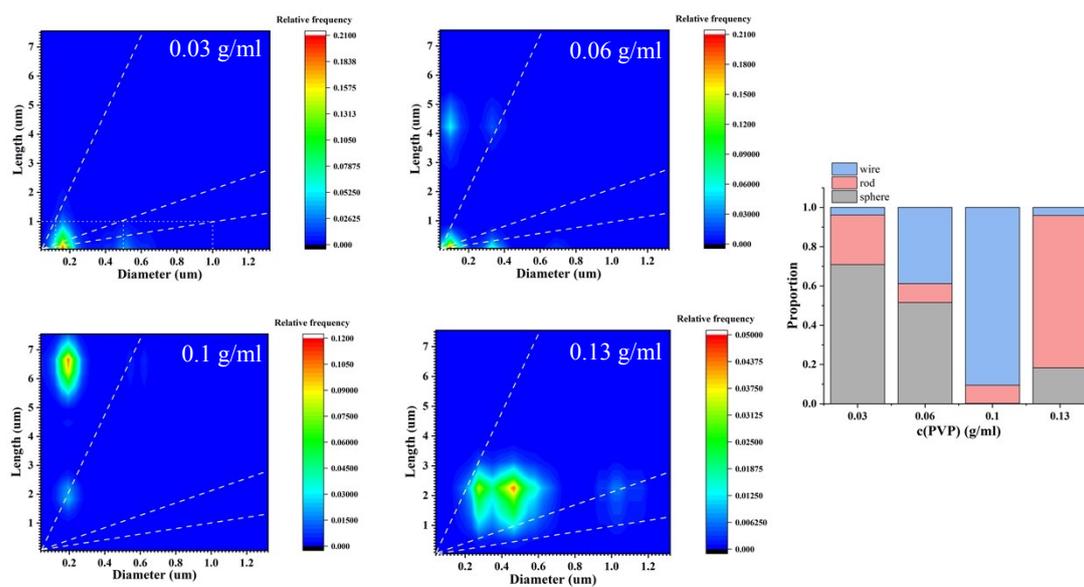
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Figure S2 Conductivity of PEO matrix with different amounts of LITFSI at 22 and 60 °C, respectively

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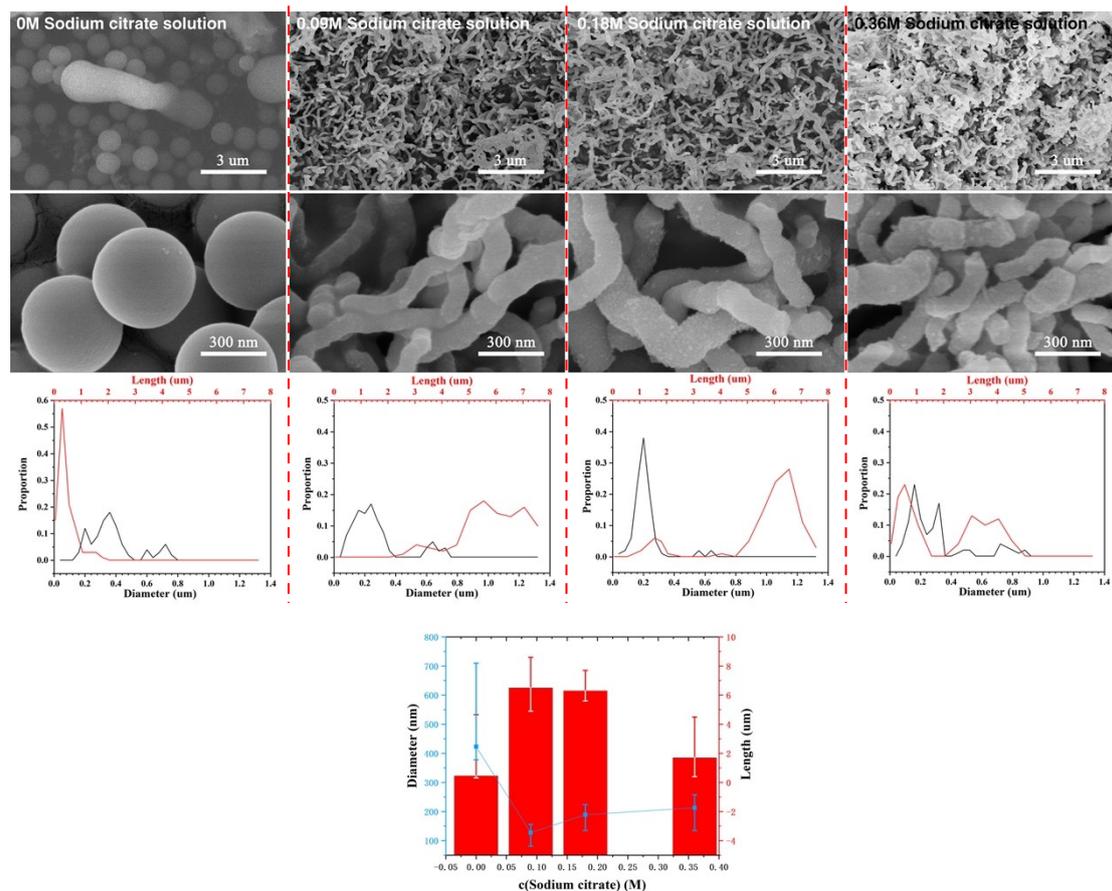


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Figure S3 The influence of the concentration of PVP on the morphology of SiO₂. The tag in the upper-right corner was the concentration of PVP. (the concentration of sodium citrate solution was fixed as 0.18 M and that of TEOS was fixed a 1:75).

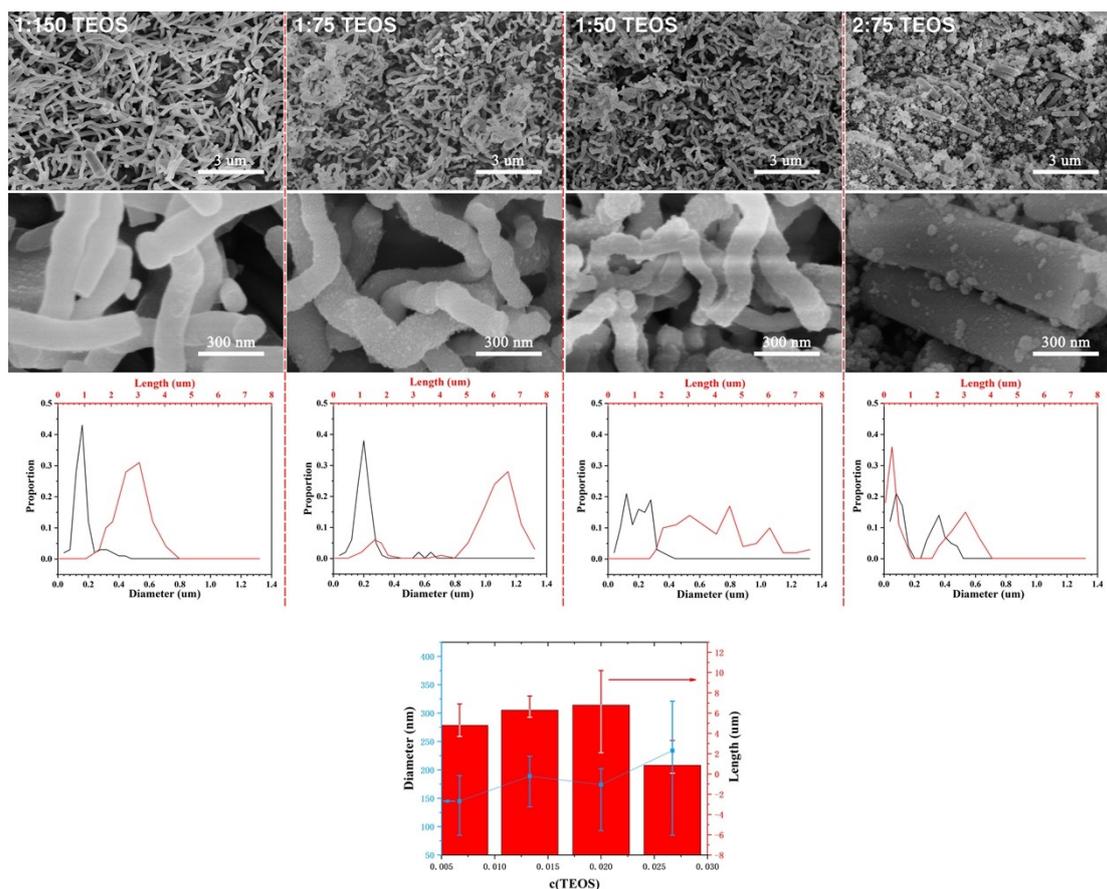
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65 **Figure S4** The influence of the concentration of sodium citrate solution on the morphology of
 66 SiO₂. The tag in the upper left corner was the concentration of sodium citrate solution. (the
 67 concentration of PVP was fixed as 0.1 g/ml and that of TEOS was fixed a 1:75)



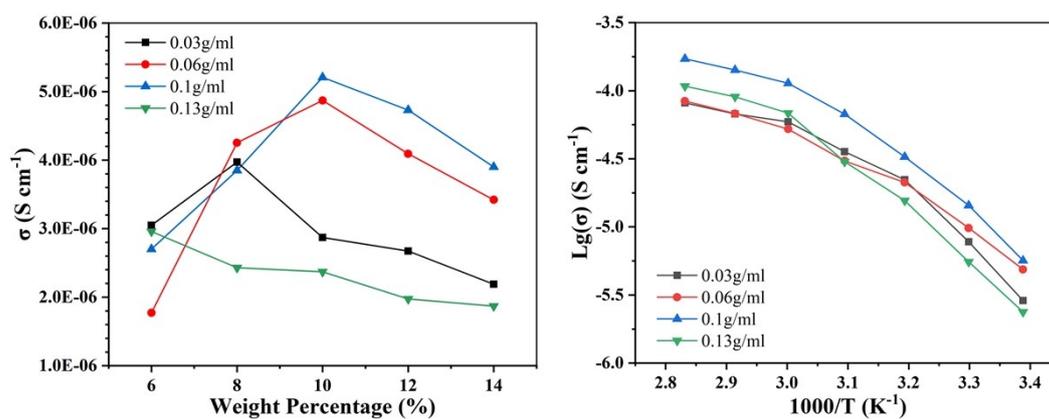
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69 **Figure S5** The influence of the concentration of sodium citrate solution on the morphology of
 70 SiO₂. The ratio in the upper left corner was the volume of TEOS to ethanol. (the concentration of
 71 sodium citrate solution was fixed as 0.18 M and that of PVP was fixed as 0.1 g/ml)

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76 **Figure S6** The ionic conductivity of PEO-SNW with various amounts of nano SiO₂ fabricated
 77 under different concentration of PVP at 22 °C and the temperature-conductivity curves of PEO-
 78 SNW with 10 wt.% nano SiO₂ fabricated under different concentration of PVP

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Table S2 VTF fitting result of different PEO-SNW

Sample	$A/S \otimes K^{1/2} \otimes c$ m^{-1}	E_a/K	$E_a/J \otimes mol^{-1}$ 1	T_0/K	$R^2(COD)$	R^2_{modify}
PEO-SNW(0.03g/ml PVP)	0.0089	297.1	2470	267.1	0.9964	0.9946
PEO-SNW(0.06g/ml PVP)	0.0226	445.2	3701	243.7	0.9977	0.9965
PEO-SNW(0.1g/ml PVP)	0.0432	388.7	3232	254.7	0.9971	0.9957
PEO-SNW(0.13g/ml PVP)	0.0999	493.7	4105	241.9	0.9939	0.9909

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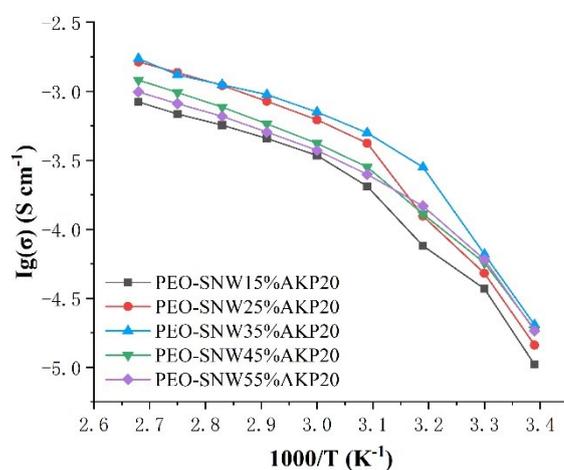
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Table S3 VTF fitting result of different samples

Sample	$A/S \otimes K^{1/2} \otimes c$ m^{-1}	E_a/K	$E_a/J \otimes mol^{-1}$ 1	T_0/K	$R^2(COD)$	R^2_{modify}
PEO	0.1662	345.8	2875	255.9	0.9824	0.9765
PEO-SNW	0.7563	451.0	3750	248.1	0.9790	0.9721
PEO-SNWAKP60	0.3227	417.9	3474	253.5	0.9625	0.9536
PEO-SNWAKP40	0.2248	250.5	2083	262.6	0.9888	0.9851
PEO-SNWAKP30	0.0623	118.6	986.1	274.9	0.9890	0.9853
PEO-SNWAKP20	0.1280	142.7	1187	271.3	0.9942	0.9922
PEO-SNWAKP10	0.1633	173.3	1441	268.7	0.9962	0.9950

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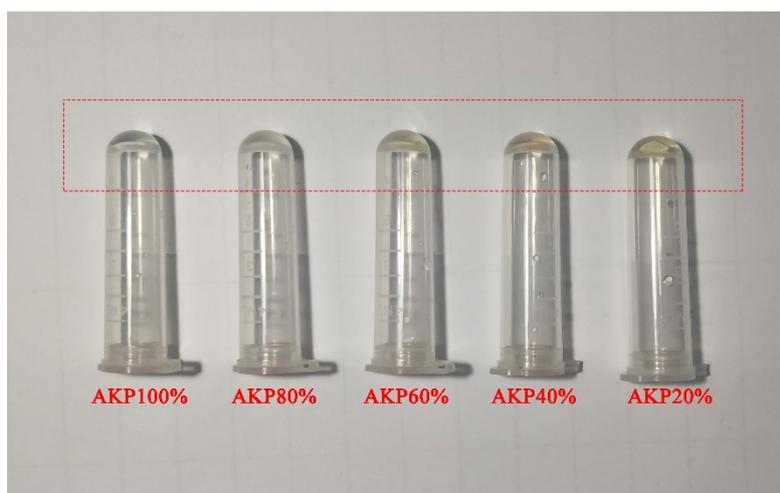


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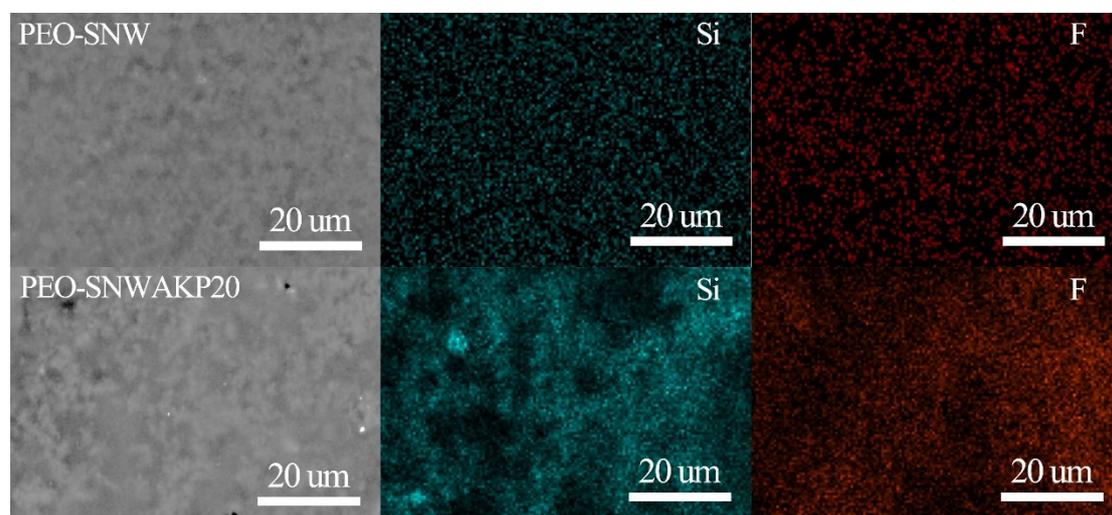
88 **Figure S7** The conductivity-temperature curves of samples with different amounts of AKP20. This
 89 ratio was set as 35 % in the manuscript, corresponding to the percentage of EO from AKP.

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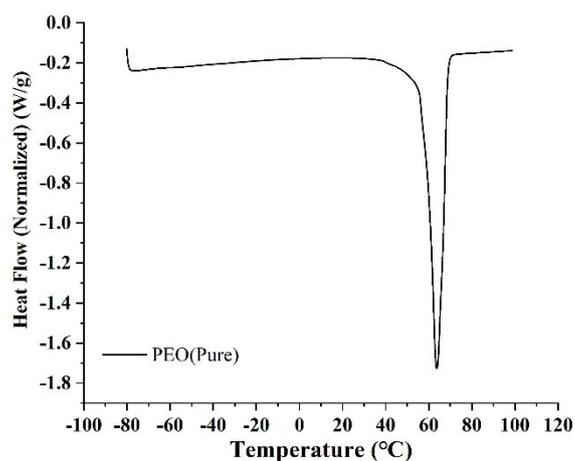
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93 **Figure S8** Bulk formed by AKP with different percentages of PEA-GPS. The percentage followed
94 is the addition amount of PEA -GPS.
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97 **Figure S9** SEM image of the surface of PEO-SNW, PEO-SNWAKP and the followings are Si, F
98 mapping of the corresponding area.
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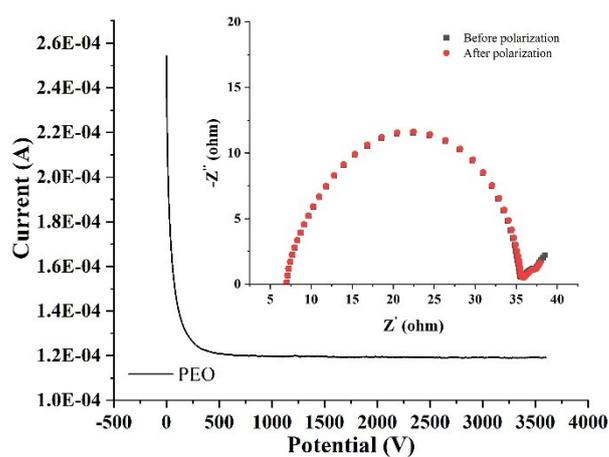


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Figure S10 DSC curve of pure PEO without adding LITFSI

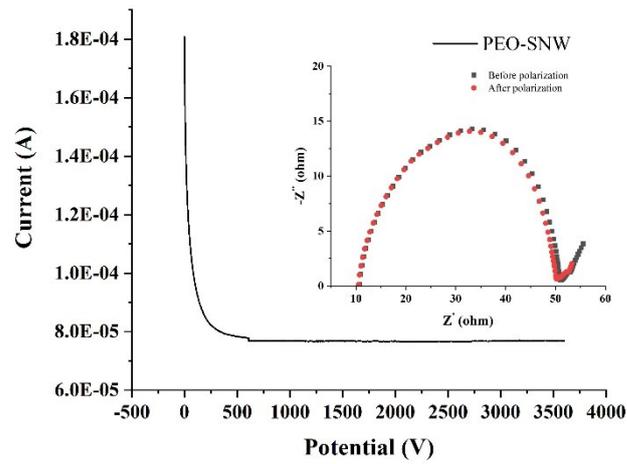
Table S4 Calculation result of different samples

Sample	$T_g/^\circ\text{C}$	$T_m/^\circ\text{C}$	$S_{m\text{-peak}}$	$m_{\text{PEO}}/\%$	$\chi_c/\%$
PEO(Pure)		64.2	12.3415	1	100
PEO	-47.2	51.2	4.45768	73.40	49.12
PEO-SNW	-48.5	50.7	4.02351	67.80	48.08
PEO-SNWAKP60	-58.8	32.4	1.51702	35.35	34.77
PEO-SNWAKP40	-64.2	41.2	2.71931	36.49	60.38
PEO-SNWAKP30	-64.7	40.6	2.58648	37.07	56.53
PEO-SNWAKP20	-64.6	40.5	2.39113	37.30	51.94
PEO-SNWAKP10	-49.4	39.8	2.06144	37.90	44.07



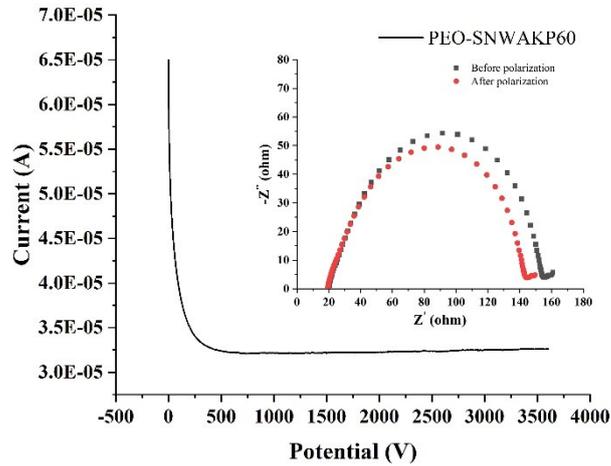
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Figure S11 i-t and corresponding IMP curves of blank PEO sample at 70 °C



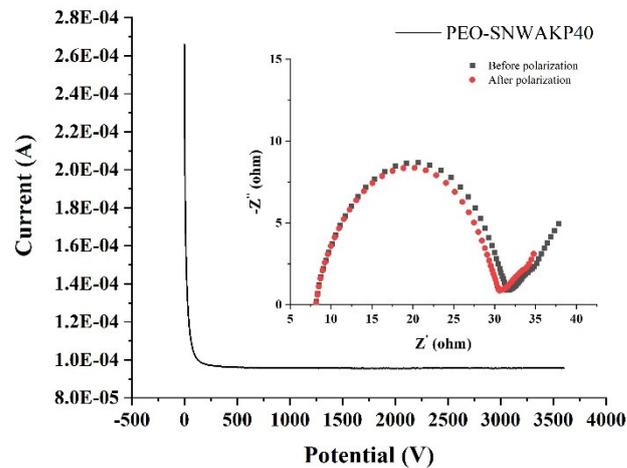
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Figure S12 i-t and corresponding IMP curves of PEO-SNW at 70 °C



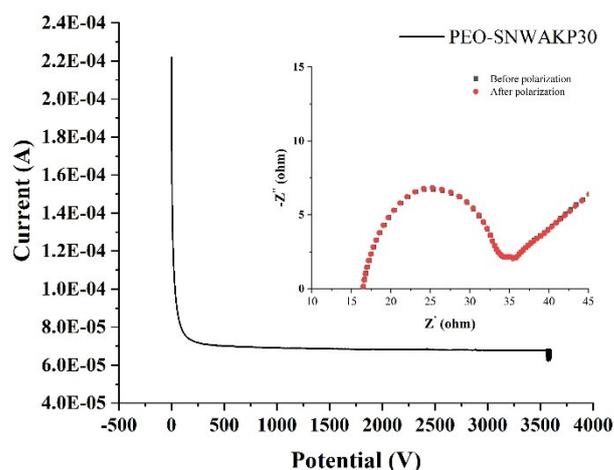
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Figure S13 i-t and corresponding IMP curves of PEO-SNWAKP60 at 70 °C



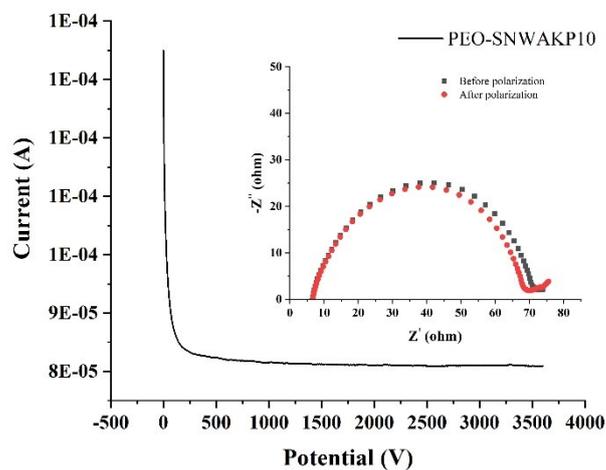
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Figure S14 i-t and corresponding IMP curves of PEO-SNWAKP40 at 70 °C



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Figure S15 i-t and corresponding IMP curves of PEO-SNWAKP30 at 70 °C



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Figure S16 i-t and corresponding IMP curves of PEO-SNWAKP10 at 70 °C

Table S5 t^+ of the representatives of different samples

Sample	V/V	I_0/A	I_s/A	R_b^0/Ω	R_b^s/Ω	R_i^0/Ω	R_i^s/Ω	t^+
PEO	0.01	2.54E-4	1.19E-4	7.188	7.139	28.45	28.65	0.1985
PEO-SNW	0.01	1.81E-4	7.69E-5	10.8	10.56	40.38	37.79	0.1648
PEO-SNWAKP60	0.01	6.50E-5	3.26E-5	19.66	19.05	127.2	122	0.1487
PEO-SNWAKP40	0.01	2.66E-4	9.57E-5	8.298	8.278	22.96	22	0.1778
PEO-SNWAKP30	0.01	2.22E-4	6.78E-5	16.44	16.59	18.24	17.69	0.2046
PEO-SNWAKP10	0.01	1.35E-4	8.09E-5	6.678	6.805	64.56	62.29	0.1553

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