

Supporting Information

Anchoring $\text{Zn}_{0.5}\text{Cd}_{0.5}\text{S}$ solid solution onto 2D porous Co-CoO nanosheets for highly improved photocatalytic H_2 generation

Xueyou Gao, Deqian Zeng*, Qingru Zeng, Zongzhuo Xie, Toyohisa Fujita, Xinpeng Wang, Gang He and Yuezhou Wei

Guangxi Key Laboratory of Processing for Non-ferrous Metals and Featured Materials, School of Resources, Environment and Materials, Guangxi University, Nanning 530004, China

*Corresponding author. Email: dqzeng@gxu.edu.cn

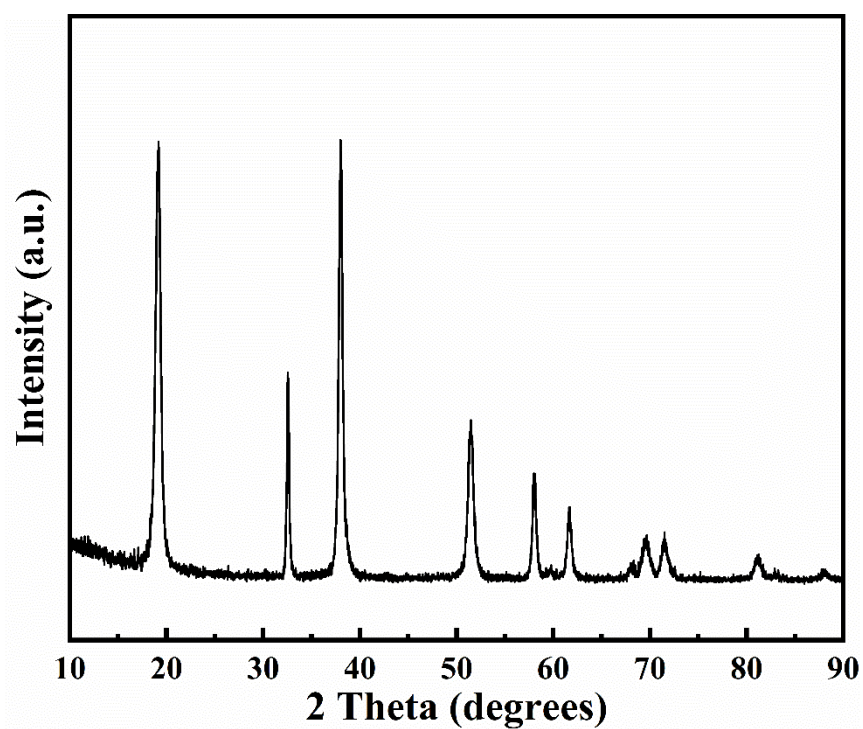


Fig. S1 XRD patterns of Co(OH)₂ nanosheets.

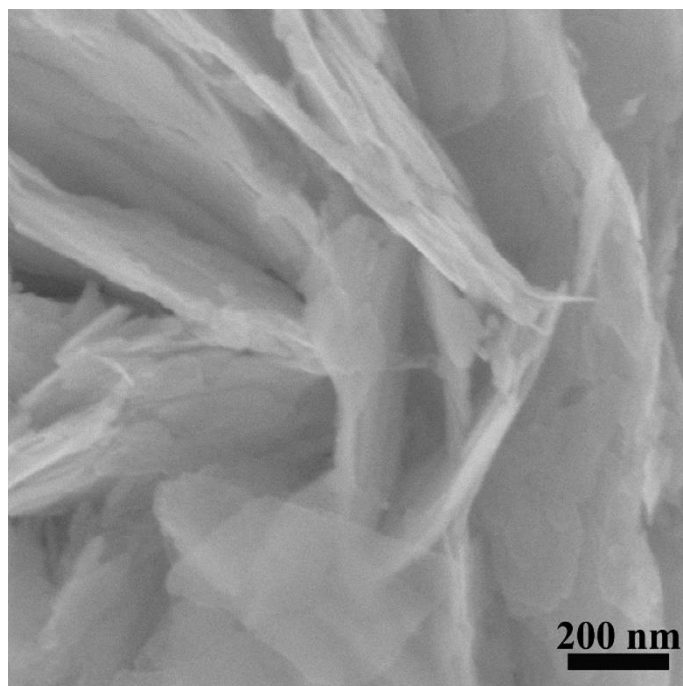


Fig. S2 SEM image of CC nanosheets.

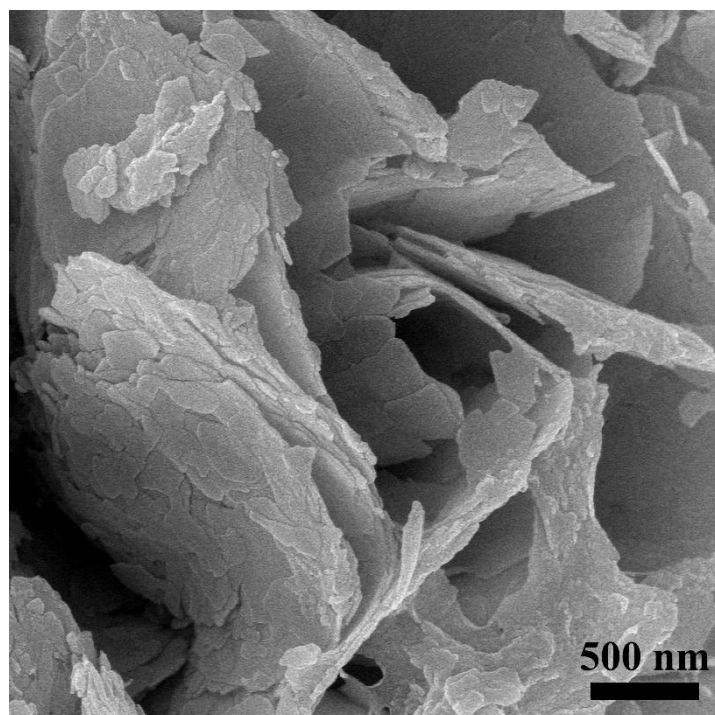


Fig. S3 SEM image of Co(OH)_2 nanosheets.

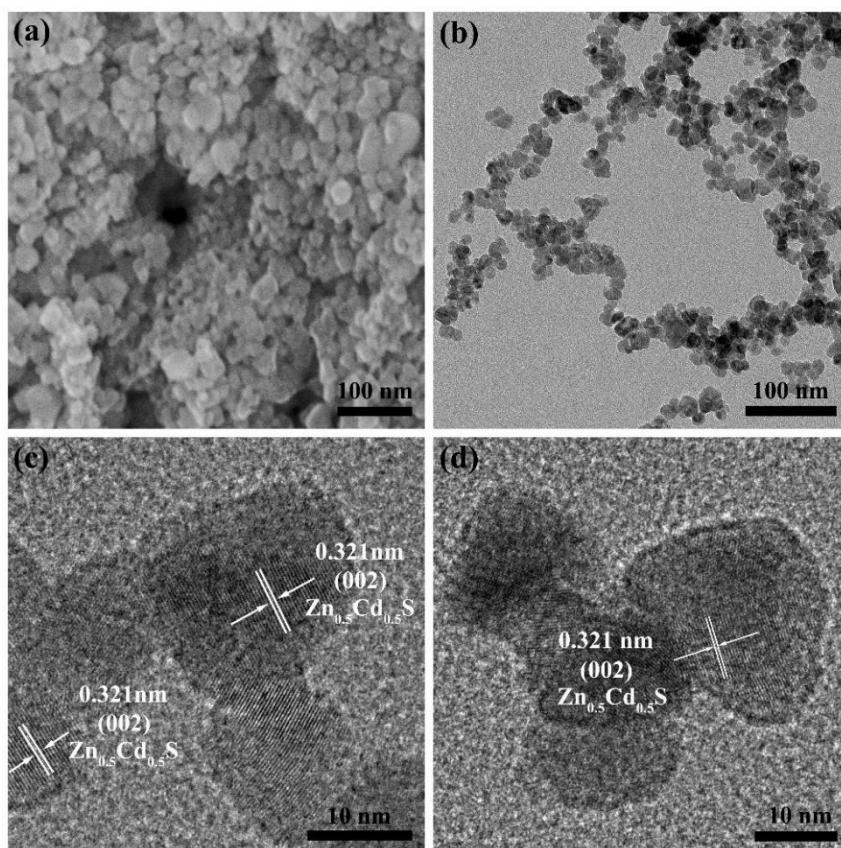


Fig. S4 (a) SEM image, (b) TEM image and (c and d) HRTEM image of $\text{Zn}_{0.5}\text{Cd}_{0.5}\text{S}$.

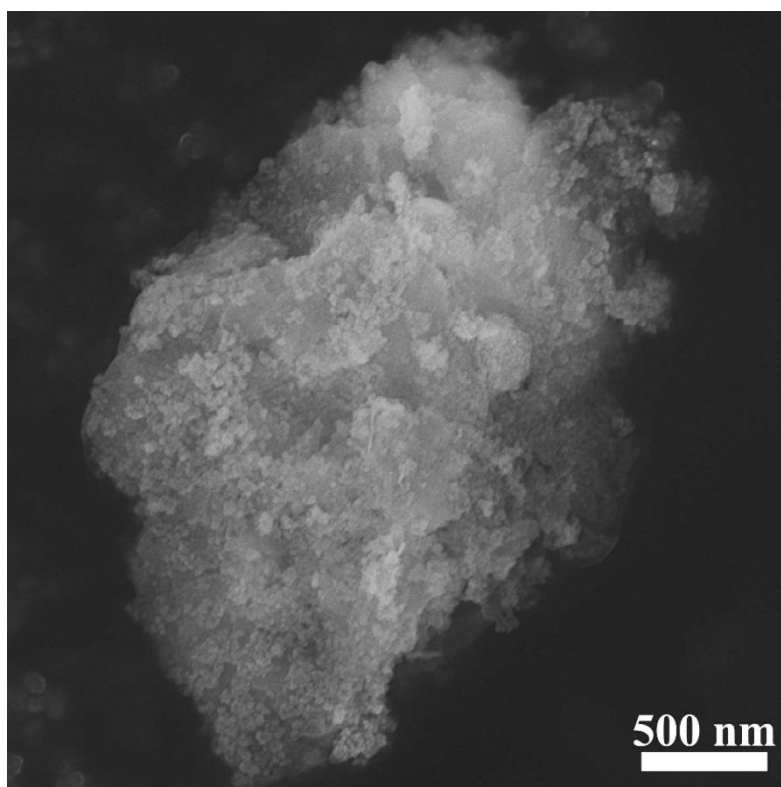


Fig. S5 SEM image of 5CC/ZCS sample

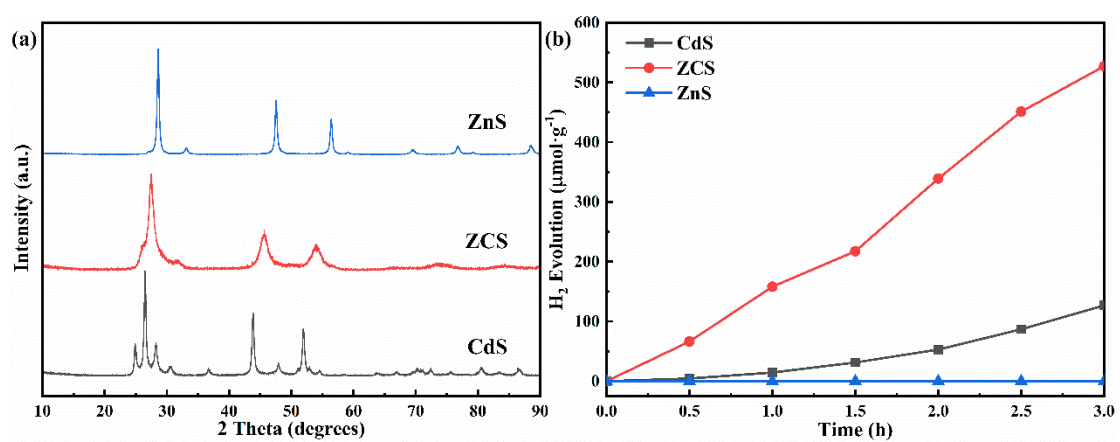


Fig. S6 XRD patterns (a) and photocatalytic activities (b) of ZnS, ZCS and CdS samples.

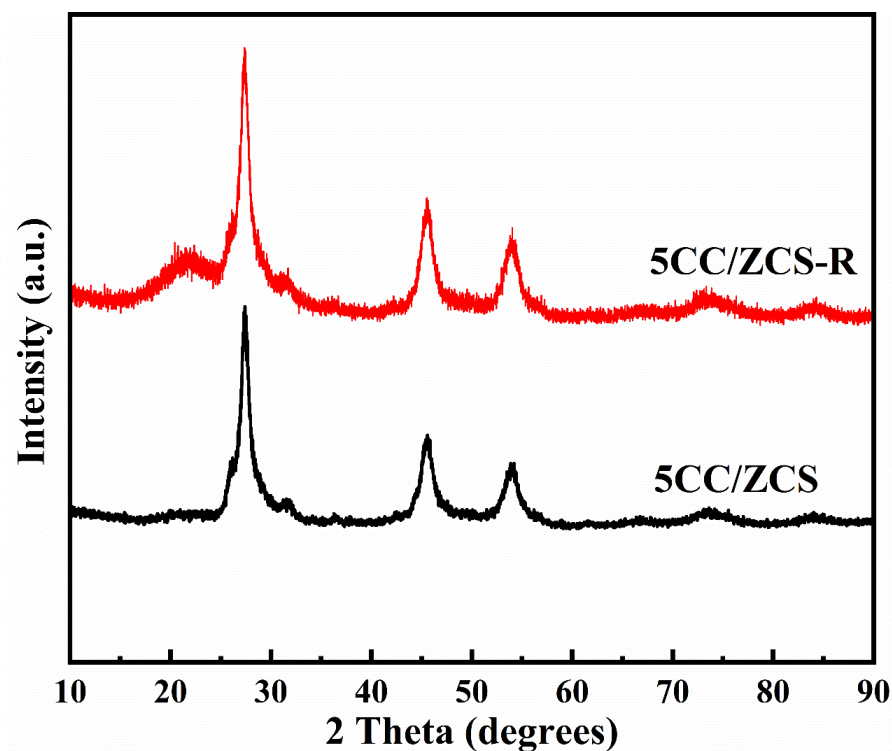


Fig. S7 XRD patterns of 5CC/ZCS before (5CC/ZCS) and after (5CC/ZCS-R) the long-time (24 h) photocatalytic reaction.

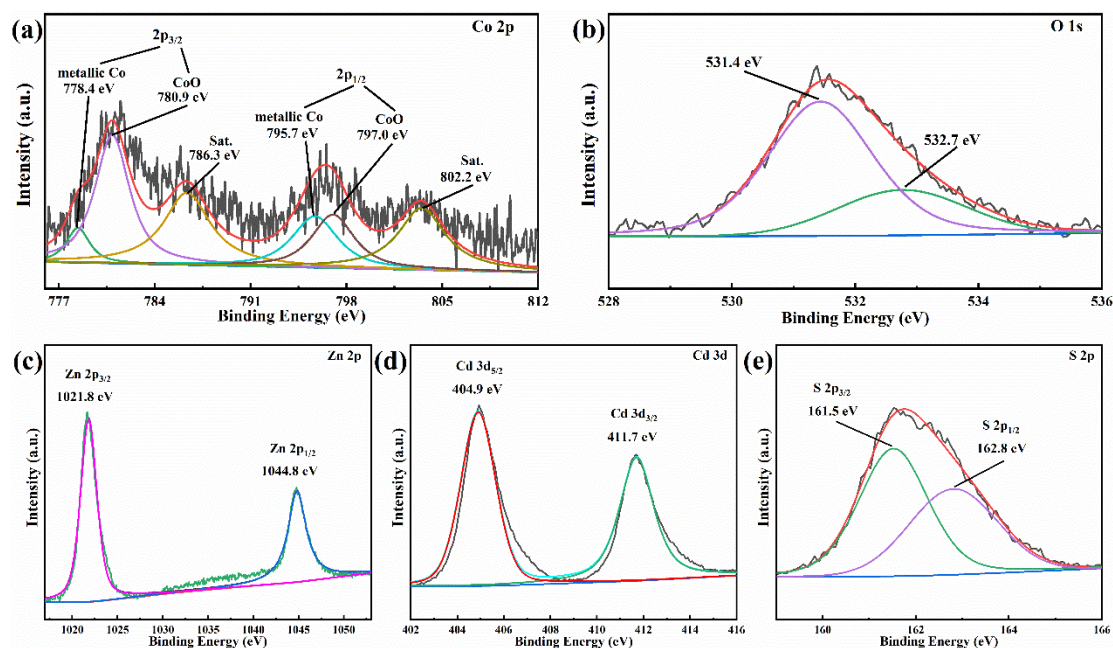


Fig. S8 High-resolution XPS spectra of (a) Co 2p, (b) O 1s, (b) Zn 2p, (d) Cd 3d and (e) S 2p for 5CC/ZCS after the long-time (24 h) photocatalytic reaction.

Table S1 Photocatalytic H₂ evolution activity of Zn_xCd_{1-x}S-based photocatalysts.

	Samples	Cocatalyst	Light source	Scaveng er	Activity ($\mu\text{mol h}^{-1} \text{g}^{-1}$)	Enhancement factor* (Reference)	Refs
1	Zn _{0.5} Cd _{0.5} S	MoS ₂ /RGO	/	Lactic acid	2310	—	S1
2	Zn _{0.5} Cd _{0.5} S	MnO ₂	300 W Xe lamp ($\lambda \geq$ 420 nm)	Na ₂ S /Na ₂ SO ₃	2524	13 (Zn _{0.5} Cd _{0.5} S)	S2
3	Zn _{0.5} Cd _{0.5} S	Cu ₂ (OH) ₂ C O ₃	300 W Xe lamp ($\lambda \geq$ 420 nm)	Na ₂ S /Na ₂ SO ₃	5514	3.04 (Zn _{0.5} Cd _{0.5} S)	S3
4	Zn _{0.8} Cd _{0.2} S	Ni(OH) ₂	300 W Xe lamp ($\lambda \geq$ 420 nm)	Triethan olamine	7160	25 (Zn _{0.8} Cd _{0.2} S) 1.2 (Pt/Zn _{0.8} Cd _{0.2} S)	S4
5	Zn _{0.25} Cd _{0.75} S	Ni(OH) ₂	300 W Xe lamp	Na ₂ S /Na ₂ SO ₃	3774	5 (CdS)	S5
6	ZnCdS	Fe _{0.3} Pt _{0.7}	300 W Xe lamp ($\lambda >$ 420 nm)	Na ₂ S /Na ₂ SO ₃	2265	3 (ZnCdS) 1.4 (Pt/ZnCdS)	S6
7	Zn _{0.5} Cd _{0.5} S	Pt	500 W Xe lamp ($\lambda >$ 400 nm)	Ascorbic acid	5500	27.9 (Zn _{0.5} Cd _{0.5} S)	S7
8	Zn _{0.5} Cd _{0.5} S	Pt	500 W Xe lamp ($\lambda >$ 400 nm)	Na ₂ S /Na ₂ SO ₃	3810	4.9 (Zn _{0.5} Cd _{0.5} S)	S7
9	Zn _{0.5} Cd _{0.5} S	CoPt ₃	300 W Xe lamp ($\lambda >$ 420 nm)	Na ₂ S /Na ₂ SO ₃	2340	4.7 (Zn _{0.5} Cd _{0.5} S) 1.2 (Pt/Zn _{0.5} Cd _{0.5} S)	S8
10	Zn _{0.3} Cd _{0.7} S	Ni ₃ C	300 W Xe lamp ($\lambda \geq$ 420 nm)	Na ₂ S /Na ₂ SO ₃	3310	6.4 (Zn _{0.3} Cd _{0.7} S)	S9
11	Zn _x Cd _{1-x} S	Ni	300 W Xe lamp	Glycerol	2253	—	S10
12	Zn _{0.5} Cd _{0.5} S	Ni nanosheets	300 W Xe lamp ($\lambda >$ 420 nm)	Na ₂ S /Na ₂ SO ₃	5930	33.7 (Zn _{0.5} Cd _{0.5} S) 15.4 (Pt/Zn _{0.5} Cd _{0.5} S)	S11
13	ZnCdS	NiB	LED lamp	Lactic acid	8137	17 (ZnCdS)	S12

14	Zn _{0.5} Cd _{0.5} S	Co-CoO nanosheets	300 W Xe lamp ($\lambda >$ 420 nm)	Na ₂ S /Na ₂ SO ₃	8152	46 (Zn _{0.5} Cd _{0.5} S) 10 (3%Pt/Zn _{0.5} Cd _{0.5} S)	This work
----	---------------------------------------	----------------------	---	---	------	---	--------------

Table S2 Decay parameters of pristine ZCS and 5CC/ZCS.

Samples	Lifetime, T (ns)	Rel (%)	T _{ave} (ns)
ZCS	T1=29.17	B1=12.17	2.69
	T2=242.84	B2=63.51	
	T3=0.67	B3=24.31	
5CC/ZCS	T1=27.32	B1=12.47	3.20
	T2=225.83	B2=67.54	
	T3=0.65	B3=19.99	

References

- S1. S. N. Guo, Y. L. Min, J. C. Fan and Q. J. Xu, *ACS Appl. Mater. Interfaces*, 2016, **8**, 2928-2934.
- S2. H. Du, X. Xie, Q. Zhu, L. Lin, Y.-F. Jiang, Z.-K. Yang, X. Zhou and A.-W. Xu, *Nanoscale*, 2015, **7**, 5752-5759.
- S3. Y. Liu, H. Ren, H. Lv, Z. Gong and Y. Cao, *Appl. Surf. Sci.*, 2019, **484**, 1061-1069.
- S4. J. Ran, J. Zhang, J. Yu and S. Z. Qiao, *ChemSusChem*, 2014, **7**, 3426-3434.
- S5. Y. Xu, Y. Gong, H. Ren, W. Liu, C. Li, X. Liu and L. Niu, *J. Alloy Compd.*, 2018, **735**, 2551-2557.
- S6. D. Shu, H. Wang, Y. Wang, Y. Li, X. Liu, X. Chen, X. Peng, X. Wang, P. Ruterana and H. Wang, *Int. J. Hydrogen Energ.*, 2017, **42**, 20888-20894.
- S7. B.-J. Ng, L. K. Putri, X. Y. Kong, K. P. Y. Shak, P. Pasbakhsh, S.-P. Chai and A. R. Mohamed, *Appl. Catal., B*, 2018, **224**, 360-367.
- S8. H. Wang, Y. Li, D. Shu, X. Chen, X. Liu, X. Wang, J. Zhang and H. Wang, *Int. J. Energ. Res.*, 2016, **40**, 1280-1286.
- S9. Z. Luo, X. Zhao, H. Zhang and Y. Jiang, *Appl. Catal., A*, 2019, **582**, 117115.
- S10. R. Gao, B. Cheng, J. Fan, J. Yu and W. Ho, *Chin. J. Catal.*, 2021, **42**, 15-24.
- S11. X. Gao, J. Yang, D. Zeng, G. He, C. Dai, Y. Bao and Y. Wei, *J. Alloy Compd.*, 2021, **871**, 159460.
- S12. L. Song, S. Zhang, D. Liu, S. Sun and J. Wei, *Int. J. Hydrogen Energ.*, 2020, **45**, 8234-8242.